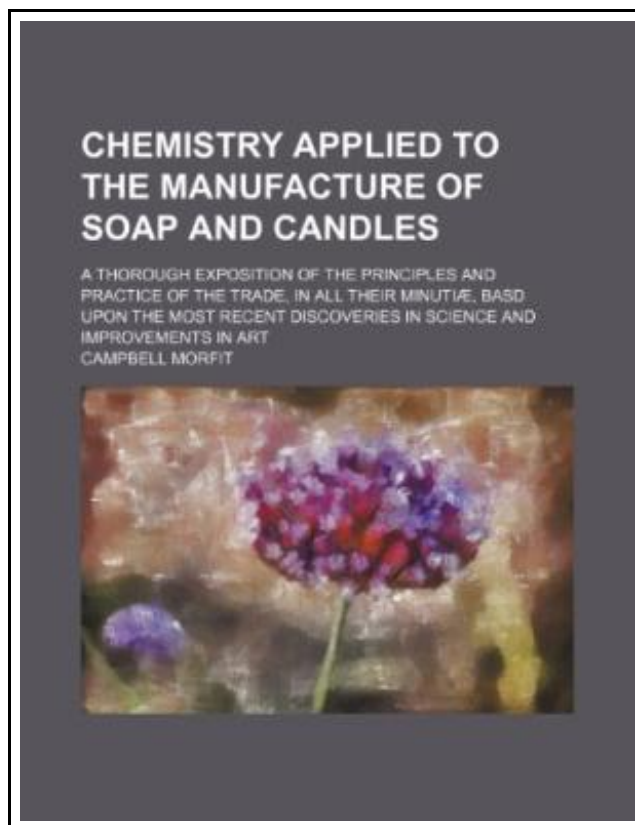


# Chemistry Applied to the Manufacture of Soap and Candles; A Thorough Exposition of the Principles and Practice of the Trade, in All Their Minutiae, Ba (Paperback)



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## ***Reviews***




*It is one of the most popular books. I am quite late in starting to read this one, but better than never. Once you begin to read the book, it is extremely difficult to leave it before concluding.*  
*(Camille Larson)*

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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1847 Excerpt: .rise to the surface, and must be collected upon a filter, and when well washed and dried, are carefully weighed, deducting that of the filter. On the other hand, unite the wash waters, evaporate them to dryness, and take the weight of the residue, which indicates, in the form of a chloride salt, the quantity of alkali contained by the soap. The difference between the two weights, subtracted from that of the soap employed, expresses the quantity of water. The marbled soap, well prepared, should give by analysis, per 100 parts, six of soda, (expressed by ten parts of chloride salt--as obtained in that state by evaporation, ) sixty-four parts fatty acid, and thirty parts water. The white soap must yield 4.6 of soda, or in the form of chloride 7.5 parts; fatty acids 50.12 parts; water 45.2 parts. By comparing the results obtained, with the above, it will be easy to determine the fraud. The first method, however, being the most simple, practicable and expeditious, is much the most preferable for general use in the arts and commerce. To estimate the amount of foreign matter, after having weighed ten grammes of the soap, previously reduced to thin shavings, treat them with boiling alcohol. If the soap be white and without admixture, the portion remaining undissolved is very minute, and a mottled soap of good quality does not leave, when operating on ten grammes, more than ten centigrammes, or about one per cent. If, however, there...

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