

Lagrangian particles evolution through RRE refraction structure

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Outline

- 1 Reminder of the different refraction structures
- 2 Regular Refraction with reflected Expansion (RRE)
- 3 Theory and equations : gas dynamics
- 4 Results given by the inert gas dynamics theory
- 5 Use of CHEMKIN II to compute chemistry calculus
- 6 Results of the chemical calculus

Reminder of the different refraction structures

From Henderson 1976 and 1978

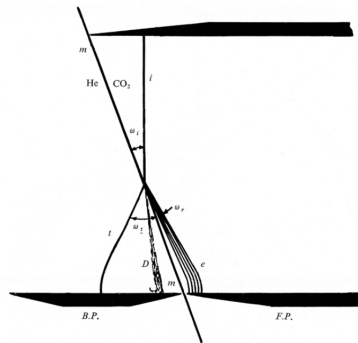
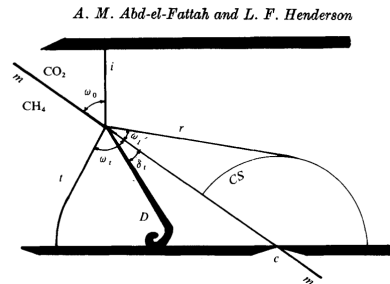


FIGURE 9. Regular refraction of a plane shock at a contaminated carbon dioxide-helium interface. For symbols see caption to figure 3.

(a) RRE struture



(b) RRR struture

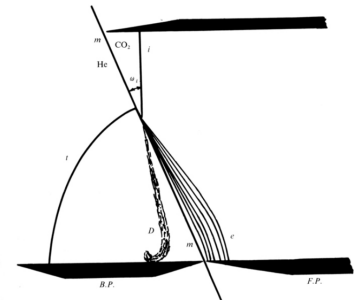


FIGURE 12. Bound-precursor irregular refraction of a plane shock at a contaminated carbon dioxide-helium interface. For symbols see captions to figures 2 and 10.

(c) BPR structure

Figure 1: Three of the already known refraction structures : RRE and BPR schemes are from Henderson, Abd-el-Fattah & Lozzi 1976; RRR scheme is from Henderson & Abd-el-Fattah 1978.

Reminder of the different refraction structures

Refraction with reflected Expansion

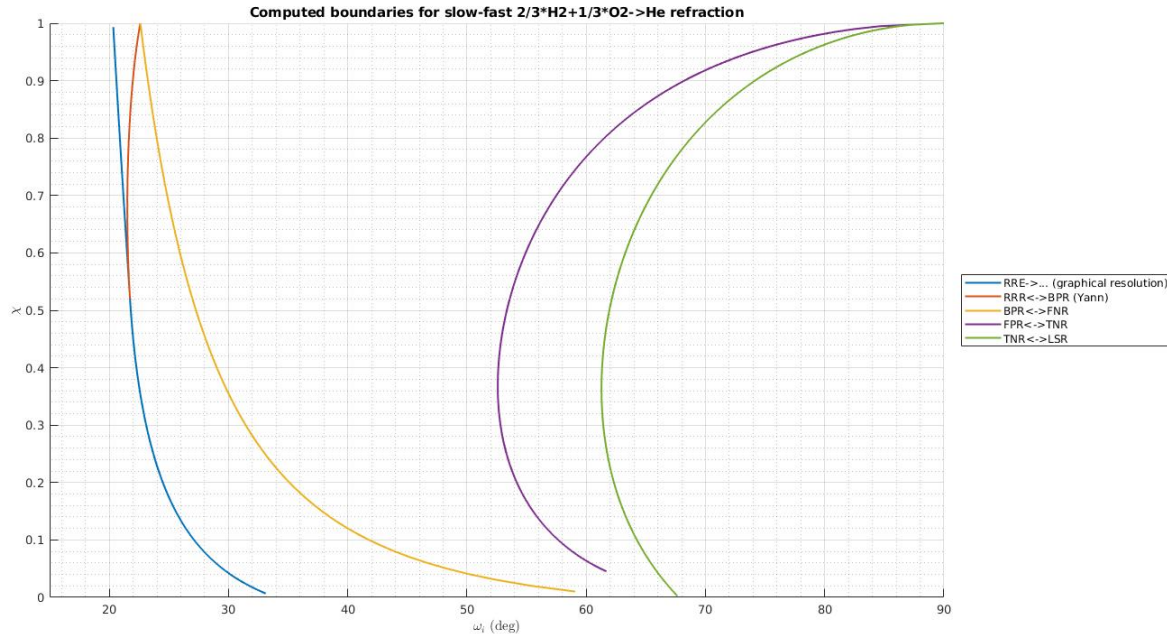


Figure 2: Boundaries of the different structures in the $\chi - \omega$ plane, for a $H_2 - O_2 // He$ system

Strength of the shock χ is related to Mach number of the shock M_{sh} .

Regular Refraction with reflected Expansion (RRE)

Relation between strength χ and Mach number M_{sh}

$$\chi = 1/\xi_i$$
$$\xi_i = \frac{1 - \gamma_I + 2\gamma_I M_{sh}}{\gamma_I + 1}$$

where γ_I is the heat ratio of phase I and M_{sh} is the normal component of the Mach number of the shock (see figure 5, slide 10). It is equal to M_{1n} the normal component of the Mach number of the incident flow. It is related to the Mach number of the flow by the following relation :

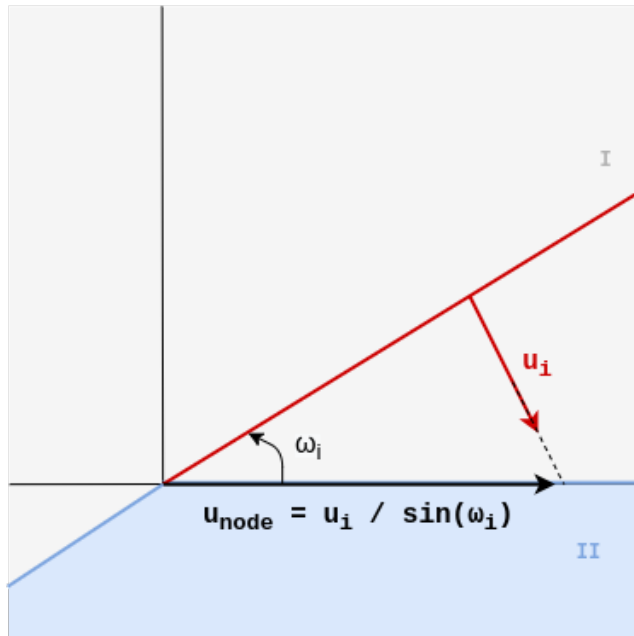
$$M_{sh} = \frac{M_1}{\sin(\omega_i)}$$

which finally leads to (with $\omega_i = 14.5^\circ$):

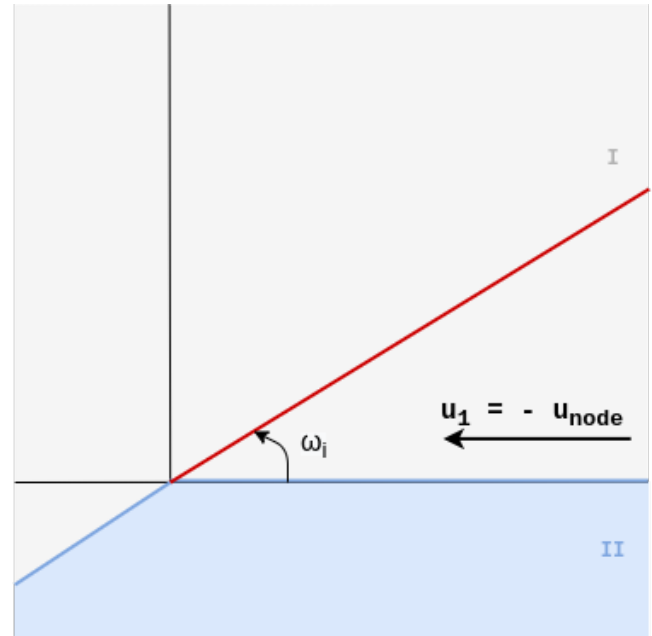
$$\chi \in [0.01; 1] \iff M_{sh} \in [1; 2.95] \iff M_1 \in [4; 11.78] \quad (1)$$

Theory and equations : gas dynamics

Change and rotation of frame of reference



(a) Frame of reference of the tube : shock is moving at u_i



(b) Frame of reference of the shock : flow is moving at u_1

Figure 3: Change of frame of reference (see legend on next slide)

Theory and equations : gas dynamics

Useful symbols

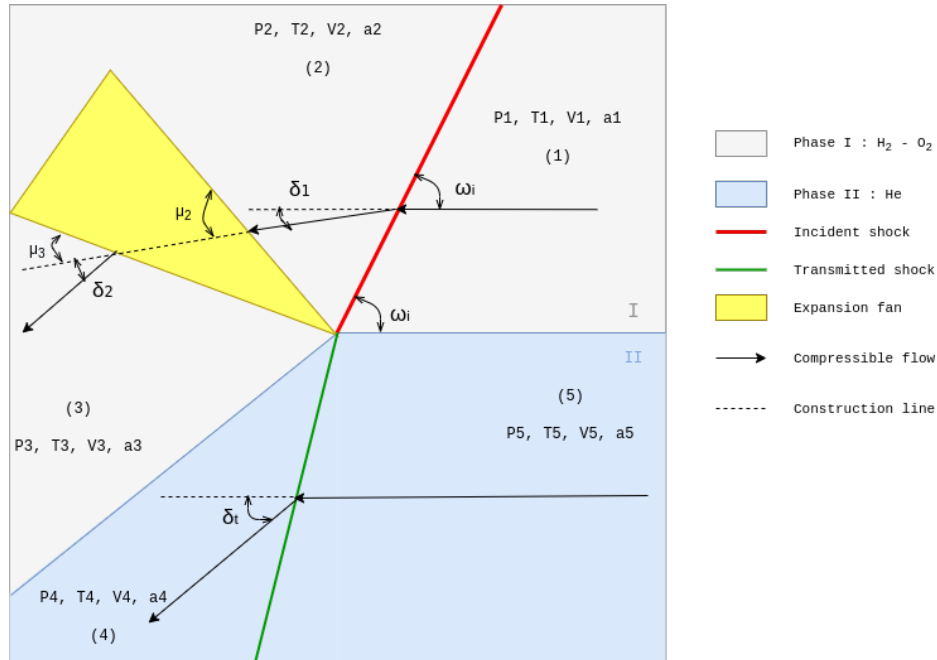


Figure 4: Symbols, zones and angles for computation

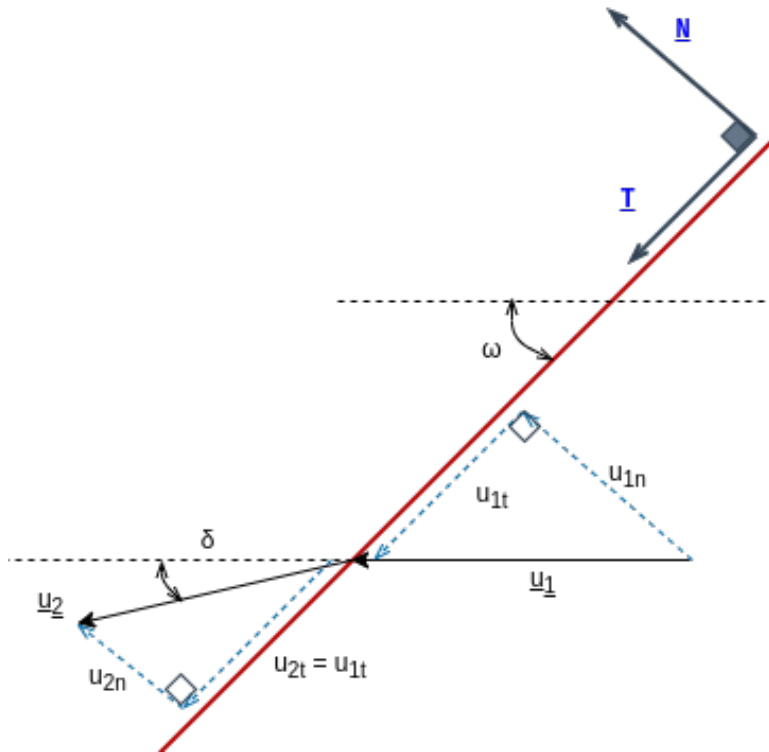
Theory and equations : gas dynamics

Initial conditions in zone (1)

Measure (unit)	Symbol	Value
Pressure (Pa)	$P1$	101 325
Temperature (K)	$T1$	600
Specific Volume	$V1$	$V1 = R_I * T1 / P1$
Mach number	$M1$	6
Angle of incidence (deg)	ω_i	14.5
Normal mach number	$M1_n$	$M1_n = M1 \sin(\omega_i)$
Speed of sound (m/s)	a_1	$a_1 = \sqrt{\gamma_I R_I T1}$

Table 1: Initial conditions

Definition of normal Mach number



$$\begin{aligned} M1_n &= \frac{u1_n}{a_1} \\ &= \frac{u1 \sin(\omega_i)}{a_1} \\ &= M1 \sin(\omega_i) \end{aligned}$$

$$\xi_i = \frac{1-\gamma_I+2\gamma_I M_{1n}}{\gamma_I+1}$$

Figure 5: Geometry associated with the oblique shock

Theory and equations : gas dynamics

From zone (1) to zone (2): oblique shock

$$\begin{aligned} M1_n &= M1 \sin(\omega_i) & \frac{V2}{V1} &= \frac{(\gamma_I+1)M1_n^2}{2+(\gamma_I-1)M1_n^2} \\ M2_n^2 &= \frac{1+\frac{\gamma_I-1}{2}M1_n^2}{\gamma_I M1_n^2 - \frac{\gamma_I-1}{2}} & \frac{P2}{P1} &= 1 + \frac{2\gamma_I}{\gamma_I+1}(M1_n^2 - 1) \\ M2 &= \frac{M2_n}{\sin(\omega_i - \delta_1)} & \frac{T2}{T1} &= \frac{P2}{P1} \frac{V2}{V1} \end{aligned}$$

where δ_1 is the angle of deflection behind the incident shock.

$$\tan(\delta_1) = 2 \cot(\omega_i) \frac{M1_n^2 - 1}{M1^2(\gamma_I + \cos 2\omega_i) + 2}$$

Theory and equations : gas dynamics

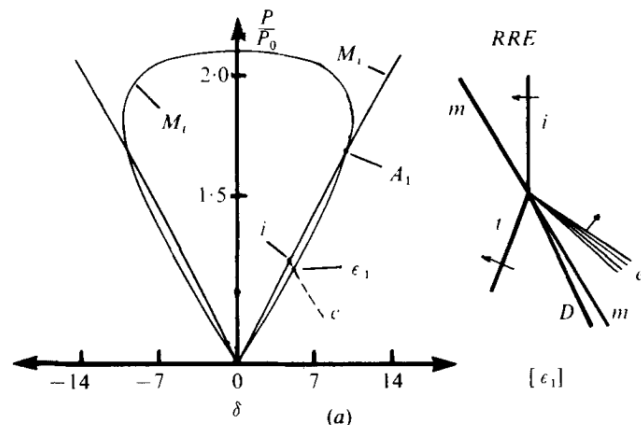
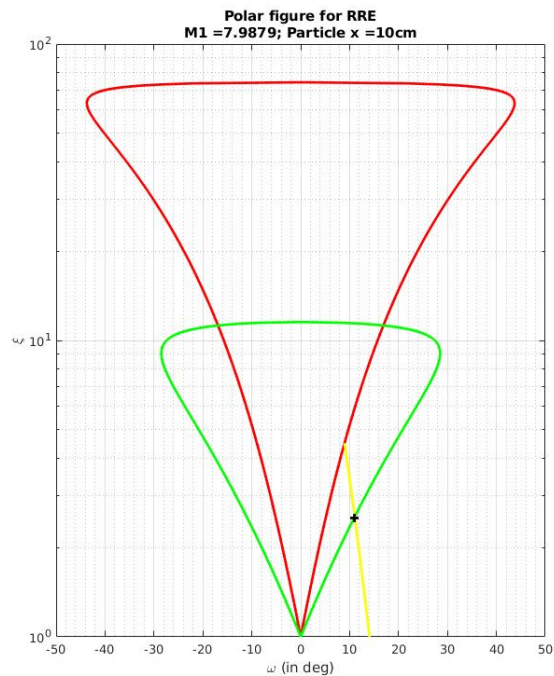
From zone (5) to zone (4) : oblique shock of unknown angle

$$\begin{aligned} M5_n &= M5 \sin(\omega_t) & \frac{V4}{V5} &= \frac{(\gamma_{II}+1)M5_n^2}{2+(\gamma_{II}-1)M5_n^2} \\ M4_n^2 &= \frac{1+\frac{\gamma_{II}-1}{2}M5_n^2}{\gamma_{II}M5_n^2-\frac{\gamma_{II}-1}{2}} & \frac{P4}{P5} &= 1 + \frac{2\gamma_{II}}{\gamma_{II}+1}(M5_n^2 - 1) \\ M4 &= \frac{M4_n}{\sin(\omega_t-\delta_t)} & \frac{T4}{T5} &= \frac{P4}{P5} \frac{V4}{V5} \end{aligned}$$

where δ_t is the angle of deflection behind the transmitted shock.
Unfortunately, ω_t , the angle between the transmitted shock and the flow in zone (5), remains unknown.

Theory and equations : gas dynamics

Membrane equilibrium between zone (4) and zone (3)



$$\delta_t = \delta_1 + \delta_2$$

$$P3 = P4$$

Figure 6: Polars for RRE structure, reference from Henderson et al. 1978 on the right

Theory and equations : gas dynamics

From zone (2) to zone (3) : Prandtl-Meyer expansion

Thanks to the polars of the shock, δ_t and ξ_t can be determined (intersection of expansion and transmitted shock polar). $P4$ thus $P3$ are known and so $M3$ thanks to Prandtl-Meyer relations.

ν is the Prandtl-Meyer function, depending on the heat ratio of the gas at stake.

$$\nu(M3) = \delta_2 + \nu(M2)$$

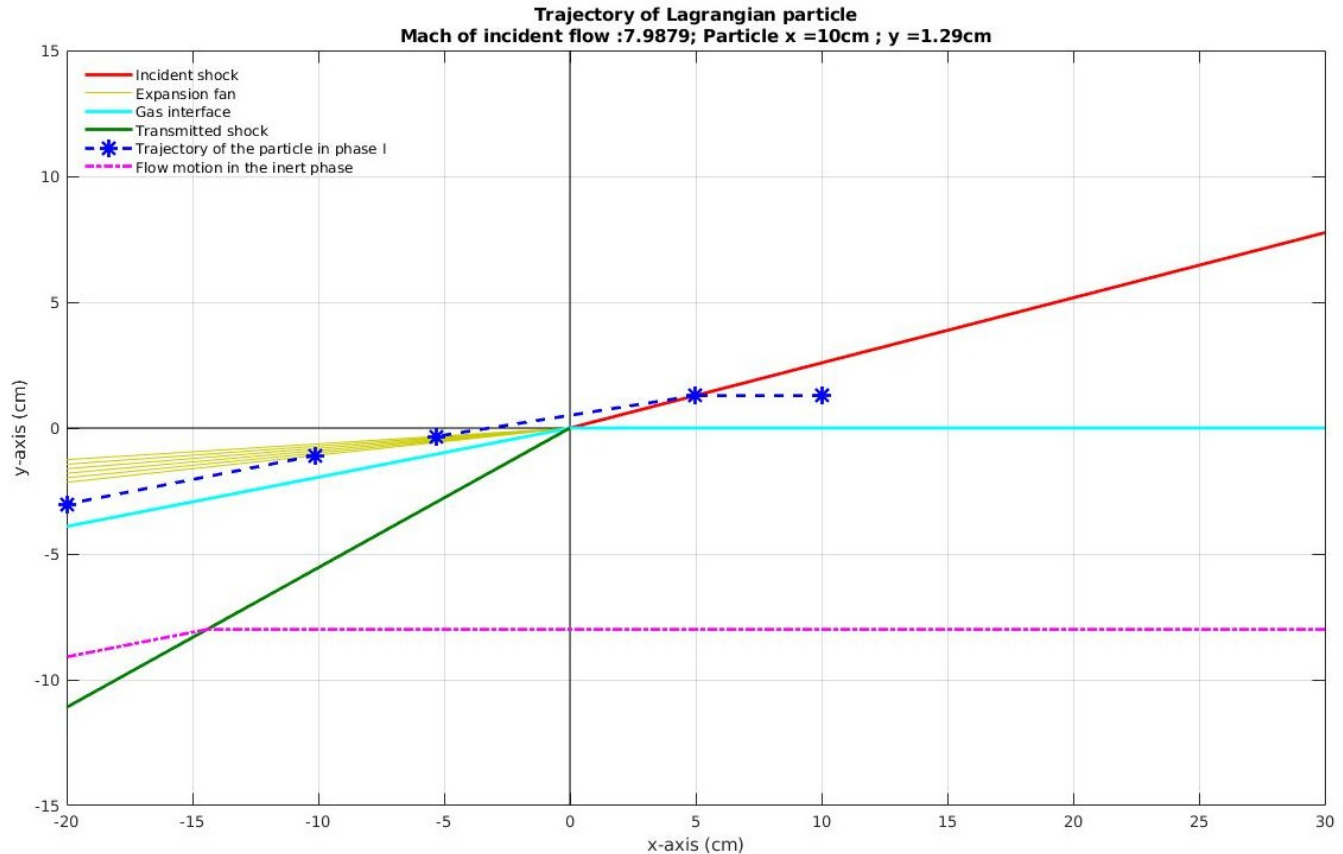
$$P3 = P4 = P5 \times \xi_t$$

$$T3 = T2 \left(\frac{P3}{P2} \right)^{\frac{\gamma_I - 1}{\gamma_I}}$$

$$V3 = R_I \frac{T3}{P3}$$

Results given by the inert gas dynamics theory

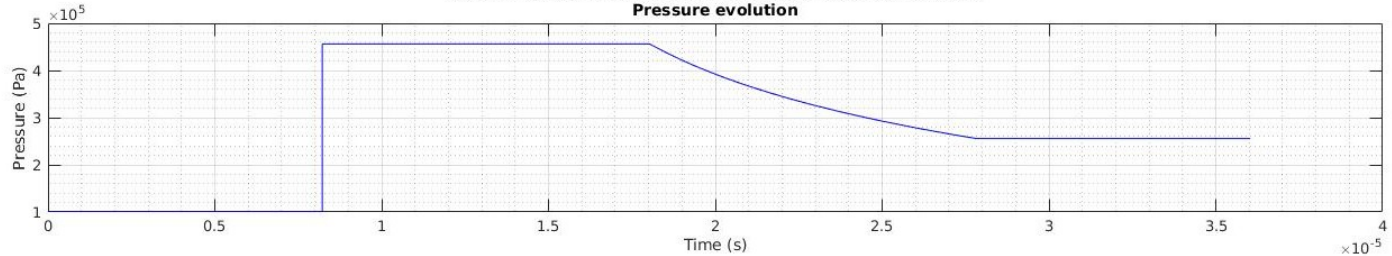
Evolution of a Lagrangian particle in the $H_2 - O_2$ phase



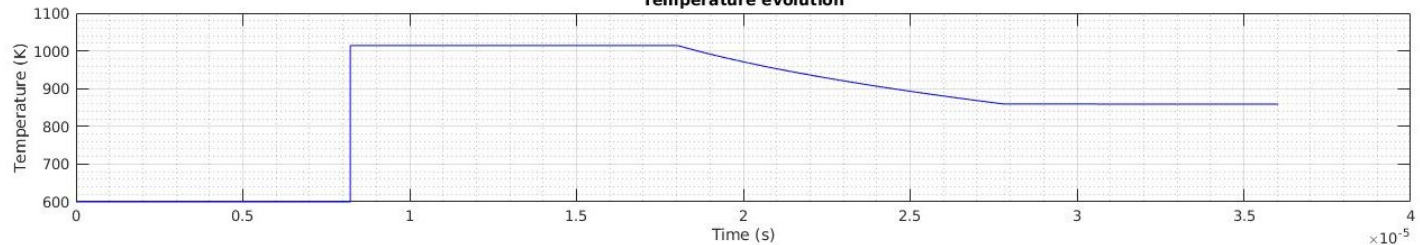
Results given by the inert gas dynamics theory

Evolution of a Lagrangian particle in the $H_2 - O_2$ phase

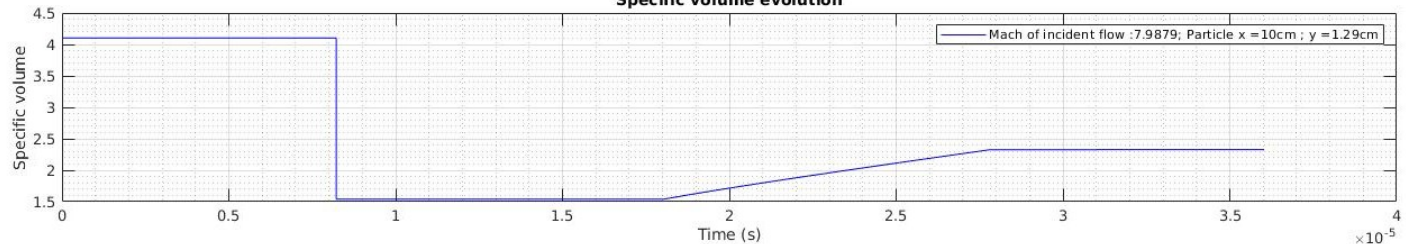
Mach of incident flow : 7.9879; Particle x = 10cm ; y = 1.29cm
Pressure evolution



Temperature evolution

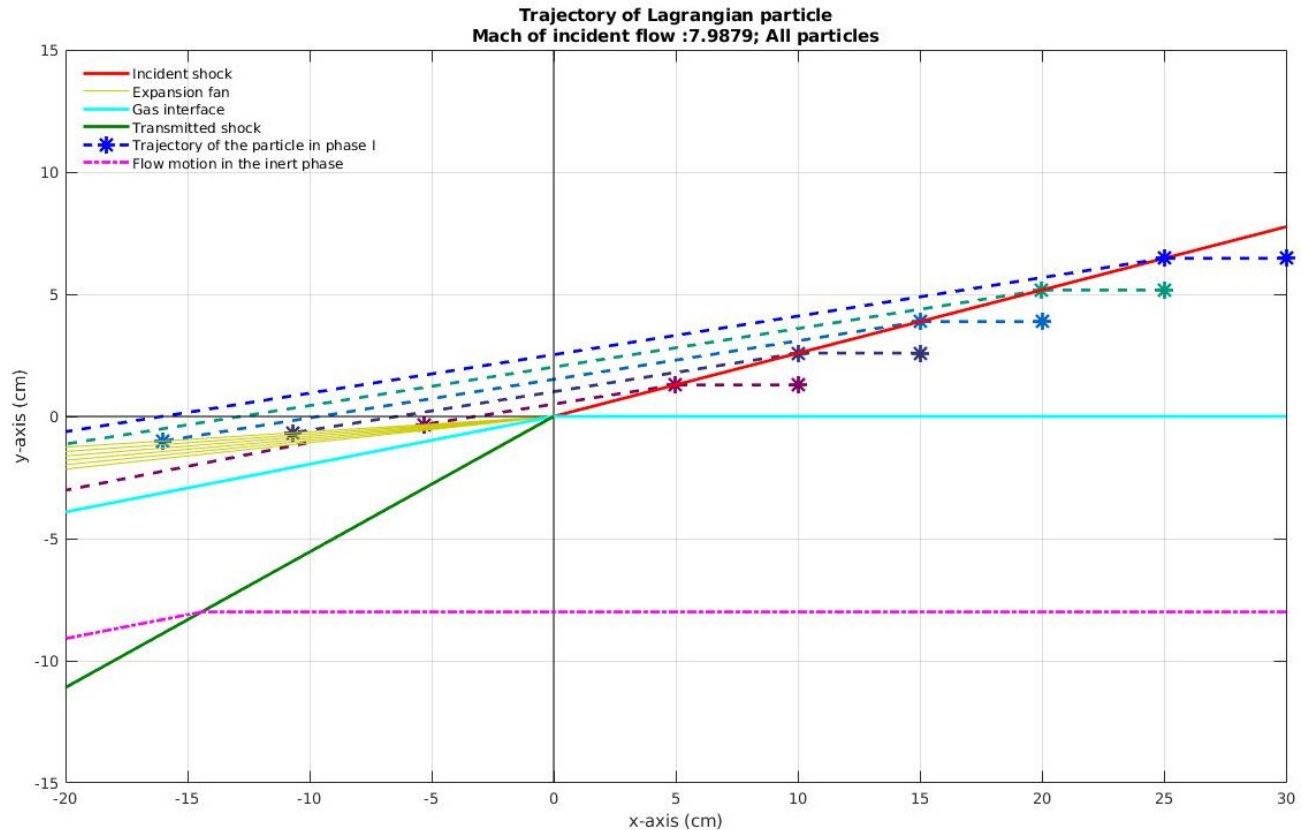


Specific volume evolution



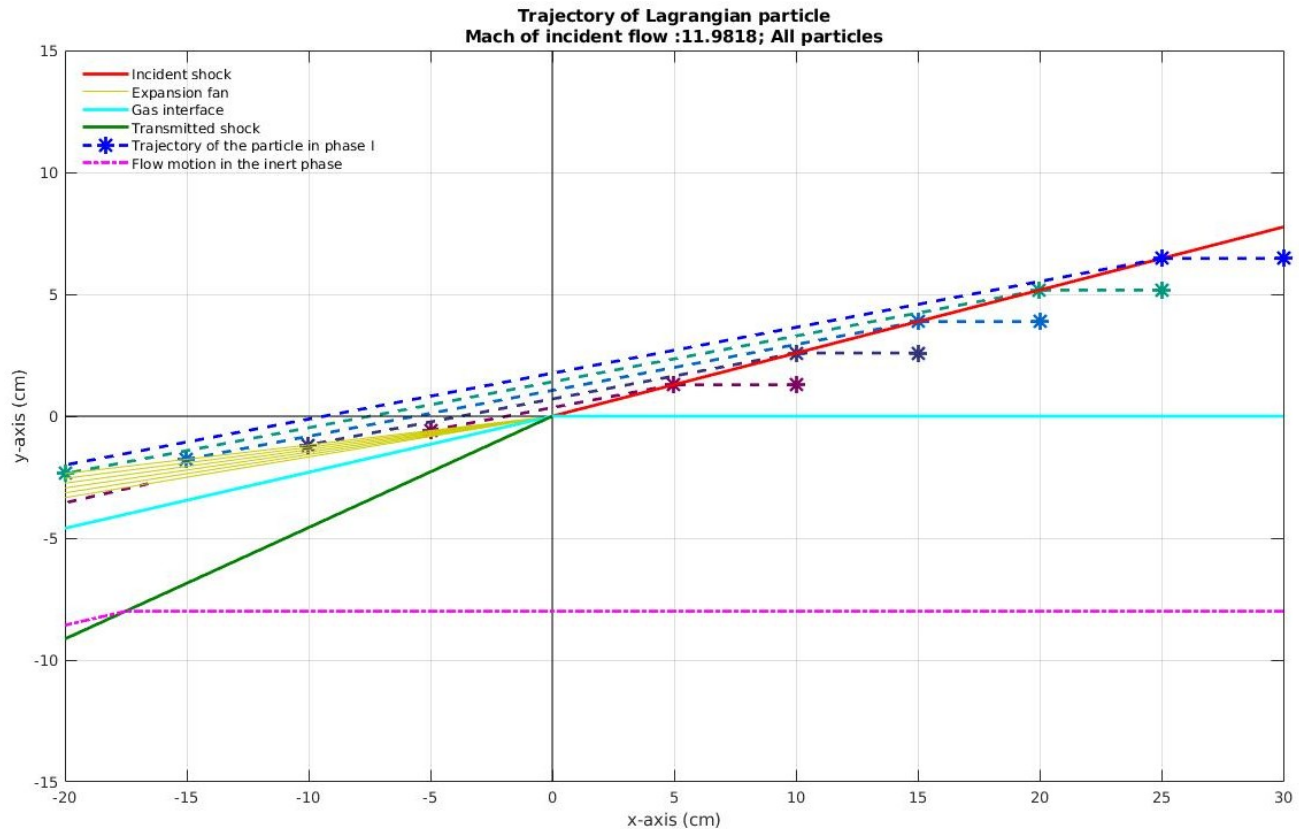
Results given by the inert gas dynamics theory

Evolution depends on position of the particle...



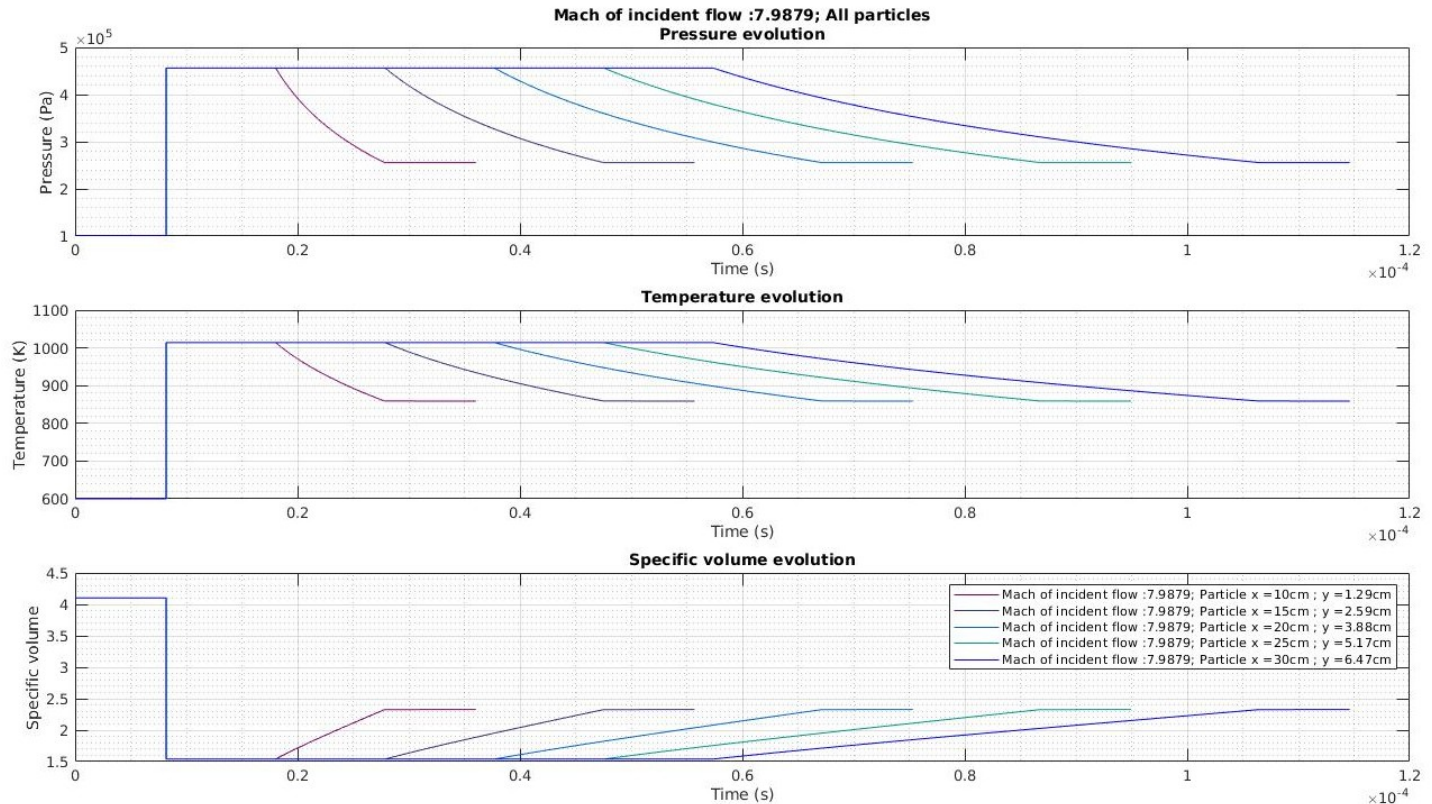
Results given by the inert gas dynamics theory

Evolution depends on position of the particle...



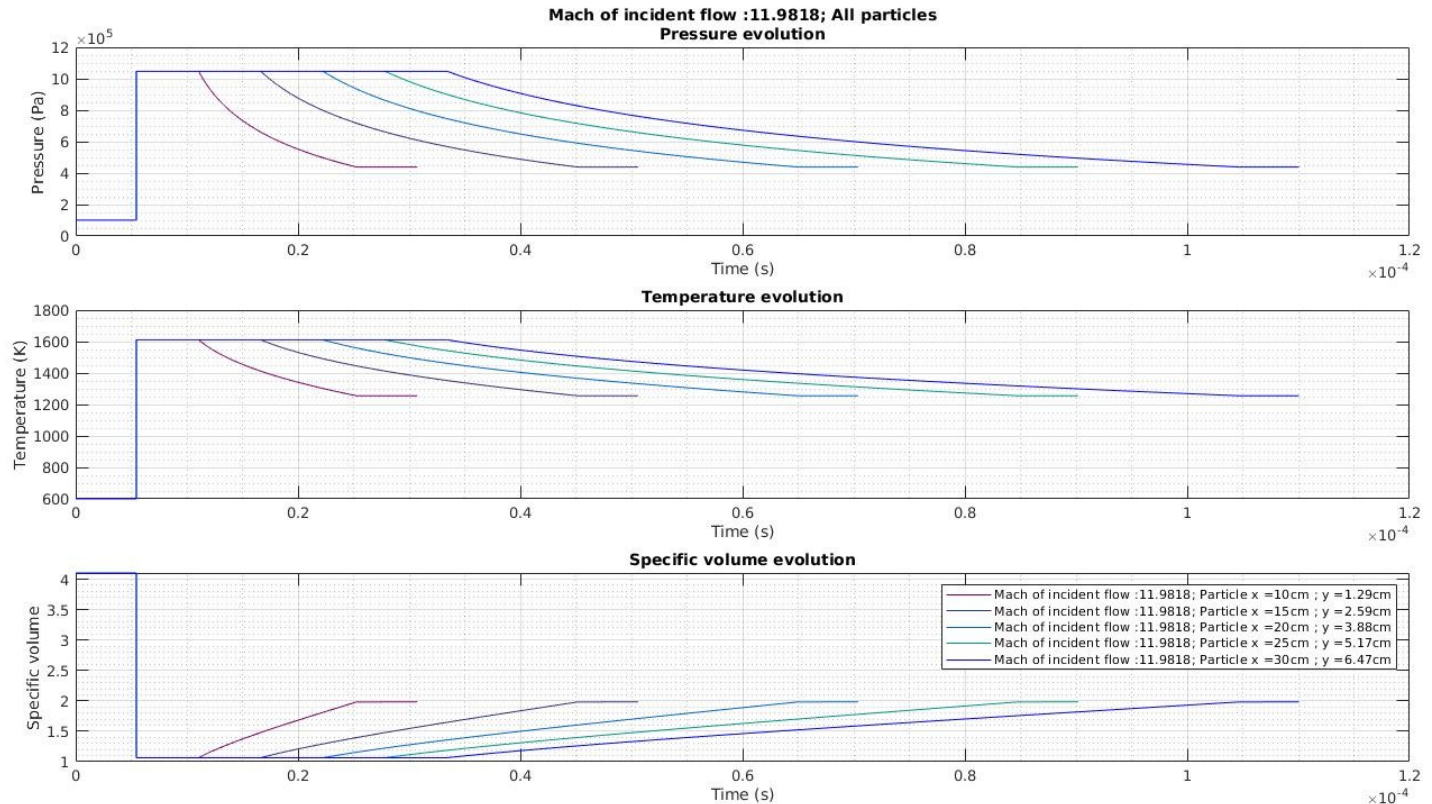
Results given by the inert gas dynamics theory

Evolution depends on position of the particle...



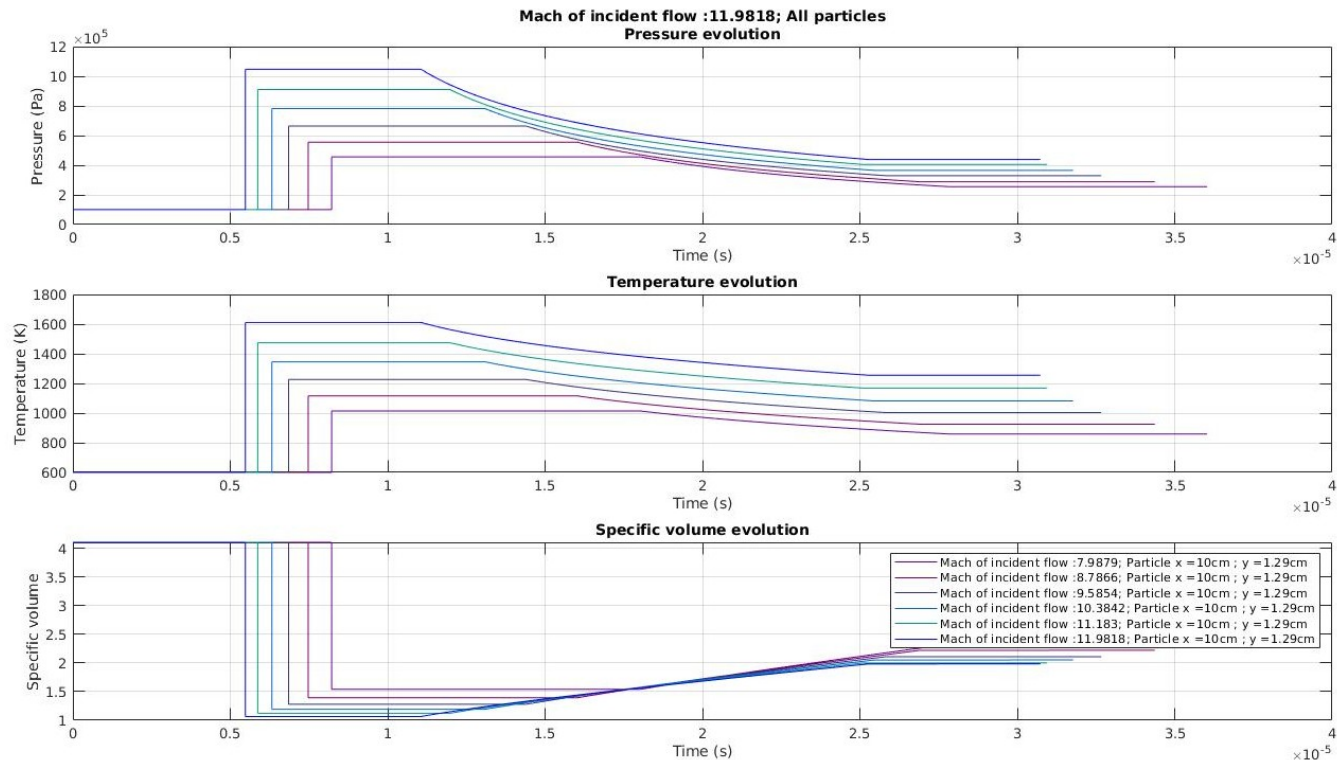
Results given by the inert gas dynamics theory

Evolution depends on position of the particle...



Results given by the inert gas dynamics theory

... and on Mach number of the shock



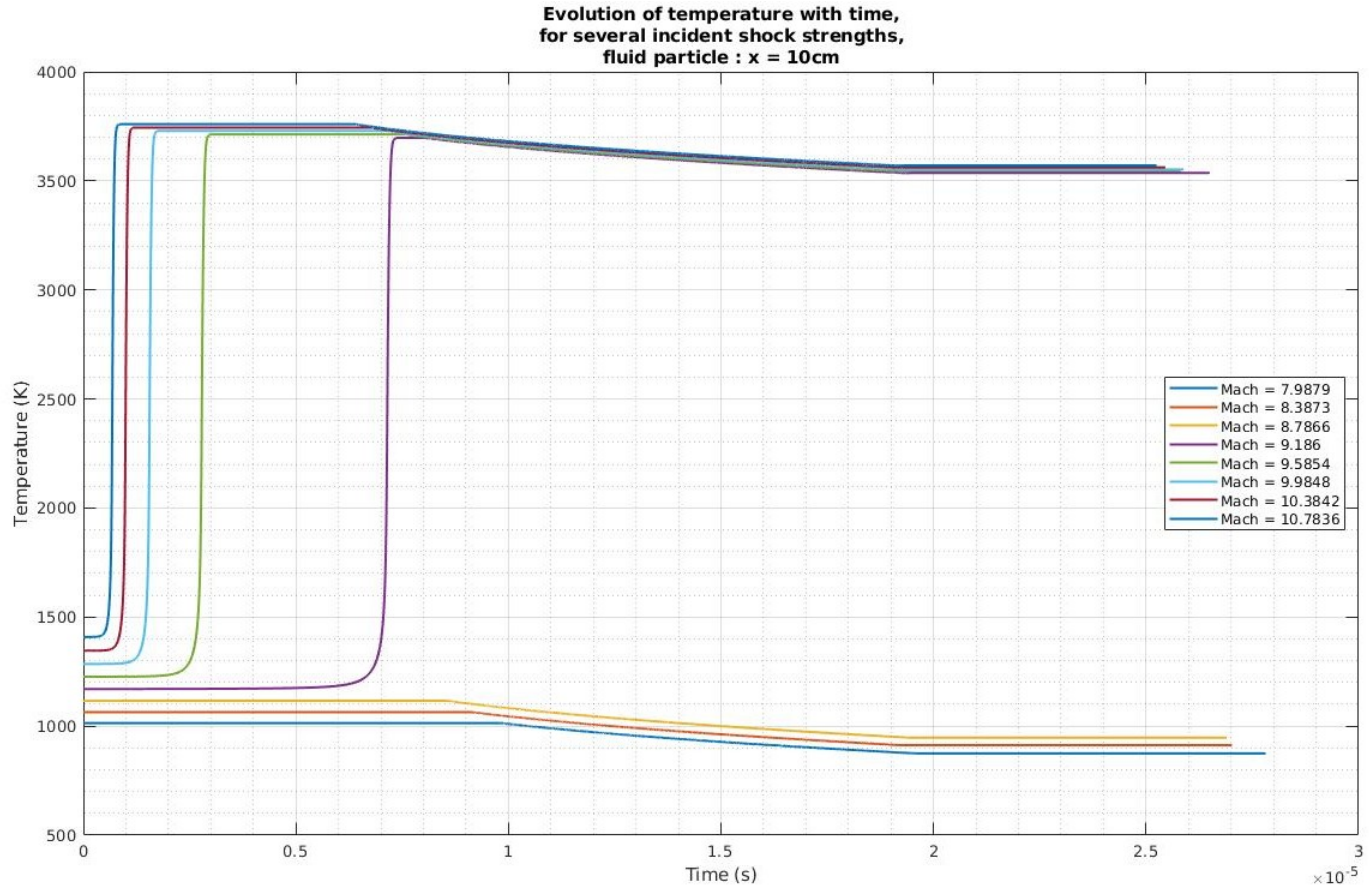
For a single particle, increasing M_{sh} leads to a rise in temperature and pressure : don't forget that $H_2 - O_2$ is a reactive phase. Is an ignition possible under certain conditions?

Main steps

- Calculate evolution of specific volume of different Lagrangian particles, for $\omega_i = 14.5^\circ$, for different Mach numbers (see slide 6).
- Use of CHEMKIN II to calculate chemical reactions in the reactive phase
- Outputs of CHEMKIN II : evolution of pressure, temperature and ratios of chemical species
- Temperature jump in CHEMKIN II output = detonation

Results of the chemical calculus

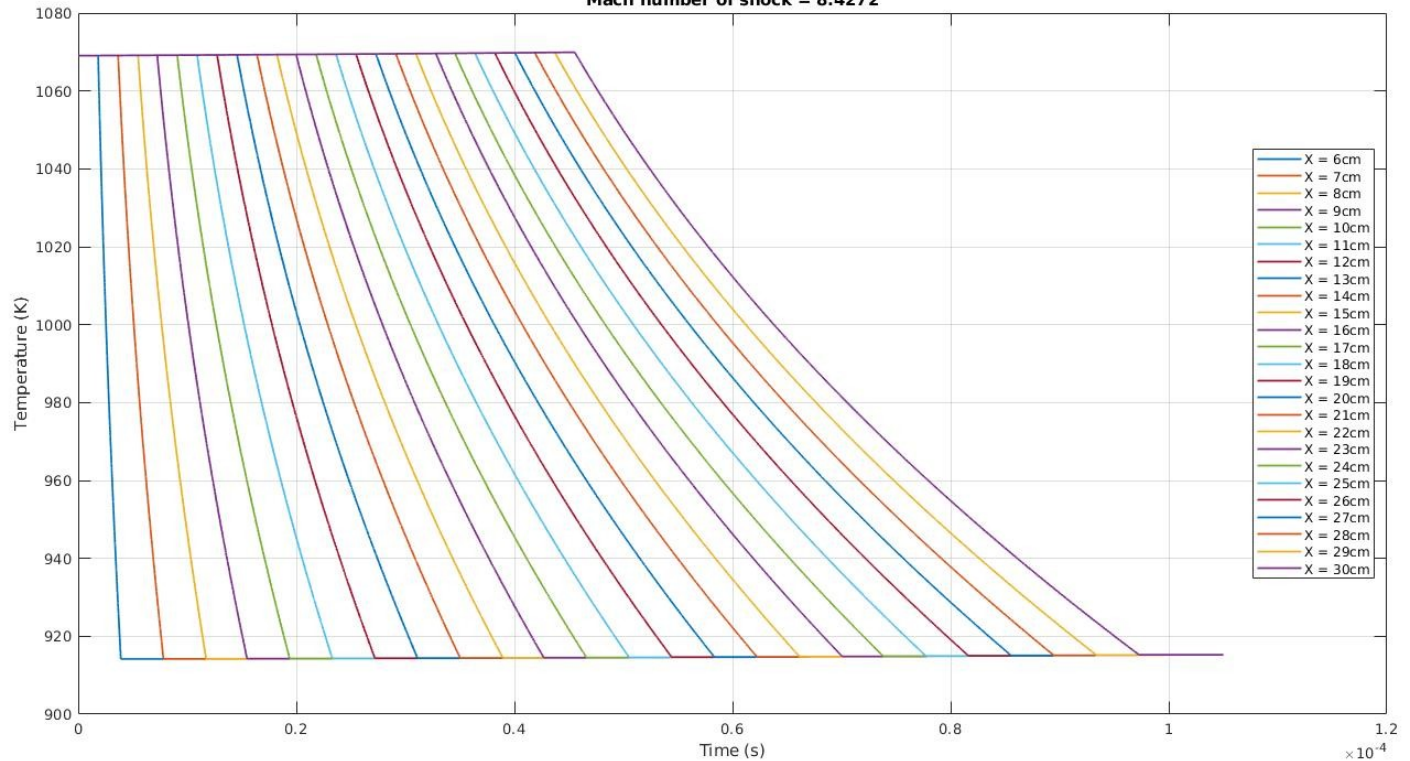
Existence of a threshold : Mach number of ignition



Results of the chemical calculus

Each particle has its own M_{ignit}

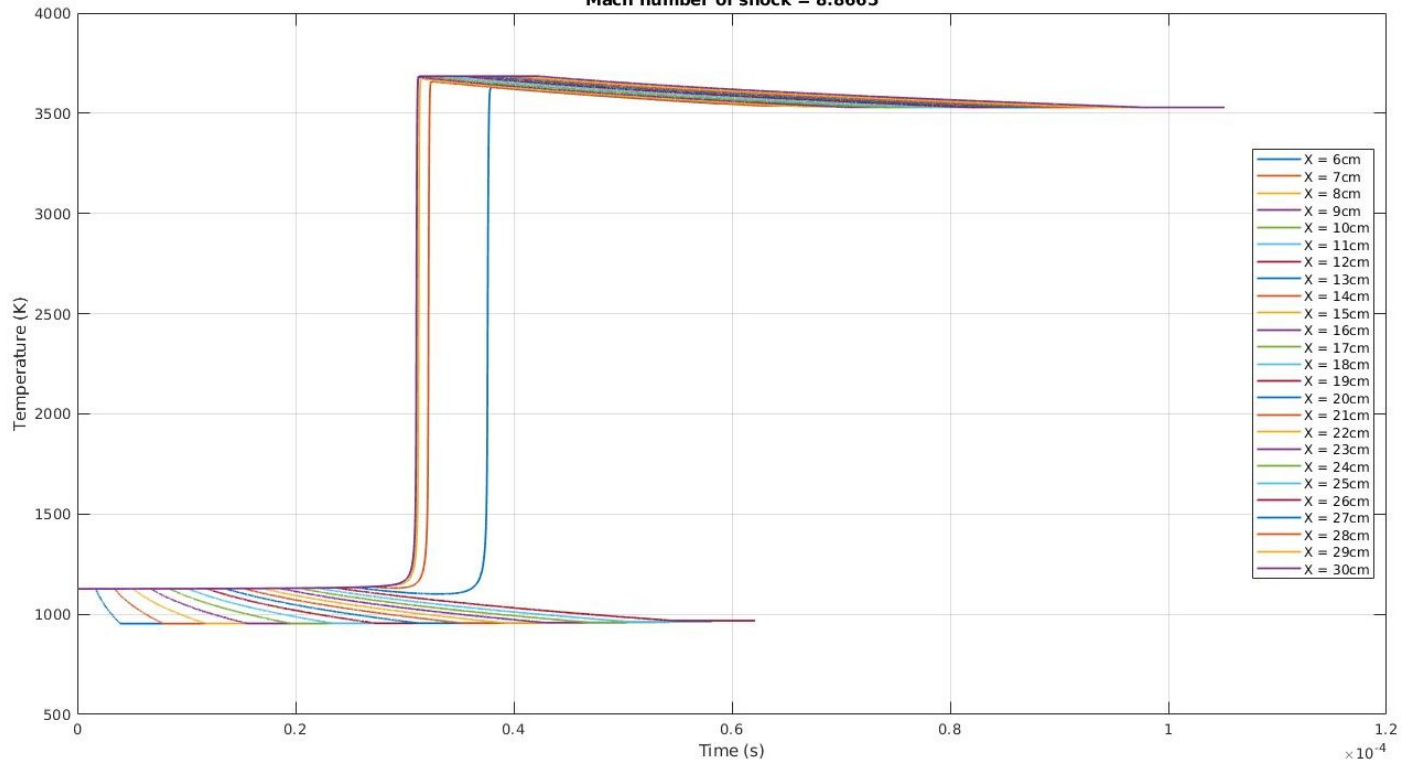
Evolution of temperature with time,
for several fluid particles,
Mach number of shock = 8.4272



Results of the chemical calculus

Each particle has its own M_{ignit}

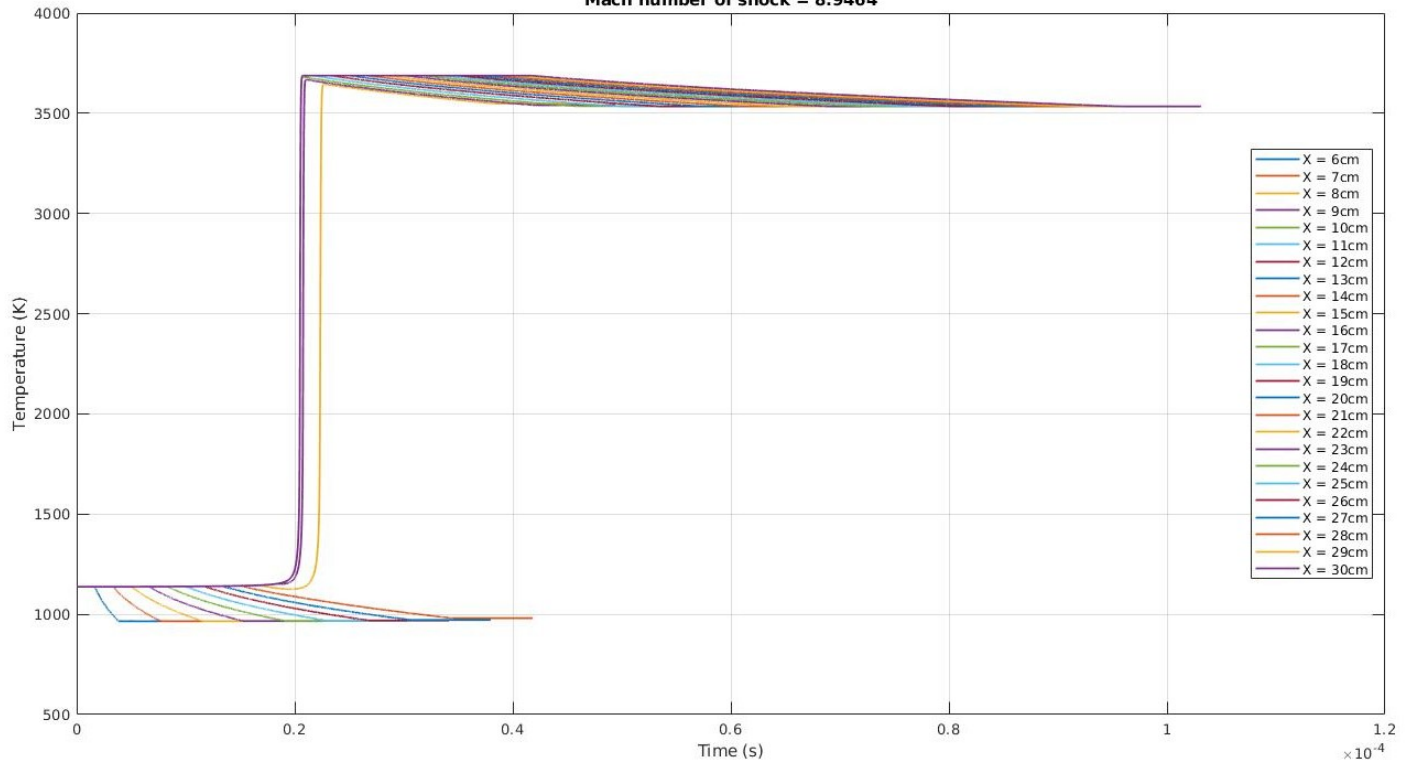
Evolution of temperature with time,
for several fluid particles,
Mach number of shock = 8.8665



Results of the chemical calculus

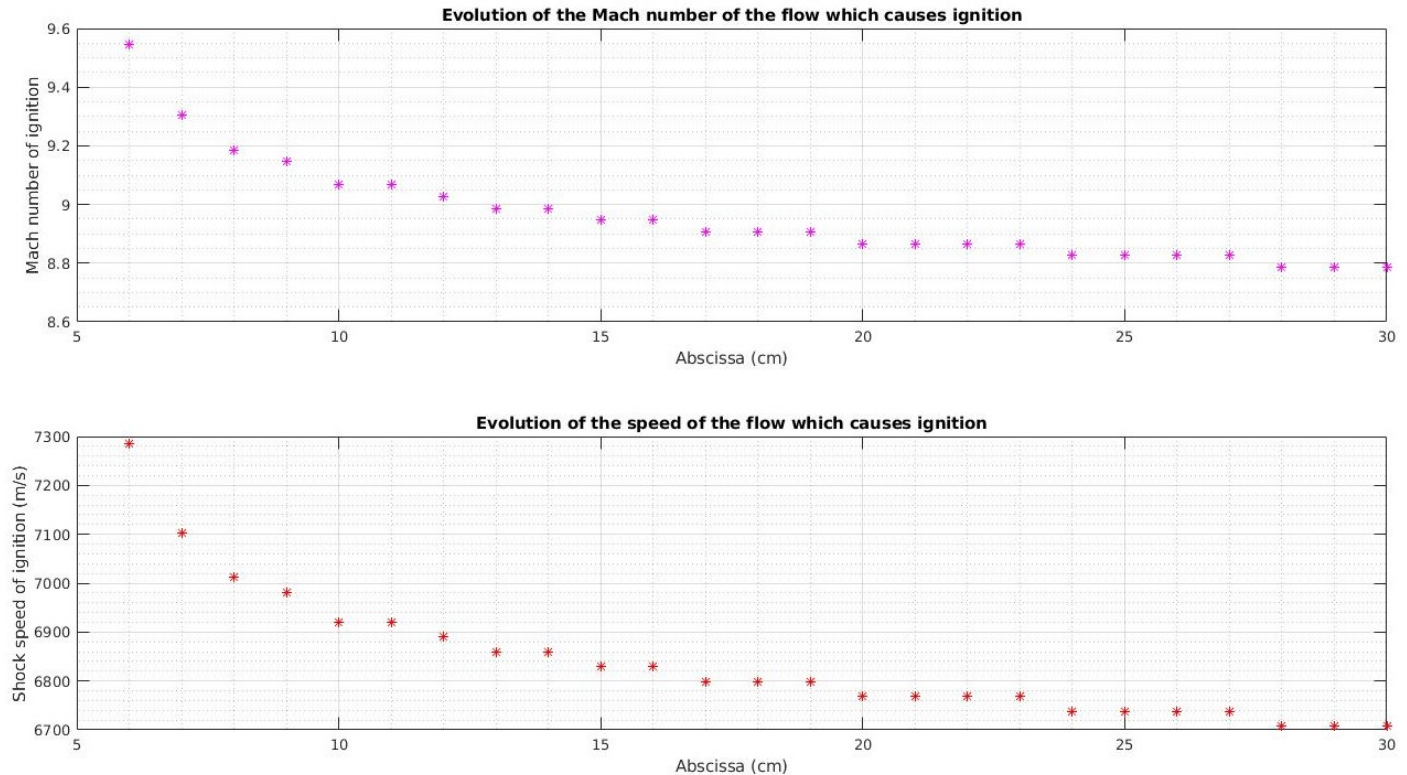
Each particle has its own M_{ignit}

Evolution of temperature with time,
for several fluid particles,
Mach number of shock = 8.9464



Results of the chemical calculus

M_{sh} plotted as a function of ordinate of the particle



Results of the chemical calculus

M_{sh} plotted as a function of ordinate of the particle

