

$P = IV = \frac{E}{\Delta t}$   
 $E = IV \Delta t$   
 $Q = (m_w C_w + m_{al} C_{al} + m_{coil} C_{coil}) \Delta T$

$\Delta T$  = change in Temp  
 $C_w$  = specific heat of water  
 $C_{al}$  = specific heat of aluminum

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## Lab 04 – Electrical Energy

### Lab Worksheet

Complete this lab worksheet and turn it in for credit. Show all your work including the calculations you performed (attach additional sheets if necessary).

Measure and record the mass of the aluminum cup.

Aluminum cup = 45.63 grams  
 $m_{al} = m_{al} = 45.63 \text{ grams}$

### I. Data Collection

Follow the instruction detailed in sections 3.4.1 and 3.4.2. Record the values specified below.

#### Trial 1

|                |               |
|----------------|---------------|
| $m_{cup+H_2O}$ | = 258.8g      |
| $m_w$          | = 213.17g     |
| $T_1$          | = 24.2°C      |
| $T_2$          | = 34.2°C      |
| $\Delta t$     | = 675 seconds |

| Trial 1        |          |          |           |
|----------------|----------|----------|-----------|
| Time (min : s) | Time (s) | Temp (C) | Volts (V) |
| 0:00           | 0        | 24.2     | 5.2       |
| 0:30           | 30       | 26.1     | 5.2       |
| 1:00           | 60       | 26.9     | 5.2       |
| 1:30           | 90       | 26.1     | 5.2       |
| 2:00           | 120      | 27.6     | 5.2       |
| 2:30           | 150      | 28.2     | 5.2       |
| 3:00           | 180      | 28.5     | 5.2       |
| 3:30           | 210      | 28.8     | 5.2       |
| 4:00           | 240      | 29.3     | 5.2       |
| 4:30           | 270      | 29.6     | 5.2       |
| 5:00           | 300      | 30       | 5.2       |
| 5:30           | 330      | 30.1     | 5.2       |
| 6:00           | 360      | 31.5     | 5.2       |
| 6:30           | 390      | 30.7     | 5.2       |
| 7:00           | 420      | 32.5     | 5.2       |
| 7:30           | 450      | 32.7     | 5.2       |
| 8:00           | 480      | 33.3     | 5.2       |
| 8:30           | 510      | 33.7     | 5.2       |
| 9:00           | 540      | 32.2     | 5.2       |
| 9:30           | 570      | 34       | 5.2       |
| 10:00          | 600      | 34.2     | 5.2       |

Make a quick graph of Temperature vs. time. Use this graph to determine if any of your techniques need to be adjusted for the second trial.

#### Trial 2

|                |               |
|----------------|---------------|
| $m_{cup+H_2O}$ | = 244.2g      |
| $m_w$          | = 198.57g     |
| $T_1$          | = 24.5°C      |
| $T_2$          | = 34.8°C      |
| $\Delta t$     | = 645 seconds |

| Trial 02       |          |          |           |
|----------------|----------|----------|-----------|
| Time (min : s) | Time (s) | Temp (C) | Volts (V) |
| 0:00           | 0        | 26.9     | 5.2       |
| 0:30           | 30       | 27.1     | 5.2       |
| 1:00           | 60       | 26.7     | 5.2       |
| 1:30           | 90       | 28.1     | 5.2       |
| 2:00           | 120      | 27.5     | 5.2       |
| 2:30           | 150      | 28.1     | 5.2       |
| 3:00           | 180      | 28       | 5.2       |
| 3:30           | 210      | 29.1     | 5.2       |
| 4:00           | 240      | 29.8     | 5.2       |
| 4:30           | 270      | 29.6     | 5.2       |
| 5:00           | 300      | 30.4     | 5.2       |
| 5:30           | 330      | 30.1     | 5.2       |
| 6:00           | 360      | 31.3     | 5.2       |
| 6:30           | 390      | 31.4     | 5.2       |
| 7:00           | 420      | 31.8     | 5.2       |
| 7:30           | 450      | 31.9     | 5.2       |
| 8:00           | 480      | 32.6     | 5.2       |
| 8:30           | 510      | 33       | 5.2       |
| 9:00           | 540      | 33.3     | 5.2       |
| 9:30           | 570      | 33.5     | 5.2       |
| 10:00          | 600      | 34       | 5.2       |
| 10:30          | 630      | 33.7     | 5.2       |
| 10:45          | 645      | 34.8     | 5.2       |

### II. Analysis

Estimate the uncertainty in each measured value. Justify your choice for each quantity.

$\delta I =$  \_\_\_\_\_

$$\delta V = \underline{\hspace{2cm}}$$

$$\delta t = \underline{\hspace{2cm}}$$

$$\delta m = \underline{\hspace{2cm}}$$

$$\delta T = \underline{\hspace{2cm}}$$

Compute the percent uncertainties in each of the following quantities:

$$\delta I = \underline{\hspace{2cm}}$$

$$\delta V = \underline{\hspace{2cm}}$$

$$\delta(\Delta t) = \underline{\hspace{2cm}}$$

$$\delta(\Delta T) = \underline{\hspace{2cm}}$$

Which quantity contributes the most uncertainty to the experiment?

Which quantity do you expect will be more precise? The electrical energy, or the thermal energy? Explain your answer.

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Plot both trials on a single graph as describe in the manual. Add error bars for the temperature and time measurements. Attach this graph to your worksheet. Comment on the consistency of the two plots.

What is the meaning of the difference between the magnitude of the slopes while the current is on, and when it is turned off?

Plot both trials on a single graph as describe in the manual. Add error bars for the temperature and time measurements. Attach this graph to your worksheet.  
Comment on the consistency of the two plots.

What is the meaning of the difference between the magnitude of the slopes while the current is on, and when it is turned off?

Compute the electrical energy lost by the circuit for both trials.

$E1 =$  \_\_\_\_\_

$E2 =$  \_\_\_\_\_

Compute the heat gained by the system for both trials.

$Q1 =$  \_\_\_\_\_

$Q2 =$  \_\_\_\_\_

Was one type of energy consistently higher? Explain why that may be.

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Compute the percent error for both trials.

Comment on the size of your percent error. Compare to the relative uncertainties you computed previously.

