2012 USABO Semifinal Exam Answers

1.	D	
2.	Ε	
3.	В	
4.	В	
5.	С	
6.	Α	
7.	D	
8.	С	
9.	D	
10.	Ε	
11.	Ε	
12.	Α	
13.	С	
14.	В	

15. D

16. B

17. B18. E

19. C

20. E

21. D

22. C

23. B

24. E

25. C

26. A

27. C

28. E

29. D

30. B

31. E

32. D

33. A

34. E

35. A

36. D

37. D

38. B

39. D

40. C

41. D

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4.	`	<u></u>
4:		C
4:		A
	4.	
	5.	
	6.	
	7.	
	8. 9.	
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	o. 1.	
	1. 2.	
	z. 3.	
	٥. 4.	
	1 . 5.	
	5. 6.	
	o. 7.	
	, . 8.	
	9.	
	o.	
	o. 1.	
	2.	
	z. 3.	
		A
	 5.	
		A
	7.	
	8.	
	9.	
	0.	
	1.	
	2.	
		ACDE
	4.	
7.	5.	ΑE
		AD

77. E

78. AC

79. AB

80. D

81. B

82. C

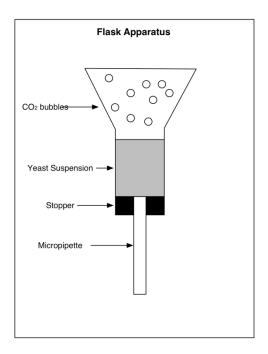
83.	Α
84.	AC
85.	С
86.	В
87.	Α
88.	В
89.	Α
90.	Α
91.	Α
92.	
93.	D
94.	
95.	С
96.	ACD
97.	В
98.	С
99.	
	. ACD
101	
102	
103	
	. CE
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Student Name	Student ID#	

Part C should be returned in its entirety with each student's scantron. Place all answers to Part C, Questions 1 and 2, on these two pages. Additional sheets of paper may be used, if necessary. Be sure that each page has the Student's Name and the Student's ID#. Please staple all pages together.

2012 USABO Semifinal Part C

1. The diagram shows an apparatus made by a student to investigate the effect of temperature on the activity of ethanol fermentation of yeast. The conical flask contains 2.5 g yeast suspended in 2% sucrose solution. The meniscus moves down the glass tube (5 mL micropipette) during fermentation.



The data shown below were collected at regular time intervals to assess the amount of suspension (mL) pushed in the glass tube due to CO₂ accumulation

Time (min)	4°C	10°C	20°C	35°C	55°C
1	0	0.2	0.4	0.7	0
2	0	1.0	1.3	1.2	0.1
3	0.1	1.9	2.2	2.8	0.2
4	0.2	3.1	3.3	4.4	0.3
5	0.3	4.0	NO	NO	0.4
			RESULT	RESULT	

A. Estimate the average rate of CO₂ production (mL CO₂/min) for the yeast suspension at 20°C using the values obtained in the period between 2 and 4 minutes.

1 ml/min

B. Estimate the specific rate of CO₂ generation [millimoles CO₂/(min·g)] at 20°C.

0.017 - 0.018 mmoles $CO_2/min \cdot g$

C. What would be the specific rate of ethanol accumulation [millimoles $CH_3CH_2OH/(min \cdot g)$], if the fermentation reaction follows the equation $C_6H_{12}O_6 \rightarrow 2C_2H_5OH + 2CO_2$?

0.017 - 0.018 mmoles CH₃OH /min·g

Partial points should be given if the student identifies that the answer to B and C will be the same, even if they don't get the numbers right (or calculate any numbers at all).

2. Complete the following table about hormones.

Hormone	Endocrine Gland	Chemical Class	Regulated By
Progesterone	Ovaries	Steroid	FSH and LH
Thyroid-stimulating	Anterior pituitary	Glycoprotein	Hypothalamic
<mark>hormone</mark>	gland		hormones
Luteinizing hormone	Anterior pituitary	Glycoprotein	Hypothalamic
			hormones
Follicle-stimulating	Anterior pituitary	Glycoprotein	Hypothalamic
<mark>hormone</mark>	gland		hormones
Calcitonin	Thyroid gland	Peptide Peptide	Blood calcium level
Prolactin	Anterior pituitary	Protein Protein	Hypothalamic Hypothalamic
	<mark>gland</mark>		<mark>hormones</mark>
Antidiuretic	Posterior pituitary	Peptide Peptide	Water/salt balance
hormone	<mark>gland</mark>		

We hope to see you as a Finalist!!