## PERIODIC TABLE OF THE ELEMENTS

| 1<br>Group               | 2                  | 1                   |                      | New Notation ———————————————————————————————————— |                    |                            |                    |                       |                     |                         |                     | ► 13<br>► IIIB       | 14<br>IVB            | 15<br>VB                         | 16<br>VIB          | 17<br>VIIB            | 18                  |        |
|--------------------------|--------------------|---------------------|----------------------|---------------------------------------------------|--------------------|----------------------------|--------------------|-----------------------|---------------------|-------------------------|---------------------|----------------------|----------------------|----------------------------------|--------------------|-----------------------|---------------------|--------|
| IA +1                    | IIA                | 1                   |                      |                                                   |                    | CAS Ver                    | sion -             |                       |                     |                         |                     | ► IIIA               | IVA                  | VA                               | VIA                | VIIA                  | VIIIA               | Shell  |
| H 1.00794                |                    |                     |                      |                                                   |                    |                            |                    |                       |                     |                         |                     |                      |                      |                                  |                    |                       | He<br>4.002602<br>2 | K      |
| 3 +1<br>Li               | 4 +2<br>Be         |                     |                      |                                                   | Atom               | nic Number<br>Symbol       |                    | Key to Ch             | _                   | Oxidation S             | States              | 5 +3<br>B            | 6 +2<br>C +4<br>-4   | 7 +1<br>N +2<br>N +3<br>+4<br>+5 | 0                  | 9 -1<br>F             | Ne 0                |        |
| 2-1                      | 9.012182<br>2-2    |                     |                      |                                                   | 2001 Ato           | mic Weight                 |                    | 118.710<br>-18-18-4   |                     | Electron<br>Configurati | on                  | 10.811<br>2-3        | 12.0107<br>2-4       |                                  | 2-6                | 18.9984032<br>2-7     | 2-8                 | K-L    |
| 11 +1<br>Na              | 12 +2<br>Mg        | 3                   | 4                    | 5                                                 | 6                  | 7                          | 8                  | 9                     | 10                  | 20migurau<br>11         | 12                  | 13 +3<br>A1          | 14 +2<br>Si +4<br>-4 | 15 +3<br>P +5                    | +6                 | +5                    | 18 0<br>Ar          |        |
|                          | 24.3050<br>2-8-2   | IIIA 🗲              |                      | VA                                                | VIA<br>VIB         | VIIA<br>VIIB               | •                  | – VIIIA<br>VIII –     | 10                  | IB<br>IB                | IIB<br>IIB          | 26.981538<br>2-8-3   | 28.0855<br>2-8-4     | 30.973761<br>2-8-5               | 32.065<br>2-8-6    | 35.453<br>2-8-7       | 39.948<br>2-8-8     | K-L-M  |
|                          | 20 +2<br>Ca        | 21 +3<br>Sc         | 22 +2<br>Ti +3<br>+4 | 23 +2<br>V +3<br>+4<br>+5                         |                    | 25 +2<br>Mn +3<br>+4<br>+7 | 26 +2<br>Fe +3     | 27 +2<br>Co +3        | 28 +2<br>Ni +3      | 29 +1<br>Cu +2          | 30 +2<br>Zn         | 31 +3<br>Ga          | 32 +2<br>Ge +4       | 33 +3<br>As +5<br>-3             | C- +6              | D +5                  | 36 ° Kr             |        |
| -8-8-1                   | 40.078<br>-8-8-2   | -8-9-2              | 47.867<br>-8-10-2    | 50.9415<br>-8-11-2                                | 51.9961<br>-8-13-1 | 54.938049<br>-8-13-2       | 55.845<br>-8-14-2  | 58.933200<br>-8-15-2  | 58.6934<br>-8-16-2  | 63.546<br>-8-18-1       | 65.409<br>-8-18-2   | 69.723<br>-8-18-3    | 72.64<br>-8-18-4     | 74.92160<br>-8-18-5              | 78.96<br>-8-18-6   | 79.904<br>-8-18-7     | 83.798<br>-8-18-8   | -L-M-N |
|                          | 38 +2<br>Sr        | 39 +3<br>Y          | 40 +4<br>Zr          | 41 +3<br>Nb +5                                    | 42 +6<br>Mo        | 43 +4<br>Tc +6<br>+7       | 44 +3<br>Ru        | 45 +3<br>Rh           | 46 +2<br>Pd +3      | 47 +1<br>Ag             | 48 +2<br>Cd         | 49 +3<br>In          | 50 +2<br>Sn +4       | 51 +3<br>Sb +5                   | T <sub>2</sub> +6  |                       | 54 0<br>Xe          |        |
| 85.4678<br>-18-8-1       | 87.62<br>-18-8-2   | 88.90585<br>-18-9-2 | 91.224<br>-18-10-2   | 92.90638<br>-18-12-1                              | 95.94<br>-18-13-1  | (98)<br>-18-13-2           | 101.07<br>-18-15-1 | 102.90550<br>-18-16-1 | 106.42<br>-18-18-0  | 107.8682<br>-18-18-1    | 112.411<br>-18-18-2 | 114.818<br>-18-18-3  | 118.710<br>-18-18 -4 | 121.760<br>-18-18-5              | 127.60<br>-18-18-6 | 126.90447<br>-18-18-7 | 131.293<br>-18-18-8 | -M-N-O |
|                          | 56 +2<br>Ba        | 57* +3<br>La        | 72 +4<br>Hf          | 73 +5<br>Ta                                       | 74 +6<br>W         | 75 +4<br>Re +6<br>+7       | 76 +3<br>Os +4     | 77 +3<br>Ir +4        | 78 +2<br>Pt +4      | 79 +1<br>Au +3          | 80 +1<br>Hg +2      | 81 +1<br>Tl +3       | 82 +2<br>Pb +4       | 83 +3<br>Bi +5                   |                    | 85<br>At              | 86 0<br>Rn          |        |
|                          | 137.327<br>-18-8-2 | 138.9055<br>-18-9-2 | 178.49<br>-32-10-2   | 180.9479<br>-32-11-2                              | 183.84<br>-32-12-2 | 186.207<br>-32-13-2        | 190.23<br>-32-14-2 | 192.217<br>-32-15-2   | 195.078<br>-32-17-1 | 196.96655<br>-32-18-1   | 200.59<br>-32-18-2  | 204.3833<br>-32-18-3 | 207.2<br>-32-18-4    | 208.98038<br>-32-18-5            | (209)<br>-32-18-6  | (210)<br>-32-18-7     | (222)<br>-32-18-8   | -N-O-P |
|                          | 88 +2<br>Ra        | 89** +3<br>Ac       | 104 +4<br>Rf         | 105<br>Db                                         | 106<br>Sg          | 107<br>Bh                  | 108<br>Hs          | 109<br>Mt             | 110<br>Ds           | 111<br>Rg               | 112<br>Uub          |                      | 114<br>Uuq           |                                  | 116<br>Uuh         |                       |                     |        |
|                          | (226)<br>-18-8-2   | (227)<br>-18-9-2    | (261)<br>-32-10-2    | (262)<br>-32-11-2                                 | (266)<br>-32-12-2  | (264)<br>-32-13-2          | (277)<br>-32-14-2  | (268)<br>-32-15-2     | (271)<br>-32-16-2   | (272)                   | (285)               |                      | (289)                |                                  | (289)              |                       |                     | -O-P-Q |
| * Lanthanides            |                    | 58 +3<br>Ce +4      | 59 +3<br>Pr          | 60 +3<br>Nd                                       | 61 +3<br>Pm        | 62 +2<br>Sm +3             | 63 +2<br>Eu +3     | 64 +3<br>Gd           | 65 +3<br>Tb         | 66 +3<br>Dy             | 67 +3<br>Ho         | 68 +3<br>Er          | 69 +3<br>Tm          | 70 +2<br>Yb +3                   | 71 +3<br>Lu        |                       |                     |        |
|                          |                    | 140.116             | 140.90765            | 144.24                                            | (145)              | 150.36                     | 151.964            | 157 .25               | 158.92534           | 162.500                 | 164.93032           | 167.259              | 168.93421            | 173.04                           | 174.967            |                       |                     |        |
| ** Actinides             |                    | -19-9-2<br>90 +4    | -21-8-2<br>91 +5     | -22-8-2<br>92 +3                                  | 93 +3              | -24-8-2<br>94 +3           | -25-8-2<br>95 +3   | -25-9-2<br>96 +3      | -27-8-2<br>97 +3    | -28-8-2<br>98 +3        | -29-8-2<br>99 +3    | -30-8-2<br>100 +3    | -31-8-2<br>101 +2    | -32-8-2<br>102 +2                |                    |                       |                     | -N-O-P |
|                          |                    | Th                  | Pa +4                | U +4                                              | Np +4              | Pu +4 +5 +6                | Am +5              | Cm                    | Bk +4               | Cf                      | Es                  | Fm                   | Md <sup>+3</sup>     | No +3                            | Lr                 |                       |                     |        |
| The new IUPAC format num |                    | 232.0381            |                      | 238.02891<br>-21-9-2                              |                    | (244)<br>-24-8-2           | (243)              | (247)<br>-25-9-2      | (247)               | (251)<br>-28-8-2        | (252)<br>-29-8-2    | (257) -30-8-2        | (258)<br>-31-8-2     | (259) -32-8-2                    | (262)              |                       |                     | -O-P-Q |

The new IUPAC format numbers the groups from 1 to 18. The previous IUPAC numbering system and the system used by Chemical Abstracts Service (CAS) are also shown. For radioactive elements that do not occur in nature, the mass number of the most stable isotope is given in parentheses. Elements 112, 114, and 116 have been reported but not confirmed.

- References
  1. G. J. Leigh, Editor, *Nomenclature of Inorganic Chemistry*, Blackwell Scientific Publications, Oxford, 1990.
  2. *Chemical and Engineering News*, 63(5), 27, 1985.
  3. Atomic Weights of the Elements, 2001, *Pure & Appl. Chem.*, 75, 1107, 2003.

Metallic solids Non-metallic solids Liquids