

Formulas

$$d = m / V$$

$$K = ^\circ C + 273.15$$

$$^{\circ}F = (9/5) (^{\circ}C) + 32^{\circ}$$

$$q = mc\Delta T \quad \& \quad q = n\Delta H$$

$$\Delta T = T_f - T_i \quad \text{or} \quad \Delta T = T_2 - T_1$$

$$PV = nRT$$

$$\frac{P_1 V_1}{n_1 T_1} = \frac{P_2 V_2}{n_2 T_2}$$

$$R = 0.08206 \text{ L atm / mol K}$$

$$M_1 V_1 = M_2 V_2$$

Conversions

$$1 \text{ mol} = 6.022 \times 10^{23}$$

$$1 \text{ cm}^3 = 1 \text{ mL (exactly)}$$

$$1 \text{ m} = 39.37 \text{ in}$$

$$1 \text{ in} = 2.54 \text{ cm (exactly)}$$

$$1 \text{ mi} = 1.609 \text{ km}$$

$$1 \text{ gal} = 3.785 \text{ L}$$

$$4 \text{ qt} = 1 \text{ gal (exactly)}$$

$$1 \text{ kg} = 2.205 \text{ lb}$$

$$1 \text{ lb} = 453.6 \text{ g}$$

$$1 \text{ cal} = 4.184 \text{ J}$$

$$760 \text{ torr} = 1 \text{ atm}$$

$$1.01 \text{ bar} = 1 \text{ atm}$$

$$760 \text{ mm Hg} = 1 \text{ atm}$$

Group or Ion	Generally soluble, with the stated exceptions.
Na ⁺ , K ⁺ , NH ₄ ⁺ Cl ⁻ , Br ⁻ , I ⁻ SO ₃ ²⁻ , SO ₄ ²⁻	None Except those containing Ag ⁺ , Hg ₂ ²⁺ , Pb ²⁺ . Except those containing Ca ²⁺ , Ba ²⁺ , Ag ⁺ , Pb ²⁺ .
CO ₃ ²⁻ , PO ₄ ³⁻ CrO ₄ ²⁻ , C ₂ O ₄ ²⁻ O ²⁻ , S ²⁻	Generally insoluble, with the stated Except those of Group 1 and NH ₄ ⁺ . Except those of Group 1 and NH ₄ ⁺ . Except those of Group 1, NH ₄ ⁺ , Ca ²⁺ , Sr ²⁺ , Ba ²⁺

Number of Electron Groups	Electron Group Geometry	Ideal Bond Angle	0 lone pairs	1 lone pair	2 lone pair
Molecular Geometry					
2	Linear	180 °	Linear	-	-
3	Trigonal Planar	120 °	Trigonal Planar	Bent	-
4	Tetrahedral	109.5 °	Tetrahedral	Trigonal Pyramidal	Bent

Metric Prefixes

$$\text{pico} = 10^{-12}$$

$$\text{nano} = 10^{-9}$$

$$\text{micro} = 10^{-6}$$

$$\text{mega} = 10^6$$







$$\text{tera} = 10^9$$

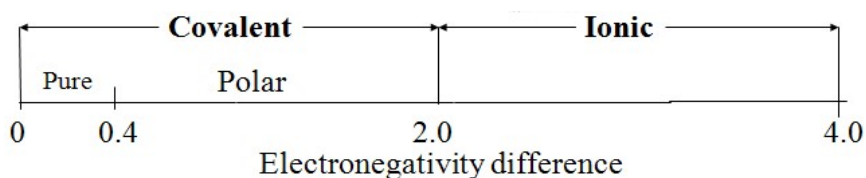
Standard Conditions (gasses)

$$\text{Pressure} = 1 \text{ atm (exactly)}$$

$$\text{Temperature} = 0 ^\circ C = 273.15 \text{ K}$$

Electron and Molecular Shapes

Electronic and Molecular Shapes	2 e regions	3 e regions	4 e regions
	 Linear	 Trigonal Planar	 Tetrahedral
		 Bent	 Trigonal Pyramidal
			 Bent



The Periodic Table of the Elements

1 Hydrogen 1 H 1.008 2.1	2 Helium 2 He 4.003 ---											13 Boron 5 B 10.81 2.0	14 Carbon 6 C 12.011 2.5	15 Nitrogen 7 N 14.007 3.0	16 Oxygen 8 O 15.999 3.5	17 Fluorine 9 F 18.998 4.0	18 Neon 10 Ne 20.180 ---
3 Lithium 3 Li 6.94 1.0	4 Beryllium 4 Be 9.012 1.5											13 Aluminum 13 Al 26.98 1.5	14 Silicon 14 Si 28.09 1.8	15 Phosphorus 15 P 30.97 2.1	16 Sulfur 16 S 32.06 2.5	17 Chlorine 17 Cl 35.45 3.0	18 Argon 18 Ar 39.95 ---
11 Sodium 11 Na 22.99 0.9	12 Magnesium 12 Mg 24.31 1.2	3 Scandium 21 Sc 44.96 1.3	4 Titanium 22 Ti 47.88 1.5	5 Vanadium 23 V 50.94 1.6	6 Chromium 24 Cr 52.00 1.6	7 Manganese 25 Mn 54.94 1.5	8 Iron 26 Fe 55.85 1.8	9 Cobalt 27 Co 58.93 1.8	10 Nickel 28 Ni 58.69 1.8	11 Copper 29 Cu 63.55 1.9	12 Zinc 30 Zn 65.39 1.6	31 Gallium 31 Ga 69.72 1.6	32 Germanium 32 Ge 72.61 1.8	33 Arsenic 33 As 74.92 2.0	34 Selenium 34 Se 78.97 2.4	35 Bromine 35 Br 79.90 2.8	36 Krypton 36 Kr 83.80 3.0
37 Rubidium 37 Rb 85.47 0.8	38 Strontium 38 Sr 87.62 1.0	39 Yttrium 39 Y 88.91 1.2	40 Zirconium 40 Zr 91.22 1.4	41 Niobium 41 Nb 92.91 1.6	42 Molybdenum 42 Mo 95.94 1.8	43 Technetium 43 Tc (98) 1.9	44 Ruthenium 44 Ru 101.07 2.2	45 Rhodium 45 Rh 102.91 2.2	46 Palladium 46 Pd 106.42 2.2	47 Silver 47 Ag 107.87 1.9	48 Cadmium 48 Cd 112.41 1.7	49 Indium 49 In 114.82 1.7	50 Tin 50 Sn 118.71 1.8	51 Antimony 51 Sb 121.76 1.9	52 Tellurium 52 Te 127.60 2.1	53 Iodine 53 I 126.90 2.5	54 Xenon 54 Xe 131.29 2.6
55 Cesium 55 Cs 132.91 0.7	56 Barium 56 Ba 137.33 0.9	57 Lanthanum 57 La 138.91 1.1	* Hafnium 72 Hf 178.49 1.3	73 Tantalum 73 Ta 180.95 1.5	74 Tungsten 74 W 183.84 1.7	75 Rhenium 75 Re 186.21 1.9	76 Osmium 76 Os 190.23 2.2	77 Iridium 77 Ir 192.22 2.2	78 Platinum 78 Pt 195.08 2.2	79 Gold 79 Au 196.97 2.4	80 Mercury 80 Hg 200.59 1.9	81 Thallium 81 Tl 204.38 1.8	82 Lead 82 Pb 207.20 1.8	83 Bismuth 83 Bi 208.98 1.9	84 Polonium 84 Po (209) 2.0	85 Astatine 85 At (210) 2.2	86 Radon 86 Rn (222) 2.4
87 Francium 87 Fr (223) 0.7	88 Radium 88 Ra (226) 0.9	** Actinium 89 Ac (227) 1.1	** Rutherfordium 104 Rf (267) ---	105 Dubnium 105 Db (268) ---	106 Seaborgium 106 Sg (271) ---	107 Bohrium 107 Bh (272) ---	108 Hassium 108 Hs (270) ---	109 Meitnerium 109 Mt (276) ---	110 Darmstadtium 110 Ds (281) ---	111 Roentgenium 111 Rg (280) ---	112 Copernicium 112 Cn (285) ---	113 Nihonium 113 Nh (284) ---	114 Flerovium 114 Fl (289) ---	115 Moscovium 115 Mc (288) ---	116 Livermorium 116 Lv (293) ---	117 Tennessine 117 Ts (294) ---	118 Oganesson 118 Og (294) ---

*lanthanides

**actinides

Cerium 58 Ce 140.12 1.1	Praseodymium 59 Pr 140.91 1.1	Neodymium 60 Nd 144.24 1.1	Promethium 61 Pm (145) 1.1	Samarium 62 Sm 150.36 1.2	Europium 63 Eu 151.97 1.1	Gadolinium 64 Gd 157.25 1.2	Terbium 65 Tb 158.93 1.1	Dysprosium 66 Dy 162.50 1.2	Holmium 67 Ho 164.93 1.2	Erbium 68 Er 167.26 1.2	Thulium 69 Tm 168.93 1.3	Ytterbium 70 Yb 173.04 1.1	Lutetium 71 Lu 174.97 1.1
Thorium 90 Th 232.04 1.3	Protactinium 91 Pa 231.04 1.5	Uranium 92 U 238.03 1.4	Neptunium 93 Np (237) 1.4	Plutonium 94 Pu (244) 1.3	Americium 95 Am (243) 1.3	Curium 96 Cm (247) 1.3	Berkelium 97 Bk (247) 1.3	Californium 98 Cf (251) 1.3	Einsteinium 99 Es (252) 1.3	Fermium 100 Fm (257) 1.3	Mendelevium 101 Md (259) 1.3	Nobelium 102 No (258) 1.3	Lawrencium 103 Lr (262) ---