

Derivation for homogeneous reacting gas mixture in closed system An adiabatic system with the volume as a function of time

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Species equation

$$\dot{\omega}_k = \frac{dc_k}{dt} \quad \frac{dm_k}{dt} = V\dot{\omega}_k W_k$$

$$v = \frac{V}{m_{tot}} = \frac{1}{\rho} \quad Y_k = \frac{m_k}{m_{tot}}$$

$$\frac{dY_k}{dt} = v\dot{\omega}_k W_k$$

Energy equation

$$\frac{du}{dt} + p\frac{dv}{dt} = 0$$

$$u = \sum Y_k u_k \quad \frac{du}{dt} = \sum Y_k \frac{du_k}{dt} + \sum u_k \frac{dY_k}{dt}$$

$$\frac{du_k}{dt} = c_{v,k} \frac{dT}{dt} \quad c_v = \sum Y_k c_{v,k}$$

$$\sum Y_k c_{v,k} \frac{dT}{dt} + \sum u_k \frac{dY_k}{dt} + p \frac{dv}{dt} = 0$$

$$c_v \frac{dT}{dt} + p \frac{dv}{dt} + \sum u_k \frac{dY_k}{dt} = 0$$

$$c_v \frac{dT}{dt} + p \frac{dv}{dt} + v \sum u_k \dot{\omega}_k W_k = 0$$

$$\frac{dT}{dt} = - \frac{p \frac{dv}{dt} + v \sum u_k \dot{\omega}_k W_k}{c_v}$$

Pressure equation, from ideal gas law

$$pv = RT$$

$$v \frac{dp}{dt} + p \frac{dv}{dt} = R \frac{dT}{dt} + T \frac{dR}{dt}$$

$$\frac{dp}{dt} = \left(R \frac{dT}{dt} + T \frac{dR}{dt} \right) * \rho$$

$$R = \frac{R_u}{W_{ave}} \quad W_{ave} = \frac{1}{\sum \frac{Y_k}{W_k}}$$

$$R = R_u * \sum \frac{Y_k}{W_k}$$

$$\frac{dR}{dt} = R_u * \sum \frac{1}{W_k} \frac{dY_k}{dt}$$

$$\frac{dp}{dt} = \left(\frac{R_u}{W_{ave}} \frac{dT}{dt} + T R_u * \sum \frac{1}{W_k} \frac{dY_k}{dt} \right) * \rho$$

Volume equation

$$R = \frac{V_{max}}{V_{min}} = \text{compression ratio}$$

f = rotation frequency

$\omega = 2\pi f$ = angular velocity

$$V = V_{ave} + (V_{max} - V_{ave}) * \cos(\omega t)$$

$$V_{ave} = \frac{V_{max} + V_{min}}{2} = \frac{V_{max} + V_{max}/R}{2} = V_{max} \left(\frac{1}{2} + \frac{1}{2R} \right)$$

$$V = V_{max} \left(\frac{1}{2} + \frac{1}{2R} \right) + \left(V_{max} - V_{max} \left(\frac{1}{2} + \frac{1}{2R} \right) \right) * \cos(\omega t)$$

$$V = V_{max} \left(\left(\frac{1}{2} + \frac{1}{2R} \right) + \left(\frac{1}{2} - \frac{1}{2R} \right) * \cos(\omega t) \right)$$

$$V = V_{max} \left(\frac{1}{2} + \frac{1}{2R} \right) \left(1 + \frac{\left(\frac{1}{2} - \frac{1}{2R} \right)}{\left(\frac{1}{2} + \frac{1}{2R} \right)} * \cos(\omega t) \right)$$

$$V = V_{ave} \left(1 + \left(\frac{2}{R+1} - 1 \right) * \cos(\omega t) \right)$$

$$R = \frac{V_{max}}{V_{min}} = \frac{1+\alpha}{1-\alpha}$$

$$R - R\alpha = 1 + \alpha \quad R - 1 = (R + 1)\alpha$$

$$\alpha = \frac{R - 1}{R + 1}$$

$$\frac{2}{R + 1} - 1 = \frac{R - 1}{R + 1} = \alpha$$

$$V = V_{ave}(1 + \alpha \cos(\omega t))$$

$$V_{ave} = \frac{V_{max} + V_{max} \frac{1 - \alpha}{1 + \alpha}}{2} = V_{max} \left(\frac{1 + \frac{1 - \alpha}{1 + \alpha}}{2} \right) = \frac{V_{max}}{1 + \alpha}$$

$$V = \frac{V_{max}}{1 + \alpha} (1 + \alpha \cos(\omega t))$$

Total mass must be constant, the system is closed and no nuclear reactions, convert volume to specific volume

$$\nu = \frac{\nu_{max}}{1 + \alpha} (1 + \alpha \cos(\omega t))$$

$$\nu_{max} = \nu_0 = \frac{R_u T_0}{p_0 W_{ave}}$$

$$\frac{d\nu}{dt} = \frac{\nu_0}{1 + \alpha} (-\alpha \omega \sin(\omega t))$$

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