

MSc Computational Science  
joint programme UvA/VU



# A DIFFUSION-BASED MODEL OF SPATIAL INTERACTIONS IN *ASPERGILLUS* SPP. GERMINATION



## FIRST EXPERIMENTS

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# OUTLINE



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Simulation space

Inhibitor

Concentration check

## 2 Experiment 1 Setup

Assumptions

Analytical verification

## 3 Experiment 2 Setup

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# GENERAL ASSUMPTIONS



## Simulation space

### » 2D lattice $\equiv$ thin layer of medium clamped between 2 impermeable sheets

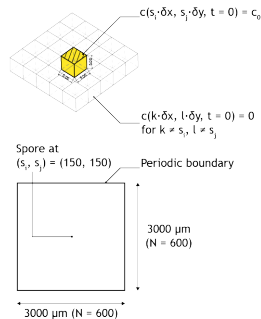
- $L = 3000 \mu\text{m}$
- Discrete blocks of size  $\delta x = \delta y = \delta z = 5 \mu\text{m}$
- $N = 600$  subdivisions
- Lattice volume  $V_L = 4.5 \times 10^7 \mu\text{m}^3 = 4.5 \times 10^{-5} \text{ mL}$
- Assuming water as medium

### » Spore $\equiv$ single block in the center of the lattice, inhibitor source

- Volume  $V_s = 125 \mu\text{m}^3$
- Top and bottom side of spore are sealed

### » Inhibitor - initial concentrations $c(x, y, t = 0)$ :

- At spore:  $c(x_s, y_s, t = 0) = c_0$
- Elsewhere:  $c(x, y, t = 0) = 0$



# GENERAL ASSUMPTIONS



## Inhibitor

### » Initial concentration is assumed to be contained within the cell wall

- Assumed cell wall thickness:  $d = 10 \text{ nm}$  [6]
- Total cell wall volume:  $V_{\text{cw}} \approx 6 \cdot 25 \mu\text{m}^2 \cdot 0.01 \mu\text{m} = 1.5 \mu\text{m}^3$
- This concentration is non-replenishable.



### » Assumed to have known molecular properties of 1-octen-3-ol:

- Molecular weight:  $m_M = 128.21 \text{ g/mol}$
- Density:  $\rho = 0.837 \text{ g/mL}$

Figure: 1-octen-3-ol

### » Assumed to have similar diffusion properties as glucose:

- Diffusion constant in water at temperature  $T = 293.15\text{K}$ :  
 $D = 600 \mu\text{m}^2/\text{s}$  [2]  
(i.e. a molecule wanders on average  $\langle x \rangle = \sqrt{6 \cdot D} = 60 \mu\text{m}$  per second)
- Permeability constant in an artificial lipid bilayer:  
 $P_s = 1.9 \times 10^{-6} \mu\text{m/s}$  (average from [3] and [1], close to [8])
- Octanol-water partition coefficient:  $K \approx 10^{-3}$



Figure: D-Glucose

# GENERAL ASSUMPTIONS



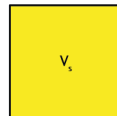
## Initial concentration check

### » Full spore volume

- Homogeneous distribution, standard temperature and pressure:

$$c_{\max} = \frac{\rho}{m_M} = \frac{0.837 \text{ g/mL}}{128.21 \text{ g/mol}} \approx 6.5 \text{ mol/mL} \quad (1)$$

- Equivalently,  $N = N_A c_{\max} V_s = 4.89 \times 10^{14}$  molecules.

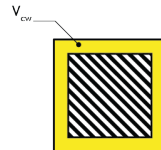


### » Cell wall volume only

- The maximum possible concentration is:

$$c_{\text{mcw}} = \frac{\rho V_{\text{cw}}}{m_M V_s} = 6.5 \text{ mol/mL} \cdot 0.012 = 0.078 \text{ mol/mL} \quad (2)$$

- Equivalently,  $N = N_A c_{\text{mcw}} V_{\text{cw}} = 7.04 \times 10^{10}$  molecules.
- The experiments use an initial concentration  $c_0 = c_{\max, \text{cw}}$ .



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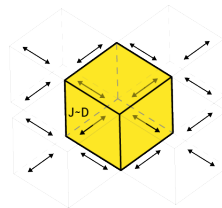
## 5 Discussion

# EXPERIMENT 1 SETUP



## Assumptions

- » At  $t = 0$ , all the inhibitor is contained in a cell wall at the interface between the spore and the medium, having the same diffusion coefficient as water.
- » To prevent numerical error, the concentration in the simulations is normalised by a factor  $\lambda = \frac{1}{c_0}$ .
- » Diffusion is simulated using the diffusion equation (Fick's Law):
$$\frac{\partial c}{\partial t} = D \nabla^2 c \quad (3)$$
- » Simulation run for  $t_{\max} = 3600$  s
- » The times for reaching threshold concentrations of  $c_t = k c_0$  ( $k = \{10^{-1}, 10^{-2}, \dots, 10^{-6}\}$ ) are recorded.

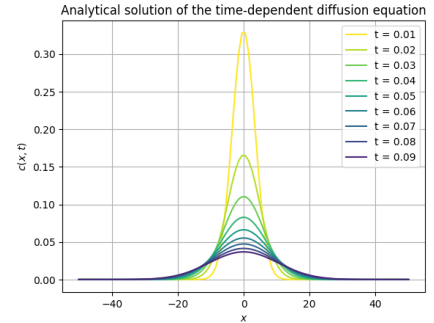


- » Assuming the initial condition is a delta pulse of concentration at coordinates  $(x_0, y_0)$
- » The concentration at an arbitrary position  $(x, y)$  at time  $t$  can also be computed through the analytical formula[5]:

$$c(x, y, t) = \frac{N^{2/3}}{4\pi Dt} \exp\left(-\frac{(x - x_0)^2 + (y - y_0)^2}{4Dt}\right) \quad (4)$$

- » At the spore site, this equation becomes:

$$c(x_0, y_0, t) = \frac{(V_s c_0)^{2/3}}{4\pi Dt} \quad (5)$$





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# EXPERIMENT 2 SETUP



## Assumptions

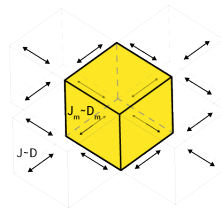
- » All the inhibitor is contained within in a cell wall with the properties of a lipid bilayer at the spore-medium interface.
- » Diffusion through the medium as before
- » Diffusion at the interface follows the flux equation

$$J_m = -D_m \nabla c \quad (6)$$

- » where  $J_m$  is the flux through the cell wall such that

$$D_m = \frac{d}{K} P_s = 1.9 \times 10^{-5} \mu\text{m}^2/\text{s} \quad (7)$$

- » Simulation run for  $t_{\text{max}} = 7200 \text{ s}$
- » The times for reaching threshold concentrations of  $c_t = kc_0$  ( $k = \{10^{-1}, 10^{-2}, \dots, 10^{-6}\}$ ) are recorded.



- » Since the concentration outside ( $c_{\text{out}}$ ) the spore with a volume  $V$  and a surface  $A$  diffuses much faster than inside ( $c_{\text{in}}$ ),  $c_{\text{out}} \approx \text{const.}$

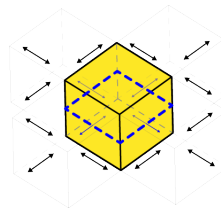
- » Then, an analytical solution is:

$$c_{\text{in}}(t) = c_{\text{out}} - \Delta c(0)e^{-t/\tau}, \quad (8)$$

where  $\tau$  is a decay constant:

$$\tau = \frac{V}{AD_m} \quad (9)$$

- » Only 2D communication in the numerical simulation  $\rightarrow A$  is actually the **circumference** of the cross section!



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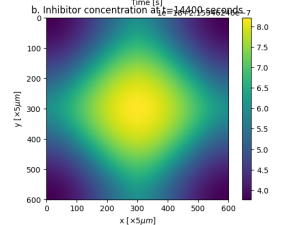
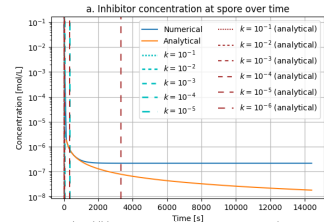
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# RESULTS



## Experiment 1

- » Results after 4 hours
- » Numerical vs. analytical
  - Alignment of numerical and analytical solution up to  $t \approx 1000$  s, then finite size starts to take effect.
  - Under the assumed  $c_0$ , the concentration in this period is low enough to make this deviation insignificant (few thousands of molecules).
- » Threshold times follow a power law
- » Under the given  $c_0$  (maximum that the cell wall can fit), the results seem unrealistic  $\rightarrow$  diffusion is too fast

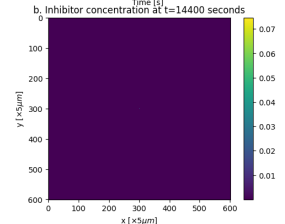
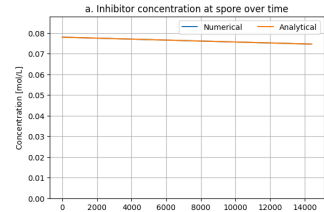


# RESULTS



## Experiment 2

- » Results after 4 hours
- » Numerical vs. analytical
  - Good alignment of numerical and analytical solution (because medium does not saturate in the given time)
- » Linear decrease in concentration
- » Diffusion appears too slow, threshold of one tenth never reached



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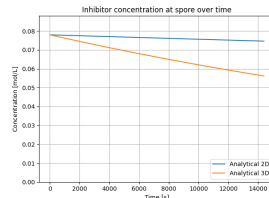
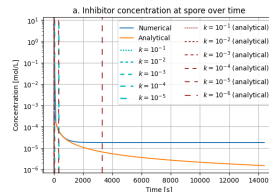
## 5 Discussion

# DISCUSSION



## Validity of results

- » The documented threshold for 1-octen-3-ol inhibition is  $10 \text{ mM} = 1 \times 10^{-5} \text{ mol/mL}$ [4].
  - Diffusion through water: threshold immediately reached
  - Diffusion through membrane: threshold not reached
- » The resulting concentrations can be rescaled to different  $c_0$ :
  - If the starting concentration is based on the entire spore volume  $V_s$ , then Experiment 1 would reach the 1-octen-3-ol threshold within 30 minutes.
  - More reasonable result, but unrealistic initial concentration.
  - Perhaps the initial concentration is only slightly above threshold?
- » The truth lies in between:
  - Permeation through the membrane is facilitated by proteins (e.g. glucose transporters)  $\rightarrow$  increased  $P_s$ ?
  - Cell wall permeability modulated during transition from dormancy.
  - In three dimensions: more surface area for diffusion





# DISCUSSION



## Future outlook

- » A spore produces relatively little 1-octen-3-ol[4, 7]. Perhaps it is not a baseline inhibitor for a single spore but only adds upon other inhibitors?
- » Therefore, a single spore cannot inhibit itself beyond its base germination time with this substance, but many spores can?
- » Future experiments could investigate the relationship between inhibitor concentration and spore density.

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