

The Bohr Effect



Or how to get Heme to pick up and drop off oxygen
Marguerite Butler Zool 430

RIGH
T

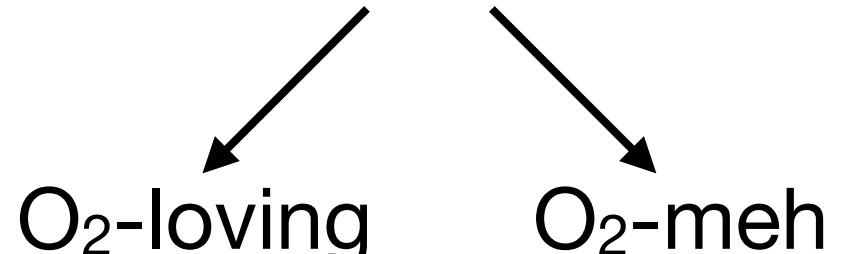


The Bohr effect

Explains how high $[CO_2]$ helps hemoglobin release O_2 in tissues that need it

O_2 Concentration

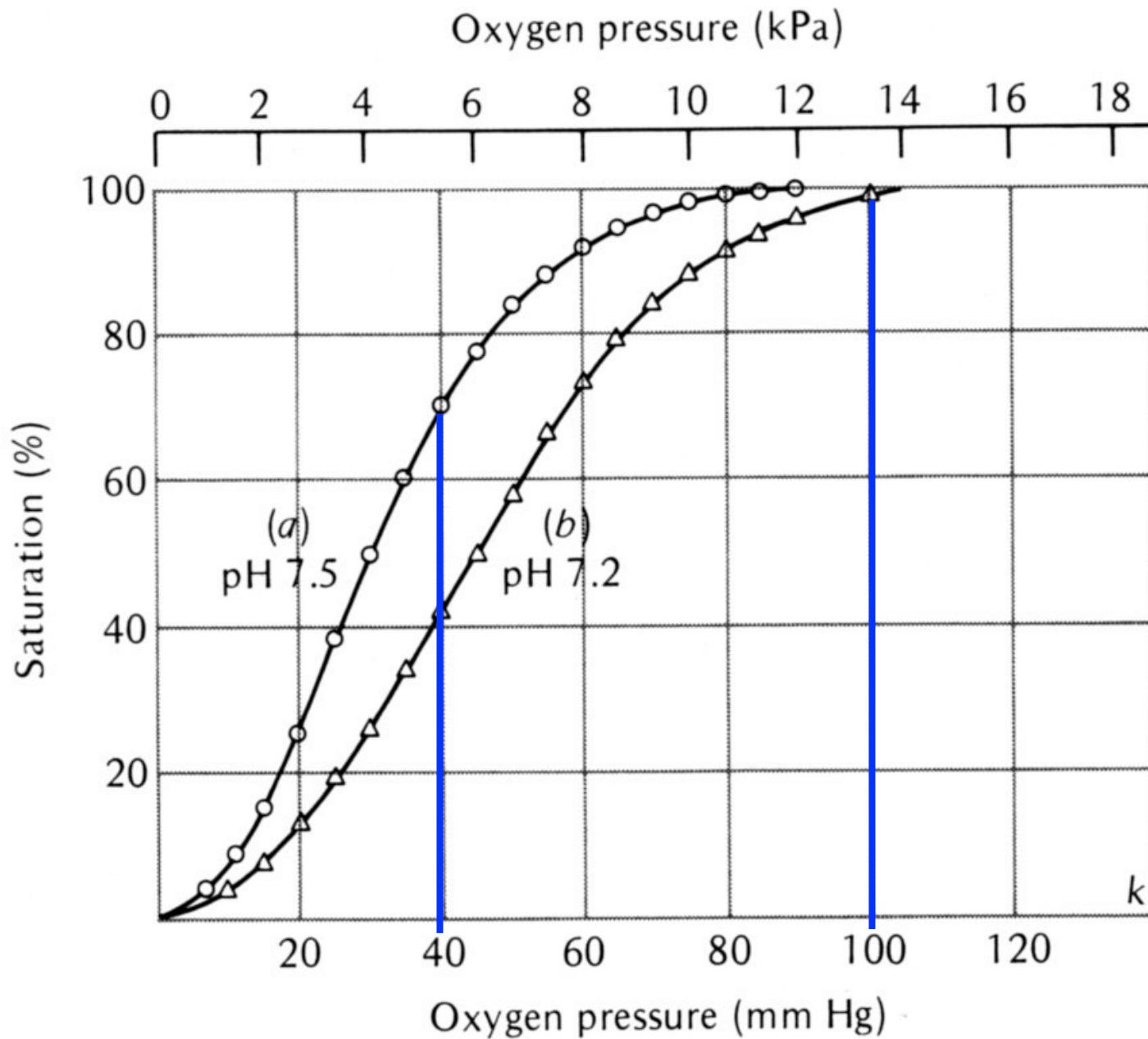
O_2 Affinity



pH

Hb-Oxygen Dissociation Curve

Hemoglobin binds O₂ reversibly



Hemoglobin

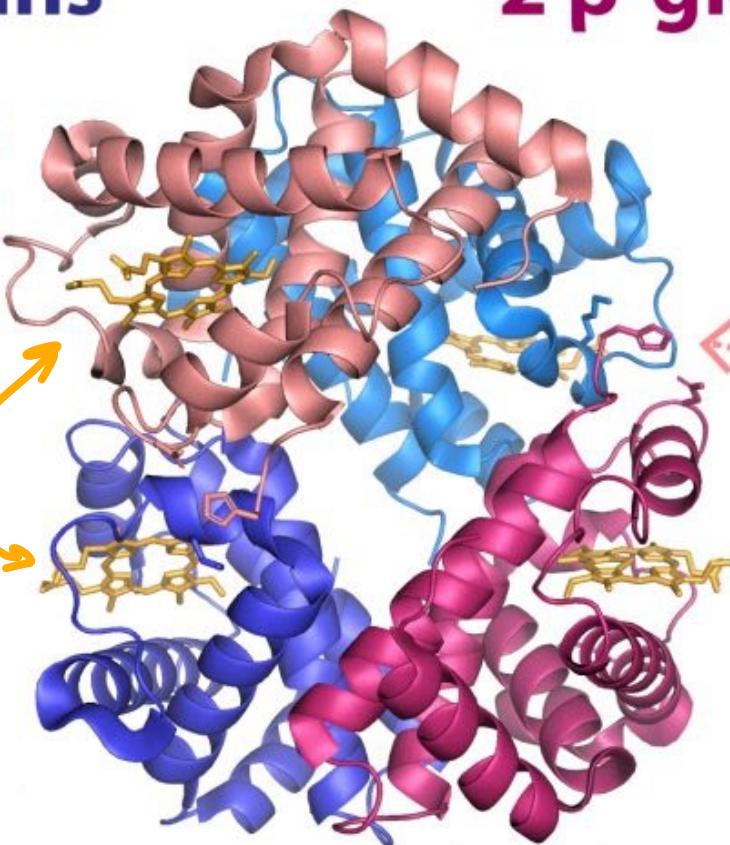
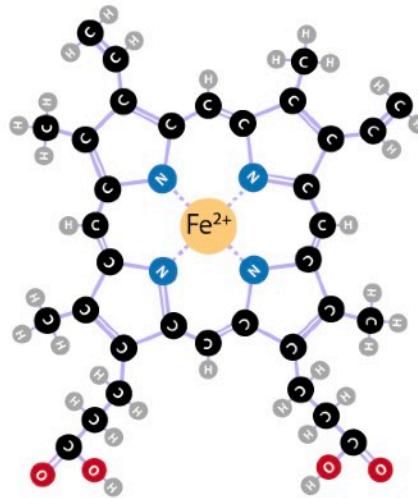
The oxygen shuttle.
Picks up O₂ at
the lungs and
.dumps O₂ at
the tissues.

Hemoglobin

it's made up of 4 separately-made protein chains

2 α-globin chains

it has 4 heme groups
that hold iron & help
hemoglobin bind &
transport oxygen



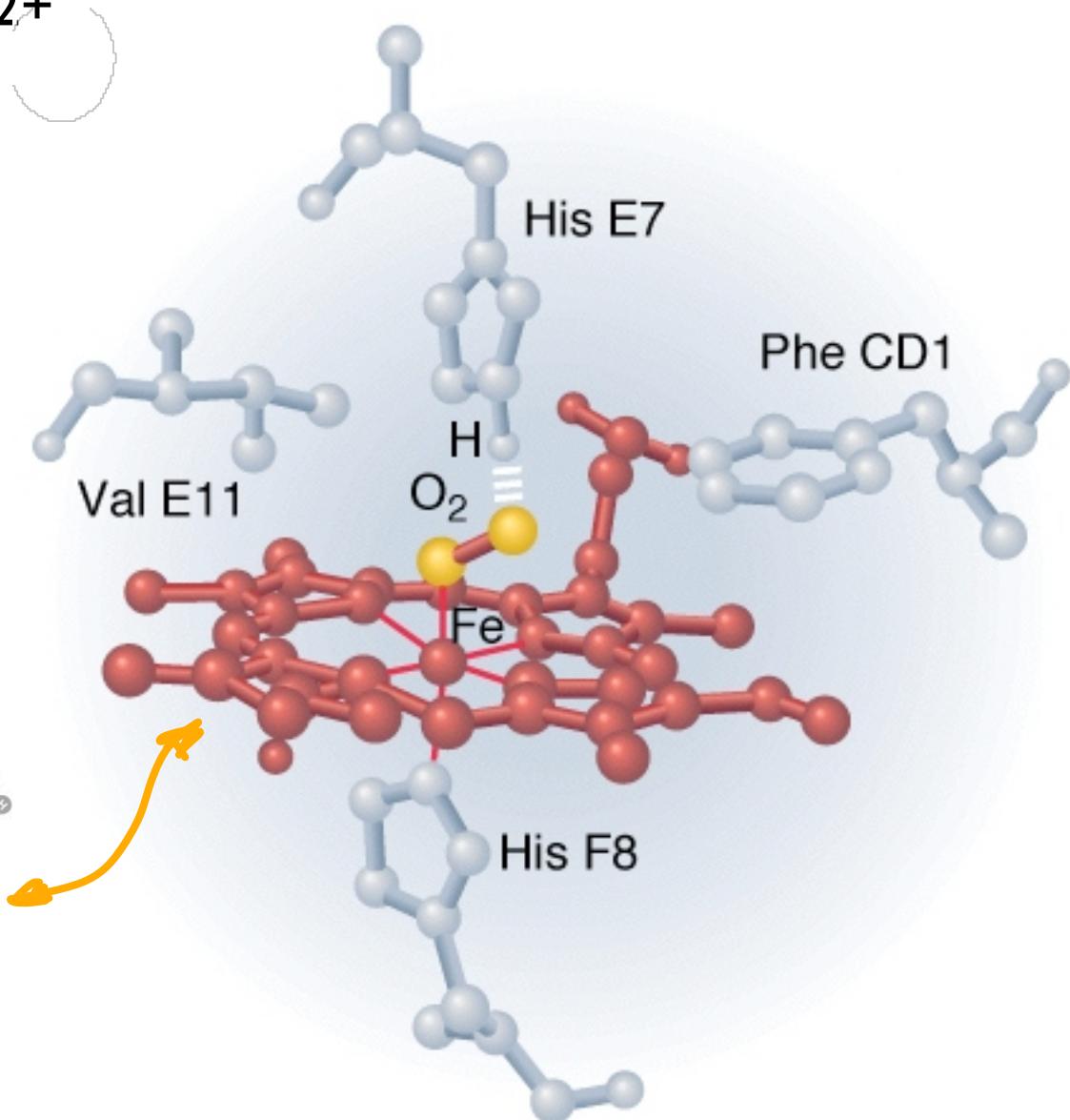
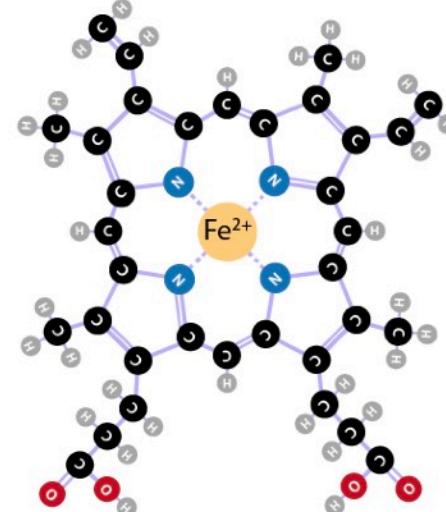
2 β-globin chains

the Bohr effect
involves
interactions
between the
chains that
happen here

these interactions can only
occur at low pH, like can be
caused by high CO_2

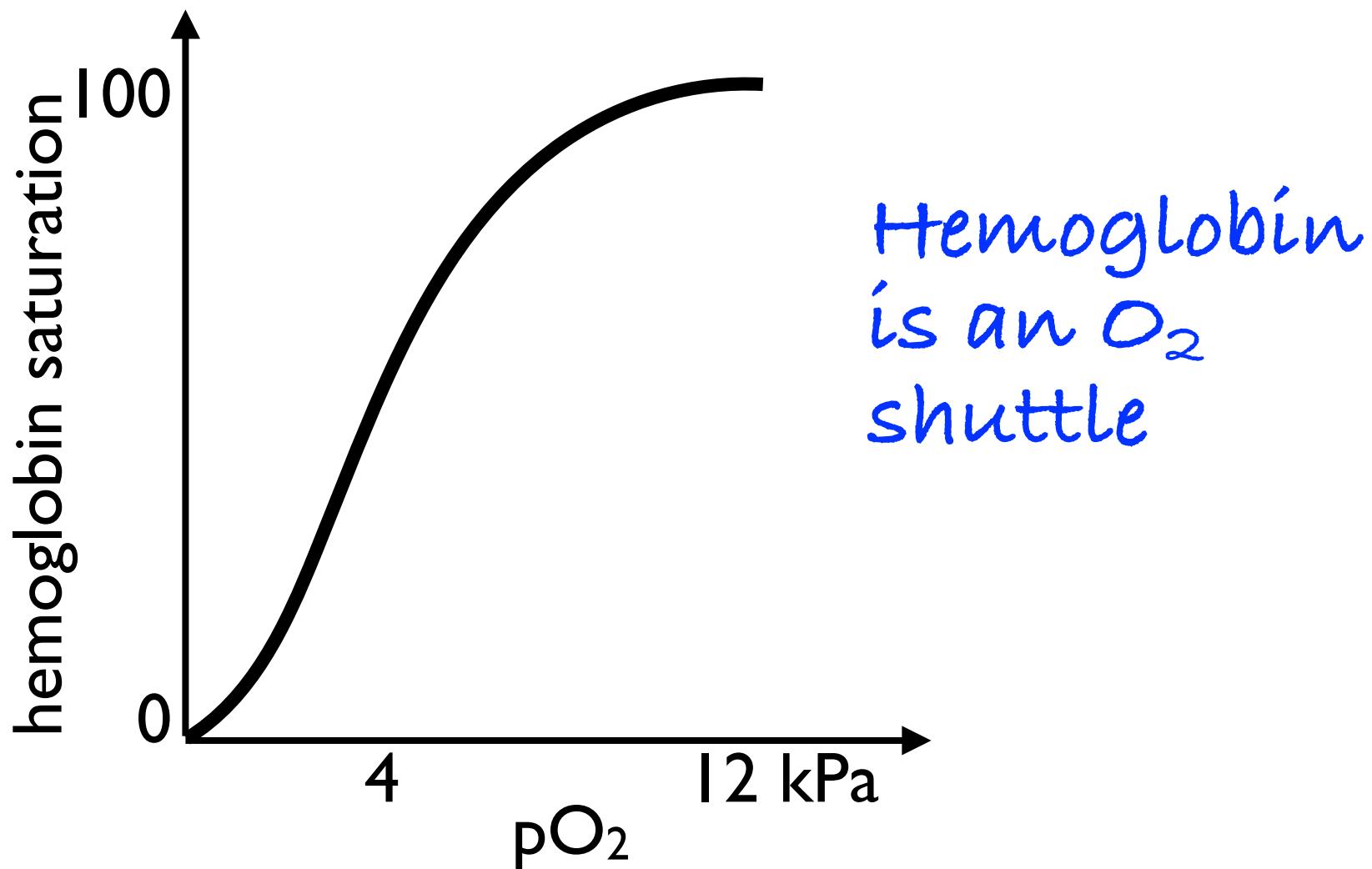
Heme Binds Oxygen Reversibly

- Heme binds O_2 via Fe^{2+}
- 4 hemes => 4 O_2 per hemoglobin
- O_2 stabilized by H-bonds with Histidine residues in globin



Subunits Bind O₂ Cooperatively

Hemoglobin has 4 subunits



Reversible, All-or-None



LOG IN



Swipe

CREATE ACCOUNT

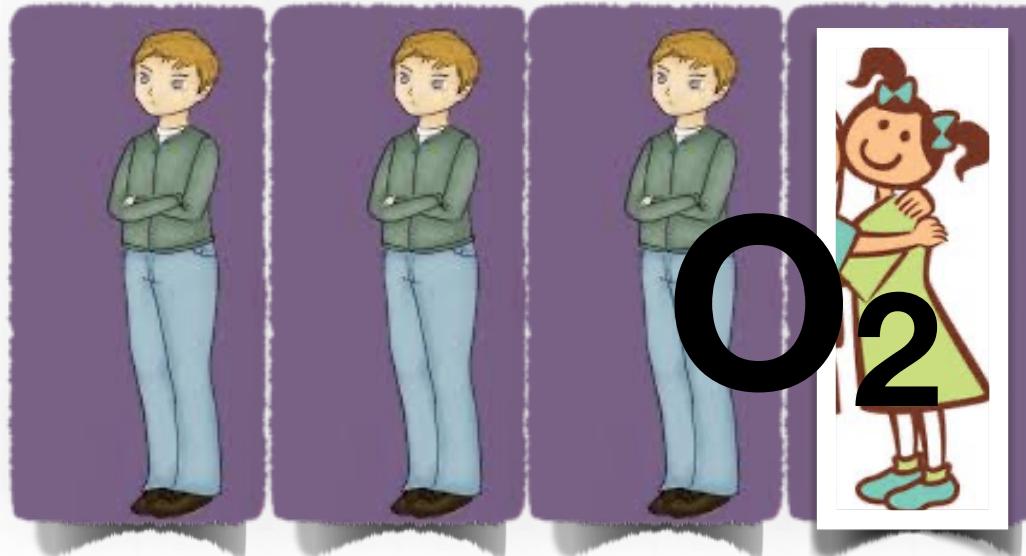


Low Affinity
“O₂ Grumpy”

High Affinity
“O₂ Sticky”

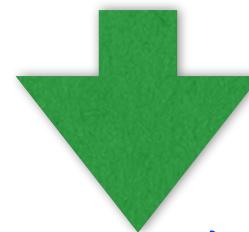
Doesn't need to
meet a lot of O₂

Subunits Bind O₂ Cooperatively



"O₂-grumpy" "O₂-sticky"

4 Subunits
if one
subunit
binds O₂



All 4 change
affinity
"O₂-sticky"

Reversible, All-or-None

What changes hemoglobin's mood?



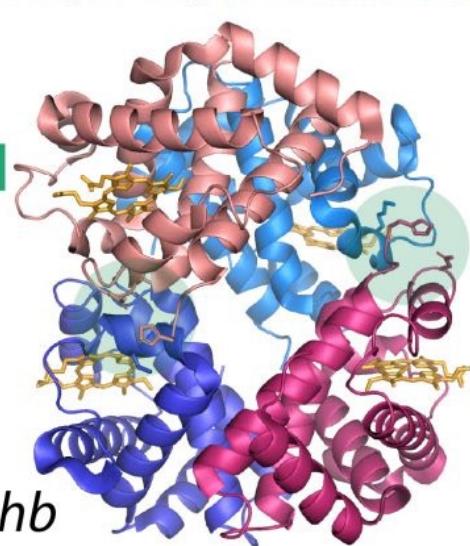
Taut
low oxygen affinity
unload



Relaxed
high oxygen affinity
load

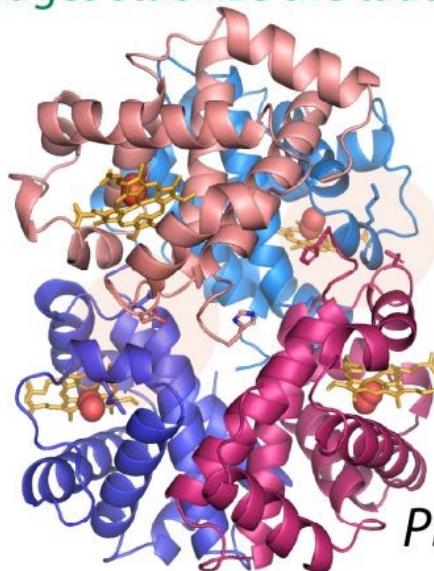
low-pH-dependent salt bridges stabilize the taut form

T



PDB 2hhb

R



PDB 1hho

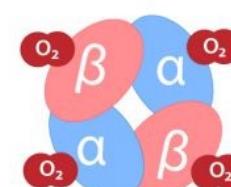
in the taut form, hemes are in a harder place for oxygen to hang out. oxygen binding kinda pulls on the heme (& attached protein) - this helps promote the other sites to bind oxygen too, giving you positive cooperativity, so that

this

is much easier than

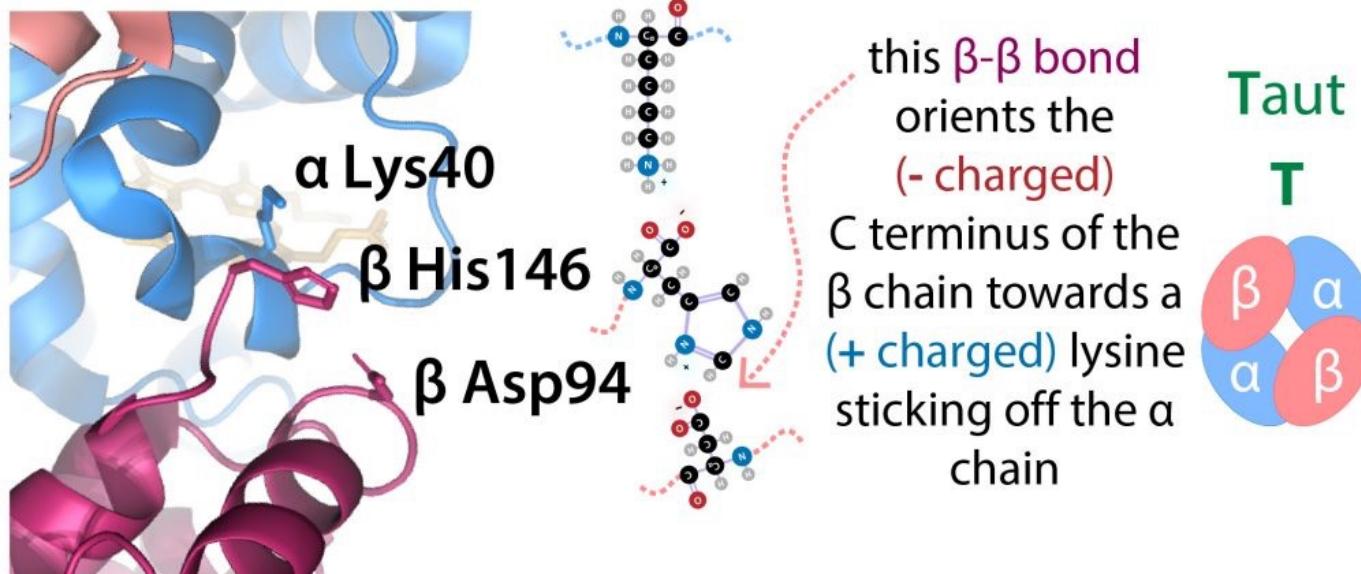


this

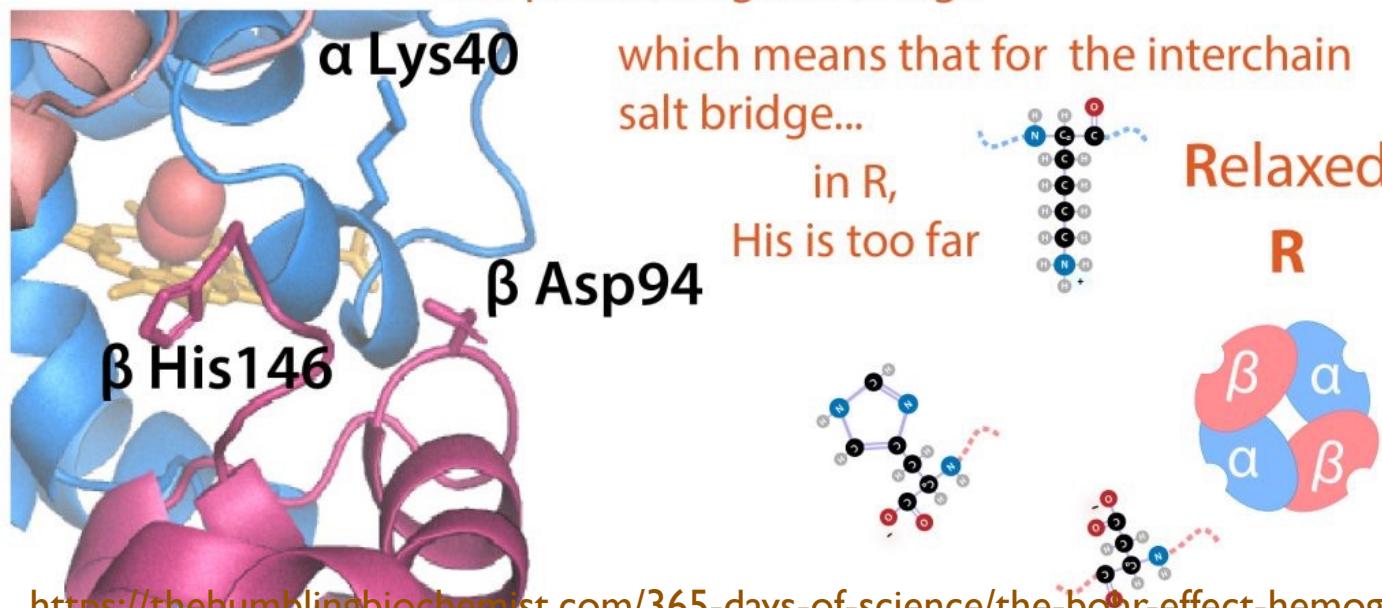


What changes hemoglobin's mood?

In the T form, salt bridges (ionic bonds) form between the chains



In the R form, His is deprotonated, so it can't form that His-positioning salt bridge



At low pH
=> lots H^+

His- residue can
form salt bridge

Blocking O_2
binding

His-
deprotonated,
blocking salt
bridge

O_2 can bind

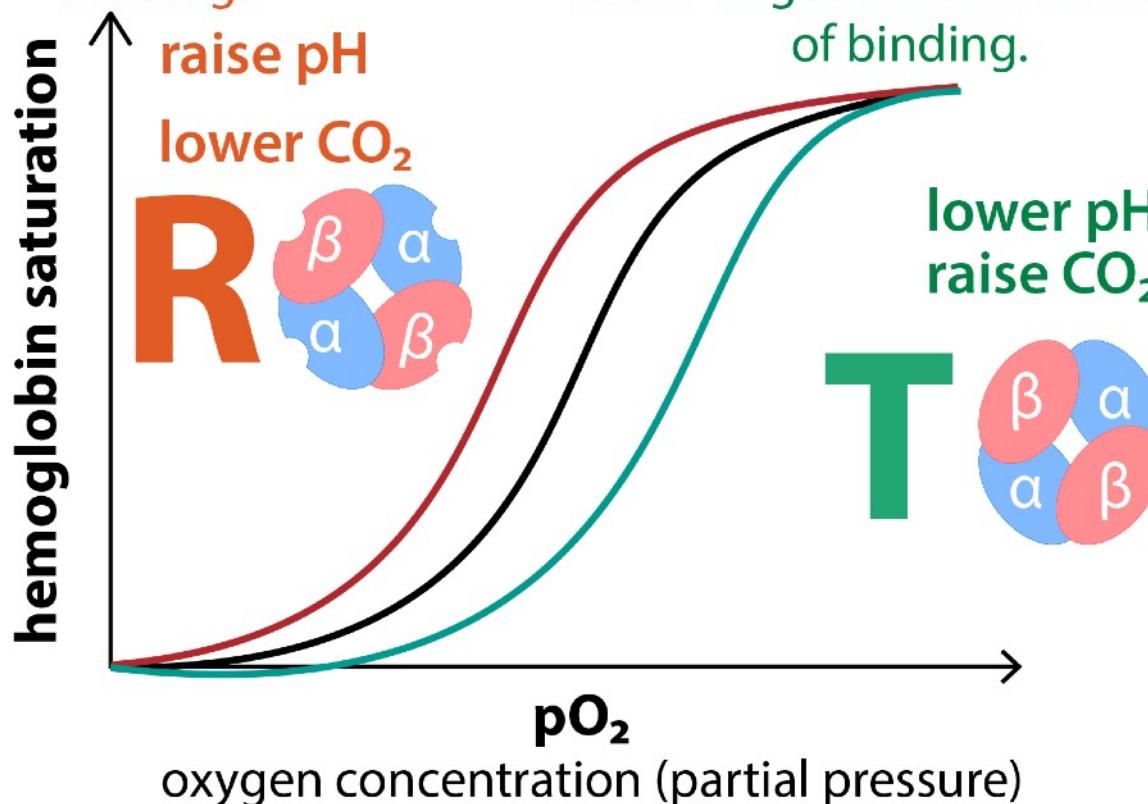
shifting the oxyhemoglobin dissociation curve

higher affinity for oxygen

they're more likely to stick each time they meet, so they don't need to meet as much in order to get the same amount of binding.

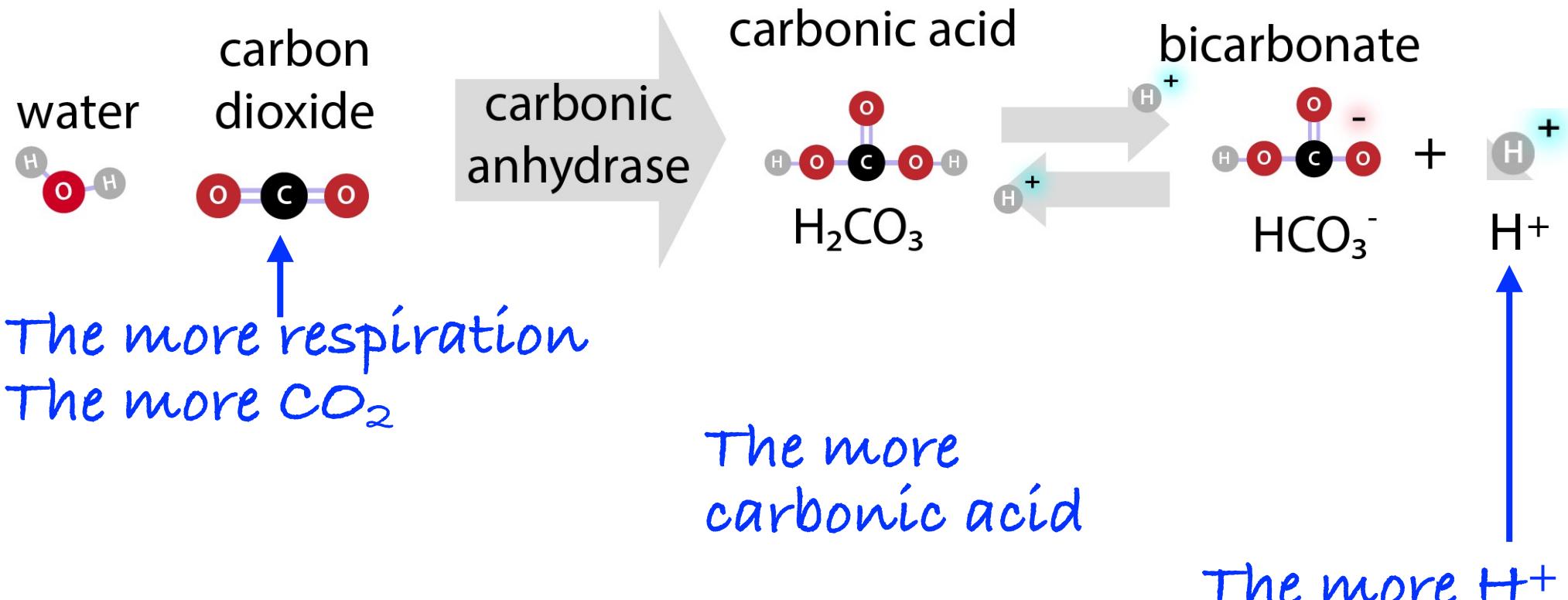
lower affinity for oxygen

each time oxygen & hemoglobin meet they have a lower probability of sticking. So they need to meet more in order to get the same amount of binding.



The Bohr effect

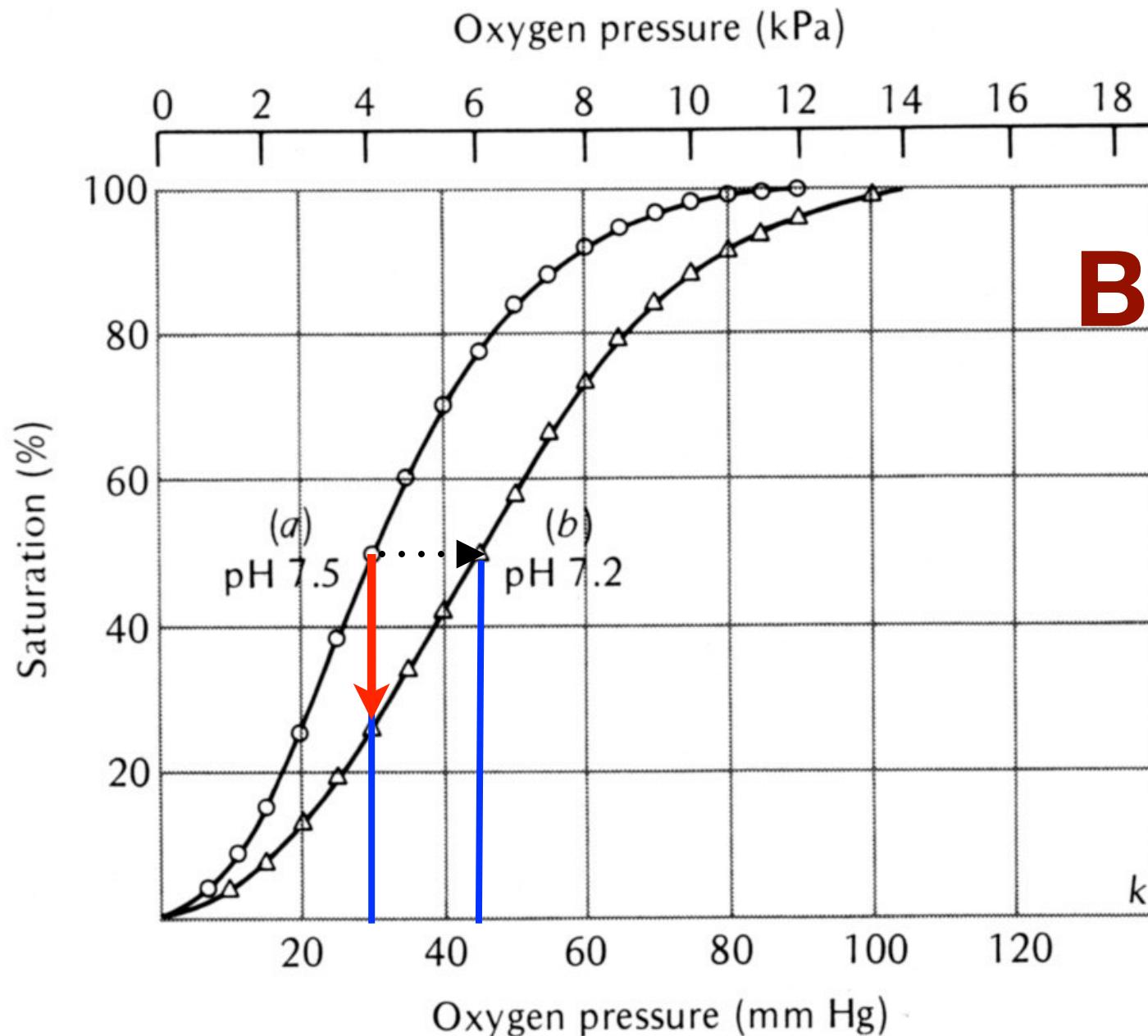
Why CO_2 and pH?



$$\text{pH} = \log_{10}\left(\frac{1}{[\text{H}^+]}\right)$$

CO_2 Acidifies blood
Lowers pH!

Subunits Bind O₂ Cooperatively



Bohr-Shift

CO₂ and pH
enhances
dumping of
oxygen at the
tissues

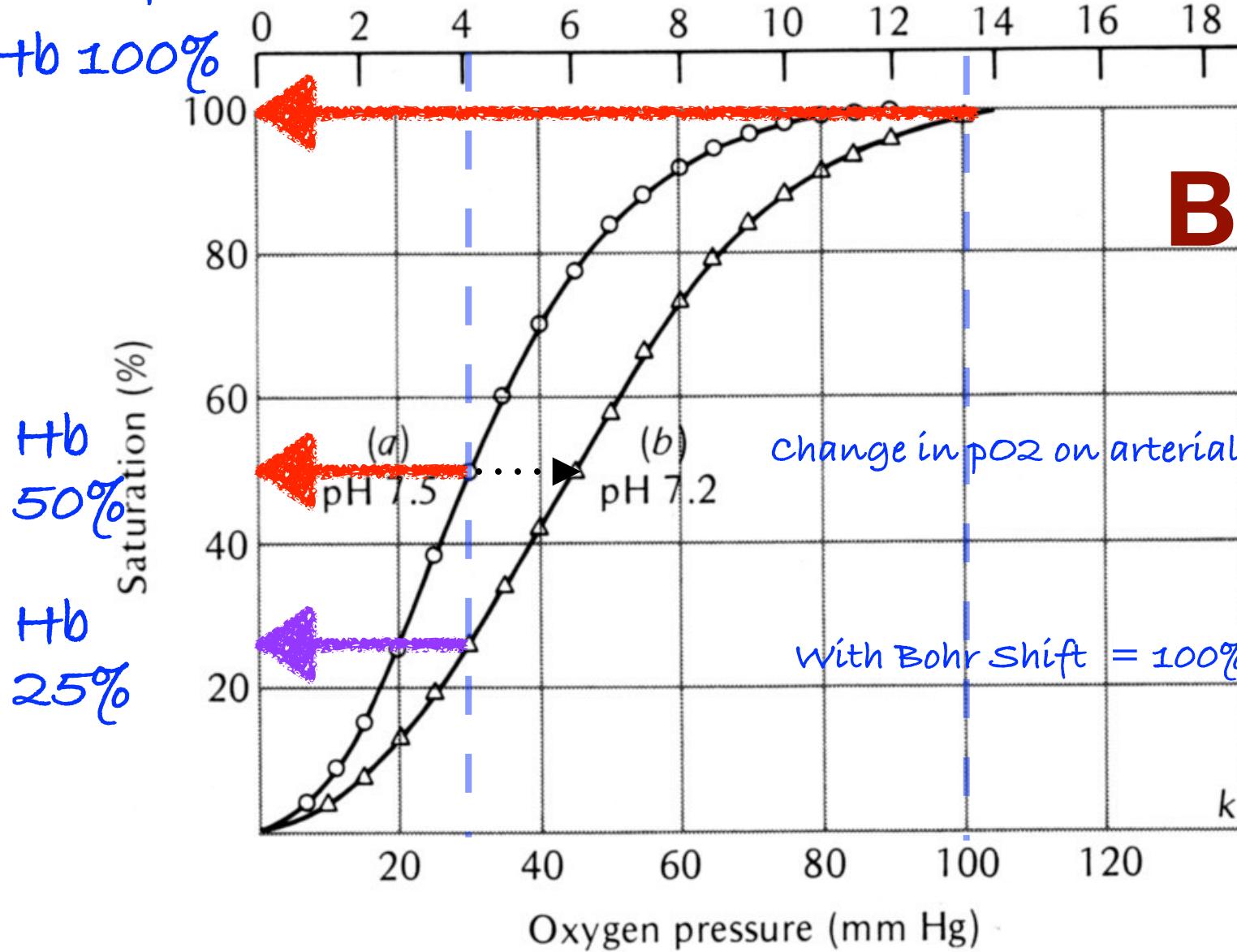
Reversible, All-or-None

Subunits Bind O₂ Cooperatively

Example:

Hb 100%

Oxygen pressure (kPa)



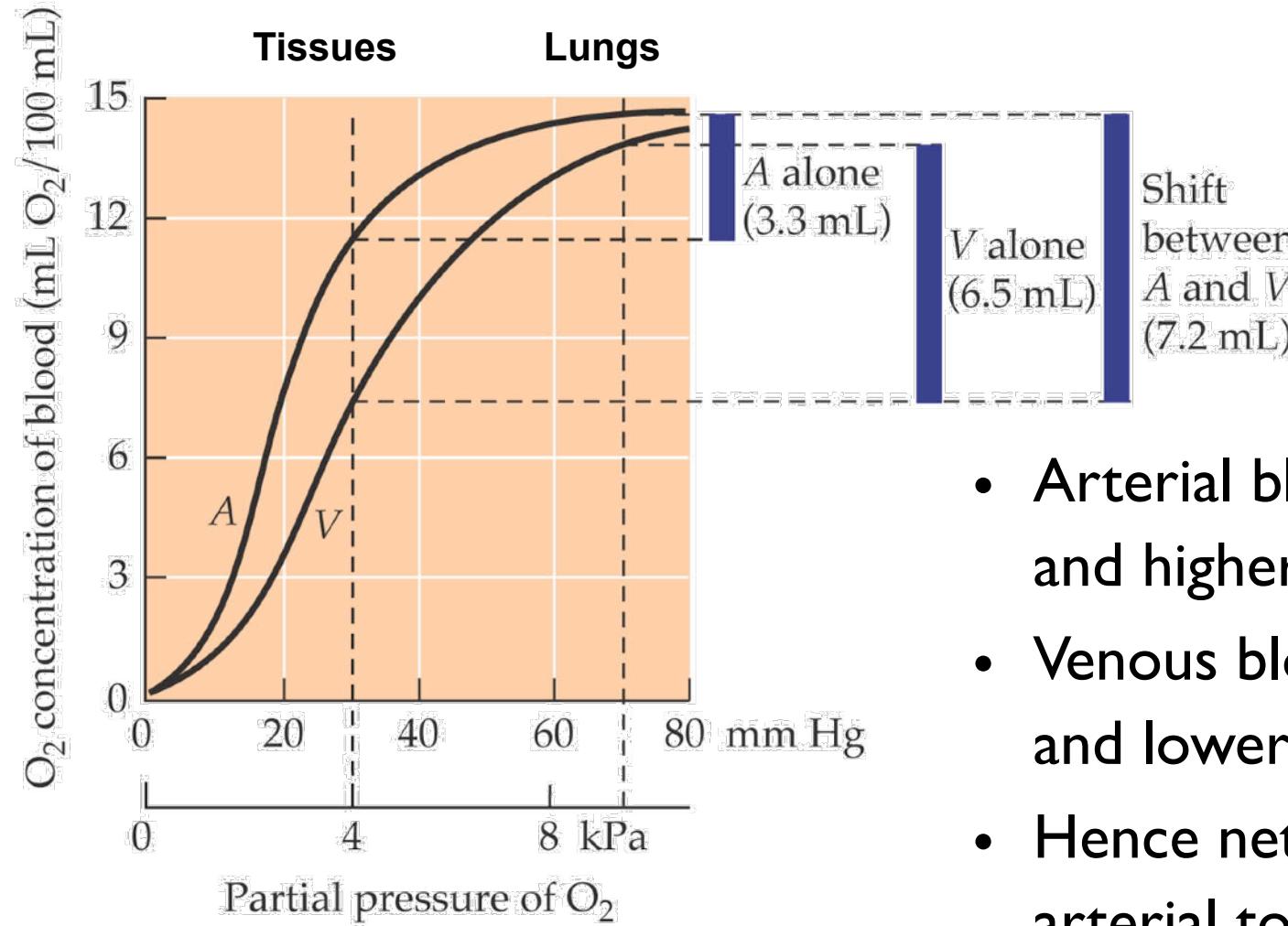
Bohr-Shift

Change in pO₂ on arterial system = 100% - 50%
= 50% delivery

With Bohr Shift = 100% - 25% = 75% delivery

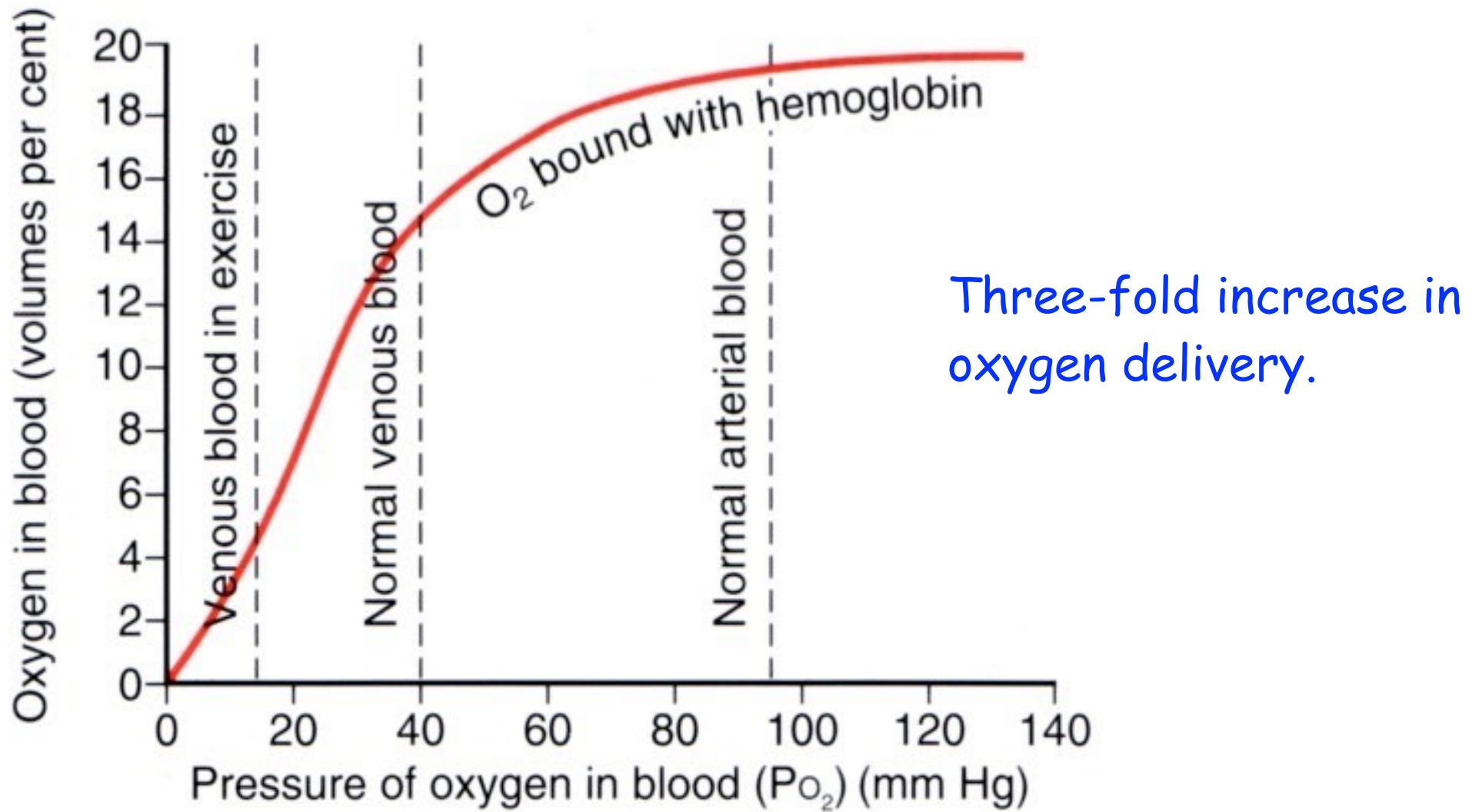
Reversible, All-or-None

Bohr Effect enhances O_2 delivery to tissues



- Arterial blood has lower CO_2 and higher pH than tissues
- Venous blood has higher CO_2 and lower pH than tissues
- Hence net O_2 delivery from arterial to venous blood enhanced by Bohr effect beyond expectations due to PO_2 alone.

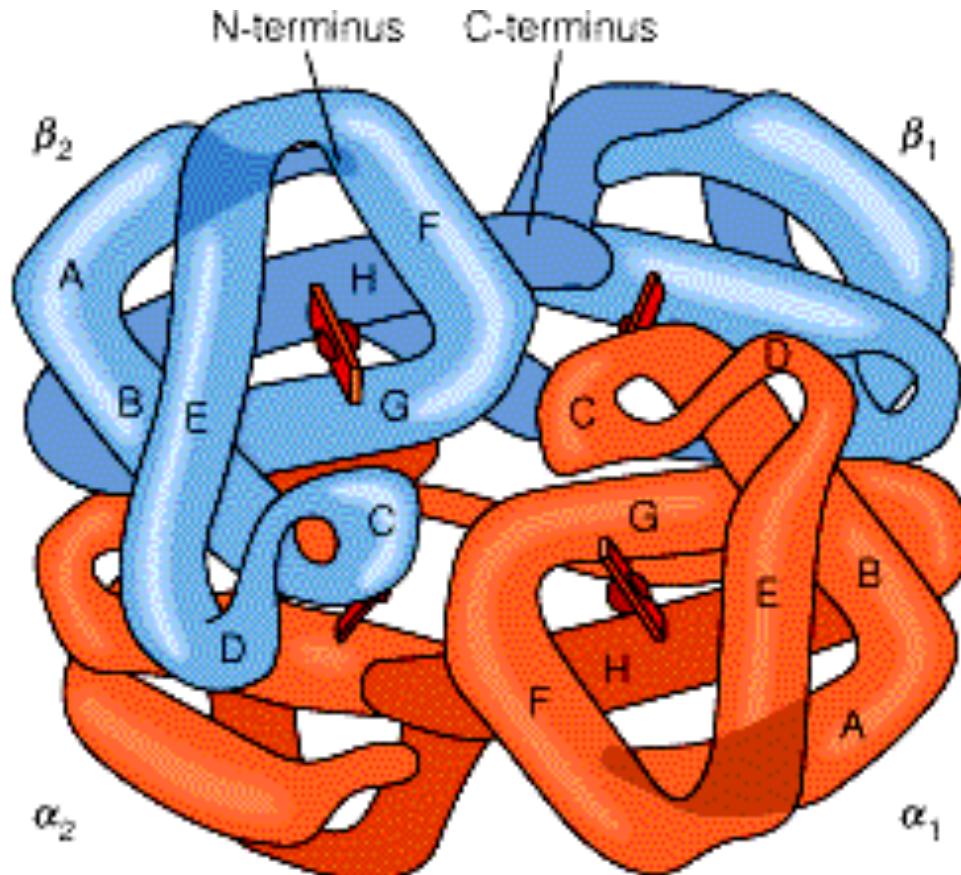
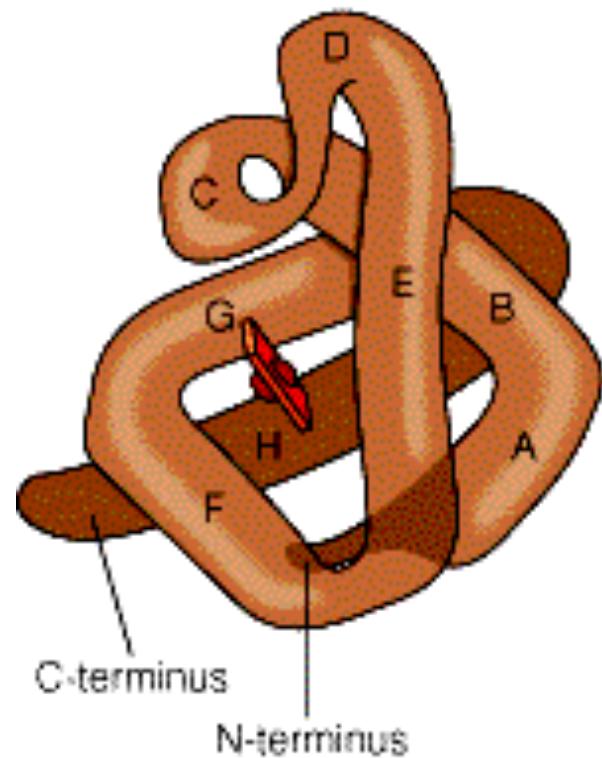
Effect of Exercise on Oxygen Delivery



Respiratory Pigments

Myoglobin: Monomer

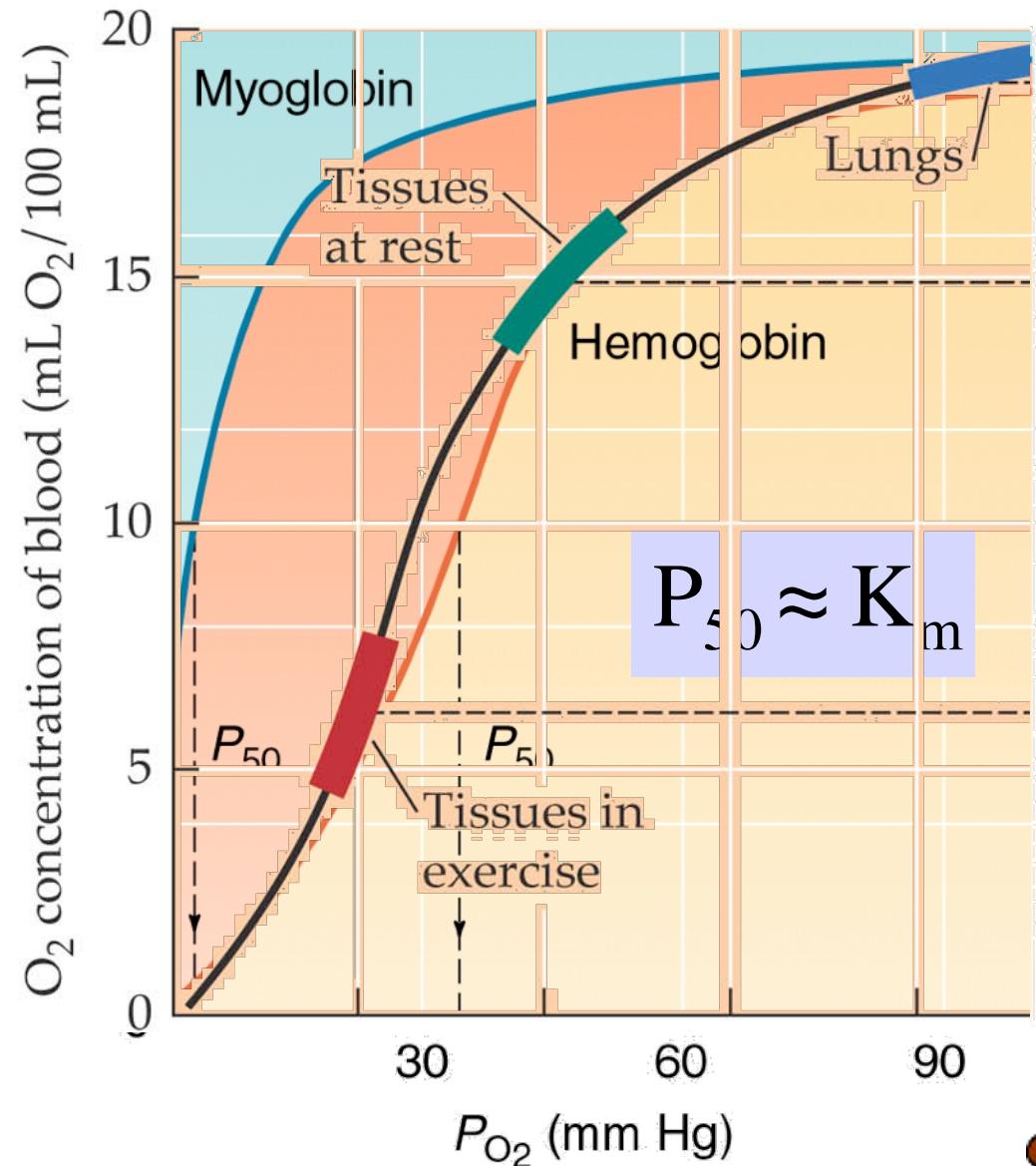
within muscle cells



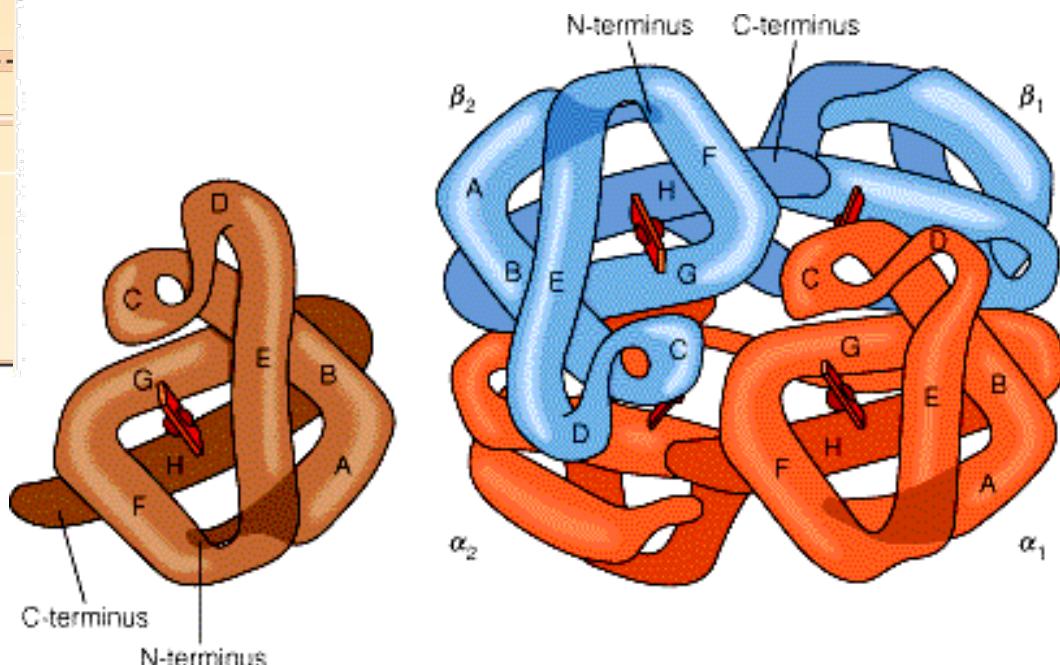
Hemoglobin: Tetramer Heterodimer: ($\alpha_1\beta_1$ & $\alpha_2\beta_2$)

within red blood cells

Oxygen binding of Myoglobin (Mb) and Hemoglobin (Hb)

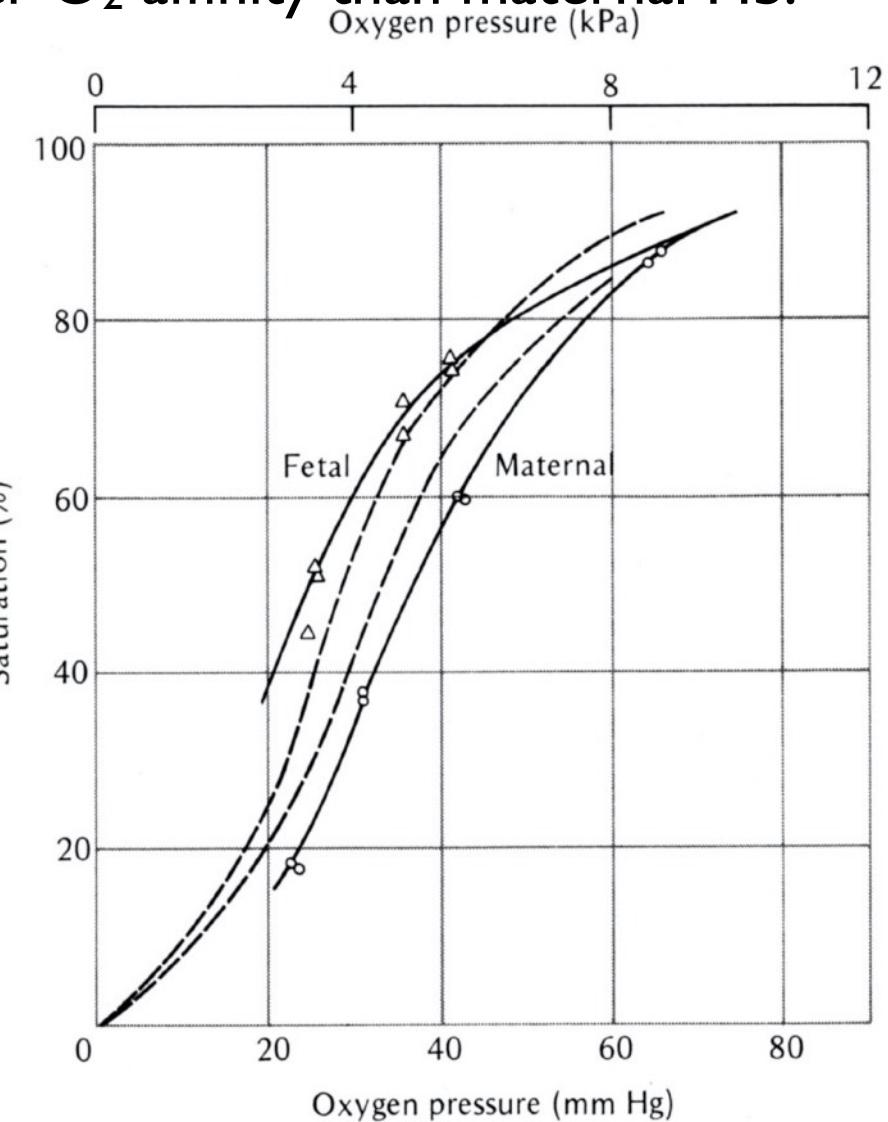
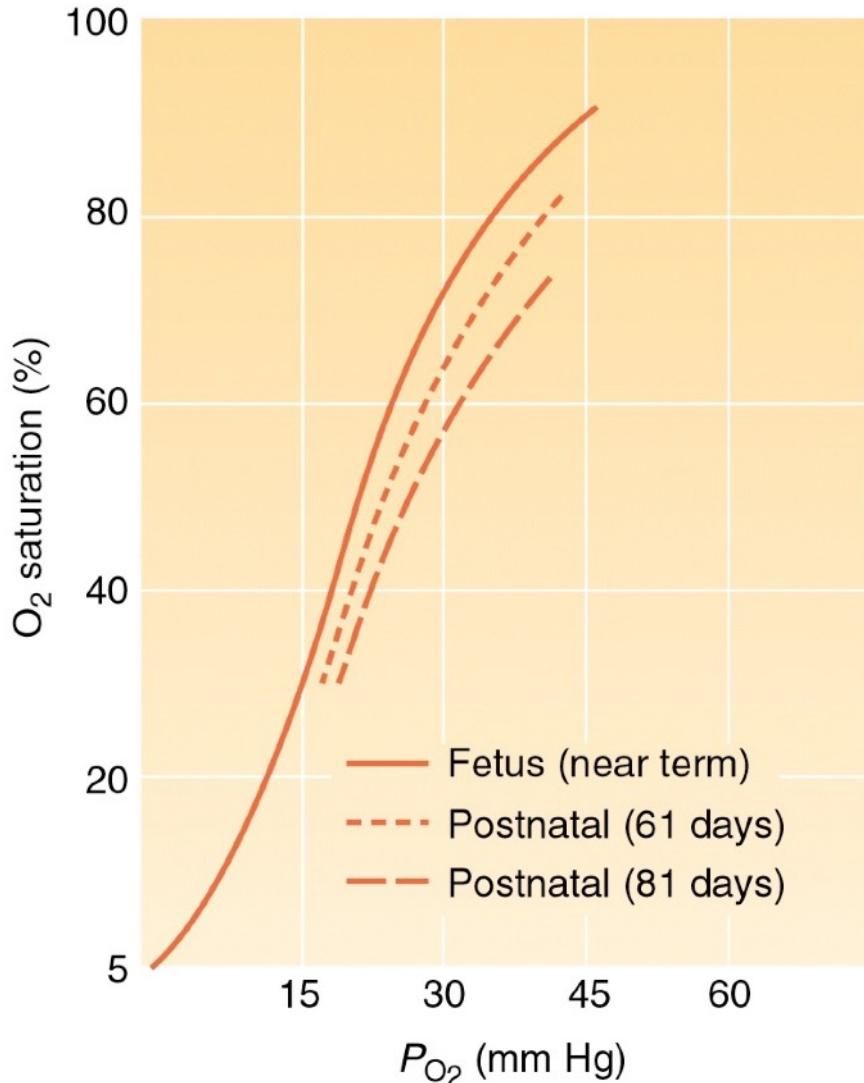


- Mb has higher affinity (lower P_{50}) than Hb
- Hb binds O₂ cooperatively
- Functional Result:
O₂ always transferred from Hb to Mb at tissues



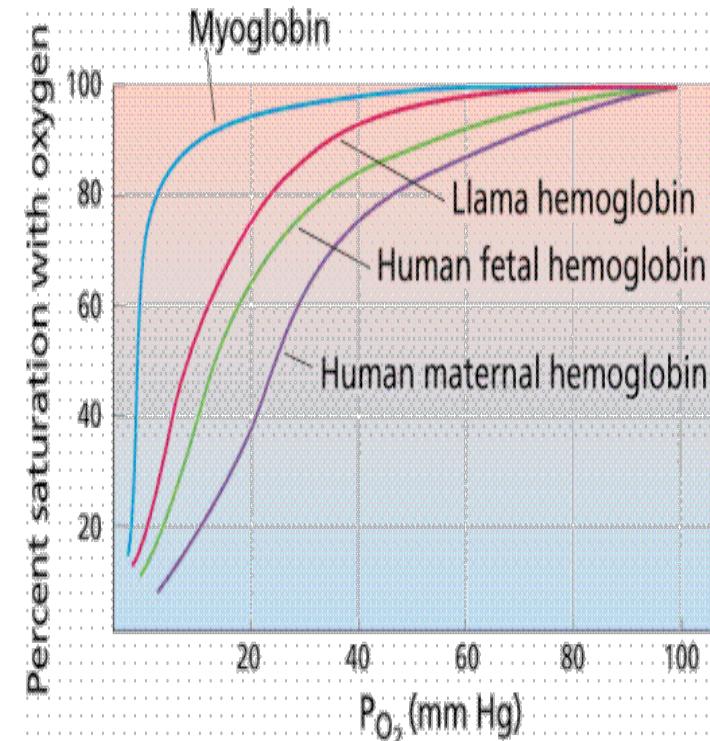
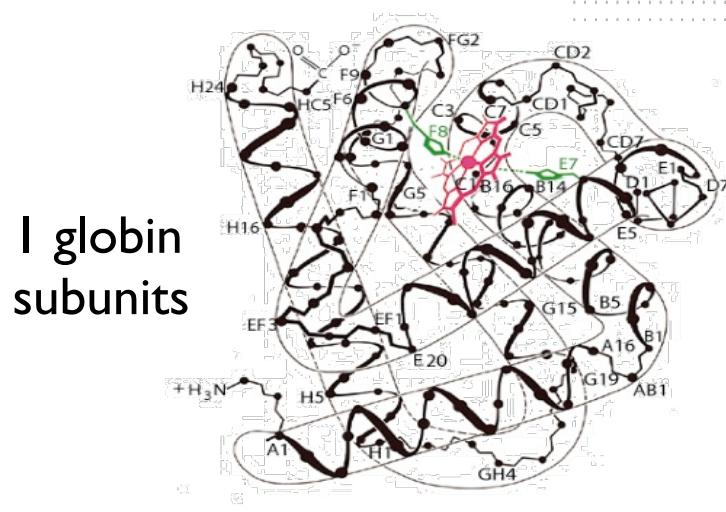
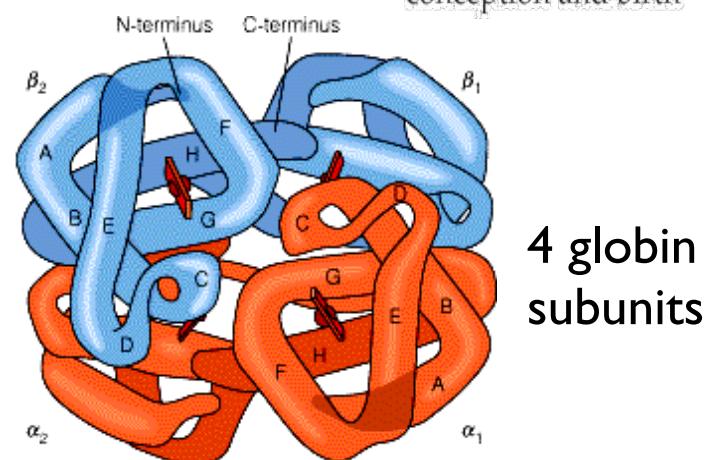
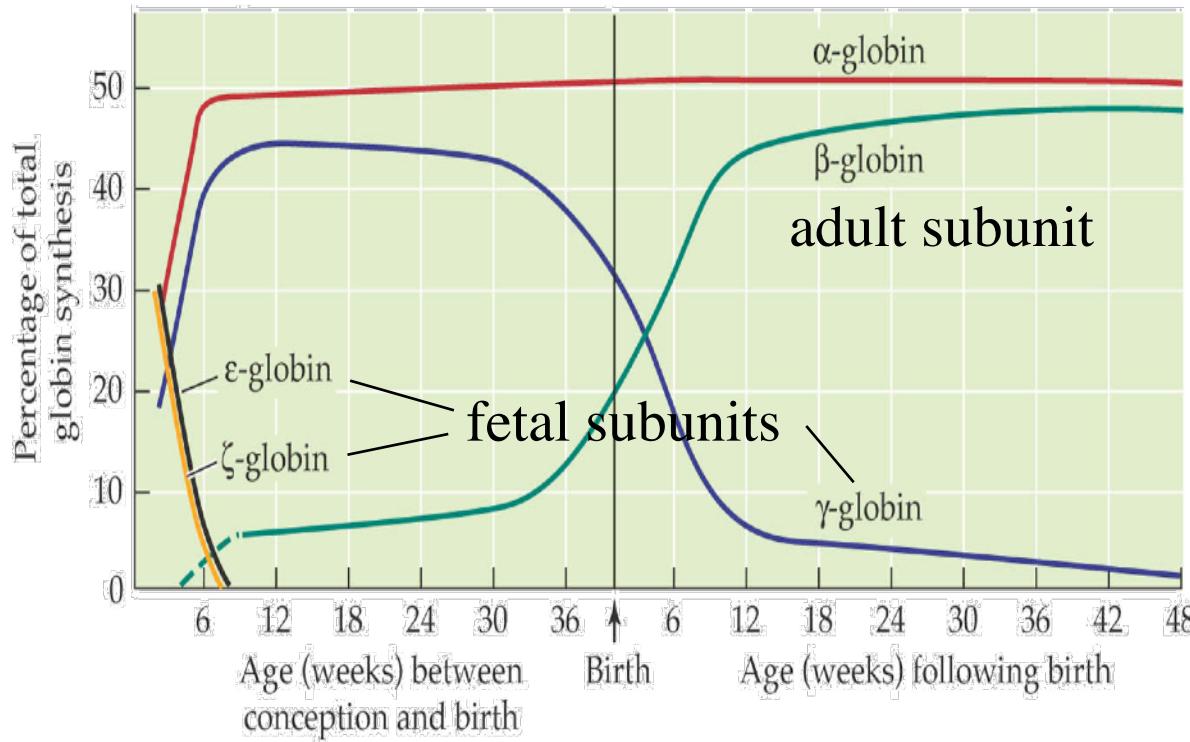
Fetal vs. Maternal Blood

- Hb subunits change with development in mammals
 - Fetal Hbs have lower P_{50} which is essential in O_2 transfer from mother to fetus
 - In other words, fetal Hb has higher O_2 affinity than maternal Hb.



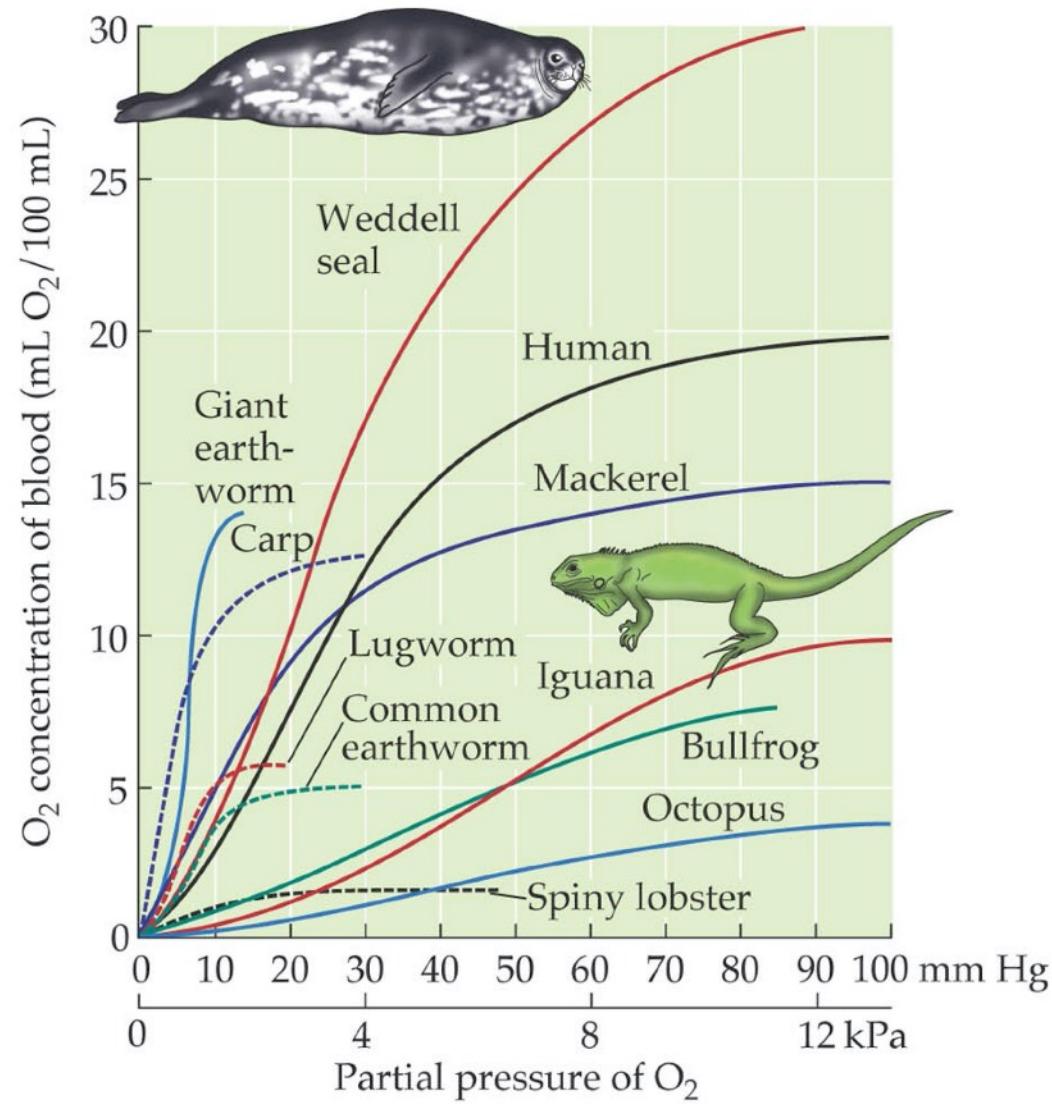
Developmental changes in Hb subunits and P_{50}

- Fetal O₂ affinity determined by globin subunit type



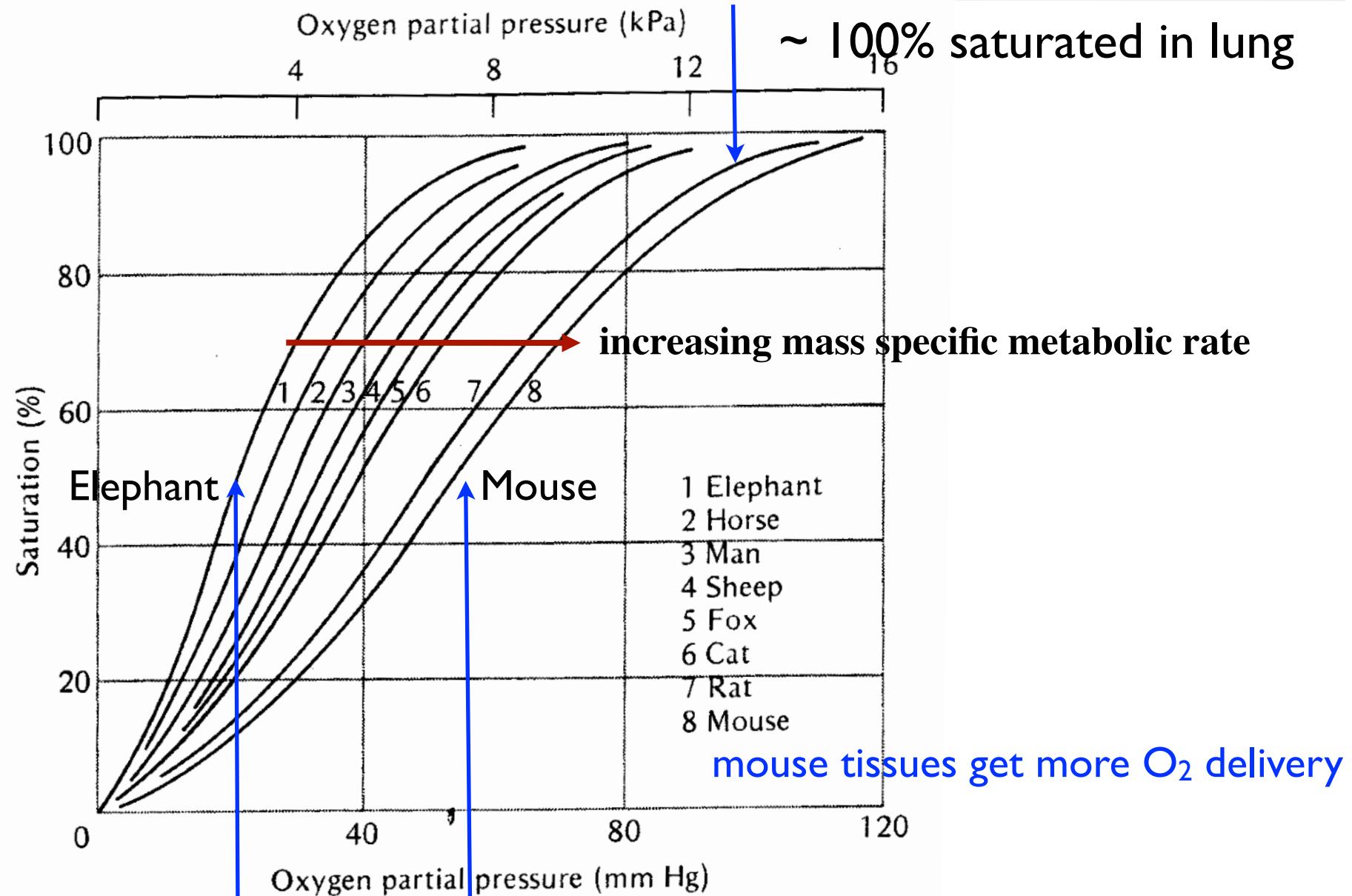
Huge interspecific variation in oxygen affinity of blood

- Adaptation to
 - Temperature
 - Hypoxia
 - Activity
- Evolutionary factors
 - Respiratory protein class



Interspecific variation of Hb function

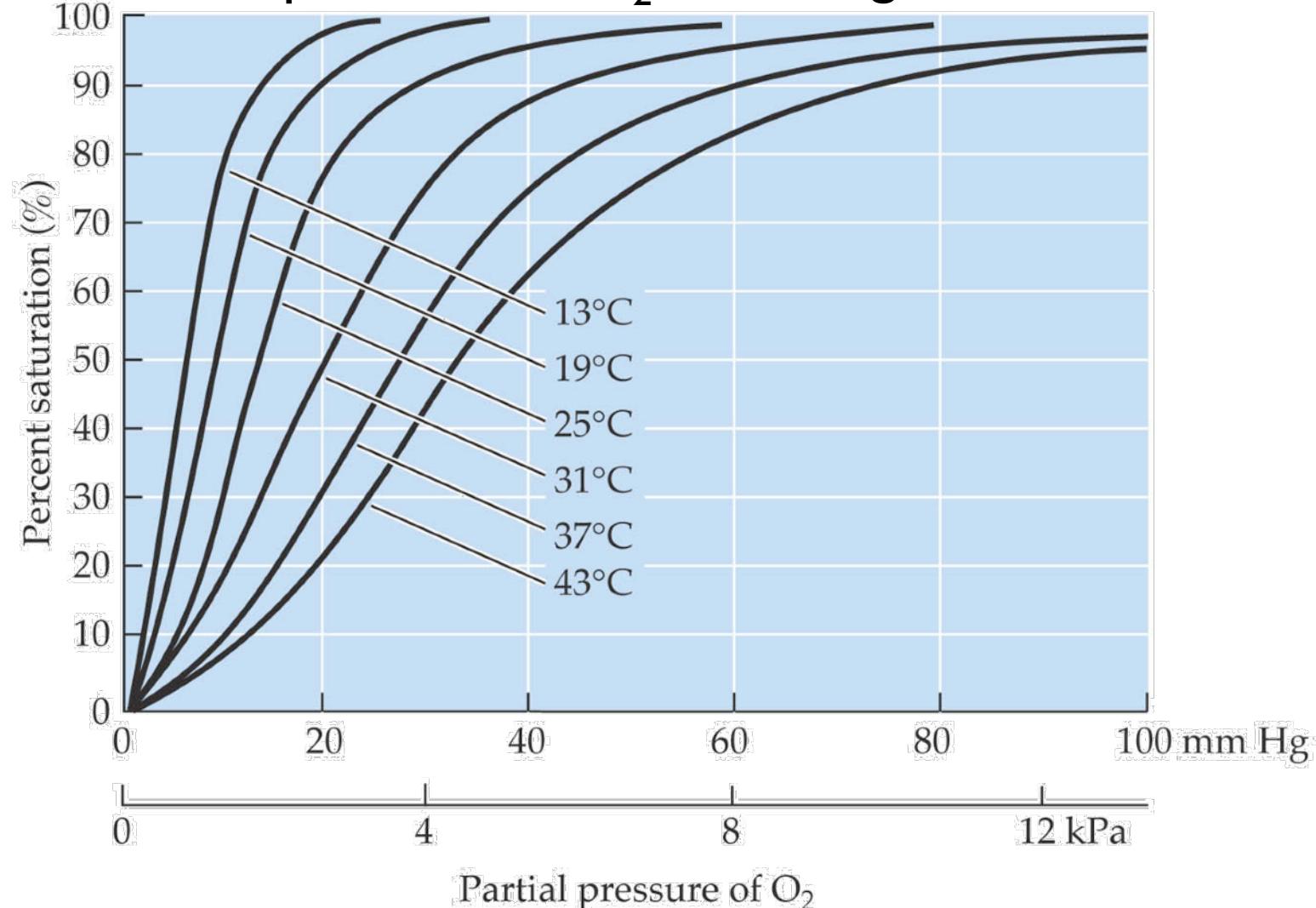
pO₂ of Mammalian Lung = 100 mm Hg (13kPa)



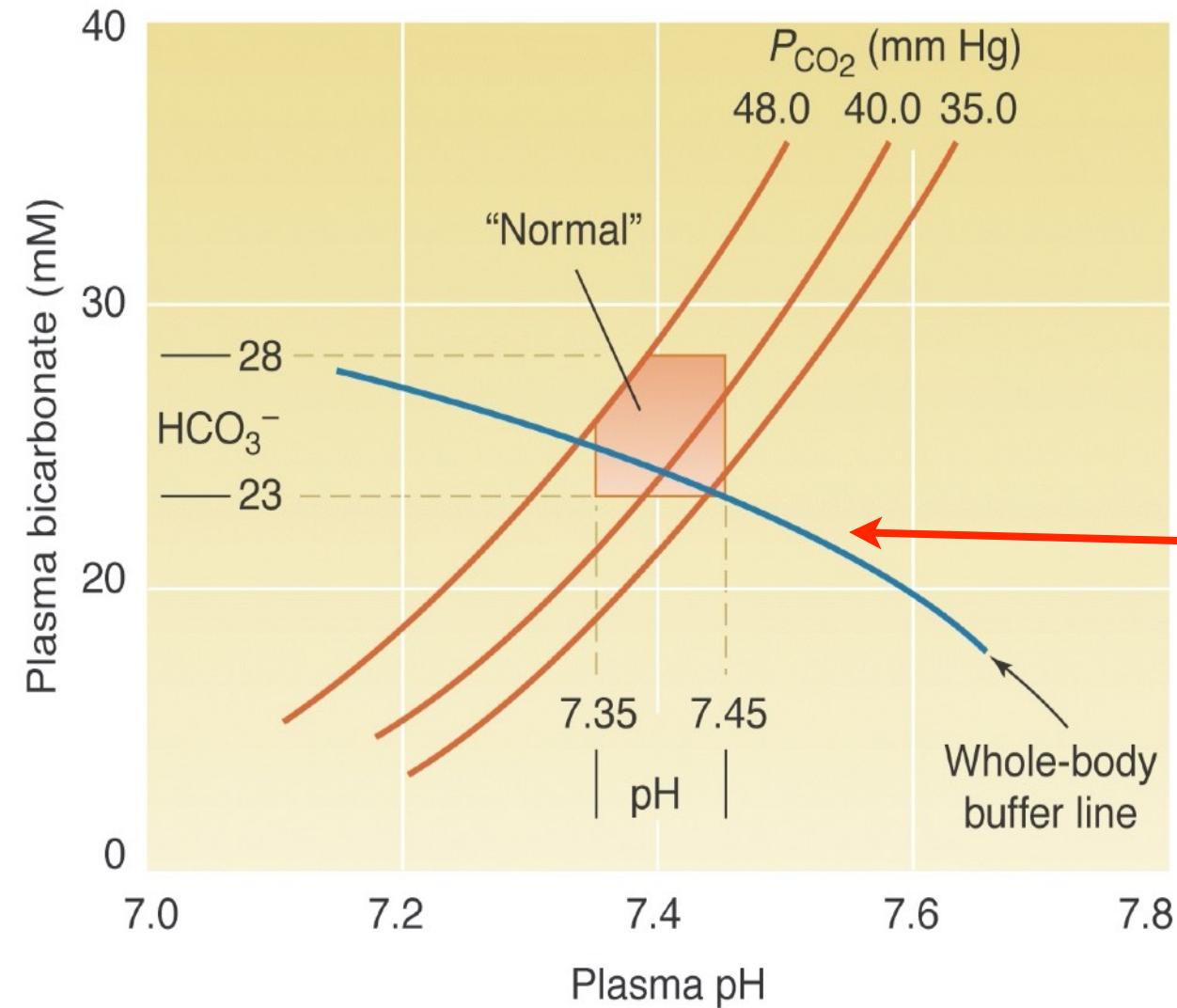
say O₂ is unloaded to tissues at 50% saturation

Increasing temperature increases Hb P₅₀

- Metabolic activity produces heat as a by-product
- Active muscles have higher temperatures than incoming blood
 - Thus, warmer temps will favor O₂ unloading



Ventilation can alter blood chemistry!



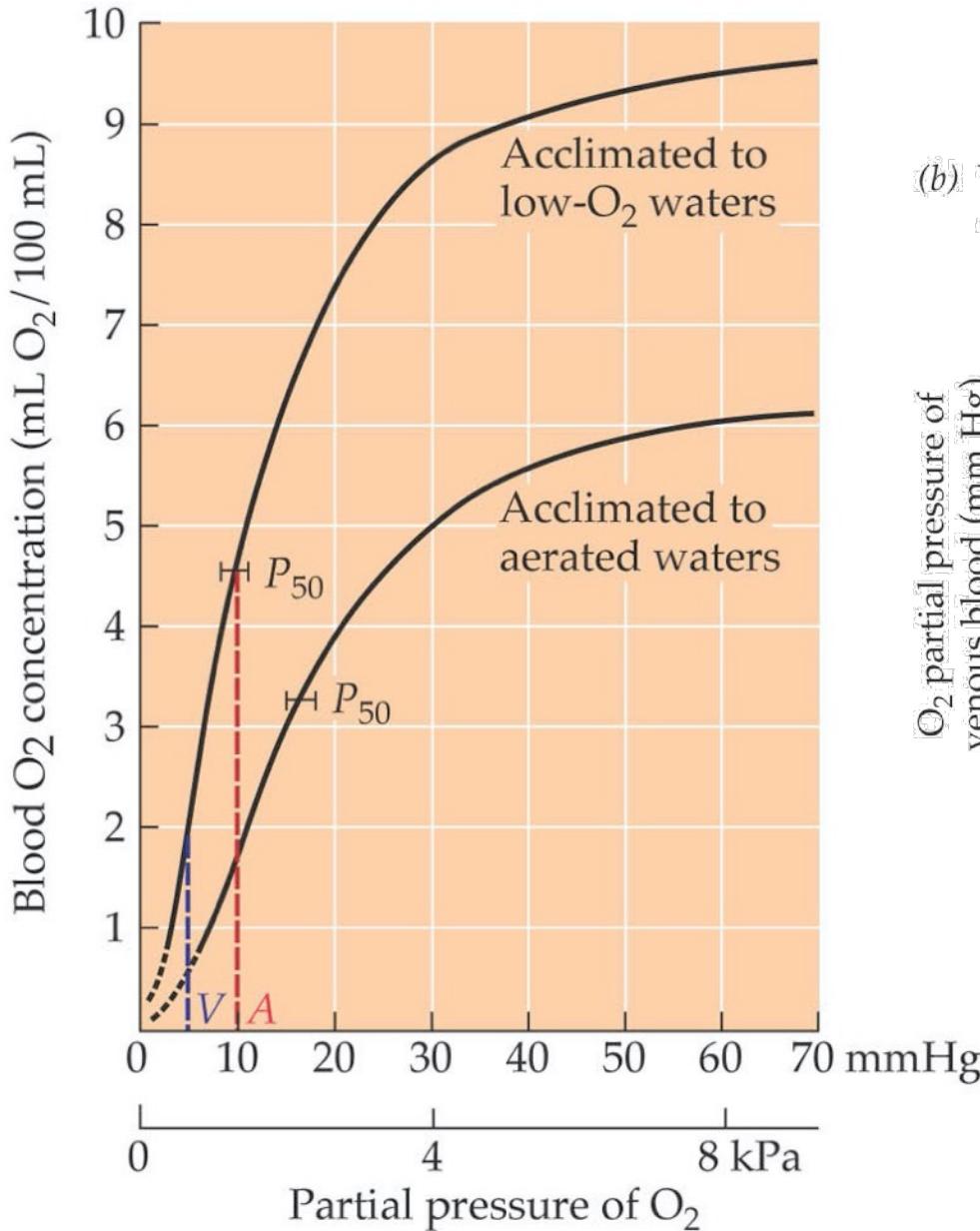
pH, Bicarb and PCO₂ in human plasma usually w/in narrow limits

hyper- and hypo- ventilation alters blood CO₂ levels. THEN plasma pH and bicarb are altered beyond the normal range

CO₂ levels are also affected by diet (Rq)

Acclimation of Hb function

Eels acclimated to hypoxic conditions



Trout acclimated to different temperatures

(b) Venous O₂ partial pressure monitored remotely with cutting-edge technology

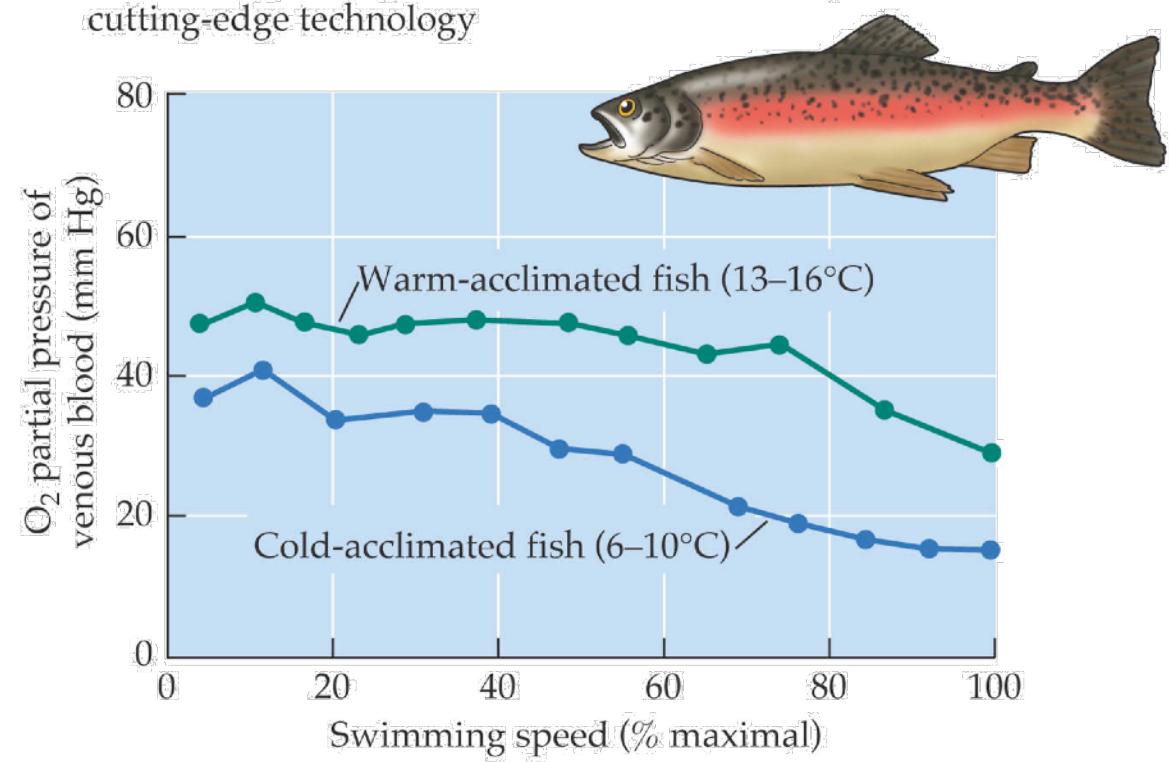


TABLE 2.2 Common respiratory pigments and examples of their occurrence in animals.

Pigment	Description	Molecular weight	Occurrence in animals
Hemocyanin	Copper-containing protein, carried in solution	300 000–9 000 000	Molluscs: chitons, cephalopods, prosobranch and pulmonate gastropods, not lamellibranchs Arthropods, crabs, lobsters Arachnomorphs: <i>Limulus</i> , <i>Euscorpius</i>
Hemerythrin	Iron-containing protein, always in cells, nonporphyrin structure	108 000	Sipunculids: all species examined Polychaetes: <i>Magelona</i> Priapulids: <i>Halicryptus</i> , <i>Priapulus</i> Brachiopods: <i>Lingula</i>
Chlorocruorin	Iron-porphyrin protein, carried in solution	2 750 000	Restricted to four families of Polychaetes: Sabellidae, Serpulidae, Chlorhaemidae, Ampharetidae; prosthetic group alone found in starfish (<i>Luidia</i> , <i>Astropecten</i>)
Hemoglobin	Iron-porphyrin protein, carried in solution or in cells; most extensively distributed pigment	17 000–3 000 000	Vertebrates: almost all, except leptocephalus larvae and some Antarctic fish (<i>Chaenichthys</i>) Echinoderms: sea cucumbers Molluscs: <i>Planorbis</i> , Pismo clam (<i>Tivela</i>) Arthropods: insects (<i>Chironomus</i> , <i>Gastrophilus</i>); crustacea (<i>Daphnia</i> , <i>Artemia</i>) Annelids: <i>Lumbricus</i> , <i>Tubifex</i> , <i>Arenicola</i> , <i>Spirorbis</i> (some species have hemoglobin, some chlorocruorin, others no blood pigment), <i>Serpula</i> (both hemoglobin and chlorocruorin) Nematodes: <i>Ascaris</i> Flatworms: parasitic trematodes Protozoa: <i>Paramecium</i> , <i>Tetrahymena</i> Plants: yeasts, <i>Neurospora</i> , root nodules of leguminous plants (clover, alfalfa)

Schmidt-Neilsen (1997)

Evolutionary loss of respiratory proteins

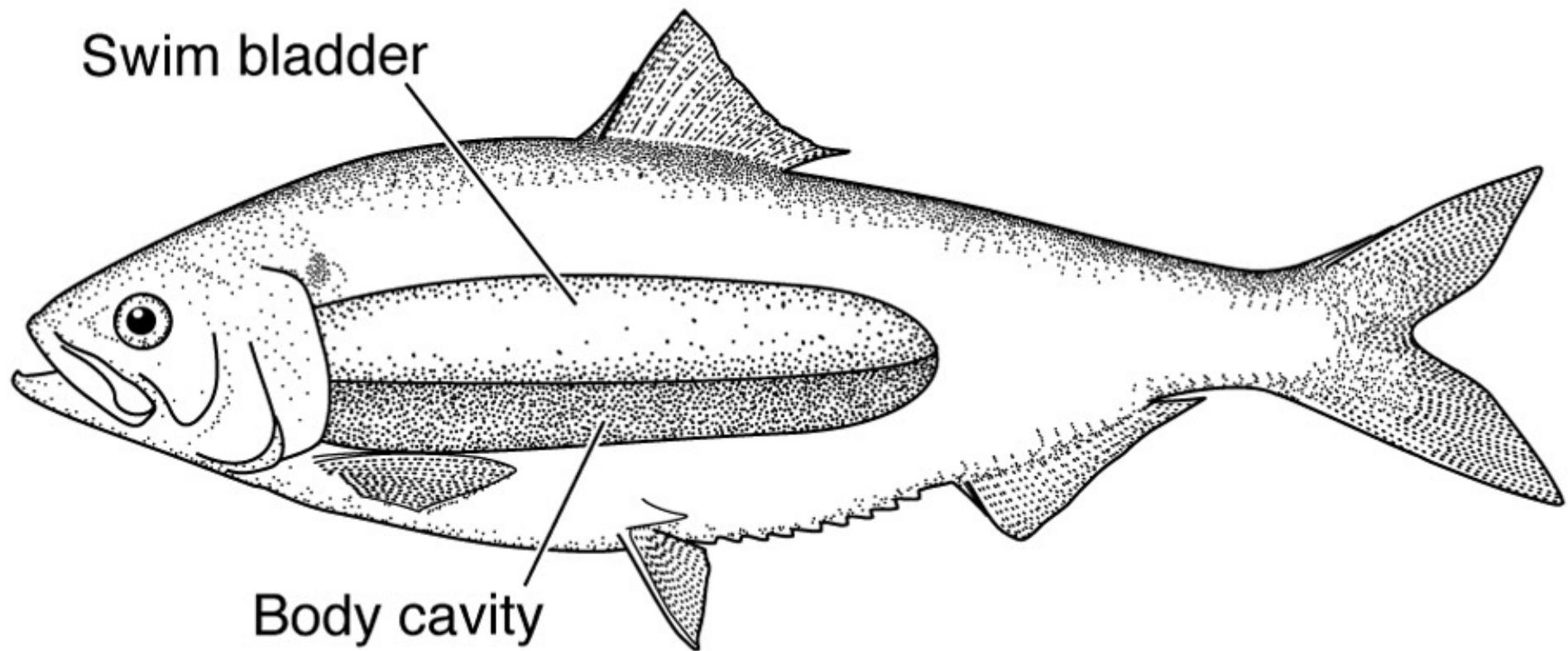
- All fishes have respiratory pigments, except one:
The Antarctic Icefish, *Chaenocephalus aceratus*



- Functional consequences - much less O_2 in blood
 - Although cold temp does incr. solubility
- Icefish adaptations:
 - Increased blood volume
 - Increased cardiac output
 - Low metabolic rate

Root Effect

Regulation of Buoyancy in Fishes with Swim Bladders/Lungs

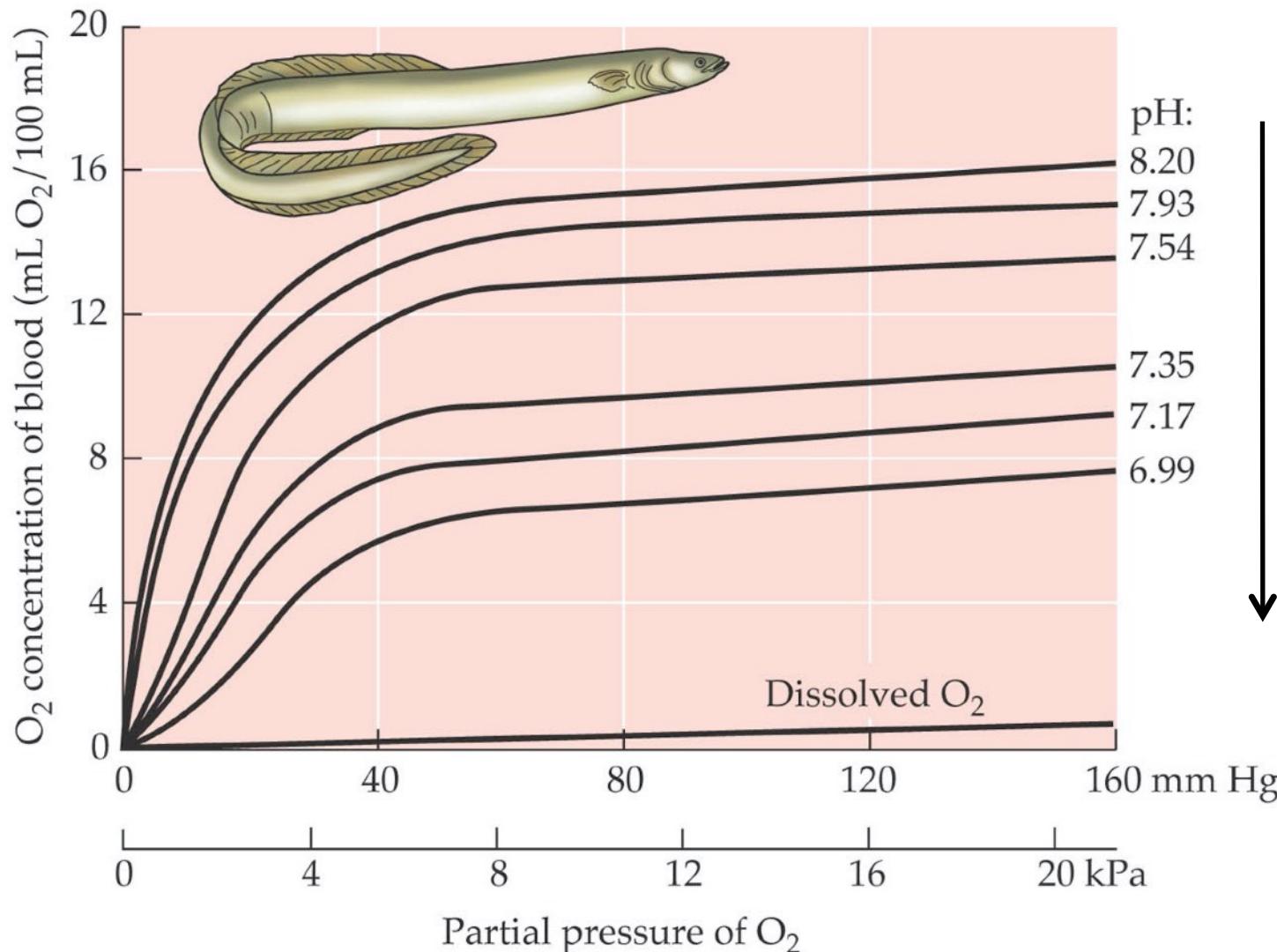


“Flesh” is slightly denser than water, so without generating any lift, a fish will sink. Most living fishes (except chondrichthyans) have swim bladders or lungs to regulate buoyancy (by controlling the amount of gas inside).

Neutral buoyancy = having the same average density as water so that animal does not sink or float in the water column.

Root Effect

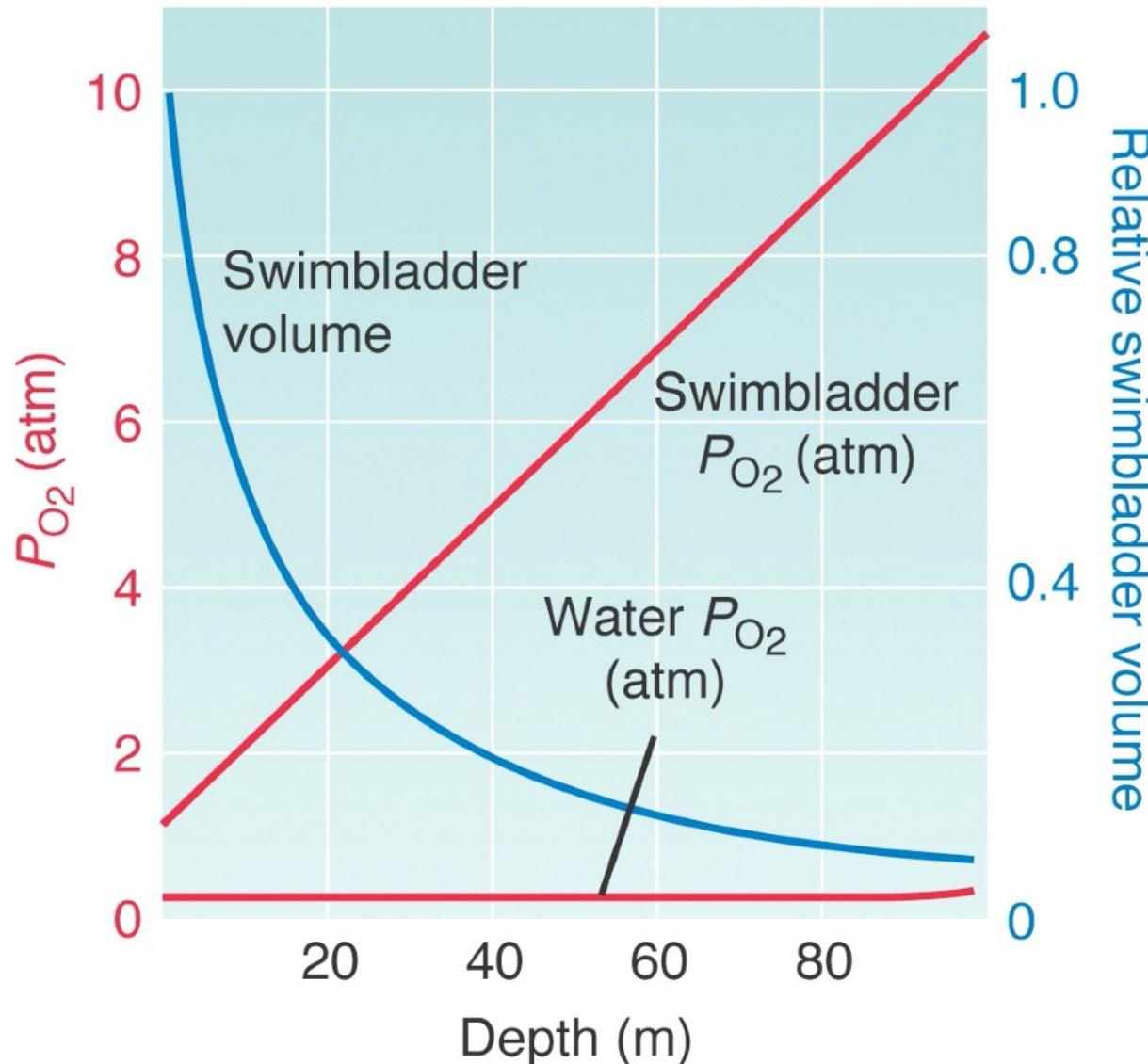
under the root effect, Hb unloads O₂



Hb O₂
carrying
capacity

Root Effect

The swim bladder challenge: working against huge O₂ gradients



- Plot assumes no change in swimbladder gas content with depth
- Result: must move O₂ into swim bladder against a pressure gradient of many atmospheres at depth

Root Effect

How to get O₂ into swim bladder against a high pressure?

say swimbladder contains O₂ at high pressure (100atm for 1000m depth), but arterial O₂ tension is no more than 0.2 atm.

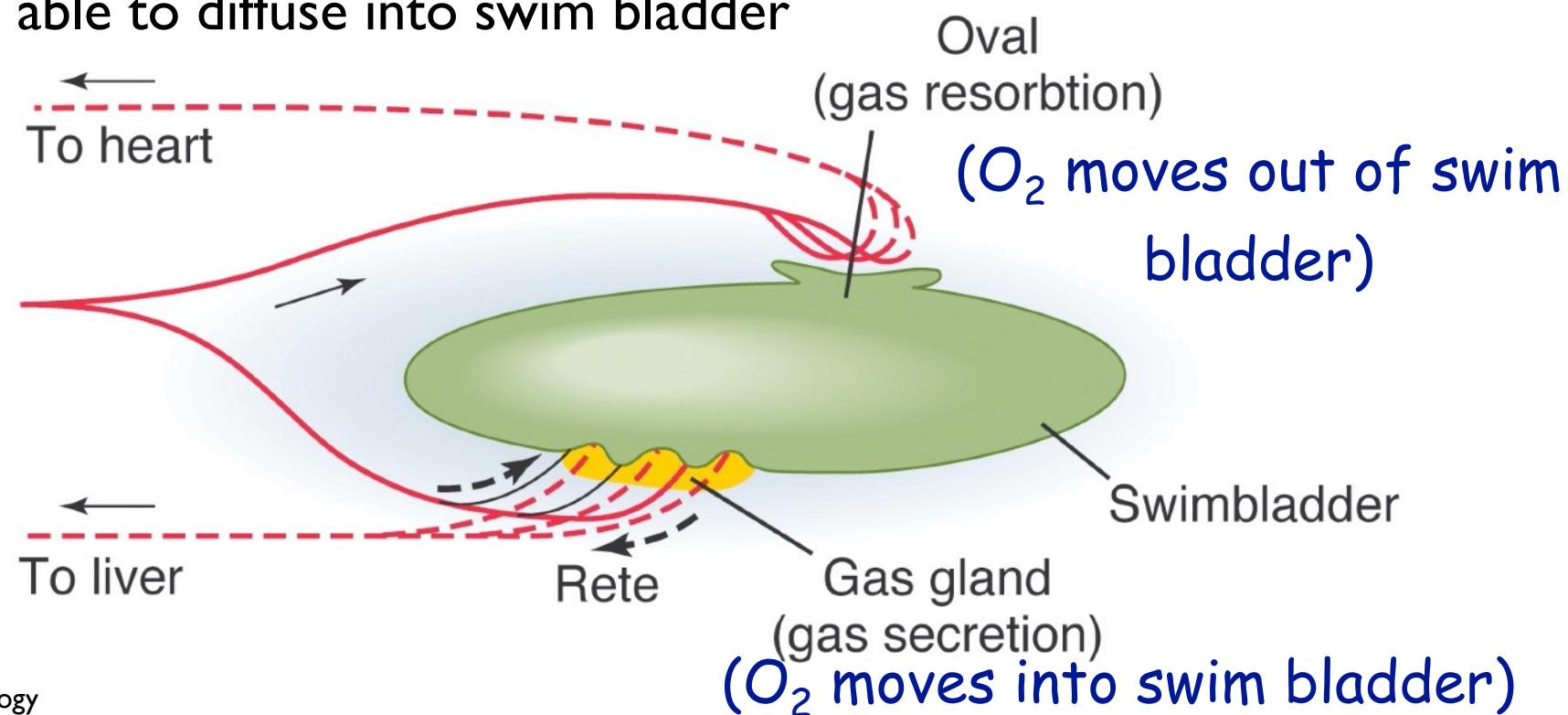
Gas secretion produces lactic acid

Lactic acid enters blood and reduces affinity for O₂ ([Root effect](#))

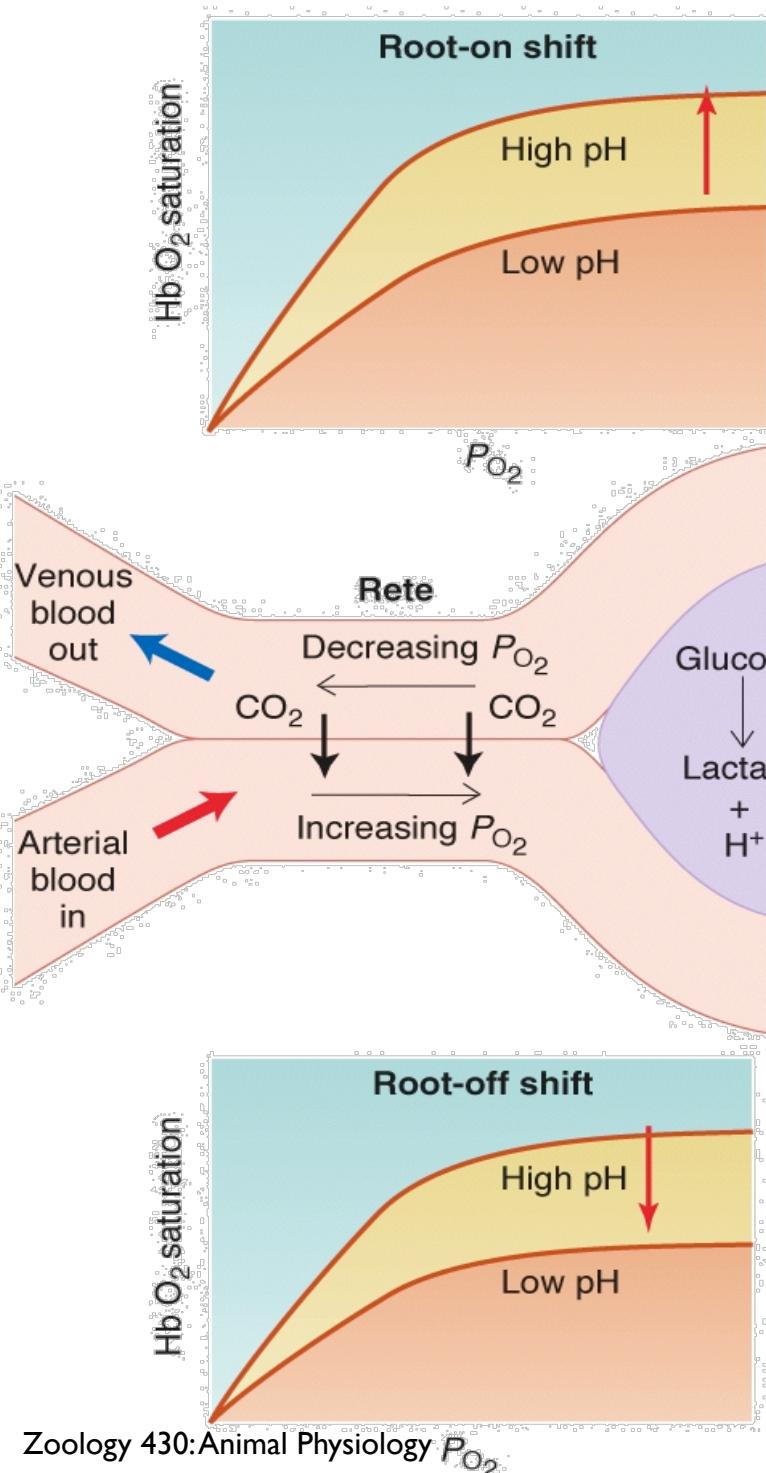
Drives O₂ off the hemoglobin

[Countercurrent flow](#) tends to concentrate lactic acid and O₂ concentration effects at the rete capillaries/gas gland interface

Eventually, O₂ able to diffuse into swim bladder



Root Effect



Oxygen bound to Hb: Decreases P_{O₂}

Root shifts are key

- Gas Gland:
 - Glycolysis
 - H⁺ production
 - Pentose phosphate
 - CO₂ production
 - Activity decreases pH and increases ionic concentration
- Root-off shift
 - lower O₂ solubility
- Rete:
 - counter current gas exchanger for CO₂
 - Drives CO₂ into arterial blood causing Root-off

Oxygen released from Hb: Increases P_{O₂}



The Bohr Effect