

SECOND YEAR: 2604
PLANETARY REMOTE SENSING 4

ELECTROMAGNETIC *RADIATION:*
emission and extinction processes

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<http://www2.le.ac.uk/departments/physics/research/earth-observation-science>

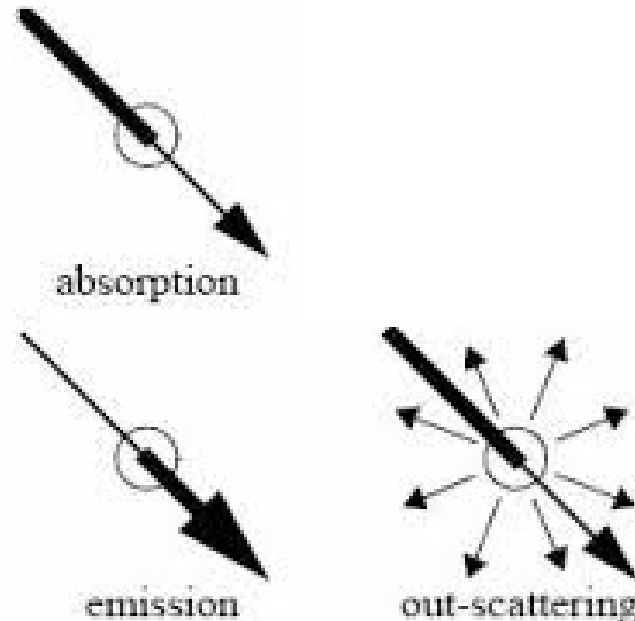


Interaction of Light and Matter

Light (EM radiation) can interact with matter in the following ways:

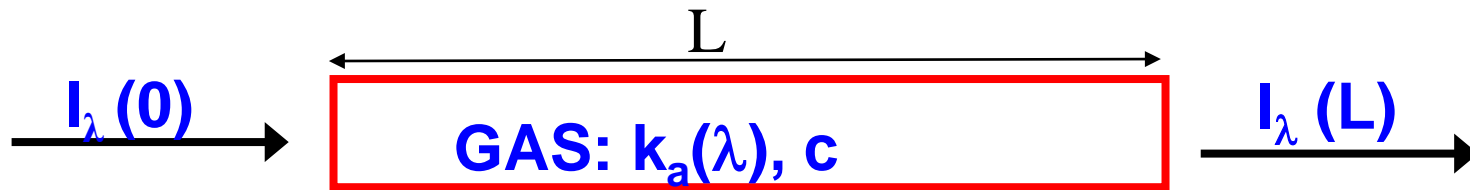
- ❑ Emission: add (generates) photons
- ❑ Absorption: removes photons
- ❑ Scattering: changes direction of photons (and sometimes energy) e.g. when impinging onto a surface, a cloud or an aerosol layer

*In such conditions
(the medium is non
transparent)
radiance is not
constant
along the ray*



BEER-LAMBERT LAW

The **Beer-Lambert Law**: if a signal of intensity I_λ penetrates a distance, dL , in a homogenous medium with **absorption coefficient**, k_a (dimensionally L^{-1}):



$$dI_\lambda = - k_a(\lambda) I_\lambda dL$$

Observed experimentally

Radiance is not constant along the ray!

Then if $k_a = \text{constant along the path } L$

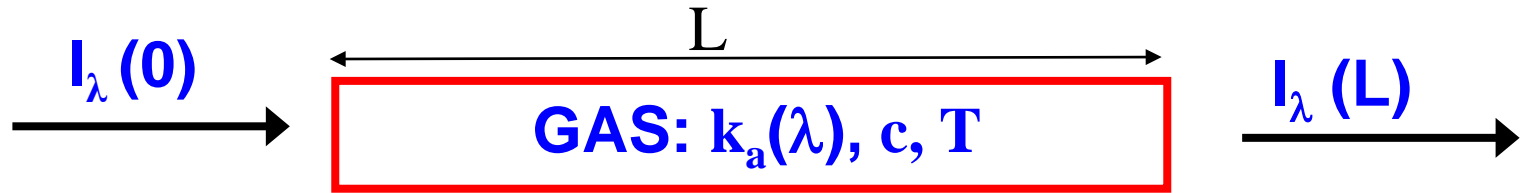
$$I_\lambda(L) = I_\lambda(0) \exp \{- k_a(\lambda) L \}$$

transmissivity:

$$\mathcal{T}(\lambda) = I_\lambda(L) / I_\lambda(0) = \exp \{- k_a(\lambda) L \} = \exp \{-\tau\}$$

with $\tau = \text{optical depth}$

Optical thickness: contributors



$$I_\lambda(L) = I_\lambda(0) \exp \{-k_a(\lambda) L\} = I_\lambda(0) \exp [-\sigma(\lambda) c L]$$

with c = the density of molecules per unit volume.

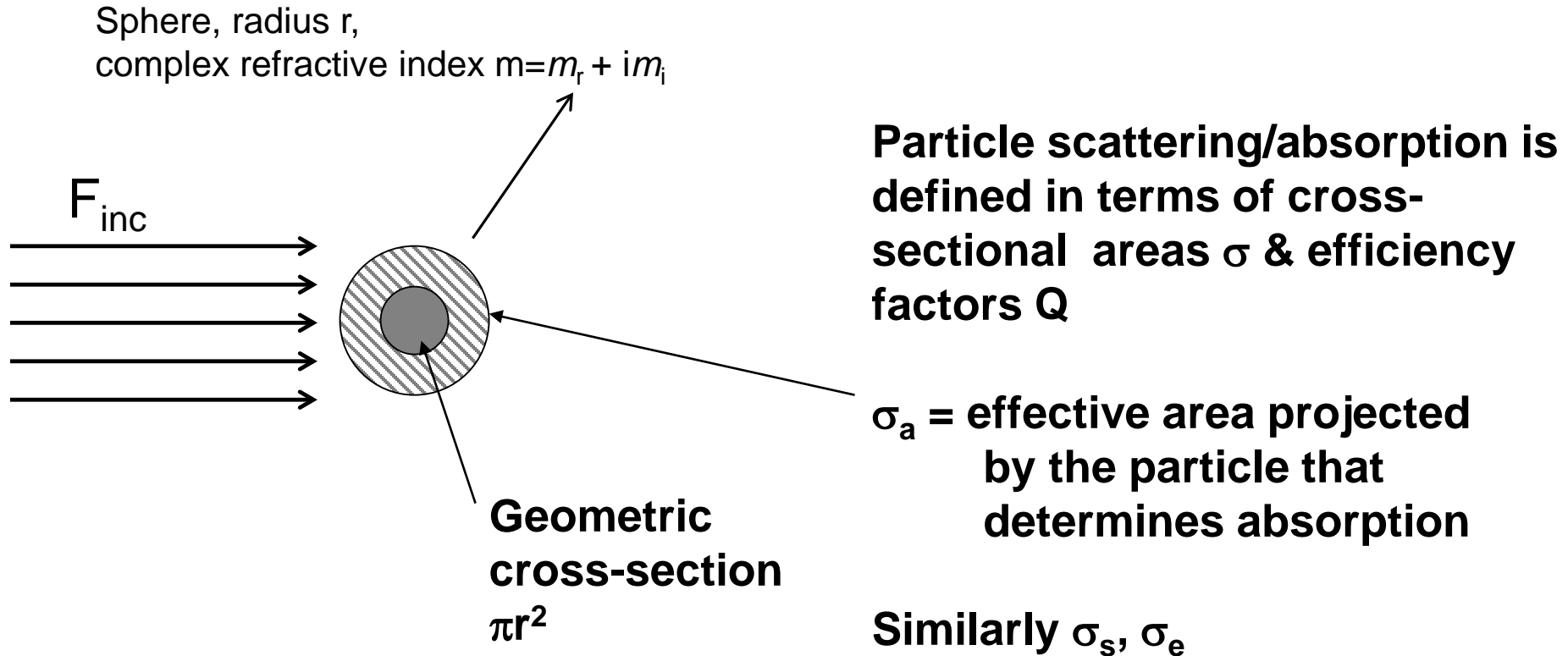
Three factors affects the optical thickness:

- **Spectroscopy**: absorption cross section $\sigma(\lambda)$ [$\text{cm}^2/\text{molecule}$]
- **Composition/density**: $c = \chi c_{\text{air}}$ [$\text{molecules}/\text{cm}^3$]
- **Photon path-length**: geometrical distance = L [km]

mixing ratio of the gas

General form with k_a changing in space: $I_\lambda(L) = I_\lambda(0) e^{-\int_0^L \sigma(\lambda, p, T) c(p, T) dL}$

Interaction of Particle with E/M Radiation: cross sections



The particles is absorbing and/or scattering some of the incident radiation → absorbed/scattered power

$$\sigma_{e,s,a} = \frac{P_{e,s,a}}{F_{inc}}$$

The **efficiency factor** then follows

$$Q_{e,s,a} = \frac{\sigma_{e,s,a}}{\pi r^2} = f(x, m)$$

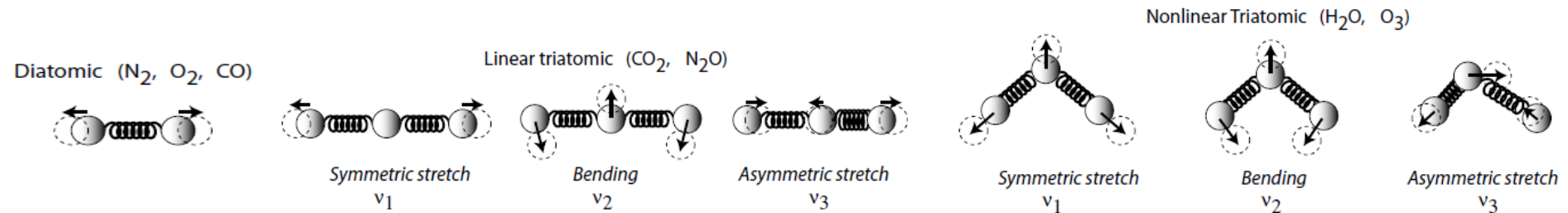
N.B.: Q can be larger than 1!!!

Spectroscopy

The study of the interaction of a medium with radiation is known as **spectroscopy**. The spectral properties of gases, liquids and solids are determined fundamentally by quantum mechanics.

Hence $\sigma_a(\nu)$ is fundamentally due to quantised transitions between energy levels in molecules.

In the infra-red, the transitions occur as distinct **bands** unique to each gas (specifically due to the vibration and rotation of each molecule).



Light gas molecules – distinct line structure grouped in bands e.g. CO , CO_2 , H_2O , CH_4

Heavy gas molecules – densely structured bands that appear apparently smooth (e.g. CFCs)

Liquids – simple broad features (droplets, aerosols, surface)

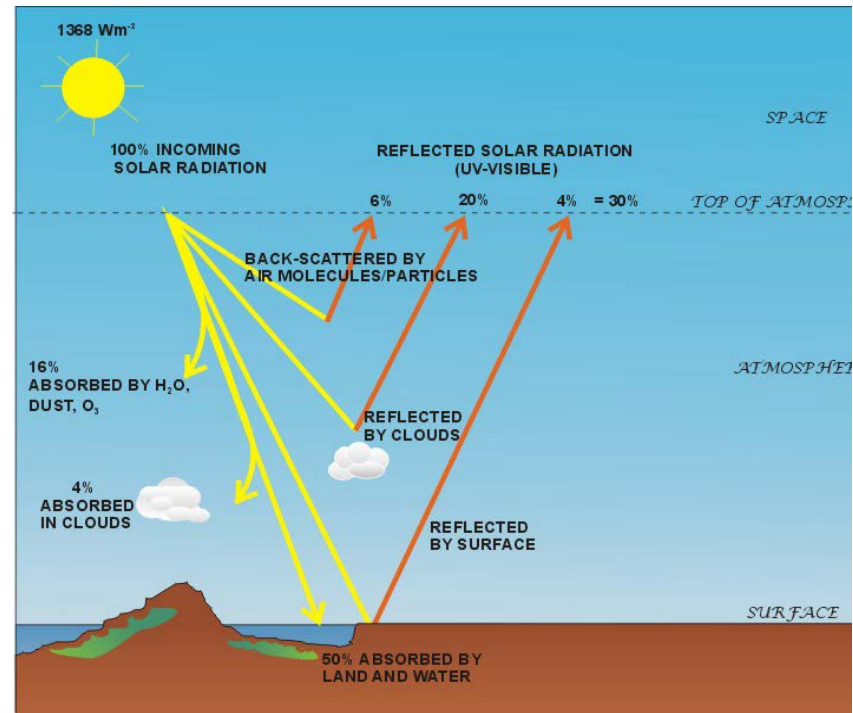
Solids – simple, broad features (ice, aerosols, surface)

Scattering
theory

Gas absorption in the Earth's atmosphere

Main Visible and near-IR absorption bands of atmospheric gases

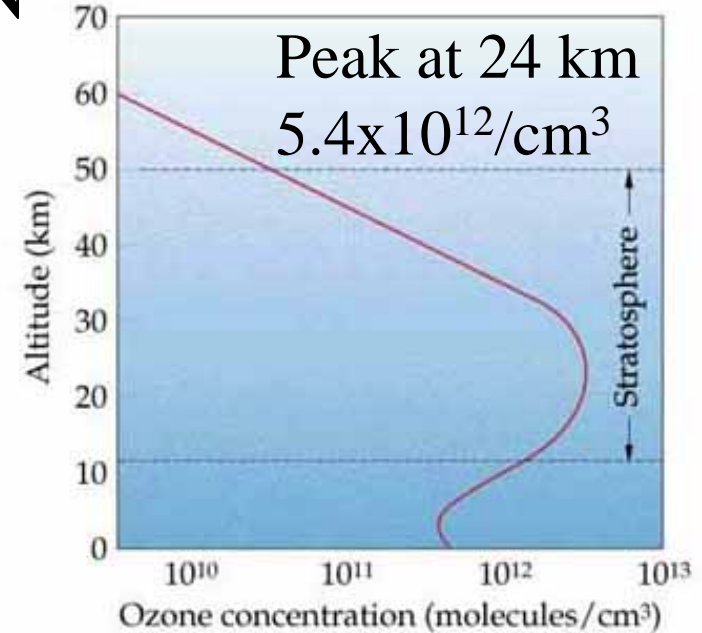
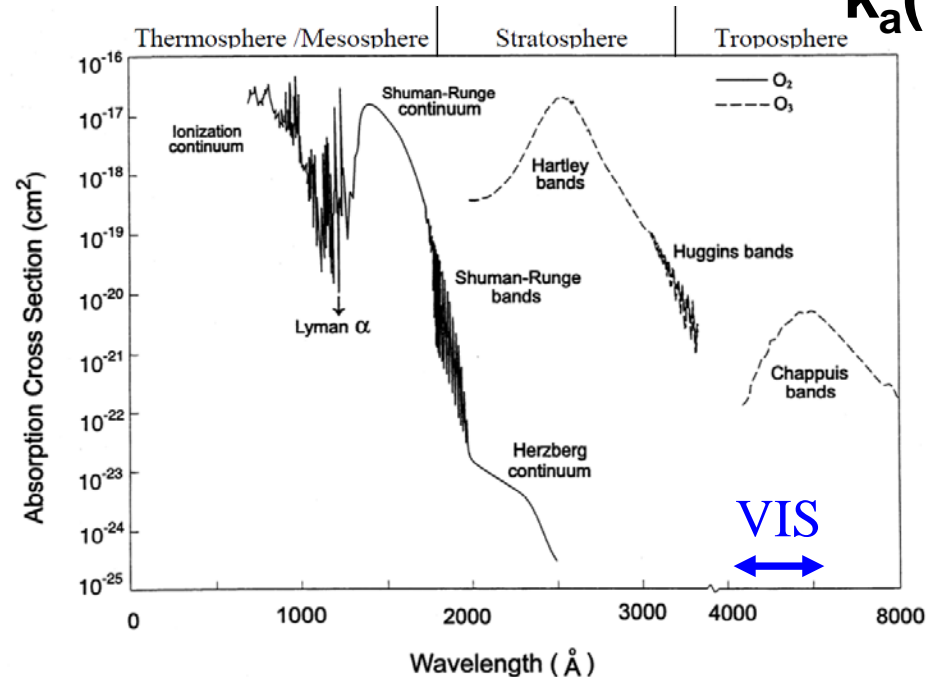
Gas	Center ν (cm ⁻¹) (λ (μ m))	Band interval (cm ⁻¹)
H₂O	3703 (2.7) 5348 (1.87) 7246 (1.38) 9090 (1.1) 10638 (0.94) 12195 (0.82) 13888 (0.72) visible	2500-4500 4800-6200 6400-7600 8200-9400 10100-11300 11700-12700 13400-14600 15000-22600
CO₂	2526 (4.3) 3703 (2.7) 5000 (2.0) 6250 (1.6) 7143 (1.4)	2000-2400 3400-3850 4700-5200 6100-6450 6850-7000
O₃	2110 (4.74) 3030 (3.3) visible	2000-2300 3000-3100 10600-22600
O₂	6329 (1.58) 7874 (1.27) 9433 (1.06) 13158 (0.76) 14493 (0.69) 15873 (0.63)	6300-6350 7700-8050 9350-9400 12850-13200 14300-14600 14750-15900
N₂O	2222 (4.5) 2463 (4.06) 3484 (2.87)	2100-2300 2100-2800 3300-3500
CH₄	3030 (3.3) 4420 (2.20) 6005 (1.66)	2500-3200 4000-4600 5850-6100
CO	2141 (4.67) 4273 (2.34)	2000-2300 4150-4350
NO₂	visible	14400-50000



In the ultra-violet and visible, incident light from the Sun is absorbed and/or scattered. A principal absorber is ozone in the stratosphere. UV-visible radiation → heating wherever it is absorbed. As a result less solar radiation reaches the surface.

Spectral absorption cross-sections of O₂ and O₃

$$k_a(\nu) = \sigma_a(\nu)N$$



- Bands of O₂ and O₃ at wavelengths < 1 μm are electronic transitions.
- These absorption bands cover a continuum because practically all absorption results in dissociation of the molecule (so the upper state is not quantized);
- Despite the small amount of O₃, no solar radiation penetrates to the lower atmosphere at wavelengths < 310 nm (because of large absorption cross-sections of O₃);

Transmissivity due to gases

Example

Calculate the transmission of O_3 lines over a range of 40 km for a line strengths of a) $10^{-17} \text{ cm}^2/\text{mol}$; b) $10^{-19} \text{ cm}^2/\text{mol}$; c) $10^{-21} \text{ cm}^2/\text{mol}$. Assume O_3 has a constant concentration given by $3 \times 10^{12} \text{ molecules cm}^{-3}$.

$$\mathcal{T}_g(\nu) = \exp [- \sigma_a(\nu) N L]$$

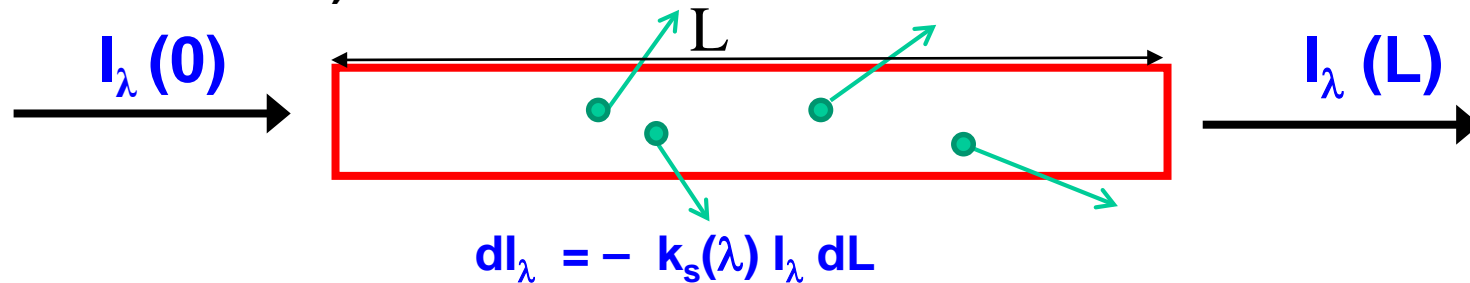
a)	$= \exp [- 10^{-17} \text{ cm}^2/\text{mol} * 3 * 10^{12} \text{ mol cm}^{-3} * 40 * 10^5 \text{ cm}]$	
	$= \exp [- 120] \approx 0.0$	$\lambda = 0.25 \text{ } \mu\text{m}$
b)	$= \exp [- 1.2] \approx 0.3$	$\lambda = 0.3 \text{ } \mu\text{m}$
c)	$= \exp [- 0.012] \approx 0.988$	$\lambda = 0.35 \text{ } \mu\text{m}$

- a) is opaque
- b) has mid-range transmission
- c) is nearly transparent.

Ozone is really screening out
near UV radiation →
relevance of ozone hole for
skin cancer

BEER-LAMBERT LAW: EXTINCTION LAW

Homogenous non-absorbing medium with scattering coefficient, k_s
(per unit distance)



and $I_\lambda(L) = I_\lambda(0) \exp \{-k_s(\lambda) L\}$ (compare absorbing case)

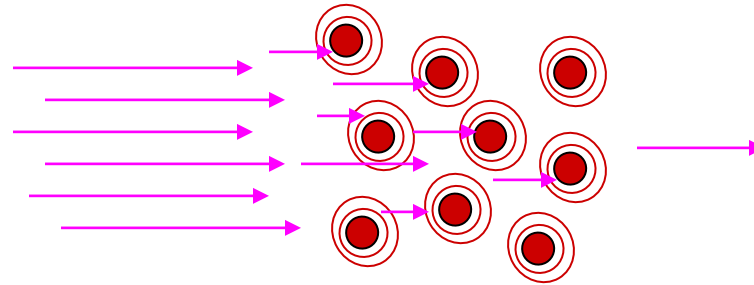
Scattering and absorbing media: $I_\lambda(L) = I_\lambda(0) \exp \{-k_e(\lambda) L\}$

with **extinction coefficient**: $k_e = k_s + k_a$

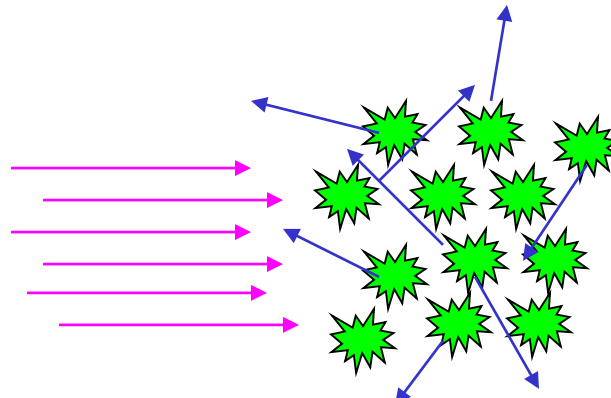
Extinction: total loss of light due to absorption and scattering of light out of path

Extinction: Scattering plus Absorption

Absorption removes radiant energy from an E/M field, converting it to other forms of energy.



Scattering does not remove energy from an E/M field, but can redirect it, thereby making it a “source” of radiation for another direction.



(Effect on transmittance is the same)

Single scatter albedo (SSA): fraction of extinction (absorption + scattering) due to scattering processes:

$$\omega_o = \sigma_s / \sigma_e$$

$$\sigma_e = \sigma_a + \sigma_s$$

$\omega_o = 1$ all scattering (conservative)

$\omega_o = 0$ all absorption (non - conservative)

How can we compute scattering and absorption cross sections of single particles?
How can we relate them to scattering and absorption coefficients?

The scattering optical thickness of our atmosphere

Compare the atmospheric transmissivity for Rayleigh scattering in UV with larger wavelength (600nm, 900 nm)

λ , nm	σ , cm ²
300	6.00×10^{-26}
400	1.90×10^{-26}
600	3.80×10^{-27}
1000	4.90×10^{-28}
10,000	4.85×10^{-32}

Transmission

$$\mathcal{T}(\lambda) = I_{\lambda}(L) / I_{\lambda}(0)$$

$$= \exp(-\tau) = \exp \{-k_s(\lambda) L\}$$

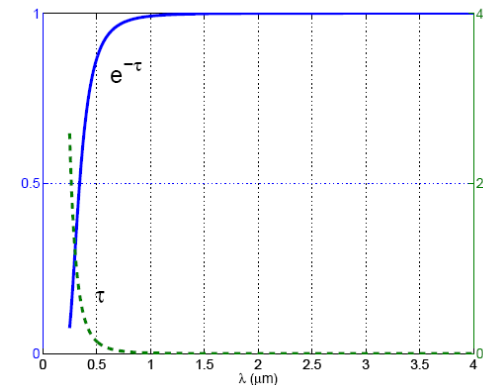
$$= \exp \{-\sigma c L\} = \exp \{-\sigma \text{VCD}\}$$

with Vertical Column Density of air
(VCD) = 2.3×10^{25} molec/cm²

Transmissivity: $\mathcal{T}(\lambda=300\text{nm}) = 0.25$
 $\mathcal{T}(\lambda=600\text{nm}) = 0.92$
 $\mathcal{T}(\lambda=1000\text{nm}) = 0.99$

$\tau = 1.4$
 $\tau = 0.08$
 $\tau = 0.01$

Transmittance and τ for Rayleigh scattering

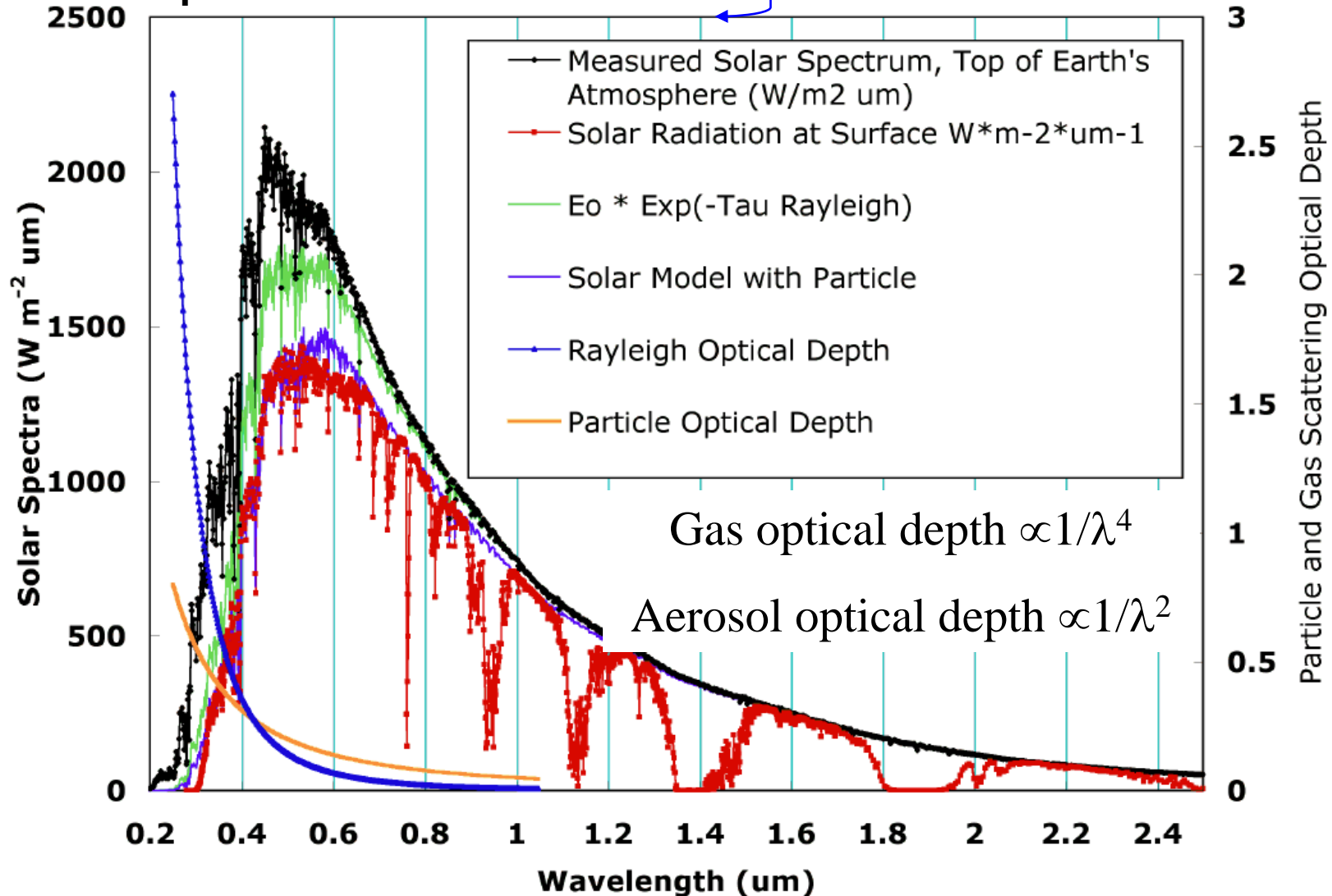


Only 25% of direct light reaches surfaces in UV at 300 nm but 99% at 1000nm.

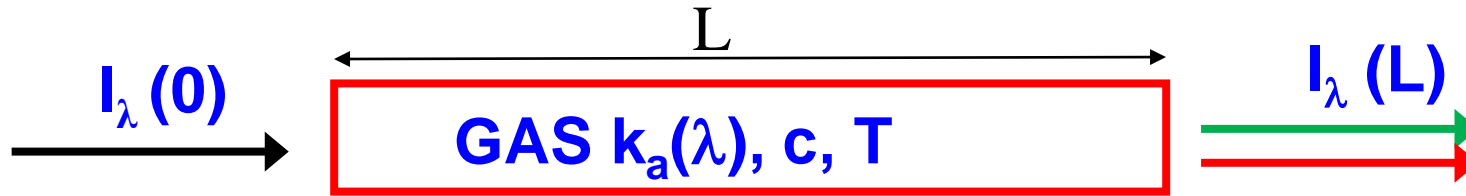
SOLAR SPECTRUM

1. Rayleigh (gas) scattering,
2. O₂, N₂, water vapour absorption
3. Aerosol particles extinction

Contributions to extinction of
solar radiation from TOA to surface



Accounting for emission



There are 2 signals emerging:

1. Radiance transmitted by the gas in the cell: $\mathcal{T}(\lambda) I_0(\lambda)$
2. Radiance emitted by the gas in the cell: $\epsilon(\lambda) \times B(\lambda, T)$

Emissivity is related to absorptance $a(\lambda)$ of the layer (fraction of radiance being absorbed)

We neglect scattering (valid for gases in the IR)

$$\mathcal{T}(\lambda) + a(\lambda) = 1 \Rightarrow a(\lambda) = 1 - \mathcal{T}(\lambda)$$

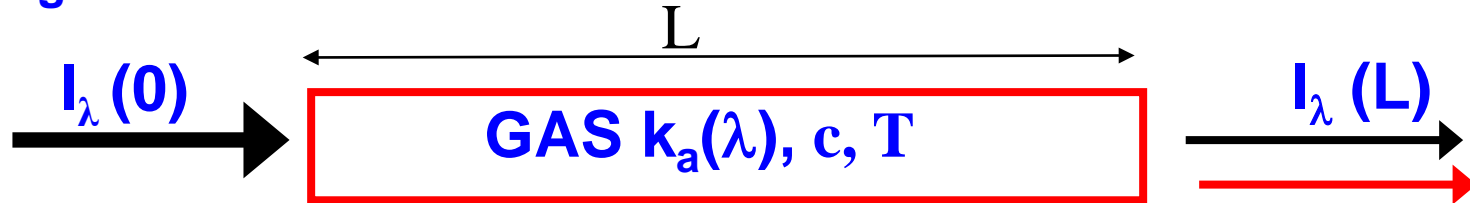
Radiation is either transmitted or absorbed

*In thermal equilibrium
(Kirchhoff's law)*

$$\epsilon(\lambda) = a(\lambda) = 1 - \mathcal{T}(\lambda)$$

IR TRANSMISSION FOR ISOTHERMAL LAYER

Total Signal



Hence, Total Signal, $I(\lambda)$ is given by

$$\Rightarrow I_\lambda(L) = \mathcal{T}(\lambda) I_\lambda(0) + (1 - \mathcal{T}(\lambda)) \times B(\lambda, T)$$

Limiting cases:

1) $\mathcal{T}(\lambda) \approx 1 \rightarrow I_\lambda(L) \approx \mathcal{T}(\lambda) I_\lambda(0)$

[Known as a WINDOW]

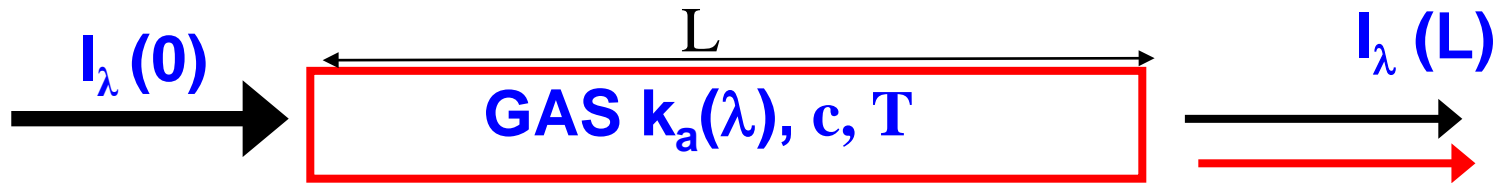
2) $\mathcal{T}(\lambda) = 0 \rightarrow I_\lambda(L) = B(\lambda, T)$

[Known as 100% absorption or saturation]

N.B. : If: a) term 1 \gg term 2 or b) if $T_{\text{gas}} = 0 \text{ K (!)}$

then Eqn. 1) would be true and conventional use of $\mathcal{T}(\lambda)$ alone is fine, e.g. uv-visible on Earth or hot source relative to cold gas.

Hot nebula



$$I_\lambda(L) = \mathcal{T}(\lambda) I_\lambda(0) + (1 - \mathcal{T}(\lambda)) \times B(\lambda, T)$$

Imagine first the case in which $I_\lambda(0)=0$, i.e. solely emission from the volume of gas (with constant source function). $I_\lambda = B_\lambda (1 - e^{-\tau_\lambda})$

We have two limiting cases:

- Optically thin case ($\tau_\lambda \ll 1$)

$$e^{-\tau_\lambda} \approx 1 - \tau_\lambda \Rightarrow I_\lambda = \tau_\lambda B_\lambda$$

(a) Opacity τ versus λ

→

(b) Intensity versus λ
- Optically thick case ($\tau_\lambda \gg 1$)

$$e^{-\tau_\lambda} \approx 0 \Rightarrow I_\lambda = B_\lambda$$

(a)

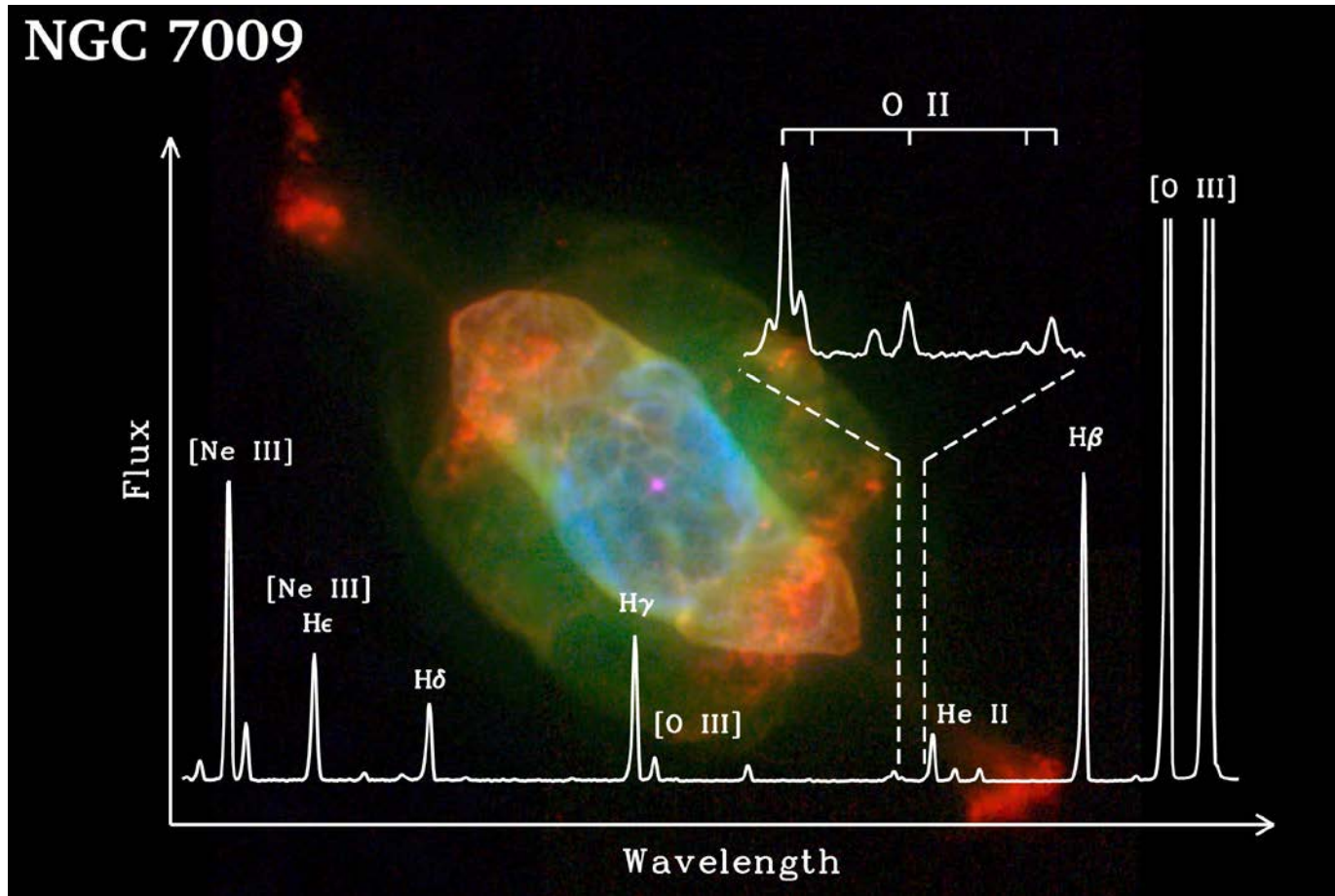
→

(b)

It behaves like a black body

Hot low density nebular gas

Example of emission lines for optically thin nebulae



The optical thickness is strongly dependent on wavelength → peaks corresponding to absorption bands of different elements in the nebula

Absorption versus emission

$$I_{\lambda}(L) = I_{\lambda}(0)e^{-\tau_{\lambda}} + B_{\lambda}(T)(1 - e^{-\tau_{\lambda}})$$

Imagine now $I_{\lambda}(0) \neq 0$, again with two extreme cases:

- Optically thin case ($\tau_{\lambda} \ll 1$) $I_{\lambda}(L) = I_{\lambda}(0)(1 - \tau_{\lambda}) + \tau_{\lambda}B_{\lambda}(T) = I_{\lambda}(0) + \tau_{\lambda}[B_{\lambda}(T) - I_{\lambda}(0)]$

(a) If $I_{\lambda 0} > B_{\lambda}$, so there is something subtracted from the original intensity which is proportional to the optical depth – we see absorption lines on the continuum intensity I_{λ} .

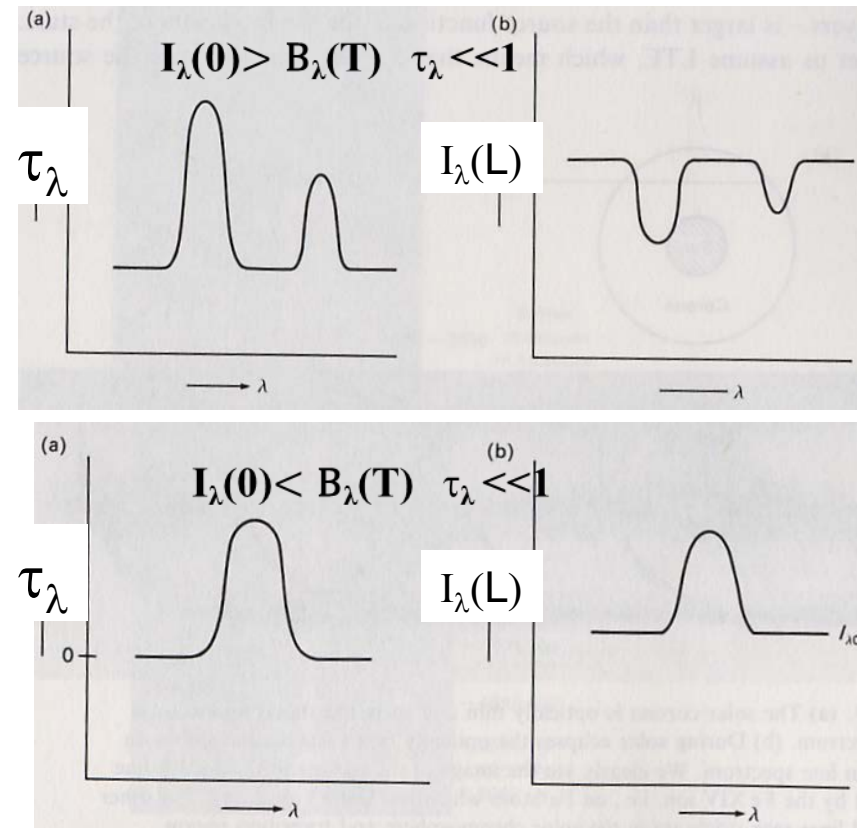
EXAMPLE: stellar photospheres

(b) If $I_{\lambda 0} < B_{\lambda}$, we will see emission lines on top of the background intensity.

Example: Solar UV spectrum

- Optically thick case ($\tau_{\lambda} \gg 1$):

Planck function as before. $I_{\lambda}(L) = B_{\lambda}(T)$



Opacity τ versus $\lambda \rightarrow$ Intensity versus λ

Absorption lines? Outward decreasing temperature

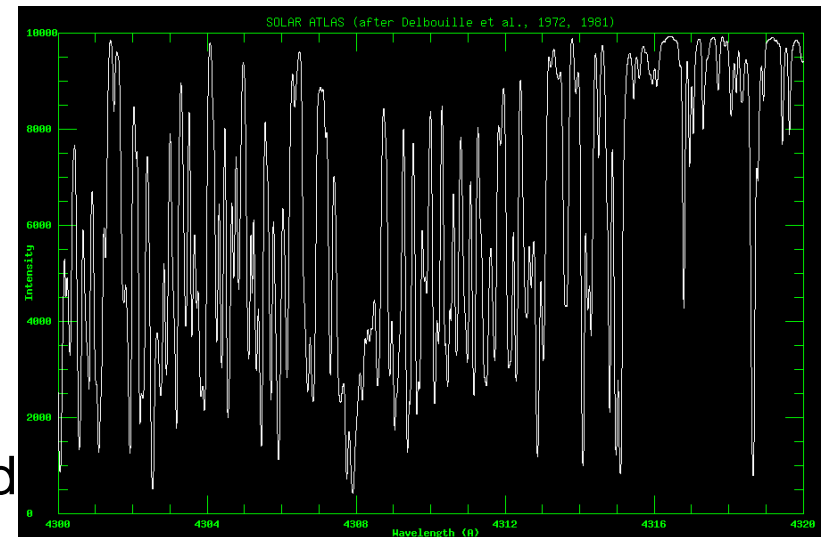
In a star absorption lines are produced if

$I_{\lambda 0} > B_{\lambda}$ i.e. the intensity from deep layers is larger than the source function from top layers.

In LTE, the source function is $B_{\lambda}(T)$, so the Planck function for the deeper layers is larger than the shallower layers.

Consequently the deeper layers have a higher temperature than the top layers (since the Planck function increases at all wavelengths with T).

(Instances occur where LTE is not valid, and the source function declines outward in parallel with an increasing temperature).



Solar Spectrum (4300-4320
angstrom=0.43-0.432 micron) →
absorption lines

Absorption versus emission spectra

Emission line spectra

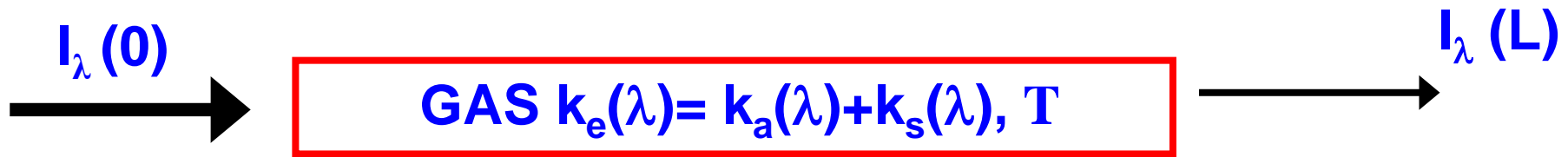
- Optically thin volume of gas with no background illumination (emission nebula)
- Optically thick gas in which the source function increases outwards (UV solar spectrum)

Absorption line spectra

- Optically thick gas in which source function declines outward, generally T decreases outwards (Stellar photospheres)
- Optically thin cold gas penetrated by background radiation (Interstellar matter between us and the star)

What do you need to know?

- ❑ Emission and Absorption
- ❑ Isothermal layer model



$$I_{\lambda}(L) = \mathcal{T}(\lambda) I_{\lambda}(0) + (1 - \mathcal{T}(\lambda)) B_{\lambda}(T)$$

transmissivity: $\mathcal{T}(\lambda) = \exp \{-k_e(\lambda) L\} = \exp \{-\tau\}$

- ❑ Lambert Beer Law (optical depth, transmissivity)