Pembahasan Soul No	la
- Proses A	
Kitatahu	
bahwa berdusarkan Fi	gurnya (dari persamaan Gas Ideal Ideal
· P=P2 = Po (Pers 1) =	T3 = T0 = 1
· T. : To : Ta (Pers 2) .	Vi=Vo Pili=nRli
· T2 = If	1. 1/4 De la partie de la carte de la cart
• /	/3 = 6 Vo ? adalah tetap karena dalam ideal) raangan tertutup
maka dapat kita cari	raangan tertutup
perubahan titik / he	$T_{1}$
1	TF. 2 To DTG DP 2 TO DP T. P
P. Vi = nR Ti , Ti	
P2 1/2 = NRT2	
Pi= nRTi= nRTo	$P_{3} V_{2} = 2 V_{0}$ $P_{3} V_{3} = nR T_{3}$ , $T_{3} = T_{0}$ $V_{3} = 6 V_{0}$
V1 V0 R2 = NRT2 = NRT5	
1/2 1/2	Pf (61/6) = nR To
dimana V2 21/0 Frantia P. = P2	$P_{f}(6/6) = nRT_{0}$ $P_{f} = \frac{nRT_{0}}{6/6} = \frac{1}{6} \left(\frac{nRT_{0}}{V_{0}}\right), P_{0} = P_{2} = \frac{nRT_{0}}{V_{0}}$
that a fr	5 Vo 6 ( Vo ) Vo
ARTO DRTE	Pf = LPo
	6
XQ 2 /2	Jadi Nilai Tg=2 To, Pg=1Po
Te = 2 To	}
maka Tf= 2 To	
mun (j= 2 10	
- Proses B	
Berdasarkan Figurny	a Dapar Kita Cari
dapat di ketahui	perubahan titik 1 Ke titik 2
0 V, = 1/2 = 1/0	birdusarkan Persamaan Gas Idea
· T2 = T3 = Tf	$P_1V_1 = nRT_1$ $P_3V_3 = nRT_3$
$P_1 = P_0 = P_3$	P2V2: NRT2 ( Makn
	$V_1 = \Omega R T_1 = \Omega R T_0$ $V_1 = V_2$
$P_2 = P_f$	P, Po MRTo MRTF
· T, = T6	V2 = NRT2 NRT5 Po PS
· 13 = 310	
The state of the s	
The second of th	3 Po Pf
The state of the s	

berdasarkan Perubahan titik 2 ke 3 leita tahu P2 1/2: nRT2 P3 = P1 = P0 Pf Vo = NRTE T3 = T2 = T5 Pala = nRT3 V3 = (3 V6) T2: P5/0 Berdasarkan Pers 3 dimana T3 = P3 V3 = Po (31/6) To : If dimana Pf: 3P6 15 = 15 = To P+ 16 - Po(3/6) Pe 3Pa Po TF= 3 To Pr. 3Po Jadi pada proses B Tf=3 To dan Pf=3 Po

```
16. Proses A
                     (Isobaric)
 Q12 = E7h - W
    z nCuDt - (-nR DT)
    2 n (CU+R) AT
    =n (Cutr) (2To-To
    = nTo((U+R)
 Qu = 0
             cadiaboutic
 Q = Q12 + Q23
   = n To ((+ + P) + 0
   = nTo (co+R)
 Proses B
 Que ETH - W (isokhorik w=0)
 Qiz = neudt
    = n Cu (Tf-To)
    = n Cu (3To - To)
    = 24 n Cu (276
    = ncu 270
Quis = Ein - w ( iso termal Dein = 0)
 023 = - (- nRTy ln (300) ]
```

Q= n RTf ln(3) = n R (3To) ln(3) = 3n R To ln(3)

atex = 2ncoTo + 3nRTolne3)

MAICAITA MBICBITB

Balok A bermassa MA dgn suhu TA menyetuh balok B bermasa MB dgn suhu TB. Kalor jenis balok A = CA dan balok B = CB.

a. Tentukan temperatur akhir Tf dari 2 balok logam!

Kita misalkan bahwa TA > TB.

Maka kita bisa menggunakan Asas black, dimana :

Q lepas = Q terima

MACA (TA-TF) = MBCB (Tf-TB)

MACATA - MACATF = MBCBTF - MBCBTB

MA CATF + MB (BTF : MA CATA + MB (BTB

Tf = MACATA + MBCBTB MACA + MBCB

## b. Tentukan perubahan suho:

Kita tinggal memasukkan persamaan Tf yg sudah didapat sebelumnya.

MACA + MB CB

ATA = MB(B(TB-TA)

$$\Delta T_A = \frac{T_B - T_A}{\left(\frac{M_A C_A}{M_B C_B} + L\right)}$$

MACA + MBCB

$$\Delta TB = \frac{TA - TB}{\left(1 + \frac{MB(B)}{MA(A)}\right)}$$

## C. Misal MA = MB dan | ATAI > | ATB|

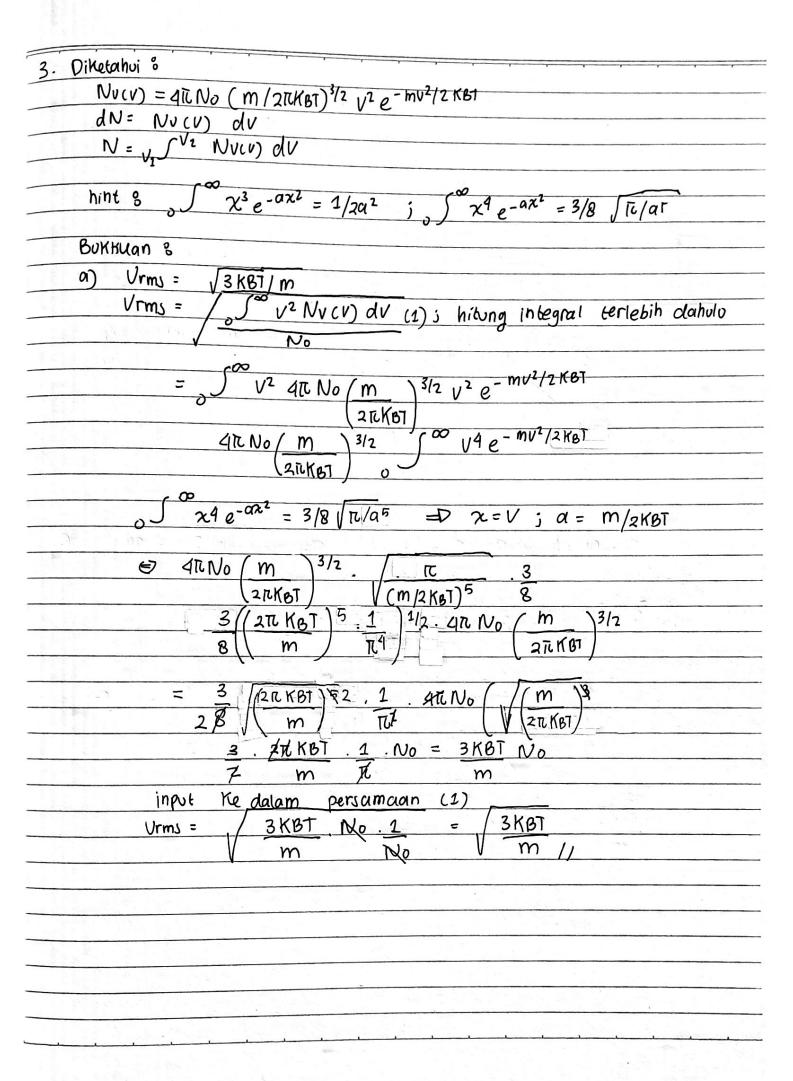
Balok logam mana ya punya Clebih Hinggi?

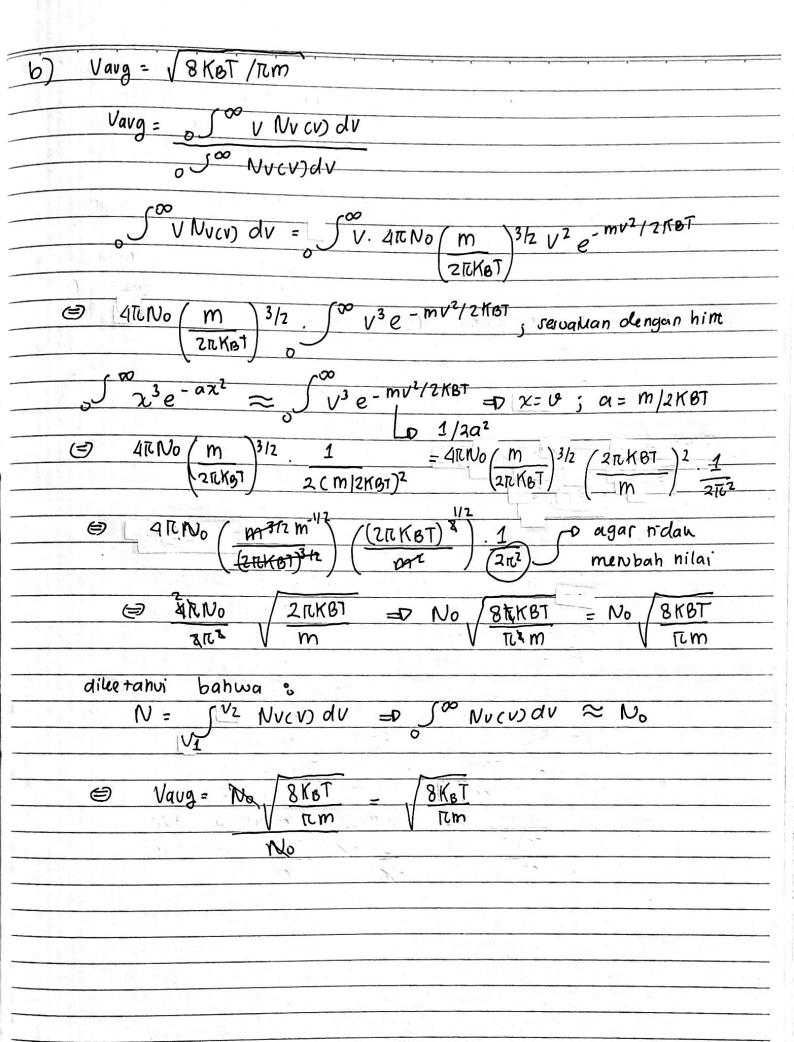
Diketahui bahwa  $\Delta TA > \Delta TB$   $\frac{\Delta TA}{\Delta TB} > 1$ 

Lalu kita gunakan asas Black lagi

$$\frac{\Delta TA}{\Delta TB} = \frac{CB}{CA} \longrightarrow \frac{CB}{CA} \nearrow 1$$

$$CB \nearrow CA$$





C) 
$$Vmp = \sqrt{2K_BT}$$
 =D  $dNv(v) = 0$  =D  $dNv(v) = 0$ 

$$dV \qquad dV$$

$$dV \qquad dV$$

$$\sqrt{2k_BT} \qquad \sqrt{2}e^{-mv^2/2K_BT} \Rightarrow dV^2e^{-mv^1/2K_BT} = 0$$

$$\sqrt{2ve^{-mv^2/2K_BT}} - (\frac{mv^3}{K_BT})e^{-mv^2/2K_BT} = 0$$

$$2ve^{-2mv^2/2K_BT} = \frac{mv^{3/2}e^{-mv^2/2K_BT}}{K_BT}$$

$$\sqrt{2} = 2K_BT \Rightarrow Vmp = V = \sqrt{2K_BT}$$

$$m \qquad Vmp = V = \sqrt{2K_BT}$$