### Problem Set 3 Solutions

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ME 257/357: Propulsion System and Gas-Turbine Analysis



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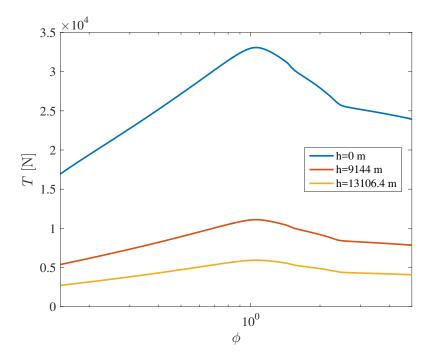


Figure 1: Thrust computed for the equilibrium combustion model.

# 1 Problem 1: Turbofan with Combustion Model (30 pts)

- (a) (10 pts) Example code uploaded to Canvas.
- (b) (20 pts)
  - (i) (10 pts) A plot of the real turbofan thrust is shown in Fig. ??. Thrust is optimized at approximately  $\phi = 1.09$ , 1.05, and, 1.09 for h = 0, 30, and 43 kft, respectively. Thrust is peaking near the stoichiometric condition since an optimal amount of fuel is being consumed for this condition given the mass flow of the air.
  - (ii) (10 pts) A plot of the temperature ratio is shown in Fig. ??. The temperature ratio peaks at approximately  $\phi = 1.09$ , 1.09, and, 1.05 for h = 0, 30, and 43 kft, respectively.  $T_4/T_{4,\text{max}} = 1$  at approximately  $\phi = 0.34$ , 0.4, and, 0.4 for h = 0, 30, and 43 kft, respectively. Comparing to Fig. ??, these equivalence ratios are suboptimal.

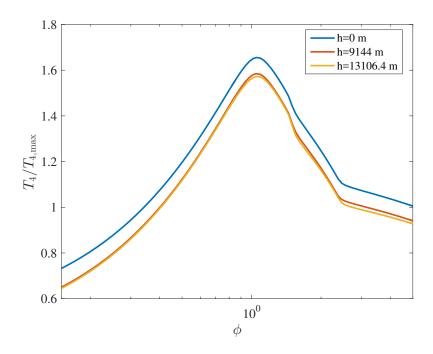


Figure 2: Temperature ratio computed for the equilibrium combustion model.

# 2 Problem 2: Cruise Conditions for the HondaJet (30 pts)

- (a) (5 pts) We will use the core airflow,  $\dot{m}_c$ , and the bypass ratio,  $\beta$  in addition to the pressure ratios and the aircraft parameters. However, we will not use the  $T_{04}=1600$  specification. We will compute  $T_{04}$  as the adiabatic flame temperature, and will find  $\dot{m}_f$  from the more accurate combustor model.
- (b) (10 pts) Note that the HondaJet has two engines and this must be accounted for. The entire drag is given by

$$D = \frac{\rho_0 U_0^2 S C_{D,0}}{2} + \frac{2W^2}{\pi e \rho_0 U_0^2 b^2} \ . \tag{1}$$

Hence the thrust required for one engine is  $T^* = D/2$ , while the mass flow through each engine is still  $\dot{m}_c = \dot{m}_{c,\rm SL} \rho_0/\rho_{0,\rm SL} = 8\rho_0/\rho_{0,\rm SL}$  kg/s. Finally, accounting for both engines burning fuel,  $\dot{m}_{f,\rm total} = 2\dot{m}_f$ 

(c) (5 pts) The ODE is simply  $\frac{\mathrm{d}W}{\mathrm{d}t} = -\dot{m}_{f,\mathrm{total}}g \ . \tag{2}$ 

 $\dot{m}_{f,\text{total}}$  depends on producing a certain amount of thrust; so, engine parameters as well as drag matters. Also, drag is a function of altitude, airspeed, and aircraft parameters.

(d) (5 pts) Using a airspeed of 200 m/s and an altitude of 43 kft, the range is found to be 6360 km.

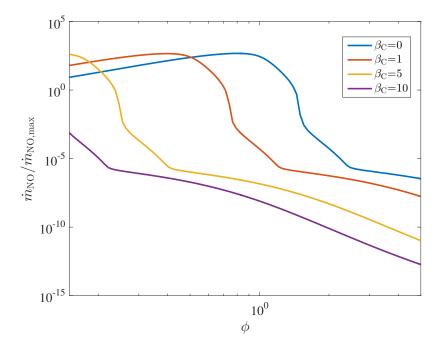


Figure 3: Mass flow rate of nitric oxide against the combustor reroute ratio.

(e) (5 pts) The Breguet range equation predicts  $s_{43\text{kft.}} \in [2810, 3590]$ , which is significantly less than that given by the more realistic combustor. Differences include the incorporation of a non-constant L/D ratio, and a significantly lower TSFC.

# 3 Problem 3: Turbofan with Combustion Model (30 pts)

- (a) (10 pts) A plot showing the normalized mass flow rate of NO is shown in Fig. ??. An equivalence ratio of  $\phi = 1.42$  is shown to meet the standard. However at this rich equivalence ratio, carbon monoxide and particulate formation may become an issue.
- (b) (15 pts)
  - (a) (5 pts)  $L_{\rm dil}$  is obtained by

$$\frac{d\dot{m}_{NO}}{dt} = k\dot{m}_{N_2}\dot{m}_O$$

$$\implies \dot{m}_{NO} = k\dot{m}_{N_2}\dot{m}_Ot$$

$$= k\dot{m}_{N_2}\dot{m}_O\frac{x}{U_4}$$

$$\implies L_{dil} = \frac{U_4\dot{m}_{NO,max}}{k\dot{m}_{N_2}\dot{m}_O}.$$
(3)

- Substituting the given values, one obtains  $L_{\rm dil} = 1.3$  km. Since the maximum dilution distance is much longer than any practical length for a combustor, it is quite feasible to quench the combustion before any significant NO is formed.
- (b) (10 pts) Figure ?? shows the normalized mass production of NO for three reroute ratios. Using the RQL combustor design. It is shown for a reroute ratio of 5 or 10, that NO formation is not a limiting factor for the design since the NO formation is much less than the allowable for  $T_4/T_{4,\rm max}=1$ .
- (c) (5 pts) At sea level,  $T_4 = 1600$  K and  $p_4 = 25.7$  bar for the real turbofan. Substituting these values into the given correlation, one obtains  $x_{NO} = 0.0019$ .

For n-dodecane  $f_{\rm st} = 0.0669$ , and since from Fig. ??  $\phi = 0.34$  for h = 0,  $f = \phi f_{\rm st} = 0.023$ . Hence,  $\dot{m}_4 = (1+f)\dot{m}_3 = 8.2$  kg/s.

Using the combustor model at the prescribed flight speed and altitude, and  $\phi=0.34$ , one finds by substituting the computed mole fraction of nitric oxide into the cantera gas object that  $Y_{\rm NO}=0.002$ . Since  $\dot{m}_{\rm NO}=Y_{\rm NO}\dot{m}_4$ ,  $\boxed{\dot{m}_{\rm NO}=16.2}$  g/s. Since  $\dot{m}_{\rm NO}>\dot{m}_{\rm NO,max}$ , this correlation shows that the Honda Jet would not meet the standards. The small difference in the models can likely attributed to the error introduced from the curve-fitting procedure and differences in the species included between the two models.