

# PRACTICE EXAM

**Difficulty: MEDIUM**

**Questions: 10**

## Thermodynamics & Fluids Exam - Chapters 3 & 4

Instructions: Please answer all questions to the best of your ability. Show your work for problem-solving questions.

### # Section 1: Multiple Choice (4 points each, 40 points total)

**Question 1:** Specific heat ratio,  $k$ , is defined as:

- A)  $c_p - c_v$
- B)  $c_v / c_p$
- C)  $c_p / c_v$
- D)  $c_p * c_v$

**Question 2:** For an incompressible substance, which property is typically assumed to be constant?

- A) Temperature
- B) Pressure
- C) Specific volume
- D) Internal energy

**Question 3:** In a polytropic process described by  $pV^n = \text{constant}$ , what type of process occurs when  $n = 0$ ?

- A) Isochoric
- B) Isothermal
- C) Isobaric
- D) Isentropic

**Question 4:** Which of the following devices is primarily designed to increase the pressure of a fluid?

- A) Nozzle
- B) Diffuser
- C) Turbine
- D) Compressor

**Question 5:** At steady state, what is true about the mass flow rate into and out of a control volume?

- A) The mass flow rate in is greater than the mass flow rate out.
- B) The mass flow rate out is greater than the mass flow rate in.
- C) The mass flow rate in is equal to the mass flow rate out.

D) The mass flow rate in and out are both zero.

## # Section 2: Short Answer (6 points each, 30 points total)

**Question 6:** Explain the two special methods used to evaluate properties of liquids and solids when compressed liquid tables are unavailable.

**Question 7:** Describe the key assumptions made when modeling a substance as incompressible.

**Question 8:** Define a polytropic process and provide two examples of processes with specific values of 'n' and their corresponding names.

**Question 9:** What is the primary purpose of a throttling device, and what usually happens to the pressure and temperature across it?

**Question 10:** What is the conservation of mass principle and why is it important in control volume analysis?

## # Section 3: Problem Solving (10 points each, 30 points total)

**Question 11:** A 2-kg block of aluminum at 100°C is placed in 1 kg of water at 20°C in an insulated container. Assuming the aluminum and water are incompressible with constant specific heats of 0.9 kJ/kg.K and 4.18 kJ/kg.K, respectively, determine the final equilibrium temperature of the system.

**Question 12:** Air enters a diffuser at 200 kPa and 25°C with a velocity of 300 m/s and exits with a velocity of 50 m/s. Assuming the process is adiabatic and kinetic energy effects are significant, determine the exit temperature of the air, assuming constant specific heats with  $c_p^* = 1.005 \text{ kJ/kg}\cdot\text{K}$ .  
\*Note: This problem only needs to be set up with the correct equation(s) ready to be solved.\*