

CHEMISTRY 9th (PRE-BOARD EXAM. 2024-25)

Section "A"

Name: ABDULLAH Roll No: 802743

Time Allowed: 03 Hours

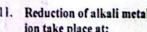
Time Allowed: 15 minutes

Max. Marks: 65

Marks: 12

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1.	Anion is formed by gaining one or more	O Neutron	O Proton	• Electron	O Positron
2.	Why is potassium iodide (KI) considered as powerful reducing agent?	Iodide ion is oxidized to iodine	Olodide ion is reduced to jodine	O lodine is reduced to iodine ion	O _{lodine} is oxidized to iodine ion
3.	The f-subshell can accommodate a maximum of	14 electrons	O10 electrons	O 06 electrons	O 02 electrons
4.	Which group of elements contains incomplete s-subshell?	Group I-A	O Group II-A	O Group III-A	O Group IV-A
5.	In Sodium Chloride, Chlorine after gaining one electron attains the electronic configuration of:	O Neon	• Argon	O Krypton	O Xenon
6.	Helium obeys duplet rule. How many electrons are in its valance shell:	0 1	2	0 3	O 4
7.	Which of the given two elements will form ionic bond:	O Na and K	Q K and Ca	• Ca and CI	O Cl and C
8.	Which of the given solution is more dilute?	• 10 ⁻² M	(10.3W	10 ³ M	10⁴M
9.	Halogens react with metals to form	O Halides	O Oxíde	O Halogen sulphides	Hydrogenated compound
10.	Which one of the given acts as oxidizing agent in the given reaction? Br ₂ + H ₂ S → 2HBR + S	O Br ₂	О Н	• \$	O H ₂ S
11.	Reduction of alkali metals ion take place at:	O Anode	OWalls of cell	O Battery	◆ Cathode
12.	The fifth element of the first	O Charmium	O Titonium	0.44	O Vandana





transition series is:

Chromium

O Titanium

O Vanaidum

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