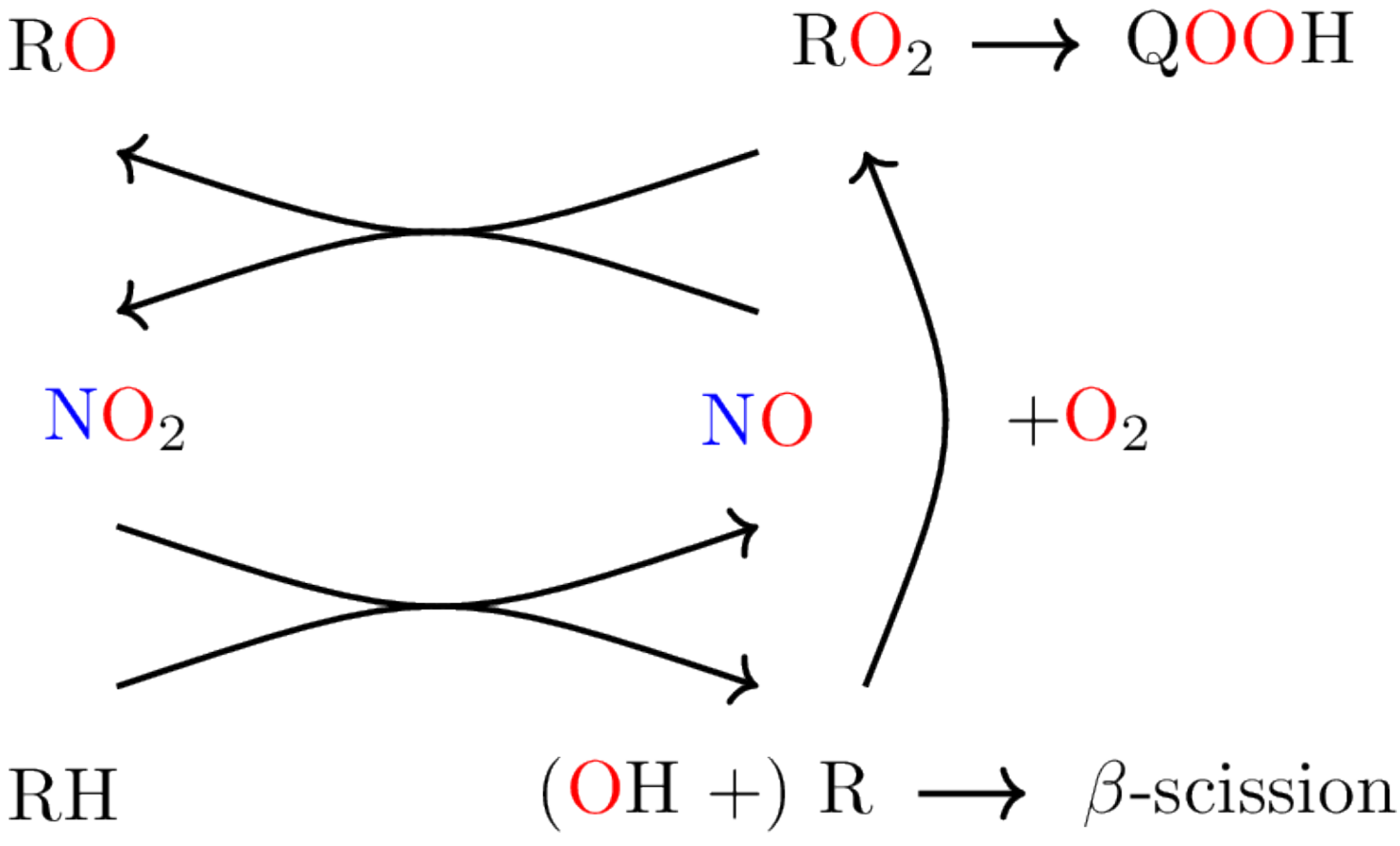
Progress in ~~Nitrogen~~ Novel Combustion Chemistry

Mark E. Fuller, Ph.D.

Physico-Chemical Fundamentals of Combustion

January 31, 2021

NO_x interactions in hydrocarbon combustion



And when RH is replaced with QOOH or OOQOOH?

Citations

- **Regular (end) citation[1]**
- **Footnote**[1]

[1] Fuller, M. E. et al. Review of Scientific Instruments **2019**, *90*, 064104.

Reaction Classes and Examples

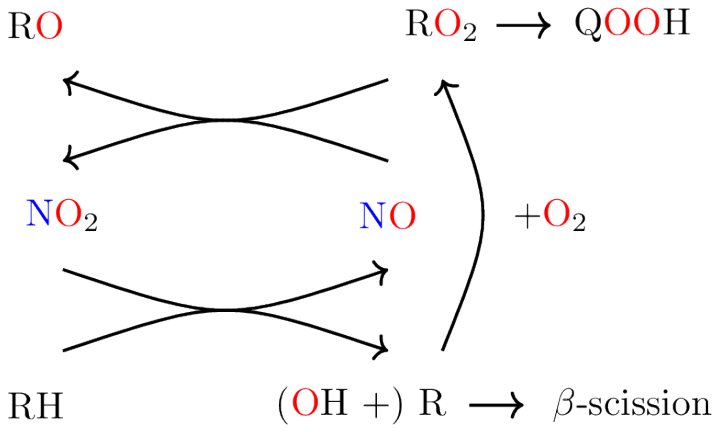
Develop mechanism by systematic inclusion of reaction classes

- **Hydrogen abstractions by NO_x to form HONO, HNO_2, HNO**
- **Unimolecular conformer formation and dissociation**
  - → $RNO_2 \rightleftharpoons R + NO_2$
  - → $RONO \rightleftharpoons RO + NO$
  - → $RONO_2 \rightleftharpoons RO + NO_2$
- **Isomerizations**
  - → $RONO \rightleftharpoons RNO_2$
- **Concerted HONO elimination**
  - → $RONO \rightleftharpoons$ alkene + HONO
- **NO_x cycling reactions**
  - → $RO_2 + NO \rightleftharpoons RO + NO_2$
  - → $RO + NO \rightleftharpoons R + NO_2$

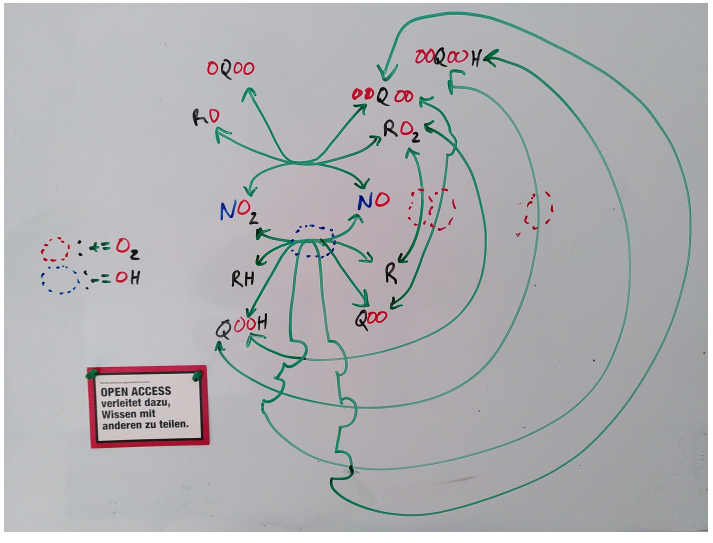
The old (slow) way forward

1. **Calculate sensitivities**
2. **Tweak/add some rates***
3. **Run simulations**
4. **Feel sad and start over**

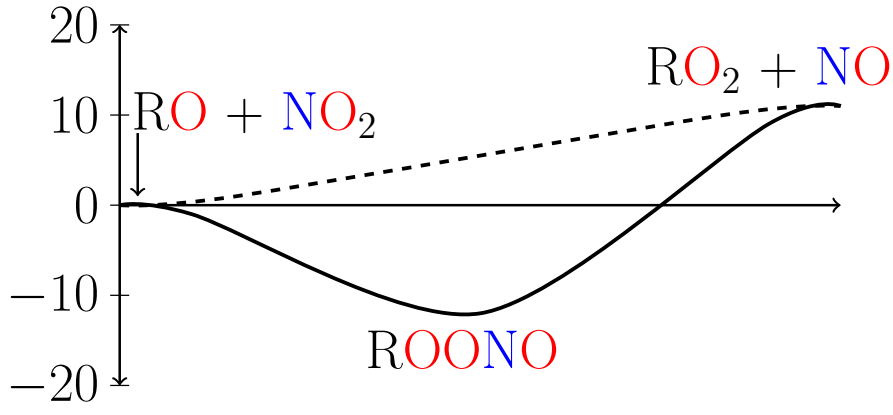
*Until this...



...becomes this (or worse!)



Progress on NO_x-Cycling



Generalized potential energy surface for alkoxy radical (RO) + NO_2 system. Energies in kcal/mol. Well-skipping occurs at virtually all combustion-relevant temperatures and pressures.

Reaction	A	n	E_a
$CH_3O_2 + NO \rightleftharpoons CH_3O + NO_2$	4.62E+15	-0.38	97.8
$C_2H_5O_2 + NO \rightleftharpoons C_2H_5O + NO_2$	2.11E+14	-0.12	-470.6
$NC_3H_7O_2 + NO \rightleftharpoons NC_3H_7O + NO_2$	1.07E+14	-0.25	-1302.0

Units: centimeters, kelvin, calories, moles

Progress on NO_x-Cycling

(1) Fuller, M. E. Energy Conversion and Management **2014**, *88*, 199–205.

(2) Fuller, M. E.; Skowron, M.; Tranter, R. S.; Goldsmith, C. F. Review of Scientific Instruments **2019**, *90*, 064104.

Mark E. Fuller, Ph.D. – fuller@pcfc.rwth-aachen.de

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RWTH Aachen University
Schinkelstr. 8
52062 Aachen

www.pcfc.rwth-aachen.de