

①

$$[A] + [B] = N \quad \rightarrow \textcircled{1}$$

$$\dot{[B]} = \frac{\beta[B][A]}{N} - \gamma[B]$$

$$\dot{[B]} = \frac{\beta[B](N - [B])}{N} - \gamma[B] \quad \{ \text{from } \textcircled{1} \}$$

$$\dot{[B]} = \frac{\beta[B]}{N} - \frac{\beta[B]^2}{N} - \gamma[B]$$

$$= -\frac{\beta[B]^2}{N} - \gamma[B] + \beta[B]$$

$$\dot{[B]} = -\frac{\beta[B]^2}{N} + (\beta - \gamma)[B]$$

$$\dot{[A]} = -\frac{\beta[B][A]}{N} + \gamma[B]$$

$$\dot{[A]}_2 = -\frac{\beta(N - [A])[A]}{N} + \gamma[B] (N - [A])$$

$$= -\frac{\beta N[A]}{N} + \frac{\beta[A]^2}{N} + \gamma[B] N$$

$$= -\gamma[A]$$

$$\dot{[A]} = -\beta [A] + \frac{\beta}{N} [A]^2 + N\gamma - \gamma [A]$$

$$[A] = 1 + \frac{1}{N} \beta [A]^2 - \gamma N$$

$$[A] = \frac{\beta}{N} [A]^2 - (\beta + \gamma) [A] + N\gamma$$

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$$\int [B] = -\frac{\beta[B]^2}{N} + (\beta - \gamma)[B]$$

$B_0(t) \quad [B](t)$

$$\int_{B_0} [B] = -\frac{\beta y^2}{N} + (\beta - \gamma)y$$

$$= -\frac{\beta y^3}{3N} + \frac{(\beta - \gamma)y^2}{2}$$

$$= -\frac{\beta([B](t))^3}{3N} + \frac{(\beta - \gamma)([B](t))^2}{2}$$

$$+ \frac{\beta(B_0)^3}{3N} + \frac{(\beta - \gamma)[B_0]^2}{2}$$

$$\frac{[B]}{[B]^2} = \beta y - \frac{\beta}{N} + \gamma y$$

$$\int \frac{[B]}{[B]^2} = \int (\beta + \gamma)y - \frac{\beta}{N}$$

$$\frac{[B]}{[B]^2} = \frac{(\beta + \gamma)y^2}{2x(y')} - \frac{\beta y}{N(y')} + \dots$$

$$[B]y^2 = (\beta + \gamma)y^2 - \frac{\beta y}{N(y')} + \frac{1}{[B](t)} \quad \text{if } y = \frac{1}{[B]}$$

factor B\* for B\*

$$= \frac{(\beta + \gamma) \cdot 1}{2 \cdot \ln[B(t)] [B]^2} - \frac{\beta}{N \cdot (\ln[B(t)] \cdot [B](t))}$$

equate it to 0. find root

$$-\frac{(\beta + \gamma)}{\ln[B_0] [B_0]^2} - \frac{\beta}{N (\ln[B_0]) [B_0]} = 0$$

$$\frac{(\beta + \gamma)}{2 \cdot \ln[B(t)] [B(t)]^2} - \frac{\beta}{N \cdot (\ln[B(t)]) [B(t)]} = 0$$

$$-\frac{(\beta + \gamma)}{\ln[B_0] [B_0]^2} - \frac{\beta}{N (\ln[B_0]) [B_0]} = 0$$

$$\frac{(\beta + \gamma)}{2} - \frac{\beta [B(t)]}{N} = (\beta + \gamma) ([B(t)])^2$$

$$- \frac{\beta t [B(t)]}{N [B_0]} = \frac{(\beta + \gamma) ([B(t)])^2}{N [B_0]}$$

$$f(n) = -\frac{(\mathbb{E}[B](t))^2}{\mathbb{E}[B_0]} - \frac{\beta \mathbb{E}[B](t)}{\mathbb{E}[B_0]N} + \frac{\beta (\mathbb{E}[B](t))}{N} \quad \text{--- (1)}$$

$$\mathbb{E}[B](t) = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a} - \frac{(\beta + \gamma)}{2} = 0$$

$$\beta^* = \frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right) \pm \sqrt{\left( \frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right) \right)^2 - \frac{4}{2x \mathbb{E}[B_0]}}$$

$\frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right)$

$$\beta^* = B_0 \left[ \frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right) \pm \sqrt{\frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right)^2 + \frac{(\beta + \gamma)}{B_0}} \right]$$

Check stability.

$$f'(n) = -\frac{2(\mathbb{E}[B](t))^2}{\mathbb{E}[B_0]} - \frac{\beta}{\mathbb{E}[B_0]N} \quad \text{eqn (1)}$$

check value

If  $f'(n) < 1$  stable  
 $f'(n) > 1$  unstable.

Substitute

$$-\left[ \frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right) \pm \sqrt{\frac{\beta}{N} \cdot \left( \frac{1+i}{B_0} \right) + \frac{(\beta+i)}{B_0}} \right]$$

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$$\frac{\beta}{[B_0]N}$$

for.  $\beta = 0.5$ ,  
 $B_0 = 1$ ,  $N = 1000$ ,  $i = 0.5$

$$-\left[ \frac{0.5}{1000} \left( \frac{1}{1} + 1 \right) \pm \sqrt{\frac{(0.5)}{1000} \left( \frac{1+1}{1} \right)^2 + (0.5+0.5)^2} \right]$$

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$$\frac{0.5}{1 \times 1000}$$

$$-\left[ \frac{0.5}{1000} \pm \sqrt{\frac{(0.5 \times 4 + 1)^2}{1000}} \right] - \frac{0.5}{1 \times 1000}$$
$$-\left[ 0.0001 \pm \sqrt{(1 + 0.0002)^2} \right] - 0.0005$$

$$= -[0.0001 \pm 1.0001] - 0.0005$$
$$= -0.0001 \mp 1.0001 - 0.0005$$

~~overnight with positive root~~ unstable  
equilibrium  
~~overnight with negative root~~ stable equilibrium

stable equilibrium

$$\beta^* = B_0 \left[ \frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right) \right]$$

$$\sqrt{\frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right)^2 + \frac{(\beta + \gamma)}{B_0}}$$

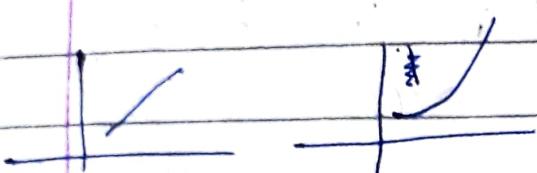
unstable equilibrium

$$\beta^* = B_0 \left[ \frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right) - \sqrt{\frac{\beta}{N} \left( \frac{1}{B_0} + 1 \right)^2 + \frac{(\beta + \gamma)}{B_0}} \right]$$

5) on  $B_0 > 0$  ~~the graph will be~~  
 $[B](t) > 0$

on different value of  $B_0$   
graph shifts horizontally  
~~or~~ vertically upwards  
(for +ve  $B_0$  values) or downwards (for -ve values of  $B_0$ )

We get a linear function  
on changing value of  $R_o$   
but on changing  $\gamma$  simultaneously  
we will get ~~a~~ a ~~exponential~~  
~~parabola~~.



$R_o$  change  $\gamma$  &  $R_o$  change.

### Simulation

3) with An gillespie : is for small  $N$  only.

we observed that a few ~~value~~  
of  $[A]$  and  $[B]$  are ~~are~~ similar  
to mean field

& for  $[B]$  a little difference is  
observed for value of  $B$ .

for correcting it, a ~~value~~ value  
of  $R_o$ . ~~N~~  $N$  can  
be increased and it starts giving  
same values ~~for~~ for both (meanfield & gillespie)

4) for first case

$$\beta = 0.95, \gamma = 0.5, N = 1000$$

graph of meanfield

$[B](t)$  is inclined to 0.

$[A](t)$  is decreasing and a parabola

$[B]$  is superimposing on  $[B](t)$

$[A]$  is showing a quite deviation to  $[A](t)$ .

for improvement we can

increase value of  $N$  as  $[A]$  for

Gillipie will start overlapping meanfield.

$$\text{for } \beta = 0.95, \gamma = 0.5, N = 1000$$

graph of meanfield

$[B](t)$  is slightly upward 0 and it is a parabola.

$[A](t)$ , it is decreasing and becomes 0. Then it is showing

-ve values which means  $< 0$ .

as no negative <sup>number of</sup> particles may also occur

A bit of

$[B]$  is superimposed to  $[B](t)$   
 $[A]$  is deviated from  $[A](t)$   
increase  $N$  for superimposing it.

A bit of critical thinking.

- ① State  $A'$  and  $B'$  might be some physical or chemical state of a particle and equilibrium between two states is observed.

This model can be applied to water

- ① Filling of tank. If rate of filling of tank is maintained with rate of outflow of water from tank then it will appear that tank is neither filling nor losing water.

Tank is neither increasing nor decreasing.  
example

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i. Conversion of number of particles of oxygen in gaseous gaseous state {B} and liquid state [A]

It might be noticed that often in nature ~~most~~ [A] & [B] are in equilibrium