# Microscopic behavior of eigenvalues and charged particles

Thomas Leblé

CIMS

Courant Instructor Day

## Random Matrix Theory - I

 $(M_{i,j})_{1 < i,j < N}$  random variables  $\longrightarrow M$ , a  $N \times N$  random matrix

#### Quantum physics

- $\bullet \ \ Interactions \leftrightarrow Hermitian \ operator.$
- Energy levels ↔ eigenvalues.
- Complicated system ↔ "Random interaction".

# Random Matrix Theory - II

#### Dyson, 1962

(...) a mathematical idealization of the notion of "all physical systems with equal probability"

#### Universality?

Some properties may be **universal** i.e. fairly independent of the law of the matrix.

## Statistical physics

(Classical, at equilibrium, in Euclidean space, . . . )

N particles  $x_1, \ldots, x_N$  in  $\mathbb{R}^d$ .

- State of the system:  $\vec{X}_N = (x_1, \dots, x_N)$ .
- Energy in the state  $\vec{X}_N$  given by  $\mathcal{H}_N(\vec{X}_N)$ .

Complicated system,  $N \approx 10^{23}$  (Avogadro) very large  $\longrightarrow$  "random state"

- Microcanonical: the energy E is fixed and  $\{\vec{X}_N, \mathcal{H}_N(\vec{X}_N) = E\}$  are equiprobable
- Canonical: the temperature  $T = \frac{1}{\beta}$  is fixed and density

$$dP(\vec{X}_N) = \frac{1}{Z_{N,\beta}} \exp\left(-\beta \mathcal{H}_N(\vec{X}_N)\right) d\vec{X}_N$$

 $Z_{N,\beta} = \text{normalizing constant}$ 

#### Connection - I

 $N \times N$  coefficients for the random matrix = independent Gaussian random variables.

- + symmetry condition (Hermitian matrix, real eigenvalues)
- ono symmetry condition (non-Hermitian matrix, complex eigenvalues)

 $\vec{X}_N = (x_1, \dots, x_N)$  the eigenvalues. Explicit computation:

$$dP(\vec{X}_{N}) = \frac{1}{Z_{N}} \exp \left( -\left( c_{1} \sum_{i \neq j} -\log|x_{i} - x_{j}| + c_{2} \sum_{i=1}^{N} N|x_{i}|^{2} \right) \right) d\vec{X}_{N}$$

 $d\vec{X}_N$  = Lebesgue measure on  $\mathbb{R}$  or  $\mathcal{C}$ .

#### Connection - II

#### Morality

Eigenvalues ⊂ (charged) particles with

- **1** Logarithmic repulsion  $\leftrightarrow \sum_{i \neq j} -\log |x_i x_j|$  (singular + long-range)
- 2 External potential  $\leftrightarrow N|x|^2$  (confining)
- **3** Some values of the (inverse) temperature  $\beta$

#### Conversely?

"True" in the Hermitian case ( $\beta$ -ensembles).

# Macroscopic behavior - I

$$\vec{X}_N = (x_1, \dots, x_N) \longrightarrow \text{Density/Empirical measure } \mu_N := \frac{1}{N} \sum_{i=1}^N \delta_{x_i}$$

$$\int f(x)d\mu_N(x) := \frac{1}{N} \sum_{i=1}^N f(x_i) \quad \text{(definition)}$$

Fact: Equilibrium measure  $\mu_{\rm eq}$  such that  $\lim_{N \to \infty} \mu_N = \mu_{\rm eq}$  almost surely.

$$\int f(x)d\mu_N(x) \approx \int f(x)d\mu_{\rm eq}(x)$$

except for very small probability (order  $exp(-N^2)$ )

# Macroscopic behavior - II

$$d\mathbb{P}_{N,\beta}(\vec{X}_N) = \frac{1}{Z_{N,\beta}} \exp\left(-\beta \left(\sum_{i \neq j} -\log|x_i - x_j| + \sum_{i=1}^N NV(x)\right)\right) d\vec{X}_N$$

- Macroscopic behavior does not depend on β, depends on V (non-universal).
- Obtained by minimizing an energy functional from classical potential theory.
- Examples: semi-circle law  $\frac{1}{2\pi}\sqrt{4-x^2}\mathbf{1}_{[-2,2]}$ , circular law  $\frac{1}{\pi}\mathbf{1}_{|x|<1}$ .

# $\beta$ large

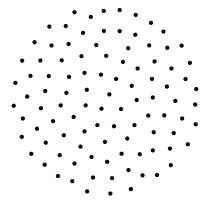


Figure:  $\beta = 400$ 

# $\beta$ small

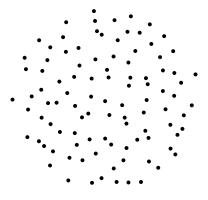


Figure:  $\beta = 5$ 

# Microscopic behavior

Observable "empirical field"  $P_N$  (law of a random point configuration).

- "Typically a square lattice"
- "1/2 triangular lattice, 1/2 square lattice"
- "1/3 lattice, 2/3 disorder"

#### **Theorem**

The empirical field behaves like minimizers of a certain "free energy" functional

$$\beta$$
 Energy + Entropy.

In particular, point processes arising in RMT must minimize the functional.

Competition: Energy favors "order" (crystal?) and entropy favors disorder (Poisson).

Microscopic behavior depends on  $\beta$ , not on V ('universal').

## Perspectives

- Other models (random polynomials, unitary matrices..).
- Understand minimizers (phase transition?).

Thank you for your attention!