

Using a Synchrotron to Explore the Potential Verwey Transition in Lead Rhodium Oxide

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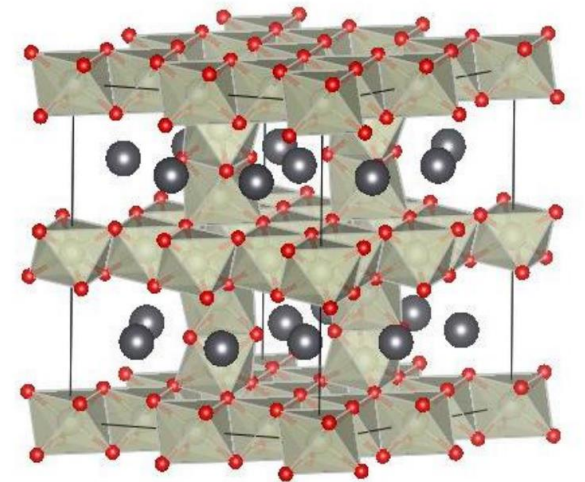
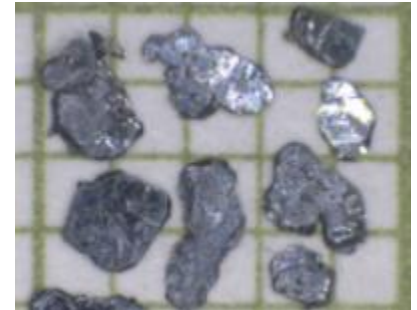
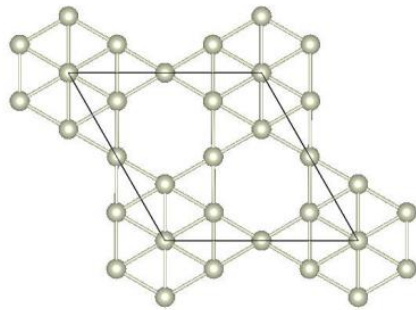
1. McMaster University, 2. Trent University, 3. Oak Ridge National Laboratory, 4. Cornell High Energy Synchrotron Source

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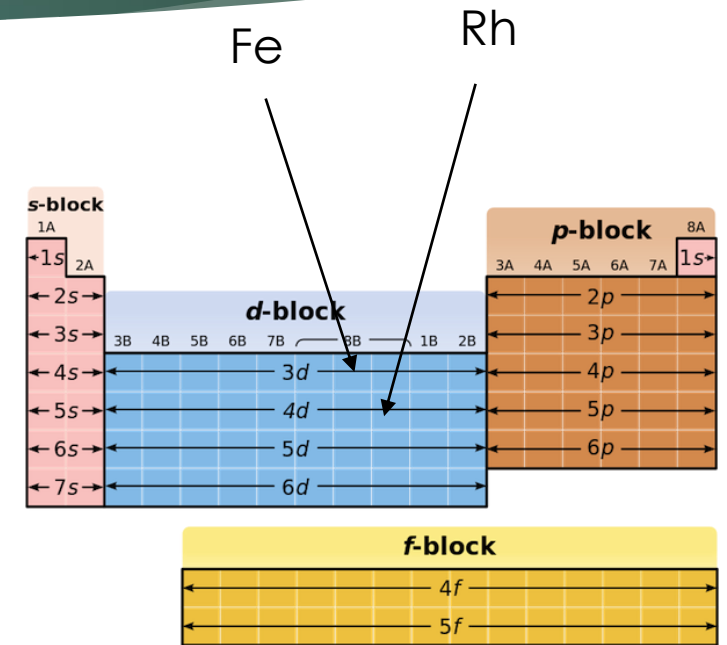
Lead Rhodium Oxide

- ▶ $\text{Pb}_3\text{Rh}_7\text{O}_{15}$
- ▶ Mixed Valence compound: $\text{Pb}_3[\text{Rh}^{3+}]_4[\text{Rh}^{4+}]_3\text{O}_{15}$
- ▶ Transition metal
- ▶ Space group: $P6_3/mcm$
- ▶ Hexagonal crystal structure
 - ▶ Lattice constants: $a = b = 10.35\text{\AA}$, $c = 13.28\text{\AA}$



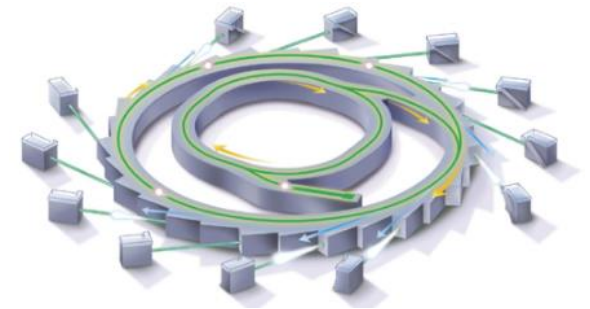
Verwey Transition

- ▶ A type of phase transition: properties change rapidly/discontinuously
- ▶ Electrical resistivity increases, charge ordering occurs
- ▶ Magnetite (Fe_3O_4) – 3d transition metal, $T_V \sim 125\text{K}$
- ▶ Lead rhodium oxide – 4d transition metal, change in space group at $\sim 185\text{K}$ (H. Mizoguchi et al, 2009)
- ▶ Use synchrotron to determine charge ordering below $\sim 185\text{K}$
- ▶ Would be first 4d or 5d transition metal oxide Verwey Transition



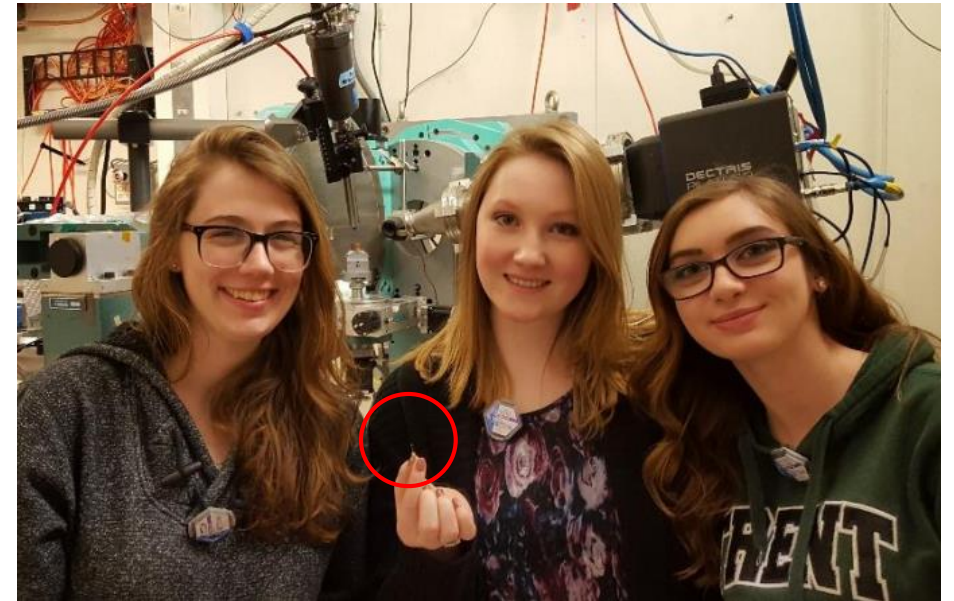
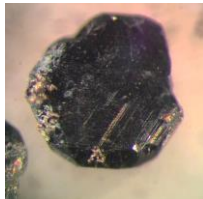
Synchrotron X-rays

- ▶ Electron changes direction, energy is emitted
- ▶ Faster electron means higher energy emitted
- ▶ X-ray wavelengths can be achieved if electron is moving fast enough
- ▶ Synchrotron accelerates electrons, then changes their direction periodically, producing X-rays
- ▶ The X-rays are directed towards beamlines



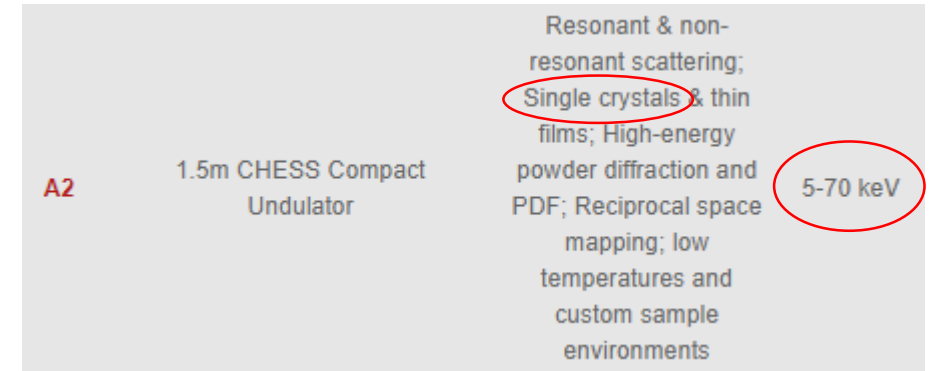
Cornell High Energy Synchrotron Source

- ▶ Travelled to Ithaca, NY to use Cornell High Energy Synchrotron Source (CHESS) beamline A2
- ▶ 99.9999995% speed of light
- ▶ Circumference: 768m
- ▶ Week long experiment, with the first day being used to mount the crystal and align the beam properly



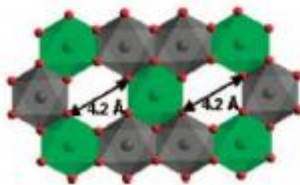
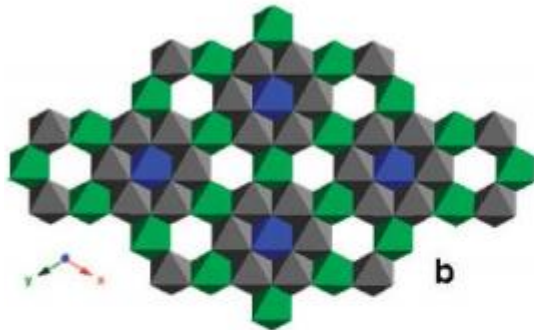
Details of the Experiment

- ▶ Beamline A2 can receive electromagnetic radiation of energies 5-70 keV
- ▶ Used 23.22 keV, K-edge for Rh
 - ▶ Energy required to eject an electron from the innermost shell
- ▶ Used liquid nitrogen stream to get sample to ~185K
 - ▶ Sometimes the mass would increase too much and sample would move, would need to realign
- ▶ Diffraction patterns produced in real-time

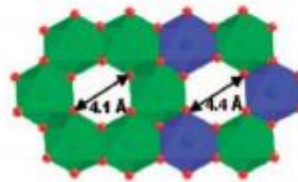
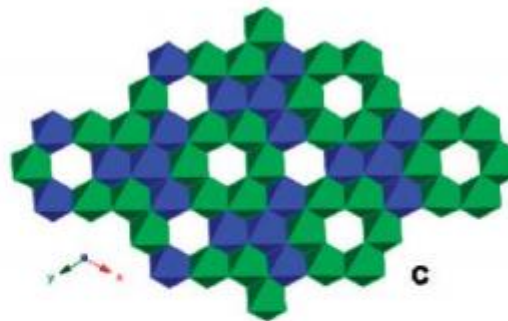


Results of the Experiment

- Predicted charge ordering:



High Temperature



Low Temperature

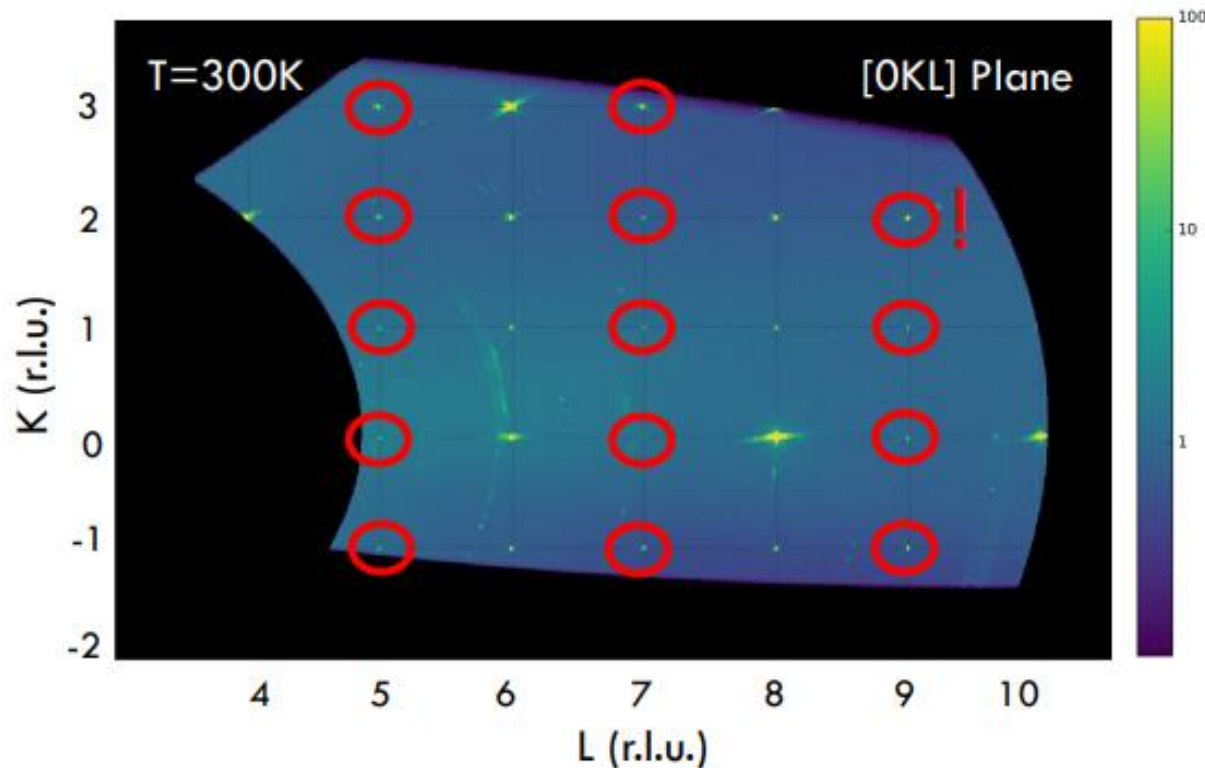
Green = Rh^{4+}

Blue = Rh^{3+}

Grey = Intermediate Charge

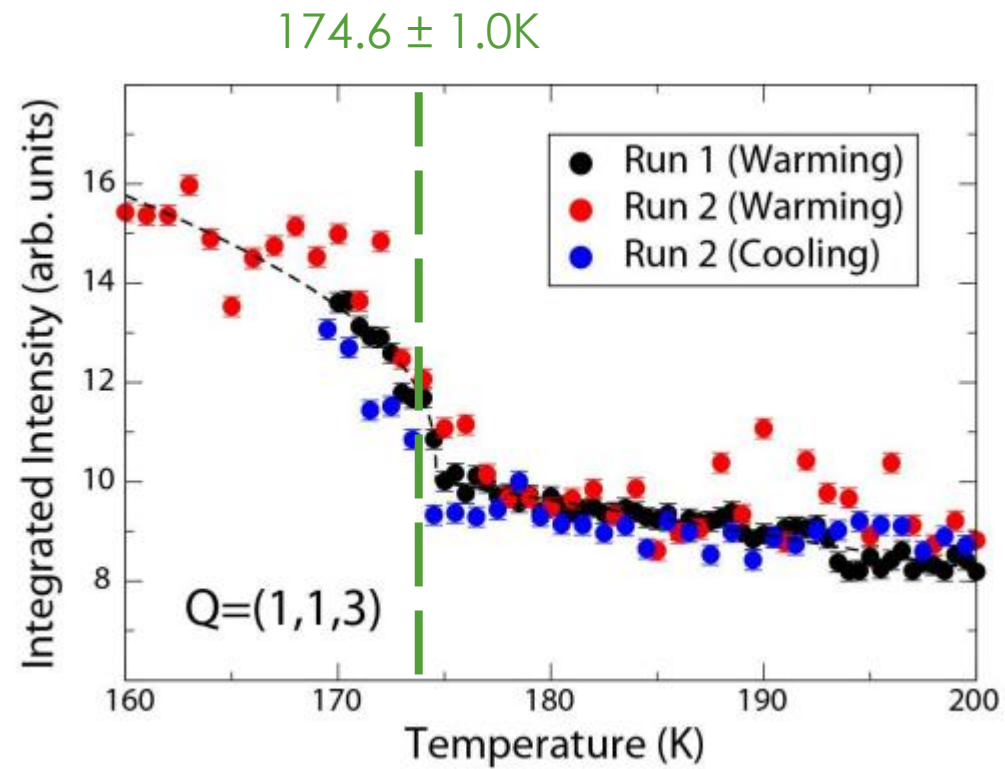
H. Mizoguchi et al,
Chem. Mater. (2009)

Results of the Experiment



- ▶ For the space group $P6_3/mcm$, peaks should only be occurring on this plane (0KL) at $2n$
- ▶ But structurally forbidden peaks are present both below AND above $\sim 185\text{K}$
- ▶ This suggests that the structure of lead rhodium oxide is more complicated than originally predicted

Results of the Experiment



Results of the Experiment

- ▶ Charge ordering seen:

Summary and Conclusions

- ▶ $\text{Pb}_3\text{Rh}_7\text{O}_{15}$ is the first 4d transition metal oxide candidate for a Verwey Transition, at $\sim 175\text{K}$
- ▶ The structure of lead rhodium oxide is more complex than predicted in 2009, at low and high temperatures
- ▶ At low temperatures ($\sim 175\text{K}$), the crystal structure becomes twinned trigonal

Acknowledgments

- ▶ Melissa Van Bussel is supported by the Trent University Physics Department and the Trent Central Student Association through the Academic, Personal, Professional and Leadership Development (APPLE) Fund.

