

# Ultra-fast oxygen conduction in Sillén Oxychlorides

Jun Meng

University of Wisconsin-Madison

[jmeng43@wisc.edu](mailto:jmeng43@wisc.edu)

October 17, 2024

PRiME 2024, Symposium: I02: Solid State Ionic Devices 15



**Dane Morgan (UW)**



**Ryan Jacobs (UW)**



**Md Sariful Sheikh (UW)**



**Jian (Jay) Liu (NETL)**

**William O. Nachlas (UW-Madison)**

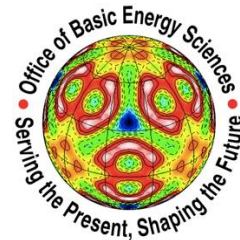


**Maciej Polak (UW)**



**Lane Schultz (UW)**

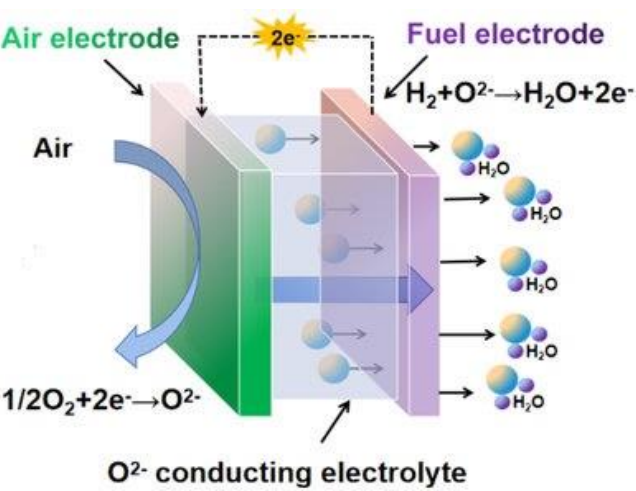
## Acknowledgement



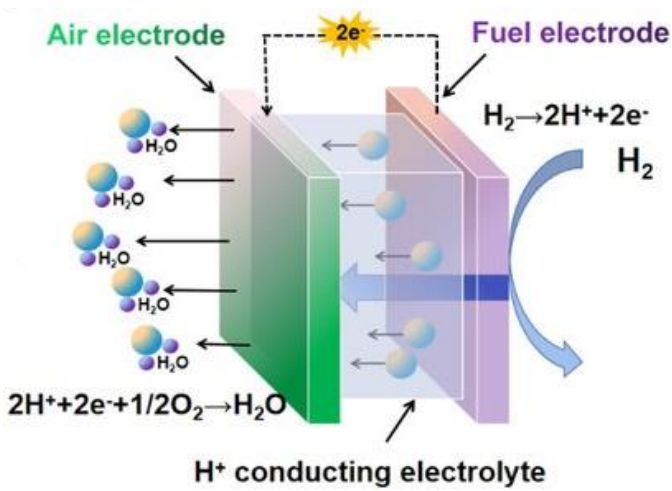
**DOE BES  
Materials  
Chemistry**



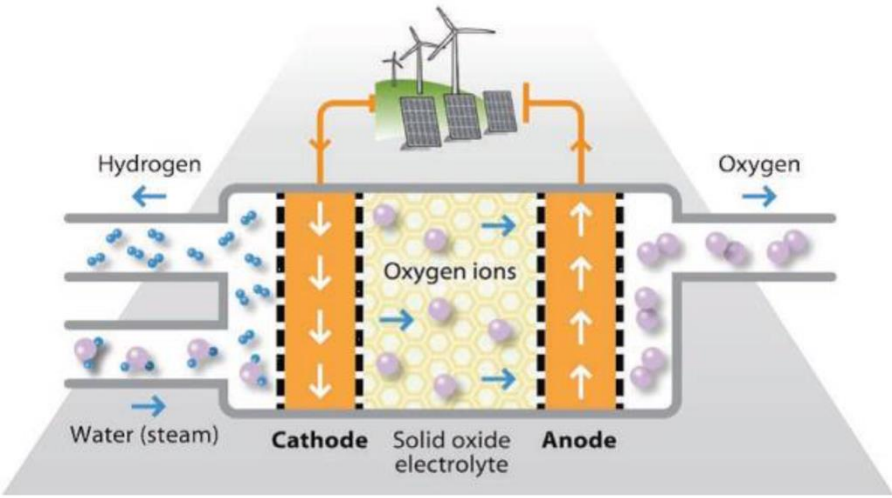
Solid Oxide Fuel Cell



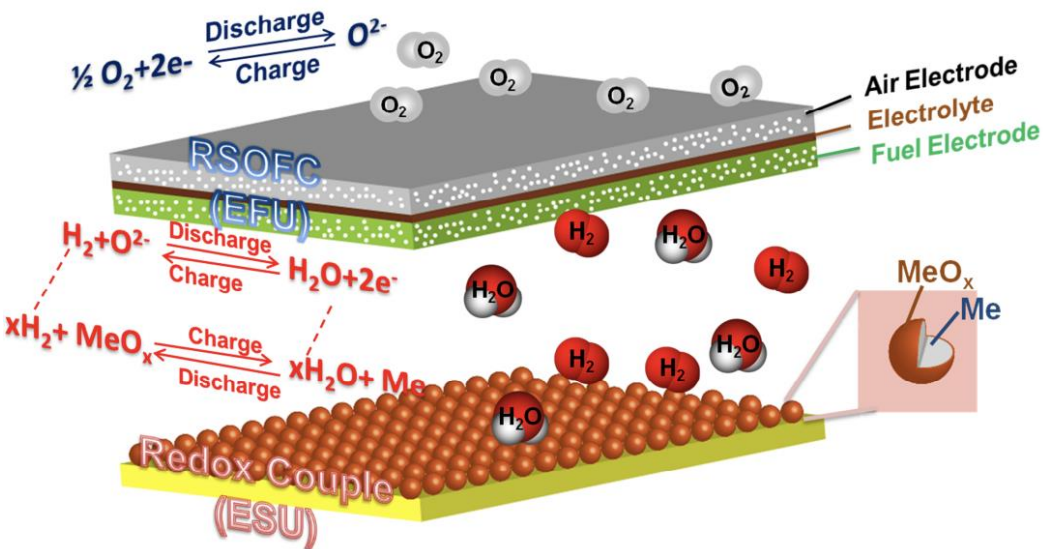
Proton-Exchange Membrane Fuel Cell



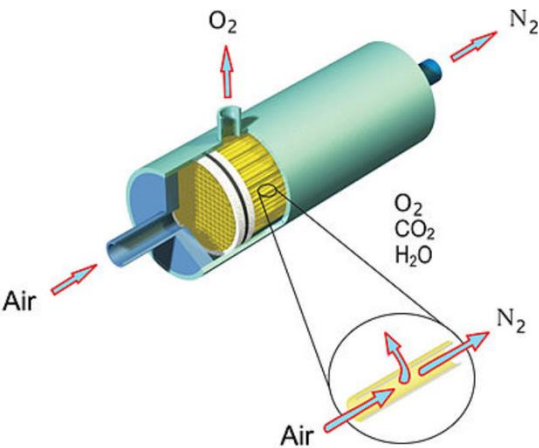
Solid Oxide Electrolysis Cell



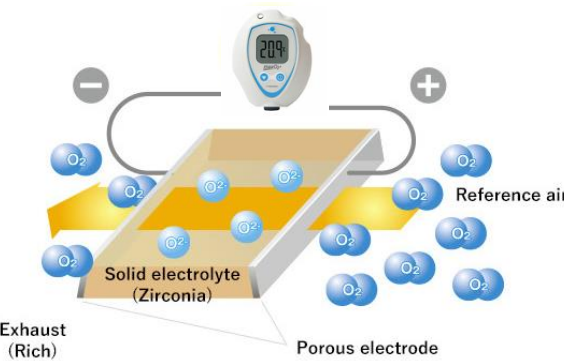
Solid Oxide Air Battery



Oxygen Separation Membrane



Oxygen Sensor

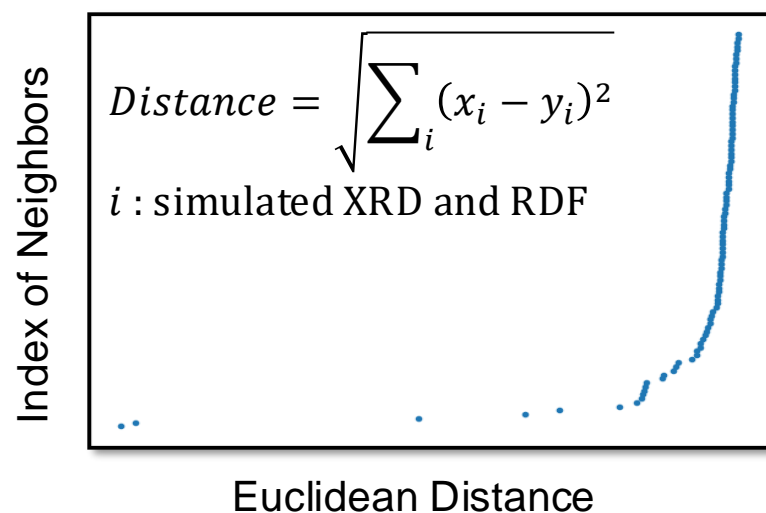


Energy Mater 2021;1:100002; ECS Trans. 2014; 58 67; Int J Energy Res. 2020; 44: 594– 611.  
[https://www.ngkntk.co.jp/english/product/sensors\\_plugs/zirconia\\_oxygen.html](https://www.ngkntk.co.jp/english/product/sensors_plugs/zirconia_oxygen.html)

# Materials discovery based on structural features

**Hypothesis:** materials with similar structural features to known oxygen conductors are promising for high oxygen mobility.

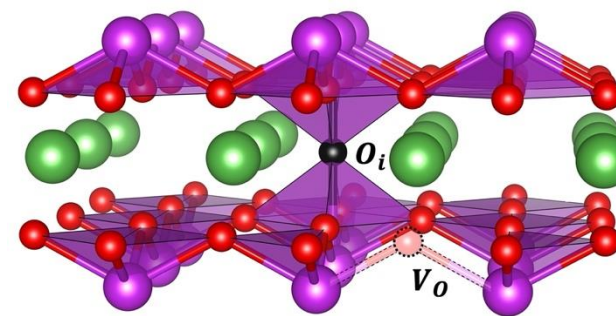
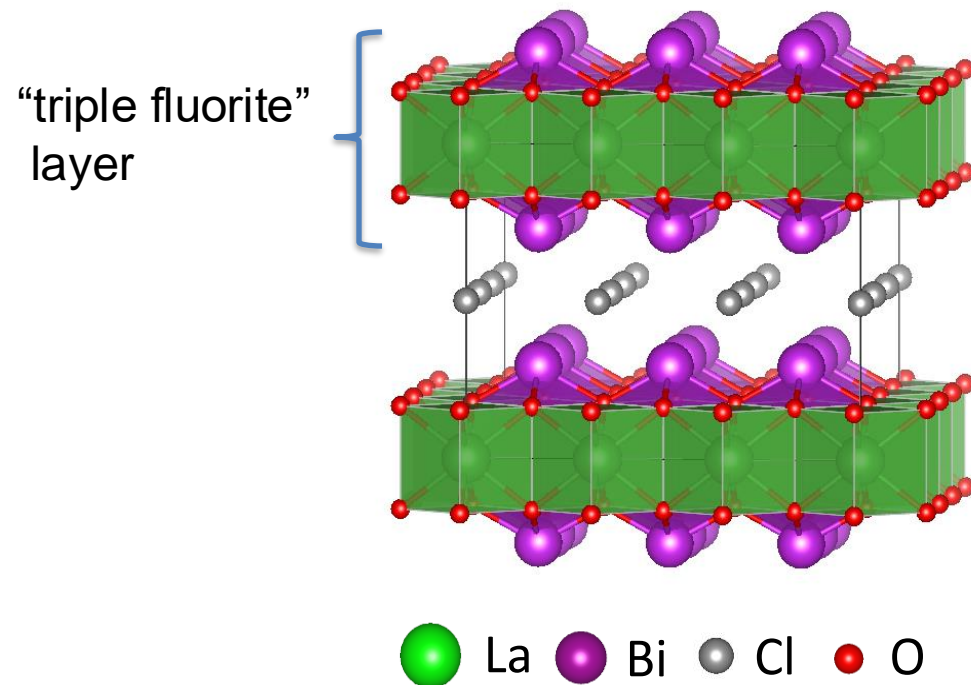
- 62,000 oxygen-containing compounds from the Materials Project were considered.
- Nearest neighbors were ranked based on their structural similarity to fluorite  $\text{BiO}_2$ , described by XRD of oxygen sublattice and RDF
- The  $\text{Bi}_2\text{MO}_4\text{X} = \text{MBi}_2\text{O}_4\text{X}$  (M= Rare-earth element, X= halogen) family was found as the nearest neighbors.  $\text{Bi}_2\text{LaO}_4\text{Cl}$  was selected as a prime candidate for further in-depth studies.



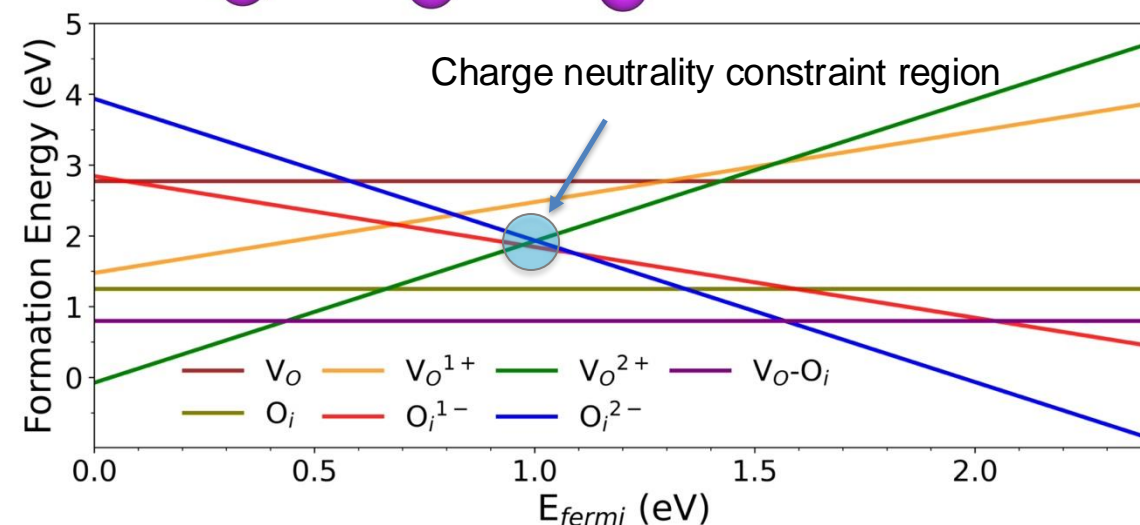
Nearest neighbors to fluorite $\text{BiO}_2$			
#	Materials ID	Formula	Distance
1	mp-546621	$\text{Bi}_2\text{ErBrO}_4$	18.67
2	mp-546350	$\text{Bi}_2\text{TmBrO}_4$	19.15
3	mp-549127	$\text{Bi}_2\text{HoClO}_4$	20.35
4	mp-549728	DyZnPO	21.85
5	mp-552738	$\text{Bi}_2\text{TmIO}_4$	22.69
6	mp-546625	$\text{Bi}_2\text{HoBrO}_4$	22.69
7	mp-3589	$\text{BPO}_4$	24.14
8	mp-1087483	ThCuPO	24.14
9	mp-6790	$\text{Ba}_2\text{Y}(\text{CuO}_2)_4$	24.17
10	mp-8789	$\text{Ca}_4\text{As}_2\text{O}$	24.25



- $\text{LaBi}_2\text{O}_4\text{Cl}$  (LBC) is a layered bismuth oxyhalide, adopts a “triple fluorite” layer  $[\text{Bi}_2\text{LaO}_4]^+$ , balanced by the  $\text{Cl}^-$  layer.
- LBC is a wide band gap insulator with  $E_{\text{gap}} = 2.65 \text{ eV}$ , consistent results achieved from exp., DFT-HSE, literatures.
- The Frenkel pair ( $V_{\text{O}} - \text{O}_i$ ) is the most thermodynamically stable defects in a wide T (300K, 873K and 1073K) range under atmospheric  $\text{P}(\text{O}_2)$ .

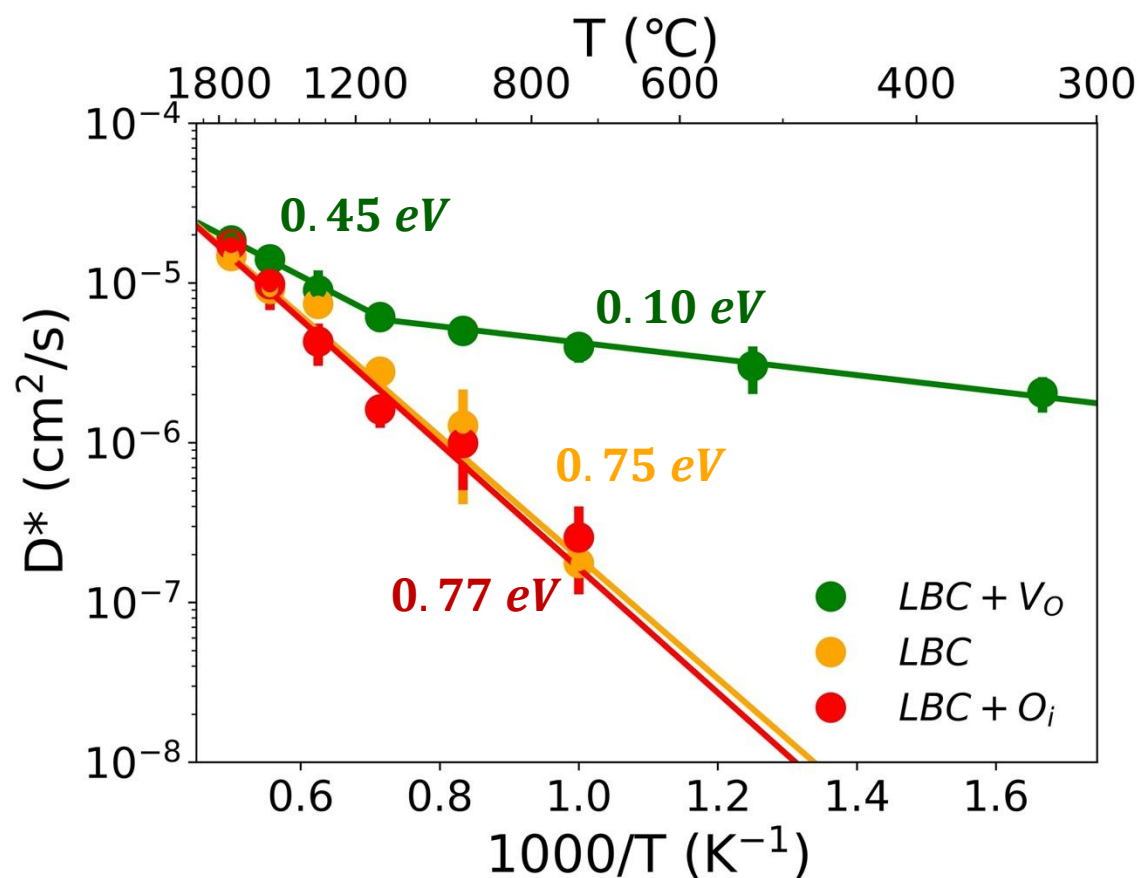


Frenkel pairs  
dominant in LBC



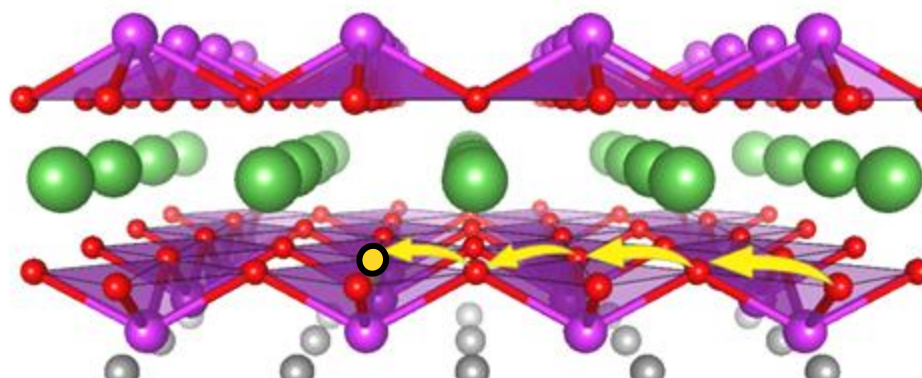
Formation energy of oxygen defects at room temperature, atmosphere

Ab initio studies indicate that vacancy diffusion in the “triple-fluorite” layer features an ultra-low migration barrier of  $\sim 0.1$  eV !

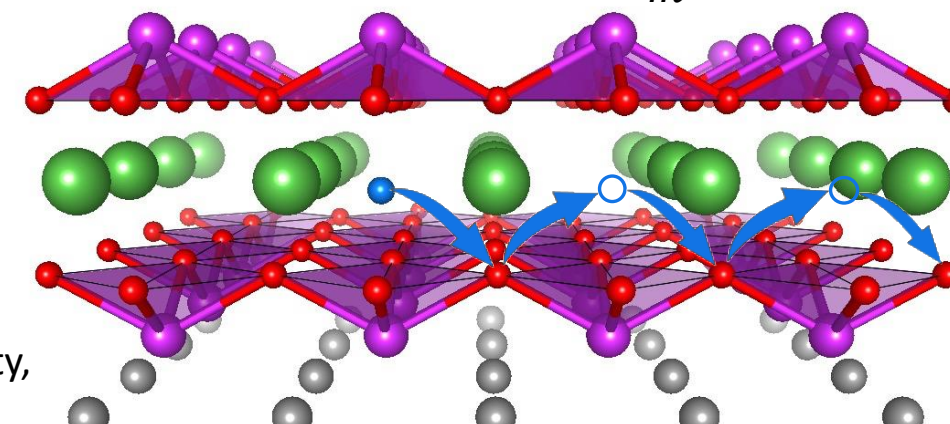


- With oxygen vacancy  $V_O$ , LBC exhibits ultra-high oxygen self-diffusivity, and ultra-low migration barrier
- With or without oxygen interstitial  $O_i$ , LBC exhibits similar oxygen self-diffusivities, and moderate migration barriers

Vacancy-mediated  $E_m$ : **0.1 eV**

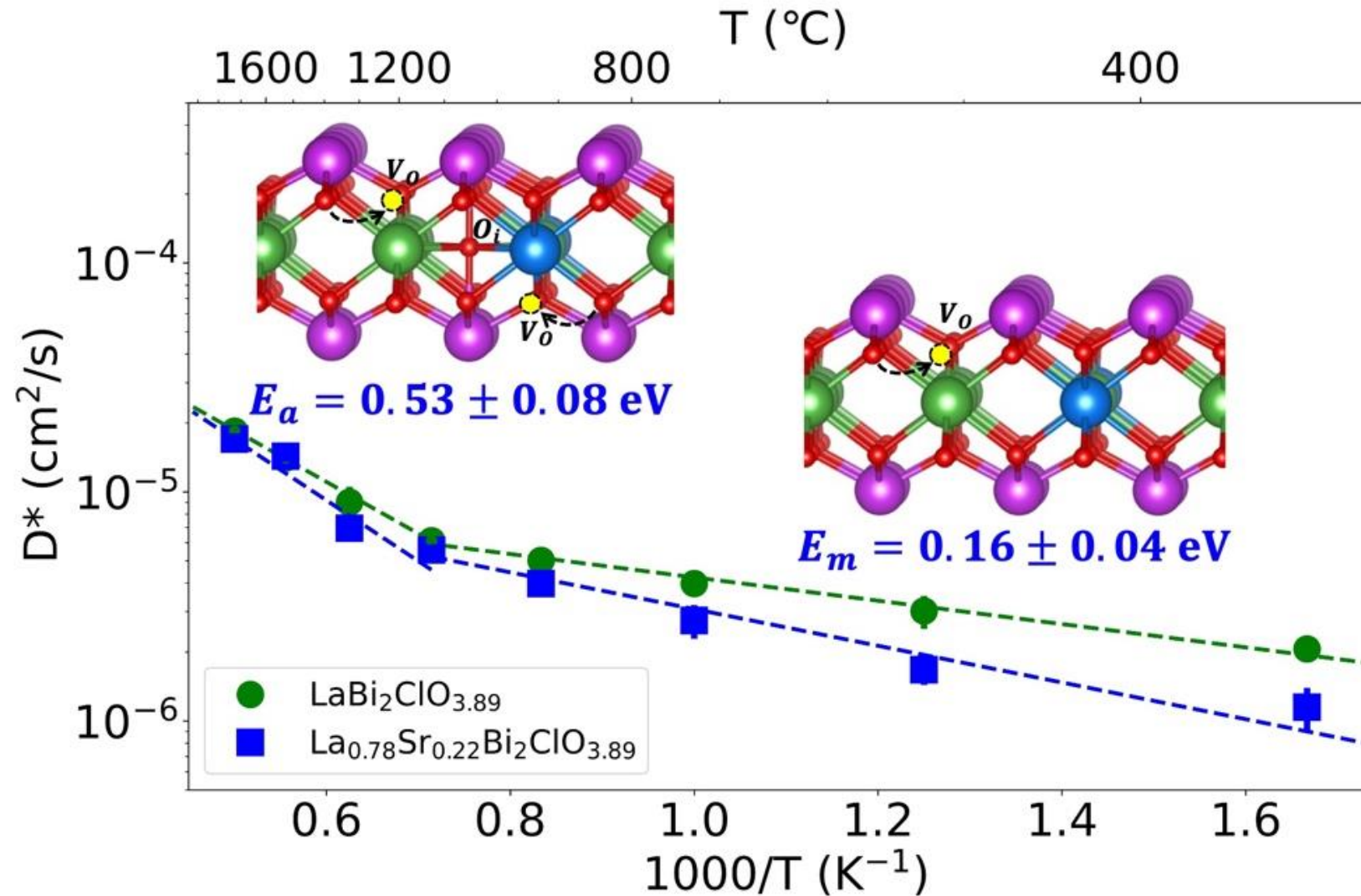


Interstitial-mediated  $E_m$ : **0.6 eV**



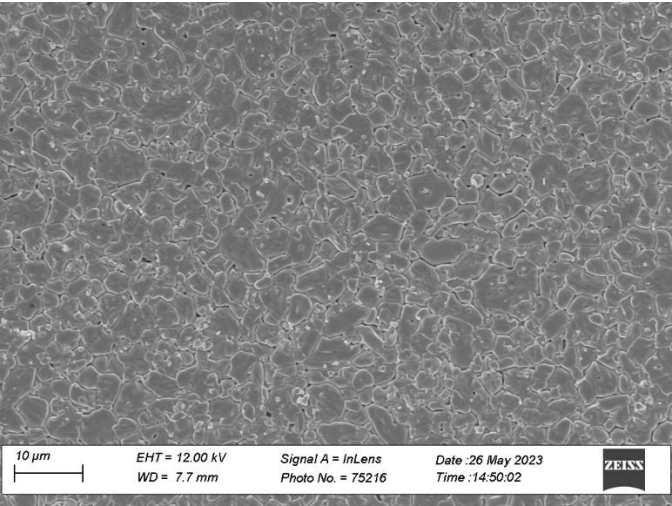

 La Bi Cl O  $V_O$   $O_i$   $O_i$  site

Similar oxygen diffusivity and diffusion barriers observed in Sr-doped LBC

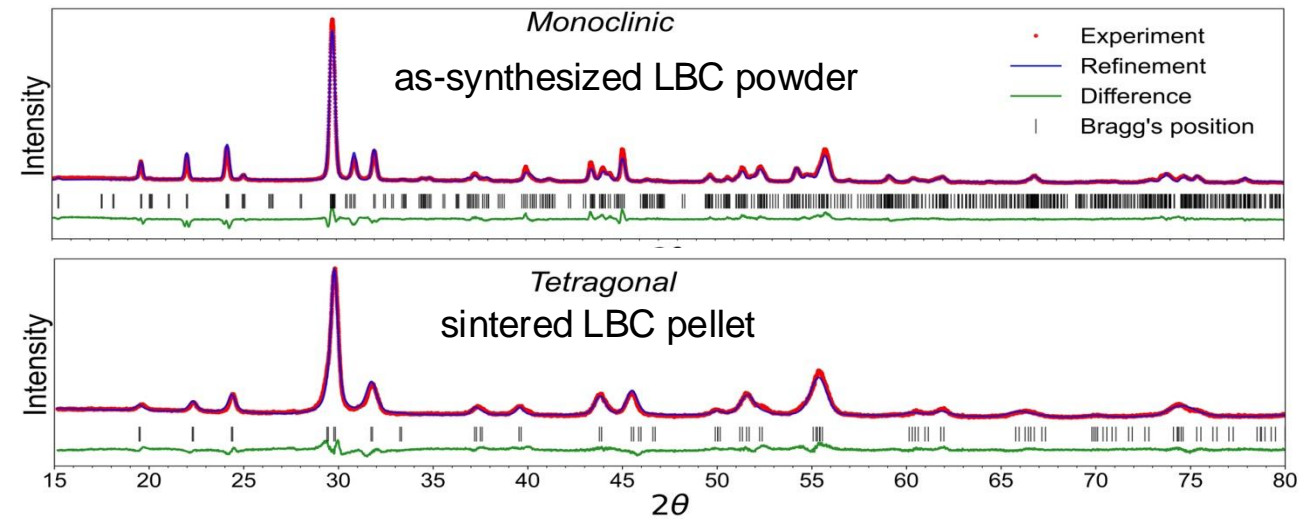




SEM of sintered LBC with grain size of 2.9  $\mu\text{m}$



XRD studies have revealed that LBC undergoes a phase transition from monoclinic to tetragonal after sintering.



Stoichiometry of LBC and Sr-doped LBC analyzed by Field Emission Electron Probe Microanalyzer (FE-EPMA) analysis, indicating small slight La and oxygen deficiencies, and excess Cl in LBC

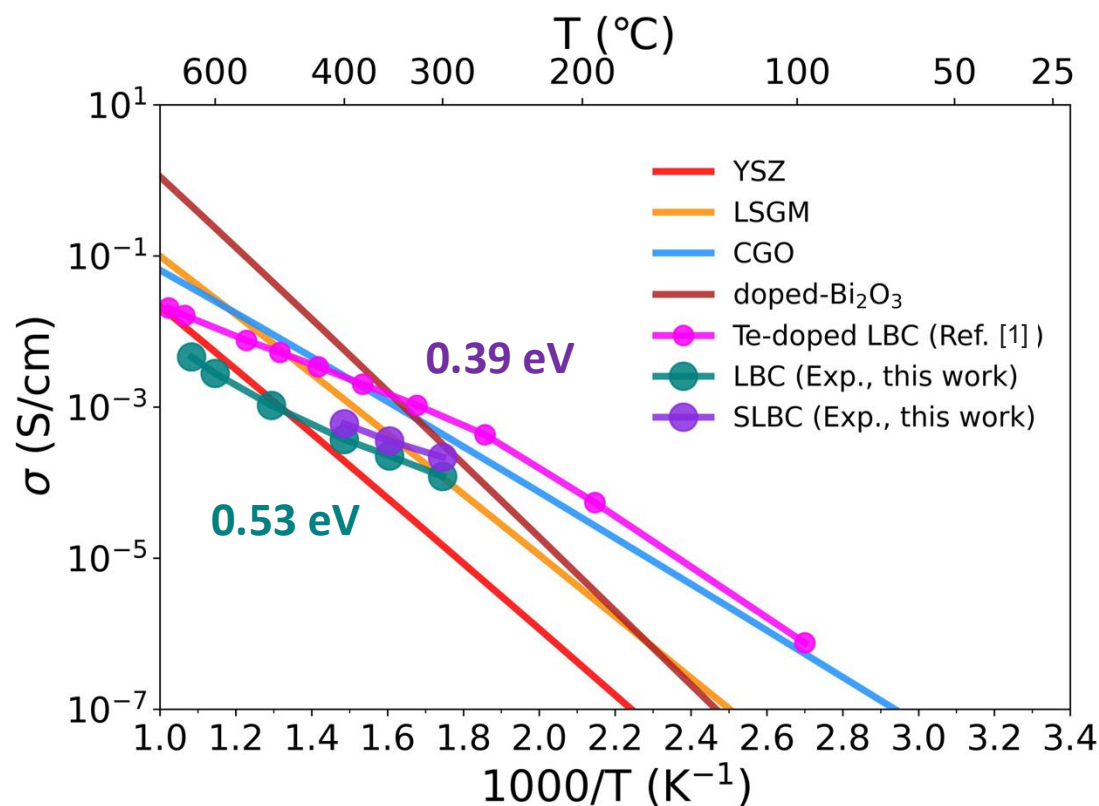
	La	Sr	Bi	O	Cl	Suggested formula
Ideal LBC	1	-	2	4	1	$\text{LaBi}_2\text{O}_4\text{Cl}$
synthesized LBC	$0.97 \pm 0.03$	-	$2.00 \pm 0.04$	$3.91 \pm 0.06$	$1.05 \pm 0.02$	$\text{La}_{0.97}\text{Bi}_2\text{O}_{3.93}\text{Cl}_{1.05}$
Ideal Sr-doped LBC	0.95	0.05	2	3.975	1	
synthesized Sr-doped LBC	$0.89 \pm 0.02$	$0.01 \pm 0.00$	$2.00 \pm 0.04$	$3.89 \pm 0.07$	$1.02 \pm 0.02$	$\text{La}_{0.89}\text{Sr}_{0.01}\text{Bi}_2\text{O}_{3.84}\text{Cl}_{1.02}$

The stoichiometric values are shown as 1 $\sigma$  ranges from the sample mean.



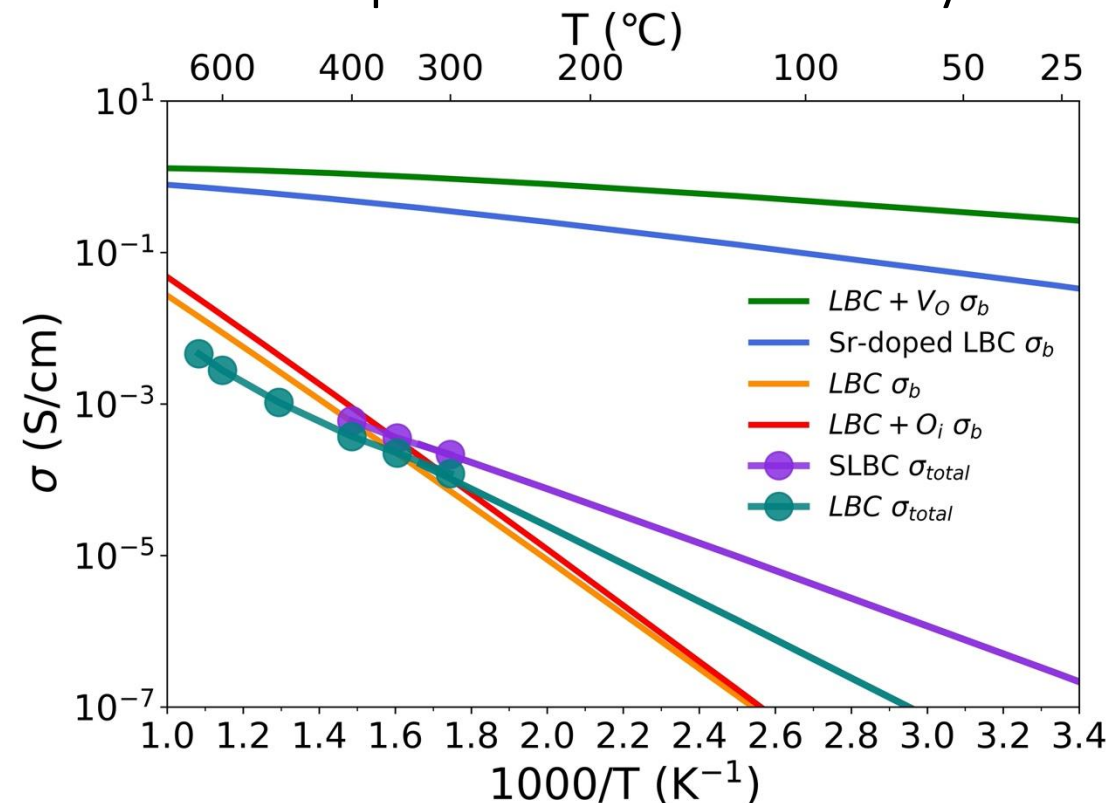
# Experimental conductivity for LBC and Sr-doped LBC

Comparison on experimental total conductivity



LBC and SLBC show comparable or higher total conductivity than YSZ and LSGM below 400 °C, with lower activation energies.

Comparison across experimental total conductivity and *ab initio*-predicted bulk conductivity



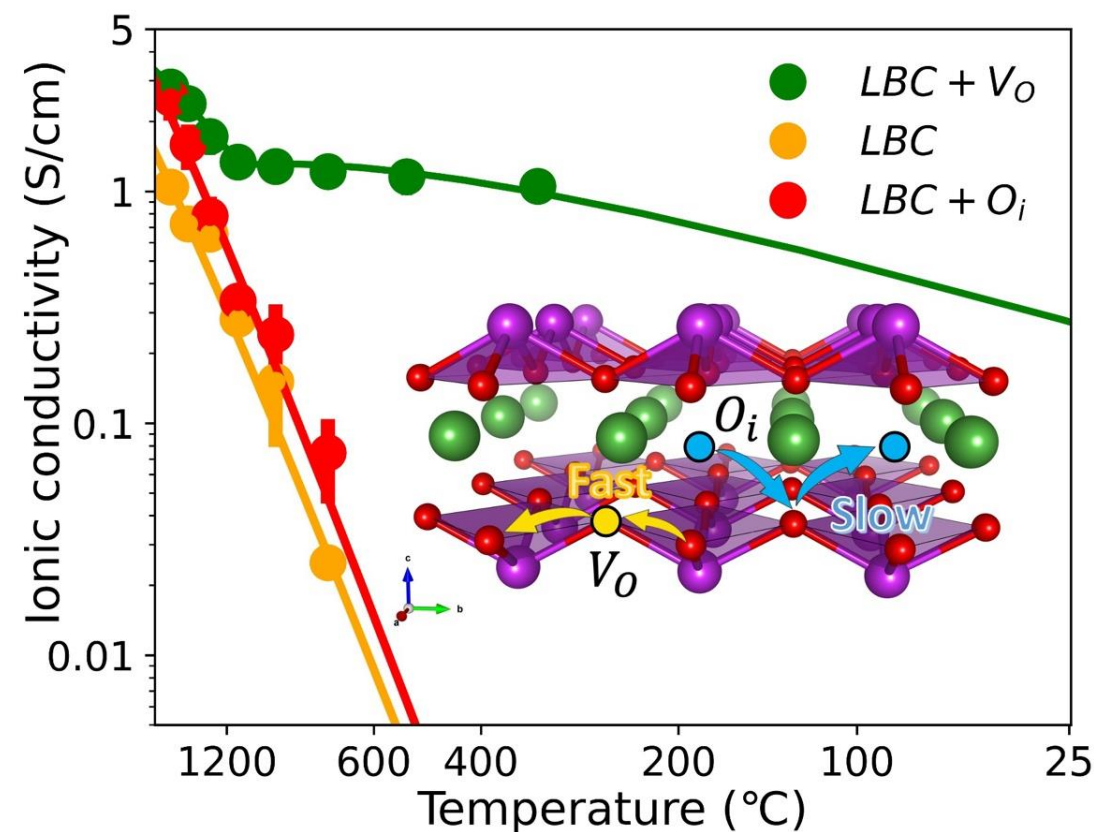
The predicted ultra-high conductivity in LBC+V<sub>O</sub> is not observed in our samples, which might be attributed to:

- The grain boundary effect
- The lack of sufficient oxygen vacancies.

[1] H. Yaguchi, *et al. Adv. Funct. Mater.* **2023**, 33, 2214082.

# Summary

- $\text{LaBi}_2\text{O}_4\text{Cl}$  (LBC) is identified as an ultra-fast oxygen conductor with ultra-low barrier of 0.1 eV, enabling ultra-high conductivity of 0.3 S/cm at 25 °C.
- LBC and Sr-doped LBC show superior conductivity and lower activation energy, although the predicted ultra-high conductivity is not fully realized yet.
- Experimental effort on aliovalent doping to create vacancies and microstructure refinement for LBC-based materials is needed to unlock its full potential of ultra-fast oxygen conduction approaching room temperature.



Scan to read the paper



## Computational Materials Group

### Faculty

Izabela Szlufarska      Dane Morgan

### Staff Scientists

Ajay Annamareddy      Maciej Polak  
Rafi Ullah      Ryan Jacobs

### Postdoc Researchers

Benjamin Afflerbach      Chen Shen  
Gaurav Arora      Jun Meng  
Muhammad Waqas Qureshi      Shuming Chen  
Siamak Attarian

### Graduate Students

Amy Kaczmarowski      Chiyoung Kim  
Lane Schultz      Ni Li  
Nuohao Liu      Sakiru Akinyemi  
Shuguang Wei      Sudipta Paul  
Xuanxin Hu      Younsoo Kim

### Undergraduate Students

Many students  
involved in the



Thank you!

