

Advanced Machine Learning Assignment: 3

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The entropy of a system whose states are \mathbf{x} , at temperature $T = 1/\beta$ is

$$S = \sum p(\mathbf{x}) \left[\ln \frac{1}{p(\mathbf{x})} \right]$$

where

$$p(\mathbf{x}) = \frac{1}{Z(\beta)} \exp[-\beta E(\mathbf{x})]$$

a

Show that

$$S = \ln Z(\beta) + \beta \bar{E}(\beta)$$

We start by filling in $p(\mathbf{x})$ into the formula for the entropy.

$$\sum p(\mathbf{x}) \left[\ln \left(\frac{1}{Z(\beta)} \exp[-\beta E(\mathbf{x})] \right)^{-1} \right]$$

We then work it out and simplify the terms:

$$\begin{aligned} &= \sum p(\mathbf{x}) \left[-\ln \left(\frac{1}{Z(\beta)} \exp[-\beta E(\mathbf{x})] \right) \right] \\ &= \sum p(\mathbf{x}) [-\ln \exp[-\beta E(\mathbf{x})] + \ln Z(\beta)] \\ &= \sum p(\mathbf{x}) [\beta E(\mathbf{x}) + \ln Z(\beta)] \\ &= \sum p(\mathbf{x}) [\beta E(\mathbf{x})] + \sum p(\mathbf{x}) [\ln Z(\beta)] \\ &= [\ln Z(\beta)] \sum p(\mathbf{x}) + \sum \frac{1}{Z(\beta)} \exp[-\beta E(\mathbf{x})] [\beta E(\mathbf{x})] \\ &= [\ln Z(\beta)] \sum p(\mathbf{x}) + \frac{\beta}{Z(\beta)} \sum \exp[-\beta E(\mathbf{x})] [E(\mathbf{x})] \end{aligned}$$

The function for \bar{E} as specified in (31.6) of the book:

$$\bar{E} = \frac{1}{Z} \sum_{\mathbf{x}} \exp(-\beta E(\mathbf{x}))$$

This means the function comes out to:

$$= [\ln Z(\beta)] \sum p(\mathbf{x}) + \beta \bar{E}(\beta)$$

The sum over p adds up to 1, this means the final function is:

$$S = \ln Z(\beta) + \beta \bar{E}(\beta)$$

This is the function we have to show, so it is proven.

b

Show that

$$S = -\frac{\delta F}{\delta T}$$

, where the free energy $F = -kT \ln Z$ and $kT = 1/\beta$

In this case of F we set $k = 1$. This means the function F comes out to:

$$F = -T \ln Z \text{ and } T = \frac{1}{\beta}$$

Then we take the derivative with the chain rule as Z includes T.

$$\frac{\delta F}{\delta T} = -\ln Z - T \frac{\delta \ln Z}{\delta T}$$

Expand the 2nd part of the derivative to include the derivative for beta

$$= -\ln Z - T \frac{\delta \ln Z}{\delta \beta} \frac{\delta \beta}{\delta T}$$

Replace T by $1/\beta$ and take the derivative. This comes out to $-\beta^2$

Substituting in the original formula gives:

$$= -\ln Z - \frac{1}{\beta} \left(\frac{\delta \ln Z}{\delta \beta} (-\beta^2) \right)$$

For $\frac{\delta \ln Z}{\delta \beta}$ fill in $-\bar{E}$ as described in (31.8) in the book. This gives:

$$= -\ln Z - \frac{1}{\beta} (-\beta^2) (-\bar{E})$$

$$= -\ln Z + (\beta) (-\bar{E})$$

This comes out to

$$-(\ln Z + \beta \bar{E})$$

which is indeed the negative of S. It is thus shown that this is the case.