

Chapter 2: Toxicology

Considers worker exposures



Definitions

Toxicology:	- entry of toxicants into organism - elimination from organism - effects on organism	Quantitative
Industrial hygiene:	prevention or reduction of entry	
Toxicant	- chemical agents - physical agents: particulates < 5 µm, noise, radiation	
Toxicity:	property related to effect on organism Problem: organisms respond via a distribution of effects	
Toxic hazard:	likelihood of damage based on exposure reduction by appropriate techniques	

Entry Routes for Toxicants

ROUTE	ENTRY	CONTROL
Ingestion	Mouth, stomach	Rules on eating, drinking, smoking
* Inhalation	Mouth, nose	Ventilation, hoods, protection equipment
Injection	Cuts in skin	Protective clothing
* Dermal Absorption	Skin	Protective clothing

* industrially most significant

Routes and elimination

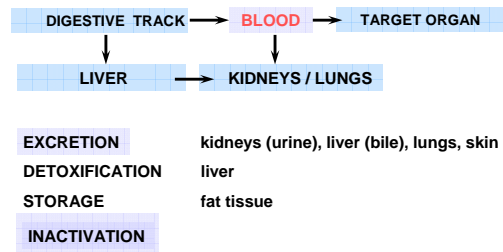
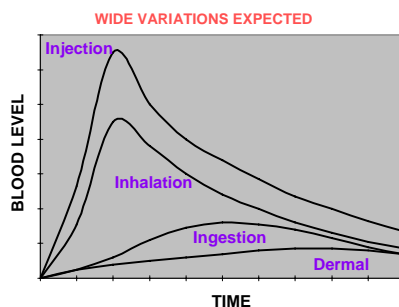


Figure 2-1

Toxic blood levels



Toxicology Experiment with Rabbits!



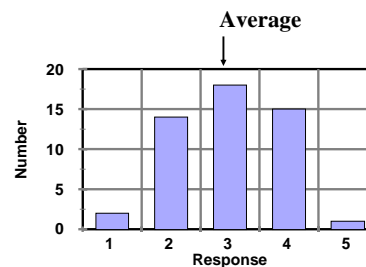
Start with 50 rabbits.
 Expose each to a fixed concentration.
 Get a variety of responses.

Determine Response Curve

Response	Number	Fraction
Least	1	2
	2	14
	3	18
	4	15
Worst	5	1
	50	1.00

$$\text{Average} = (1 \times 2 + 2 \times 14 + 3 \times 18 + 4 \times 15 + 5 \times 1) / 50 = 149 / 50 = 2.98$$

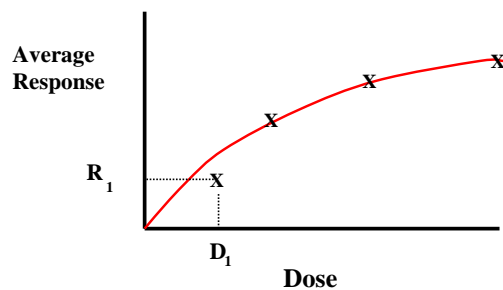
Plot Bar Chart



Repeat experiment at different doses.

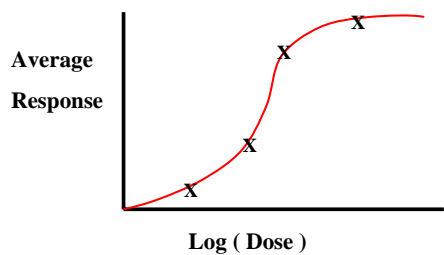
Dose	Average Response
D_1	$R_1 = 2.98$
D_2	R_2
D_3	R_3
D_4	R_4

Plot Response vs. Dose



This form not very useful, particularly at low doses.

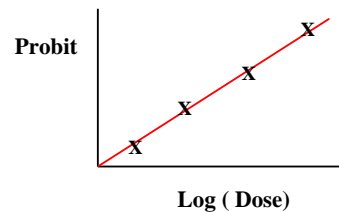
Take the log of the dose.



Get S-shaped curve - better at low dose values

Transform into Probit

Change S-shape into straight line using a mathematical transformation called a probit.
See Table 2-4 in text for numerical conversion.



Probit Equations

Using probits, most response vs. dose curves can be represented in the form:

$$Y = k_1 + k_2 \ln(V)$$

where Y = Probit variable

k_1, k_2 are constants

V = causative variable

See Table 2-5 for a list of probit equations for toxic exposures, fires and explosions.

Conversion from Probit to %

$$P = 50 \left[1 + \frac{Y-5}{|Y-5|} \operatorname{erf} \left(\frac{|Y-5|}{\sqrt{2}} \right) \right]$$

P = Percentage

Y = probit

erf = error function (available on spreadsheet)

This is very useful for spreadsheet calculations!

Threshold Limit Values

THRESHOLD DOSE: NO DETECTABLE EFFECT

Threshold Limit Value TLV: worker's lifetime
This is for 8 hours per day, 40 hours per week.
NOT CONTINUOUS EXPOSURE!

TLV - TWA *	Time weighed average
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TLV - STEL	Short term exposure limit
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TLV - C	Ceiling limit
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See Table 2-7 for detailed definitions of these.

See Table 2-8 for specific values for a number of chemicals. More values are available for TWAs than for STEL or C.

Threshold Limit Values - 2

Published by ACGIH: American Conference of Governmental Industrial Hygienists, a professional organization without legal authority.

Cannot be used as indication of relative toxicity.

Cannot be used for air pollution exposures.

➡ **Some toxicants have zero thresholds** ⬅

TLV – Example Values Table 2-8

Acetone	750 ppm
Ammonia	25 ppm
CO	25 ppm
Chlorine	0.5 ppm
Gasoline	300 ppm
Hexane	50 ppm
Phosgene	0.1 ppm

For flammables, TLV is ¼ of lower flammable limit.

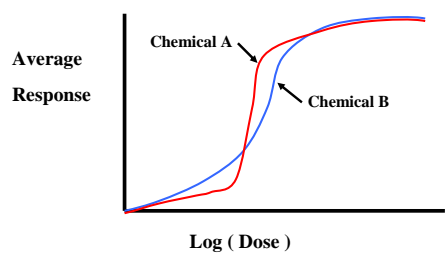
Conversion from mg/m³ to ppm

$$\text{ppm} = \left(\frac{22.4}{M} \right) \left(\frac{T}{273} \right) \left(\frac{1}{P} \right) (\text{mg/m}^3) = 0.08205 \left(\frac{T}{PM} \right) (\text{mg/m}^3)$$

Equation (2-7)

For liquid mixtures ppm = mg/m³, but this is not true for vapors!

Problem with Relative Toxicity



PEL - Permissible Exposure Level

Published by OSHA, and have legal authority.

Defined the same as TLV.

Most PELs are same as TLVs.

Not updated as regularly as TLVs.

Most companies use lowest of the two values.

For some chemicals, i.e. benzene, vinyl chloride, a specific OSHA regulation has been published. Each regulation is unique, but most require EXPLICIT data that workers are not exposed.

See OSHA.gov web site for regulations.