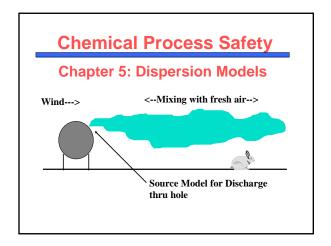
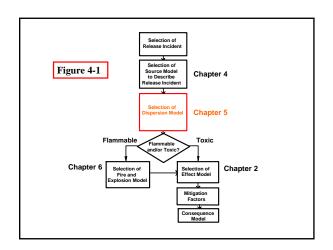
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### **Dispersion Models**

What? Describe how vapors are transported downwind of a release. Valid between 100 m to 10 km.

- Below 100 m use ventilation equations Chapt. 3.
- Above 10 km → almost unpredictable.

Why? To determine the consequences.

Results: Downwind concentrations (x,y,z)

Area affected

**Downwind evacuation distances** 

# Dispersion PLUME wind PUFF Continuous release DOWNWIND DILUTION BY MIXING WITH FRESH AIR ATMOSPHERIC DISPERSION Wind speed Atmospheric stability: vertical temp. profile Roughness ground: buildings, structures, trees, water Height of release above ground level Momentum and buoyancy: effective height

## Atmospheric stability MAINLY DETERMINED BY VERTICAL TEMPERATURE GRADIENT heat & radiation balance troposphere and surface convective air flows season time of day night heat & radiation balance troposphere and surface Figure 5-3 Figure 5-3 Adiabatic temperature (Centigrade) Adiabatic temperature gradient humid air: 0.5 °C / 100 m

## Unstable: Sun heats ground faster than heat can be removed so that air temperature near the ground is higher than the air temperature at higher elevations. Neutral: The air above the ground warms and the wind speed increases, reducing the effect of solar input. Stable: The sun cannot heat the ground as fast as the ground cools - temperature at ground is lower. STABILITY CLASSES A - F A Extremely unstable B Moderately unstable C Slightly unstable D Neutral E Slightly stable F Moderately stable

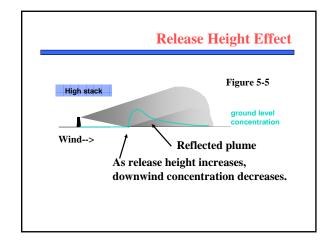


 Table 5-1
 Atmospheric Stability Classes for Use

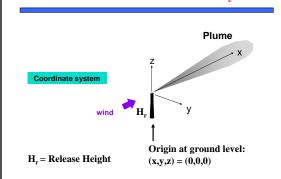
 with the Pasquill-Gifford Dispersion Model 1.2

				Nighttime c	onditions <sup>4</sup>
Surface wind speed		aytime insolation	n³	Thin overcast	≤3/8
(m/s)	Strong	Moderate	Slight	low cloud	cloudiness
<2	A	A-B	В	F5	F <sup>5</sup>
2-3	A-B	В	С	E	F
3-4	В	B-C	С	D6	E
4-6	С	C-D	D6	D6	D6
>6	С	D6	D6	D6	D6

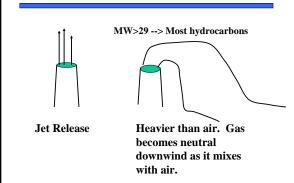
See qualifying notes below table!







### **Release Momentum and Buoyancy**



### **Dispersion Models**

Dispersion models are based on a mass balance.

Two approaches:

1. Use eddy diffusivities, K, to represent turbulence.

Advantage: nice tidy theoretical model.

Disadvantage: K = K(x,y,z), and impossible to measure.

See text.

2. Use dispersion coefficients which represent the standard deviations in the concentration profiles.

Advantage: easy to measure and correlate.

### Gaussian form of plume equation Equation 5-49

$$\langle C \rangle (x, y, z) = \frac{Q_m}{2\pi\sigma_y\sigma_z u} \exp\left(-\frac{y^2}{2\sigma_y^2}\right) \times \left\{ \exp\left[-\frac{(z - H_z)^2}{2\sigma_z^2}\right] + \exp\left[-\frac{(z + H_z)^2}{2\sigma_z^2}\right] \right\}$$

Top View of plume

--> Wind

 $\langle C \rangle(x, y, z) = \text{Ave. conc. } (20-30 \text{ min ave})$ 

 $Q_m$  = Release rate (mass/time)

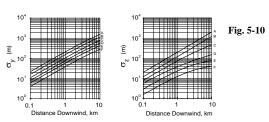
 $\sigma_y, \sigma_z$  = Dispersion coefficients = f(stability class, downwind distance)

u = Wind speed (length/time)

y, z =Coordinates (length)

 $H_r$  = Release height (length)

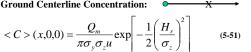
### Atmospheric stability



Dispersion coefficients for plume model for rural releases. See Figure 5-11 for urban releases, and Table 5-2 for plume equations. Figure 5-12 and Table 5-3 are for puff model.

### **Simplified Cases - Plume**

**Ground Centerline Concentration:** 



Ground, centerline, release height  $H_r = 0$ 

$$< C > (x,0,0) = \frac{Q_m}{\pi \sigma_v \sigma_z u}$$
 (5-48)

*X* is implicit in the dispersion coefficients!

### **Maximum Concentrations - Plume**

Always occurs at release point.

For releases above ground, max. concentration on ground occurs downwind.

$$\left(\sigma_z\right)_{x,\text{max}} = \frac{H_r}{\sqrt{2}}$$

$$< C>_{\text{max}} = \frac{2Q_m}{e\pi \iota H_r^2} \left(\frac{\sigma_z}{\sigma_y}\right)$$

- 1. Use left equation to determine  $\sigma_{z}$
- 2. Use Figures 5-10 or 5-11 to get x.
- 3. Determine  $\sigma_{_{\scriptscriptstyle V}}\,$  from Figures 5-10 or 5-11.
- 4. Calculate *<C>* from right equation.

### Equation 5-54

$$< C > (x, y, z, t) = \frac{Q_m^t}{\sqrt{2}\pi^{3/2}\sigma_x\sigma_y\sigma_z} \exp\left[\frac{1}{2}\left(\frac{y}{\sigma_y}\right)^2\right] \times \left\{\exp\left[-\frac{1}{2}\left(\frac{z - H_x}{\sigma_z}\right)^2\right] + \exp\left[-\frac{1}{2}\left(\frac{z + H_x}{\sigma_z}\right)^2\right]\right\}$$



u, t not explicit in equation

x is implicit thru dispersion coefficients

Coordinate system moves with puff center at x=ut

Assume  $\sigma_x = \sigma_y$ 

### **Simplified Cases - Puff**

Concentration on ground below puff center

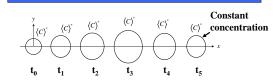
$$< C > (0,0,0) = \frac{Q_m^*}{\sqrt{2}\pi^{3/2}\sigma_x\sigma_y\sigma_z} \exp\left[-\frac{1}{2}\left(\frac{H_r}{\sigma_z}\right)^2\right]$$
 (5-56)

Same as above, with  $H_r = 0$ . Puff center on ground.

$$< C > (0,0,0) = \frac{Q_m^*}{\sqrt{2\pi^{3/2}\sigma_x\sigma_y\sigma_z}}$$
 (5-41)

Puff center always at release height.

### **Location of Puff**

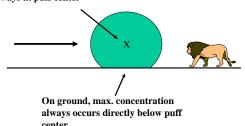


---> Wind = constant at u

Center of puff located at: x = u t

### **Maximum Concentration - Puff**

Always at puff center



### **Example:**

10 kg/s of H<sub>2</sub>S is released 100 m off of ground. Estimate the concentration 1 km downwind on ground? It is a clear, sunny day, 1 PM, wind speed = 3.5 m/s. Assume rural conditions.

Plume, due to continuous nature of release!

From Table 5-1, Stability Class B.

From Figure 5-10,  $\sigma_v = 130 \text{ m}$ 

From Figure 5-10,  $\sigma_z = 120 \text{ m}$ 

Use Equation 5-51 for a plume.

### **Example:** Apply Equation 5-51

Applies to ground concentration directly downwind of release:

$$\langle C \rangle(x,0,0) = \frac{Q_m}{\pi \sigma_y \sigma_z u} \times \exp \left[ -\frac{1}{2} \left( \frac{H_r}{\sigma_z} \right)^2 \right]$$

$$\langle C \rangle (x, 0, 0) = \frac{10.0 \text{ kg/s}}{(3.14)(130 \text{ m})(120 \text{ m})(3.5 \text{ m/s})} \times \exp \left[ -\frac{1}{2} \left( \frac{100 \text{ m}}{120 \text{ m}} \right)^2 \right]$$

 $\langle C \rangle (x,0,0) = 41.2 \times 10^{-6} \text{ kg/m}^3 = 41.2 \text{ mg/m}^3$ 

Use Equation 2-7 to get 29.7 ppm. TLV-TWA is 10 ppm.

### **Example:** Where is max. concentration?

Use Equation 5-53:

$$(\sigma_z)_{x,\text{max}} = \frac{H_r}{\sqrt{2}} = \frac{100 \text{ m}}{1.414} = 70.7 \text{ m}$$

Use equation in Table 5-3 to determine downwind distance:

$$\sigma_z = 0.12x$$

$$70.7 \text{ m} = 0.12x$$

$$x = 590 \text{ m}$$

At this location, from Figure 5-10:

$$\sigma_{\rm v} = 92 \text{ m}$$

Use Equation 5-52 to calculate max. concentration:

$$< C>_{\max} = \frac{2Q_m}{e\pi u H_r^2} \left(\frac{\sigma_z}{\sigma_y}\right) = \frac{(2)(100 \text{ kg/s})}{(2.718)(3.14)(3.5 \text{ m/s})(100 \text{ m})^2} \left(\frac{70.7 \text{ m}}{92 \text{ m}}\right)$$

$$< C>_{\text{max}} = 5.14 \times 10^{-4} \text{ kg/m}^3 = 514 \text{ mg/m}^3 = 370 \text{ ppm}$$

### **Example:** What is max. discharge to result in 10 ppm?

Maximum will occur at same location:  $(\sigma_z)_{x,\text{max}} = \frac{H_r}{\sqrt{2}}$ 

10 ppm = 13.9 mg/m<sup>3</sup> (Equation 2-7)

Substitute into Equation 5-52:

$$< C>_{\text{max}} = \frac{2Q_m}{e\pi u H_r^2} \left( \frac{\sigma_z}{\sigma_y} \right)$$

13.9 × 10<sup>-6</sup> kg/m<sup>3</sup> = 
$$\frac{2Q_m}{(2.71)(3.14)(3.5 \text{ m/s})(100 \text{ m})^2} \left(\frac{70.71 \text{ m}}{92 \text{ m}}\right)$$

 $Q_{m} = 2.7 \text{ kg/s}$  Not very much!

### **Example:**

 $10~\rm kg$  of  $\rm H_2S$  is released instantly on the ground. What is concentration at fenceline 100 m away? Same conditions as before.

From Table 5-1, stability class is B.

At x = 0.1 km, from Figure 5-12:  $\sigma_y = 10 \text{ m}$   $\sigma_z = 16 \text{ m}$ 

Use Equation 5-41 for a ground release, centerline conc.:

$$< C>(0,0,0) = \frac{Q_m^*}{\sqrt{2\pi^{3/2}\sigma_x\sigma_y\sigma_z}}$$

**Assume**  $\sigma_x = \sigma_y$ 

$$Q_{m}^{\circ} = 10 \text{ kg} = 10 \times 10^{6} \text{ mg}$$

$$<$$
C $> = 79.4 mg/m3 = 571 ppm$ 

### **Example:**

How long does it take for puff to reach fenceline?

$$x = ui$$

$$t = \frac{x}{u} = \frac{100 \text{ m}}{3.5 \text{ m/s}} = 28.6 \text{ s after release.}$$

Very little time for an emergency response!

### **Example:**

What size release will result in  $10\ ppm$  at fenceline?

Same procedure as for plume. Answer is 0.175 kg = 175 gm.

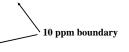
Not very much! Better to contain chemicals than to mitigate after a release!

## Isopleths

What: Lines of constant concentration

Plume:

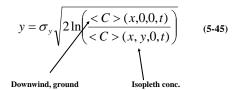
Puff:



### **Determining Isopleths: Plume and Puff**

Divide equation for centerline concentration by equation for ground level concentration. Solve for y, which is crosswind direction.

centerline conc.



### **Procedure to Determine Isopleths - 1**

1. Determine concentrations along centerline at fixed points downwind.

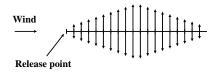
Wind

Release point

2. Use equation (5-45) to find y at each fixed point.

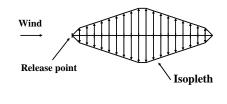
### **Procedure to Determine Isopleths - 2**

3. Plot +y and - y at each fixed point.



### **Procedure to Determine Isopleths - 3**

4. Connect the points.



### Worst Case on Ground- Plume and Puff

Want maximum <C> downwind

Plume: Centerline, ground,  $\mathbf{H_r} = \mathbf{0}$   $< C > (x,0,0) = \frac{Q_m}{\pi \sigma_v \sigma_z u}$ 

Puff: Center of puff on ground,  $H_r = 0$ 

$$< C > (x,0,0) = \frac{Q_m^*}{\sqrt{2}\pi^{3/2}\sigma_x\sigma_v\sigma_z u}$$

Specify:  $Q_m$  and  ${Q_m}^*$  at maximum

 $\sigma$ 's are minimum --> F stability

u is minimum within stability class --> 2 m/s

(EPA suggests 1.5 m/s!)

### **Dense Gas Dispersion**

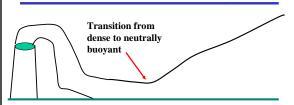


Britter-McQuaid Dense Gas Dispersion Model:

Based on dimensionless groups and available experimental data.

See pages 195-199.

### **Dense Gas Dispersion**



Results in shorter distances if only dense gas. Results in longer distances if dense gas coupled with Gaussian dispersion.

### **Toxic Effect Criteria**

What concentration should we use for emergency releases?

Cannot use PEL or TWA since these are for continuous work exposures – values are too low for short-term exposures.

<u>ERPGs</u> – Emergency Response Planning Guidelines, issued by American Industrial Hygiene Association

**EEGLs** – Emergency Response Guidance Levels, issued by National Academy of Sciences/National Research Council.

**LOC** – Level of Concern from EPA

Toxic Endpoints promulgated by EPA as part of RMP

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## **ERPG:** Emergency Response Planning Guideline

ERPG-1: max. airborne concentration below which it is believed nearly all individuals can be exposed for up to 1-hr without experiencing effects other than mild transient adverse health effects or perceiving a clearly defined objectionable odor.

ERPG-2: max. airborne conc. below which it is believed nearly all individuals can be exposed up to 1 hr without experiencing or developing irreversible or other serious health effects or symptoms that could impair their ability to take protective action.

## **ERPG:** Emergency Response Planning Guideline

ERPG-3: max. airborne concentration below which it is believed nearly all individuals can be exposed for up to 1-hour without experiencing or developing life-threatening health effects.

Table 5-6	ERPG-1	ERPG-2	ERPG-3
Ammonia	25 ppm	200 ppm	1000 ppm
Chlorine	1 ppm	3 ppm	20 ppm
Monomethylamine	10 ppm	100 ppm	500 ppm
Toluene	50 ppm	300 ppm	1000 ppm

### **EEGL: Emergency Exposure Guidance Levels**

A concentration of a gas, vapor, or aerosol that is judged acceptable and allows exposed individuals to perform specific tasks during emergency conditions lasting from 1 to 24 hours.

Table 5-7	1-hr EEGL	24-hr EEGL
Ammonia	100 ppm	
Chlorine	3 ppm	0.5 ppm
Toluene	200 ppm	100 nnm

## IDLH: Immediately Dangerous to Life and Health

A concentration that poses a threat of exposure to airborne contaminants when that exposure is likely to cause death or immediate or delayed permanent adverse health effects or prevent escape from such an environment.

Available from OSHA (osha.gov) or NIOSH Pocket Guide.

Ammonia 50 ppm Chlorine 10 ppm Toluene 500 ppm

### **LOC:** Level of Concern

The max. concentration of an extremely hazardous substance in air that will not cause serious irreversible health effects in the general population when exposed to the substance for relatively short duration.

See EPA for values.

### **Toxic Endpoints**

Must be used for dispersion modeling required for the EPA RMP. Values are, in order of preference, ERPG-2 or LOC.

Ammonia 0.14 mg/L
Chlorine 0.0087 mg/L
Chloroform 0.49 mg/L

Table 5-8

Table 5-9:	ORDER IS REVERSED	IN TABLE!
Primary	Secondary	Benzene
ERPG-3:	EEGL (30-min)	1000 ppm
	IDLH	500 ppm
ERPG-2:	EEGL (60 min)	150 ppm
	LOC	
	PEL-C	
	TLV-C	
	5 x TLV-TWA	50 ppm

RPG-1: PEL-STEL	
	50 ppm
TLV-STEL	5 ppm
3 x TLV-TWA	30 ppm