# Numerical ODE integrators in Stan/Torsten

Yi Zhang, Charles Margossian

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# Nonlinear ODEs without analytical solution

```
functions {
 real[] reaction(real t, real[] x, real[] p, real[] r, int[]
i){
   /* k1r = k -1 */
   real k1 = p[1];
   real k1r = p[2];
   real k2 = p[3];
   real S = x[1];
   real E = x[2];
   real ES = x[3];
   real P = x[4];
   real dSdt = k1r * ES - k1 * S * E;
    real dEdt = k1r * ES + k2 * ES - k1 * S * E:
    real dESdt = k1 * S * E - k1r * ES - k2 * ES;
    real dPdt = k2 * ES;
    return {dSdt, dEdt, dESdt, dPdt};
```

# Numerical integrators

- Runge-Kutta 4th/5th (rk45)
  - non-stiff equations
  - Most popular, try this if you don't know the nature of the ODE, or what you're doing, or both.
- Backward differentiation formula (bdf)
  - stiff equations
  - More expensive to use
- Adams-Moulton
  - non-stiff equations
  - higher-order of accuracy(do you really need it?)
  - scales better with number of steps

# Numerical integrators

| Integrators | Stan                | Torsten                            |
|-------------|---------------------|------------------------------------|
| rk45        | integrate_ode_rk45  | pmx_integrate_ode_rk45             |
| BDF         | integrate_ode_bdf   | pmx_integrate_ode_bdf              |
| Adams       | integrate_ode_adams | <pre>pmx_integrate_ode_adams</pre> |

- ▶ ODE\_RHS: ODE right-hand-side f in  $y' = f(y, t, \theta, x_r, x_i)$ .
- y0: initial condition at time t0.
- t0: initial time.
- ts: times at which we require a solution.
- theta: parameters to be passed to f.
- x\_r: real data to be passed to f.
- x\_i: integer data to be passed to f.



### Exercise

We consider the kinetics of an autocatalytic reaction [1]. The structure of the reactions is

$$A \xrightarrow{k_1} B$$

$$B + B \xrightarrow{k_2} C + B$$

$$B + C \xrightarrow{k_3} C + A,$$

where  $k_1$ ,  $k_2$ ,  $k_3$  are the rate constants and A, B and C are the chemical species involved. The corresponding ODEs are

$$x'_{1} = -k_{1}x_{1} + k_{3}x_{2}x_{3}$$

$$x'_{2} = k_{1}x_{1} - k_{2}y_{2}^{2} - k_{3}x_{2}x_{3}$$

$$x'_{3} = k_{2}y_{2}^{2}$$

Given  $k_1 = 0.04$ ,  $k_2 = 1.0e4$ ,  $k_3 = 3.0e7$ , we seek the initial condition for  $x_1(t = 0)$ .



### Exercise

- Write the functions block for f.
- ▶ What's the initial conditions for  $x_2$  and  $x_3$ ?
- Which numerical integrator to use? Why? Did you try other options?

### Exercise

#### How to build & run?

## Edit/Add cmdstan/make/local

#### Build in cmdstan

```
make ../example-models/examples/chemical_reactions/chem
```

#### Run

```
./chem sample adapt delta=0.95 random seed=1104508041 data \rightarrow file=chem.data.R init=chem.init.R
```

### Reference



H. H. Robertson.

Numerical analysis, an introduction, chapitre The solution of a set of reaction rate equations.

Academic Press, 1966.