1. Law of mass action

O The reaction is:

3A -> P That is A+A+A->P

The mass balance:

rate of change of concentration of A

= rate of addition - rate of consumption

known. I

use Law of Mass Action.

z k·a·a·a = ka³

$$\Rightarrow \frac{da}{dt} = 1 - ka^3$$
.

(2) The reaction is:

A+2b -> P

Since A starts out as half of B, and A+ZB->P, me have

 $a(t) = \frac{1}{2}b(t)$ at any time t. of B

> b(t)= Agg Zalt)

$$\frac{da}{dt} = 0 - ka^*b \cdot b = -kab^2 = -4ka^3$$

$$\frac{dN}{dt} = rN(1 - \frac{N}{K})$$
$$= rN - r\frac{N^{2}}{K}$$

When N is big,
$$-\frac{N^2}{K}$$
 dominates, $\frac{dN}{dt} < 0$.

Suppresses over growth.

Let
$$T = \frac{t}{r}$$
, $y = \frac{N}{K}$

measure time in the unit of -

measure papulation size in the unit of k.

$$\frac{dN}{dt} = \lim_{\Delta t \to 0} \frac{\Delta N}{\Delta t} = \lim_{\Delta t \to 0} \frac{K\Delta Y}{\int_{\Gamma} \Delta t} = Kr \lim_{\Delta t \to 0} \frac{\Delta Y}{\Delta t}$$

 $\Rightarrow R \frac{dy}{d\tau} = r (y(1-y))$ $\frac{dy}{dz} = y(1-y)$ (A new DE without any parameter! If we can some colve this DE, we can back out any N given any values of K, r. One equation rules all.

dimensionless dimensional