

Lecture 16: Integral balance laws

Logistics: - no HW

- try to post new HW

Last time: - Velocity gradients

$$\text{spatial: } \underline{\underline{\epsilon}} = \nabla_{\underline{x}} \underline{\underline{\sigma}} = \dot{\underline{\underline{F}}} \underline{\underline{F}}^{-1} \quad \left. \begin{array}{l} \\ \end{array} \right\} \nabla_{\underline{x}} \underline{V} = \nabla_{\underline{x}} \underline{\underline{\sigma}} \underline{\underline{F}}$$
$$\text{material: } \dot{\underline{\underline{F}}} = \nabla_{\underline{x}} \underline{V}$$

$$- \underline{\underline{\epsilon}} = \underline{\underline{\epsilon}}_d + \underline{\underline{\omega}}$$

$$\underline{\underline{\epsilon}}_d = \text{sym}(\underline{\underline{\epsilon}}) = \frac{1}{2} (\nabla \underline{\underline{\sigma}} + \nabla \underline{\underline{\sigma}}^T) \quad \text{rate of strain}$$

$$\underline{\underline{\omega}} = \text{skew}(\underline{\underline{\epsilon}}) = \frac{1}{2} (\nabla \underline{\underline{\sigma}} \underline{\underline{\epsilon}} + \nabla \underline{\underline{\epsilon}}^T) \quad \text{spin}$$

- Reynolds Transport. Thm

$$\frac{d}{dt} \int_{\Omega_c} \phi \, dV_{\underline{x}} = \int_{\Omega_c} \frac{\partial \phi}{\partial t} \, dV_{\underline{x}} + \oint_{\partial \Omega_c} \phi \underline{\underline{\sigma}} \cdot \underline{n} \, dV_{\underline{x}}$$

\Rightarrow we don't need to know ϕ

Today: - Balance laws

Discrete \rightarrow continuum

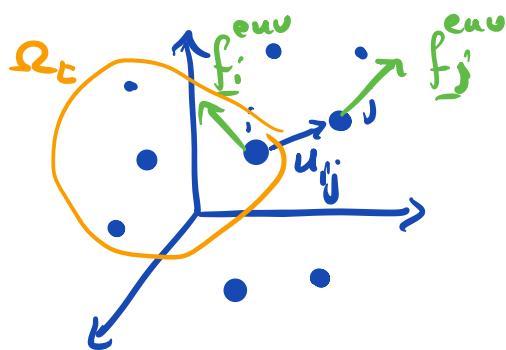
- Balance laws in integral form
- Continuum Thm 0

Discrete system

N particles

m_i = masses

\underline{x}_i = locations



$U_{ij} = U_{ji}$ intrachain energy

$f_{ij}^{\text{int}} = -\nabla_x U_{ij}$ intrachain force

f_i^{env} : environ. forces

Mass conservation: $\dot{m}_i = 0$

Newton's 2nd law: $m_i \ddot{\underline{x}}_i = \underline{f}_i^{\text{env}} + \sum_{j \neq i} \underline{f}_{ij}^{\text{int}}$

For any subset Ω_t of particles: $i \in I \subset \{1, \dots, N\}$

total mass: $M[\Omega_t] = \sum_{i \in I} m_i$

lin mom.: $L[\Omega_t] = \sum_{i \in I} m_i \underline{\dot{x}}_i$

ang. mom.: $J[\Omega_t] = \sum_{i \in I} \underline{x}_i \times m_i \dot{\underline{x}}_i$

internal energy: $U[\Omega_t] = \sum_{i \in I} U_{ij}$

$$\text{kinetic energy: } K[\Omega_t] = \sum_{i \in I} \frac{1}{2} m_i |\dot{\underline{x}}_i|^2$$

We have the following discrete balance laws:

$$\text{Mass is conserved: } \frac{d}{dt} M[\Omega_t] = 0$$

Change in lin. and ang. mom are equal to resultant ext. force and torque on Ω_t .

$$\frac{d}{dt} L[\Omega_t] = \sum_{i \in I} [f_i^{\text{ext}} + \sum_{j \notin I} f_{ij}^{\text{int}}]$$

$$\frac{d}{dt} j[\Omega_t] = \sum_{i \in I} \dot{x}_i \times [f_i^{\text{ext}} + \sum_{j \notin I} f_{ij}^{\text{int}}]$$

Change in internal and kinetic energy

is due to the power of external forces

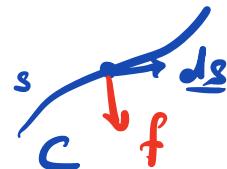
$$\frac{d}{dt} (U[\Omega_t] + K[\Omega_t]) = \sum_{i \in I} \dot{x}_i \cdot \underbrace{[f_i^{\text{ext}} + \sum_{j \notin I} f_{ij}^{\text{int}}]}_{\text{f}}$$

Reminder:

Work is energy transferred by application of a force along a distance.

$$W = F s$$

$$W = \int_C \underline{f} \cdot d\underline{s} = \int_{t_0}^{t_1} \underline{f} \cdot \underbrace{\frac{d\underline{s}}{dt}}_{\underline{v}} dt$$



$$= \int \underline{f} \cdot \underline{v} dt = \int \frac{dW}{dt} dt$$

$$\Rightarrow \frac{dW}{dt} = \underline{f} \cdot \underline{v}$$

Power :

$$P = \frac{dW}{dt} = \underline{v} \cdot \underline{f}$$

To generalize balance laws to continuum

$\Sigma \rightarrow \int$ for mass, lin & ang mom.

but continuum energy balance is more complicated.

continuum velocity is mean velocity

\Rightarrow loose information about velocity fluctuations

Introduce new variables:

Temperature :- measure of the magnitude
of the velocity fluctuations

Heat : measure of the energy in fluctuations

Balance laws in integral form

continuum def. mass, lin. & ang. mom.

$$\text{of } \Omega_t \in \mathcal{B}_t : M[\Omega_t] = \int_{\Omega_t} p(x, t) dV_x$$

$$L[\Omega_t] = \int_{\Omega_t} p(x, t) \underline{v}(x, t) dV_x$$

$$j[\Omega_t] = \int_{\Omega_t} (x - z) \times p(x, t) \underline{v}(x, t) dV_x$$

Conservation of mass:

In absence of relativistic effects or
radioactive decay the mass of body

~~object~~ does not change as it deforms:

$$\boxed{\frac{d}{dt} M[\Omega_t] = 0} \quad \text{for all } \Omega_t \subset B_t$$

Laws of inertia

In fixed reference frame

$$\frac{d}{dt} L[\Omega_t] = \int_{\Omega_t} p(x, t) b(x, t) dV_x + \int_{\partial\Omega} t(x, t) dA_x$$

$$\frac{d}{dt} j[\Omega_t] = \int_{\Omega_t} \dot{x} \times p(x, t) b(x, t) dV + \int_{\partial\Omega} \dot{x} \times t(x, t) dt,$$

Continuum Thermodynamics

Assume existence of abs. temperature field

$\Theta(x, t) \geq 0$ at all $x \in B_t$. It is a measure of the velocity fluctuations of atoms in vicinity of x .

Thermal energy or heat content is the energy associated with temp. / fluctuations.

Bodies can gain/lose heat in two ways:

I) Body heating: $Q_b[\Omega_t] = \int_{\Omega_t} p(x,t) r(x,t) dV_x$

II) Surface heating: $Q_s[\Omega_t] = - \int_{\partial\Omega_t} q(x,t) \cdot n dA_x$

where $r(x,t)$ is heat supply/loss per unit mass.
and $q(x,t)$ is the heat flux vector.

Net heating

$$\dot{Q}[\Omega_t] = \dot{Q}_b[\Omega_t] + \dot{Q}_s[\Omega_t] = \int_{\Omega_t} p r dV_x - \int_{\partial\Omega_t} q \cdot n dA_x$$

Kinetic Energy: $K[\Omega_t] = \int_{\Omega_t} \frac{1}{2} \rho |\underline{v}|^2 dV_x$

Power of external forces:

$$P[\Omega_t] = \int_{\Omega} \rho \underline{b} \cdot \underline{v} dV_x + \int_{\partial\Omega} \underline{t} \cdot \underline{v} dA_x$$

Net working $W[\Omega_t]$ of external forces

on Ω_t is the mech. power that is not

converted into motion.

$$\dot{W}[\Omega_t] = P[\Omega_t] - \frac{d}{dt} K[\Omega_t]$$

$\dot{W}[\Omega_t] > 0$: mechanical energy is stored

$\dot{W}[\Omega_t] < 0$: mechanical energy is released

Internal energy & the first law

Energy content of body not assoc. with kinetic energy is called the internal energy.

Here we assume thermal, mechanical(elastic) internal energy is

$$U[\Omega_t] = \int_{\Omega_t} p \phi \, dV_x$$

ϕ is internal energy density per unit mass

First law of Thermo

$$\frac{d}{dt} U[\Omega_t] = Q[\Omega_t] + W[\Omega_t]$$

or

$$\frac{d}{dt}(U[\Omega_t] + K[\Omega_t]) = Q[\Omega_t] + P[\Omega_t]$$

Note: Discrete energy balance does not have the net heating term

In some cases, the power of an external force can be written as $P[\Omega_t] = -\frac{d}{dt}G[\Omega_t]$ where $G[\Omega_t]$ is the potential energy for the external force

$$\frac{d}{dt}(U[\Omega_t] + K[\Omega_t] + G[\Omega_t]) = Q[\Omega_t] + P$$

↑ ↑ ↑
 internal kinetic potential

Entropy and 2nd Law

The second law expresses the fact that a body has ~~no~~ a limit on the rate of heat uptake but no limit on rate of heat release.

$$Q[\Omega_t] \leq \Sigma[\Omega_t]$$

where $\mathbb{E}[\Omega_t]$ is upper bounded on the net heating.

In absence of net working $W[\Omega_t] = 0$

$$\frac{dU}{dt}[\Omega_t] = Q[\Omega_t] \leq \mathbb{E}[\Omega_t]$$

rate of energy storage is limited

Entropy of a body is defined

$$\boxed{\frac{d}{dt} S[\Omega_t] = \frac{\mathbb{E}[\Omega_t]}{\Theta[\Omega_t]}} \quad \Theta \text{ is mean temp.}$$

entropy is the quantity whose rate of change is the upper bounded per unit temperature.

In terms of net heating

$$\boxed{\frac{d}{dt} S[\Omega_t] \geq \frac{Q[\Omega_t]}{\Theta[\Omega_t]}}$$

Clausius-Planck
Inequality

In thermo books: $dS = \frac{dQ_{\text{rev}}}{T} \quad dS \geq \frac{dQ}{T}$

For a non-hom. body we introduce $s(x, t)$
entropy density per unit mass

$$S[\Omega_t] = \int_{\Omega_t} \rho s \, dV_x$$

Generalization of 2nd law to inhom. sys.

$$\frac{d}{dt} \int_{\Omega_t} \rho s \, dV_x \geq \int_{\Omega_t} \frac{\rho \Gamma}{\theta} \, dV_x - \int_{\partial \Omega} \frac{q \cdot n}{\theta} \, dA_x$$

Clausius-Duhem inequality

\Rightarrow places restrictions on constitutive relations.

Note: Precise form of 2nd law in continuum mechanics is not settled yet!

But all proposed forms lead to same restrictions!