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Biological detection by optical oxygen sensing

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Recent developments in the area of biological detection by optical sensing of molecular oxygen (O_2) are reviewed, with particular emphasis on the quenched-phosphorescence O_2 sensing technique. Following a brief introduction to the main principles, materials and formats of sensor technology, the main groups of applications targeted to biological detection using an O_2 transducer are described. These groups include: enzymatic assays; analysis of respiration of mammalian and microbial cells, small organisms and plants; food and microbial safety; monitoring of oxygenation in cell cultures, 3D models of live tissue, bioreactors and fluidic chips; *ex vivo* and *in vivo* O_2 measurements; trace O_2 analysis. For these systems, which enable a range of new bioanalytical tasks with different samples and models in a minimally invasive, contact-less manner, with high sensitivity, flexibility and imaging capabilities in 2D and 3D, relevant practical examples are presented and their merits and limitations discussed. An outlook of future scientific and technological developments in the field is also provided.

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1. Introduction

Molecular oxygen (O_2) is of paramount importance for living organisms and biological systems. For anaerobic organisms O_2

is generally toxic and must be excluded or controlled accurately at low levels.^{1,2} Obligate anaerobes die in the presence of O_2 or create microoxic zones, whereas facultative anaerobes can use both anaerobic respiration (fermentation) and aerobic respiration for growth.¹ In photosynthetic organisms O_2 fluxes reflect their efficiency and metabolic state (light-dependent and dark processes).^{3,4} Aerobic cells and organisms require a constant

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supply of O₂ to generate energy in the form of ATP, perform the necessary physiological functions and drive numerous enzymatic reactions which convert metabolic substrates and produce vital products.⁵ O₂ levels in mammalian tissue are normally tightly regulated and maintained within narrow physiological limits.^{6–8} Higher eukaryotic organisms including mammals and humans have developed special systems to store, transport and regulate O₂ in tissues by means of haemoglobins, myoglobins, red blood cells and vasculature. A decreased O₂ or imbalance between O₂ supply and demand may lead to pathological states and ultimately death of the cell and whole organism.⁷ On the other hand, aerobic cells also possess oxygen-independent energy production pathways (*e.g.* glycolysis, Krebs cycle, glutaminolysis) which allow them to survive stress conditions by compensatory regulation and adaptation of their energy requirements and metabolism to changing environments. Living organisms are also able to 'sense' low and high O₂ (hypoxia and hyperoxia causing energy and oxidative stress, respectively) and respond in an adaptive manner and thus survive. However maladaptive responses to O₂ are also common, *e.g.* the Warburg effect in cancer cells, tissue remodelling. Recent research also demonstrates the pivotal role of O₂ regulation in plants,⁹ microbes and the complex interspecies communities such as deep sea vents, soil or gut biota.¹⁰

O₂ can therefore be viewed as an informative marker of the presence, viability, metabolic status and physiological behaviour of living systems that consume, release or depend on O₂ levels in the environment. The generic nature of this marker can be used to probe many functional characteristics of biological specimens, their behaviour under various conditions (both internal and external), metabolic parameters and responses to stimuli, drug treatment or transformation. Direct and indirect involvement in the many functions of cells, tissues and whole organisms, make O₂ one of the most important analytes, particularly in general cell biology, tissue and animal physiology, detection of aerobic cells, biomedical and clinical diagnostics, microbial safety, food science, environmental monitoring, and many other areas of science and human practice.

As an analyte, O₂ is a small, non-polar, gaseous, paramagnetic molecule, which has moderate solubility in aqueous solutions. Under air-saturated conditions (20.86% O₂, normal atmospheric pressure) dissolved O₂ concentration is 219 μM at 35 °C, and is strongly influenced by temperature and salinity (*e.g.* 166 μM at 5% salinity, 35 °C).^{1,11} Unlike other biological analytes such as ions, pH, metabolites, cellular markers, O₂ is not confined in a specific compartment of the cell or the biological sample, it diffuses fast across cell membranes, tissues, solution phase and even solid matter. Many polymers have moderate and high permeability and solubility for O₂, and this must be taken into account, especially for common plasticware made of polystyrene, polycarbonate and silicone.¹¹ Diffusion coefficient of O₂ in aqueous media at 25 °C is $2 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$, and O₂ capacity of air is approximately 50 times higher than for water.¹² As a consequence, altered mass exchange or contact with the gaseous phase (ambient air) may have a profound impact on O₂ measurements and skew the results. It is also impossible to 'freeze'

homeostatic O₂ levels within the sample and analyse them afterward (*post factum*). Therefore, the sensor, the detection system and the biological sample need to be carefully tuned to each other, so that they operate in an analytically plausible manner and under biologically relevant conditions.

Realisation of the variety of analytical tasks with biological samples of different types and characteristics requires a spectrum of dedicated sensor materials, analytical methodologies, measurement instrumentation and accessory tools. A large number of different platforms have been developed for O₂ detection and quantification in biological specimens, among which optical and particularly quenched-luminescence sensing techniques represent an important and very versatile group.^{13–15}

The scope of this review is rather broad as it spans across various roles of O₂ in biological systems, analytical applications of this biomarker, new biological knowledge revealed by its use, and advancement of the optical O₂ sensing technology and materials. It is difficult to cover all the aspects in equal detail, so we had to focus on the most important developments directly related to biological detection, and perhaps leave aside some excellent work on sensor materials, analytical methodologies and detection platforms, if at this stage they do not demonstrate novel biological data or clear practical advances in biological detection. The number of studies in the area, the groups producing these results and chemical, life science and medical journals publishing them are growing exponentially.

In this review, we mainly analyse the new *applications* of quenched phosphorescence O₂ sensing in the area of biological detection developed over the last 5–7 years, examining their main findings, merits, limitations and *status quo*, and trying to identify the directions of future research. The details of sensor materials and detection systems used in these applications will be discussed only briefly, these aspects are well covered in recent reviews.^{13,16–21} The authors would also like to apologise for possible oversight or exclusion from this collection of some studies that other researchers may consider relevant to the subject.

2. O₂ detection techniques

Traditional O₂ quantification relies on sampling and analysis by chemical (Winkler titration²²), physical (cartesian diver,²³ manometric²⁴) or instrumental (gas chromatography²⁵) methods. However, high demand for continuous monitoring, minimally or non-invasive measurements, *in situ* and *in vivo* analyses and imaging of O₂ in complex biological samples have led to the development of a spectrum of new detection platforms. These include the relatively simple and low cost electrochemical sensors (Clark-type O₂ electrode²⁶) actively used nowadays, optochemical sensors and biosensors, which have already become indispensable in many laboratory and industrial applications,¹³ and more sophisticated instrumental techniques such as Electron Paramagnetic Resonance,²⁷ functional Magnetic Resonance Imaging,²⁸ Positron Emission Tomography,²⁹ pulse oximetry,³⁰ and intrinsic optical signal imaging.³¹ In this family, optical techniques and particularly phosphorescence,^{32,33} fluorescence,³⁴

Table 1 Common optical techniques used for O₂ detection in biological samples

Method ^{ref.}	Detection principle	Probe type	Advantages	Limitations
Luciferase reporter system ³⁵	O ₂ -dependent expression and bioluminescence of luciferase enzyme	Endogenous luciferase and substrate	Sensitive, easy to measure	Not quantitative, requires cell transformation (reporter gene) and substrate
Phosphorescence quenching ³³	Quenching of phosphorescence of exogenous probe introduced in the sample	Synthetic phosphorescent materials	Direct, accurate, quantitative method. Stable calibration. Various read-out parameters (intensity, lifetime, ratiometric intensity). Many different formats and applications including imaging	Exogenous substance in the sample. Possible toxicity, altered function, O ₂ photoconsumption
Colorimetric indicators ^{326–328}	Changes in absorption upon O ₂ binding or red-ox reactions	Exogenous dyes	Possibility of visual detection, simple chemistry	Slow response, interferences, low resolution, instability. Not fully reversible
Hypoxia stains misonidazole and pimonidazole ^{38,39}	Administration of a redox dye, visualisation of hypoxic areas (dye accumulation) by immunohistochemistry	Bioreducing dye	Staining of hypoxic regions tissue. Analysis of tissue samples <i>post mortem</i>	Indirect, end-point method. Does not allow for quantification or real-time monitoring of O ₂
Hypoxia staining with fluorescent protein expressed under HIF promoter ³²⁹	Fluorescent protein is expressed in cells upon activation of HIF signalling	Red fluorescent protein or its modifications	Direct and real-time staining of hypoxic regions. No need in antibodies	Viral transfection is preferable. Dependence on activity of HIF system within the cell (cell-specific). Semi-quantitative
Hypoxia staining with fluorescent probes ^{254,330–332}	Administration of redox dye, visualisation by fluorescence	Redox dye, convertible by nitroreductase or related endogenous enzymes	Direct and real-time staining of hypoxic regions. No need in antibodies	Semi-quantitative. Dependence on the endogenous enzyme activity
Measurement <i>via</i> fluorescence of GFP constructs ^{34,40,333}	Detection of reversible (“photoactivated”) red-shifted O ₂ dependent fluorescence	Green fluorescent protein or its modifications	Intracellular, <i>in vivo</i> imaging. Targeted delivery to cellular compartments	Require genetic manipulation or cell transfection. Low sensitivity to O ₂ (detectable at <2% O ₂), semi-quantitative. Dependence of probe maturation on O ₂
Delayed fluorescence of PPIX ³⁷	Dynamic quenching of PPIX delayed fluorescence by O ₂	Overproduced endogenous PPIX upon administration of 5-aminolevulinic acid	Minimally invasive. Direct measurement of mitochondrial O ₂ . No toxic effects reported	Weak signals, low S/N requires cell stimulation. Low photostability of PPIX. Applicable only to eukaryotic cells and tissues
Photoacoustic spectroscopy ³⁶	Transient absorption of excited triplet states of an O ₂ sensitive dye	Exogenous dyes with long-lived triplet states (<i>e.g.</i> metalloporphyrin or methylene blue)	Deep tissue penetration. 3D imaging	Two NIR lasers (exciting and probing). Complex and expensive set up. Substantial doses of the probe

bioluminescence,³⁵ and photoacoustics³⁶ based O₂ detection represent a large and important group.

The intrinsic properties of the O₂ molecule make its direct detection rather difficult. Therefore, special ‘indicator substances’ are employed for selective and sensitive detection of O₂ in complex samples, which produce a characteristic and easily detectable signal without the need for tedious sampling and processing steps. Such indicators can be *endogenous* chemical or biological molecules present in the sample, for example haemoglobin in red blood cells³⁰ or protoporphyrin IX (PPIX³⁷). Genetically encoded fluorescent protein constructs³⁴ can be introduced into cells by DNA or viral transfection or cell transformation. Otherwise, exogenous, synthetic indicators, represented by O₂-sensitive dyes, conjugates, supramolecular structures and composite materials (coatings, particles, probes) on their basis, can be introduced in the test sample to achieve O₂ detection.¹⁸ The indicator produces a characteristic response as a result of a physico-chemical (binding, reduction–oxidation), biological (accumulation of glucose analogues or bioreduction of redox dye in hypoxic cells³⁸) or photophysical (phosphorescence quenching,³³ acoustic,³⁶ delayed fluorescence³⁷) process that is linked to O₂ concentration or to the related parameter of the sample.

Table 1 gives an overview of different optical methods used for O₂ detection, highlighting their attractive features and limitations.

Unfortunately, many of these systems for O₂ detection appear to be indirect and/or mediated by the complex (bio)-chemical reactions or processes which are slow, not fully reversible, non-stoichiometric or influenced by many factors (Table 1). Some of these methods are tailored to specific applications with predefined analytical specifications (medical or diagnostic), which require special samples, measurement conditions and instrumentation, or provide a semi-quantitative or end-point O₂ readout.^{29,30,38,39} Endogenous O₂ indicators are also in small numbers^{37,40} and have limited use. In this regard, O₂ sensors and probes based on quenching of exogenous photoluminescent dyes possess many attractive features and advantages over the other platforms for O₂ detection.¹⁸

3. O₂ sensing by phosphorescence quenching

Since its discovery,⁴¹ quenched-phosphorescence O₂ sensing has evolved as a powerful analytical method with long-ranging applications. First O₂ sensing systems introduced in 1960s³² and 1980s³³ targeted the very demanding biomedical applications.

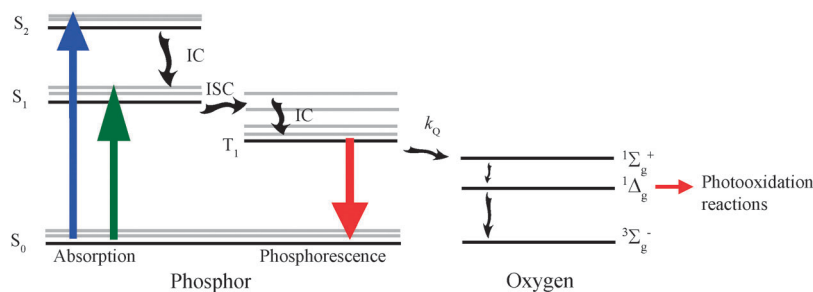


Fig. 1 Simplified Jablonski diagram showing the main energy transitions during the process of phosphorescence quenching by O₂. S₀, S₁, S₂ – ground state, first and second excited singlet states of the indicator, and T₁ – its excited triplet state. IC – internal conversion; ISC – intersystem crossing.

One was monitoring of blood oxygenation during intensive care by fibre-optic microsensors with solid-state O₂-sensitive coatings and fluorescence intensity measurements.^{42,43} Another was the phosphorescent lifetime based analysis of tissue oxygenation using a soluble probe, tailored for the *in vivo* O₂ imaging format.^{33,44}

Although providing a step change in O₂ determination, the uptake of these technologies was rather slow. Partly because first sensing materials and detection approaches were far from being optimal, measurement instrumentation was not readily available, and applications themselves required special technical expertise and detailed validation to demonstrate their practical benefits. During the 1990s and 2000s, optical O₂ sensor technology underwent major development, which revolved around its main building blocks: sensor materials and indicator dyes, detection instrumentation and measurement schemes, and analytical methodologies. Quenched-phosphorescence O₂ sensing has largely benefited (and continues to do so) from the technological advancements in allied disciplines, particularly synthetic, polymer and supramolecular chemistry,^{16,17,20,21} semiconductor optoelectronics, fluorescence spectroscopy,⁴⁵ chemical biology⁴⁶ and nanotechnology,^{47,48} optical and bio-imaging.^{49–51} This cross-fertilisation produced a range of advanced sensor chemistries, materials, measurement approaches and detection platforms.¹⁶ Intensive R&D has addressed the main bottlenecks of the O₂ sensor technology and produced new tools and systems with enhanced capabilities, some of which have become commercial products.^{13,18,20,52–54} Development of new applications and their dissemination are now keys for harvesting the full potential of optical O₂ sensing.

3.1. Detection principles

Quenched-phosphorescence O₂ detection relies on direct, non-chemical, reversible sensing of O₂ via a photochemical process of collisional quenching of excited state indicator dye molecules by molecular oxygen (Fig. 1). Upon light excitation, the luminophore absorbs photons and, through the processes of internal conversion and intersystem crossing, quickly relaxes to the excited triplet state. The emission from the triplet state is generally slow (a forbidden process which requires a change of spin). As a result, some excited dye molecules undergo quenching through collisional interaction with O₂ which reduces the yield and lifetime of the phosphorescence in a

concentration-dependent manner. It is considered that the paramagnetic ground state O₂ molecule (³Σ_g[−], open-shell triplet state) accepts energy from the luminophore's excited triplet state, recovering the ground state luminophore and generating a very short lived state (³Σ_g⁺) and subsequently singlet oxygen (¹Δ_g).¹⁶ Singlet oxygen, the main product of the quenching process, has relatively short lifespan in condensed media and high reactivity. It quickly deactivates back to the ground state O₂ (through interaction with solvent molecules, but also showing weak phosphorescence at 1270 nm⁵⁵), or reacts with neighbouring chemical structures causing their oxidation.⁵⁶

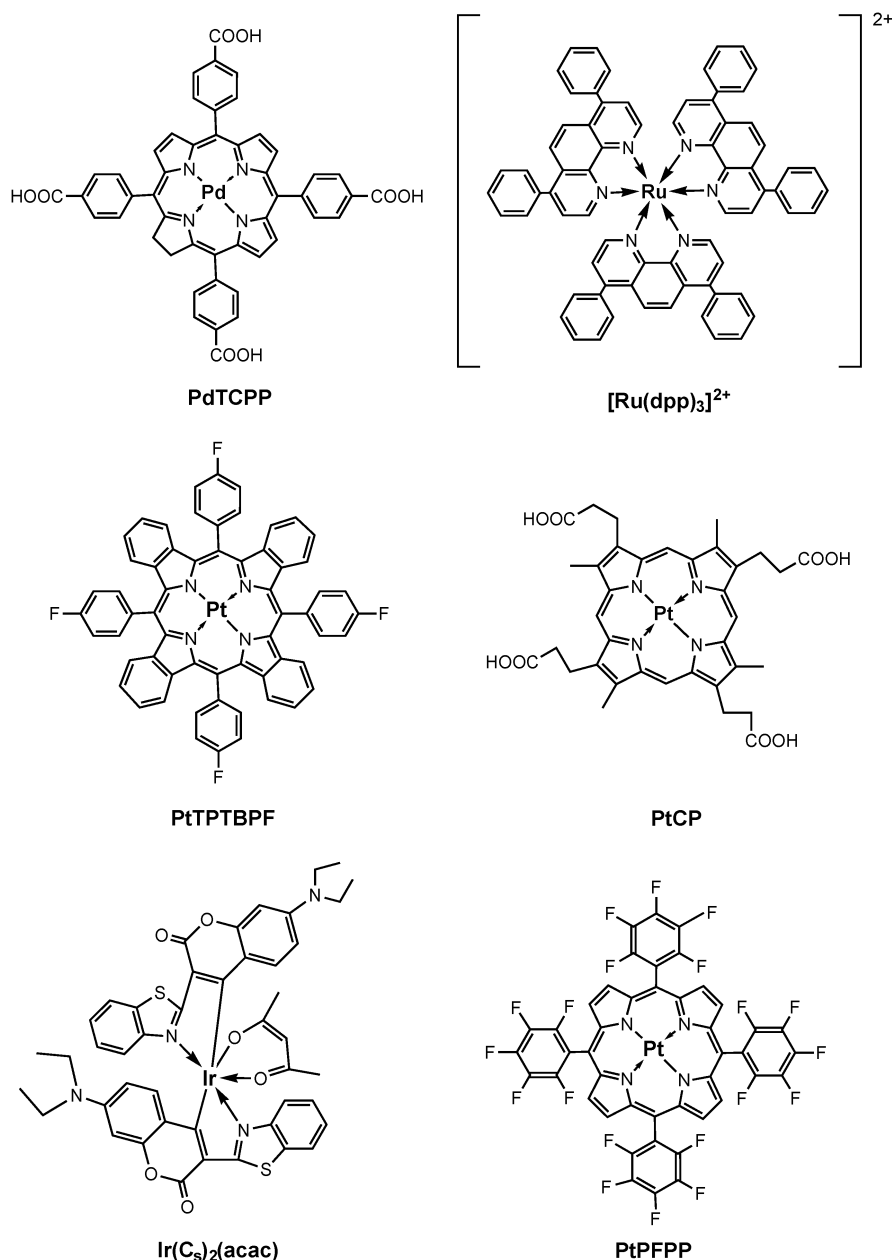
Photoluminescent O₂-sensitive materials are usually based on synthetic indicator dyes (macrocyclic complexes of heavy metal ions) with long-decay emissions (phosphorescence or delayed fluorescence) and lifetimes in the microsecond range. In recent years, many good reviews were published,^{16–19,21,57} which describe over a hundred of different indicator structures and their photophysical and O₂ sensing characteristics. Considering that such indicators can be used in different types of sensor materials, this gives us a pool of hundreds of different O₂-sensors. However, when it comes to specific applications, many of these indicator dyes and sensor materials become redundant or obsolete. In fact, the field of O₂ sensing is currently dominated by a rather small group of dyes, which have spectral and O₂ sensing characteristics, availability, costs, analytical performance, biocompatibility and the level of practical validation far superior to the other members. Most popular phosphorescent dyes are shown in Table 2 and Fig. 2.

For many research and laboratory applications, O₂ indicators emitting in the visible range (550–700 nm) are very appropriate, especially for screening of biological samples of similar type such as cell cultures. Thus, Ru(dpp)₃²⁺, Ir(C_s)₂(acac), PtCP and PtPFPP dyes are compatible with standard fluorescence spectrometers and readers and can be measured even in simple intensity mode to trace relative changes in concentration or O₂ consumption rate. Some of these dyes also allow lifetime-based O₂ detection on such instruments, which makes them more efficient, robust and quantitative and therefore preferred. For the dyes used on a disposable basis on such instruments, photostability is not so critical, however high brightness ($\epsilon\Phi$ product, Table 2) is always an advantage. Spectral sensitivity of photodetectors should be considered as it usually declines sharply at long wavelengths.

Table 2 Popular O₂-sensitive photoluminescent dyes and their characteristics (taken from ref. 16, 61 and 334)

Indicator dye	$\lambda_{\text{max}}^{\text{exc}}$ (nm)	$\lambda_{\text{max}}^{\text{em}}$ (nm)	τ_0 (μs)	Brightness, ($\epsilon\Phi$) ($\text{M}^{-1} \text{cm}^{-1}$)	Photostability
PdTCPP	415, 524	690	640	1900	Moderate
PdTCPTBP	442, 632	790	251	6000	Moderate
PtPFPP	390, 504, 538	647, 710	60	28 424	High
PtTPTBPF	430, 615	773	50	87 600	High
[Ru(dpp) ₃] ²⁺ Cl ₂	463	618	6.4	10 467	High
PtCP	380, 535	650	67	56 000	Modest
Ir(C ₅) ₂ (acac)	472, 444	563	11.3	50 112	Modest

Abbreviations: PdTCPP – Pd-*meso*-tetra-(4-carboxyphenyl)-porphyrin dendrimer; PdTCPTBP – Pd-*meso*-tetra-(4-carboxyphenyl)tetrabenzoporphyrin dendrimer; PtPFPP – Pt(II)-tetrakis(pentafluorophenyl)porphine; PtTPTBPF – Pt-*meso*-tetra-(4-fluorophenyl)tetrabenzoporphyrin; PtCP – Pt-coproporphyrin; Ru(dpp)₃ – tris(4,7-diphenyl-1,10-phenanthroline)ruthenium(II) chloride; ϵ – molar absorptivity, Φ – phosphorescence quantum yield. τ_0 – unquenched phosphorescence lifetime.

**Fig. 2** Chemical structures of some O₂-sensitive indicator dyes, described in Table 2.

For measurements in animal tissue and complex specimens (coloured, highly scattering and fluorescent samples), the choice is driven towards the indicators that are excitable and emitting in the red and very-near infrared spectral region (600–900 nm), such as benzoporphyrins. In high-resolution microscopy imaging, high photostability and brightness are the main criteria for indicator selection, which point towards PtPFPP and PtTPTBPF dyes¹⁶ or systems with light-harvesting antennae.^{58–60} Shorter emission lifetimes are also preferred for imaging, providing faster acquisition times and better temporal resolution in time-lapse experiments. In this regard, Pt-porphyrins are better than Pd-complexes which have 3–10 times longer τ_0 (and K_{s-v}) values.¹⁶ Ru-dyes are also attractive, but they often show low sensitivity to O₂, cyto- and phototoxicity⁵⁷ due to their cationic and non-biogenic nature, making them less preferred than porphyrins.

The alternative O₂-sensitive dyes have less favourable photo-physical properties (highlighted in Table 2) or biocompatibility, making them hard to compete. Nonetheless, the relatively small group of high-performance indicators provides the basis for a broad range of O₂-sensitive materials, the main groups of which are outlined in Table 3. Notably, Tables 2 and 3 are non-exhaustive, meaning that new 'blockbuster' indicators and sensor materials may appear in the future.

Photophysical and O₂ sensing properties of an indicator dye are dependent on its micro-environment (*e.g.* polymer matrix medium or sensor material), macro-environment and properties of the sample (temperature, presence of other quenchers, light-absorbing components, and sometimes pH, ionic strength, binding agents^{61,62}). This knowledge can be used to perform rational modification of the indicator, encapsulation/quenching medium or the method of incorporation in a sample and tune the properties of an O₂-sensitive material.^{16,63} For example, sensitivity to O₂ and the measurement range are determined by the Stern–Volmer quenching constant which, in turn, is determined by the unquenched phosphorescence lifetime of the indicator and characteristics of encapsulation media. For example, Pt(II)-porphyrins (PtPFPP and PtCP) are more sensitive to O₂ than corresponding Pd(II) complexes.¹⁸ The addition of a dendrimeric shell to the phosphorescent PdTCPTBP moiety linearized and stabilised its O₂ calibration, increased hydrophilicity and solubility in water, decreased τ_0 and quenching constant and protected from quenching interferences.^{61,64}

Polymers are widely used as encapsulation and quenching media. Examples include common plastics (polystyrene, fluorinated polymers, polysulfone, polycarbonate, plasticized PVC), co-polymers, silicones, ormosils, sol-gels, and hydrogels. Proper selection of the polymer enables to achieve the desired specifications of sensor material, particularly its sensitivity, measurement range, response time, method of dye inclusion, biocompatibility, and sensor format (solid-state film coatings or nanoparticle probes). Polymer matrices with high permeability and diffusion of O₂ provide higher sensitivity.^{21,47,63}

Furthermore, available indicator dyes and sensor materials can be bundled with other indicators, reporters or reference dyes to produce O₂ sensitive materials with new functional features. Examples include new supramolecular structures,

bioconjugates, polymeric composites, nanoparticle formulations, and materials for multi-parametric analyses. These are covered in more detail in recent reviews.^{32,33,45,48,49}

3.2. Practical realisation and measurement set-ups for optical O₂ sensing

With respect to analysis of biological specimens, the main measurement tasks include: (1) sample oxygenation, *i.e.* O₂ concentration; (2) O₂ consumption rate (OCR), *i.e.* O₂ flux associated with a biochemical process in the system; (3) spatial distribution and gradients of O₂ within the sample; and (4) dynamics of O₂ changes in time and space.

The need to perform various analytical tasks with different biological specimens by O₂ sensing has also led to the development of a number of measurement set-ups and detection options, which enable O₂ measurement in a sample to be carried out in a number of different ways. Some typical examples are presented in Table 4 and Fig. 3.

Fibre-optic (micro)sensors provide flexibility and miniaturization, allowing us to access small compartments of the sample and even penetrate into tissue (needle sensors). However the sensor is usually non-detachable,²⁰ which makes them invasive (minimally though).

Point measurement of O₂ can be realised by applying a solid-state sensor coating to a small area in an assay vessel in contact with a liquid sample or headspace.²⁰ In solid-state materials indicator molecules are shielded from quenching interferences with a gas-diffusion barrier (polymeric matrix), they do not contaminate the sample, have high specific signals that can be used in various environments and facilitate quantitative O₂ measurements even in intensity mode. These spot sensors are relatively easy to fabricate and tune, and they can be measured from outside the vessel, *i.e.* contactless. However, they are not very suitable for studying complex respiring samples with profound heterogeneity and localised O₂ gradients.

Soluble probes can be distributed across respiring objects, such as cultured cells or tissue to enable O₂ analysis in different parts of the sample. They can be used without much adaptation in various assay substrates, miniature samples, and even be introduced into the cell or the 3D respiring sample such as tissue or whole animal. Such probes provide the highest degree of flexibility. However, their signals are usually much lower than for solid-state sensors (the latter can accommodate up to 1% w/w of the dye¹⁶), they contaminate the sample, and are harder to use on a large scale and with macroscopic samples.

While solid-state O₂ sensors and probes have their merits and limitations, micro- and nano-particle sensors combining the features of both can overcome the limitations. In addition, magnetic O₂-sensitive microparticles can be distributed in the sample, but then concentrated and manipulated with a magnet.⁶⁵

In some of the set-ups, solid-state (micro)sensors and probes are interchangeable and provide similar analytical performance (*e.g.* OCR assays in cuvettes or microplates with localised sensor coating or probe distributed evenly). Measurement of relatively uniform biological samples (microbial, mammalian cells) is

Table 3 The main types of phosphorescent O₂ sensitive materials

O ₂ -sensitive material	Measurement tasks	Advantages	Limitations
<i>Thin-film solid-state sensors:</i> Non-detachable solid-state (micro)sensors	Point measurement of dissolved and gaseous O ₂ concentration	Minimally invasive. Scanning capability (with micromanipulator). No sample contamination, multi-point systems (2–4 channels)	Less durable and robust than spot sensors. Fragile, invasive, costly
Detachable spot sensors (coatings, inserts)	Point measurements of dissolved or gaseous O ₂ concentration	Provide non-invasive, contact-less sensing, disposable and long-term use. No sample contamination, high optical signals, low cost, robustness. Many sensors can be probed with one instrument	Integration in sample is not easy. Limited flexibility and applicability – adhesion, solvents, microscopic samples. Point readings not always representative
Substrates with built-in sensors (coated plates, tubes, flasks)	Process control, screening of multiple samples	Convenient, ready for use. Optimised for disposable use in certain applications. Affordable	Limited flexibility, suit only certain applications. Assay redesign is difficult
Planar optodes (sensor sheets)	2D visualisation of O ₂ distribution and gradients	2D imaging capability	Fragile sensor films, require solid support. Difficult interfacing with samples
<i>Soluble probes:</i> General use (macro)molecular probes	Measurement of dissolved O ₂ in bulk samples	Compatible with various samples, substrates and tasks, especially small samples and screening applications. Adjustable signals and formats	Contaminate samples. Weaker signals than with solid-state sensors
Micro- and nano- particle probes	Measurement of dissolved O ₂ in bulk samples	Easy use by dispensing. Allow manipulation (e.g. magnetic beads). Flexible, easy to manufacture. Increased brightness and photostability	Contaminate samples. Stability, reproducibility and aggregation issues
Targeted probes (cell/tissue-permeable)	Cell/tissue penetration or binding. Control of local oxygenation and O ₂ gradients	Allow analysis of cell populations, tissue and individual cells	May impair cellular function. Cell-specific accumulation and distribution
Imaging probes	Measurement of vascular and tissue O ₂	High photostability, low intrinsic and phototoxicity, sufficient brightness and retention in cells	Often show cell specificity, lack of photostability, limited validation
Multi-functional and multi-modal probes	Provide additional reference signal or parameter	Increased flexibility, more stable (ratiometric) and multi-parametric measurements	Complex synthesis, high costs. Stability, batch-to-batch variability
<i>In vivo</i> probes	Allow use in live animals and tissues	Improved safety and biocompatibility, low toxicity, high level of validation	Require complex measurement set up and animal licences. Many accessory techniques

Table 4 Main measurement set-ups used in optical O₂ sensing with biological specimens

Measurement set-up	Brief description and sensor type	Main applications
Fibre-optic probe – microsensor	Optical fibre with sensor material on the tip, and protective cladding (optional). Normally solid-state coating, or probe solution in a microchamber on the tip	Point monitoring of sample O ₂ . OCR measurement. Scanning of O ₂ distribution (with micromanipulator)
Air-tight cuvette	Sealed cuvette (no headspace, air-tight), with or without stirrer. Extracellular probe or solid-state sensor (sticker or microsensor)	OCR measurements with enzymes, isolated mitochondria, suspension cells, small organisms
Partially sealed cuvette or chamber	Microplate wells sealed with mineral oil. Capillary cuvettes, ⁹³ vials with sensors, ⁹² extracellular probes or solid-state sensor coatings	Respiration and OCR measurements with cells, enzymes, small organisms isolated mitochondria. Analysis of cell metabolism, bioenergetics, drug treatment, cell physiology
Biochips with static samples	Sealable microplate, ⁹² XF analyser system, ¹⁰⁵ microchannel biochips ²⁹²	Respiration and OCR measurements with cells, enzymes, small organisms isolated mitochondria. Analysis of cell metabolism, bioenergetics, drug treatment, cell physiology
Fluidic systems and perfusion cells	O ₂ measurement under flow conditions in a chamber with biological material. Extracellular and intracellular probes, solid-state sensor coatings and microsensors ¹⁷³	O ₂ consumption measurements. Responses of cell and tissue samples to stimulation, drug treatment, hypoxia, etc.
Open samples in contact with atmosphere	An open plate or imaging mini-dish with respiring samples, in medium exposed to gaseous atmosphere. Mostly intracellular probes ¹⁸	<i>In situ</i> control of oxygenation in cells, spheroids, tissues. Monitoring of steady-state oxygenation of cells and respiratory responses to metabolic stimulation.
<i>In vivo</i> O ₂ measurements	O ₂ sensor/probe and detection system are tailored to a particular biological model and measurement task ^{18,75}	Studies of cell physiology and hypoxia. O ₂ imaging Levels, maps of O ₂ concentration, their dynamics. Physiological studies

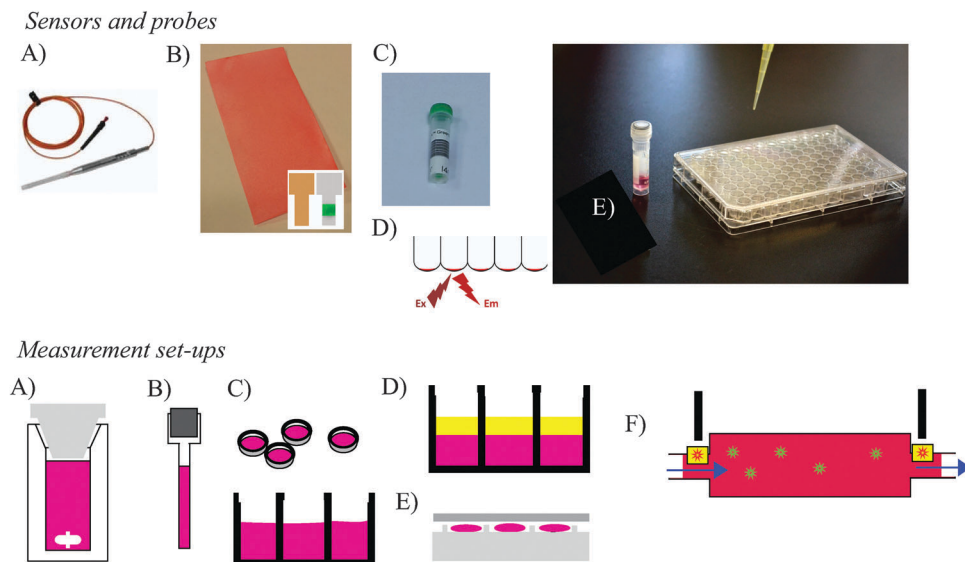


Fig. 3 Examples of common materials and measurement set-ups used for O_2 sensing in biological samples. Sensors and probes: (A) fibre-optic micro-sensor. (B) Planar sensor sheet and stickers with sensor dots. (C and D) Vial and microplate with sensor coating and measurement set-up. (E) Soluble probe used with standard microplates. Measurement set-ups: (A) sealed stirred cuvette. (B) Glass capillary with liquid barrier; (C) open microplate and imaging mini-dishes. (D) Microplate wells sealed with oil. (E) Low-volume sealable microplate. (F) Perfusion chamber, can operate under flow or static conditions with soluble (green) or solid-state sensors (yellow-red).

well developed and some applications are becoming standard, such as cell-based assays, drug screening, microbial tests and bioreactor monitoring. In other cases with complex and non-standard samples, such as whole organism ecosystems, micro-sensors, customisation of both the O_2 sensor/probe and detector parts is often required, which may be minor or very significant.

For quantitative monitoring of O_2 concentration using a sensor, it is necessary to establish the relationship between the measured optical parameter and O_2 concentration, *i.e.* calibration. The main read-out parameters in optical O_2 sensing are phosphorescence intensity (I) and lifetime (τ). They relate to O_2 concentration according to the Stern–Volmer equation:⁶⁶

$$I_0/I = 1 + K_{S-V}[O_2] \quad \text{or} \quad \tau_0/\tau = 1 + K_{S-V}[O_2] \quad (1)$$

$$[O_2] = (I_0/I - 1)/K_{S-V} \quad \text{or} \quad [O_2] = (\tau_0/\tau - 1)/(k_q\tau_0). \quad (2)$$

The intensity and lifetime values at zero O_2 are designated I_0 and τ_0 , while the sensitivity of O_2 -sensitive material is determined by the quenching constant (K_{S-V}), which is a function of τ_0 , the immediate environment of the dye, sterical factors and

temperature (expressed *via* the bimolecular quenching rate constant, k_q).⁶⁶ For heterogeneous sensor materials and systems with non-ideal behaviour, more complex mathematical equations or fitting functions are used.⁶⁷ For the dual-fluorophore sensor systems, the intensity ratio of the O_2 -insensitive and O_2 -sensitive signals (I_1/I_2) is used for quantitation, this relationship resembles linearised Stern–Volmer.^{21,68,69} Typical O_2 calibrations are shown in Fig. 4. Depending on the sensor material, instrumentation and the biological sample, calibration procedures may vary. Specific examples can be found in ref. 62, 70 and 71.

A broad variety of O_2 sensitive materials having different physical states, spectral characteristics, sensitivity to O_2 , chemical–physical properties, and biocompatibility, is further augmented by the versatile photoluminescent detection. This allows design of simple, portable and affordable (bio)analytical systems for in-field, *in situ*, or laboratory use which can be operated by non-skilled personnel,^{20,72} as well as sophisticated equipment tailored to a particular measurement task or application.^{73,74} O_2 sensors can also be integrated in existing widely available detection platforms and multi-functional instrumentation,

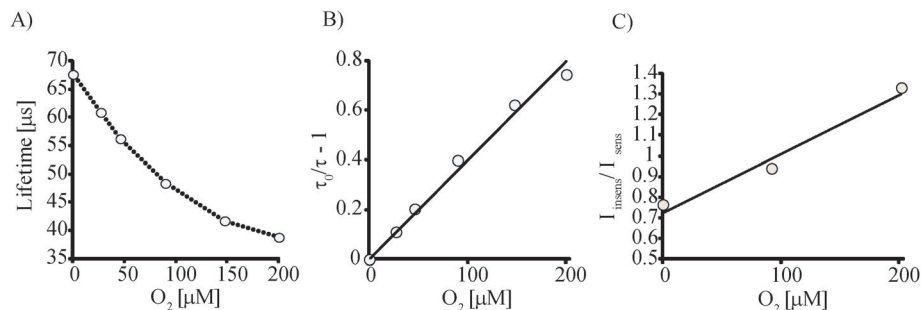


Fig. 4 Examples of calibrations of O_2 sensing materials operating in lifetime (A and B) and ratiometric intensity modes.

particularly conventional fluorescence readers and spectrometers, live cell and *in vivo* imaging systems. Similar to fluorescent bioimaging, multi-photon excitation is a preferred mode for O₂ sensing and imaging deep into tissue.⁷⁵

Sensor intensity signals can be influenced by many parameters including dye concentration, measurement geometry, optical properties of the sample, drift of optoelectronic components, and photobleaching. Therefore, stable and precise calibration is difficult to achieve. These instability issues can be addressed using the sensors containing an additional spectrally distinct dye (O₂-insensitive reference) and ratiometric measurements at two different wavelengths. This compensates for many of the above factors,⁶⁸ though not completely.^{21,76,77} Phosphorescence lifetime-based sensing is rather insensitive to variations of indicator concentration and experimental conditions that can influence intensity signals, it generally provides more stable measurements and enables to use once-off calibration.

Phosphorescence lifetime-based sensing schemes require more specialised instrumentation, with time-gated or frequency-modulated excitation and time or phase correlated detection, respectively.⁷⁸ Such platforms are also quite common in biomedical research. They enable assessment of biological activity and functional characteristics of different biological samples in a highly informative and versatile manner. Still, instrumental errors, environmental factors and measurement artefacts can lead to differences in experimental results and calibrations, so it is highly recommended to perform critical assessment, testing and calibration for such systems.

General instrumentation and data processing for phosphorescence intensity and lifetime-based O₂ sensing are described in many reviews,^{20,45,51,79,80} and will not be discussed in detail here. At the same time, new detection systems and set-ups for specific biological application appearing with high frequency will be discussed in Section 4. The large leap in digital imaging equipment over the last few years, from high-spec gated CCD cameras to low-cost CMOS chips and mobile phone cameras, is also affecting the O₂ sensing area, with a number of systems and applications already described and commercialised.^{81,82}

Last but not least, for many applications an important requirement for the O₂ sensor is that it does not bring or increase the risk of contamination for the sample under investigation. Sensors/probes usually need to be provided in a sterile form, while reusable sensors should withstand factory, on-line or at-line sterilisation. Standard sterilisation procedures, such as gamma irradiation, chemical sterilisation, heating or autoclaving, often cause damage to the sensor material,⁸³ so it has to be specially designed and optimised for this. Storage stability, biofouling,⁸⁴ transportation are other practical limitations for many O₂ sensors, especially liquid probes, and nanoparticle formulations.

4. Detection of biological activity by quenched-phosphorescence O₂ sensing

Biological applications of optical O₂ sensors often differ from their conventional industrial applications, in which measured

O₂ concentration is the main parameter of interest. In biological detection, a primary signal from the sensor on its own rarely provides the sought information about the sample. To extract specific and physiologically relevant information, the input of additional sample parameters and criteria is usually required for the assay.

The additional information fed back into the sensor-based system can be, for example: key physical-chemical and biological characteristics of the sample and its micro- and macro-environment (*e.g.* mass, geometrical dimensions, temperature, humidity); known theoretical relationships between measured optical signals and sample parameters; initial or final conditions, results of experiments produced with control samples or artificial signal modulation; algorithms for converting sensor primary signal into biological parameters which account for key variables and potential sources of error.

This reflects the general strategy of analysing complex biological specimens by O₂ sensing, while it is the task of the operator to keep it to a reasonable minimum that ensures stable and robust measurements and generation of unambiguous end-results. Detailed optimisation of the measurement system and independent verification of results it produces are very important. Indeed, not paying due attention to these factors, many of which have biological nature, may lead to measurement artefacts, misinterpretation of experimental results or too broad generalisations (discussed in more detail in Section 4.13). If assay results are valid only for a particular condition or a sample type, this should be clearly specified. In other cases, primary optical signals can be applied for assessment, without calculating O₂ or OCR values.

In the following sub-sections we review the main groups of applications of quenched-phosphorescence O₂ sensing targeted at biological detection, assessment of activity and functional characteristics of biological specimens, with particular examples and their analysis.

4.1. Enzymatic assays

Living systems constantly maintain and reproduce themselves through a complex network of metabolic, biosynthetic and biodegradation pathways driven by enzyme-catalysed reactions, many of which are O₂ dependent. Even the reactions that do not directly consume or release O₂ are often coupled with other enzymatic reactions that are influenced by O₂. Quenched-phosphorescence O₂ sensors allow for investigation of enzymes that directly consume O₂. Indirect assessment, when the target enzyme-catalysed reaction is coupled with an oxidase consuming the product of the first reaction, is also possible.¹³ Altogether, this provides a large scope for designing various enzymatic assays based on the monitoring of consumption or release of dissolved O₂. Such assays can be configured to perform: (1) analysis of enzyme substrates – key metabolites such as glucose, lactate, amino acids, cholesterol; (2) measurement of activity of clinically and industrially relevant enzymes; (3) mechanistic studies of enzymes, their substrates and inhibitors; and (4) screening of combinatorial libraries of new chemical entities

and biocatalysts (engineered enzymes with altered catalytic properties).

One well developed area of enzymatic assays is the determination of glucose in biological fluids (blood plasma, urine, saliva, tear, sweat), used in routine clinical tests, diagnostics of human diseases, intensive and personal care (40% of all clinical tests with blood samples include glucose determination⁸⁵), biotechnology. Large clinical analysers, portable devices (glucometers) and specialised systems (*e.g.* implantable biosensors for continuous monitoring, insulin pumps for diabetic care) conduct glucose assays in a number of different formats using enzymatic platforms.

Glucose determination using glucose oxidase and O₂ sensing is well established, solid-state sensors, fibre-optic and soluble probes were applied and demonstrated satisfactory performance.^{47,53,72} However, electrochemical and colorimetric biosensors that do not use O₂ detection and optical transduction^{19,86} are currently more successful in this huge market than optical biosensor systems because: (1) O₂ transducers operate by measuring a decrease in initially high O₂ concentration ($\sim 200\ \mu\text{M}$ in air-saturated samples¹¹) being less sensitive than *e.g.* measurement of H₂O₂ or mediator production;⁸⁵ vast excess of ambient O₂ and its diffusion into the sample can also interfere; (2) greater susceptibility of the phosphorescence quenching method to environmental parameters such as temperature, ambient light, oxygenation of blood, sample properties; (3) higher complexity and slower development of optical O₂ transducers compared to their electrochemical and colourimetric counterparts, especially at the start of establishment of this market; (4) non-optimal analytical performance, robustness and costs of early stage optical biosensors. As a result, traditional enzyme biosensors with optical O₂ transducers for glucose and other metabolites have not been very successful so far.

On the other hand, O₂ sensor technology is now more competitive and mature, many high-performance components are available for designing O₂ transducers and biosensors of new generation.^{41,43,55} Many of their initial bottlenecks have now been addressed, such that modern phosphorescence based O₂ transducers can now provide: (1) internal referencing and stable calibration through lifetime based O₂ sensing; (2) reliable operation under ambient light conditions and variable sample optical properties; (3) multi-parametric systems with internal compensation for key sample variables, particularly temperature and dissolved O₂; (4) flexible, multi-functional sensor materials compatible with biological samples, affordable and scalable in fabrication; (5) miniaturized, modular and integrated detection architecture; (6) contactless interrogation with the sample, intelligent communication with the detector, software and user/operator (also wireless); (7) improved analytical performance matching or exceeding that of the alternative (*e.g.* electrochemical) biosensors.

A new glucose biosensor uses glucose oxidase, optical O₂ transducer, reference O₂ sensor to correct for dissolved O₂, and a microdialysis unit. Implanted through skin, this microsensor allows long-term transcutaneous measurements of glucose in blood plasma during intensive care, showing advantages over

the electrode-based analog.⁸⁷ Another recent example is a biosensor for H₂O₂ with Ru-based O₂ transducer and immobilised catalase enzyme.⁸⁸

Significant progress has been achieved with laboratory and screening enzymatic assays which rely on optical O₂ sensing. An important development was their adaptation for standard bioassay substrates, particularly 96- and 384-well microtiter plates, using samples sealed with mineral oil.⁸⁹ Also PtCP and PtPFPP based sensors and probes can be used on conventional time-resolved fluorescence readers (originally developed for *in vitro* diagnostic tests using lanthanide chelate labels) without any significant modifications, to perform lifetime-based O₂ sensing by the Rapid Lifetime Determination (RLD) method.⁹⁰ In RLD emission intensity signals (F_1 , F_2) are collected at two different delay times (t_1 , t_2) after the excitation pulse, from which lifetime is calculated as: $\tau = (t_2 - t_1)/\ln(F_1/F_2)$.

Such platforms provide robust, sensitive and simple measurement of enzymatic reactions in complex samples, with a quantitative and accurate readout and stability to optical and quenching interferences.⁹¹ Using automated dispensing and liquid handling (multi-channel pipettes and robotics) and parallel analysis of a large number of samples in kinetic mode (scanning periodically the microplate), a large number of samples can be processed rapidly and conveniently, thus enabling an array of screening applications.⁹²

Another significant development has been the high-sensitivity enzymatic assays. In particular, the fluorescence based platform LightCycler[®] (Roche), dedicated for real-time monitoring of nucleic acid amplification, has been adapted for enzymatic assays with optical O₂ transducers. The glass capillary cuvettes providing a good barrier for ambient O₂ are combined with their efficient temperature control (active ventilation with heated air) and precise mechano-optical alignment in the carousel (accommodates up to 48 samples), provide unsurpassed sensitivity and accuracy in the monitoring of enzymatic O₂ consumption. It has been demonstrated with a number of enzymatic assays of high practical significance, including the detection of monoamine oxidase enzymes, cyclooxygenases, cytochrome P450 isoforms (drug metabolism), their substrates (dopamine and other catecholamines), and inhibitors.⁹³

These measurement formats are also applicable to coupled enzymatic reactions. Thus, cholinesterase–choline oxidase system and soluble O₂ probes were used for the detection of cholinesterase inhibitors.⁹⁴ These simple and sensitive assay systems operating in the screening format can be used in security and biodefence applications to monitor chemical warfare and relevant toxins.⁹⁵ In the coming years, we anticipate a burst of activity in development and application of enzyme biosensors.

4.2. Analysis of mitochondrial respiration

A mitochondrion is a small organelle (size of 0.5–1 μm) present in eukaryotic cells in significant numbers. Mitochondria host the most efficient energy generation pathway, oxidative phosphorylation (OxPhos), and are the main producers of cellular ATP.⁵ They consume most of cellular O₂ ($\sim 90\%$, but may vary for different cell types), while all the other processes utilize the

remaining small portion.⁵ Mitochondria participate in many other vital processes within the cell, including Ca^{2+} signalling, ion fluxes, redox homeostasis,⁹⁶ generation of reactive oxygen species (ROS) and trans-membrane potentials, initiation of apoptosis and autophagy. They have their own chromosome (mtDNA) encoding some, but not all, of their proteins.⁹⁷

Although tightly connected to the internal cellular life, mitochondria can be isolated from cells and tissues retaining their integrity and many vital functions.⁹⁷ They can be manipulated and used to study the complex machinery of the mitochondrion, its diverse range of physiological functions and particularly OxPhos – a multi-enzyme system located in the inner mitochondrial membrane. An essential part of the OxPhos is the Electron Transport Chain (ETC), which consists of complexes I–IV. O_2 molecules act as the terminal acceptor of the ETC at complex IV – cytochrome *c* oxidase. The ETC is largely responsible for the formation and maintenance of physiological gradients of potential, $\Delta\psi_{\text{m}}$, and $[\text{H}^+]$, $\Delta\text{pH}_{\text{m}}$, across the mitochondrial membrane, producing the proton motive force, pmf, which drives the synthesis of ATP molecules by complex V (ATP-synthase). This is shown schematically in Fig. 5.

As a result, by supplying the different substrates, specific inhibitors of complexes I–V and uncouplers (Fig. 5) and measuring the OCR, one can extract detailed mechanistic information about the activity of different respiration states (states 1–4 respiration⁹⁸), the relative contribution of complex I and complex II pathways, the degree of coupling and spare respiratory capacity upon uncoupling, the mode of action of a drug on the mitochondria and its particular target(s) within. Many other processes linked to or influenced by the OxPhos can also be studied.⁹⁷

Analysis of respiration of isolated mitochondria allows studying of mitochondrial (dys)function, toxicological effects of drugs, detailed bioenergetic and mechanistic studies.⁵

Traditionally OCR has been measured in a sealed, stirred chamber with built-in Clark-type O_2 electrode.⁹⁷ Although accurate and quantitative, this set-up cannot provide the required throughput and speed (mitochondrial preparations are only stable for a few hours on ice), stability to interferences and reliable measurements at low O_2 levels.

The introduction of quenched phosphorescence O_2 sensing has been an important step forward.³³ Although studied for many decades, many fundamental questions still remain, including the dependence of respiration on O_2 concentration and regulation of OxPhos by internal and external factors. The Oxyphor G2 probe⁶¹ was recently applied to analyse O_2 -dependent respiration of mitochondria in a sealed system.⁹⁹ K_{M} for O_2 was shown to decrease significantly at lower pH, an effect which may have physiological consequences giving the advantage to transformed cells over normal cells during cancer progression.

Another milestone was the analysis of mitochondrial respiration in standard microtiter plates using a PtCP based probe MitoXpress[®]-Xtra (Luxcel Biosciences) on a standard fluorescence plate reader and simple intensity or preferably lifetime based O_2 sensing.⁹⁸ Such assays are conducted in partially sealed samples: a layer of mineral oil is applied on top of each sample, thus priming the assay and depletion of O_2 which is then related to the OCR. They currently operate in 96- and 384-well plates, with parallel measurement of samples with different drugs and conditions. Respirometric screening for mitochondrial toxicity is typically performed on two respiration media: one without ADP (state 2 respiration) to identify mitochondrial uncouplers, and another with ADP (state 3 respiration) to identify mitochondrial inhibitors. Inhibition is initially tested at one compound concentration (typically 1 μM), and for positive hits IC₅₀ is subsequently determined. Uncouplers normally

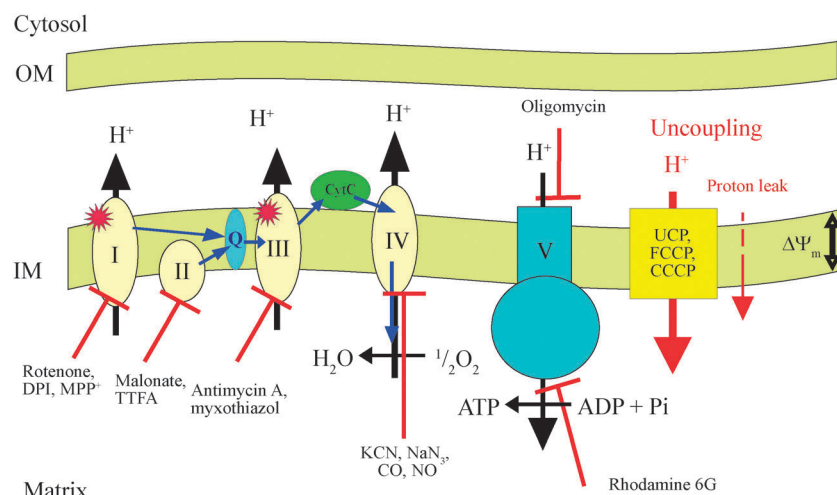


Fig. 5 General representation of OxPhos machinery in the mitochondria of eukaryotes. The ETC consists of four enzymes: complexes I (NADH-ubiquinol-oxidoreductase), II (succinate-ubiquinol oxidoreductase), III (ubiquinol-cytochrome *c* oxidoreductase) and IV (cytochrome *c* oxidase) located in the inner mitochondrial membrane.⁹⁷ Transfer of electrons (blue arrows) is mediated by coenzyme Q_{10} (Q) and cytochrome *c* (cyt *c*) and results in O_2 consumption at complex IV. The proton, $\Delta\text{pH}_{\text{m}}$, and potential, $\Delta\psi_{\text{m}}$, gradients are used to produce ATP by complex V (F_0F_1 ATP-synthase). Electron leaks at complexes I and III (indicated by stars) generate superoxide radical ($\bullet\text{O}_2^-$) – the main source of ROS. Proton movement across the inner membrane is possible through the action of protonophores (FCCP, CCCP), uncoupling proteins (UCP) or intrinsic proton leak. Mitochondrial function is also linked to Krebs cycle (supplies succinate), Ca^{2+} signalling, apoptosis and other cellular processes.⁹⁷ OM – outer mitochondrial membrane, IM – inner mitochondrial membrane.

produce bell-shaped responses (increased respiration due to uncoupling followed by decrease due to toxicity), and to avoid false-negative results they must be tested at several different concentrations. This method has been adopted by pharmaceutical companies as a screen for mitochondrial toxicity of drug candidates.^{89,98}

Multi-parametric assessment of mitochondrial function of animal tissue and organs was also undertaken by combined measurement of OCR with MitoXpress[®]-Xtra probe and mitochondrial swelling, membrane potential and cytochrome *c* release, which provided mechanistic information on drug induced injury of liver tissue.¹⁰⁰ Solid-state sensor systems including coated microplates¹⁰¹ and a Seahorse XF analyser (see Section 4.3) were also used in this application.¹⁰²

At the same time, analysis of isolated mitochondria has a number of limitations, mainly the lack of relevance to the context of the whole cell. For drugs that cannot easily pass cell membranes and reach mitochondrial components the observed effects appear to be overestimated. Many drugs with specific transport mechanisms, tissue specificity, metabolism inside the cell or effects on other metabolic pathways are difficult to study. The results are also influenced by the isolation procedure which determines the degree of 'coupling' and quality of mitochondrial preparations. Many of these issues can be overcome by analysing whole cells.

4.3. Mammalian cell respiration

Mammalian cell cultures are in the centre of life sciences and biomedical research. They allow researchers to study cellular processes in their integrity and conditions resembling the physiological environment (adherent state, contacts with neighbouring cells). Respiration of intact cells accounts for transport of key nutrients including O₂, the multiple regulatory networks and compensatory mechanisms, other energy generating, metabolic and signalling pathways and secondary factors acting in the cells. Non-mitochondrial O₂ consumption can also be probed.

General strategies for microplate based OCR measurements with cells are similar to those described in Sections 4.1 and 4.2. However, weak respiration of mammalian cells (<1 nM O₂ per min per 10⁶ cells⁵) is more difficult to measure, especially for adherent cells the maximal density of which is limited by the confluent layer. Low respiration rates and back-diffusion of atmospheric O₂ make determination of absolute OCRs more difficult, however relative changes in OCR can be measured reliably. So far, many types of mammalian cells including liver, muscle, neuronal cell lines, primary cells and induced pluripotent stem cells were measured in standard microplates with samples sealed with mineral oil⁸⁹ and detection on a standard fluorescence reader.^{98,103,104} Moreover, photophysical characteristics of a MitoXpress[®]-Xtra probe allow multiplexing with a long-decay pH-sensitive lanthanide probe pH-Xtra[™] (Luxcel Biosciences), and parallel measurement of OCR and extracellular acidification (ECA) in the same sample without cross-talk.¹⁰⁴ Other fluorescent probes and cell-based assays can be added in the multiplex or parallel format.

Seahorse Bioscience designed an integrated opto-mechanical system for OCR and ECA measurements with adherent cells, called an XF analyser. It operates with dedicated microplates having specially shaped wells and moving cartridges with solid-state phosphorescent O₂ and pH sensors and light guides.¹⁰⁵ During the measurement cycle, the cartridges create micro-chambers at the bottom of each well sealing the cells and providing rapid and sensitive quantification of OCR and ECA (due to a high cell volume to sample volume ratio). Complex modelling is used to compensate for O₂ leaks and work out absolute OCR values for the respiring sample.¹⁰⁶ The analyser exists in 24 and 96-well plate modifications, with two or three-parametric (with CO₂ sensors) detection, built-in thermostats, drug vessels and injectors for each well. This system has been used with many different cell lines, primary cells and β -cell islets and also with suspension cells and isolated mitochondria where its benefits are not so evident.^{96,105}

The above two platforms enable the simultaneous assessment of the two main energy production pathways within the cells: OxPhos (OCR) and glycolysis (ECA). Further multiplexing can be applied in the analysis of drug-induced toxicity, mitochondrial (dys)function, disease models, ageing, biotechnology and many therapeutic areas.^{107,108}

4.3.1. Analysis of drug-induced toxicity. Different transport, degradation and action mechanisms for a drug in isolated mitochondria and in intact cells often result in different patterns of toxicity. Consequently, organ toxicity and drug safety assessment is now shifting towards cell-based models. OCR assays can provide valuable information about the mode of drug action and possible toxicological effects *in vivo*,^{109,110} especially when bundled together with other markers of toxicity.

Examples of action of different model drugs on cells and prognostic values of different biomarkers are shown in Fig. 6. Thus, total cellular ATP assay, which provides information on general viability and non-specific cytotoxicity, misses many toxins with strong effects on cell metabolism and bioenergetics. OCR is much more sensitive and selective identifying both uncouplers (FCCP) and inhibitors (antimycin, rotenone) of respiration. For drugs that act on the glycolytic ATP pathway (okadaic acid), changes in OCR may not be so prominent, but clearly detectable by combined OCR/ECA assay.⁹⁷

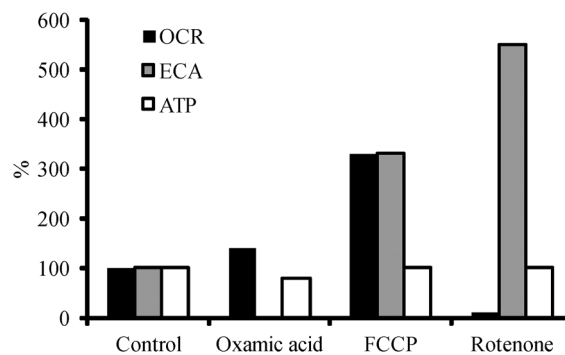


Fig. 6 Responses of drugs with different mode of action on cell metabolism assessed by the ATP, OCR and ECA assays. Modified from ref. 104.

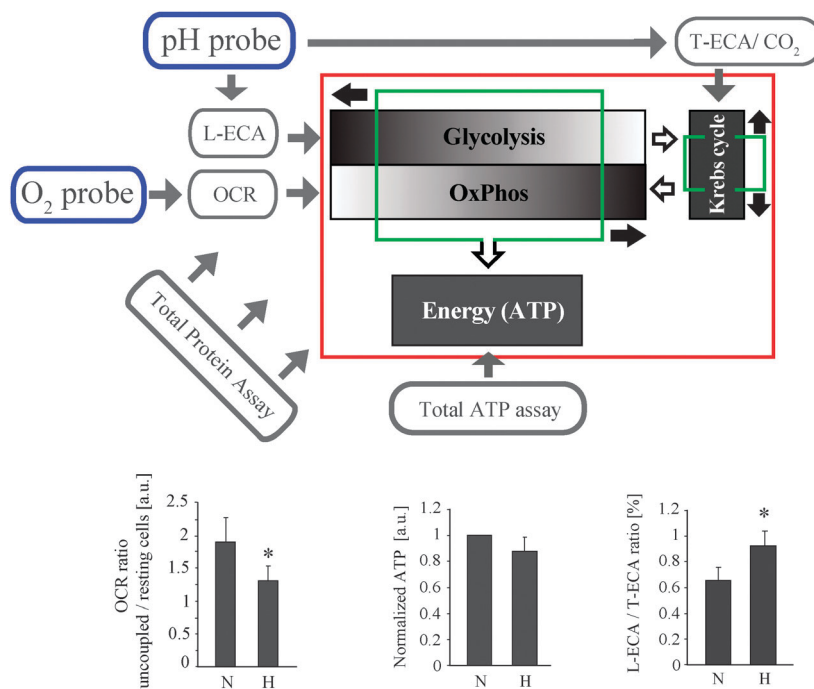


Fig. 7 Representation of the main cell energy generating pathways of the cell and CEB concept. Normal balance of glycolysis, OxPhos and Krebs cycle (green rectangles) can be perturbed by mutations, drugs, or other factors. The balance of different pathways can be probed using indicated methods: OxPhos/OCR – with an O_2 probe in sealed samples; glycolysis/L-ECA (lactate component) – with a pH probe in unsealed samples; Krebs cycle/T-ECA (total – lactate and CO_2) – with a pH probe in sealed samples; total ATP/general cell viability – with a commercial bioluminescent kit; variation of biomass in different samples – with total protein assay (modified from ref. 114). Bottom panel: demonstration of CEB with neural PC12 cells maintained under 21% (normoxia, N) and 3% O_2 for 30 days (hypoxia, H). Altered OCR, ATP and L-ECA/T-ECA ratios reflect a reduction in OxPhos, increase in glycolytic flux in H-cells and reduced spare capacity (modified from ref. 117).

A two-parametric OCR/ECA platform for cell-based screening of drug-induced mitochondrial dysfunction and organ toxicity using a MitoXpress[®]-Xtra probe and standard 384-well plates was developed and validated with 200 commercial compounds on a common suspension cell line, HL60.¹⁰³ Similar tests can be carried out on an XF analyzer.¹¹¹ The two systems show comparable data, XF is a standalone unit with dedicated consumables for enhanced sensitivity, whereas the MitoXpress[®] platform operating on standard microplates and plate readers targets flexibility, high throughput and low cost.

Multi-parametric analysis of respiration and other markers revealed new targets for another pharmacological agent – bafilomycin A1, a commonly used inhibitor of V-ATPase and potential anti-cancer drug¹¹² (see Section 4.5). In addition to V-ATPase inhibition, bafilomycin was shown to enhance OCR through mild mitochondrial uncoupling in neuronal cells.

A mechanistic toxicological study was described with microcystins, hepatotoxins produced by cyanobacteria during algal blooms which can contaminate fresh and drinking water. Microcystins are taken up by liver cells *via* organic anion transporting polypeptides (OATP) and inhibit cellular protein phosphatases inducing toxic effects. The study with microcystin-LR and measurement of OCR, ROS, cellular ATP, ECA and phosphatase activity in different cell lines and primary hepatocytes revealed new targets for microcystins in the mitochondria and an uncoupling effect on respiration.¹¹³ Furthermore, a new screening test for the presence of microcystins in environmental

samples was developed, which uses ordinary cell lines (normally immune to microcystins as they lack OATP) and facilitated delivery of the toxin with the aid of transfection reagent. Showing sub-nM sensitivity, this *in vitro* test offers a viable alternative to current tests for microcystins on animals and primary hepatocytes, or by ELISA.

4.3.2. Assessment of cell bioenergetics. The above platforms based on optical O_2 and pH probes/sensors can be incorporated into a larger suite of assays to provide comprehensive assessment of cell bioenergetics and related processes. Plate reader platforms are particularly amenable to multi-parametric analysis, one such concept called ‘Cell Energy Budget’ (CEB) is presented schematically in Fig. 7.

According to CEB, OCR/ECA assays (see Section 4.3.1) were extended with ECA measurement in sealed (T-ECA) and unsealed (L-ECA) samples. This allowed an assessment of the individual contribution of anaerobic glycolysis (produces lactate which is extruded from the cells) and Krebs cycle (produces CO_2 which remains in sealed samples but escapes from unsealed). Total ATP assay was introduced to monitor overall cell viability (energy depletion and death), and total protein assay – to correct for differences in biomass (cell number, size, growth rate) in different samples. Furthermore, the cells are measured both in resting and uncoupled (with FCCP added) states, thus determining basal and maximal respiration and glycolytic fluxes (*i.e.* spare capacity).

A typical CEB experiment includes: (1) kinetic OCR assay with O_2 probe (MitoXpress[®]-Xtra) and sealed samples to measure

basal and maximal respiration; (2) kinetic L-ECA assay with pH probe (pH-Xtra) and unsealed samples to measure glycolytic flux; (3) kinetic T-ECA assay with pH probe (pH-Xtra) and sealed sample to determine Krebs cycle activity as the difference between T-ECA and L-ECA; (4) end-point measurement of total cellular ATP with a luminescent kit, and (5) total protein assay followed by normalisation of measured OCR, ECA and ATP values for biomass content. Each assay is performed in several replicates (3–5), providing statistically plausible data.

In this form, CEB provides a powerful tool for the analysis of different cells (normal, diseased, treated, transformed or different types), their metabolic features and modulators of cell bioenergetics and mitochondrial function.¹¹⁴ It minimises the risk of missing minor impairments in key pathways due to compensation by alternative pathways. Such ‘masking’ can occur in the basal, but not the uncoupled state. Similar studies can also be performed on Seahorse XF analyser, the new model of which also measures O₂, pH and CO₂.

The CEB concept was applied for analysis of mitochondrial protein PNC1 which is involved in mtDNA biogenesis,^{114,115} Krebs cycle enzyme fumarate hydratase implicated in renal cell cancer and hereditary leiomyomatosis^{114,116} and the effect of long-term hypoxia (physiological normoxia) on the function of neural cancer cells.¹¹⁷

Simple OCR measurements were also applied in many physiological studies. Thus, increased biogenesis and PGC-1 α levels were seen to increase O₂ consumption leading to sustained hypoxia and stabilisation of HIF-1 α protein, thus proving that PGC-1 α and HIF-1 α pathways are cross-regulated.¹¹⁸ Studies on cancer metabolism,^{119–121} mitochondrial dysfunction due to loss of function of cellular proteins,^{115,116,122,123} hypoxia,¹¹⁷ differentiation and function of osteoclast cells in rheumatoid arthritis,¹²⁴ function of primary¹²⁵ and stem cells,¹²⁶ ageing¹²⁷ were also studied using this method.

Overall, OCR assays with mammalian cell models can be used to study general mechanisms of aerobic respiration, ROS, NO and Ca²⁺ signalling, uncoupling proteins, thermoregulation and obesity, neurodegeneration, cardiovascular and metabolic diseases, hypoxia research. Also drug discovery and development are now shifting from relatively non-specific ‘gunshot’ screening of compound libraries on cell models to more focused approaches which provide mechanistic information and quantitative structure–activity relationships (SARs) about particular targets, metabolic pathways and signalling networks in the cell and whole tissue. The basic panel can be expanded and customised with other functional assays. In this regard, respirometric cell-based assays and the CEB platform provide powerful tools. Another important area is security and bio-defence, where such cell-based assay platforms can be used to detect low toxins, chemical and biological warfare at low levels, with high sensitivity and selectivity.

At the same time, traditional 2D cell cultures also possess drawbacks, the main being their limited resemblance of 3D environment in live tissue and organs. Replacement of cell monolayers and detached cells with mixed cultures, 3D scaffold and spheroid models, tissue slices, and the use of microfluidic

biochips and *ex vivo* models¹²⁸ can address this (see Sections 4.6–4.8).

4.4. O₂ in microbial cultures

Microbiology deals with a broad range of microorganisms from viruses to fungi. Some of these are pathogenic and possess significant hazard to human health, for example, deadly infections, food borne (*E. coli* O157), clinical (MRSA, rotavirus) and environmental (*Campylobacter*, coliforms) pathogens. These have to be monitored at very low concentrations, down to one viable cell, prevented from uncontrolled circulation and proliferation, and eradicated if possible. Other microorganisms are very beneficial for mankind, actively used in human practice (baker yeasts, lactic bacteria) and biotechnology (microbial fermentations, production and degradation of biopolymers, drugs), or being inherent part of our body (*e.g.* microflora in the gut, skin). Viruses do not have their own metabolism and utilise infected cells for replication, however their invasion changes homeostatic and functional conditions of host cells and this can be used for indirect detection *via* O₂.

Aerobic and microaerobic bacteria consume significant amounts of O₂, and therefore are dependent on aeration conditions. Compared to mammalian cells, common aerobic bacteria are characterised by high proliferation rates (typical doubling times >24 h and <30 min, respectively) and respiration.¹²⁹ At the same time, bacterial cultures exhibit different growth phases: lag phase, exponential growth, stationary (substrate limitation) and death (depletion of nutrients) phases,¹³⁰ and are able to quickly adjust their metabolism to new environmental conditions, nutrients and O₂ availability, stress factors. Stressed or injured bacteria (*e.g.* after dehydration, sterilisation or drug treatment) can stay in the lag phase for a long time before they enter active growth and respiration. Such behaviour makes their detection *via* O₂ respiration more difficult. Nonetheless, when placed in a nutrient-rich media which promotes growth, microbial samples can quickly enter exponential growth and produce a robust change in oxygenation conditions. The change, which usually occurs at a point when critical cell numbers and OCR are reached, is easily detectable using an O₂ sensor.

Respirometric analyses of bacterial cultures by quenched-phosphorescence O₂ sensing^{131,132} are gaining popularity. Dedicated products such as BD Biosensor™ plates (Becton Dickinson), SensorDish® Reader and plates (Presens), the vial and plate based systems GreenLight® 910, 930 and 960 (Mocon-Luxcel Biosciences partnership)^{133,134} are commercially available. They allow rapid, high throughput analysis of microbial cultures and complex samples, high sensitivity (down to a single viable cell), broad dynamic range (10⁰–10⁸ cfu ml^{−1}), simple add and measure procedure, general convenience, affordable costs and broad range of applications. Therefore micro-respirometry in *liquid* media provides an attractive alternative to the conventional colony counting method on solid media (agar plating¹³⁵) and other rapid microbiological methods.¹³⁶ O₂ profiles can be used to determine the absence (sterility) or the presence of *viable* aerobic microorganisms, their proliferation rate and metabolic activity. Enumeration of bacteria in original samples can be achieved by applying pre-determined analytical relationship (calibration) to respirometric data. Doubling times can be

determined by analysing the sample at two or more dilutions, and used for predictive identification of microbial species.¹³⁷ Yeast and fungi are also detectable using this method, though their respiration can be significantly slower than for bacteria.¹³⁸

Metabolic and chemical interactions between different organisms is an emerging area, for example in gut health, (neuro)gastroenterology, studies of biogeochemical communities in soil.² Secondary metabolites often possess the functions of antibiotics, toxins, elicitors or signals in such systems.¹⁰ Antifungal bacterial metabolite 1,4-diacetylphloroglucinol was studied in cultures of *S. cerevisiae* by high-throughput OCR measurements, uncovering its new uncoupling effect on yeast mitochondria.¹³⁹

Detection of intrinsic bioluminescence of *Vibrio fischeri* bacteria (e.g. Microtox[®] kits) is currently used as a standard acute toxicity testing method for chemical and environmental samples.¹⁴⁰ Being relatively fast, sensitive and simple, this test is limited to one particular organism (prokaryote), problematic with samples that are turbid, absorb light, quench or interfere with the luminescent reaction, and operates on a dedicated instrument. An alternative approach has been proposed based on respirometric detection with MitoXpress[®]-Xtra probe in standard microplates on a fluorescent plate reader.¹⁴¹ Using a similar procedure to the MicroTox, this respirometric test produces comparable results with *V. fischeri*, but improved sample throughput, automation and miniaturization. It is not limited to luminescent bacterial strains, and can use many other test organisms and their panels which may include prokaryotic and eukaryotic cells, small invertebrates and vertebrates.¹⁴² Such 'respirometric profiling' provides more detailed toxicological data and can be used for predictive identification of different groups of toxicants.

As growth and respiration depend on the culturing conditions (media, temperature, pH, ions, additives),¹⁴³ selective determination of particular microbial species and genera can be achieved using appropriate media and condition. However, the diversity of bacterial species and strains having similar metabolic and growth patterns and complex mixtures present in real samples limit the potential of their definitive selective determination (other microbiological methods face the same challenges). Also when a mixture of different bacteria is placed in growth promoting medium, faster growing and more robust species can overgrow the others and 'mask' their respiration.

Nonetheless, in a recent study, a panel of 9 common species of aerobic bacteria was investigated by high-throughput micro-respirometry in 384-well plates with 16 partially selective media.¹³⁷ For each medium and bacterial strain, growth profiles were recorded at different dilutions, and standard curves, doubling times and growth patterns in different media determined. Selective, sensitive and rapid (one day experiment) determination of bacteria in pure cultures and simple mixtures was thus demonstrated as a proof of concept. Existing selective tests with other transducers (e.g. Biolumix system with colorimetric pH/CO₂ sensors¹⁴⁴), can be adapted for O₂ sensing, bringing additional benefits.

For comparison, a standard selective microbiological test (ISO method 07.100.30, 4831–4833) usually involves: (1) pre-enrichment (24 h) to resuscitate stressed bacteria and

encourage their growth; (2) selective enrichment (24–48 h) to increase target pathogen numbers to detectable levels and suppress non-target cells; (3) selective and differential plating (24–48 h) on chromogenic agar to identify pathogens of interest; (4) confirmation by biochemical or serological identification. Definitive identification of bacterial strains (*E. coli* O157 and alike) can also be achieved by the molecular methods (immunological and nucleic acid tests), but these require skilled personnel, special equipment, and have higher costs.

Similar to the cell-based assays (see Section 4.4), the respirometric microbial assays also allow multiplexing with other tests. For example, they can be used as part of the pre-enrichment and/or selective enrichment steps, to identify negative samples and exclude them from further testing, or combined with traditional tests with chromogenic metabolic substrates.¹³⁷ Coupling with optical sensors for pH, CO₂ or ammonia, which operate in a similar format as the O₂ sensor¹⁰⁴ can also improve the selectivity of such microbial assays.

Unicellular bacteria can form or be incorporated into organised aggregate structures – biofilms. Microbial biofilms secrete substances supporting their growth, similar to an extracellular matrix in mammalian tissue. Functional properties of microbial cells in the dispersed state and in biofilms are different, and this knowledge is important for many other areas particularly infectious diseases. *Pseudomonas aeruginosa*, a facultative anaerobic bacterium with reduced growth and metabolic activity at low O₂ levels, is almost inescapable in patients with cystic fibrosis and impaired lung function. Creating hypoxic regions in tissue, it survives antibiotic treatments (e.g. tobramycin, ciprofloxacin, and tetracycline) which preferentially kill the physiologically active bacteria living at high oxygen levels.¹⁴⁵ Using O₂ sensing and imaging, *P. aeruginosa* films were investigated as a model of infections in cystic fibrosis.^{146,147} O₂ dynamics was also studied in *Staphylococcus aureus* biofilms by ratiometric measurements of 1 µm silica particles with ruthenium and Nile blue dye, and showed a significant role in organ transplantation, also increased antibiotic resistance.¹⁴⁸

Other applications of microbial micro-respirometry include screening for antimicrobial drugs and drug resistance, development of new growth media (especially selective media for particular pathogens); development and combinatorial testing of nutraceuticals and probiotics (beneficial for human and animal health by promoting microflora and improving healthy living), optimisation and control of treatment methods (processing, sterilisation, preservation) for food, pharmaceutical, environmental and clinical samples, beverages and microbial cultures (see below).

4.5. Food and microbial safety

Food and other perishable products usually degrade rapidly under high O₂ conditions, through oxidation (of lipids, destruction of ascorbic acid), enzymatic reactions (fast ripening, browning of fruit and vegetables), and microbial spoilage by bacteria, yeast and fungi many of which are active consumers of O₂.¹⁴⁹ Microbial spoilage and contamination with common food borne pathogens is of major concern for the food, packaging and retail industries,

customers and society in general. The following pathogens are specifically cited by the legislation: *Listeria monocytogenes*, *Salmonella* spp. (including typhimurium and enteritidis), *E. coli* (indicator of faecal contamination), *Enterobacteriaceae*, *Staphylococci*, *Bacillus cereus*, and zero tolerance limits have been established for some of them.¹⁵⁰ Packaging under modified atmosphere (MAP) with low O₂ is therefore used for many foods. In this context, O₂ levels and dynamics in food packs are important indicators of quality, safety and microbial load of food products.¹⁵¹ Optical O₂ sensing can greatly facilitate such monitoring.

Non-destructive monitoring of O₂ in packs is achieved by incorporating phosphorescent sensors either directly at the packaging line (inserts, stickers, labels^{149,152,153}) or using pre-made packaging materials integrated with sensors. Such 'smart' packs can be interrogated using an optical scanner which reads sensor signals and determines O₂ concentration. Disposable, low-cost solid-state sensors made from food-compatible materials (polymeric composites) are the preferred format for this application. Temperature of the pack also needs to be measured, e.g. using a contactless IR temperature sensor, which is used for correction of optical readings. Measurements in many packs (having different and variable shapes and sizes) can be conducted with one instrument in a high throughput manner, repetitively and without affecting package integrity. This approach has proven its utility for quality and safety assessment of packaged products, including modified atmosphere packaged meats, fish, cheese,¹⁵⁴ green produce, ready-to-eat foods.^{153,155} Solid-state O₂ sensors were also used with beverages to control beer brewing,¹⁵⁶ quality and sensory characteristics of pre- and post-pasteurized lager.¹⁵⁷ These sensor systems enable the identification of faulty packs, problems with packaging materials and processes, manufacturing, storage and transportation issues, batch testing and shelf life studies with a large number of packs.^{149,152} Commercial systems for such applications include Optech[®] Platinum (Mocon-Luxcel), Fibox 3 (PreSens, Germany), OxySense Oxygen Analysers (OxySense, USA), FirestingO2 (Pyro Sciences, Germany). Colorimetric O₂ sensors and 'consume within' indicators¹⁴⁹ derived from them have been developed (Embedded Timer[™], Insignia Technologies), which also have significant potential for food and consumer safety.

Non-destructive assessment of microbial load and growth in food packs is also possible, but has its limitations. Enzymatic and chemical (by scavengers) consumption of O₂, intrinsic respiration of food products (green produce, vegetables), diffusion of atmospheric O₂ across packaging material, volume of headspace should all be taken into consideration as they also affect residual O₂ levels. Due to low O₂, high CO₂ and antimicrobial preservatives in packaged food products, monitoring of microbial respiration is difficult and may result in poor predictability of the actual microbial load and related hazard. Microbial growth on the surface of the product may lead to variability of results. On the other hand, for products packaged at moderate O₂ levels (convenience foods, green produce), residual O₂ levels in sealed packs showed clear correlation with their microbial load and dynamics under different packaging and storage conditions.¹⁴⁹ Knowing these relationships,

non-destructive measurements with a phosphorescent O₂ sensor can be used for assessment of particular products.

However, analysis of microbial safety of foods using sampling methods¹⁵⁸ is much more efficient (though destructive). It allows for more optimal conditions for respirometric measurements: (1) homogenisation of food samples and addition of nutrient-rich media (usually 1:10 w/w) eliminate stress conditions in the pack and promote fast microbial growth; (2) respirometric assay is performed with air-saturated samples (diluted with medium) and at optimal temperature (e.g. 30 °C or 37 °C); (3) monitoring of O₂ depletion in liquid medium under seal (i.e. without headspace). All this ensures rapid, sensitive, reproducible and robust assay with crude food homogenates, with smooth respiration profiles, and accurate determination of microbial load (Total Viable Counts, TVCs).

Such respirometric TVC assays are performed in microtiter plates^{132,158} or disposable vials at constant temperature (30 °C), on a phosphorescent reader which determines the onset/threshold time (TT) of the probe/sensor signal for each sample. TT is the time from the start of the experiment when the phosphorescent signal increases above the basal (air-saturated) level and reaches the pre-set threshold level. TT usually produces a linear relationship with the logarithm of microbial load in the original sample, log(cfu/ml), making it a convenient parameter for quantification. Samples with higher TVCs produce shorter TT values (reciprocal relationship), so that highly contaminated samples can be identified within a few hours in real time. Representative oxygenation profiles (raw data) produced by samples containing aerobic bacteria and processed results are presented in Fig. 8.

The respirometric platform for rapid TVC tests has been shown to work reliably with complex and coloured samples, crude homogenates and different food matrices.¹⁵⁸ It is more stable and robust than alternative turbidimetry (absorbance at 600 nm), bioluminescence (total ATP) or flow cytometry methods.¹⁵⁹ For industrial applications, it is currently provided in several configurations: (1) a high-throughput GreenLight[®] 960 system which uses a soluble O₂ probe, 96/384 well plates and measurement of samples sealed under oil on a standard TR-F plate reader; (2) a medium-throughput GreenLight[®] 930 system operating with disposable plastic vials (2 mL volume, bar-coded) with built-in sensors, and an automated carousel detector/incubator which accommodates up to 48 vials, provides TVC quantification and classification of test samples in real-time; (3) a low-throughput GreenLight[®] 910 system, a simplified version of the 930 system which measures one vial at a time manually. The GreenLight[®] tests (marketed by Mocon and Luxcel Biosciences) have been certified by AOAC/MicroVal for food safety applications – determination of TVCs in raw meat and other products (AOAC licence no. 061002). Such testing can be combined with other quality testing methods used by the industry, including the non-destructive sensing of O₂ and CO₂ in packages. This has been demonstrated with packaged ready to eat salads,¹⁵³ raw fish and some other food products.^{153,154,160}

An ultra-high throughput imaging system was developed by Techno-Path for testing of raw and processed milk. It uses microplates with phosphorescent sensors, operates in a simple

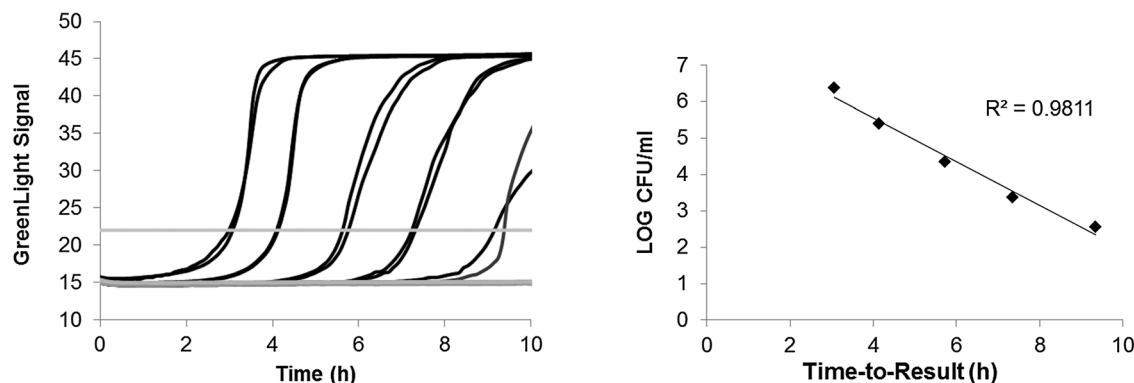


Fig. 8 Oxygenation profiles (left) of food samples spiked with different concentrations of *Bacillus cereus* and the resulting calibration (right). Threshold and air-saturated signal levels (phosphorescence lifetime) are shown with grey lines. Samples were measured on GreenLight[®] platform at 30 °C in duplicates (courtesy of Luxcel Biosciences).

'mix and measure' procedure and can process up to 1000 samples per day.

Theoretical and practical sensitivity of respirometric microbial assays is one viable cell per sample measured.¹⁵⁸ For 96-well plates with 0.1 mL samples diluted 1 : 10 during homogenisation, statistically reliable detection of microbial load in food specimens is 10^2 – 10^3 cfu g⁻¹ and higher. For assays in higher sample volume the sensitivity increases proportionally: for GreenLight[®] 930/910 systems working with 2 mL and 15 mL vials, the limits of detection are approximately 50 and 7 cfu g⁻¹. For homogeneous liquid samples, dilution can be reduced by adding concentrated medium or nutrients directly into the sample, thus improving the sensitivity 10-fold.

High sensitivity, flexibility and other useful features of the respirometric microbial testing make it attractive for other biodetection applications. Thus, surface swabs can be analysed in larger vials and used for hygiene monitoring at workplaces. Detection of microbial warfare (directly or after retention and concentration on air or liquid filters), coliforms in freshwater and groundwater (require sensitivity at 1 cell in a 100 mL sample), Biological Oxygen Demand,¹⁶¹ analysis of wastewater and environmental samples can be conducted. Monitoring systems to protect from terrorist attacks with biological weapons and to detect biological warfare can also be developed.

There is a potential to adapt these platforms for *selective* determination of microbial pathogens in food. As outlined in Section 4.4, such assays still require improvement of selectivity and adaptation to 'real life' conditions and complex samples. Active microbiological research on characteristic metabolic features of common pathogens,¹³⁹ rational design and high-throughput optimisation of new selective media are expected to accelerate this work and bring to use selective respirometric tests for common food borne pathogens.

4.6. Control of cell oxygenation and O₂ gradients in cell cultures

Since Lavoisier's discovery of toxic effects of high O₂ concentrations, numerous studies have been conducted on animals and cell models.^{162,163} At the cellular level, elevated O₂ was shown to have

mutagenic and apoptotic effects, promote tumorigenesis,^{164–168} and affect senescence and proliferation.¹⁶⁹ However, most of routine cell culture is still performed at ambient O₂ (5% CO₂, 19% O₂, 76% N₂), *i.e.* several times higher than physiological normoxia in mammalian tissue. This mismatch imposes stress on the cells through increased ROS generation, reduced levels of antioxidants,¹⁷⁰ altered metabolism and bioenergetics (generation of highly glycolytic cancer cell lines, the Warburg effect, cell transformation – see Section 4.3).

Researchers are now switching to tissue culture in workstations operating at low O₂, special hypoxia chambers and instruments with built-in atmospheric control (*e.g.* plate readers, microscopes). Although more physiological, these systems usually control atmospheric O₂ levels, but not within samples where respiring cells consume O₂ in the surrounding media and act as O₂ sinks. Significant *local* O₂ gradients have been reported for adherent cells grown at 21% O₂,¹⁷¹ and they are more prominent at low O₂ and in larger objects such as 3D spheroids and tissue slices.^{128,172} Micro-gradients can be eliminated by continuous perfusion, rigorous stirring or shaking, culturing cells on gas-permeable membranes facilitating mass exchange,¹⁷³ however this is not always applicable. Reliable *in situ* control of cell oxygenation is highly needed, especially for static cell cultures in flasks and microplates, and this can be achieved with appropriate O₂ probes and sensors.

Extracellular probes are not so efficient, as they get distributed in bulk sample and provide information on average oxygenation.¹⁷⁴ Sensor coatings also have biocompatibility issues so that adherent cells usually grow around the sensor but not on it. Microsensors require micromanipulators to bring them to the cell layer, their fragility and point measurements are significant limitations. On the other hand, cell-penetrating O₂ probes have been successful in providing reliable, accurate and minimally invasive monitoring of oxygenation of cells with minimal toxicity impact on cellular function.¹⁷¹ This application imposes special requirements on the probe, cell loading and measurement strategies, which have been addressed through intensive research. A number of approaches are described here.

Phagocytic cells such as macrophages are known for their ability to ingest particulate matter. This can be exploited with

micro- and nano-particle probes which have no or little intrinsic ability to penetrate cells. Phosphorescent nanoparticles (40 nm Pt-Fluospheres, Invitrogen) were used to measure changes in O_2 during phagocytosis of bacteria,¹⁷⁵ revealing a coordinated inflammatory response of macrophage populations to the invading pathogens. Polystyrene microspheres consumed by RAW264.7 macrophages were used to measure oxygenation and OCR of an individual cell in a microchamber device on a FLIM microscope with gated CCD camera.¹⁷⁶ Although working with phagocytic cells, such loading has moderate/low efficiency, and requires high concentrations ($>100 \mu\text{g ml}^{-1}$) stressing the cells.

For non-phagocytic cells more sophisticated loading strategies are required. Microinjection, gene gun,^{68,177} partial cell permeabilisation (with Ru-dye introduced in esophageal epithelial cells¹⁷⁸) and electroporation have been tested, but these techniques are invasive, damaging to the cell and rather inefficient. Facilitated delivery with transfection reagent (MitoXpress[®]-Xtra probe and EndoPorter[®], respectively) was a significant improvement,⁹¹ however this method is rather slow (loading times 24–28 h), and has moderate efficiency and cell-specificity so that some cell types cannot be loaded and analysed. The new generation of cell-penetrating small molecule and nanoparticle-based probes¹⁸ has greatly improved the situation.

Traditional O_2 sensing materials usually have low intrinsic ability to enter the cells, localise in a predictable manner and stay inside for sufficient time permitting optical measurements without causing significant cyto- and phototoxic insult.¹⁷⁹ Anionic polycarboxylic porphyrins show weak penetration across the cell membrane which bears a negative charge. Hydrophobic dyes have poor solubility in water, bind non-specifically to and migrate across various biological structures and surfaces. Other dyes do penetrate into the cell, but often show toxicity, probably due to membrane or nuclear localisation and interaction with DNA (e.g. cationic porphyrins, Ir- and Ru-dyes^{179,180}). Hence, significant tuning of physical-chemical properties of the phosphorescent structures (hydrophilicity, water solubility, charges, special functional groups) and introduction of a suitable 'delivery vector' are usually required. This can be achieved by rational chemical design based on the knowledge of intracellular transport and trafficking mechanisms for different chemical moieties, development of new dyes and delivery strategies, rigorous experimental testing and optimisation with different cell types or more complex models. A number of such probe structures and studies have been performed in recent years (reviewed in ref. 18). Currently, the core-shell polymeric and cationic nanoparticles impregnated with hydrophobic dyes, and conjugates of hydrophilic Pd and Pt-porphyrin derivatives with peptides and other vectors^{181–183} are the most efficient for loading and sensing O_2 in mammalian cells. Several key factors determine cell loading efficiency: probe concentration, loading time and conditions (temperature, media, additives), mechanism(s) of transport (cell specificity), migration and leakage from the cell, specific brightness and photostability, cytotoxicity and photo-induced toxicity.^{70,181} These parameters need to be carefully investigated for each probe and optimised for the measurements.

The probes such as MitoXpress[®]-Intra, MitoImage[™]-NanO2 and MitoImage[™]-MM2 (Luxcel Biosciences), provide efficient passive loading of cells by simply adding them to the medium at low $\mu\text{g ml}^{-1}$ concentrations, 6–16 h incubation, and subsequent analysis of cells by phosphorescence quenching.^{181,184} Similar to the extracellular probes and sensors used in OCR measurements and micro-respirometry (see Sections 4.1–4.3), cell-penetrating probes based on PtPFPP and PtCP dyes are compatible with available detection instrumentation and can be measured on standard TR-F readers supporting the RLD mode to analyse O_2 in cell populations.⁹¹ Residing in the cell, they can provide information on cellular hypoxia at different times and conditions, and allow manipulation (washing and treatment of cells) and long-term experiments.¹⁸ Other indicator dyes, including the short-decay ruthenium complexes and longwave PtTBP and PdTCPP dyes, are not so compatible with such readers, which rely on xenon flash-lamp excitation (not suitable for detection of lifetimes $<10 \mu\text{s}$) and PMT detectors (poor sensitivity at wavelengths $>700 \text{ nm}$), however instrumentation development is expected to improve the situation.⁴⁵

A new real-time method for measurement of oxygenation of cells and their responses to metabolic stimulation was developed recently which is carried out in an open plate (unsealed samples exposed to the atmosphere).⁹¹ Under static conditions, oxygenation of the cell layer is a function of OCR, hence this can be used to trace and quantify changes in cell respiration. A number of sample parameters also affect the oxygenation, including atmospheric pO_2 , height and viscosity of the medium layer (diffusion barrier for ambient O_2), temperature, and sample geometry. However, these parameters are easy to keep constant or use them to make the optical response more easy to detect.¹⁷¹ In this high-throughput metabolic assay, pre-treatment of cells and multiple additions of effectors (small aliquots of drugs) during the monitoring are possible, as shown in Fig. 9.

The cell-permeable and biocompatible O_2 probes with enhanced photostability, brightness, reference and light antenna moieties for multi-modal detection (e.g. MitoImage[™] NanO2, MitoImage[™] MM2 and some others¹⁸) have enabled O_2 analyses on live-cell fluorescence imaging and high-content screening platforms, on a scale of a single cell. High-resolution O_2 imaging facilitates the analysis of heterogeneous samples,

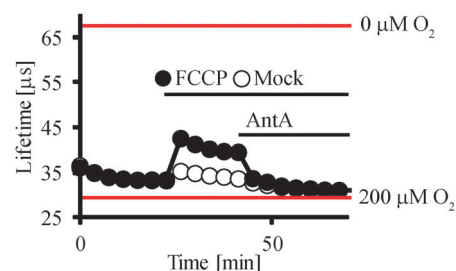


Fig. 9 Oxygenation profiles of MEF cells monitored with cell-penetrating probe MitoXpress[®]-Intra. After the initial monitoring of resting cells, uncoupler (FCCP at $1 \mu\text{M}$) or mock ($0 \mu\text{M}$) were added at 25 min, followed by the addition of inhibitor AntA at 40 min. Oxygenation conditions are presented in phosphorescence lifetime scale, the range of lifetime and O_2 values is indicated with red lines.

differential and asynchronous responses, development and differentiation of stem cells, and transformation of cancerous and diseased cells. It can also be used with 3D cell models, artificial tissue, organs and *in vivo* (see Sections 4.7 and 4.8). Modern imaging systems supporting ratiometric intensity, microsecond FLIM and/or two-photon excitation modes and capable of providing a quantitative readout of O₂ concentration in 2D, 3D and 4D are now widely available.^{51,70,75,80}

Using the intracellular O₂ sensing technique with neuronal cancer PC12 cells experiencing local hypoxia, activation of the HIF pathway by intrinsic respiration under ambient pO₂ (19–21%) was demonstrated.¹⁷¹ In another study, a transient activation of OxPhos in PC12 cells upon depletion of extracellular Ca²⁺ was revealed (not detectable by traditional OCR measurements), and by applying different pharmacological treatments and probes the detailed mechanism of this response was elaborated.¹⁸⁵ Similarly, analysis of the mode of action of the common V-ATPase inhibitor, Bafilomycin A1, revealed its previously unknown uncoupling effect on OxPhos and flickering depolarization of the mitochondria.¹¹² Mild pharmacological uncoupling with bafilomycin was seen to stabilise HIF- α protein(s) under physiological conditions and induce adaptive responses in colon cancer cell lines.¹⁸⁶ Long-term exposure of neuronal cells to hypoxia (3% O₂ for 30 days) revealed different levels of intra and extracellular O₂ in PC12 and SH-SY5Y cell lines: after adaptation to hypoxic conditions PC12 cells with lower O₂ showed reduced HIF- α signalling.¹¹⁷ In the latter study, headspace O₂ was monitored using a solid-state OpTech™ sensor (Mocon), along with cellular O₂ analysis.

Due to active consumption of O₂ in the mitochondria and its supply by passive diffusion inside the cell, 'intracellular' O₂ gradients may develop and play physiological roles. Probed using different cell models and techniques, the existence of such gradients under physiological conditions is still controversial and hotly debated.^{187–189} Several groups attempted to measure or calculate such gradients, and reported them to be extremely small, below the resolution of current experimental techniques. Thus, in static 2D culture of endothelial cells and the O₂ probe delivered by endocytosis, intracellular O₂ levels were equal to extracellular levels, *i.e.* no gradient.⁵⁹ Using the mitochondria-targeted GFP sensor construct, O₂ gradients were also undetectable in Hep3B cells.¹⁹⁰ In contrast, other studies reported the existence of dynamic gradients and anoxic cores in cardiomyocytes.^{188,189} Intracellular O₂ probes were used to measure O₂ gradients in lung fibroblasts and neuroblastoma cells which were found to be 1–3 mmHg.³⁷ In 2D culture of mouse fibroblast cells with active respiration, the gradient between the plasma membrane and intracellular space measured on a TR-F reader was reported to be $\sim 3 \mu\text{M}$ (at 9% pO₂ or 90 μM).¹⁷⁴ Although these values are close to the measurement error, it can be hypothesized that in cells with active OxPhos, large size and shape resembling their native 3D environments, O₂ gradients under physiological conditions can be significant, while in small cells with low basal respiration (*e.g.* endothelial cells⁵⁹) they hardly exist.

The above examples demonstrate the utility and application potential of O₂ sensing techniques based on cell-penetrating

probes. Still better probes are actively sought which can provide more selective, controlled distribution in the cells (cell surface, mitochondria, cytosolic compartments), low toxicity, high brightness, operational and storage stability and convenient use. Active work is on-going, particularly with porphyrin dyes¹⁹¹ and nanoparticle structures, and new studies are published more and more frequently clarifying important biological questions.

4.7. Tissue models: spheroids, cell aggregates and tissue explants

Cell models which mimic the conditions in 3D tissue include multi-cellular spheroids, cells grown in 3D scaffolds, tissue slices, organ explants and artificial tissue.¹²⁸ Multicellular spheroids and cell aggregates¹⁰⁷ are common models in drug screening, toxicity assessment and developmental biology,^{128,172} showing *in vivo*-like 3D micro-environment, cell to cell communications, diffusion limited supply of nutrients, pH, CO₂, and particularly O₂. Mapping of O₂ in such models is important, and phosphorescence O₂ sensing techniques are well-suited for that.

Cell-penetrating probes are particularly useful, they can be loaded in spheroid structures and/or cells composing them, and retained for prolonged time. Quenched-phosphorescence O₂ sensing enabled us to demonstrate limited O₂ availability and formation of necrotic core in cancer cell spheroids. Oxygenation of spheroids from primary rat neurons stained with the MitoImage™ probe was visualised by ratiometric intensity imaging under two-photon excitation.¹⁹²

In-depth analysis of objects with sizes >100 μm faces many challenges, particularly for loading with probes and imaging using optical techniques. Using different types of cell-penetrating probes, such conditions were optimised for spheroid cultures of neuronal cells (neurospheres). This method has allowed real-time monitoring of oxygenation of the whole spheroid, and multiplexing with a number of cell biomarkers and fluorescent probes (for intracellular pH, Ca²⁺, membrane potentials, cell surface markers).¹⁹³ Representative images are shown in Fig. 10.

The so-called *ex vivo* models, which include excised tissue slices, biopsies and perfused organs, are used in tumour and developmental biology, studies of cell adhesion, migration, epithelial morphogenesis, and drug testing.¹²⁸ Such specimens retain natural 3D organisation and cell-cell interactions, but lack normal O₂ supply by vasculature. They are normally maintained in a 'carbogen' atmosphere (95% O₂, 5% CO₂) to avoid anoxia and cell death. To preserve optimal functional characteristics and viability of such samples, more reliable and accurate control of O₂ and its dynamics is necessary.

Skeletal muscle fibres isolated from frog and microinjected with the Oxyphor™ probe were studied on a custom-built wide-field camera based phosphorescent microscope working in a frequency domain.⁶¹ Dynamics of intracellular O₂ during muscle contraction¹⁹⁴ and at different contraction frequencies,¹⁹⁵ the relationship between aerobic metabolism and fatigue,¹⁹⁶ temperature dependence of intracellular pO₂¹⁹⁷ in muscle fibres were also investigated by this method. Penetration of the PdTCPP probe into the interstitial space of rat spinotrapezius muscles upon topical application allowed the study of average O₂ consumption¹⁹⁸ and

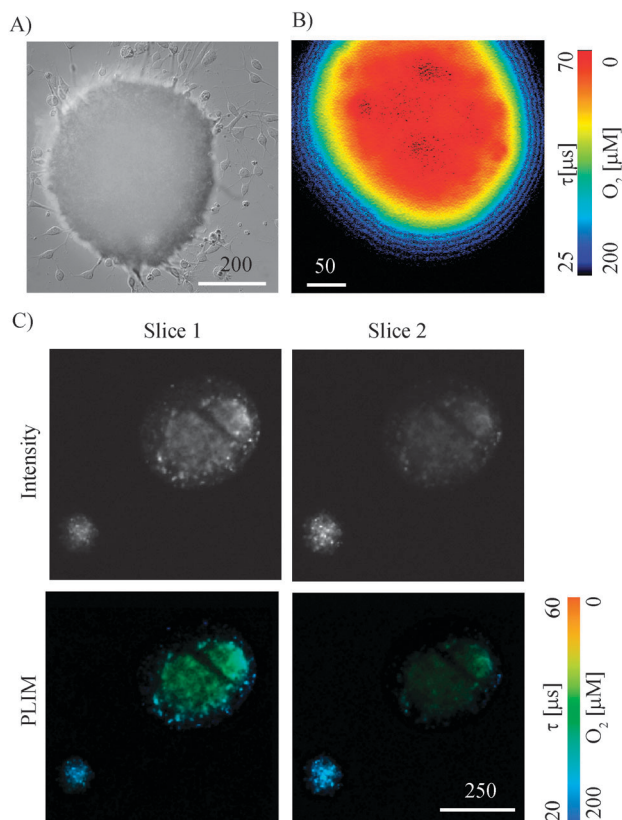


Fig. 10 Phosphorescence intensity (loading) and lifetime (O_2 concentration) images for neurospheres produced by microsecond FLIM microscopy and cell-penetrating O_2 probes. (A) Differential interference contrast image of a proliferating neurosphere; (B) widefield FLIM image of a neurosphere stained with MitolImage™ NanoO2 probe; (C) optical sections (10 μm each) of confocal FLIM-TCSPC and intensity images of neurosphere stained with PtPFPP conjugate. The false-colour FLIM images provide information on the O_2 levels across the spheroid. Scale bar is in μm .

dependence of oxygen availability in skeletal muscle,¹⁹⁹ revealing a sigmoid dependence of VO_2 on pO_2 .²⁰⁰

Carotid body is a specialised organ which senses changes in arterial oxygen and regulates respiration and lungs function. *Ex vivo* analysis of type I oxygen sensing cells was conducted by dynamic multifocal microscopy imaging of intracellular O_2 (with PEPP0²⁰¹ probe), together with Ca^{2+} and membrane potential, revealing the heterogeneous responses of different cells in the carotid body.²⁰²

Respiration and physiology of salivary glands isolated from an insect (blowfly *Calliphora vicina*) were studied by injecting the PtPFPP-based 300 nm polystyrene microparticles and lifetime-based detection of O_2 by phase fluorometry on a custom-made microscope.^{203,204} Using this set-up, authors also performed quantitative imaging of intracellular pH and extracellular O_2 in this tissue under resting and stimulated conditions and demonstrated the regulation of intracellular pH by V-ATPase and carbonic anhydrase.²⁰⁴

Cell-impermeable O_2 sensors were also applied for mapping O_2 in 3D cell cultures, including the PDMS microparticles (5–40 μm) with Ru-dye and intensity imaging of hydrogel scaffolds,²⁰⁵ silica

beads with Ru-dye to image oxygenation of chondrogenic cells in aggregates and spheroids.²⁰⁶ More recently, ratiometric Ru-dye based nanosensors with Nile Blue²⁰⁷ or coumarin 540⁶⁹ reference dyes were produced and used for O_2 imaging in 3D culture of colon intestinal cancer cells and the micro-fluidic tumor model by fluorescence microscopy. Fibre-optic O_2 sensors were applied to measure O_2 in porcine muscle in a tissue perfusion system.²⁰⁸ OCR and viability measurements in whole porcine kidneys allowed differentiation of healthy and damaged organs, thus providing a useful quality control tool for organ transplantation.²⁰⁹ Unfortunately, in some of these studies, qualitative rather than quantitative O_2 analysis was conducted, which reduces their value.

New materials for tissue engineering and organ transplantation must be analysed for their biocompatibility and functional properties. A Ru-based solid-state O_2 sensor was applied to study O_2 diffusion across pulmonary surfactant layers to reveal that O_2 diffusion in surfactant layers is significantly higher than in the water phase.²¹⁰

4.8. *In vivo* and *ex vivo* measurements

O_2 levels in different tissues and cells of our body normally range 0.5–10%.^{8,211} They are highest in the lungs from which O_2 is supplied to other tissues by the blood through a network of vessels and micro-capillaries. O_2 demand is determined by the metabolic state and respiratory activity of tissue (mainly OxPhos and mitochondrial function), while supply – by blood flow, oxygenation (influenced by many biochemical parameters), O_2 diffusion and steady-state O_2 gradients within tissue. O_2 distribution and dynamics are specific for each tissue and critical for their physiological functions.^{188,212–219} Normal homeostatic conditions *in vivo* are called *physiological* normoxia. Hypoxia and anoxia denote substantially reduced and zero levels of O_2 . These abnormal conditions lead to pathological states, development of disease and ultimately death.

Quenched phosphorescence methods have long been applied to *in vivo* measurements of O_2 .^{6,217,220} We mainly focus on the new applications with *quantitative* O_2 measurements and corresponding methodologies revealing novel biological results. The non-biological and qualitative studies, as well as alternative hypoxia-specific probes (Table 1), are omitted.

High intrinsic absorption and scattering of animal tissue demand probes with longwave or multi-photon excitation and emission in the range 600–900 nm, with high brightness and photostability for high-resolution 3D imaging. Extracellular, cell-impermeable probes injected intravenously in live animals are most commonly used. These probes are required to be inert, have good biocompatibility and distribution in the tissue of interest, long clearance times, low accumulation in organs and systemic toxicity. Recent modelling demonstrated that for microscopic and high resolution measurements, the phosphors do not have to be confined in the region of interest and can be distributed throughout the volume.²²¹

Probes based on PdTCPP and Pd-TBP dendrimers (Oxyphor™ family¹⁸ from Oxygen Enterprises) are most commonly used. They have moderate solubility and inertness, optimal spectral

characteristics and sensitivity to O₂, distribution and retention in the body, and are seemingly non-toxic. Initially they were used in low-resolution point and 2D measurements under one-photon excitation. The new structures, such as PtP-C343 probe with light-harvesting antennae and PEGylated shell, have enabled high-resolution, 3D and multiphoton O₂ imaging with excellent analytical performance⁷⁵ (though they are difficult to synthesise, expensive to use and require specialised set-up – two-photon microsecond FLIM).

These intravital O₂ probes were applied to study cerebral vasculature, neuronal activity,^{73,222–225} tumor oxygenation,²²⁶ microcirculation,²¹⁷ muscle physiology,^{194,197,227} for mapping O₂ tension in retina²²⁸ and tomographic imaging at the macroscopic level.²²⁹ Topical application of the PdTCPP probe and one-photon time-gated phosphorescence quenching microscopy^{62,228,230,231} were used in *ex vivo* studies of pO₂ near arterioles and oxygen consumption in rat mesentery,²³⁰ where no difference in pO₂ profiles between small and large arterioles and similar pO₂ values between interstitial and intravascular compartments were observed. O₂ loss from mesenteric arterioles was also shown to be moderate and close to theoretical predictions and results produced by other groups.²³² These applications are reviewed in more detail in ref. 217.

Point measurements of O₂ were also used. Thus, *in vivo* function of AMP-activated kinase, a master regulator of cell bioenergetics, is poorly understood. It was examined using the Oxyphor R2 probe and frequency-domain phosphorimetry. Analysis of microvascular dynamics of pO₂ during contractions of mouse muscles demonstrated that O₂ delivery is affected by AMPK activation.²²⁷ Similarly, myocardial function was studied directly in the beating heart using the Oxyphor G3 probe injected in three myocardial areas.²⁰⁰ Several set-ups were proposed for pO₂ mapping in retina,^{228,233–236} and optimisation measurements to avoid vascular damage and degeneration of the optic nerve.²³⁷ Modulation of vascular pO₂ in retina by light was demonstrated,²³⁵ and used in the concept of wireless intraocular microrobots.²³⁸

Hypoxia is one of the hallmarks of cancer,²³⁹ it contributes to tumour aggressiveness and metastatic potential, and efficiency of anti-cancer therapy. Tumour oxygenation was long thought to be regulated by diffusion, however, phosphorescent imaging with the Oxyphor G2 probe has revealed that various tumours (fibrosarcomas, carcinomas and gliomas) display dynamic patterns of oxygenation with spatial and temporal heterogeneity.²²⁶ New Oxyphor R4 and G4 probes, which can operate under protein-rich conditions such as blood plasma and interstitial space and have relatively long retention times (hours),⁶⁴ also demonstrated increased accumulation in tumours. This can be explained by the higher permeability and the retention effect of tumour tissue.²⁴⁰ Tumor targeting of O₂ probes (e.g. with antibodies or other 'vectors') can also facilitate imaging of tumours.²⁴¹ Imaging of mouse tumours with Oxyphor G4 on a widefield microscope showed that anaesthesia with isoflurane affects tissue oxygenation.⁷⁴

The most advanced probe of this type, PtP-C343 ($\tau_0 = 60 \mu\text{s}$, $k_Q = 150 \text{ mmHg}^{-1} \text{ s}^{-1}$),^{59,60} was used to measure oxygenation of vasculature in live animal brain by two-photon microsecond

FLIM and analyse function of regions of the brain in 3D.^{59,242} Simultaneous imaging of blood flow and oxygenation in rat cerebellum with penetration depth of up to 300 μm and spatial resolution of $\sim 1 \mu\text{m}$ was demonstrated,²²⁴ and used to create a biophysical model of multicompartment in cerebral vasculature.^{223,243} Analysis of vasculature and tissue oxygenation in rat somatosensory cortex under resting and stimulated conditions showed that stimulus-evoked increase of tissue pO₂ is dependent on the baseline O₂ levels.²²⁰ pO₂ in brain vasculature and Ca²⁺ dynamics during neuronal activation by inhalation of an odorant,²²⁴ and monitoring of pO₂ transients in capillaries and local neuronal activity in rodent brain were studied.²²⁵ Amazingly, two-photon microscopy also revealed local O₂ gradients around individual erythrocytes in blood microvessels.²²⁵

Planar sensors (PtOEP dye in polystyrene on a polyester support) were used to image O₂ in the dorsal skinfold chamber model with amelanotic melanoma,²⁴⁴ confirming lower mean O₂ values in tumour regions than in normal skin. The sensor data were compared with the electrode method²⁴⁵ which produced higher O₂ values for capillary blood samples, and this was explained by O₂ consumption by the electrode. Similarly, ischemia-reperfusion experiments with sensor film on forearm demonstrated the anticipated fluctuations of O₂ in microcirculation.²⁴⁵ Sensor foil was also proposed for the study of skin cancer development, wound healing,^{214,246} and the role of stratum corneum in epidermal oxygen barrier.²⁴⁶

Planar sensor foil was applied to monitor oxygenation of rat brain cortex^{247,248} (qualitatively, using a light-conducting PMMA cylinder, ring of 405 nm LEDs and a small CCD camera assembled on animal head), detection of radial O₂ gradients across cortex arterioles.^{247,248} Another ratiometric planar film sensor was applied to monitor brain activity and also proposed for use in conjunction with electrophysiology to monitor other analytes in interstitial space during spontaneous seizure-like events.²⁴⁹ Similarly, progression of trauma patients towards hemorrhagic shock was monitored *via* oxygenation of the peripheral tissue of model animals (piglets) using a needle-type microsensor containing a solution Oxyphor G3 at its tip.²⁵⁰

A new cell-penetrating probe, NanO2-IR, was applied topically to quickly stain rat brain neocortex, and then monitor through a cranial window responses to peripheral sensory stimulation (whisker model) in the live animal. Dynamics of oxygenation in selected regions of brain tissue was monitored on a standard animal imager in 2D, and correlated with the data produced by a fluorescent potentiometric probe revealing considerable differences.²⁵¹

A few other phosphors were proposed for *in vivo* O₂ imaging. Ir(btp)₂(acac) bound to albumin was applied for semi-quantitative intensity-based O₂ imaging in rodent tumour tissue.^{182,252} This probe showed broad distribution in tissues and efficient internalisation, it was later modified with coumarin C343 to enable ratiometric imaging²⁵³ (though its photostability may be an issue). Similarly, Ru-complexes were proposed for imaging of tissue hypoxia.²⁵⁴

O₂ sensing by quenching of delayed fluorescence of endogenous protoporphyrin IX (PPIX) was described. The procedure

involves administration of 5-aminolevulinic acid – a precursor in PPIX biosynthesis, which increases the levels of mitochondrial PPIX (this approach is used in photodynamic tumour therapy, PDT²⁵⁵), and detection of red emission of PPIX sensitive to O₂.^{37,256–260} Known localisation of PPIX enables to probe mitochondrial/intracellular O₂.³⁷ The drawbacks of this method are poor photostability and brightness of PPIX, phototoxicity and physiological consequences of PPIX overload.²⁶¹ Point measurements with an optical fibre (1000 µm) and 510 nm excitation have limited spatial resolution and in-depth penetration, giving average pO₂ values, but they are well-suited for O₂ measurements on the skin surface. Measurements of rat skin produced O₂ values were comparable to those for skeletal muscle.²⁵⁸ Analysis of perfused heart demonstrated that mitochondrial O₂ levels *in vivo* are higher and more heterogeneous than were previously thought.²⁵⁹ O₂ monitoring in perfused liver²⁶⁰ and in tumor tissue undergoing PDT²⁶² (which depends on the available O₂²⁶¹) were also reported. As a model of vascular damage in PDT, the chorioallantoic membrane of chicken egg demonstrated good correlation between the damage during PDT and tissue hypoxia.²⁶²

In the future, increased use of such materials and sensing techniques, and development of new probes with improved biocompatibility, brightness and targeting properties, suitable for high-resolution, quantitative O₂ imaging and compatible with standard detection platforms such as multiphoton microscopes and animal imagers are envisaged.

4.9. O₂ in plants

Plants produce O₂ by photosynthesis and consume by respiration, these light-dependent and ‘dark’ metabolic pathways are coupled and cross-regulated. Evolved from aquatic photosynthetic organisms, plants also have specialised mechanisms to regulate metabolism and cope with low O₂ availability (hypoxia is a common condition in plant tissues, seeds and roots during soil flooding⁹). Understanding of this regulation is important for basic plant science and agriculture.

Plants possess a rigid carbohydrate cell wall which makes it rather difficult to interrogate plant tissue and conduct O₂ measurements. This natural barrier can be overcome by using needle-type microsensors, microinjection, or delivery of soluble O₂ probes through the root and circulating fluids. Strong autofluorescence (chlorophyll in leaves, carotene and other pigments) can interfere with sensor signals, particularly in intensity and phase measurements. The rooted zone of plants, called the rhizosphere, is a dynamic system which includes symbiotic bacteria, nematodes and other organisms, regulates nutrient storage, nitrogen fixation, biosynthesis pathways of the plant, and the global cycle of biogenic elements and chemicals.

Similar to animal cells and tissues, phosphorescent sensors are useful for bioenergetic studies of plant compartments.²⁶³ Potato tubers with active metabolism and diffusion-limited supply of O₂ creating hypoxic conditions are popular models. During their lifespan tubers experience different environmental conditions, from deep hypoxia in flooded soil to 21% O₂ during storage. Regulation of sugar metabolism by O₂ during tuber

development was measured with an O₂ micro-sensor to elaborate the roles of pyruvate and alternative oxidase.²⁶⁴ Similar to animal cells, plants contain hypoxia-responsive elements, particularly the ethylene-response factor (ERF) which regulates a spectrum of genes during tuber development.⁹ O₂-dependent expression of these genes in different compartments was investigated using O₂ microsensors.^{265,266} A micromanipulator allowed microinjections in cells and in-depth measurement. Similarly, hypoxic regulation of metabolism in the roots of pea and *Arabidopsis* and dynamics of O₂ gradients in the root zone upon stimulation with pyruvate were analysed.²⁶⁷

A simple imaging system, VisiSens (PreSens) operating with planar sensor foil, was used to study oxygenation of plant surface and underlying vasculature.^{268,269} Providing 20 µm spatial resolution, it was applied to seed, leaf, stem sections and the rhizosphere of *Cabomba caroliniana* (water plant), *Tradescantia fluminensis* (ground plant).²⁶⁸ Photosynthetic O₂ production and consumption by respiration under alternating light and dark conditions, upon inhibition of photosynthesis by 3-(3,4-dichlorophenyl)-1,1-dimethylurea,²⁶⁹ and in plants infected by bacteria were also studied.

Exchange of gases, particularly O₂, plays important roles in ripening of fruits and their behaviour under ambient and preservation conditions. Optical O₂ sensing was used to monitor soft fruits and vegetables and their surroundings. Hypoxia in tomato fruits was probed with a fiber-optic microsensor.²⁷⁰

Analysis of the quality and germination potential of plant and vegetable seeds is another important application.²⁷¹ This can be performed in dedicated micro-well plates with O₂ sensors. One seed with a small amount of water is placed in each well, and then a uniform gas-tight sealing of the wells is applied followed by the monitoring of seed respiration (several hours per day). The gas-liquid system is optimised to activate metabolism and germination of the seed, but not to ‘flood’ or suffocate it. Rational assay design (plate material, seal, volume ratio of the well/added liquid/seed, respiration activity) enabled us to achieve reproducible results. Fig. 11 shows typical respiration profiles, with a robust change of metabolism from anaerobic to aerobic for the viable but not dead seeds, and their different germination potential. After the non-invasive testing, healthy seeds can be extracted by puncturing the seal and planted. This helps to avoid planting of non-viable material, optimise storage conditions and treatments, test efficiency of herbicides and other factors on the quality of elite seed material. Commercial systems are being developed by Astec Global and other companies.

A solid-state PtPFPP based sensor coupled with neutron radiography was applied to monitor O₂ and H₂O in root zone during night and day phases,²⁷² providing a powerful platform for plant biochemistry and metabolism.

4.10. Control of O₂ in fluidic systems, cell based biochips and bioreactors

Perfusion chambers, microfluidic devices and biochips allow rapid, miniaturised and parallel analyses of test cells and tissue, with exposure to various conditions. These systems

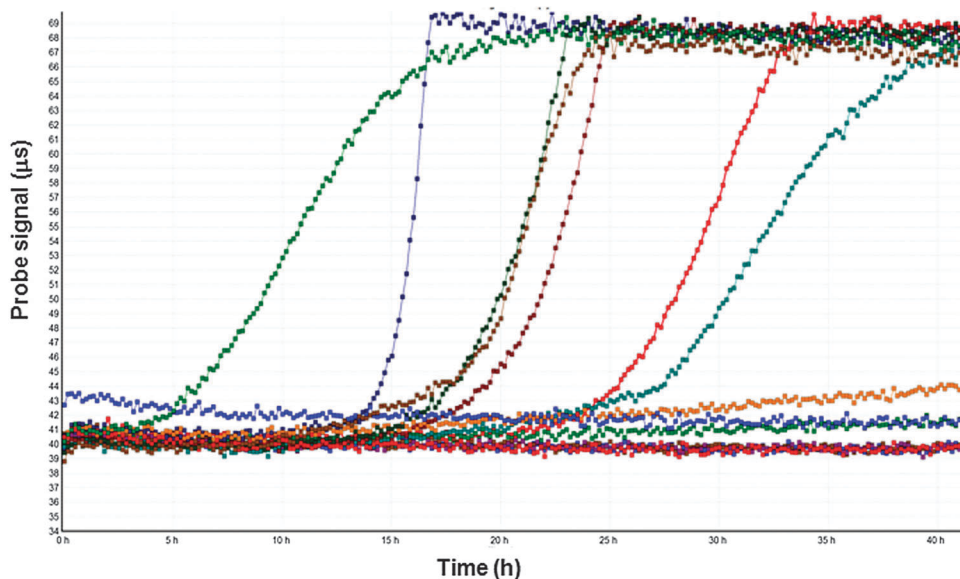


Fig. 11 Assessment of viability of individual seeds by O_2 micro-respirometry (courtesy of Luxcel Biosciences).

can provide large amounts of data, save valuable biomaterials, time and costs,¹⁷³ but are also facing many challenges. To produce plausible biological data and interpret the results correctly, reliable control of oxygenation conditions is required for such systems, to avoid cell stress by transient or sustained hypoxia, anoxia or hyperoxia. Even in macroscopic cultures, respiring samples face O_2 limitation (see Section 4.6), altered cell function or death.¹⁷² Stable supply of O_2 can be achieved by dynamic flow conditions,^{146,173} but small dimensions of such devices, the need for servicing and priming (particularly seeding the cells and letting them to adhere to the surface), require stopped flow conditions which may lead to fast anoxia.¹⁷³ This can be overcome by using biochip materials highly permeable to O_2 such as PDMS.²⁷³ Biochips integrated with O_2 probes or solid-state sensors^{173,273–275} and tailored to O_2 sensing and imaging experiments^{82,146,173} were also described. Control of cell oxygenation combined with ‘metabolomics-on-a-chip’ analysis using 1H -NMR provides a powerful tool for studies of tissue function.²⁷⁶

New ways to manipulate using O_2 concentration and regulate cellular function can also be implemented. Thus, O_2 gradients can be produced by mixing gases inside the PDMS chips controlled with an O_2 sensor,^{277,278} or by chemical consumption O_2 in channels separated by a gas-diffusion membrane.^{147,279} The effect of O_2 availability on the action of an anti-cancer drug on cells was studied.²⁷⁹ A microfluidic device with polycarbonate coating (to minimise back diffusion of O_2 through PDMS) was applied to study hypoxia-dependent migration of breast cancer cells in 3D collagen scaffolds,²⁸⁰ showing that $\sim 3.9\%$ hypoxia (compared to 8.5% O_2 in normal breast tissue) promoted directional motility of cancer cells. Biochips with O_2 sensors allowed modelling of O_2 gradients and shear stress, and design of novel bioreactors and 3D scaffolds.^{275,281} Fluidic inserts with Ru-based sensors were used to modulate O_2 levels and study heterogeneity of adherent cells in standard microtitre plates.²⁸²

Many existing microfluidic devices can be adapted for O_2 sensing with soluble O_2 probes. Small-size microchannel based devices such as μ -slides (Ibidi) provide high sensitivity in OCR measurements under static conditions, cell perfusion and multiple treatments, and multiplexed detection of other biomarkers (fluorescent probes, immunostaining) to study drug effects on cell bioenergetics.²⁸³

O_2 sensors have been used in macro-scale bioreactors to monitor dissolved and headspace O_2 during fermentations with microbial cells.²⁸⁴ High rates of O_2 utilisation make O_2 availability one of the main limiting factors, and feedback control through optimal aeration and stirring provides faster and higher yield of target products and reduces energy and costs. Conversely, stress of the cells due to O_2 limitation results in reduced biomass and/or expression of target products (proteins or metabolites).

In array-type micro-bioreactors O_2 monitoring is also necessary. Customised sensor systems, coated microplates and SensorDish reader (PreSens)²⁸⁵ were applied to study cultivation conditions and media with yeast,^{286,287} bacterial, mammalian and plant cell cultures.^{288,289,290}

Design of new biofermentation systems and micro-bioreactors provides good scope for soluble O_2 probes, *e.g.* monitoring of yeast fermentation with ratiometric nanoparticle sensors.²⁹¹ Magnetic microparticle sensors manipulated inside the fermenter were also demonstrated with bacterial cultures.⁶⁵

Generally, macroscopic solid-state sensors with an integrated optoelectronic module or optical fibre connection mounted on the reactor vessel can be used in fermentations performed on a large scale, with bioproducts intended for human consumption (for which even minute contamination with sensor material is highly undesirable – regulated by FDA). Soluble O_2 probes are better suited for laboratory systems, process optimisation and pilot studies performed on a small scale in a combinatorial and high throughput format.

4.11. Respiration of whole organisms, biochemical and environmental toxicology

Total respiration of an organism is a useful biochemical indicator of its development, metabolism, adaptation to changing conditions.²⁹² Toxicological assessment of chemical, biological and environmental samples using such models is a growing research area. Small aquatic organisms such as *Daphnia magna*, *Artemia salina* and *Danio rerio* (zebrafish), as well as soil worms *C. elegans*¹⁴² are convenient models for acute toxicity testing. They have been applied to analyse heavy metal ions, pesticides, environmental samples such as fresh, sea and wastewater, soil by optical micro-respirometry, showing superiority to traditional toxicity assays based on mortality or morbidity assessment of model organisms.^{293,294} OCR is a generic and sensitive marker of toxicity detecting sub-lethal effects *via* both inhibition or activation respiration. The format described for cells (standard microplate, MitoXpress[®]-Xtra probe, TR-F reader detection) is also applicable to these organisms, allowing the generation of dose-response curves and patterns of toxicity (on a panel of different organisms) and processing of large number of samples in an automated fashion. This is an attractive alternative to toxicity testing using higher animals (fish, mammals).

Fatty acid metabolism is important in the adaptation of fishes to water composition and is related to mitochondrial function. It was analysed in developing dover sole fish (*Solea solea*) under hypoxia with a solid-state O₂ sensor,²⁹⁵ revealing that depending on the diet, larvae have different metabolic rates, tolerance to hypoxia and metamorphosis rate. Spatio-temporal dynamics of O₂ at the surface of corals was probed with specially designed magnetic sensor microparticles.²⁹⁶ Cyanobacteria *Synechocystis* sp. PCC6803 monitored using a planar O₂ sensor showed a higher photosynthetic activity in the exponential phase than in the stationary phase,²⁹⁷ which can be explored in biofuel production by direct photoconversion. Biodegradation of many hydrocarbons (*e.g.* chlorobenzene by 1,2-dioxygenase enzymes) is highly dependent on O₂. Using O₂ sensors, microaerophilic organisms living at reduced O₂ levels (<60 μM O₂) were shown to have potential for industrial biodegradation processes.²⁹⁸ A fiber-optic O₂ minisensor was applied in long-term (450 h) batch-monitoring of dissolved organic carbon biodegradation.²⁹⁹

Environmental O₂ affects behaviour and migration of many soil residents such as *C. elegans* nematodes.³⁰⁰ The NanO2 probe and wide-field microsecond FLIM microscopy were used to study O₂-dependent behaviour of the worms fed with *E. coli* on agar plates. Under ambient air conditions the worms were seen to gather in clumps where they create low O₂ (0–5%) conditions.⁷⁰

4.12. Trace and low oxygen analysis in marine, aquatic, soil and environmental biology

Since the emergence of life on the planet until present days, O₂ levels in the atmosphere, ocean and the lithosphere have been closely connected. Starting from very low levels (<0.01% v/v),³⁰¹ O₂ pollution by evolved photosynthetic cyanobacteria and plants,

has brought us to current levels of 20.86% O₂ in the atmosphere. This transition was accompanied by a major change in living systems from anaerobic to aerobic metabolism.³⁰² Anaerobic and deeply hypoxic environments still exist in many parts (deep biosphere) and are populated with primitive organisms. They often host large communities of different species (anaerobes, microaerophiles and others^{1,2}). Functional organisation and metabolism in these niches with living matter are poorly understood, hard to access or reproduce under laboratory conditions, and their roles in the emergence of carbon-based life and higher organisms (studied much better) remain elusive.

The challenges here are the precise control of very low O₂ levels and respiration rates.^{130,302} Polarographic (electrode-based) and many other methods become unusable for this. Standard O₂ sensors can work reliably at 0.05% O₂, the phosphorescent method can be easily adapted for much lower ranges. There are many phosphors with long emission lifetimes (>500 μs) and polymeric matrices providing high K_{S-V}. Thus, the Fibox 3 trace system (PreSens) based on the PtPFPP dye in a fluorinated polymer,³⁰³ PtTPTBPF based microsensors³⁰⁴ and polymeric films of fullerene C₇₀^{305,306} were suggested for trace O₂ analysis.

The cryosphere, which includes glaciers, ice sheets, and Antarctic sub-glacial lakes, is an extreme environment which also hosts communities of microorganisms, with O₂ concentration determined by the balance between photosynthesis, respiration and chemosynthetic processes such as sulphide oxidation. Detection of O₂ in the cryosphere using optical sensors has been demonstrated,³⁰⁶ and optimisation of sensor design resulted in fast response times, high sensitivity, minimal temperature dependence and signal drift.³⁰³

Open ocean covers 92% of the sea floor and has an average depth of >200 m. Photosynthesis in such an environment is determined by light penetration, available nutrients and substrates, and is limited to the 'photic zone' of ~100 m. In deep ocean suboxic (<4.5 μM O₂ and <1 μM H₂S) and anoxic zones are common, but they are stratified.³⁰⁷ Such zones occupy only ~1% of global ocean, but contribute to about 1/3 of the loss of nitrogen in the ocean *via* denitrification.³⁰⁸ For deep sea benthic layers, specialised sensor systems have been developed which measure O₂ consumption in multiple sites on the sea floor.^{309–311} These sensor systems are usually optimised for long-term measurements, autonomous or cabled and coupled with other measurement devices (*e.g.* photocamera, acoustic, sedimentation sensors). Analysis of sub-mm niches with reducing conditions in benthic fauna require high resolution imaging of O₂ (also CO₂ and pH).³¹² It was hypothesized that in heterogeneous sediments such micro-niches play significant roles in the accumulation of trace metals, scavenging and fossilisation. Here needle-type and planar O₂ sensors demonstrated high stability, quick response times, non-invasiveness and imaging capabilities.³¹² Film sensors were also used to probe underground water capillary fringes.³¹³

In marine biology, macro-methods allow monitoring of net biological production, air-sea O₂ exchange, salinity, geological processes. Seasonal factors and O₂ availability regulate the life of aquatic communities of protozoa and higher organisms,

which requires investigation.² Traditional electrode-based systems³¹⁴ are now complemented by optical O₂ sensors. Fibre-optic O₂ and CO₂ sensors (e.g. Aanderaa, Ocean Optics, PreSens), allow the analysis of marine environments, discrete colonies and individual (micro)organisms on a micro-scale. A dedicated autonomous FerryBox system^{315,316} allows monitoring for >1 year, is insensitive to temperature and salinity, flexible and affordable.^{307,317} A dual O₂–CO₂ sensor for aquatic systems is also described.³¹⁸ In freshwater systems steep O₂ gradients and stratification effects can also be prominent. Respiration of freshwater communities of bacteria and plankton in a cuvette format were also measured, with sensitivity at about 2.2 million cells per ml.³¹⁹ Analysis of trace O₂ and microaerobic organisms is expected to develop rapidly.

4.13. Data processing and measurement artefacts

With increasing use of the O₂ sensing techniques, processing of experimental data is becoming important, especially for measurements with complex and gentle biological samples and in applications requiring high accuracy and reliability of O₂ quantification or dealing with large amount of experimental data (screening or imaging).

A dedicated data processing algorithm was developed for screening OCR assays in microplates, in which raw profiles of phosphorescence lifetime or intensity signals are transformed into the double-reciprocal plots and then processed using the pairwise reading method. This approach provides more accurate and statistically plausible OCR values, it was initially demonstrated with simple enzymatic reactions and then applied to screening of compound libraries with isolated mitochondria.³²⁰ It also works well with sealed cuvettes and capillary platforms.⁹³

O₂ imaging, especially FLIM, 3D and time-lapse experiments also require special algorithms to cope with massive volume of data, and process them on-line. Some vendors of laboratory equipment are starting to offer dedicated templates, scripts and software for selected applications based on quenched-phosphorescence O₂ sensing. Examples include fluorescence readers from BMG, FLIM microscopes from Becker&Hickl and LaVision, O₂ probes from Luxcel, GreenLight[®] systems from Mocon.

Measurement artefacts are also quite common in this area. They are usually associated with incorrect quantification of O₂, misinterpretation of O₂ sensing data, and, the worst case, of the biological results produced. Some common sources of errors are:

(1) Effects of temperature equilibration of the sample or absence of reliable *T*-control during measurements particularly at elevated temperatures (37 °C). For example, in microplate assays with multiple samples, plate equilibration is evident over the first 5–20 min. Some instruments produce non-uniform *T*-maps and pronounced plate effects. This must be considered by optimisation, physical modelling of sample and sensor behaviour, or disregarding the initial part from quantitative analysis.

(2) Unstable or inaccurate O₂ calibration of the system, which includes the sensor, sample and instrument, for given experimental conditions. Calibration can be performed inadequately or affected by sample optical properties, interfering substances, pH, ionic strength. Sensors with high temperature

coefficients, complex shape of calibration (non-linear Stern–Volmer dependence⁶⁷) or limited stability should be avoided and replaced with more stable and robust materials.

(3) Low probe concentrations or loading efficiency, non-uniform loading of the sample or probe localisation in specific compartments (e.g. vasculature, interstitial space, but not inside cells). Photounstable probes can bleach rapidly. Low and unstable optical signals and high binning may result in significant measurement noise, skewed calibration, or parts of the sample being unavailable for O₂ analysis.

(4) Cyto- and photo-toxicity of the probe due to its non-biogenic nature, photophysical characteristics distribution within the cell (e.g. in membranes, mitochondria, nuclei) or high doses used. Shortwave excitation (e.g. with 365, 405 and even 488 nm lasers) also causes direct photodamage of cells. Optimisation of the probe, loading method, and use of red and two-photon excitable (600–900 nm) phosphors^{18,75} or up-converting nanoparticles³²¹ help to minimise this.

(5) Incorrect measurement settings on the instrument or blind conversion of sensor signals into O₂ concentration (common for many commercial systems) may produce unusual respiration profiles, OCR values and biological results.¹⁰¹ Excessive excitation can cause photosensitization and photo-damage of the sample and sensor material through generation of high levels of singlet oxygen and ROS. Thus, phosphorescent imaging with large excitation area and high repetition rate of excitation pulses leads to significant photoconsumption of O₂ resulting in measurement artefacts.²¹⁷ This can be overcome using the Small Excitation Area method.³²²

(6) Perturbed mass exchange and O₂ diffusion in the sample can strongly influence the accuracy of measurements. Uncontrolled leak of ambient O₂ through vessel material (plastic), broken seal, contact with gaseous phase, additional barrier introduced by the measurement system,³²³ or photoconsumption of O₂ can strongly influence the results. Back-diffusion of ambient O₂ can be reduced by barrier materials (e.g. glass-coated plates), special assay vessels,¹⁰⁵ or compensated by physical modelling.^{106,171}

Another group of artefacts is more specific to particular O₂ sensing formats and bioassays. For example, when changes in O₂ concentration occur very fast (e.g. in samples with high microbial load, OCR or enzyme activity), intensity based sensing may deliver a false-negative result as both fully aerated and deoxygenated conditions produce constant signal. Lifetime-based sensing overcomes this as it unambiguously informs on O₂ levels.

Planar sensor foil with O₂-impermeable support also has drawbacks. For optical measurements the specimens are usually adhered tightly to the foil (sometimes flattened and sliced), and this may impose mechanical stress, alter natural aeration conditions and produce different results.²⁴⁴ Artefacts in tissue O₂ measurement caused by mechanical stress were also reported for needle-type sensors and microelectrodes.⁶

Measurement of cellular responses in an open plate with an intracellular O₂ probe also requires conditions where resting cells partly deoxygenate themselves (basal oxygenation). If they

appear to be fully oxygenated or deoxygenated, changes in respiration upon stimulation may be undetectable: uncoupling of deoxygenated cells or inhibition of air-saturated cells cannot be seen on the profiles.¹⁸⁴ Cell oxygenation levels and optical responses can be tuned by optimising cell density or performing measurements at reduced pO₂. Mechanical distortion of the plate and effector addition can affect steady-state deoxygenation, gentle dispensing and handling of the plate helps eliminate this.

In summary, when introducing a new probe, method, measurement set-up or application based on quenched-phosphorescence O₂ sensing, they should be carefully examined for presence of such issues and degree of their severity. Analysis of primary signals from the sensor (intensity and lifetime), control samples, blanks and artificial simulations proved very useful for working out adequate measurement, signal acquisition and data processing strategies.

5. Future outlook

The above sections and examples show the high utility and tremendous potential of optical O₂ sensing in the field of biological detection. Many exciting developments, studies with different models and objects and ground breaking biological results have been produced so far. The area is set to develop rapidly for the next 5–10 years.

In basic life sciences and medicine, detailed mechanistic studies of cellular function, physiology of cell, tissue and whole organisms, stem cell technologies, tissue engineering, hypoxia research will be high on the agenda. O₂ sensing technologies are indispensable research tools here, gradually approaching their ultimate targets – diagnostics, therapeutics and human use. On the application side, growing use of biodetection systems based on O₂ sensing is envisaged, particularly in rapidly emerging large-volume applications such as microbial and food safety, security and biodefence, drug screening and toxicity testing, control of agricultural and pharmaceutical products, medical devices and bioprocessing, environmental monitoring.

The following main trends in development of such O₂ sensing systems are seen: (1) broader use of established O₂ sensing platforms, particularly fibre-optic (micro)sensors, phosphorescent readers, live-cell and animal imaging systems, intravital systems, as routine research and laboratory tools; (2) deeper integration of optical O₂ sensing technologies with advanced photonics, nanomaterials, micro and nanofabrication³²⁴ on the one hand, and cell and tissue engineering, stem cell technology, regenerative medicine and *in vivo* experimentation on the other,^{128,325} (3) wider use of imaging modalities and digital photography, development of hi-spec and super-resolution platforms, as well as portable low-cost bioimaging solutions (*e.g.* microscope on a chip⁴⁹); (4) development of interactive and intelligent sensor systems providing remote control, wireless communication and autonomous operation; (5) flexible architecture and convergence of optical O₂ sensor technology with other bioanalytical platforms. Further miniaturisation and development of sensor chips, in-field, personal care and implantable devices; (6) design of advanced sensing materials with improved

biocompatibility, analytical performance and flexibility produced by rational design for *in vivo* and *ex vivo* use; and (7) development and deployment of specialised bioanalytical platforms, integrated systems and solutions for the life, medical and environmental sciences.

Abbreviations

ADP	Adenosine-diphosphate
AMP	Adenosine-monophosphate
AOAC	Association of official analytical chemists
ATP	Adenosine-triphosphate
CCD	Charge coupled device
CMOS	Complementary metal-oxide semiconductor
ECA	Extracellular acidification
ELISA	Enzyme-linked immunosorbent assay
ERF	Ethylene-response factor
ETC	Electron transport chain
FCCP	Carbonyl cyanide 4-(trifluoromethoxy)phenyl-hydrazine
FDA	Food and drug administration
FLIM	Fluorescence/phosphorescence lifetime imaging microscopy
GFP	Green fluorescent protein
HIF	Hypoxia inducible factor
LED	Light-emitting diode
MAP	Modified atmosphere packaging
MRSA	Methicillin-resistant <i>Staphylococcus aureus</i>
NIR	Near infrared
OATP	Organic anion transporting polypeptide
OCR	Oxygen consumption rate
OxPhos	Oxidative phosphorylation
PDMS	Poly(dimethylsiloxane)
PDT	Photodynamic therapy
PEG	Poly(ethyleneglycol)
PMT	Photomultiplier tube
PPIX	Protoporphyrin IX
RLD	Rapid lifetime determination
ROS	Reactive oxygen species
SAR	Structure–activity relationships
TCSPC	Time-correlated single photon counting
TR-F	Time-resolved fluorescence
TVC	Total aerobic viable counts

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