

Contents lists available at SciVerse ScienceDirect

Renewable and Sustainable Energy Reviews

journal homepage: www.elsevier.com/locate/rser



Performance, materials and coating technologies of thermochromic thin films on smart windows



M. Kamalisarvestani ^{a,*}, R. Saidur ^a, S. Mekhilef ^b, F.S. Javadi ^a

- ^a Department of Mechanical Engineering, University of Malaya, 50603 Kuala Lumpur, Malaysia
- ^b Department of Electrical Engineering, University of Malaya, 50603 Kuala Lumpur, Malaysia

ARTICLE INFO

Article history: Received 6 December 2011 Received in revised form 5 May 2013 Accepted 20 May 2013 Available online 25 June 2013

Keywords: Smart windows Thermochromic glazing Thin film Nano particle Glass coating

ABSTRACT

A significant amount of energy is consumed to maintain thermal comfort in buildings, a huge portion of which is lost through windows. Smart coating, thin films with spectrally selective properties on the surface of glass, is the innovative solution to the problem. Thermochromic smart windows change their color and optical properties in response to temperature variations. The performance, materials, coating technologies and energy modeling of thermochromic windows are reviewed in the present study. The effect of doping vanadium dioxide (VO₂) coatings with different dopants such as tungsten, fluorine, gold nanoparticles and etc. is elaborated. Various deposition techniques, specifically hybrid chemical vapor deposition (AA/APCVD) and physical vapor deposition (PVD) methods are elucidated. Different dopants and techniques show different results on metal to semiconductor transition (MST) and the critical temperature. The "change in visible and infra-red transmission and reflectance" is the touchstone of performance for the different afforded chromogenic intelligent windows.

© 2013 Elsevier Ltd. All rights reserved.

Contents

| 1. | Introduction | . 353 |
|-----|--|-------|
| 2. | Smart windows | 354 |
| 3. | Thermochromic windows (TCW) | 356 |
| | Thermochromic materials and dopants | |
| 5. | Thermochromic coating technologies | 360 |
| | Energy modeling of thermochromic windows | |
| 7. | Conclusions | 361 |
| Ack | nowledgment | 361 |
| Ref | erences | 361 |

1. Introduction

A significant amount of energy is consumed for maintaining thermal comfort in buildings. The energy used to maintain thermal comfort in buildings is mostly exploited to keep HVAC devices running. The building energy consumption in developed countries accounts for 20–40% of the total energy use. About 41% of primary energy the U.S. (as the second largest consumer of world energy representing 19% of global consumption), consumed in 2010 was for buildings sector. Consequently, this amount

accounted for 7% of global energy use in 2010. Approximately 60% of all used energy in building sector was consumed for space heating, space cooling, lighting and ventilation in 2010 [1]. Buildings in China, as the largest consumer of world energy, consumed 26% of total primary energy in 2006; the figure is anticipated to rise to more than 30% by 2020 [2]. The building energy consumption is even more dominant in hot and humid regions, using one-third to half of the electricity produced in some countries [3–5].

In addition, building sector was the culprit of around 40%, 18% and 8% of energy-related carbon dioxide emissions in 2010 for the US, China and worldwide, respectively [1,6]. Therefore, energy saving measures should be taken in order to reduce buildings energy losses and CO₂ emissions.

There are two approaches in building energy saving strategies, the active strategies and the passive ones. Improving HVAC

^{*} Corresponding author: Tel: +603 79677611; mobile: +60176617504. *E-mail addresses*: masoudkamalis@ymail.com, mkamali@siswa.um.edu.my (M. Kamalisarvestani).

| Nomen | clature | $\Delta T \ T_t$ | Change in transmittance Transition temperature (critical temperature) |
|-------|-----------------------------------|------------------|--|
| TC | Thermochromic | PVD | Physical vapor deposition |
| EC | Electrochromic | CVD | Chemical vapor deposition |
| TCW | Thermochromic window | APCVD | Atmospheric pressure chemical vapor deposition |
| ECW | Electrochromic window | AACVD | Aerosol assisted chemical vapor deposition |
| SPD | Suspended particle device | ΔR | Change in reflectance |
| MST | Metal to semiconductor transition | | |

systems and building lighting can actively increase the building's energy efficiency, whereas measures amending the properties and thermal performance of building envelopes such as adding thermal insulation to wall, using cool coatings on roofs and coated window glazing are among the passive methods. Any building element, such as wall, roof and fenestration which separates the indoor from outdoor is called building envelope [7–10].

Windows are known as one of the most energy inefficient components of buildings [11]. Preventing these losses by improving the windows thermal performance will result in reduced electricity costs and less greenhouse gas emissions. While controlling transmitted Infrared (IR) radiation, an ideal window should be capable of sufficient transmission of visible light [12]. The most significant parameters influencing the heat transfer through windows include outdoor conditions, shading, building orientation, type and area of window, glass properties and glazing characteristics [13]. Improving glazing characteristics of windows such as thermal transmittance and solar parameters is the most important criterion to be considered in building windows standards [14]. Several international standards have been published to evaluate the performance of windows and glasses in building in order to achieve minimum requirement by considering energy performance improvement of building, ISO 10291:1994 [15]. ISO 12567:2005 [16,17], ISO 9050:2003 [18], and ISO 14438:2002 [19] are the examples of such standards.

Based on international standard (ISO 9050) [18], light transmittance, solar direct transmittance, total solar energy transmittance, ultraviolet transmittance and related glazing factors are the essential parameters for determining the light and energy performance of glazing in building. Some studies have been done based on this standard (or its European equivalent EN 410) [20] in order to determine the optical properties of coated glass products [21], modeling the solar energy transmittance of windows by considering to the angular behavior, and calculate the solar control parameters [22].

Generally, international standards specify the criteria and the essential characteristics to be considered worldwide. International standards can apply directly or modify based on the local conditions. There are many international and local standards related to the energy and lighting performance of windows, some of which are tabulated in Table 1.

The thermal dynamics and lighting potential of glazing should be considered in building energy calculations. Energy efficiency in building envelopes is generally calculated based on the ratio of the temperature difference across a building compartment and the heat flux (*R*-value) or the rate of heat transfer through a building element at certain conditions (*U*-factor). In cold climates, low *U*-factors or high *R*-values prevent heat from escaping from buildings, and in hot climates, they prevent heat from entering buildings [23,24].

2. Smart windows

Smart windows, defined as the type of windows that partially block the unwanted solar radiation, can help building to maintain higher energy performance levels. The energy performance can be improved by increasing heat gain in cold weather and decreasing it in hot weather by adopting windows' radiative and thermal properties dynamically [25]. Adding controllable absorbing layer on the surface of the glass can change the optical properties of the glass by controlling the incident solar heat flux [26]. Therefore, smart windows lead to reduced HVAC energy consumption and size and electric demand of the building [11,27,28].

There is a wide range of modern intelligent glazing options for energy saving purposes including Low-e coatings [29,30], micro blinds, dielectric/metal/dielectric (D/M/D) films [31,32], and switchable reflective devices including electrochromic windows (ECW) [33–35], gasochromic windows [36], liquid crystal glazing [11], Suspended-Particle Devices (SPD) [37] and thermochromic windows (TCW).

Low emissivity (low-E) coatings are spectrally selective films that are aimed to let the visible light pass through and block the IR and UV wavelengths which generally create heat [10]. Because of its high IR-reflectance, this type of glazing has been developed greatly, and many have studied their different properties during the last two decades [30,38–40]. Typically, there are two types of low-e coatings, the tin oxide based hard coating and the silver based soft coating with higher IR reflectance and lower transmittance than the other one. However, the visible transmittance of hard coatings can boost up with anti-reflecting property of silicon dioxide [29].

D/M/D films on glass exhibit great energy saving effects by reflecting the IR radiation by their reflective metal film and transmitting visible and near IR radiation through the two antireflective dielectric coatings [41]. Design, fabrication and properties of D/M/D films have been studied thoroughly by many researchers focusing on the optimization [31,32,42–45]. Beside the optimized performance, cost of these films in terms of their material and the fabrication technique is also important [41].

The switchable reflective devices (also called dynamic tintable windows) are categorized in to passive and active systems. In passive devices, the switching process is activated automatically in accordance with the environmental conditions. This environmental factor can be light in case of photochromic windows; or temperature and heat in thermochromic windows (TCW). Alternatively, the active systems require an external triggering mechanism to perform the modulation. For instance, electricity is the actuating signal in electrochromic windows (ECW). The active switchable glazing systems offer supplementary options compared to the passive systems whereas their dependency on power supply and wiring should be reckoned with as a drawback. Chromic materials, liquid crystals, and suspended particle windows are the three most common active-controlled intelligent windows [11]. The latter two share the disadvantage of their dependency on an electric field to be maintained when a transparent mode is desired; resulting in excessive electricity consumption. This is not the case in EC glazing that wants electricity only for transition [46]. However, chromic materials are classified into four types: electrochromic (EC), gasochromic, photochromic and thermochromic (TC). The first two belong to active glazing, responding to electricity and hydrogen

gas, respectively, as a function of solar irradiation [11,47]. Smart windows are apt to glazing the cooling load demanding buildings with large solar gain [48], though providing a see-through mode is a must in any application.

The EC effect which was first explained in 1969 is a characteristic of a device which varies its optical properties when an external voltage triggers the EC material. The EC device modulates its transmittance in visible and near IR when a low DC potential is applied [34,49].

It is usually consisted of several layers deposited on glass. The glass substrates are usually coated with transparent conducting films with natural colors-mostly tin oxide doped with either indium (ITO) or fluorine (FTO). The three major deposited layers cover the coated glass substrate as follows: *The Electrochromic film* (cathodic electro-active layer with reversible transmittance modulation characteristic) which gets a darker color when the external circuit transfers electrons into the EC lattice to compensate for the positive ions injected from the adjacent ion storage layer; *Ion conductor* (ion conducting electrolyte); and *Ion storage layer* (anodic electro-active layer) that becomes darker while releasing positive ions [33,34,50–52].

The electro-active layers (also named electrochromics) switch between their oxidized and reduced forms causing variations in their optical properties and colors as well. Ideally, it is desired that electrochromics act more reflective rather than absorptive in their colored state compared with their bleached mode [49].

EC windows should provide daylight while acting as a barrier to heat. Obviously, this type of window is not capable of providing

both effects simultaneously [35]. The EC function can be controlled by thermal load, temperature and sunlight. The latter is stated to be the best governing parameter, especially from the comfort point of view [53–56]. All the more, self-powered EC windows are also developed using semitransparent PV cells, which provide the required activating electricity [57–65].

The function of gasochromic devices is also based on electrochromism in EC windows. The main difference is that instead of DC voltage, a hydrogen gas (H₂) is applied to switch between colored and bleached states. Compared to their counterpart, gasochromic devices are cheaper and simpler because only one EC layer is enough and the ion conductor and storage layers are not needed anymore. Although, gasochromic devices exhibit some merits such as better transmittance modulation, lower required voltage, staying lucid in the swap period, and adjustability of any middle state between transparent and entirely opaque; only a few numbers of EC materials can be darkened by hydrogen. Furthermore, strict control of the gas exchange process is another issue [66].

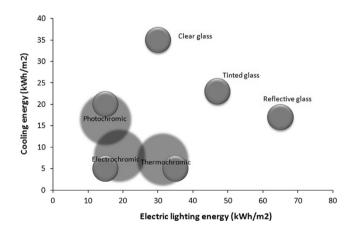
Commonly used in wrist watches, LC technology is getting more popular as a means of protecting privacy in some interior applications such as bathrooms, conference halls and fitting rooms in stores. Two transparent conductor layers, on plastic films squeeze a thin liquid crystal layer, and the whole set is pressed between two layers of glass. Normally, the liquid crystal molecules are situated in random and unaligned orientations scattering light and cloaking the view to provide the interior space with privacy. When the power is switched on the two conductive layers provide

Table 1Energy and lighting performance standards of windows.

| Name | Description | Country |
|-------------------------|---|-----------------------|
| ASTM E-2141-06 | Standard test methods for assessing the durability of absorptive electrochromic coatings on sealed insulating glass units | International [11] |
| ISO 9050:2003 | Glass in building—determination of light transmittance, solar direct transmittance, total solar energy transmittance, ultraviolet transmittance and related glazing factors | International [22] |
| ISO 10291:1994 | Glass in building—determination of steady-state <i>U</i> values (thermal transmittance) of multiple glazing—Guarded hot plate method | International |
| ISO 10292:1994 | Glass in building—calculation of steady-state U values (thermal transmittance) of multiple glazing | International |
| ISO 10293:1997 | Glass in building—determination of steady-state U values (thermal transmittance) of multiple glazing- Heat flow meter method | International |
| EN 1096-1:2012 | Glass in building—coated glass—Part 1: definitions and classification | European |
| EN 1096-2:2012 | Glass in building—coated glass—Part 2: requirements and test methods for class A, B and S coatings | European |
| EN 1096-3:2012 | Glass in building—coated glass—Part 3: requirements and test methods for class C and D coatings | European |
| EN 1096-4:2004 | Glass in building—coated glass—Part 4: evaluation of conformity/Product standard | European |
| EN ISO 14438:2002 | Glass in building—determination of energy balance value—calculation method (ISO 14438:2002) | European |
| BS EN 14449- 1:2005 | Glass in building. Laminated glass and laminated safety glass. Evaluation of conformity/product standard | British |
| BS EN 14179- 2:2005 | Glass in building. Heat-soaked thermally-toughened soda lime silicate safety glass. Evaluation of conformity/product standard | British |
| BS EN 1748-1- 2:2004 | Glass in building. Special basic products. Borosilicate glasses. Evaluation of conformity. Product standard | British |
| BS EN 12337- 2:2004 | Glass in building—chemically strengthed soda lime silicate glass—evaluation of conformity/Product standard | British |
| BS EN 14178- 2:2004 | Glass in building—basic alkaline earth silicate glass products—evaluation of conformity/Product standard | British |
| BS EN 1096 | Coated glass | British |
| BS EN 12898: 2001 | Glass in building, determination of the emissivity | British |
| BS EN 410: 1998 | Glass in building—determination of luminous and solar characteristics of glazing | British |
| BS EN 673: 1998 | Glass in building: determination of thermal transmittance (value). Calculation method | British |
| BS EN ISO 12567 | Thermal performance of windows and doors | British |
| BS EN ISO 14438:2002 | Glass in building. Determination of energy balance value. Calculation method | British |
| ES ISO 12567- 1:2012 | Thermal performance of windows and door—determination of thermal transmittance by the hot box method—complete windows and doors | Ethiopian |
| ES ISO 12567- 2:2012 | Thermal performance of windows and door—determination of thermal transmittance by the hot box method—roof windows and other projecting windows | Ethiopian |
| ES ISO 14438:2012 | Glass in building. Determination of energy balance value. Calculation method–calculation method | Ethiopian |

Table 2Comparison between different smart windows.

| Smart windows | Activation | Colour | Advantages | Disadvantages | Thermal performance | Optical performance | Application |
|-----------------------|---------------------|---------|--|---|----------------------------|---|--|
| Photochromic | Light | Bleach | No activation electricity, No foggy effect, Automatic | Manufacturing difficulties for large sizes, limited application, Cannot reduce heat gain, More darker in the winter than the summer | | UV protector, darken and lighten at sunlight and dusk | Sunglasses, Supramolecular chemistry, data storage |
| тс | Heat | Colored | High energy saving, No activation electricity | No outside visibility, manufacturing difficulties for large sizes | Reflecting heat gain | Reflecting infrared light | Duracell battery state indicators, Thermochromic paints, Thermochromic papers are used for thermal printers, thermochromism polymer |
| EC | Voltage/ current | Colored | Energy saving | Activation electricity is needed | Control the amount of heat | Reversibly changing color when a burst of charge is applied, control the amount of light | Electrochromic devices such as windows and smart glasses, Automobile industry |
| Liquid Crystal | Voltage | Bleach | Control privacy | No energy saving, No outside visibility, Activation electricity is needed | High heat gain | Transmit incident light | Electrooptical devices, hyperspectral imaging |
| Suspended Particle | Current | Bleach | Instantaneous control of light, Outside visibility, Have wide range of transmittance | Limited in size, Activation electricity is needed | Energy saving | UV protector, Reduction of infrared light, | Polaroid camera, Windows |



 $\textbf{Fig. 1.} \ \ \text{Comparison of electric lighting energy and cooling energy between different glazing types [24].}$

an electric field via their electrodes. The field causes the crystal molecules to be positioned in an aligned direction causing a change in transmittance [67,68]. The LC technology suffers from the disadvantage of high power demand in transparent mode, resulting in an electricity usage of 5–20 W/m². These devices have problems, in long term UV stability and high cost disadvantages as well [11]. The technology using liquid crystals in intelligent windows is called Polymer dispersed liquid crystals (PDLC).

SPDs have many things in common with LC devices: they are both fast in switching between phases, high electricity consumptive and dependent on an electric field. They consist of the liquid like active layer formed by adsorbing dipole needle-shaped or spherical particles (molecular particles), i.e. mostly polyhalide, suspended in an organic fluid or gel sandwiched between two sheets of glass coated with transparent conductive films. Normally, the device is in the dark reflective state because of the random

pattern of the active layer's light absorbing particles. When the electric field is applied, the particles will align resulting in the clear transmissive state. As soon as the power turns off the device switches to its dark state. Typically, the transmission of SPDs varies between 0.79–0.49 and 0.50–0.04, with 100 to 200 ms switching time and 65–220 V AC requirements [11]. The critical comparison between different smart windows is summarized in Table 2.

Shown in Fig. 1, clear glass, tinted glass, reflective glass and three chromogenic fenestration technologies are compared regarding to their cooling and electric lighting energy. As it is observed EC and TC windows demand the lowest cooling energy; and as previously reported ECWs require less energy for lighting than TCWs [69]. However, the necessity of wiring in EC glazing and the better ability of TCWs to maintain the visible transmission (when doped properly) [70] besides their simple structure [71] have given TCWs a cutting edge as a low-priced [72] alternative to the other counterparts.

Different technologies such as electrochromic, gasochromic, liquid crystal and suspended particle devices have been widely reviewed neglecting the TCWs [11,37,46,73]. As a result, there was a need to review thermochromic windows comprehensively. The following sections address the application of TC thin films in fenestration. Part 3 introduces thermochromism, TCWs, their structure, and their performance. Part 4 discusses TC materials, the dopants and the nanoparticles. In part 5 thin film deposition and fabrication techniques are addressed. In part 6 the energy modelings done on TCW are discussed. And lastly, in part 7 there will be a conclusion covering the whole papers reviewed.

3. Thermochromic windows (TCW)

In the first glance, the word "Thermochromic" might seem strange. But the word originates from the Greek roots: "Thermos" meaning warm or hot; and "Chroma" which means color. Generally,



Fig. 2. Sequential color switching of a thermochromic laminated glass [4].

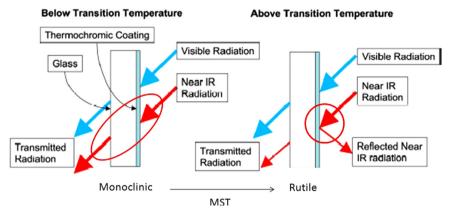


Fig.3. Schematic representation of thermochromic materials applied as an intelligent windows coating [77].

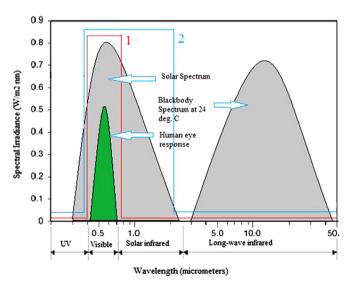


Fig. 4. The spectral transmittance of perfect TCW in cold and hot weather conditions (adopted from [81]).

thermochromic (TC) materials change color in response to temperature variations. Thermochromic window, an example of which is shown in Fig. 2, is a type of glass with TC materials that can reduce the energy demand of buildings by changing the device's reflectance and transmission properties and reducing the unwanted solar energy gain [74,75].

The TC thin film is initially in its monoclinic state (cold state) at lower temperatures (usually room temperature). Monoclinic materials behave as semiconductors, less reflective especially in near-IR (NIR) radiation. As the temperature becomes higher than a certain point, the TC material changes its nature from monoclinic to rutile state. This phenomenon is called metal to semiconductor

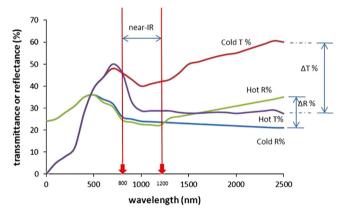


Fig. 5. Typical transmittance change (ΔT) and reflectance change (ΔR) of thermochromic films through MST (adapted from [91]).

Table 3The Ideal optical performance of thermochromic windows (adapted from [91]).

| State | Monoclinic/co | old $(T < T_t)$ | Rutile/hot (7 | Rutile/hot $(T > T_t)$ | | | |
|-------------------|---------------|-----------------|---------------|------------------------|--|--|--|
| Wavelength | Visible (%) | NIR (%) | Visible (%) | NIR (%) | | | |
| Transmittance (T) | 60–65 | 80 | 60–65 | 15 | | | |
| Reflectance (R) | 17 | 12 | 17 | 77 | | | |

transition (MST). In rutile state (hot state) the material acts like a semi-metal, reflecting a wide range of solar radiation [76] (Fig. 3).

As illustrated in Fig. 4, the majority of heat gain in solar spectrum takes place at NIR range (800–1200 nm) [78–80].

The red line (line 1) indicates the transmittance of perfect TCW in cold state. Visible light should be transmitted and near infrared radiation should be reflected. Long-wave radiation is also reflected back to indoor. This transmittance approach leads to reduction of

solar heat gain and is apt in nearly all climates. The blue line (line 2) indicates the transmittance of perfect TCW in hot state. Visible and near infrared radiation are transmitted, while long-wave infrared is reflected to inside. This transmittance mode is suitable in low temperature climates where solar heat gain is desired. Therefore, in high temperatures TCWs reduce NIR and far-IR transmittance, while in low temperatures they allow these parts of solar radiation to pass [82] (Fig. 5). The MST is fully reversible, co-occurred with large variations in electrical and optical properties in NIR range [83]. Transition temperature of pure vanadium dioxide as the most common TC material in TCWs is about 68 °C which is considered a high temperature. In order to make this type of glazing feasible, the MST temperature (T_t) should decrease to near the ambient temperature. Doping metal ions into the lattice of TC materials can alter T_t [84,85]. The size and charge [84,86,87] of dopant ion, film's strain [88,89] as well as the variations in electron carrier density are the determinant factors prevailing on the fall or rise of T_t [90].

The ideal spectral behavior of TCWs is presented in Table 3. The visible transmission and reflectance should be equal in both sides of transition while the infra-red variations are from 0% to 65%. The change in transmittance (ΔT %) and reflectance (ΔR %) can be formulated in Eqs. (1) and (2) [92]:

$$\Delta T\% = (T_h - T_c) \times 100 \tag{1}$$

$$\Delta R\% = (R_h - R_c) \times 100 \tag{2}$$

where T_h is transmittance at hot state, T_c is transmittance at cold state, R_h is reflectance at hot state and R_c is reflectance at cold state.

The most critical weakness of VO₂ coatings is their low transmittance in visible range (T_{vis}). Many studies have reported values between 40% and 50%, which is well below the acceptable value of 60% [93,94]. The reported values of transmittance and reflectance measured before and after T_t for VO₂ coatings are presented in Table 4.

Besides low luminance visibility, low energy-saving efficiency also makes application of VO₂ coatings limited. The change in

transmittance before and after T_t at 2500 nm known as the switching efficiency (η_T) is the benchmark of energy-saving efficiency. This value is influenced by doping [107,108], microstructure [80,95,109–111], and film thickness [80,88]. The most paramount factor among them is film thickness that affects switching efficiency most significantly. However, increasing the film thickness has an adverse effect on T_{vis} . As it is observed in Table 4, the ideal film thickness is between 40 and 80 nm. Choosing the most suitable dopant (for reducing T_t and improving T_{vis}), the most appropriate coating technology (to acquire the optimum thickness and sufficient TC transition), adding efficient anti reflecting coating (to increase T_{vis}) and reducing the coating costs are the crucial steps to overcome the limited application of TWCs. In the following sections these factors will be discussed.

4. Thermochromic materials and dopants

Comparatively, almost all inorganic materials exhibit color change with temperature. Electronic properties of such materials at different temperatures cause this thermochromic effect. Though, some TC materials exhibit more drastic color and property changes with temperature. Some of the most renowned TC materials and their corresponding transition temperature are introduced in Table 5.

The TC materials for glazing purposes, their properties and fabrication are not new in glazing industry, having been studied by the early 1970s [115,116]. The most promising TC material for windows is vanadium dioxide (VO₂). It is known to be in four polymorphic forms: monoclinic VO₂(M) and rutile VO₂(R) and two metastable forms VO₂(A) and VO₂(B) by monoclinic to rutile transition temperature of 68 °C [117]. VO₂ TCWs suffer from low luminous transmittance, the drawback which could be solved by fluorination [118] or applying SiO₂ anti-reflective (AR) coating [82,119]. Using ZrO₂ coating with its appropriate refractive index is reported to enhance the luminous transmittance while maintaining the TC switching [120]. The most popular materials used in TC coatings, and their corresponding effects are presented in Table 6.

Table 4Transmittance and reflectance values reported for various VO₂ coatings before and after transition temperature.

| Thermochromic coating | Belov | Below T_t (Cold) | | | | Above T_t (Hot) | | | |
|--|----------------------|-------------------------|----------------------------------|----------------------------------|----------------------|-------------------------|----------------------------------|----------------------------------|-------------------|
| | T _{vis} (%) | R _{vis} (%) | T _{IR} ¹ (%) | R _{IR} ¹ (%) | T _{vis} (%) | R _{vis} (%) | T _{IR} ¹ (%) | R _{IR} ¹ (%) | - ₍ %) |
| [95]: 280 nm tungsten-vanadium co-sputtered thin films on glass substrates by magnetron sputtering, T_c =20 °C, T_h =70 °C | 41 | 6 | 60 | 14 | 41 | 8 | 12 | 30 | 48 |
| [96]: RF reactive sputtered VO ₂ thin films, $T_c = 26 ^{\circ}\text{C}$, $T_b = 90 ^{\circ}\text{C}$ | 38 | _ | 67 | _ | 48 | _ | 10 | _ | 47 |
| [97]: deposited VO ₂ thin films by CVD | 78 | _ | _ | _ | 74 | _ | _ | _ | |
| [98]: sputter deposited VO ₂ thin films, $T_c=20$ °C, $T_h=80$ °C | 38 | _ | 72 | _ | 44 | _ | 10 | _ | 62 |
| [99]: RF reactive sputtered VO ₂ thin films, T_c =20 °C, T_h =90 °C | _ | _ | 59 | _ | _ | _ | 5 | _ | 54 |
| [100]: RF sputtered CeO ₂ –VO ₂ bilayers on SiO ₂ substrates, T_c =25 °C, T_h =100 °C | 37 | 24 | 59 | 25 | 20 | 15 | 3 | 43 | 56 |
| [101]: 50 nm VO ₂ thin films by the sol–gel process | 20 | _ | 60 | _ | 35 | _ | 5 | _ | 55 |
| [102]: VO ₂ films produced by reactive e-beam evaporation | 48 | _ | _ | _ | _ | _ | _ | _ | _ |
| [103]: 70 nm deposited VO_2 films by APCVD, $T_c=25$ °C, $T_h=65$ °C | _ | _ | 19^{2} | _ | _ | _ | 11 ² | _ | 18 ² |
| [104]: 40 nm ZnO-doped VO ₂ thin films | 60 | 38 | 85 | 10 | 65 | 32 | 32 | 30 | 53 |
| [105]: VO ₂ thin films sputtered onto corning glass, $T_c=25$ °C, $T_h=100$ °C | 38 | _ | _ | _ | 32 | _ | _ | _ | _ |
| [72]: 50 nm VO ₂ thin films, $T_c = 22 ^{\circ}\text{C}, T_h = 100 ^{\circ}\text{C}$ | 50 | 42 | 76 | 20 | 55 | 39 | 20 | 41 | 56 |
| [106]: RF sputtered VO ₂ thin films with anti reflecting coating, T_c =20 °C, T_h =90 °C | 55 | - | 52^{3} | - | 50 | - | 33^{3} | - | 19 ³ |
| [76]: | 65 | 20 | 80 | 10 | 65 | 20 | 50 | 13 | 30 |
| • 80 nm thick VO_2 films prepared by APCVD, T_c =25 °C, T_h =80 °C | | | | | | | | | |
| • 80 nm thick Tungsten–doped VO ₂ films prepared by APCVD, T_c =25 °C, T_h =40 °C | 55 | 30 | 64 | 14 | 55 | 28 | 28 | 30 | 36 |
| ¹ : Measured at 2500 nm wavelength | $T_{\rm vis}$: | Visib | le transi | mittance | $T_{\rm IR}$: | Infra- | | _ | |
| ² : Measured at 1500 nm wavelength ³ : Measured at 1000 nm wavelength | R_{vis} : | Visib | le reflec | tance | $R_{\rm IR}$: | | mittance red refl | - | |

Table 5 Thermochromic materials.

| TC material | T_t (°C) | Cold state color | Hot state color | Reason of transition |
|--|------------|--------------------------|---------------------|---|
| Cuprous mercury iodide (Cu ₂ HgI ₄) [112] | 55 | Bright red | Dark brown | Cu(I)-Hg(II) charge transfer |
| Silver mercury iodide (Ag ₂ HgI ₄) [112] | 47– 51 | Yellow | Orange | Cu(I)–Hg(II) charge transfer |
| Mercury(II) iodide Bis (dimethylammonium) tetrachloronickelate [113] | 126 110 | Red Raspberry- red | Pale yellow Blue | Reversible change transition |
| Bis (diethylammonium) tetrachlorocuprate [112] | 52– 53 | Bright green | Yellow | Relaxation of the hydrogen bond an change of arrangement of the copper atom's d -orbitals d |
| Nickel sulfate | 155 | Green | Yellow | |
| Chromium(III) oxide: aluminium(III) oxide (1:9) [114] | 400 | Red | Grey | Changes in its crystal field |
| chromium-rich pyropes | 80 | | Green | |
| normally reddish-purplish | 80 | | Green | |
| titanium dioxide | | White | Yellow | |
| zinc oxide | | White | Yellow | |
| indium(III) oxide | | Dark yellow | Yellow brown | |
| Lead(II) oxide | | Dark yellow | Yellow brown | |

Table 6 Effect of some materials on the thermochromic coatings.

| Material | Effect on MST | T_t | Film color |
|--------------------------------|--|----------------------|----------------------------------|
| Pure VO ₂ crystals | - | 68 °C [117] | Brown/yellow [122] |
| Un-doped VO ₂ films | _ | 50-66 °C [89] | Brown/yellow |
| Tungsten doping | 23% Δ <i>T</i> at | 20 °C [76] 1.56 at% | Blue [76] |
| | 2500 nm [108] | 25 °C [133] 2.7 mol% | |
| Gold nanoparticle | 35–40% ΔT and 10% | 15-20 °C [134] | Green/blue [124,125] |
| • | ΔR at 2500 nm [134] | • • | Varies in different temperatures |
| Fluorine doping | 15 % ΔT and 5 % ΔR [127] | 60 °C [127] | Brown/yellow [127] |
| | | 25 °C [28] | |
| TiO ₂ | 21.2% IR- ΔT and 84% Vis- ΔT [130] | 50-60 °C [128] | Brown [92,128] |
| | 5–10% Δ <i>T</i> at | • • | |
| | 2500 nm [128] | | |
| CeO ₂ | 5–10% Δ <i>T</i> at | 50-60 °C [128] | Brown/yellow[128] |
| | 2500 nm [128] | . , | |

Fluorine and tungsten-doped VO₂ are good choices for energy-saving windows [108,121]. However, tungsten (W) by reducing the T_t of VO₂ to the ideal temperature of 25 °C at 2 at% (lowering the T_t by 25 °C per dopant atomic percent incorporated) is the most effective dopant ion for VO₂ [76,122]. In another experiment, it is shown that a 2.7 mol% content of WO₃ has also made the film's T_t drop to room temperature [123].

Gold nanoparticles have been used as dopant in recent studies. It gives a pleasant green/blue color to the films and affects the MST and T_t . However, because of its high price, gold doping cannot be prevalent [124,125].

Fluorine doping of VO₂ films have been done by PVD techniques [28,121,126]. Incorporation of fluorine in the films using AACVD gave the films a lighter color but still the unpleasant yellow/brown color of un-doped film did not change [127].The changes in transmission (15%) and reflectance (5%) are less than those of tungsten-doped films. The key effect of fluorine is the better transmission in the visible range. In contrast to previous studies [28], it is reported that the transition temperature (~60 °C) did not change noticeably from the un-doped case [127].

Performed titanium dioxide (TiO₂) and cerium dioxide (CeO₂) nanoparticles have been also used in producing VO₂ films. By utilizing these photocatalytic nanoparticles, films with both photocatalytic and TC characteristics can be afforded. Through MST, the transmittance changes insignificantly (5–10%) at

2500 nm, while the reflectance changes about 30%. The nanoparticles used do not change the color of the film [128]. Moreover, CeO₂ can be used as a protective coating over VO₂ films to plunge the chemical deteriorations [129].

 ${
m TiO_2}$ is used to fabricate double- (or multi-) layered VO₂ thin films. The anti-reflection feature of ${
m TiO_2}$ can enhance the visible transmittance up to 84% [130]. Deposition of platinum (Pt) on VO₂ films can improve the IR reflection while maintaining the TC characteristic. T_t has been also lowered by 9.3 °C to around 58 °C. Pt has depressed the visible transmittance which can be fortified by adding ${
m SiO_2}$ antireflection coating [131]. Earlier, ${
m SiO_2}$ was used to improve the luminous transmittance to 75% (in 700 nm) compared with 33% of plain VO₂ case. ${
m SiO_2}$ has also lowered T_t to 61 °C [132].

There are some materials which can be used as deposition precursor. The most common precursors are vanadium alkoxides [89,135–137] and oxovanadium reagents [138]. Vanadium chloride should react with a source of oxygen (such as methanol or ethanol) the product of which is V_2O_5 ; heating the product in a reducing atmosphere results in deposition of VO_2 [139]. Addition of tetraoctyl ammonium boromide (TOAB) as surfactant to the vanadium dioxide matrix can control the distribution, shape and size of gold nanoparticles. It is reported that surfactant TOAB can reduce T_t from 50–54 °C to 42–45 °C [140] and from 52 °C to 34 °C [134].

Surfactants are molecules, which can alter the structure of films due to their ability to affect the surface tension of liquids and deposition mechanism in hybrid CVD processes. Using surfactants can decrease T_t of VO₂ films [125,134,140].

Thermochromic coatings can also be used on surfaces other than window. For example, Karlessi et al. introduced TC coatings as buildings external surfaces to reduce the heating and cooling loads in winter and summer [109].

5. Thermochromic coating technologies

Deposition of TC thin films on different substrates by physical vapor deposition (PVD), sputtering and sol–gel [28,74,121,141–145] techniques was executed by the early 2000s. In particular, VO_2 as the most appropriate TC material had been deposited using sol–gel [135,146–148], sputtering [71,80,149], pulsed laser techniques and chemical vapor deposition methods (CVD). These methods have been widely reviewed in previous studies [87,94,150,151].

As it is mentioned before, doping VO₂ films with tungsten can enhance their TC property. The doping and production methods include sputtering and PVD [28,141,152] sol–gel [74,135,144,146–148], APCVD [76,79,89,122,153,154] and AACVD [155,156]. Despite the technological developments in surface engineering, until now, low visible light transmission, high transition temperatures, low durability, poor transition performance, high deposition costs and unpleasant visible colors did not let these techniques be marketable.

For instance, in an attempt to decrease T_t by doping Sn on VO₂ films using PVD, it was reported that the un-doped films showed much lower T_t [141]. Even the recent production of w-doped VO₂ and V₂O₅ films prepared by sputtering has not shown good switching performance [110]. There have been some efforts to optimize the TC switching of films by controlling the deposition conditions. The best results are 76% for ΔT and 75% for ΔR ; though, T_t was still high [80].

Reactive sputtering is the most widely PVD method used to produce thermochromic thin films by means of ion beam sputtering [157], DC magnetron sputtering [152,158,159] and RF magnetron sputtering [78,160,161].

On the other hand, chemical vapor deposition (CVD) shows potential in manufacturing glasses in commercial scale. Specifically, atmospheric pressure chemical vapor deposition (APCVD) is well matched with float-glass mass production lines. The deposition rates of VO₂ films are noticeably fast in this method [76]. By the same token, the resulting physical properties such as durability and adhesion make this method more suitable [91]. There have been many researches investigating various CVD techniques and doping effects [76,79,122,162].

In a study conducted by Binions et al. [89] APCVD method was used to produce tungsten-doped VO_2 films by reaction of vanadyl acetylacetonate and tungsten hexachloride with oxygen. It is reported that tungsten doping causes a considerable drop in T_t as well as a great change in near infra-red optical properties [75]. The film thickness was also found to influence the transmission [89] and extent of TC switching [89,136]. The film thickness can be adjusted by changing the deposition time [134].

Using APCVD, TiO₂ on VO₂ on SiO₂-coated glass and VO₂ on TiO₂ on SiO₂-coated glass have been grown and compared. In both cases, the T_t did not change significantly. The ΔT % in the first multilayer film was 92% noticeably more than that of the case of VO₂ over TiO₂ (71%). The two films showed ΔR % of 55% and -55%, respectively [92]. In another study, two deposition methods were used: low pressure metal-organic CVD (MOCVD), and argon

annealing of VO₂ (B) films by MOCVD. It was found that the film microstructure regulates the MST markedly [111].

Manning et al. [79,122] used water, vanadium and tungsten sources to prepare VO_2 thin films by CVD. The resulting films had the disappointing transmission of 11% to 40%. As previously suggested, the acceptable transmission level should be above 60% so that the window's optical properties can live up to the building's aesthetical and lighting standards [152].

Vernardou et al. [162] have studied the aftereffect of doping monoclinic VO₂ by atmospheric-pressure direct liquid injection metal-organic CVD (DLI-MOCVD). It was shown that T_t of tungsten-doped VO₂ drops from 60 °C to 35 °C. Aerosol assisted CVD (AACVD) is also another deposition process reducing V₂O₃ to V₂O₅ [155]. However, the AACVD techniques do not feature good mechanical properties and can be easily removed from the surfaces [163].

Hybrid atmospheric pressure and aerosol assisted (AA/APCVD) method can be used to prepare VO_2 thin films with reduced transition temperatures [124,134]. In this technique, the multifunctional feature of AACVD is mixed with the mechanical sturdiness of APCVD [124]. This method has been also used to prepare gold doped VO_2 films. Due to the surface plasmon resonance (SPR) bands, the gold nanoparticles change the color of films from unpleasant yellow/brown color to more appealing green/blue colors. In addition, it reduces the T_t and causes a surge in reflectance. The hybrid method results in more adherent films compared with AACVD films. As MST takes place, a 10% change in reflectance and 30–40% change in transmittance is observed at 2500 nm. The reason for this change may be the metallic nature of gold nanoparticles [140]. Hybrid AA/APCVD is also used to prepare VO_2 films by using a suspension of TiO2 and CeO2 [128].

Blackman et al. [76] used Water, vanadium chloride, and tungsten chloride for synthesizing tungsten-doped VO₂ thin films by APCVD. Among the different afforded un-doped films, the optical properties of which having been measured between 300 nm to 2500 nm wavelengths, the 80 nm-thick films showed the best performance with 60% transmission at 570 nm (visible range) and 35% change in transmission at 2500 nm (far infra-red range) through transition. Testing the tungsten doped films, the effect of different tungsten atomic percentages on reducing T_t has been evaluated. As it was previously reported [157], by increasing the tungsten content T_t changes (linearly [108]) approximately by -20 °C/at% until 3 at% after which the reduction will be pseudo linearly. In addition, the powdery yellow/brown color of films can change to more adhering, more transparent and bluer/greener films by adjusting the vanadium precursor to water ratio and increasing the tungsten content of films to more than 2.5 at%.

In choosing the deposition technique, several factors should be taken into mind including the limitations of the material to be deposited, substrate material, deposition rate, the cost of required equipments, scalability, environmental considerations and the desired film features such as thickness, micro structure, mechanical strength, and optical, thermal and thermochromic performance.

In the first glance, comparing the different VO₂ coating methods results in baffling outcomes. Compared to other techniques, PVD methods require expensive equipments and high vacuum pressure. In addition, they are time and energy consumptive and suffer from low film growth rates. To the contrary, PVD methods are more suitable for synthesizing ultra thin films, require lower temperatures, are more environmentally-friendly, compatible to a wider range of substrates and more convenient for developing multi-layer thin films [150,164]. On the other hand, CVD is believed to have the potential for producing glass in commercial scale, due to its compatibility to float-glass production line [165] and fast deposition rates [165,166]. However, PVD and CVD are considered as energy consuming techniques and are not

cost-efficient. Gao et al. reviewed solution processes for VO₂ films preparation and concluded that these methods are cheap, suitable for scaling up and easy to be utilized in practical applications. In contrast, the film characteristics such as thickness and microstructure are not precisely controllable by solution based methods [94].

To sum up, both PVD and CVD methods are suitable for experimental scale due to their scalability and superiority in terms of controlling process parameters and film features. However, they are restricted by the expensive machines and processes they require, while solution processes do not suffer from these drawbacks in industrial scale. There are two suggested approaches to these shortcomings: (1) to modify the PVD and CVD methods and to design more inexpensive equipments and coating processes (2) to improve the solution based methods so that the film features including visibility, micro structure, color, and film thickness can be adjusted by controlling the process parameters.

6. Energy modeling of thermochromic windows

There are various software tools which have been used in previous researches for window simulation including energy-10 [12], the window simulation tool Winsel [167], simulation tool SOLENE [168], self developed simulation software (SDS) based on the ASHRAE tables [13], DOE building energy simulation program [69], TRNSYS building energy simulation program [66] and simulation package Integrated ENergy Use Simulation (IENUS) [27]. Different smart windows configurations have been also simulated and modeled [167,170–175] for evaluating their energy saving effect and optical performance.

Based on theory, TCWs are capable of curtailing the buildings energy consumption by allowing visible day light, limiting undesired solar gain in the hot seasons and allowing favorable solar heat gain during the cold seasons. There a few number of studies which modeled and calculated the energy performance of TCWs. Saeli et al. modeled TCWs by energy plus software and demonstrated the technology's energy saving effectiveness both in lighting and ventilation. The total energy consumption reduced more for commercial windows (100% of wall area glazing) than the residential case (25% of wall area glazing). Since high temperature lets the window be longer in its rutile state, it was also discovered that the technology works best in cities with warmer climates [91]. They also compared TCWs with conventional glazing and showed the effectiveness of TC coating in saving energy [176].

In another study Xu et al. compared the cooling energy consumption of white glass and four Low-E glasses by using the software TRNSYS 16 and showed the best performance was attained by the double glass when the VO₂ films were coated on the inside surface of the outer pane with 85% energy saving compared to white glass. However, the heating energy consumption was the highest for TCWs [177]. The result of this study also shows that TCWs are more suitable for cooling demand climates.

To the contrary, Ye et al. evaluated the energy consumption of different windows by an energy analysis program "BuildingEnergy" and showed that $\rm VO_2$ glazing has no apparent energy saving benefit and solar control advantage over conventional glazing. They concluded that, controlling the emissivity of the window is more beneficial than regulating the solar transmissivity. It was also concluded that, the energy saving effect of TCWs in summer is due to low transmittance of solar radiation and the higher absorptivity in the metal state results in higher energy consumptions which consequently makes the phase transition useless to the energy saving performance [151]. The inconsistency and scarcity of the studies in modeling the energy performance of TCWs emphasizes the necessity of more research in this field.

7. Conclusions

It's been decades since the time Vanadium dioxide thermochromic coating was reported, however, TCWs have not been commercialized due to shortcomings such as low luminance visibility, unattractive colors, low energy-saving efficiency and high coating costs. Nano technology, suitable dopants and adding efficient anti reflecting coating can reduce transition temperature (near room temperature) and improve visible transmittance (more than 60%). Appropriate and cost-efficient coating technologies provide optimum thickness (40–80 nm), sufficient thermochromic transition (more than 50%) and reduces the coating costs.

The coatings can be doped with different nanoparticles. Each dopant induces a special effect on the coating. Tungsten lowers the transition temperature, gold nanoparticles bring more pleasant film colors, fluorine increases the visible transmittance and titanium dioxide adds self-cleaning and mechanical strength to the films. The most common preparation methods are PVD, sol-gel techniques, and CVD. CVD is fast and suitable for mass production. AACVD and APCVD are the two most up-to-date deposition routes having multifunctional characteristics and mechanical strength, respectively. A prudent manner is to combine the qualities of both methods by employing hybrid AA/APCVD. To recapitulate, both PVD and CVD methods are suitable for experimental scale due to their scalability and superiority in terms of controlling process parameters and film features. However, they are restricted by the expensive machines and processes they require, while solution processes do not suffer from these drawbacks in industrial scale. There are two suggested approaches to these shortcomings: (1) to modify the PVD and CVD methods and to design more inexpensive equipments and coating processes (2) to improve the solution based methods so that the film features including visibility, micro structure, color, and film thickness can be adjusted by controlling the process parameters.

All in all, the major aims that must be reckoned with thermochromic glazing are to maximize the change in infra-red (predominantly 800–1200 nm) Reflectivity and transmission, tapering the transition temperature to near room temperature and maintaining the proper visible transmission to conserve the lighting energy. By the same token, the emissivity of the films should be modulated. Since the emissivity of the coatings are high in both monoclinic and rutile states, this technology does not work well in cooler climates currently. In addition, the energy simulation of thermochromic windows also highlights the fact that this type of glazing is more efficient in warmer climates.

Finally, it should be mentioned that there are a few works contributed to the energy modeling and heat transfer analysis of thermochromic thin films. This can be emphasized in the future attempts.

Acknowledgment

The authors would like to acknowledge the financial support from the High Impact Research Grant (HIRG) scheme (UM-MoHE) project no UM.C/HIR/MOHE/ENG/24 (D000024-16001) to carry out this research.

References

- [1] DOE U. Buildings energy databook. Energy Efficiency & Renewable Energy Department 2011.
- [2] Fridley, DG., Estimating total energy consumption and emissions of China's commercial and office buildings. 2008.
- [3] Al-Rabghi OM, Hittle DC. Energy simulation in buildings: overview and BLAST example. Energy Conversion and Management 2001;42(13):1623–35.

- [4] Wilde PD, Voorden MVD. Providing computational support for the selection of energy saving building components. Energy and Buildings 2004;36 (8):749–58.
- [5] Kwak SY, Yoo SH, Kwak SJ. Valuing energy-saving measures in residential buildings: a choice experiment study. Energy Policy 2010;38(1):673–7.
- [6] Hong T. A close look at the China design standard for energy efficiency of public buildings. Energy and Buildings 2009;41(4):426–35.
- [7] Bojic M, Yik F, Sat P. Influence of thermal insulation position in building envelope on the space cooling of high-rise residential buildings in Hong Kong. Energy and Buildings 2001;33(6):569–81.
- [8] Cheung CK, Fuller R, Luther M. Energy-efficient envelope design for high-rise apartments. Energy and Buildings 2005;37(1):37–48.
- [9] Synnefa A, Santamouris M, Akbari H. Estimating the effect of using cool coatings on energy loads and thermal comfort in residential buildings in various climatic conditions. Energy and Buildings 2007;39(11):1167–74.
- [10] Sadineni SB, Madala S, Boehm RF. Passive building energy savings: a review of building envelope components. Renewable and Sustainable Energy Reviews 2011;15(8):3617–31.
- [11] Baetens R, Jelle BP, Gustavsen A. Properties, requirements and possibilities of smart windows for dynamic daylight and solar energy control in buildings: a state-of-the-art review. Solar Energy Materials and Solar Cells 2010;94 (2):87–105.
- [12] Correa G, Almanza R. Copper based thin films to improve glazing for energy-savings in buildings. Solar Energy 2004;76(1-3):111–5.
- [13] Hassouneh K, Alshboul A, Al-Salaymeh A. Influence of windows on the energy balance of apartment buildings in Amman. Energy Conversion and Management 2010;51(8):1583–91.
- [14] Tarantini M, Loprieno AD, Porta PL. A life cycle approach to Green Public Procurement of building materials and elements: a case study on windows. Energy 2011;36(5):2473–82.
- [15] ISO, ISO 10291,2,3—Glass in building—Determination of steady-state U values (thermal transmittance) of multiple glazing 1994, Internationa Standard Organization.
- [16] ISO, ISO 12567-1—Thermal performance of windows and doors—determination of thermal transmittance by hot box method—roof windows and other projecting windows, 2005, International Standard Organization.
- [17] ISO, ISO 12567-2—Thermal performance of windows and doors—determination of thermal transmittance by hot box method—roof windows and other projecting windows, 2005, International Standard Organization.
- [18] ISO, ISO 9050—determination of light transmittance, solar direct transmittance, total solar energy transmittance, ultraviolet transmittance and related glazing factors, 2003, International Standard Organization.
- [19] ISO, ISO 14438—Glass in building—determination of energy balance value— Calculation method 2002.
- [20] Karlsson J, Roos A. Modelling the angular behaviour of the total solar energy transmittance of windows. Solar Energy 2000;69(4):321–9.
- [21] Hutchins MG, et al. Measurement and prediction of angle-dependent optical properties of coated glass products: results of an inter-laboratory comparison of spectral transmittance and reflectance. Thin Solid Films 2001;392 (2):269–75.
- [22] Alvarez G, et al. Spectrally selective laminated glazing consisting of solar control and heat mirror coated glass: preparation, characterization and modelling of heat transfer. Solar Energy 2005;78(1):113–24.
- [23] IEA. Energu efficiency requirement s in building codes, energy efficiency politics for new building. International Energy Agency 2008.
- [24] Kamalisarvestani M, Mekhilef S, Saidur R. Analyzing the optical performance of intelligent thin films applied to architectural glazing and solar collectors. Sustainability in Energy and Buildings.Springer; 813–26.
- [25] Asdrubali F, Baldinelli G. Theoretical modelling and experimental evaluation of the optical properties of glazing systems with selective films. Building Simulation 2009;2(2):75–84.
- [26] Dussault J-M, Gosselin L, Galstian T. Integration of smart windows into building design for reduction of yearly overall energy consumption and peak loads. Solar Energy 2012;86(11):3405–16.
- [27] Gugliermetti F, Bisegna F. Visual and energy management of electrochromic windows in Mediterranean climate. Building and Environment 2003;38 (3):479–92.
- [28] Burkhardt W, et al. W-and F-doped VO₂ films studied by photoelectron spectrometry. Thin Solid Films 1999;345(2):229–35.
- [29] Hammarberg E, Roos A. Antireflection treatment of low-emitting glazings for energy efficient windows with high visible transmittance. Thin Solid Films 2003;442(1-2):222-6.
- [30] Rosencrantz T, et al. Increased solar energy and daylight utilisation using anti-reflective coatings in energy-efficient windows. Solar Energy Materials and Solar Cells 2005;89(2-3):249–60.
- [31] Cheng CH, Ting JM. Transparent conducting GZO, Pt/GZO, and GZO/Pt/GZO thin films. Thin Solid Films 2007;516(2-4):203-7.
- [32] Durrani S, et al. Dielectric/Ag/dielectric coated energy-efficient glass windows for warm climates. Energy and Buildings 2004;36(9):891–8.
- [33] Zinzi M. Office worker preferences of electrochromic windows: a pilot study. Building and Environment 2006;41(9):1262–73.
- [34] Granqvist CG. Handbook of inorganic electrochromic materials. Elsevier Science Ltd.; 1995.
- [35] Lee ES, DiBartolomeo D. Application issues for large-area electrochromic windows in commercial buildings. Solar Energy Materials and Solar Cells 2002;71(4):465–91.

- [36] Wittwer V, et al. Gasochromic windows. Solar Energy Materials and Solar Cells 2004;84(1-4):305–14.
- [37] Lampert CM. Chromogenic smart materials. Materials Today 2004;7 (3):28–35.
- [38] Huang S, et al. Determination of optical constants of functional layer of online Low-E glass based on the Drude theory. Thin Solid Films 2008;516 (10):3179–83.
- [39] Ren Z, et al. Electrical and corrosion properties of the Ti5Si3 thin films coated on glass substrate by APCVD method. Journal of Non-Crystalline Solids 2011;357(15):2802–9.
- [40] Schaefer C, Bräuer G, Szczyrbowski J. Low emissivity coatings on architectural glass. Surface and Coatings Technology 1997;93(1):37–45.
- [41] Leftheriotis G, Yianoulis P, Patrikios D. Deposition and optical properties of optimised ZnS/Ag/ZnS thin films for energy saving applications. Thin Solid Films 1997;306(1):92–9.
- [42] Al-Shukri A. Thin film coated energy-efficient glass windows for warm climates. Desalination 2007;209(1-3):290–7.
- [43] Dima I, et al. Influence of the silver layer on the optical properties of the $TiO_2/Ag/TiO_2$ multilayer. Thin Solid Films 1991;200(1):11–8.
- [44] Lee CC, Chen SH, Jaing C. Optical monitoring of silver-based transparent heat mirrors. Applied Optics 1996;35(28):5698–703.
- [45] Leftheriotis G, Yianoulis P. Characterisation and stability of low-emittance multiple coatings for glazing applications. Solar Energy Materials and Solar Cells 1999;58(2):185–97.
- [46] Lampert CM. Smart switchable glazing for solar energy and daylight control. Solar Energy Materials and Solar Cells 1998;52(3-4):207–21.
- [47] Bahaj ABS, James PAB, Jentsch MF. Potential of emerging glazing technologies for highly glazed buildings in hot arid climates. Energy and Buildings 2008;40(5):720–31.
- [48] Sullivan R, et al. Energy performance of evacuated glazings in residential buildings. CA (United States): Lawrence Berkeley National Lab.; 1996.
- [49] Syrrakou E, Papaefthimiou S, Yianoulis P. Eco-efficiency evaluation of a smart window prototype. Science of The Total Environment 2006;359(1-3): 267–82
- [50] Papaefthimiou S, Leftheriotis G, Yianoulis P. Study of electrochromic cells incorporating WO₃, MoO₃, WO₃-MoO₃ and V₂O₅ coatings. Thin Solid Films 1999:343:183–6.
- [51] Papaefthimiou S, Leftheriotis G, Yianoulis P. Advanced electrochromic devices based on WO₃ thin films. Electrochimica Acta 2001;46(13-14): 2145-50.
- [52] Papaefthimiou S, Leftheriotis G, Yianoulis P. Study of WO3 films with textured surfaces for improved electrochromic performance. Solid State Ionics 2001:139(1-2):135–44.
- [53] Karlsson, J, et al. Control strategies and energy saving potentials for variable transmittance windows versus static windows. In: Proceedings of Eurosun, Copenhagen, Denmark, 2000.
- [54] Sullivan, R, et al. Effect of switching control strategies on the energy performance of electrochromic windows. 1994.
- [55] Sullivan, R, et al. The energy performance of electrochromic windows in heating-dominated geographic locations. 1996.
- [56] Sullivan R, Rubin M, Selkowitz S. Energy performance analysis of prototype electrochromic windows. CA (United States): Lawrence Berkeley National Lab.; 1996.
- [57] Bechinger C, Gregg B. Development of a new self-powered electrochromic device for light modulation without external power supply. Solar Energy Materials and Solar Cells 1998;54(1-4):405–10.
- [58] Benson, DK, et al. Stand-alone photovoltaic (PV) powered electrochromic window, 1995, Google Patents.
- [59] Huang LM, et al. Photovoltaic electrochromic device for solar cell module and self-powered smart glass applications. Solar Energy Materials and Solar Cells 2011.
- [60] Deb SK, et al. Stand-alone photovoltaic-powered electrochromic smart window. Electrochimica Acta 2001;46(13-14):2125–30.
- [61] Deb SK. Opportunities and challenges in science and technology of WO₃ for electrochromic and related applications. Solar Energy Materials and Solar Cells 2008;92(2):245–58.
- [62] Gao W, et al. First a-SiCá: áH photovoltaic-powered monolithic tandem electrochromic smart window device. Solar Energy Materials and Solar Cells 1999;59(3):243–54.
- [63] Gao W, et al. Approaches for large-area a-SiC:H photovoltaic-powered electrochromic window coatings. Journal of Non-Crystalline Solids 2000;266:1140–4.
- [64] Hauch A, et al. New photoelectrochromic device. Electrochimica Acta 2001;46(13-14):2131-6.
- [65] Pichot F, et al. Flexible solid-state photoelectrochromic windows. Journal of the Electrochemical Society 1999;146:4324.[66] Georg A, et al. Switchable glazing with a large dynamic range in total solar
- energy transmittance (TSET). Solar Energy 1998;62(3):215–28.
- [67] Doane, JW, G Chidichimo, NAP Vaz. Light modulating material comprising a liquid crystal dispersion in a plastic matrix, 1987, Google Patents.
- [68] Fergason, JL. Encapsulated liquid crystal material, apparatus and method, 1992, Google Patents.
- [69] Niklasson GA, Granqvist CG. Electrochromics for smart windows: thin films of tungsten oxide and nickel oxide, and devices based on these. Journal of Materials Chemistry 2006;17(2):127–56.
- [70] Kanu SS, Binions R. Thin films for solar control applications. Proceedings of the Royal Society of London Series A 2010;466(2113):19.

- [71] Granqvist CG, et al. Progress in chromogenics: new results for electrochromic and thermochromic materials and devices. Solar Energy Materials and Solar Cells 2009;93(12):2032–9.
- [72] Mlyuka N, Niklasson G, Granqvist CG. Thermochromic multilayer films of VO₂ and TiO₂ with enhanced transmittance. Solar Energy Materials and Solar Cells 2009;93(9):1685–7.
- [73] L CM. Large-area smart glass and integrated photovoltaics. Solar Energy Materials and Solar Cells 2003;76(4):489–99.
- [74] Greenberg CB. Undoped and doped VO₂ films grown from VO (OC₃H₇)₃. Thin Solid Films 1983;110(1):73–82.
- [75] Parkin IP, Manning TD. Intelligent thermochromic windows. Journal of Chemical Education 2006;83(3):393.
- [76] Blackman CS, et al. Atmospheric pressure chemical vapour deposition of thermochromic tungsten doped vanadium dioxide thin films for use in architectural glazing. Thin Solid Films 2009;517(16):4565–70.
- [77] Kiria P, Hyettb G, Binionsa R. Solid state thermichromic materials. Advances Material Letters 2010;1(2):20.
- [78] Jin P, Tanemura S. Formation and thermochromism of VO₂ films deposited by RF magnetron sputtering at low substrate temperature. Japanese Journal of Applied Physics Part 1 1994;33:1478 1478.
- [79] Manning TD, Parkin IP. Atmospheric pressure chemical vapour deposition of tungsten doped vanadium (IV) oxide from VOCl₃, water and WC_{I6}. Journal of Materials Chemistry 2004;14(16):2554–9.
- [80] Guinneton F, et al. Optimized infrared switching properties in thermochromic vanadium dioxide thin films: role of deposition process and microstructure. Thin Solid Films 2004;446(2):287–95.
- [81] McCluney R, Center FSE. Fenestration solar gain analysis. Citeseer 1996.
- [82] Lee MH, Cho JS. Better thermochromic glazing of windows with antireflection coating. Thin Solid Films 2000;365(1):5–6.
- [83] Morin F. Oxides which show a metal-to-insulator transition at the Neel temperature. Physical Review Letters 1959;3(1):34–6.
- [84] Phillips TE, Murphy RA, Poehler TO. Electrical studies of reactively sputtered Fe-doped VO₂ thin films. Materials Research Bulletin 1987;22(8):1113–23.
- [85] Béteille F, et al. Switching properties of V1-xTixO₂ thin films deposited from alkoxides. Materials Research Bulletin 1997;32(8):1109–17.
- [86] MacChesney J, Guggenheim H. Growth and electrical properties of vanadium dioxide single crystals containing selected impurity ions. Journal of Physics and Chemistry of Solids 1969;30(2):225–34.
- [87] Nag J, Haglund Jr R. Synthesis of vanadium dioxide thin films and nanoparticles. Journal of Physics: Condensed Matter 2008;20:264016.
- [88] Xu G, et al. Thickness dependence of optical properties of VO₂ thin films epitaxially grown on sapphire (0 0 0 1). Applied Surface Science 2005;244(1-4):
- [89] Binions R, et al. Doped and un-doped vanadium dioxide thin films prepared by atmospheric pressure chemical vapour deposition from vanadyl acetylacetonate and tungsten hexachloride: the effects of thickness and crystallographic orientation on thermochromic properties. Journal of Materials Chemistry 2007;17(44):4652–60.
- [90] Pierce J, Goodenough J. Structure of orthorhombic V_{0.95}Cr_{0.05}O_{2}. Physical Review B: Condensed Matter 1972;5(10):4104.
- [91] Saeli M, et al. Energy modelling studies of thermochromic glazing. Energy and Buildings 2010;42(10):1666–73.
- [92] Evans P, et al. Multi-functional self-cleaning thermochromic films by atmospheric pressure chemical vapour deposition. Journal of Photochemistry and Photobiology A: Chemistry 2007;189(2-3):387–97.
- [93] Li S-Y, Niklasson GA, Granqvist. C-G. Thermochromism of VO₂ nanoparticles: calculated optical properties and applications to energy efficient windows. MRS proceedings.Cambridge Univ Press; 2011.
- [94] Gao Y, et al. Nanoceramic VO₂ thermochromic smart glass: a review on progress in solution processing. Nano Energy 2012;1(2):221–46.
- [95] Zhou S, et al. Microstructures and thermochromic characteristics of low-cost vanadium-tungsten co-sputtered thin films. Surface and Coatings Technology 2012;206(11):2922–6.
- [96] Kivaisi R, Samiji M. Optical and electrical properties of vanadium dioxide films prepared under optimized RF sputtering conditions. Solar Energy Materials and Solar Cells 1999;57(2):141–52.
- [97] Binions, R, IP Parkin. Novel chemical vapour deposition routes to nanocomposite thin films. 2011.
- [98] Sobhan M, et al. Thermochromism of sputter deposited W < sub > x < / sub > V < sub > 1- x < /sub > 0 < sub > 2 < /sub > films. Solar Energy Materials and Solar Cells 1996;44(4):451–5.
- [99] Saitzek S, et al. New thermochromic bilayers for optical or electronic switching systems. Thin Solid Films 2004;449(1):166–72.
- [100] Saitzek S, et al. Thermochromic CeO < sub > 2 < /sub > -VO < sub > 2 < /sub > bilayers: role of ceria coating in optical switching properties. Optical Materials 2007;30(3):407–15.
- [101] Speck K, et al. Vanadium dioxide films grown from vanadium tetraisopropoxide by the sol–gel process. Thin Solid Films 1988;165(1):317–22.
- [102] Babulanam S, et al. Thermochromic VO < sub > 2 </sub > films for energy-efficient windows. Solar Energy Materials 1987;16(5):347–63.
- [103] Evans P, et al. Multi-functional self-cleaning thermochromic films by atmospheric pressure chemical vapour deposition. Journal of Photochemistry and Photobiology A: Chemistry 2007;189(2):387–97.
- [104] Kang L, et al. Thermochromic properties and low emissivity of ZnO: Al/VO < sub > 2 </sub > double-layered films with a lowered phase transition temperature. Solar Energy Materials and Solar Cells 2011;95(12):3189–94.

- [105] Kana J, et al. Thermochromic VO < sub > 2 < /sub > thin films synthesized by rf-inverted cylindrical magnetron sputtering. Applied Surface Science 2008;254(13):3959–63.
- [106] Lee M-H. Thermochromic glazing of windows with better luminous solar transmittance. Solar Energy Materials and Solar Cells 2002;71(4):537–40.
- [107] Chen B, et al. Al < sup > 3+ < /sup > -doped vanadium dioxide thin films deposited by PLD. Solar Energy Materials and Solar Cells 2009;93(9):1550-4.
- [108] Shi J, et al. Preparation and thermochromic property of tungsten-doped vanadium dioxide particles. Solar Energy Materials and Solar Cells 2007;91 (19):1856–62.
- [109] Karlessi T, et al. Development and testing of thermochromic coatings for buildings and urban structures. Solar Energy 2009;83(4):538–51.
- [110] Luo Z, et al. Microstructures and thermochromic properties of tungsten doped vanadium oxide film prepared by using VOX-W-VOX sandwich structure. Materials Science and Engineering: B 2011;176(9):762–6.
- [111] Sahana M, Dharmaprakash M, Shivashankar S. Microstructure and properties of VO₂ thin films deposited by MOCVD from vanadyl acetylacetonate. Journal of Materials Chemistry 2002;12(2):333–8.
- [112] Thought of Amherst. 2010 [cited 2013 2013-05-04]; Available from: (http://www3.amherst.edu/~thoughts/contents/amberger-thermochromism.html).
- [113] Bukleski M, Petruševski VM. Preparation and properties of a spectacular thermochromic solid. Journal of Chemical Education 2009;86(1):30.
- [114] Bamfield P, Hutchings MG. Chromic phenomena: technological applications of colour chemistry. Royal Society of Chemistry; 2010.
- [115] Ryabova L, Serbinov I, Darevsky A. Preparation and properties of pyrolysis of vanadium oxide films. Journal of the Electrochemical Society 1972;119:427.
- vanadium oxide films. Journal of the Electrochemical Society 19/2;119:42/.
 [116] MacChesney J, Potter J, Guggenheim H. Preparation and properties of vanadium dioxide films. Journal of the Electrochemical Society 1968;115:52.
- [117] Leroux C, Nihoul G, Van Tendeloo G. From VO_ {2}(B) to VO_ {2}(R): theoretical structures of VO_ {2} polymorphs and in situ electron microscopy. Physical Review B: Condensed Matter 1998;57(9):5111.
- [118] Gutarra A, et al. Electrochromism of sputtered fluorinated titanium oxide thin films. Applied Physics Letters 1994;64(13):1604–6.
- [119] Babulanam S, et al. Thermochromic VO₂ films for energy-efficient windows. Solar Energy Materials 1987;16(5):347–63.
- [120] Xu G, et al. Optimization of antireflection coating for VO2-based energy efficient window. Solar Energy Materials and Solar Cells 2004;83(1):29–37.
- [121] Burkhardt W, et al. Tungsten and fluorine co-doping of VO₂ films. Thin Solid Films 2002;402(1-2):226–31.
- [122] Manning TD, et al. Intelligent window coatings: atmospheric pressure chemical vapor deposition of tungsten-doped vanadium dioxide. Chemistry of Materials 2004;16(4):744–9.
- [123] Cho J-H, et al. Thermochromic characteristics of WO < sub > 3 < /sub > doped vanadium dioxide thin films prepared by sol-gel method. Ceramics International 2012:38:5589–93.
- [124] Binions R, et al. Hybrid aerosol assisted and atmospheric pressure CVD of gold doped vanadium dioxide. Chemical Vapor Deposition 2008;14(1 2):33–9
- [125] Saeli M, et al. Nano-composite thermochromic thin films and their application in energy-efficient glazing. Solar Energy Materials and Solar Cells 2010:94(2):141–51.
- [126] Khan K, Niklasson G, Granqvist C. Optical properties at the metal insulator transition in thermochromic VO₂-xFx thin films. Journal of Applied Physics 1988;64(6):3327–9.
- [127] Kiri P, et al. Fluorine doped vanadium dioxide thin films for smart windows. Thin Solid Films 2011;520(4):1363–6.
- [128] Warwick MEA, et al. Hybrid chemical vapour and nanoceramic aerosol assisted deposition for multifunctional nanocomposite thin films. Thin Solid Films 2011.
- [129] Saitzek S, et al. Thermochromic CeO₂–VO₂ bilayers: role of ceria coating in optical switching properties. Optical Materials 2007;30(3):407–15.
- [130] CHEN Z, et al. VO₂-based double-layered films for smart windows: optical design, all-solution preparation and improved properties. Solar Energy Materials and Solar Cells 2011;95(9):2677–84.
- [131] Kang L, et al. Pt/VO₂ double-layered films combining thermochromic properties with low emissivity. Solar Energy Materials and Solar Cells 2010;94(12):2078–84.
- [132] Chen HK, et al. The preparation and characterization of transparent nanosized thermochromic VO₂–SiO₂ films from the sol–gel process. Journal of Non-Crystalline Solids 2004;347(1-3):138–43.
- [133] Cho, J-H, et al. Thermochromic characteristics of WO₃-doped vanadium dioxide thin films prepared by sol–gel method. Ceramics International, (0).
- [134] Saeli M, et al. Templated growth of smart coatings: hybrid chemical vapour deposition of vanadyl acetylacetonate with tetraoctyl ammonium bromide. Applied Surface Science 2009;255(16):7291–5.
- [135] Livage J. Optical and electrical properties of vanadium oxides synthesized from alkoxides. Coordination Chemistry Reviews 1999:391–403190 1999:391–403.
- [136] Maruyama T, Ikuta Y. Vanadium dioxide thin films prepared by chemical vapour deposition from vanadium (III) acetylacetonate. Journal of Materials Science 1993;28(18):5073–8.
- [137] Parkin IP, et al. Thermochromic coatings for intelligent architectural glazing. Journal of Nano Research 2008;2:1–20.
- [138] Barreca D, et al. Vanadyl precursors used to modify the properties of vanadium oxide thin films obtained by chemical vapor deposition. Journal of the Electrochemical Society 1999:551146 1999:551.

- [139] Bramwell S, et al. Bulk magnetization of the heavy rare earth titanate pyrochlores-a series of model frustrated magnets. Journal of Physics: Condensed Matter 2000;12:483.
- [140] Saeli M, et al. Templated growth of smart nanocomposite thin films: hybrid aerosol assisted and atmospheric pressure chemical vapour deposition of vanadyl acetylacetonate, auric acid and tetraoctyl ammonium bromide. Polyhedron 2009;28(11):2233–9.
- [141] Lee MH, Kim MG, Song HK. Thermochromism of rapid thermal annealed VO₂ and Sn-doped VO₂ thin films. Thin Solid Films 1996;290:30–3.
- [142] Nygren M, Israelsson M. A DTA study of the semiconductor-metallic transition temperature in. Materials Research Bulletin 1969;4(12):881–6.
- [143] Jin P, Tanemura S. Relationship between transition temperature and x in V1- xWxO₂ films deposited by dual-target magnetron sputtering. Japanese Journal of Applied Physics 1995;34(5A):2459–60.
- [144] Takahashi I, Hibino M, Kudo T. Thermochromic properties of double-doped VO₂ thin films prepared by a wet coating method using polyvanadate-based sols containing W and Mo or W and Ti. Japanese Journal of Applied Physics 2001:139140 2001:1391.
- [145] Takahashi I, Hibino M, Kudo T. Thermochromic V_ {1-ninmbi x} W_ {ninmbi x} O_ {2} thin films prepared by wet coating using polyvanadate solutions. Japanese Journal of Applied Physics 1996;35:L438–40.
- [146] Beteille F, Livage J. Optical switching in VO₂ thin films. Journal of Sol–Gel Science and Technology 1998;13(1):915–21.
- [147] Livage J, et al. Sol-gel synthesis of oxide materials. Acta Materialia 1998;46
- [148] Livage J, et al. Optical properties of sol-gel derived vanadium oxide films. Journal of Sol-Gel Science and Technology 1997;8(1):857-65.
- [149] Cui HN, et al. Thermochromic properties of vanadium oxide films prepared by dc reactive magnetron sputtering. Thin Solid Films 2008;516(7):1484–8.
- [150] Kiri P, Hyett G, Binions R. Solid state thermochromic materials. Advanced Materials Letters 2010;1(2):86–105.
- [151] Ye H, Meng X, Xu B. Theoretical discussions of perfect window, ideal near infrared solar spectrum regulating window and current thermochromic window. Energy and Buildings 2012:164–7249 2012:164–72.
- [152] Sobhan M, et al. Thermochromism of sputter deposited WxV1-xO₂ films. Solar Energy Materials and Solar Cells 1996;44(4):451-5.
- [153] Binions R, Piccirillo C, Parkin IP. Tungsten doped vanadium dioxide thin films prepared by atmospheric pressure chemical vapour deposition from vanadyl acetylacetonate and tungsten hexachloride. Surface and Coatings Technology 2007;201(22-23):9369–72.
- [154] Qureshi U, Manning TD, Parkin IP. Atmospheric pressure chemical vapour deposition of VO₂ and VO₂/TiO₂ films from the reaction of VOCl3, TiCl4 and water. Journal of Materials Chemistry 2004;14(7):1190–4.
- [155] Piccirillo C, Binions R, Parkin IP. Synthesis and functional properties of vanadium oxides: V₂O₃, VO₂, and V₂O₅ deposited on glass by aerosol assisted CVD. Chemical Vapor Deposition 2007;13(4):145–51.
- [156] Piccirillo C, Binions R, Parkin IP. Synthesis and characterisation of W-doped VO₂ by aerosol assisted chemical vapour deposition. Thin Solid Films 2008;516(8):1992–7.
- [157] Jorgenson GV, Lee JC. Doped vanadium oxide for optical switching films. Solar Energy Materials 1986;14(3-5):205–14.

- [158] Ghanashyam Krishna M, Debauge Y, Bhattacharya A. X-ray photoelectron spectroscopy and spectral transmittance study of stoichiometry in sputtered vanadium oxide films. Thin Solid Films 1998;312(1-2):116–22.
- [159] Talledo A, Granqvist C. Infrared absorption in lithium-intercalated vanadium pentoxide films. Journal of Physics D: Applied Physics 1994:244527 1994:2445.
- [160] Xue-Jin W, et al. Surface oxidation of vanadium dioxide films prepared by radio frequency magnetron sputtering. Chinese Physics B 2008;17:3512.
- [161] Kana JBK, et al. Thermochromic VO₂ thin films synthesized by rf-inverted cylindrical magnetron sputtering. Applied Surface Science 2008;254 (13):3959–63.
- [162] Vernardou D, Pemble ME, Sheel DW. Tungsten doped vanadium oxides prepared by direct liquid injection MOCVD. Chemical Vapor Deposition 2007;13(4):158–62.
- [163] Binions R, Carmalt CJ, Parkin IP. Aerosol-assisted chemical vapour deposition of sodium fluoride thin films. Thin Solid Films 2004;469:416–9.
- [164] Livage J. Vanadium pentoxide gels. Chemistry of Materials 1991;3(4):578–93.
- [165] Guzman G, et al. Electrical switching in VO₂ sol–gel films. Journal of Materials Chemistry 1996;6(3):505–6.
- [166] Galy J. A proposal for (B) VO < sub > 2 < /sub > ⇒(A) VO < sub > 2 < /sub > phase transition: a simple crystallographic slip. Journal of Solid State Chemistry 1999;148(2):224–8.
- [167] Jonsson A, Roos A. Evaluation of control strategies for different smart window combinations using computer simulations. Solar Energy 2010;84 (1):1–9.
- [168] Miguet F, Groleau D. A daylight simulation tool for urban and architectural spaces—application to transmitted direct and diffuse light through glazing. Building and Environment 2002;37(8):833–43.
- [169] Reilly MS, et al. Modeling windows in DOE-2.1E. Energy and Buildings 1995:22(1):59–66.
- [170] Clarke J, Janak M, Ruyssevelt P. Assessing the overall performance of advanced glazing systems. Solar Energy 1998;63(4):231–41.
- [171] Syrrakou E, Papaefthimiou S, Yianoulis P. Eco-efficiency evaluation of a smart window prototype. Science of the Total Environment 2006;359(1):267–82.
- [172] Dussault J-M, Gosselin L, Galstian T. Integration of smart windows into building design for reduction of yearly overall energy consumption and peak loads. Solar Energy 2012.
- [173] Gelin K, et al. Thermal emissivity of coated glazing—simulation versus measurements. Optical Materials 2005;27(4):705–12.
- [174] Jonsson A, Roos A, Jonson EK. The effect on transparency and light scattering of dip coated antireflection coatings on window glass and electrochromic foil. Solar Energy Materials and Solar Cells 2010;94(6):992–7.
- [175] Jonsson A, Roos A. Visual and energy performance of switchable windows with antireflection coatings. Solar Energy 2010;84(8):1370–5.
- [176] Saeli M, et al. Optimisation of thermochromic thin films on glass; design of intelligent windows. Advances in Science and Technology 2011;75:79–90.
- [177] Xu X, et al. Simulation and improvement of energy consumption on intelligent glasses in typical cities of China. Science China Technological Sciences 2012;55(7):1999–2005.