oblig Y

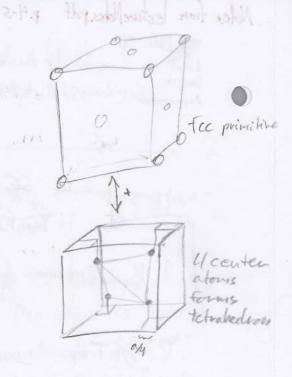
2).

assume for-si (a= 3.42 %)

& There are atoms at:

Cornus (8.1/8): 000 001 011 111 010 101

Inside (4) 1/4 1/4 3/4 1/4 3/4 3/4 3/4 3/4 3/4 3/4 3/4



Basis vectors: $\vec{a} = (0 \ 2 \ 2) \ \vec{b} = (2 \ 20) \ \vec{c} = (4 \ 4 \ 4)$

Summing any integer-scaled basis vectors (linear combination with integers as weight). gives the position of an atom.
All atoms can be represented in this way.

x= n, a + n, b + n, c = 2n, (0 /4 /4) + 2n, (40 /4) + n, (44 /4)

No= 8.1/2 + 6.1/2 + 1.4 = 8 eight atoms / unit cell each Si-atom has 14 electrons. 8.14 = 112 = Ne

b) Na = 8. 1/8 + 6/2 = 4 Ne = 4. 14 = 56

