4T Modelling Thermodynamics and Equilibrium of mixture of radiation field, ionic gas and electron gas

Constants and Functions

electron-ion collision cross sections typically 10^(-11) m^2 https://www.nist.gov/system/files/documents/srd/jpcrd386.pdf

```
(* Initial parameters and constants *)
c = 3.0 * 10^{(8)}
ep0 = 8.8541878128 * 10^{(-12)};
kb = 1.380649 * 10^{(-23)};
ee = 1.602 * 10^{(-19)};
na = 6.02 * 10^23; (* avogadros number*)
a = 2 * 10^{(-10)};
mp = 1.67262192 * 10^{(-27)};
md = 2.014102 * mp; (*mass deuterium*)
mt = 3.016049 * mp; (*mass tritium*)
mal = 4.002603 * mp; (**)
hbar = 1.05457182 * 10^{(-34)};
me = 9.1093837 * 10^{(-31)};
mme = me * na; (* molar electron mass*)
ui = 13.6 * ee;(* ionization energy for hydrogen*)
ar = 5.670374 * 10^{(-8)}; (* stefan-boltzmann constant *)
sigmaie = 1.0 * 10^{(-11)}; (* e- ion collision cross section m^2*)
sigmaii = Pi * a * a; (* ion ion collision cross section based on atomic size *)
(*Nuclear Constants for burning and transfer*)
sigt = (8 / Pi) * (ee^2 / (me * c * c))^2; (*thompson cross section*)
```

Out[72]=

 $3. \times 10^8$

Model Parameters

```
(*The model parameters*)
Ni = 1; (* number of moles of ions in total*)
ni0 = 1.0; (*initial number of moles of ions*)
ne = Ni; (* number of electrons*)
```

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In[84]:= (* Saha equation*) (* Here calculate the ionization fraction *)
     x[N_, T_, Ne_] :=
        ((2 * Pi * me * kb / hbar)^(3/2)) * (T^(3/2) / Ne) * Exp[-(ui/(kb * T))];
      (* Ne is electron number density *) (* typical ne 10^{(-20)})
      (* time between collisions \dots relaxation time*)
      taun[N_, T_] := (1 / (N * sigmaii)) * Sqrt[mp / (kb * T)];
      taue[Ne_, Te_] := (1 / (Ne * sigmaie)) * Sqrt[me / (kb * Te)]
```

In[0]:=

Radiation and Perfect Gas

```
Out[0]=
        and Gas Perfect Radiation
 In[87]:= pr[N_, Vo_, T_] := (N * kb * T / Vo) + (ar * T^4) / 3;
        Et[N_{,} Vo_{,} T_{]} := (3/2) * (N * kb * T) + (Vo * ar * T^4);
        St[N_{,} Vo_{,} T_{]} := N * kb * Log[Vo * T^{(3/2)}] + (4/3) * ar * Vo * T^{3}
```

Radiation and Ionic Gas

```
(* using saha equation to get energy
and pressure for ionic gas and radiation field*)
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Planck and Bose Einstein Distribution

```
(* Planck distribution*)
In[90]:= npl[nu_, T_] := 1.0 / (Exp[2 * Pi * hbar * nu / (kb * T)] - 1.0)
      (* Bose-Einstein distribution*)
In[91]:= nbe[nu_, mu_, T_] := 1.0 / (Exp[((2*Pi*hbar*nu) / (kb*T)) - mu] - 1.0)
```

```
In[92]:= fbe[ep_, al_] := 1.0 / (Exp[ep + al] - 1.0);
```

Rates for Nuclear Reactions

```
(* rates for nuclear reactions*)
In[135]:=
        Adt = 5.0 * 10^{(-22)}; (* m^3/s*)
        kdt = 0.1;
        T0dt = 1.0;
        sigvdt[T_] := Adt * (T0dt - Exp[-kdt * T]);
        (*DT fusion reaction cross section sigv*) (*T is the temperature in keV*)
        Plot[sigvdt[x1], {x1, 0, 120}]
Out[139]=
        5.0 × 10<sup>-22</sup>
        4.8 \times 10^{-22}
        4.6 \times 10^{-22}
        4.4 \times 10^{-22}
        4.2 \times 10^{-22}
        4.0 \times 10^{-22}
        3.8 \times 10^{-22}
                                 40
                                                           100
                                                                    120
        (* functions *)
        nuc[ne_] := c * sigt * ne(* compton strength*);
        lambda[te_, ne_] := 25.127 - Log[(ne^0.5) / (1000.0 * te)];
        fali[te_] := te / (te + 32);
        (*mme molar mass of electron*)
        pei[ne_, ni_, nal_, te_, ti_] :=
           (6 / Sqrt[Pi]) * nuc[ne] * ni * ((mme / md) + (mme / mt) + 4 * ((ni0 - ni) / ni) * (mme / mal)) *
            ((me * c * c / (2 * te))^{(3/2)}) * Log[lambda[te, ne]] * (ti - te);
```

Experiments

```
In[93]:= 2 * Pi * hbar / kb
Out[93]=
4.79924 × 10<sup>-11</sup>
```

In[94]:= **nur = 3.0 * 10^ (17)** $myl = \{nbe[nur, 0.005, T], nbe[nur, 0.1, T], nbe[nur, 0.2, T], nbe[nur, 0.3, T]\}$ Plot[myl, {T, 10 * 10^6, 500 * 10^6}]

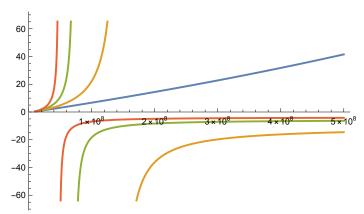
Out[94]=

 $\textbf{3.}\times\textbf{10}^{\textbf{17}}$

Out[95]=

$$\left\{\frac{1}{-1}, \frac{1}{-1}, \frac{$$

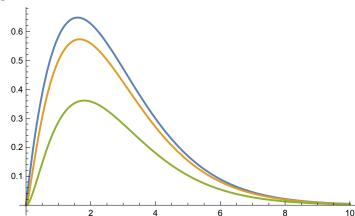
Out[96]=



In[104]:=

Plot[$\{x1^2 * fbe[x1, 0], x1^2 * fbe[x1, 0.1], x1^2 * fbe[x1, 0.5]\}, \{x1, 0, 10\}$]

Out[104]=



In[105]:=

A = 2;

k = 1;

x0 = 1.0;

In[108]:=

Out[109]=

