

Semi-Classical Quantization of Molecular Vibrations

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Abstract—This paper was written for an introductory undergraduate class in computational physics in 1997. It focuses on basic computational/numerical techniques for quadrature and root finding. It then applies these techniques to finding the energy levels of molecular hydrogen H_2 using a semi-classical quantization approach. The source code listings in FORTRAN77 can be found in the appendices.

I. QUADRATURE

A. Aims

THE main aim is to write a FORTRAN program to numerically integrate the integral:

$$\int_0^1 \frac{\ln(1+x)}{x} dx = \frac{\pi^2}{12} \quad (1)$$

- using the Trapezoidal rule,
- using Simpson's method,
- and to investigate the accuracy of both the Trapezoidal rule and Simpson's method for different step sizes.

B. Procedure

A FORTRAN program was written (quad.f, source code in Appendix A), which implemented both the Trapezoidal and Simpson's rules for the quadrature of equation 1. The program calls the functions to calculate the integral a number of times each with a different number of divisions n , and hence step size h . Both the Trapezoidal and Simpson's rules are evaluated at number of divisions starting from $n = 10$ to $n = 10^6$, with each new evaluation increasing by a single order of magnitude. This allows the investigation of the accuracy of the different techniques as the number of divisions increases.

The error is calculated is the absolute error:

$$Error = \text{abs}(Result - Answer)$$

where Result is the value produced by the numerical technique and Answer, is the analytic solution to the integral known to be $\frac{\pi^2}{12} \simeq 0.822467033$.

Since the integrand is not defined at $x = 0$, the FORTRAN function used to evaluate this returns a value of 1.0 when $x = 0$, this is because

$$\lim_{x \rightarrow 0} \frac{\ln(1+x)}{x} = 1$$

C. Results

1) *Trapezoidal Rule*: The output from the program quad.f using the Trapezoidal rule is shown below in Table I

TABLE I
RESULTS OF USING TRAPEZOIDAL RULE.

n	h	Result	Error
10	.1000000000	.8227225585	.255525E-03
100	.0100000000	.8224695905	.255709E-05
1000	.0010000000	.8224670590	.255711E-07
10000	.0001000000	.8224670337	.255709E-09
100000	.0000100000	.8224670334	.255251E-11
1000000	.0000010000	.8224670334	.263123E-13

TABLE II
RESULTS USING SIMPSON'S RULE.

n	h	Result	Error
10	.1000000000	.8224677660	.732614E-06
100	.0100000000	.8224670335	.744940E-10
1000	.0010000000	.8224670334	.799361E-14
10000	.0001000000	.8224670334	.310862E-14
100000	.0000100000	.8224670334	.310862E-14
1000000	.0000010000	.8224670334	.643929E-14

2) *Simpson's Rule*: The output from the program quad.f using Simpson's rule is shown below in Table II.

D. Discussion

Looking at the results for the Trapezoidal rule calculations in Table 1, we see that as we increase the number of divisions n by one order of magnitude, the error decreases by 2 orders. Therefore, we see that when using the Trapezoidal rule for quadrature, the error is of order $O(n^2)$, as expected.

Making a comparison with the results for the Simpson's rule calculations shown in Table 2, we see that that as the number of divisions n is increased by an order of magnitude, the error decreases by 4 orders. Therefore, when using Simpson's rule, the error is of order $O(n^4)$. The only problem is that the magnitude of the error stops decreasing after $n = 10000$. The most likely explanation for this is that we have reached the floating point limit on the number of decimal places which the computer which ran the program can handle.

Therefore we see that that Simpson's rule is more accurate than the trapezoidal rule for quadrature.

II. ROOT FINDING

A. Aims

The main aim is to write a FORTRAN program which makes use of the false position method to calculate the non zero root of the equation

$$\int_0^x t^2 dt = x \quad (2)$$

and to compare with the analytic solution. The program will make use of the Simpson's rule to evaluate the integral.

Procedure

Firstly equation 2 must be solved analytically.

$$\begin{aligned}\int_0^x t^2 dt &= x \\ \left[\frac{t^3}{3} \right]_0^x - x &= 0 \\ \frac{x^3}{3} - x &= 0 \\ x \left(\frac{x^2}{3} - 1 \right) &= 0\end{aligned}$$

Therefore equation 2 has three roots

$$x = 0, \pm\sqrt{3}$$

We are interested in finding the positive root, $x = +\sqrt{3} \simeq 1.732050808$, of equation 2 numerically.

The procedure involved writing a FORTRAN program to find the roots of the equation

$$\int_0^x t^2 dt - x = 0$$

. The algorithm used to find the root of this equation was the false position method. Simpson's rule was also used in the program to evaluate the integral in the equation.

The program outputs a table of results, with varying number of divisions n for the Simpson rule calculations and varying *Tolerance* for the false position method. The values of n range from $n = 10$ to $n = 10^7$ with each one being an order of magnitude of the previous value. The value for the tolerance of the false position algorithm is simply $1/n$.

The source code for the program `root.f` can be found in Appendix B.

B. Results

The results from the `root.f` program using the false position method is shown in Table III

TABLE III
RESULTS USING THE FALSE POSITION METHOD.

n	<i>Tolerance</i>	<i>Result</i>	<i>Error</i>
10	.1000000000	1.6831581899	.488926E-01
100	.0100000000	1.7301619523	.188886E-02
1000	.0010000000	1.7318112238	.239584E-03
10000	.0001000000	1.7320371397	.136679E-04
100000	.0000100000	1.7320476140	.319354E-05
1000000	.0000010000	1.7320506340	.173520E-06
10000000	.0000001000	1.7320507671	.404475E-07

C. Discussion

Looking at the results in Table 3, we see that as the number of divisions n (for Simpson's rule) is increased in order of magnitude and the tolerance (for false position method) is made smaller, the error decreases by order $O(n)$.

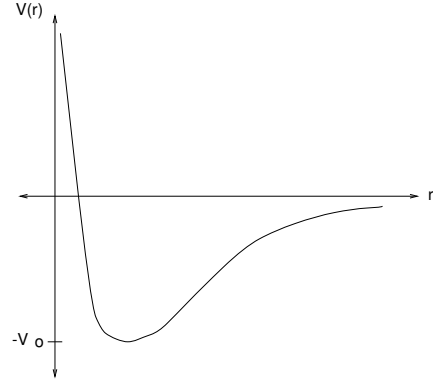
III. MOLECULAR VIBRATIONS

A. Background

When two atoms are bound together in a molecular structure such as the two Hydrogen atoms in a H_2 molecule, they vibrate. Since the nuclei (protons in this case), are much heavier than the electrons the approximation that the protons are infinitely heavier than the electrons can be made. Therefore the potential between the protons depends only on the distance between them.

At large distances, the potential is attractive to a van der Waals interaction and repulsive at short distances due to the Coulombic electrostatic repulsion of like charges, and because of the Pauli exclusion principle which states that no two fermions can occupy the same quantum state.

The interaction of the potential for a diatomic molecule such as H_2 can be summarised in the graph below.



For a quantum system such as this one, Schrodinger's Equation is usually used to solve for the allowed energies E_n of the molecular system.

$$\left[\frac{\hbar}{2m} \frac{d^2}{dr^2} + V(r) \right] = E_n \psi_n \quad (3)$$

Due to the fact that the mass of the protons is approximated to be infinite, the problem can be solved using classical mechanics and then applying quantisation rules of the "old" quantum theory.

The total energy in classical mechanics is given by the sum of the kinetic and potential energies.

$$E = \frac{p^2}{2m} + V(r) \quad (4)$$

Solving for the momentum p we get:

$$p(r) = \pm \sqrt{2m(E - V(r))} \quad (5)$$

To quantize this classical motion, we consider the potential in phase space. The area enclosed by the phase space trajectory is called the "action" is given by $S(E)$, where E is the energy. According the "old" quantum theory quantisation rules, for a given energy E_n , the action must be only half integral multiples of π .

$$S(E_n) = \oint \frac{p(r)}{\hbar} dr \quad (6)$$

$$S(E_n) = 2\sqrt{\frac{2m}{\hbar^2}} \int_{r_{in}}^{r_{out}} \sqrt{E_n - V(r)} dr \quad (7)$$

$$S(E_n) = \left(n + \frac{1}{2}\right) \pi \quad (8)$$

B. Aims

The main aim is to solve the scaled action equation, to find the quantised scaled energy levels ϵ_n for the different quantum levels n .

$$s(\epsilon_n) = \gamma \int_{x_{in}}^{x_{out}} [\epsilon_n - v(x)]^{1/2} dx = \left(n + \frac{1}{2}\right) \pi. \quad (9)$$

The actual energy levels can then be calculated from $E = V\epsilon$. The aim is to solve the equation with a quadratic potential $v(x)$ both analytically and numerically. Then the potential is to be replaced with the Morse potential which can only be solved numerically. The numerical results of the energy then need to be compared with the experimental results obtain for the quantised energy levels of the H_2 molecule.

C. Quadratic Potential Procedure

Using a quadratic potential $V(r)$, equation 9 needs to be solved.

$$V(r) = 4V_0 \left(\frac{r}{a} - 1\right) \left(\frac{r}{a} - 2\right)$$

We need to find the roots of $\epsilon_n - v(x) = 0$, $x_{in}(\epsilon_n)$ and $x_{out}(\epsilon_n)$, where $x = r/a$, and $v(x) = V(x)/V_0$ and therefore $v(x) = 4(x-1)(x-2)$.

$$\begin{aligned} \epsilon_n - v(x) &= 0 \\ \epsilon_n - 4(x-1)(x-2) &= 0 \\ -4x^2 + 12x - 8 + \epsilon_n &= 0 \end{aligned}$$

The roots of a quadratic are given by

$$\begin{aligned} x_{\pm} &= \frac{-b \pm \sqrt{b^2 - 4ac}}{2a} \\ x_{\pm} &= \frac{-12 \pm \sqrt{144 + 16(\epsilon_n - 8)}}{-8} \\ x_{\pm} &= \frac{-3 \pm \sqrt{\epsilon_n + 1}}{-2} \end{aligned}$$

Therefore the roots are given by:

$$\begin{aligned} x_- = x_{in}(\epsilon_n) &= \frac{3 - \sqrt{\epsilon_n + 1}}{2} \\ x_+ = x_{out}(\epsilon_n) &= \frac{3 + \sqrt{\epsilon_n + 1}}{2} \end{aligned}$$

With the roots of the $\epsilon_n - v(x) = 0$, know available, equation 9 can be solved analytically. Alternatively the a convenience equation can be used to solve equation 9.

$$\int_{x_-}^{x_+} \sqrt{ax^2 + bx + cd} = \frac{(4ac - b^2)\pi}{8a\sqrt{-a}}$$

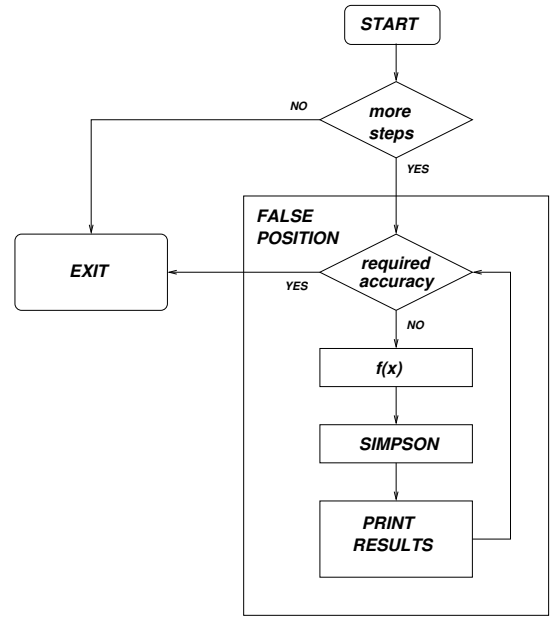


Fig. 1. Flow chart of quadratic integration algorithm

$$\begin{aligned} s(\epsilon_n) &= \gamma \int_{x_{in}}^{x_{out}} [\epsilon_n - v(x)]^{1/2} dx = \left(n + \frac{1}{2}\right) \pi \\ \gamma \int_{x_{in}}^{x_{out}} \sqrt{-4x^2 + 12x - 8 + \epsilon_n} dx &= \left(n + \frac{1}{2}\right) \pi \\ \gamma \left[\frac{4(-4)(\epsilon_n - 8) - 144}{-64} \right] &= \left(n + \frac{1}{2}\right) \pi \\ \gamma \left[\frac{\epsilon_n + 1}{4} \right] &= \left(n + \frac{1}{2}\right) \pi \\ \epsilon_n &= \frac{4\left(n + \frac{1}{2}\right)}{\gamma} - 1 \end{aligned}$$

Therefore the analytic solution for the energy ϵ_n is given by

$$\epsilon_n = \frac{4\left(n + \frac{1}{2}\right)}{\gamma} - 1$$

Using this information, a FORTRAN program was written, to solve equation 9 for the quadratic potential $v(x) = 4(x-1)(x-2)$. The source code for the program quadratic.f can be found in Appendix C. The program takes the quantum number n , and a value for the constant γ as input.

Figure 1 shows a very high level architectural design of the FORTRAN program quadratic.f shown in Appendix C.

D. Quadratic Potential Results

The table below shows the results of the quad.f program for the values of $n = 0, 1, 2, 3, 4$, and with $\gamma = 1$. Since the accuracy of this method was investigated for different step sizes and tolerance values in the previous section, the results here all have 10^6 divisions for Simpson's rule and a tolerance of 10^{-6} for the false position algorithm.

The column labelled ϵ_n is the analytic solution to equation 9 for $v(x)$ given by equation 10, whereas the column labelled ϵ_n^* is the solution given by the FORTRAN program quadratic.f.

To look at the numerical stability of the program, we vary the number of divisions N in the Simpson quadrature algorithm,

TABLE IV
VALUES OF ϵ_n FOR A QUADRATIC POTENTIAL WITH $\gamma = 1$.

n	ϵ_n	ϵ_n^*	Error
0	1	1.0000000008	.826897E-09
1	5	5.0000000025	.248070E-08
2	9	9.0000000041	.413422E-08
3	13	13.0000000058	.578873E-08
4	17	17.0000000074	.744295E-08

and the tolerance in the false position algorithm. Taking just one of the values of n , we have the following results for $n = 1$ and $\gamma = 1$, in the table below.

TABLE V
 ϵ_n FOR $n = 1$, $\gamma = 1$, FOR VARYING TOLERANCES FOR QUADRATURE AND ROOT FINDING ALGORITHMS.

N	Tolerance	ϵ_n	Error
10	.1000000	5.0801607150	.801607E-01
100	.0100000	5.0024839897	.248399E-02
1000	.0010000	5.0000784582	.784582E-04
10000	.0001000	5.0000024808	.248084E-05
100000	.0000100	5.0000000785	.784505E-07
1000000	.0000010	5.0000000025	.248070E-08

E. Quadratic Potential Discussion

Looking at Table IV, we see good agreement with the values calculated analytically for ϵ_n and those calculated numerically, with an error of approximately 10^{-8} for all cases except for when the $n = 0$, when the error is approximately 10^{-9} . The technique can then be used solve the problem of finding the quantized energy levels with more complicated and more physically realistic potentials such as the Morse potential, which there exists no analytic solution and must therefore be solved numerically.

Looking at Table V, we see the results for varying tolerances for both the quadrature (Simpson's) algorithm, N , and the root finding algorithm (false position), $Tolerance$, for $n = 1$ and $\gamma = 1$. We see that the algorithms used result in numerically stable solutions by observing that the error consistently decreases, as N is increased and as the $Tolerance$ is decreased.

F. Morse Potential Procedure

The procedure was to modify the program used to calculate ϵ_n from using a quadratic potential, to a potential that was closer in shape to what had been observed in experiments. This is the Morse potential given by

$$V_{Morse}(r) = V_0 \left[\left(1 - e^{-(r-r_{min})/a} \right)^2 - 1 \right] \quad (10)$$

The Morse potential can be normalised ($x = r/a$ and $v(x) = V(r)/V_0$) to get

$$v(x) = \left(1 - e^{-(x-x_{min})} \right)^2 - 1 \quad (11)$$

The next step is to analytically find the turning points $x_{in}(\epsilon_n)$ and $x_{out}(\epsilon_n)$ of the equation $\epsilon_n - v(x) = 0$.

$$\epsilon_n - \left[\left(1 - e^{-(x-x_{min})} \right)^2 - 1 \right] = 0$$

Let $z = e^{-(x-x_{min})}$

$$\begin{aligned} \epsilon_n - [(1-z)^2 - 1] &= 0 \\ \epsilon_n - (z^2 - 2z) &= 0 \\ z^2 - 2z - \epsilon_n &= 0 \end{aligned}$$

Solve for z using

$$\begin{aligned} z_{\pm} &= \frac{-b \pm \sqrt{b^2 - 4ac}}{2a} \\ z_{\pm} &= \frac{2 \pm \sqrt{4 - 4(1)(-\epsilon_n)}}{2} \\ z_{\pm} &= 1 \pm \sqrt{1 + \epsilon_n} \end{aligned}$$

Now substitute z back in to get

$$\begin{aligned} e^{-(x_{\pm} - x_{min})} &= 1 \pm \sqrt{1 + \epsilon_n} \\ -(x_{\pm} - x_{min}) &= \ln(1 \pm \sqrt{1 + \epsilon_n}) \\ x_{\pm} &= x_{min} - \ln(1 \pm \sqrt{1 + \epsilon_n}) \end{aligned}$$

Therefore $x_{in}(\epsilon_n) = x_{min} - \ln(1 + \sqrt{1 + \epsilon_n})$ and $x_{out}(\epsilon_n) = x_{min} - \ln(1 - \sqrt{1 + \epsilon_n})$.

The following procedure was then followed:

- The `quadratic.f` program was modified to change the potential from quadratic to Morse and to add the new turning points calculated above.
- The source code for the new program `morse.f` can be found in Appendix D.
- For the case of $n = 0$, the program was run with a number of different values of a , until a value was found such that the value given for the energy $E_n = V_0 \epsilon_n$ was very close to value of $E_0 = -4.477$ as given in Table 1.5 of Koonin (experimental result).
- Using the final value of a , the first four quantised energy levels were calculated with the program and compared with the experimental results for the H_2 spectrum.
- The choice of starting values for ϵ_n , is restricted by the turning points, $x_{\pm} = x_{min} - \ln(1 \pm \sqrt{1 + \epsilon_n})$. We can see from this equation that $-2 < \epsilon_n \leq 1$.

G. Morse Potential Results

1) *Determining the parameter a:* The table below shows the results of inputting different values of a into the `morse.f` program. We have the following values:

- $\gamma = 33.6567a$
- $r_{min} = 0.74166 \text{ \AA}$, $x_{min} = 0.74166a$
- $V_0 = 4.747 \text{ eV}$, ($E_n = V_0 \epsilon_n$)
- $n = 0$
- $N = 1000$ (Number of divisions in Simpson's Rule)
- $T = 10_{-3}$ (Tolerance in false position method)

From experimental results looking at the spectrum of the Hydrogen molecule, we know that for the lowest energy state (when

TABLE VI
VALUES OF ϵ_0 FOR DIFFERENT VALUES OF THE PARAMETER a .

a	ϵ_0
0.50	-.9414586271
0.60	-.9510928378
0.55	-.9467075966
0.52	-.9436775546
0.51	-.9425895401
0.515	-.9431387536

$n = 0$), we have $E_0 = -4.477$ eV. Therefore, for a given value of a , we are looking for $\epsilon_0 = -4.477/4.747 \simeq -0.943121971$.

We see from Table VI that we get closest to the value of ϵ_0 when $a = 0.515$.

Using this value of a , we can then run the program for the first 4 values of $n = 0, 1, 2, 3$ and compare to the experimental results. Since the experimental results are only available to 3 decimal places, it is meaningless to attempt to calculate solutions with more accuracy since we don't have more accurate experimental data to compare with. In all cases the calculations were done with 1000 divisions in Simpson's rule and a tolerance of 10^{-3} for the root finding false position algorithm.

The results of the running the program for the first four energy levels are shown in Table VII, where E_n^* is the numerical result calculated using the program and E_n is the experimental result from Table 1.5 of Koonin.

TABLE VII
FIRST FOUR QUANTIZED ENERGY LEVELS OF THE SPECTRUM OF THE H_2 MOLECULE.

n	ϵ_n	E_n^*	E_n	Error
0	-.9431387536	-4.4770796631	-4.477	.796631E-04
1	-.8344090650	-3.9609398314	-3.962	.106017E-02
2	-.7323362062	-3.4763999710	-3.475	.139997E-02
3	-.6369210780	-3.0234643571	-3.017	.646436E-02

H. Morse Potential Discussion

As can be seen from Table 7, the values of the energy levels of the Hydrogen molecule for quantum levels $n = 0, 1, 2, 3$, have a good agreement with the experimental results. The error was of magnitude 10^{-2} for $n = 1, 2, 3$ and of magnitude 10^{-4} for $n = 0$.

The agreement between calculation and experiment is good, especially considering the use of the old quantum theory in the numerical calculations.

More insight could have been gained into the accuracy of the numerical techniques used to do the calculations if the experimental data presented for the Hydrogen molecule spectrum was more accurate.

With all the data now available we can make a plot of the equation

$$V_{Morse}(r) = V_0 \left[\left(1 - e^{-(r-r_{min})/a} \right)^2 - 1 \right].$$

Using the values for γ , a , r_{min} , and V_0 , shown in the previous section we can plot the Morse potential for the Hydrogen molecule in Figure 2

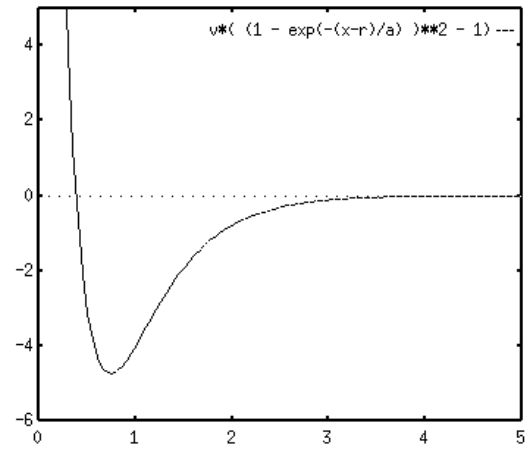


Fig. 2. Morse Potential for Hydrogen Molecule

Using the data for the first four quantized energy levels, we can plot these on top of the Morse potential plot, shown in Figure 3.

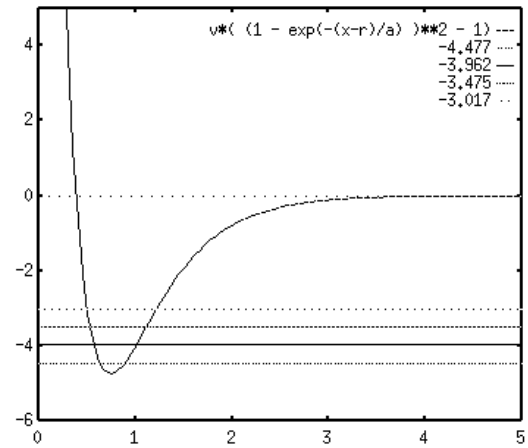


Fig. 3. First Four Quantized Energy Levels

APPENDIX
CODE LISTING: QUAD.F

```

1  C-----
2  C PROGRAM : QUADRATURE - TRAPEZOIDAL AND SIMPSONS RULES
3  C
4  C SUBJECT : 640-364 COMPUTATIONAL PHYSICS
5  C NAME : MICHAEL PAPASIMEON
6  C DATE : 30/07/1997
7  C
8  C PURPOSE : This program does a numerical integration
9  C : between 0 and 1 for the function ln(1+x)/x
10 C : using both the Trapezoidal and Simpsons
11 C : rules for quadrature.
12 C-----
13
14     program main
15         call all_trapezoidals
16         call all_simpsons
17     end
18
19 C-----
20 C FUNCTION : f
21 C PURPOSE : Given a real number x, returns the result of
22 C : evaluating ln(1+x)/x.
23 C : Care is taken when x = 0, since this causes
24 C : a division by zero error, and hence a value
25 C : of 1 is returned in this case.
26 C-----
27
28     double precision function f(x)
29     implicit none
30     real*8 x
31     if ( x .eq. 0.0d0 ) then
32         f = 1.0
33     else
34         f = dlog(1+x)/x
35     endif
36     return
37     end
38
39 C-----
40 C SUBROUTINE : all_trapezoidals
41 C PURPOSE : Calls the trapezoidal subroutine 6 each time
42 C : with an increase in the number of divisions
43 C : (increased by a factor of 10). The result
44 C : is a table of the result of the quadrature
45 C : including an output of the percentage error.
46 C-----
47
48     subroutine all_trapezoidals
49     implicit none
50     integer*4 n, i
51     real*8 actual, pi, a, b
52
53     a = 0.0d0
54     b = 1.0d0
55     pi = 4*atan(1.0d0)
56     actual = (pi**2)/dfloat(12)
57
58     write(*,*) '-----'
59     write(*,*) '          TRAPEZOIDAL_RULE'
60     write(*,*) '-----'
61     write(*,*) 'Lower_Limit_:a= ',a
62     write(*,*) 'Upper_Limit_:b= ',b
63     write(*,*) 'Actual_Result_= ',actual
64     write(*,*) '-----'
65     write(*,*) '          n          h          Result          Error'

```

```

66     write(*,*) '-----',
67
68     n = 10
69     do 10 i = 0, 5, +1
70         call trapezoidal(n,a,b,actual)
71         n = n*10
72 10 continue
73     write(*,*) '-----',
74
75     end
76
77 C-----
78 c SUBROUTINE : trapezoidal
79 c PURPOSE : Performs numerical quadrature using the
80 c : trapezoidal rule for the function f
81 c : between a and b and prints out the result
82 c : for the number of divisions given as a
83 c : parameter.
84 C-----
85
86     subroutine trapezoidal(n, a, b, answer)
87     implicit none
88     integer*4 n
89     real*8 a, b, answer
90     integer*4 i
91     real*8 error, h, sum, x, f
92
93     i = 0
94     sum = 0.0d0
95     x = 0.0d0
96     h = (b-a)/dfloat(n)
97
98     do 100 i = 1, (n-1), +1
99         x = a + i*h
100        sum = sum + 2.0d0*f(x)
101 100 continue
102
103        sum = sum + f(a) + f(b)
104        sum = (0.5d0)*h*sum
105        error = abs(sum - answer)
106
107        write(*,200) n, h, sum, error
108 200 format(i10, f14.10, f14.10, g14.6)
109
110        end
111
112 C-----
113 c SUBROUTINE : all_simpsons
114 c PURPOSE : Calls the simpson subroutine 6 each time
115 c : with an increase in the number of divisions
116 c : (increased by a factor of 10). The result
117 c : is a table of the result of the quadrature
118 c : including an output of the percentage error.
119 C-----
120
121     subroutine all_simpsons
122     implicit none
123     integer*4 n, i
124     real*8 actual, pi, a, b
125
126     a = 0.0d0
127     b = 1.0d0
128     pi = 4*atan(1.0d0)
129     actual = (pi**2)/dfloat(12)
130
131     write(*,*) '-----',
132     write(*,*) 'SIMPSONS_RULE',
133     write(*,*) '-----',
134     write(*,*) 'Lower Limit: a=', a

```

```

135 write(*,*) 'Upper_Limit:=',b
136 write(*,*) 'Actual_Result:=',actual
137 write(*,*) '-----',
138 write(*,*) 'n=====h=====Result=====Error',
139 write(*,*) '-----',
140
141 n = 10
142 do 10 i = 0, 5, +1
143     call simpson(n,a,b,actual)
144     n = n*10
145 10 continue
146     write(*,*) '-----',
147
148 end
149
150 C-----
151 c SUBROUTINE : simpson
152 C-----
153
154 subroutine simpson(n, a, b, answer)
155 implicit none
156 integer*4 n, i, factor
157 real*8 a, b, answer
158 real*8 error, h, sum, x, f
159
160 i = 0
161 factor = 4
162 sum = 0.0d0
163 x = 0.0d0
164 h = (b-a)/dfloat(n)
165
166 do 300 i = 1, (n-1), +1
167     x = a+i*h
168     if (factor .eq. 2) then
169         sum = sum + 2*f(x)
170         factor = 4
171     else
172         sum = sum + 4*f(x)
173         factor = 2
174     endif
175 300 continue
176     sum = sum + f(a) + f(b)
177     sum = (h*sum)/3.0d0
178     error = dabs(sum - answer)
179
180     write(*,400) n, h, sum, error
181 400 format(i10, f14.10, f14.10, g14.6)
182
183 end

```


APPENDIX
CODE LISTING: ROOT.F

```

1  C-----
2  C PROGRAM : ROOT FINDING - FALSE POSITION METHOD
3  C
4  C SUBJECT : 640-364 COMPUTATIONAL PHYSICS
5  C NAME : MICHAEL PAPASIMEON
6  C DATE : 06/08/1997
7  C
8  C PURPOSE : This program does a numerical integration
9  C : between 0 and 1 for the function ln(1+x)/x
10 C : using both the Trapezoidal and Simpsons
11 C : rules for quadrature.
12 C-----
13
14     program main
15     implicit none
16     integer*4 n, i
17     real*8 result, false_position, tolx, answer, error
18
19     answer = dsqrt(3.0d0)
20
21     n = 10
22     tolx = 1.0d0/dfloat(n)
23     do 10 i = 0, 6, +1
24         result = false_position(n, 1.0d0, tolx)
25         error = dabs(result - answer)
26         write(*,15)n, tolx, result, error
27 15 format(i10, f14.10, f14.10, g14.6)
28         n = n*10
29         tolx = 1.0d0/dfloat(n)
30 10 continue
31     end
32
33 C-----
34 C FUNCTION : g
35 C PURPOSE : returns t**2
36 C-----
37
38     double precision function g(t)
39     implicit none
40     real*8 t
41     g = t**2
42     return
43     end
44
45 C-----
46 C SUBROUTINE : simpson
47 C PURPOSE : calculates the integral between a and b
48 C : for the function g(x)
49 C-----
50
51     double precision function simpson(n, a, b)
52     implicit none
53     integer*4 n, i, factor
54     real*8 a, b
55     real*8 h, sum, x, g
56
57     i = 0
58     factor = 4
59     sum = 0.0d0
60     x = 0.0d0
61     h = (b-a)/dfloat(n)
62
63     do 300 i = 1, (n-1), +1
64         x = a+i*h
65         if (factor .eq. 2) then

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66         sum = sum + 2.0d0*g(x)
67         factor = 4
68     else
69         sum = sum + 4.0d0*g(x)
70         factor = 2
71     endif
72 300 continue
73     sum = sum + g(a) + g(b)
74     sum = (h*sum)/3.0d0
75
76     simpson = sum
77     return
78
79 end
80
81 C-----
82 c FUNCTION : f
83 c PURPOSE : calculates and returns the value of the function
84 c : of the integral between a and b of the function
85 c : g(x) determined by the simpson function minus
86 c : c. When a = 0 and b = x, we have
87 c : f = Integrate[t**2 dt,0,x] - x
88 C-----
89
90     double precision function f(n,a,b)
91     implicit none
92     integer*4 n
93     real*8 a, b, simpson
94     f = simpson(n,a,b) - b
95     return
96 end
97
98 C-----
99 c FUNCTION : false_position
100 c PURPOSE : calculates and returns the root of the function
101 c : f, after position x1, to a tolerance of tol
102 c : using the false position method. The parameter
103 c : n determines the accuracy to which the function
104 c : f can be calculated to.
105 C-----
106
107     double precision function false_position(n, x1, tol)
108     implicit none
109     integer*4 n
110     real*8 x1, tol
111     real*8 f
112     real*8 x2, f1, f2, x3, f3, h, a
113
114     a = 0.0d0
115     h = 0.33
116
117     x2 = x1 + h
118     f1 = f(n,a,x1)
119     f2 = f(n,a,x2)
120
121     do while (f1*f2 .ge. 0.0d0)
122         x2 = x2 + h
123         f2 = f(n,a,x2)
124     end do
125
126     x3 = x2 - f2*(x2-x1)/(f2-f1)
127     f3 = f(n,a,x3)
128
129     do while (dabs(f3) .gt. tol)
130         if (f1*f3 .lt. 0.0d0) then
131             x2 = x3
132         else
133             x1 = x3
134         endif

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```
135         f1 = f(n,a,x1)
136         f2 = f(n,a,x2)
137         x3 = x2 - f2*(x2-x1)/(f2-f1)
138         f3 = f(n,a,x3)
139     end do
140
141     false_position = x3
142     return
143
144 end
```

APPENDIX
CODE LISTING: QUADRATIC.F

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1  C-----
2  C PROGRAM : main
3  C-----
4
5  program main
6  implicit none
7  integer i, n, step
8  real*8 gamma, en, answer, error
9  real*8 result, tol, false_position
10
11  write(*,*) 'Enter n, gamma, answer'
12  read(*,*) n, gamma, answer
13
14  en = -0.1d0
15  step = 10
16  tol = 1.0d0/step
17
18  write(*,*) '-----'
19  write(*,*) 'Molecular_Vibrations:_Quadratic_Potential'
20  write(*,*) '_'
21  write(*,*) 'step=_step_size_for_simpson_integration'
22  write(*,*) 'tol=_tolerance_for_false_position_method'
23  write(*,*) 'n=_', n
24  write(*,*) 'gamma=_', gamma
25  write(*,*) 'Energy=_Quantised_energy_level_En'
26  write(*,*) '-----'
27  write(*,*) 'step_tol_Energy_Error'
28  write(*,*) '-----'
29
30  do 100 i = 0, 5, +1
31      result = false_position(n, gamma, step, en, tol)
32      error = dabs(answer - result)
33      write(*,20) step, tol, result, error
34  20 format(i8, f9.7, f14.10, e14.6)
35      step = step*10
36      tol = 1.0d0/step
37  100 continue
38
39  write(*,*) '-----'
40
41  end
42
43  C-----
44  C FUNCTION : xin
45  C-----
46
47  double precision function xin(en)
48  implicit none
49  real*8 en
50
51  xin = (3.0d0-dsqrt(en + 1.0d0))/2.0d0
52  return
53
54  end
55
56  C-----
57  C FUNCTION : xout
58  C-----
59
60  double precision function xout(en)
61  implicit none
62  real*8 en
63
64  xout = (3.0d0+dsqrt(en + 1.0d0))/2.0d0
65  return

```

```

66
67     end
68
69 C-----
70 C FUNCTION : f
71 C-----
72
73     double precision function f(en, n, gamma, step)
74     implicit none
75     integer*4 n, step
76     real*8 xin, xout, simpson
77     real*8 en, gamma, xi, xo, pi
78
79     pi = 4*atan(1.0d0)
80     xi = xin(en)
81     xo = xout(en)
82
83     f = gamma*simpson(step,xi,xo,en) - pi*(dfloat(n) + 0.5d0)
84     return
85
86     end
87
88 C-----
89 C FUNCTION : g
90 C PURPOSE : integrand
91 C-----
92
93     double precision function g(en, x)
94     implicit none
95     real*8 en, x, v, arg
96
97     arg = en - v(x)
98
99     if (dabs(arg) .lt. 1.0E-14) then
100         g = 0.0d0
101     else
102         g = dsqrt(arg)
103     endif
104     return
105
106     end
107
108 C-----
109 C FUNCTION : v
110 C PURPOSE : potential
111 C-----
112
113     double precision function v(x)
114     implicit none
115     real*8 x
116
117     v = 4*(x-1)*(x-2)
118     return
119
120     end
121
122 C-----
123 C FUNCTION : simpson
124 C-----
125
126     double precision function simpson(step, a, b, en)
127     implicit none
128     integer*4 step, i, factor
129     real*8 a, b, en
130     real*8 h, sum, x, g
131
132     i = 0
133     factor = 4
134

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```

135     sum = 0.0d0
136     x = 0.0d0
137     h = (b-a)/dfloat(step)
138
139     do 300 i = 1, (step-1), +1
140         x = a+i*h
141         if (factor .eq. 2) then
142             sum = sum + 2.0d0*g(en,x)
143             factor = 4
144         else
145             sum = sum + 4.0d0*g(en,x)
146             factor = 2
147         endif
148 300 continue
149     sum = sum + g(en,a) + g(en,b)
150     sum = (h*sum)/3.0d0
151
152     simpson = sum
153     return
154
155 end
156
157 C-----
158 c FUNCTION : false_position
159 c PURPOSE : calculates and returns the root of the function
160 c : f, after position x1, to a tolerance of tol
161 c : using the false position method. The parameter
162 c : step determines the accuracy to which the function
163 c : f can be calculated to.
164 C-----
165
166     double precision function false_position(n,gamma,step,
167 > start,tolx)
168     implicit none
169     integer*4 step, n
170     real*8 start, tol, gamma
171     real*8 f
172     real*8 x1, x2, f1, f2, x3, f3, h
173
174     h = 0.3d0
175
176     x1 = start
177     x2 = x1 + h
178     f1 = f(x1,n,gamma,step)
179     f2 = f(x2,n,gamma,step)
180
181     do while (f1*f2 .ge. 0.0d0)
182         x2 = x2 + h
183         f2 = f(x2,n,gamma,step)
184 90 format(f15.8, f15.8)
185     end do
186
187     x3 = x2 - f2*(x2-x1)/(f2-f1)
188     f3 = f(x3,n,gamma,step)
189
190     do while (dabs(f3) .gt. tol)
191         if (f1*f3 .lt. 0.0d0) then
192             x2 = x3
193         else
194             x1 = x3
195         endif
196         f1 = f(x1,n,gamma,step)
197         f2 = f(x2,n,gamma,step)
198         x3 = x2 - f2*(x2-x1)/(f2-f1)
199         f3 = f(x3,n,gamma,step)
200     end do
201
202     false_position = x3
203     return

```

204

205

end

APPENDIX
CODE LISTING: MORSE.F

```

1 C-----
2 C PROGRAM : main
3 C-----
4
5 program main
6 implicit none
7 integer i, n, step
8 real*8 gamma, en, answer, error, xmin, a
9     real*8 result, tol, false_position
10
11     write(*,*) 'Enter n, a'
12     read(*,*) n, a
13
14     answer = -0.943121971d0
15     gamma = 33.6567d0*a
16
17     en = -1.0d0
18     step = 10
19     tol = 1.0d0/step
20     xmin = 0.74166d0*a
21
22     write(*,*) '-----'
23     write(*,*) 'Molecular_Vibrations_:Quadratic_Potential'
24     write(*,*) '_'
25     write(*,*) 'step=_step_size_for_simpson_integration'
26     write(*,*) 'tol=_tolerance_for_false_position_method'
27     write(*,*) 'n=_', n
28     write(*,*) 'gamma=_', gamma
29     write(*,*) 'Energy_=Quantised_energy_level_En'
30     write(*,*) '-----'
31     write(*,*) 'step_____tol_____Energy_____Error'
32     write(*,*) '-----'
33
34     do 100 i = 0, 5, +1
35         result = false_position(n, gamma, step, en, tol, xmin)
36         error = dabs(answer - result)
37         write(*,20) step, tol, result, error
38 20 format(i8, f9.7, f14.10, e14.6)
39         step = step*10
40         tol = 1.0d0/step
41 100 continue
42
43     write(*,*) '-----'
44
45     end
46
47 C-----
48 c FUNCTION : xin
49 C-----
50
51 double precision function xin(en, xmin)
52 implicit none
53 real*8 en, xmin
54
55     xin = xmin - dlog(1.0d0 + dsqrt(en + 1.0d0))
56     return
57
58     end
59
60 C-----
61 c FUNCTION : xout
62 C-----
63
64 double precision function xout(en, xmin)
65 implicit none

```



```

66     real*8 en, xmin
67
68 c write(*,*)'[xout]:[en]_',en
69     xout = xmin - dlog(1.0d0 - dsqrt(en + 1.0d0))
70     return
71
72     end
73
74 C-----
75 c FUNCTION : f
76 C-----
77
78     double precision function f(en, n, gamma, step, xmin)
79     implicit none
80     integer*4 n, step
81     real*8 xin, xout, simpson
82     real*8 en, xmin, gamma, xi, xo, pi
83
84     pi = 4*atan(1.0d0)
85     xi = xin(en, xmin)
86     xo = xout(en, xmin)
87
88     f = gamma*simpson(step,xi,xo,en,xmin) - pi*(dfloat(n) + 0.5d0)
89     return
90
91     end
92
93 C-----
94 c FUNCTION : g
95 c PURPOSE : integrand
96 C-----
97
98     double precision function g(en, x, xmin)
99     implicit none
100    real*8 en, xmin, x, v, arg
101
102    arg = en - v(x, xmin)
103
104    if (dabs(arg) .lt. 1.0E-14) then
105        g = 0.0d0
106    else
107        g = dsqrt(arg)
108    endif
109    return
110
111    end
112
113 C-----
114 c FUNCTION : v
115 c PURPOSE : potential
116 C-----
117
118
119     double precision function v(x, xmin)
120     implicit none
121     real*8 x, xmin
122
123     v = (1 - dexp(-(x - xmin)))*2.0d0 - 1.0d0
124     return
125
126     end
127
128 C-----
129 c FUNCTION : simpson
130 C-----
131
132     double precision function simpson(step, a, b, en, xmin)
133     implicit none
134     integer*4 step, i, factor

```

```

135     real*8 a, b, en, xmin
136     real*8 h, sum, x, g
137
138     i = 0
139     factor = 4
140     sum = 0.0d0
141     x = 0.0d0
142     h = (b-a)/dfloat(step)
143
144     do 300 i = 1, (step-1), +1
145         x = a+i*h
146         if (factor .eq. 2) then
147             sum = sum + 2.0d0*g(en,x,xmin)
148             factor = 4
149         else
150             sum = sum + 4.0d0*g(en,x,xmin)
151             factor = 2
152         endif
153 300 continue
154     sum = sum + g(en,a,xmin) + g(en,b,xmin)
155     sum = (h*sum)/3.0d0
156
157     simpson = sum
158     return
159
160     end
161
162 C-----
163 c FUNCTION : false_position
164 c PURPOSE : calculates and returns the root of the function
165 c : f, after position x1, to a tolerance of tolx
166 c : using the false position method. The parameter
167 c : step determines the accuracy to which the function
168 c : f can be calculated to.
169 C-----
170
171     double precision function false_position(n,gamma,step,
172 > start,tolx,xmin)
173     implicit none
174     integer*4 step, n
175     real*8 start, tolx, gamma
176     real*8 f, xmin
177     real*8 x1, x2, f1, f2, x3, f3, h
178
179     h = 0.00001d0
180
181     x1 = start
182     x2 = x1 + h
183     f1 = f(x1,n,gamma,step,xmin)
184     f2 = f(x2,n,gamma,step,xmin)
185
186     do while (f1*f2 .ge. 0.0d0)
187         x2 = x2 + h
188         f2 = f(x2,n,gamma,step,xmin)
189     end do
190
191     x3 = x2 - f2*(x2-x1)/(f2-f1)
192     f3 = f(x3,n,gamma,step,xmin)
193
194     do while (dabs(f3) .gt. tolx)
195         if (f1*f3 .lt. 0.0d0) then
196             x2 = x3
197         else
198             x1 = x3
199         endif
200         f1 = f(x1,n,gamma,step,xmin)
201         f2 = f(x2,n,gamma,step,xmin)
202         x3 = x2 - f2*(x2-x1)/(f2-f1)
203         f3 = f(x3,n,gamma,step,xmin)

```

```
204     end do
205
206     false_position = x3
207     return
208
209     end
```