Design Project: Robust Toggle Switch

Milo Knowles

May 11, 2020

1 IFFL Controller for Temperature Robustness

In the IFFL controller, we introduce two new proteins R_1 and R_2 , which serve as negative inducers for X_1 and X_2 , respectively. See Figure 1 for a system diagram of the controller.

1.1 Reactions

$$X_1 \stackrel{k_1}{\underset{k_2}{\rightleftharpoons}} X_1^d$$
 Dimerization reaction (1)

$$X_2 \stackrel{k_1}{\rightleftharpoons} X_2^d$$
 Dimerization reaction (2)

$$X_1^d + D_2 \stackrel{k_3}{\underset{k_4}{\longleftarrow}} C_2$$
 X_1 represses production of X_2 (3)

$$X_2^d + D_1 \stackrel{k_3}{\underset{k_4}{\rightleftharpoons}} C_1$$
 X_2 represses production of X_1 (4)

$$X_1^d + R_1 \stackrel{k_3}{\underset{k_4}{\rightleftharpoons}} C_{r1}$$
 Negative inducer R_1 sequesters X_1 (5)

$$X_2^d + R_2 \stackrel{k_3}{\underset{k_1}{\rightleftharpoons}} C_{r2}$$
 Negative inducer R_2 sequesters X_2 (6)

$$D_1 \xrightarrow{\beta} D_1 + X1$$
 Translation of X_1 , neglecting mRNA dynamics (7)

$$D_2 \xrightarrow{\beta} D_2 + X2$$
 Translation of X_2 , neglecting mRNA dynamics (8)

$$X_1 \xrightarrow{\gamma} \phi$$
 Degradation or dilution (9)

$$X_2 \xrightarrow{\gamma} \phi$$
 Degradation or dilution (10) (11)

R2 -# X2 X1 R1 -# X1 X2

Figure 1: A system diagram for the IFFL controller. On the left, X_2 represses production of X_1 , but is also repressed by the negative inducer R_2 . Symmetrically on the right, X_1 represses production of X_2 , but is also repressed by the negative inducer R_1 . When a change in temperature reduces the dissociation constant for X_1 and X_2 it also reduces the dissociation constant for the negative inducers R_1 and R_2 . This increases the amount of free X_1 and X_2 (i.e not sequestered by the negative inducers), which compensates for the reduced repressive ability of X_1 and X_2 .

1.2 System of ODEs

We can model the system with the following system of ODEs:

$$\frac{dX_1}{dt} = -2k_1X_1^2 + 2k_2X_1^d + \beta D_1 - \gamma X_1 \tag{12}$$

$$\frac{dX_2}{dt} = -2k_1X_2^2 + 2k_2X_2^d + \beta D_2 - \gamma X_2 \tag{13}$$

$$\frac{dX_1^d}{dt} = k_1 X_1^2 - k_2 X_1^d - k_3 X_1^d D_2 + k_4 C_2 - k_3 R_1 X_1^d + k_4 C_{1r}$$
(14)

$$\frac{dX_2}{dt} = -2k_1X_2^2 + 2k_2X_2^d + \beta D_2 - \gamma X_2$$

$$\frac{dX_1^d}{dt} = k_1X_1^2 - k_2X_1^d - k_3X_1^d D_2 + k_4C_2 - k_3R_1X_1^d + k_4C_{1r}$$

$$\frac{dX_2^d}{dt} = k_1X_2^2 - k_2X_2^d - k_3X_2^d D_1 + k_4C_1 - k_3R_2X_2^d + k_4C_{2r}$$
(13)

$$\frac{dC_1}{dt} = k_3 X_2^d D_1 - k_4 C_1 \tag{16}$$

$$\frac{dC_1}{dt} = k_3 X_2^d D_1 - k_4 C_1$$

$$\frac{dC_2}{dt} = k_3 X_1^d D_2 - k_4 C_2$$

$$\frac{dC_{1r}}{dt} = k_3 R_1 X_1^d - k_4 C_{1r}$$
(18)

$$\frac{dC_{1r}}{dt} = k_3 R_1 X_1^d - k_4 C_{1r} \tag{18}$$

$$\frac{dC_{2r}}{dt} = k_4 R_2 X_2^d - k_4 C_{2r} \tag{19}$$

(20)

1.3 Model Assumptions

• The rate constants k_3 and k_4 are the same for all reactions involving complexes C_1 , C_2 , C_{1r} , and C_{2r} .

Conservation Laws

In addition, we have the following four conservation laws:

$$D_{1,tot} = D_1 + C_1 (21)$$

$$D_{2,tot} = D_2 + C_2 (22)$$

$$R_{1,tot} = R_1 + C_{1r} (23)$$

$$R_{2,tot} = R_2 + C_{2r} (24)$$

which capture the fact that the DNA can either be free or in a complex, and similarly, that the negative inducers R_1 and R_2 can either be free or in their respective complexes.

Quasi-Steady-State Approximations 1.5