The radioactive isotope  $^{233}$ Pa can be produced following neutron capture by  $^{232}$ Th when the resulting  $^{233}$ Th decays to  $^{233}$ Pa. In the neutron flux of a typical reactor, neutron capture in 1 g of  $^{232}$ Th produces  $^{233}$ Th at of a rate of  $2.0 \times 10^{11}$  s<sup>-1</sup>.

- a) What are the activities (in Ci) of <sup>233</sup>Th and <sup>233</sup>Pa after this sample is irradiated for 1.5 hours?
- b) The sample is then placed in storage with no further irradiation so that the  $^{233}$ Th can decay away. What are the activities (in Ci) of  $^{233}$ Th and  $^{233}$ Pa after 48 hours of storage?
- c) The decay of <sup>233</sup>Pa results in <sup>233</sup>U, which is also radioactive. After the above sample has been stored for 1 year what is the <sup>233</sup>U activity in Ci? (Hint: it should not be necessary to set up an additional differential equation to find the <sup>233</sup>U activity.)

Recall: 1 Ci =  $3.7 \times 10^{10}$  s<sup>-1</sup>

Use the following masses for parts (a) and (b):

 $\begin{array}{l} {\rm n:~1.008665~u} \\ {}^1{\rm H:~1.007825~u} \\ {}^2{\rm H:~2.014102~u} \\ {}^{56}{\rm Fe:~55.934939~u} \\ {}^{98}{\rm Y:~97.922203~u} \\ {}^{135}{\rm I:~134.910048~u} \\ {}^{235}{\rm U:~235.043924~u} \end{array}$ 

Also, recall:  $1\mathbf{u} \cdot c^2 = 931.502 \text{ MeV}$ 

a) Calculate the Q-value of the reaction:

$$^{235}{
m U}\,+\,n 
ightarrow\,^{135}{
m I}\,+\,^{98}{
m Y}\,+\,3n$$

b) Calculate the average binding energy per nucleon in MeV of <sup>2</sup>H, <sup>56</sup>Fe, and <sup>235</sup>U.

a) Solve the first order differential equation

$$\frac{dy}{dx} + 3y = 0$$

b) Solve the second order differential equation (A and B are constants)

$$\frac{d^2y}{dx^2} - A^2y = B$$

The boundary condition is  $y(\pm \frac{1}{A}) = 0$ .

#### Answers

### Problem 2

a) The Q-value of a reaction is given by:

$$Q = [m(x) + m(X) - m(y) - m(Y)]c^{2}.$$

$$\begin{aligned} ^{235}\mathrm{U} + n &\to ^{135}\mathrm{I} + ^{98}\mathrm{Y} + 3n \\ Q &= [m(^{235}\mathrm{U}) + m(n) - m(^{135}\mathrm{I}) - m(^{98}\mathrm{Y}) - 3m(n)]c^2 \\ Q &= [235.043924\mathrm{u} + 1.008665\mathrm{u} - 134.910048\mathrm{u} - 97.922203\mathrm{u} - 3(1.008665\mathrm{u})]c^2 \\ Q &= 0.194343\mathrm{u} \cdot c^2 \\ \hline Q &= 181.031 \ \mathrm{MeV} \end{aligned}$$

b) The binding energy B(Z,A) for any nuclei can be found approximately from the equation:

$$M(Z, A) = Zm(^{1}H) + (A - Z)m_n - B(Z, A)/c^{2},$$

where M(Z, A),  $Zm(^{1}H)$  and  $m_n$  are experimentally calculated values. Per nucleon, the binding energy can be expressed as:

$$B_A(Z, A) = [Zm(^1H) + (A - Z)m_n - M(Z, A)]c^2/A.$$

 $^{2}\mathbf{H}$ 

$$B_A(1,2) = [m(^1H) + (2-1)m_n - M(1,2)]c^2/2$$

$$B_A(1,2) = [(1.007825\mathrm{u}) + (1.008665\mathrm{u}) - (2.014102\mathrm{u})]c^2/2$$

$$B_A(1,2) = 0.001194\mathbf{u} \cdot c^2$$

$$B_A(1,2) = 1.112 \text{ MeV}$$

 $^{56}$ Fe

$$B_A(26,56) = [26m(^1\text{H}) + (56 - 26)m_n - M(26,56)]c^2/56$$

$$B_A(26,56) = [26(1.007825\mathrm{u}) + 30(1.008665\mathrm{u}) - (55.934939\mathrm{u})]c^2/56$$

$$B_A(26, 56) = 0.009437 \mathbf{u} \cdot c^2$$

$$B_Z(26, 56) = 8.791 \text{ MeV}$$

 $^{235}U$ 

$$B_A(92,235) = [92m(^1\text{H}) + (235 - 92)m_n - M(92,235)]c^2/235$$

$$B_A(92,235) = [92(1.007825\mathrm{u}) + 143(1.008665\mathrm{u}) - (235.043924\mathrm{u})]c^2/235$$

$$B_A(92, 235) = 0.00814924u\mathbf{u} \cdot c^2$$

$$B_Z(92, 235) = 7.591 \text{ MeV}$$

a)

$$\frac{dy}{dx} = -3y\tag{1}$$

$$\frac{dy}{y} = -3 dx \tag{2}$$

$$\int \frac{dy}{y} = -3 \int dx \tag{3}$$

$$ln y = -3x + C$$
(4)

$$y = e^{-3x+C} \tag{5}$$

(6)

$$y = Ce^{-3x}$$

Solve the second order differential equation

$$\frac{d^2y}{dx^2} - A^2y = B$$

For the homogeneous equation,  $\frac{d^2y}{dx^2} - A^2y = 0$ , try  $y_c = X_1e^{Ax} + X_2e^{-Ax}$  as the complementary solution ( $X_1$  and  $X_2$  are constants). Then

$$\frac{d^2y}{dx^2} = X_1 A^2 e^{Ax} + X_2 A^2 e^{-Ax}$$

and

$$(X_1 A^2 e^{Ax} + X_2 A^2 e^{-Ax}) - A^2 (X_1 e^{Ax} + X_2 e^{-Ax}) = 0$$
$$A^2 (X_1 e^{Ax} + X_2 e^{-Ax}) - A^2 (X_1 e^{Ax} + X_2 e^{-Ax}) = 0$$

This is true.

For the inhomogeneous equation,  $\frac{d^2y}{dx^2} - A^2y = B$ ,  $y_P = -B/A^2$  is the only particular solution satisfying the equation. The general solution is the sum of the complementary and particular solutions,  $y = y_c + y_p$ .

$$y = X_1 e^{Ax} + X_2 e^{-Ax} - \frac{B}{A^2}$$

Now we can solve for  $X_1$  and  $X_2$ .

$$y(\frac{1}{A}) = X_1 e^{A(\frac{1}{A})} + X_2 e^{-A(\frac{1}{A})} - \frac{B}{A^2} = 0$$

and

$$y(-\frac{1}{A}) = X_1 e^{A(-\frac{1}{A})} + X_2 e^{A(\frac{1}{A})} - \frac{B}{A^2} = 0$$

We can note that in these two equations,  $X_1$  and  $X_2$  can be interchanged freely, and so must be equal. We will say,  $X_1 = X_2 = X$ . Then, we have

$$0 = Xe^{A(-\frac{1}{A})} + Xe^{A(\frac{1}{A})} - \frac{B}{A^2}$$
$$0 = X(e^{-1} + e^1) - \frac{B}{A^2}$$
$$X = \frac{B}{A^2(\frac{1}{e} + e)}$$

The we plug this X into the final solution, which gives

$$y = \frac{B}{A^2(\frac{1}{e} + e)}(e^{Ax} + e^{-Ax}) - \frac{B}{A^2}$$

Note that  $e^{Ax} + e^{-Ax}$  is similar in form to  $\cosh(Ax)$  which we could have also used to solve this problem.