

# NE250\_HW01\_mnegus-prob2-ntbk

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## 1 NE 250 – Homework 1

### 1.1 Problem 2

9/22/2017

The mean free path of a particle is given by the formula

$$\lambda = \frac{1}{\Sigma_t}$$

when scattering is considered to be isotropic.

The macroscopic scattering cross section can be reexpressed in terms of the number density of the material,  $n$ , and the microscopic cross section,  $\sigma_t$ . For mixtures, the macroscopic cross section is the sum of the macroscopic cross sections of its components.

$$\Sigma_{x,\text{mix}} = \sum_i \Sigma_{x,i} = \sum_i n_i \sigma_{x,i}$$

Additionally, the number density of the material can be found from the material density,  $\rho$ , the molar mass of the material,  $m$ , and Avogadro's number.

$$n = \frac{\rho N_A}{m_i}$$

Then the mean free path is

$$\lambda = \frac{1}{\sum_i \frac{\rho_i N_A \sigma_{t,i}}{m_i}}$$

```
In [1]: def mfp(energy, medium, xs, rho, m):
        # Mean free path [cm]
        return 1/macroXS_mix(energy, medium, rho, m, xs)

def macroXS_mix(energy, medium, rho, m, xs):
    # Macroscopic cross section of a mixture of isotopes [cm^2]
    Sig_t_mix = 0
    if energy == '14MeV': E = 0
    elif energy == '1MeV': E = 1
    elif energy == '0.05eV': E = 2
    for isotope in rho:
```

```

        Sig_t_mix += macroXS(rho[isotope],m[isotope],xs[isotope][E])
    return Sig_t_mix

def macroXS(rho_i,m_i,xs_i):
    # Macroscopic cross section of an isotope [cm^2]
    N_A = 6.022e23
    Sig_t = rho_i*N_A*xs_i*1e-24/m_i
    return Sig_t

```

### 1.1.1 Densities

**Air** Air is 75.5% nitrogen, 23.1% oxygen, and 1.3% argon by mass according to [NIST](#). We will assume that the nitrogen is entirely  $^{14}\text{N}$ , oxygen is entirely  $^{16}\text{O}$ , and  $^{40}\text{Ar}$ , which comprise more than 99.5% of their respective element naturally.

```

In [2]: # Density [g/cm^3]
rho_air = 1.20e-3
rho_a = {'N14': 0.755*rho_air,
        'O16': 0.231*rho_air,
        'Ar40': 0.013*rho_air
        }

```

**Water** Water is 11.2% hydrogen and 88.8% oxygen by mass. We are assuming the hydrogen is entirely  $^1\text{H}$  and the oxygen is entirely  $^{16}\text{O}$ .

```

In [3]: # Density [g/cm^3]
rho_water = 1
rho_w = {'H1': 0.112*rho_water,
        'O16': 0.888*rho_water
        }

```

**Uranium** Natural uranium is 99.3% U238 and 0.7% U235 by mass according to the [World Nuclear Association](#).

```

In [4]: # Density [g/cm^3]
rho_uranium = 19.1
rho_u = {'U235': 0.007*rho_uranium,
        'U238': 0.993*rho_uranium
        }

```

### 1.1.2 Molar Masses

```

In [5]: # Molar mass [g/mol]
m = {'H1': 1.008,
     'N14': 14.003,
     'O16': 15.995,
     'Ar40': 39.962,
     'U235': 235.044,

```

```

        'U238': 238.051
    }

```

### 1.1.3 Cross Sections

Microscopic cross sections are from ENDF/B-VII.1, found using [KAERI](#)

```

In [6]: # Cross sections at 14 MeV, 1 MeV, and 0.05 eV [b]
xs = {'H1': (0.687, 4.246, 20.673),
      'N14': (1.569, 2.345, 11.264),
      'O16': (1.592, 8.154, 3.852),
      'Ar40': (2.290, 3.135, 1.116),
      'U235': (5.831, 6.800, 472.885),
      'U238': (5.858, 7.068, 9.862)
    }

```

### 1.1.4 Calculations

```

In [7]: def loop_media(energy):
        print('Mean Free Path in')
        for medium in ['air', 'water', 'uranium']:
            if medium == 'air': rho = rho_a
            elif medium == 'water': rho = rho_w
            elif medium == 'uranium': rho = rho_u
            print('\t{}\t{} cm'.format(medium, round(mfp(energy, medium, xs, rho, m)

```

#### *a) 14 MeV neutrons in air, water, and uranium*

```

In [8]: energy = '14MeV'
        loop_media(energy)

```

```

Mean Free Path in
air      12773.78 cm
water    10.08 cm
uranium   3.53 cm

```

#### *b) 1 MeV neutrons in air, water, and uranium*

```

In [9]: energy = '1MeV'
        loop_media(energy)

```

```

Mean Free Path in
air      5643.26 cm
water     1.8 cm
uranium   2.93 cm

```

#### *c) 0.05 eV Neutrons in air, water, and uranium*

```
In [10]: energy = '0.05eV'  
         loop_media(energy)
```

Mean Free Path in

air	2086.21 cm
water	0.66 cm
uranium	1.57 cm