## 1) Molecular Polarizability - an Introduction

- When a beam of light is incident on a transparent material medium of refractive index different from that of its surroundings, the medium gets polarized.
- If a light wave of electric intensity  $\vec{E}$  goes through a molecule in the medium, it induces an optic moment in the molecule. The molecule is said to be polarized.
- Theoretically; therefore, polarizability should be a function of the incident light frequency.
- $\alpha$  can be regarded as a constant; unless, when measuring n.
- $P = \alpha E$ , P is the dipole moment, E is the electric field,  $\alpha$  is the polarizability of the molecule.
- In isotropic materials  $\alpha$  can be treated as a scalar of constant magnitude.
- If the material is anisotropic, the polarizability acquires directional property.
- For optically inactive molecules,  $\alpha$  is a symmetrical tensor.
- An optically active molecule, rotates the plane of polarization after it goes through the molecule. Dextorotatory component rotates the light clockwise, and Levorotatory components rotates the light counterclockwise.
- The mean molecular polarizability is given by:  $\alpha = \frac{2}{3} (b_L + b_T + b_V)$ . Where  $b_L$  is the longitudinal link polarizability,  $b_T$  is the link polarizability in the plane of the molecule or group containing the link and at right angles to it and  $b_V$  is the one normal to the plane.
- Individual chemical bonds can be associated with polarizability components along their lengths.

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