

# A novel method for determination of aragonite saturation state on the continental shelf of central Oregon using multi-parameter relationships with hydrographic data

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[1] We developed a multiple linear regression model to robustly determine aragonite saturation state ( $\Omega_{arag}$ ) from observations of temperature and oxygen ( $R^2 = 0.987$ , RMS error 0.053), using data collected in the Pacific Northwest region in late May 2007. The seasonal evolution of  $\Omega_{arag}$  near central Oregon was evaluated by applying the regression model to a monthly (winter)/bi-weekly (summer) water-column hydrographic time-series collected over the shelf and slope in 2007. The  $\Omega_{arag}$  predicted by the regression model was less than 1, the thermodynamic calcification/dissolution threshold, over shelf/slope bottom waters throughout the entire 2007 upwelling season (May–November), with the  $\Omega_{arag} = 1$  horizon shoaling to 30 m by late summer. The persistence of water with  $\Omega_{arag} < 1$  on the continental shelf has not been previously noted and could have notable ecological consequences for benthic and pelagic calcifying organisms such as mussels, oysters, abalone, echinoderms, and pteropods. **Citation:** Juranek, L. W., R. A. Feely, W. T. Peterson, S. R. Alin, B. Hales, K. Lee, C. L. Sabine, and J. Peterson (2009), A novel method for determination of aragonite saturation state on the continental shelf of central Oregon using multi-parameter relationships with hydrographic data, *Geophys. Res. Lett.*, 36, L24601, doi:10.1029/2009GL040778.

## 1. Introduction

[2] Since the preindustrial, atmospheric loading of  $\text{CO}_2$  from fossil fuel combustion and land use changes has driven an anthropogenic ocean uptake of  $146 \pm 20$  Pg C (updated from Sabine and Feely [2007]) and a corresponding average surface water pH change of 0.1 units [Feely *et al.*, 2004]. Accelerating emission rates and reduced buffering capacity will decrease pH by as much as 0.3–0.4 units by the end of this century under business-as-usual scenarios [Orr *et al.*, 2005]. Effects of these “ocean acidification” changes on marine organisms are still under intense study [Kleypas *et al.*, 2006; Fabry *et al.*, 2008; Doney *et al.*, 2009], but increased ocean  $\text{CO}_2$  content will result in a reduced

saturation state for calcium carbonate minerals and potentially deleterious impacts for organisms that form  $\text{CaCO}_3$  shells, including corals, pteropods, foraminifera, and commercially important shellfish and their larvae.

[3] The saturation state ( $\Omega$ ) of  $\text{CaCO}_3$  minerals is determined by the relationship:

$$\Omega = [\text{Ca}^{2+}][\text{CO}_3^{2-}]/K'_{sp}, \quad (1)$$

where  $K'_{sp}$ , the stoichiometric solubility product, is a function of temperature, salinity, pressure, and the particular mineral phase (aragonite or calcite). In a thermodynamic sense,  $\Omega > 1$  indicates mineral precipitation is favored and  $\Omega < 1$  indicates dissolution is favored, although biogenic calcification is subject to “vital effects” such as organic shell coatings and species-specific calcification mechanisms, and calcification/dissolution can occur when ambient-water  $\Omega$  values indicate opposing thermodynamic effects [Langdon *et al.*, 2003; Tunnicliffe *et al.*, 2009]. However, recent experiments indicate that  $\Omega < 1$  adversely impacts some organisms; Fabry *et al.* [2008] reported net dissolution in live pteropods within 48 hours of exposure to undersaturated water. Because aragonitic  $\text{CaCO}_3$  has a metastable crystalline structure and is  $\approx 50\%$  more soluble than calcite [Mucci, 1983], organisms that form aragonitic shells will likely be affected first, and perhaps most severely, by ocean acidification.

[4] Transient episodes of reduced aragonite  $\Omega$  ( $\Omega_{arag}$ ) have already been noted in productive eastern boundary upwelling systems such as the California current system [Feely *et al.*, 2008a]. Understanding the duration, intensity, and overall ecological impact of these events is a key need in economically and socially important coastal fisheries regions. Here we present an approach, updated from Feely *et al.* [2008b], to determine  $\Omega_{arag}$  from temperature and  $\text{O}_2$ , using data collected on a 2007 survey of North American Pacific coastal waters. We justify the approach with a statistical evaluation, and apply it to a hydrographic time-series from the central Oregon coast to evaluate seasonal changes in  $\Omega_{arag}$ .

## 2. Algorithm Development

[5]  $\Omega_{arag}$  is a function of temperature ( $T$ ), salinity ( $S$ ), pressure ( $P$ ), and the  $[\text{Ca}^{2+}]$  and  $[\text{CO}_3^{2-}]$  of seawater (equation (1)). Because  $[\text{Ca}^{2+}]$  changes are proportionally small in seawater, variations in  $\Omega_{arag}$  are largely determined by changes in  $[\text{CO}_3^{2-}]$ , which can be predicted from observations of dissolved inorganic carbon (DIC) and total

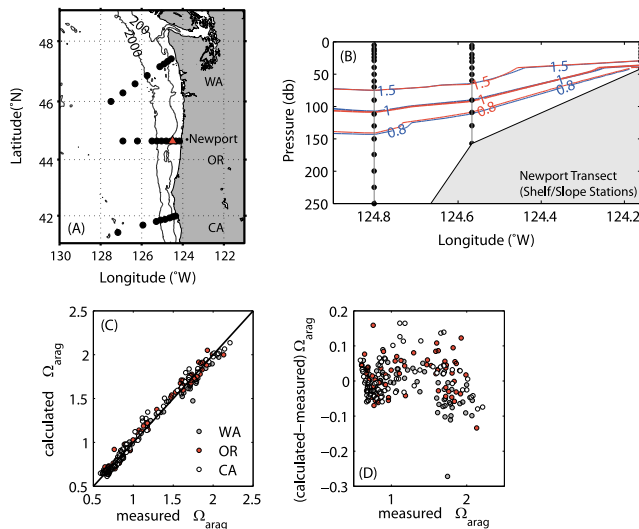
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**Figure 1.** (a) Region map showing location of three coastal transects used in developing the algorithm and location of NDBC buoy 46050 (red triangle). Sampling locations for the Newport time-series shown in Figure 2 are similar to those shown here, but with higher resolution over shelf/slope areas and a reduced seaward extent. (b)  $\Omega_{arag}^e$  (blue) and  $\Omega_{arag}^e$  (red) contours for transect collected near Newport, Oregon in late May 2007, with profiles (vertical lines) and sampling depths indicated. (c) Measured  $\Omega_{arag}^e$  and calculated  $\Omega_{arag}^e$ , color coded by transect location. Lack of geographic bias in residuals indicates that the algorithm applies for WA, OR, and N. CA coastal areas. (d) Residual ( $\Omega_{arag}^e - \Omega_{arag}^e$ ) versus  $\Omega_{arag}^e$  for PNW data, color coded as in C. All  $\Omega_{arag}^e$  values were determined using the regression model described by equation (3).

alkalinity (TA). DIC concentrations are governed by physics (solubility, surface gas exchange) and biology (photosynthesis/respiration) and therefore should be a function of  $T$ ,  $S$ , and either  $O_2$  or  $NO_3^-$  [Anderson and Sarmiento, 1994; Lee *et al.*, 2000]. TA can also be modeled as a function of  $T$  and  $S$  [Lee *et al.*, 2006]. We would therefore expect a predictive relationship for  $\Omega_{arag}$  as a function of  $T$ ,  $S$ ,  $P$ ,  $O_2$ ,  $NO_3^-$ , or a subset of these parameters.

[6] A hydrographic survey of the U.S. west coast in 2007 [Feely *et al.*, 2008a] allowed an opportunity to develop predictive relationships for  $\Omega_{arag}$  based on contemporaneous  $T$ ,  $S$ ,  $P$ ,  $O_2$ , and  $NO_3^-$  measurements. We first evaluated a linear additive model of the following form:

$$\Omega_{arag}^e = \beta_0 + \beta_1 T + \beta_2 S + \beta_3 P + \beta_4 O_2 + \beta_5 NO_3^-, \quad (2)$$

where  $\Omega_{arag}^e$  is the empirically predicted aragonite saturation state, and the coefficients  $\beta_i$  are empirical constants. We determined coefficients for equation (2) using an ordinary least-squares regression of  $\Omega_{arag}^e$  observations collected in the Pacific Northwest (PNW) region (transects of Washington, Oregon, and N. California coastal waters, Figure 1a), using only data in the 30–300 m depth range to minimize localized effects of surface warming, gas exchange and riverine inputs and to include only relevant source water masses for the shelf/slope region. Although all resulting regression coefficients

were significant, tests of collinearity among the independent variables (via pair wise regression and the variance inflation factor test, see Table 1) indicated that  $S$ ,  $O_2$ , and  $NO_3^-$  were too closely related, leading to potential errors in least-squares regression coefficients [Kutner *et al.*, 2004]. Stepwise regression and regression statistics ( $R^2$ , RMS error) subsequently identified  $O_2$  as the most robust predictor of the three collinear variables.

[7] A multiple linear regression of  $T$ ,  $P$ , and  $O_2$  yielded significant regression coefficients and reasonable regression statistics (Table 1). However, residuals for this relationship showed a strong bias, i.e., overestimation of  $\Omega_{arag}^e$  at minimum and maximum  $T$  and  $O_2$  (see Figure S1 of the auxiliary material).<sup>1</sup> This bias is likely the result of the non-linear dependence of  $CO_3^{2-}$  on TA and DIC, which arises in coastal waters with high  $pCO_2$  and significant contributions to TA from non-carbon species. We examined several possible non-linear terms and found that the bias could be minimized through the addition of an interaction term between  $T$  and  $O_2$  (Figure S1); when this term is added,  $P$  and  $T$  are no longer significant as predictor variables. To reduce large magnitudes of the product of  $T \cdot O_2$  and subsequent errors in the least-squares regression analysis (Table 1) [Kutner *et al.*, 2004], we normalized each term by subtracting a reference value for each variable, i.e.,

$$\Omega_{arag}^e = \alpha_0 + \alpha_1 (O_2 - O_{2,r}) + \alpha_2 (T - T_r) \cdot (O_2 - O_{2,r}). \quad (3)$$

Where  $\alpha$ 's indicate regression coefficients and  $T_r$  and  $O_{2,r}$  are values representative of upwelling source water in the PNW region ( $T_r = 8^\circ\text{C}$  and  $O_{2,r} = 140 \mu\text{mol/kg}$ , see Figure 2 and Table 1). The resulting model had improved regression statistics (Table 1) and resulted in  $\Omega_{arag}^e$  predictions that correctly reproduce both the magnitude and depth-distribution of  $\Omega_{arag}^e$  observations for the effective range experienced over the shelf/slope areas ( $\approx 0.6$  to  $2.2$ ) of the PNW region (Figure 1).

### 3. Model Evaluation and Caveats

[8] We evaluated the skill of the model described by equation (3) by comparison of the unexplained error in  $\Omega_{arag}^e$  and the ability to constrain  $\Omega_{arag}^e$  given analytical uncertainties in DIC and TA (2 and  $3 \mu\text{mol/kg}$ , respectively [Feely *et al.*, 2008a]). Uncertainty in  $\Omega_{arag}^e$  was determined by a Monte Carlo approach, in which DIC and TA inputs into the Matlab<sup>®</sup> program CO2SYS [van Heuven *et al.*, 2009] were varied randomly about chosen values for the PNW data, with standard deviations equal to analytical uncertainties. The  $1\sigma$  values of 1000 individually calculated  $\Omega_{arag}^e$  determinations,  $0.017/0.034$  for minimum/maximum  $\Omega_{arag}^e$  values in the PNW data ( $0.61/2.22$ , respectively), represent the theoretical lower limit for unexplained random error,  $\varepsilon$ , in any model used to predict  $\Omega_{arag}^e$ . The RMS error determined for the equation (3) model is close to, but still slightly higher than, the limit calculated for analytical uncertainties alone ( $\varepsilon$ ). Although adding new terms to the regression model causes the RMS error to approach  $\varepsilon$ , the contribution of these additional terms to the explained

<sup>1</sup>Auxiliary materials are available in the HTML. doi:10.1029/2009GL040778.

**Table 1.** Summary of Model Parameters, Coefficients, and Indicators Used in Model Selection<sup>a</sup>

Parameters	VIF <sup>b</sup>	R <sup>2</sup>	RMS Error	Coefficients $\pm$ STD Error <sup>c</sup>	Comments
$T, S, P, O_2, NO_3^-$	3.9, 24, 2.9, 35, 9.3	0.966	0.090	$\beta_0 = 6.3 \pm 1.7$ $\beta_1 = 9.5 \cdot 10^{-2} \pm 1.0 \cdot 10^{-2}$ $\beta_2 = -1.94 \cdot 10^{-1} \pm 5.0 \cdot 10^{-2}$ $\beta_3 = 8.6 \cdot 10^{-4} \pm 1.5 \cdot 10^{-4}$ $\beta_4 = 2.82 \cdot 10^{-3} \pm 4.5 \cdot 10^{-4}$ $\beta_5 = -3.7 \cdot 10^{-3} \pm 1.7 \cdot 10^{-3}$	$O_2, S,$ and $NO_3^-$ collinear (VIF > 5)
$T, O_2, P$	2.8, 3.8, 3.0	0.965	0.084	$\beta_0 = -0.521 \pm 7.0 \cdot 10^{-2}$ $\beta_1 = 7.74 \cdot 10^{-2} \pm 8.3 \cdot 10^{-3}$ $\beta_2 = 5.18 \cdot 10^{-3} \pm 1.3 \cdot 10^{-4}$ $\beta_3 = 1.16 \cdot 10^{-3} \pm 1.3 \cdot 10^{-4}$	Residuals show bias at high/low $O_2$ and $T$ (see Figure S1)
$O_2$	N/A	0.946	0.088	$\beta_0 = 1.145 \pm 6 \cdot 10^{-3}$ $\beta_1 = 4.99 \cdot 10^{-3} \pm 7 \cdot 10^{-5}$	Residuals show bias, as above
$(O_2 - O_{2,r}),$ $(T - T_r)(O_2 - O_{2,r})$	<b>1.5, 1.5</b>	<b>0.987</b>	<b>0.053</b>	$\alpha_0 = 9.242 \cdot 10^{-1} \pm 4.4 \cdot 10^{-3}$ $\alpha_1 = 4.492 \cdot 10^{-3} \pm 5.0 \cdot 10^{-5}$ $\alpha_2 = 9.40 \cdot 10^{-4} \pm 3.4 \cdot 10^{-5}$	$T_r = 8^\circ\text{C};$ $O_{2,r} = 140 \mu\text{mol/kg};$
$(T - T_r),$ $(O_2 - O_{2,r}),$ $(T - T_r)(O_2 - O_{2,r}),$ $(S - S_r)(O_2 - O_{2,r}),$ $(P - P_r)(O_2 - O_{2,r})$	9, 33, 7, 23, 19	0.990	0.043	$\alpha_0 = 9.079 \cdot 10^{-1} \pm 4.6 \cdot 10^{-3}$ $\alpha_1 = 3.37 \cdot 10^{-2} \pm 7.0 \cdot 10^{-3}$ $\alpha_2 = 3.4710^{-3} \pm 1.8 \cdot 10^{-4}$ $\alpha_3 = 7.49 \cdot 10^{-4} \pm 5.9 \cdot 10^{-5}$ $\alpha_4 = -1.32 \cdot 10^{-3} \pm 1.3 \cdot 10^{-4}$ $\alpha_5 = 5.8 \cdot 10^{-6} \pm 1.2 \cdot 10^{-6}$	$T_r = 8^\circ\text{C};$ $O_{2,r} = 140 \mu\text{mol/kg};$ $P_r = 200 \text{ dbar } S_r = 34$

<sup>a</sup>Bold denotes selected model. 227 observations used in each model.

<sup>b</sup>VIF: Variance inflation factor. See *Kutner et al.* [2004] for a full description, but briefly, the VIF is an objective measure of the inflation in coefficient uncertainty from poorly scaled or singular matrices (e.g., due to rounding errors during matrix inversion). The VIF is calculated as  $(1 - R^2)^{-1}$  for the regression of each variable versus the remaining independent variables; values given in the order parameters are listed. Values >5 indicate potential collinearity among predictor variables.

<sup>c</sup>Coefficients with standard error estimates for robust-fit multiple linear regression, which reduces the weight of outliers in the regression analysis. Coefficients correspond to order in which parameters appear. Following equations (2) and (3) in text,  $\beta$  and  $\alpha$  values are coefficients for regressions without/with a reference value subtracted.

variance is marginal (Table 1). To avoid overfitting, we rejected these models. A simple model based only on  $O_2$ , which was the strongest predictor variable of  $\Omega_{arag}^e$  ( $R^2 = 0.946$ , RMS error 0.088; Table 1) was also considered. However, the  $O_2$  model had a higher RMS error and a strong bias in residuals similar to that observed for the multiple linear regression of  $T, P$ , and  $O_2$  (Figure S1). Based on these observations, we chose the equation (3) model.

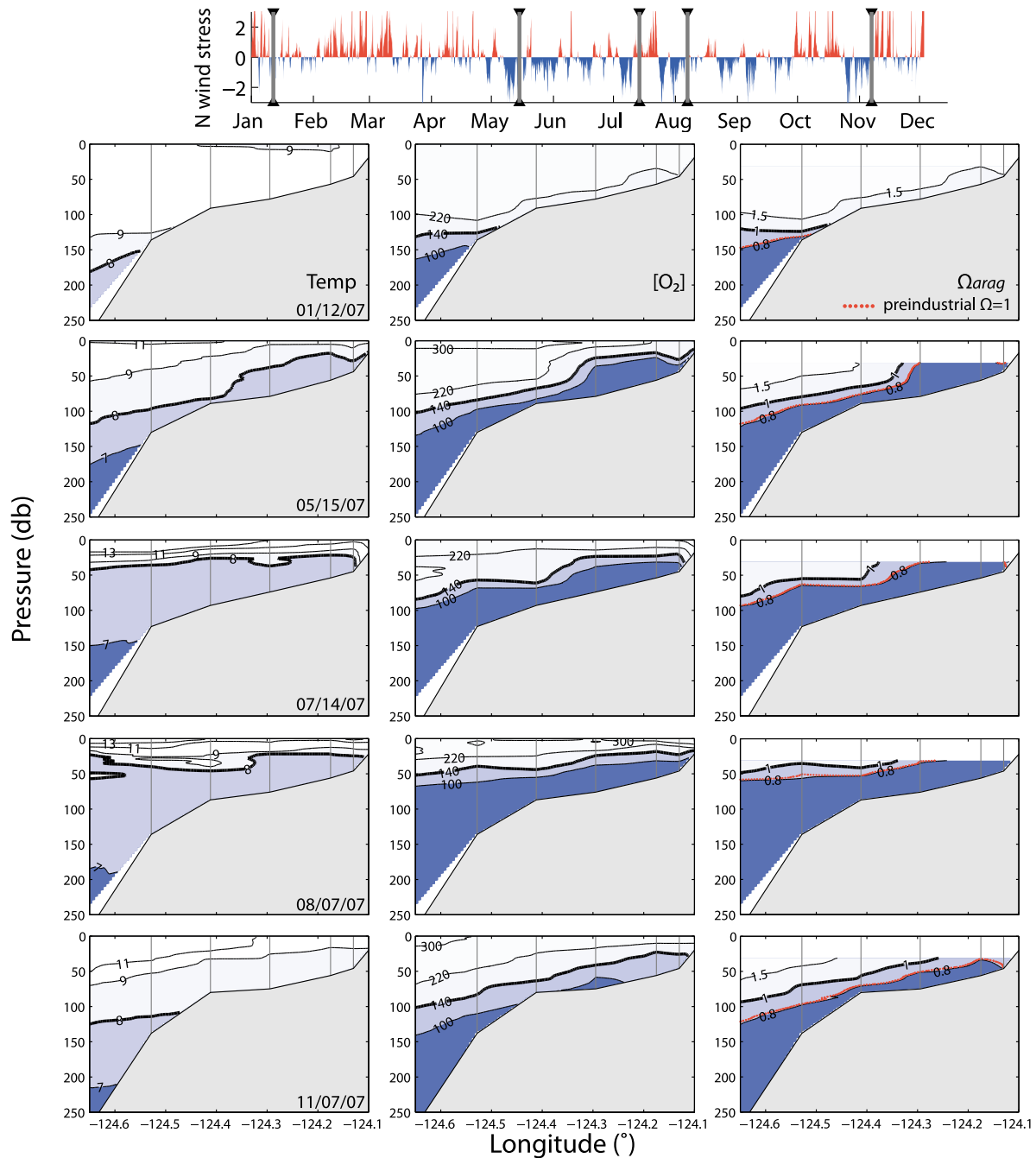
[9] Because the only  $\Omega_{arag}$  data available for algorithm development in this region are from late May 2007, we note there may be important caveats to a seasonal application of equation (3). However, three lines of evidence indicate seasonal application is justified. First, biologically-driven changes in  $\Omega_{arag}$  for the 30–300 m depth range (i.e., due to remineralization of organic matter over the productive summer months) are to a first order driven by changes in DIC rather than TA, since diatoms typically dominate coastal upwelling systems [*Lassiter et al.*, 2006]. DIC and  $O_2$  changes are expected to be proportional in remineralization zones that are not anoxic [*Hales et al.*, 2005; *Anderson and Sarmiento*, 1994], and therefore changes in DIC should be inherently captured in an algorithm involving  $O_2$ . Second, the  $T$ - $S$  (and  $T$ - $O_2$ ) range experienced spatially in the PNW data is similar to the range observed seasonally near Newport (see Figure S2), suggesting that the water masses present in the seasonal data are present in the regional PNW data. Finally, algorithm developments for the Southern California Bight region suggest no significant bias of algorithm development using only late May data (i.e., difference of measured and predicted values for August 2008 was 0.075 (S. Alin, unpublished data, 2009)). As more  $\Omega_{arag}$  data become available, the algorithm for this region

can be tested and refined. Nevertheless, these arguments point toward the ability to model the seasonal  $\Omega_{arag}$  dynamics near Newport with the data in hand.

[10] One potential time frame when algorithm predictions could deviate from observations is between February and May. PNW coastal waters experience intense river inputs during the rainy winter months, and the TA:DIC signature of these freshwaters is often different than in the open ocean [*Park et al.*, 1969]. Proportionality of  $[Ca^{2+}]$  to salinity, an assumption used in calculating  $\Omega_{arag}$ , may also change during these months. Consequently, we do not present predictions for this time period.

#### 4. Seasonal Evolution of $\Omega_{arag}^e$ on the Oregon Coast

[11] We calculated the seasonal evolution of  $\Omega_{arag}^e$  on the shelf and slope near Newport, Oregon with the model described by equation (3) and a time-series of  $T$  and  $O_2$  data (described by *Peterson and Keister* [2003]) collected on biweekly to monthly intervals in 2007. The central Oregon coast is located in the northern end of the California Current system and experiences seasonal upwelling during spring and summer months. The region has been well-studied with regard to the physical forcing driving seasonal and interannual variability in water properties (cf. the 2006 *Geophysical Research Letters* special issue devoted to this region). Selected sections of  $\Omega_{arag}^e$  (Figure 2) show a distinct seasonal cycle that is tightly coupled to upwelling dynamics near Newport. In January, the  $\Omega_{arag}^e = 1$  saturation horizon sits near the shelf break ( $\approx 125$  m), roughly at the depth horizon of the  $140 \mu\text{mol/kg } O_2$  contour and the



**Figure 2.** (bottom) Selected sections of  $T$  ( $^{\circ}\text{C}$ , left),  $[\text{O}_2]$  ( $\mu\text{mol/kg}$ , center), and  $\Omega_{\text{arag}}^e$  (right), for January to November 2007.  $\Omega_{\text{arag}}^e$  was calculated from  $T$  and  $\text{O}_2$  data (averaged in 1 db bins) using the regression model described by equation (3). Also shown is an estimated preindustrial  $\Omega_{\text{arag}}^e = 1$  line. Locations of profiles indicated by vertical lines. Note that  $\Omega_{\text{arag}}^e$  are only shown for depths greater than 30 m (see text for explanation). Propagated uncertainty in  $\Omega_{\text{arag}}^e$ , based on uncertainties of  $T$  and  $\text{O}_2$  data ( $0.003^{\circ}\text{C}$  and  $0.45 \mu\text{mol/kg}$ , respectively) is  $0.002$ . (top) Northward wind stress ( $\text{dynes cm}^{-2}$ ) from NDBC buoy 46050 (see Figure 1 for location), with upwelling-favorable winds (southward winds, negative wind stress) denoted in blue and downwelling-favorable winds (northward winds, positive wind stress) denoted in red. Dates of sections are designated by grey bars.

$9^{\circ}\text{C}$  isotherm. The  $1.5 \Omega_{\text{arag}}^e$  horizon is 25 m shallower, at  $\approx 100$  m. The onset of upwelling season begins in early May with the physical spring transition [Huyer *et al.*, 1979], during which wind forcing becomes predominantly equatorward, and offshore transport becomes positive. The offshore transport is compensated by the upwelling of cold, dense waters that are rich in DIC and nutrients, and poor in

$\text{O}_2$ . The strong upwelling event in mid-May (strongly negative N wind stress, blue lines in Figure 2 (top)) results in sharply up-warped iso-surfaces, and the outcropping of the  $0.8 \Omega_{\text{arag}}^e$  horizon to the upper 30 m from the mid-shelf (80 m isobath) to the coast. After the spring transition, persistent upwelling-favorable winds pull the  $1.0 \Omega_{\text{arag}}^e$  and the  $140 \mu\text{mol/kg}$   $\text{O}_2$  contour onto the shelf where they

remain through mid-November (Figure 2). Occasional poleward wind stress events (Figure 2, top) result in relaxation from upwelling, but the source water remains over shelf/slope regions. The late May transect used to formulate the algorithm (Figure 1b) occurred during one of these relaxation events.

[12] Throughout the remainder of the season  $\Omega_{arag}^e$  and  $O_2$  distribution show depletion on similar hydrographic surfaces, presumably as a result of biological activity (e.g., 1.0/1.5  $\Omega_{arag}^e$  and 140/220  $\mu\text{mol/kg}$   $O_2$  contours retain similar behavior). Between May and November the 1.0  $\Omega_{arag}^e$  contour reaches 30 m near-continuously over the inner shelf (i.e., from the 80 m isobath shoreward), with the exception of early October, when a strong downwelling event confines the low- $\Omega_{arag}^e$  water to the shelf-bottom (not shown). Over the outer shelf and slope, the 1.5  $\Omega_{arag}^e$  horizon shoals to less than 30 m by mid-July and the 1.0 horizon shoals to 50 m by mid-August (Figure 2). After the onset of persistent downwelling-favorable winds in mid-November the 1.0  $\Omega_{arag}^e$  and 140  $\mu\text{mol/kg}$  contours retreat back to the shelf-break/slope region, similar to conditions predicted for January 2007.

[13] The coupling of low  $\Omega_{arag}$  state and physically-driven upwelling dynamics would be expected, given the high DIC (low pH) signature associated with upwelling source waters [Hales et al., 2005]. The absolute magnitude of  $\Omega_{arag}$  over the coastal shelf regions, however, is largely unknown, due to a lack of depth-resolved DIC and TA measurements. This model therefore provides previously unattainable insight into both the magnitude of  $\Omega_{arag}$  and how it relates to seasonal hydrography changes on the central Oregon shelf. The range in  $\Omega_{arag}^e$  experienced seasonally over the shelf (e.g., 0.5–1.4 and 0.8–1.8 for the mid-shelf at 80 and 30 m, respectively) is also much greater than the uncertainty in model predictions (0.053). This favorable signal to noise ratio makes the region particularly amenable to this approach, compared to open ocean subtropical regions where the seasonal range is considerably less [Doney et al., 2009].

[14] An obvious question to ask is: What is the anthropogenic contribution to  $\Omega_{arag}$  on the central Oregon shelf? We used the density-anthropogenic  $\text{CO}_2$  relationship presented by Feely et al. [2008a, supplement] to correct observed DIC in PNW waters for anthropogenic  $\text{CO}_2$  input (20–40  $\mu\text{mol/kg}$ ) and calculated a “preindustrial”  $\Omega_{arag}^e$  for our data. A parallel algorithm with the same form as equation (3) was fitted to the data ( $R^2 = 0.989$ ) and used to predict the preindustrial  $\Omega_{arag}^e = 1$  horizon for the time-series data (Figure 2). This preindustrial  $\Omega_{arag}^e = 1$  threshold very closely follows the 2007  $\Omega_{arag}^e$  0.8 isoline. Therefore, within the ability to estimate anthropogenic  $\text{CO}_2$  content in coastal waters ( $\pm 50\%$  [Feely et al., 2008a]), undersaturation over shelf/slope bottom waters is likely a natural phenomena, but an anthropogenic reduction in  $\Omega_{arag}$  by 0.2 units has caused a shoaling of the 1.0 horizon by  $\approx 25\text{m}$  (shelf/slope) to  $\approx 40\text{m}$  (offshore). Exposure of pelagic communities to undersaturated water may therefore be lengthened or intensified by anthropogenic  $\text{CO}_2$  input.

## 5. Implications

[15] The persistence of water with  $\Omega_{arag} < 1$  over the shelf throughout the May–November upwelling season has

not been previously noted. Although it is unclear how organisms on the central Oregon coast are directly affected by these conditions, laboratory experiments have indicated potentially deleterious impacts for organisms exposed to waters with  $\Omega_{arag} < 1$  [Kleypas et al., 2006; Fabry et al., 2008; Doney et al., 2009]. A clear application of the regression model presented here is to explore effects of low  $\Omega_{arag}$  on shelf communities when DIC and TA data are unavailable. Preliminary examination of historical pteropod abundance data from the Oregon coast from the last 20 years (B. Peterson, unpublished data, 2009) indicates that pteropods are generally found where upwelling water is not; their abundances are maximum in offshore waters outside of the upwelling region and peak over the shelf only during winter or El Niño events, when upwelling is suppressed. In-depth examination of these data and other historical records may provide insight into adaptations organisms use to cope with low  $\Omega_{arag}$  conditions.

[16] Bakun [1990] and Snyder et al. [2003] have suggested that upwelling intensity is likely to increase under future warming climate scenarios. Because the transit time of upwelling source waters from last atmospheric exposure to the sites of local upwelling are on the order of decades [Feely et al., 2008a], additional anthropogenic  $\text{CO}_2$  is already “in the pipeline” in the ocean interior, and will continue to decrease coastal  $\Omega_{arag}$  well into this century, regardless of atmospheric  $\text{CO}_2$  rise scenarios. Impacts of these changes will be better understood as studies of the seasonality in  $\Omega_{arag}$  and effects on coastal organisms emerge. The  $\Omega_{arag}^e$  relationship presented here (equation (3)) will need to be adjusted on 5–10 year intervals to account for the additional anthropogenic  $\text{CO}_2$  input.

[17] A key advantage of the ability to estimate  $\Omega_{arag}$  using commonly available hydrographic parameters ( $T$ ,  $O_2$ ) is the capability to hindcast  $\Omega_{arag}$  from historical datasets to explore relationships with previously documented ecological/physical observations, provided corrections for reduced anthropogenic  $\text{CO}_2$  in prior data, if significant, can be taken into account. For example, regression model development efforts by T. Kim et al. (Prediction of East/Japan Sea acidification over the past 40 years using a multiple-parameter regression model, submitted to *Global Biogeochemical Cycles*, 2009) highlight the importance of ventilation events for determining subsurface (50–500 m)  $\Omega_{arag}$  in a 50-year hydrographic time-series in the East/Japan Sea. Continued refinement of  $\Omega_{arag}^e$  regression models for the PNW and other coastal regions (Kim et al., submitted manuscript, 2009; S. R. Alin et al., manuscript in preparation, 2009) as more  $\Omega_{arag}$  data become available will significantly enhance our understanding of the sensitivity of coastal regions to future  $\text{CO}_2$ -chemistry changes and warming.

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