

# Chapter 5 - Second Law of Thermodynamics

## Lecture 17 Sections 5.9

MEMS 0051 Introduction to Thermodynamics

Mechanical Engineering and Materials Science Department  
University of Pittsburgh



# Student Learning Objectives

Chapter 5 - Second  
Law of  
Thermodynamics

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Learning Objectives

5.9 - Real vs. Ideal  
Machines

Summary

At the end of the lecture, students should be able to:

- Analyze real and ideal machines (heat engines, refrigerators, heat pumps) using the thermodynamic temperature scale



- ▶ Using the thermodynamic temperature scale, we can express the efficiency of the following devices in terms of temperature
- ▶ Heat engine:

$$\eta_{\text{Carnot}} = 1 - \frac{T_L}{T_H}$$

- ▶ Refrigerator:

$$\beta_{\text{Carnot}} = \frac{T_L}{T_H - T_L}$$

- ▶ Heat pump:

$$\beta'_{\text{Carnot}} = \frac{T_H}{T_H - T_L}$$



- ▶ Real machines have irreversibilities, which means the efficiency is less than that of ideal machines
- ▶ Heat engine:

$$\eta_{\text{real}} = 1 - \frac{Q_L}{Q_H} \leq 1 - \frac{T_L}{T_H}$$

- ▶ Refrigerator:

$$\beta_{\text{real}} = \frac{Q_L}{Q_H - Q_L} \leq \frac{T_L}{T_H - T_L}$$

- ▶ Heat pump:

$$\beta'_{\text{real}} = \frac{Q_H}{Q_H - Q_L} \leq \frac{T_H}{T_H - T_L}$$



# Example #1

- ▶ An inventor claims to have developed a power cycle that outputs 410 [kJ] of work for an energy input of 1,000 [kJ]. The system operates between a high-temperature reservoir of 500 [K] and rejects heat to a low temperature reservoir of 300 [K]. Evaluate the inventor's claim.



## Example #2

- ▶ A refrigerator maintains the cold compartment at  $-5\text{ }^{\circ}\text{C}$  while the ambient air is  $22\text{ }^{\circ}\text{C}$ . The rate of heat removed from the cold compartment is  $8,000\text{ [kJ/hr]}$  while the power input is  $3,200\text{ [kJ/hr]}$ . Determine C.O.P. of the real and ideal refrigerator.



## Example #3

- ▶ A building uses a heat pump to maintain a temperature of  $22\text{ }^{\circ}\text{C}$  while the ambient temperature is  $10\text{ }^{\circ}\text{C}$ . The heat delivered by the heat pump is  $500\text{ [MJ]}$  per day. Determine the minimum theoretical work required for this heat pump to operate.



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Summary

At the end of the lecture, students should be able to:

- ▶ Analyze real and ideal machines (heat engines, refrigerators, heat pumps) using the thermodynamic temperature scale
- ▶ The performance of a real machine is quantified in terms of heat inputs/outputs, whereas that of an ideal machine is quantified using the Carnot efficiency via the thermodynamic temperature scale.





# Suggested Problems

- ▶ 5.101, 5.104, 5.106, 5.107, 5.109, 5.110, 5.112, 5.114, 5.120

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