

# MoBioTools: A Toolkit to Setup QM/MM Calculations

#### **Tutorials**

- 1 Reduction Potential: Guanine in Acetonitrile.
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- 3 Absorption Spectrum: p-Diaminoazobenzene Integrated into the human  $Na_v1.4$  Channel.
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#### 1 REDUCTION POTENTIAL: GUANINE IN WATER.

In this section, we will determine the one-electron oxidation potential of guanine in water. In order to obtain this property we will run classical molecular dynamics (MD) simulations to explore the conformational space of both the neutral and the cationic version of the system using AMBER. Afterwards, we will use MoBioTools to obtain an ensemble of geometries from each of the trajectories. From the neutral ensemble of geometries we will obtain an average value of the vertical ionization energy (VIE) while the average vertical attachment energy (VAE) will be obtained from the cationic ensemble of configurations. Under the framework of the Marcus theory these two properties can give an approximate value of the reduction free energy of the half-reaction:

$$\Delta G_{red} = \frac{1}{2} \left( \langle VIE \rangle_R - \langle VAE \rangle_O \right) - G_e(g) \tag{1}$$

The term  $G_e(g) = -0.867$  kcal/mol is the free energy of the electron in the gas phase according to the Fermi-Dirac statistics. Additionally, the one-electron oxidation potential  $\Delta E_{red}$  can be related to  $\Delta G_{red}$  by the following equation:

$$\Delta E_{red} = \frac{\Delta G_{red}}{nF} - E_{red,SHE}^0 \tag{2}$$

where  $E_{red,SHE}^0$  is the absolute one-electron oxidation potential of the reference electrode, which in this case is the standard hydrogen electrode ( $E_{red,SHE}^0 = 4.28 \text{ V}$ ).

#### 1.1 Setup

We will start drawing a guanine molecule in a visualization program. These structures can be taken from any other reference or source. In our case, we have drawn it with the use of the Avogadro package and we have saved the structure as a PDB (see Figure 1). At this point, it is important to highlight a relevant issue. Traditional force fields (FF) are usually parameterized for neutral species. Thus, the use of traditional FFs are not suitable for reproducing the behaviour of a cationic species. In this context, a reparameterization of both the neutral and the cationic species is strongly recommended. In the original paper, we have used the Seminario method for that reparameterization after an optimization of the guanine moiety in both states so that the behaviour of the neutral species is different from the cation. In this tutorial, we will use GAFF2 to model the FF of both species. In addition, we will use ESP charges derived from quantum optimizations of the species to model the electrostatic interactions in a realistic way. However, results will not be accurate at all due to that lack of differentiation in the behaviour of the cation and the neutral molecule.

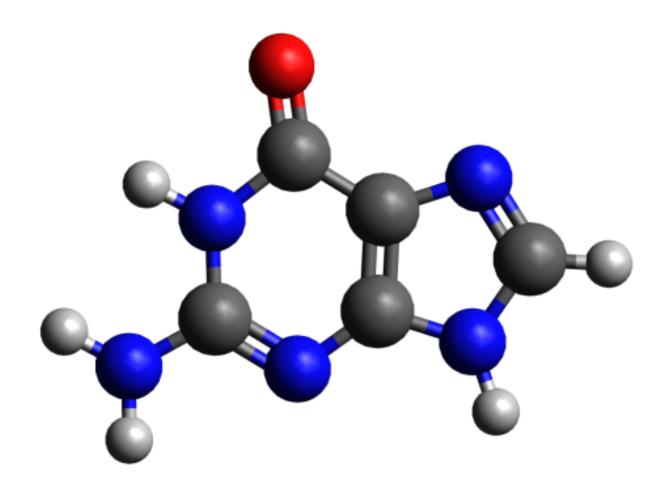


Figure 1: Guanine molecule.

Choose a quantum chemistry package to perform a geometry optimization of guanine in the neutral and cationic state. In our case, we have performed geometry optimizations using the ORCA package at the B3LYP/6-311G(d) level of theory and the CPCM solvent model to account for solvent effects:

```
! Opt B3LYP 6-311G(d) CPCM(Water)
%pal nprocs 8 end
%maxcore 3000
%scf
  maxiter 15000
end
%geom
  maxiter 1500
end
* xyzfile 0 1 guanine.pdb
```

Listing 1: orca\_opt.in

Then, we have used the application orca\_2aim to obtain the wave function of both species: orca\_2aim orca\_opt

And finally, we have used Multiwfn to obatin the ESP charges:

```
orca_opt.wfx
7
18
1
y
0
0
```

Listing 2: orca\_opt.wfx

With this procedure, we obtained two files with the ESP charges of the neutral and the cationic species. At this point, we will obtain the atom types and the associated parameters from the topology of the PDB of the optimized species. In order to do this, we will use the antechamber application implemented in AMBER typing the following command:

```
antechamber -i filename.pdb -fi pdb -o filename.mol2 -fo mol2 -c bcc -nc 0 -
    m 0
antechamber -i filename.pdb -fi pdb -o filename.mol2 -fo mol2 -c bcc -nc 1 -
    m 1
```

where the first line corresponds to the neutral species and the second line to the cationic molecule. If we now open these mol2 files we will see a column with the calculated charges by antechamber:

We have to replace that column for the one obtained from our ESP calculation at the correspondent species (see Figure 2). Let's now check if there is any parameter left and required in our FF. For that, there exists the parmchk2 application, which assigns the remaining parameters in terms of their similarity.

```
parmchk2 -i filename.mol2 -o filename.frcmod -f mol2
```

Now, it is the time to build our solvated system. For that purpose, we will use the tleap application also implemented in AMBER. First, generate an input (filename.in) like the following:

```
source leaprc.gaff2
source leaprc.water.tip3p
loadamberparams filename.frcmod
UNL = loadmol2 filename.mol2
solvateoct UNL TIP3PBOX 12
(addions2 UNL Cl- 0)
saveamberparm UNL filename.prmtop filename.rst7
quit
```

Listing 3: filename.in

and type the following command:

```
tleap -i filename.in
```

In the input, the line in parenthesis is only required in the cationic system since it is important to neutralize the positive charge of guanine. The generated files (filename.prmtop and

```
g«TREPOS»MOLECULE
   16
          17
                                 0
SMALL
bec
@<TRIPOS>ATOM
      1 H
2 M
3 C
4 M1
                                                                                      0.3946861346
                       -1.8338
                                     5.9500
                                                   3.6618 hm
                                                                        1 UNL
                                     5.7100
                                                  2.9690 na
                                                                                     -0.4027638526
0.2227787333
-0.6028472170
                       -1.7290
                                                                        1 UNL
                       -1.6840
                                                   1.6220 oc
                                                                        1 UNL
                                      5.9980
                       -2.7340
                                     5.5590
                                                   0.9730 nd
                                                                        1 UNL
       5 C1
                       -3.5110
                                                   1.9478 od
                                     4.9490
                                                                        1 UNL
      6 CZ
7 O
                                                   1.0440 c
                       -4.7710
                                     4.2630
                                                                        1 UNL
                       -5.4730
                                     4.0880
                                                   0.8550 o
                                                                        1 UNL
                       -5.1900
-4.5050
      8 M2
                                     3.8110
                                                   3.1140 n
                                                                        1 UML
      9 C3
                                                                        1 UNL
                                      3.9510
                                                   4.2990 cd
      10 H3
                                                   5.4000 nh
                       -5.0700
                                      3.4300
                                                                        1 UNL
      11 84
                       -3.3370
                                      4.5660
                                                   4.3850 nc
                                                                        1 UNL
     12 64
                                     5.8368
                       -2.8980
                                                   3.1970 oc
                                                                        1 UNL
      13 H1
                       -6.8818
                                     3.3286
                                                   3.1150 hn
                                                                        1 UML
                                                                                      6.4349481222
      14 HZ
                       -0.8500
                                     6.5310
                                                   1.1800 hs
                                                                        1 UNL
                                     2.9480
                                                   5.3910 hn
      15 H3
                       -5.9530
                                                                        1 UNL
      16 H4
                       -4.5830
                                      3.5150
                                                   6.2830 hn
                                                                        1 UNL
g«TREPOS»80ND
                    2 1
                  3 1
12 1
4 2
14 1
     2
     5
                  12 2
     8
            5
                   Ø
     9
            6
     10
            8
     12
            8
                   13.
     13
            9
                   10 1
           10
                   15
     16
           18
                   16 1
e<TRIPOS>SUBSTRUCTURE
                      1 TEMP
     1 UNL
                                                                 e ROOT
```

Figure 2: mol2 file where the highlighted region correspond to the charges that must be substituted by the ones calculated by a quantum chemistry package.

filename.rst7) are the parameters and the coordinates files which will be given to AMBER in order to run the classical molecular dynamics situation. Note that this process must be done for the neutral and the cationic species so that two different simulations can be performed, each of them in the correspondent conformational space.

# 1.2 Molecular Dynamics

At this point, we are able to perform the MD simulations. The sampling procedure will be divided in four different stages: minimization, heating, pressure equilibration and production. For the minimization, the following input (01\_Min.in) will be given to AMBER:

```
Minimization input file in explicit solvent
&cntrl
   ! Minimization options
                ! Turn on minimization
  ntx=1,
  irest=0,
  maxcyc=10000, ! Maximum number of minimization cycles
  ncyc=5000, ! Number of steps of Steepest descent algorithm
  ntr=0,
  ntb=1,
                  ! Periodic boundary conditions
   ! Potential energy function options
              ! Cutoff
   cut = 12.0,
   fswitch=10.0, ! Switching function
   ! Control how often information is printed to the output file
               ! Print energies every 100 steps
  ntpr=100,
  ntwx=0.
```

Listing 4: 01\_Min.in

In order to perform the minimization, type the following command:

```
pmemd.cuda_SPFP -0 -i 01_Min.in -o 01_Min.out -p filename.prmtop -c filename
.rst7 -r 01_Min.ncrst -ref filename.rst7 -inf 01_Min.mdinfo &
```

Once the minimization is done, let's perform the heating using the following input (02\_Heat.in):

```
Heat
 &cntrl
                  ! Not a minimization
  imin=0,
 ntx=1,
  irest=0,
  nstlim=500000, ! Number of steps
  dt = 0.002,
                  ! Timestep
  ntf=2,
                   ! SHAKE algorithm
                  ! SHAKE algorithm
  ntc=2,
  tempi=0.0,
                  ! Initial Temperature
               ! Final Temperature
  temp0=300.0,
```

```
ntpr=100,
                   ! Frequency of printing
  ntwx=100,
                   ! Fequency of snapshot
  cut=12.0,
                    ! Cutoff
  fswitch=10.0,
                   ! Switching function
 ntb=1,
                   ! Periodic boundary conditions
 ntp=0,
                    ! No pressure control
 ntr=0,
 ntt=3,
                   ! Langevin thermostat
  gamma_ln=2.0,
                   ! Collision frequency
 nmropt=1,
                   ! Restraints
 ig=-1,
                    ! Random seed
 /
&wt type='TEMP0', istep1=0, istep2=250000, value1=0.0, value2=300.0 /
&wt type='TEMP0', istep1=250001, istep2=500000, value1=300.0, value2=300.0 /
&wt type='END' /
                               Listing 5: 02_Heat.in
  and type:
pmemd.cuda_SPFP -O -i O2_Heat.in -o O2_Heat.out -p opt.prmtop -c O1_Min.
   ncrst -r 02_Heat.ncrst -ref 01_Min.ncrst -x 02_Heat.nc -inf 02_Heat.
   mdinfo &}
  For the pressure equilibration, we will use the model input
Equil
&cntrl
  imin=0,
 ntx=7,
  irest=1,
  nstlim=500000,
  dt = 0.002,
 ntf=2,
 ntc=2,
  temp0=300.0,
 ntpr=50000,
 ntwx = 50000,
 ntwr=50000,
  cut=12.0,
  fswitch=10.0,
  ntb=2,
                   ! Pressure control (Berendsen barostat)
 ntp=1,
 ntr=0,
 ntt=3,
  gamma_ln=2.0,
  pres0=1.0,
                   ! Pressure 1 bar
  comp=44.6,
                   ! Compressibility coefficient
  taup=2.0,
                   ! Pressure relaxation time
  ig=-1,
```

```
Listing 6: 03_Equil.in
   and type:
pmemd.cuda_SPFP -O -i 03_Equil.in -o 03_Equil.out -p opt.prmtop -c 02_Heat.
   ncrst -r 03_Equil.ncrst -ref 02_Heat.ncrst -x 03_Equil.nc -inf 03_Equil.
   mdinfo &}
   Finally, for the production stage we will use the input displayed below
Prod
 &cntrl
  imin=0,
  ntx=7,
  irest=1,
  nstlim=100000000,
  dt = 0.002,
  ntf=2,
  ntc=2,
  temp0 = 300.0,
  ntpr=20000,
  ntwx=20000,
  ntwr=20000,
  cut=12.0,
  fswitch=10.0,
  ntb=2,
  ntp=1,
                     ! Pressure control (Berendsen barostat)
  ntr=0,
  ntt=3,
  gamma_ln=2.0,
  pres0=1.0,
                     ! Pressure 1 bar
  comp=44.6,
                    ! Compressibility coefficient
                     ! Pressure relaxation time
  taup=2.0,
  ig=-1,
                                Listing 7: 04_Prod.in
   and type:
```

#### 1.3 MoBioTools

mdinfo &

The O4\_Prod.nc file contains the production trajectory. The full MD protocol must be done twice, one for the neutral guanine and another one for the cation parameterization. Those two O4\_Prod.nc files will be the files that we will use to obtain the ensemble of geometries of the neutral and the cationic species using MoBioTools.

```
&main
tpl = orca
top = filename.prmtop
traj = 04_Prod.nc
qmmask = :UNL
geoms = variabless
solvmask =:WAT
&end
```

Listing 8: orca.main.in

```
&header
&end

&route
!B3LYP/6-311G*
&end
&externchg
&end
&chgspin
0 1 (1 2)
&end
```

Listing 9: orca.tpl.in

If many snapshots are needed, you can use the following script:

```
#!/bin/bash
for number in ` seq 200 20 5000`; do
    sed -i "s/variabless/$number/g" orca.main.in
    main_qminputs.py -i orca.main.in -t orca.tpl.inp
    sed -i "s/$number/variabless/g" orca.main.in
done
```

Listing 10: more\_geoms.sh

#### 1.4 One-electron Oxidation Potential

Now, it is the time to run all the ORCA inputs generated for both, the neutral and the cationic species. For each trajectory, two sets of calculation must be performed: one for the neutral species and one for the cation. In this way, four sets of calculations must be conducted: neutral species from the neutral trajectory, cationic species from the neutral trajectory, neutral species from the cationic trajectory and cationic species from the cationic trajectory. Once these calculations are finished, the only thing required is to take the electronic energies of each geometry and use Eq. 1 to obtain the one-electron oxidation potential of guanine. In out case, this procotol (with Seminario reparameterization) revealed a one-electron oxidation potential of  $1.28 \pm 0.65$  V.

#### 2 ENERGY DECOMPOSITION ANALYSIS: TYROSINE IN WATER.

This section aims at illustrating the ability of the MoBioTools toolkit to easily process MD trajectories into input files for the calculation of interaction energies and their different quantum mechanical energy contributions for a system consisting of two fragments, at the QM/MM level of theory. The chosen test system consists of the aminoacid tyrosine in a water solution, which undergoes configurational sampling by means of classical MD, followed by the Energy Decomposition Analysis (EDA) of the resulting geometry ensemble. An electron density based QM/MM-EDA scheme[1, 2, 3] is used for such purpose, allowing the decomposition of the total interaction energy in its electrostatic, polarization (induction and dispersion), and Pauli or exchange-repulsion components.

### 2.1 Setup

The geometry file of tyrosine (tyr.xyz) is taken from reference4. This file has to be transformed into .pdb format. In order to do that, we open the file with PyMOL and click on File > Export Molecule > Select tyr > Save

There are some changes that have to be performed in the obtained tyr.pdb file:

- First column: Change the record type HETATM by ATOM by typing sed -i 's/HETATM/ATOM /g' tyr.pdb
- 2. Third column: Change atom name from N, C, C, ... to N1, C2, C3, ...
- 3. Fourth column: When the number is added to the atom name, the column is moved one placed, so it is necessary to delete one space.
- 4. Fifth column: Change to 1 because there is only one residue.

Then, we generate tyr.mol2 file using antechamber module from AmberTools20 by writing

```
antechamber -i tyr.pdb -fi pdb -o tyr.mol2 -fo mol2 -c bcc -at gaff -nc 0
```

where the options -i, -fi, -o, -fo, -c, -at and -nc correspond to the input name, input format, output name, output format, method to determine the charges, atom type, and the total charge of the system, respectively.

Although this tyr.mo12 file contains charges, we calculate Merz-Singh-Kollman (MK) charges with B3LYP functional and 6-311g(d) basis set using Gaussian09. When this calculation has finished, we copy MK charges to the tyr.mo12 file (last column).

The next step is to check if there is any missing parameter using parmchk2 module from AmberTools20 by this command

```
parmchk2 -i tyr.mol2 -o tyr.frcmod -f mol2
```

Where the options -i, -o, and -f correspond to the input file, the output file and the input format, respectively.

Then, we generate the library file with tleap module from AmberTools20 by typing

```
tleap -f tleap.inp

source leaprc.gaff2  # source leaprc file for gaff
source leaprc.water.tip3p  # source leaprc file for tip3p
loadamberparams tyr.frcmod  # load tyr parameters
UNK=loadmol2 tyr.mol2  # load whole system
saveoff UNK tyr.lib  # save library of tyr
quit
```

Listing 11: tleap.inp

And finally, we solvate the system using again *tleap* module with the same command as before

```
tleap -f tleap2.inp
    source leaprc.gaff2
                                                      # source leaprc file
                                                        for gaff
                                                      # source leaprc file
    source leaprc.water.tip3p
                                                         for tip3p
    loadamberparams tyr.frcmod
                                                      # load tyr parameters
    loadoff tyr.lib
                                                      # load tyr library
    TYR=loadpdb tyr.pdb
                                                      # load whole system
    solvateOct TYR TIP3PBOX 12.0
                                                      # solvate system
                                                        with water with
                                                         truncated octahedron
                                                        box of 12.0 A
    addions2 TYR Na+ 0
                                                      # add ions
    addions2 TYR Cl- 0
                                                      # add ions
    savepdb TYR tyr_solv.pdb
                                                      # save system
    saveamberparm TYR tyr_solv.prmtop tyr_solv.rst7 # save parameters
    quit
```

Listing 12: tleap2.inp

# 2.2 Molecular Dynamics

Once all inputs are prepared, the next step is to run molecular dynamic (MD) simulations. The following amber inputs are used

### Minimization

```
ncyc = 5000,
                   ! 5000 steepest-descent steps, better
                   ! for strained systems
                   ! Periodic boundaries for constant volume
   ntb=1,
   iwrap=1,
                   ! Wrap coordinates into primary box
   ! Potential energy function options
   cut = 12.0,
   fswitch=10.0,
   ! Control how often information is printed to the output file
   ntpr=100,
                 ! Print energies every 100 steps
   ntwx=0,
                 ! No amber mdcrd trajectory file written
                           Listing 13: 1_Min.in
Heating
&cntrl
   imin=0,
                 ! No minimization
   irest=0,
                 ! This is NOT a restart of an old MD simulation
   ntx=1,
                 ! So our inpcrd file has no velocities
   ! Temperature control
   ntt=3,
                  ! Langevin dynamics
   gamma_ln=1.0, ! Friction coefficient (ps^-1) Langevin thermostat
                  ! collision frequency
   tempi=0.0,
                 ! Initial temp -- give it some small random
                  ! velocities
   temp0=303.15, ! Target temperature
   ! Potential energy control
   cut = 12.0,
              ! nonbonded cutoff, in angstroms
   fswitch=10.0, ! Force-based switching
   ! MD settings
   nstlim=500000, ! 1ns
   dt=0.002, ! time step (ps)
   ntb=1,
                 ! Periodic boundaries for constant volume
   ntp=0,
                 ! No pressure control
   ! SHAKE
   ntc=2.
                 ! Constrain bonds containing hydrogen
   ntf=2,
                  ! Do not calculate forces of bonds containing
                  ! hydrogen
   ! Control how often information is printed
                ! Print energies every 1000 steps
   ntpr=100,
   ntwx=100,
                 ! Print coordinates every 5000 steps to the
```

! trajectory

```
! Wrap coordinates when printing them to the same unit cell
   iwrap=1,
                ! Randomize the seed for the pseudo-random number
   ig=-1,
                 ! generator
   ! Restraint options
                ! NMR restraints and weight changes read
   nmropt=1,
! Allow the thermostat to change its target temperature throughout
the simulation.
&wt type='TEMP0', istep1=0, istep2=250000, value1=0.0, value2=300.0 /
&wt type='TEMP0', istep1=250001, istep2=500000,
value1=300.0, value2=300.0 /
&wt type='END' /
                           Listing 14: 2_Heat.in
Production
&cntrl
                ! No minimization
   imin=0,
                 ! This IS a restart of an old MD simulation
   irest=1,
```

```
! So our inpcrd file has velocities
ntx=5,
! Temperature control
ntt=3,
               ! Langevin dynamics
gamma_ln=1.0, ! Friction coefficient (ps^-1) Langevin thermostat
               ! collision frequency
temp0=303.15, ! Target temperature
ig=-1,
               ! Randomize the seed for the pseudo-random number
               ! generator
! Potential energy control
cut=12.0,
          ! nonbonded cutoff, in Angstroms
fswitch=10.0, ! Force-based switching
! MD settings
nstlim=50000000, ! 100 ns total
dt = 0.002,
               ! time step (ps)
! SHAKE
ntc=2.
              ! Constrain bonds containing hydrogen
ntf=2,
               ! Do not calculate forces of bonds containing
               ! hydrogen
! Control how often information is printed
             ! Print energies every 1000 steps
ntpr=1000,
ntwx = 25000,
             ! Print coordinates every 25000 steps to the
               ! trajectory
ntwr=10000,
             ! Print a restart file every 10K steps (can be
```

```
! less frequent)
   ntxo=2,
                  ! Write NetCDF format
   ioutfm=1,
                  ! Write NetCDF format (always do this!)
   ! Wrap coordinates when printing them to the same unit cell
   iwrap=1,
   ! Constant pressure control.
   barostat=2,
                ! MC barostat... change to 1 for Berendsen
                  ! 1=isotropic, 2=anisotropic, 3=semi-isotropic
   pres0=1.0,
                 ! Target external pressure, in bar
   ! Set water atom/residue names for SETTLE recognition
   watnam='WAT', ! Water residues are named WAT
   owtnm='0', ! Water oxygens are named 0
/
&ewald
   vdwmeth = 0,
& w t.
  type='END'
```

Listing 15: 3\_Prod.in

#### 2.3 MoBioTools

At this stage, we have obtained a classical MD trajectory from which we will extract an ensemble of geometries to perform the EDA. To do so, three single point calculations must be carried out for each of the processed geometries as suggested by the supermolecular approach to interaction energies, one corresponding to the interacting complex and the remaining ones, to the isolated fragments. In this tutorial, we will make use of the Gaussian16 program to perform such calculations.

In order to automate the generation of the Gaussian16 input, we will make use of the MoBioTools toolkit. To do so, we simply need the gaussian.main.inp, on which will specify the amber trajectory and parameters, as well as the number of geometries to be processed, and the Gaussian template, gaussian.tpl.in.

```
&main
tpl = gaussian
top = tyr_solv.prmtop
traj = tyr_solv.nc
qmmask = @1-24
geoms = 0 100 2
solvmask = :WAT
closest = 10
```

&end

&header

Listing 16: gaussian.main.in

```
%NprocShared=1
%mem=2GB
%chk=gau.chk
&end
&route
#P M062X/6-31G(d) charge nosym integral=ultrafinegrid
&end
&externchg !Include MM point charges in the gaussian input
&basis
6 - 31G(d)
&end
&bsse
mon1 = 01-24
mon2 = @25 - 12297
&end
&chgspin
0,1,0,1,0,1
&end
```

Listing 17: gaussian.tpl.inp

As one can see from the gaussian.main.inp file, we are choosing 50 geometries from the tyr\_solv.nc trajectory. For the set up of the QM/MM calculation, we have specified a QM region containing tyrosine, as well as the 10 closest water molecules to the amino acid, as specified in the gaussian.main.inp. The keyword bsse is included in gaussian.tpl.inp, to automatically generate the input corresponding to each fragment, as well as that of the complex. The counterpoise approximation is also automatically set to account for the bsse. The externchg keyword sets everything out of the QM region as external point charges.

#### 2.4 EDA

Once the generated Gaussian input files have been run, we will be able to perform the EDA. The EDA-NCI program (https://github.com/marcos-mandado/EDA-NCI) requires as input the electron density matrices corresponding to the fragments and complex that constitute the interacting systems, which are read from a formatted Gaussian checkpoint file, and a simple template as the one shown here below.

Listing 18: tmp

However, in order to properly account for the MM part of the system, one must incorporate the MM potential of the fragments and complex into the corresponding nuclear potential of the QM Hamiltonian, the charge section included in the Gaussian input files must be transferred to the corresponding Gaussian formatted checkpoint files using the charge.exe utility. To execute it, one must simply type:

```
./charge.exe comp.com mon1.com mon2.com comp.fchk mon1.fchk mon2.fchk

Now that the .fchk files have been properly modified we can perform the EDA by typing:
./EDA.exe $<$ tmp
```

As a result, an output file containing a summary on the main interaction energy components is produced:

```
_____
```

```
Interaction energy terms (in kcal/mol)
```

```
Electrostatic = -32.553
Pauli = 25.912
Polarization = -19.968
TOTAL = -26.588

Disp+Res-pol = -12.514
Induction = -7.455
```

Listing 19: EDA-NCI output summary

Assuming data normality, we can obtain the distribution of each component of the interaction energy based on the sample's mean and standard deviation:

For further details on the usage and installation of EDA-NCI and its associated utilities, please check https://github.com/marcos-mandado/EDA-NCI.

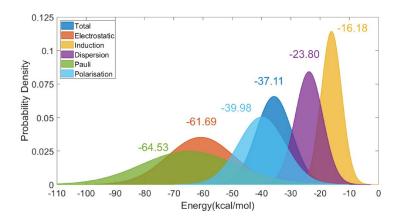


Figure 3: Normal probability distributions for each interaction energy component in kcal/mol obtained from a sample of 50 geometries from a MD trajectory. The Pauli exchange-repulsion is represented with opposite sign. Sample means are reported in kcal/mol

# 3 Absorption Spectrum: P-diaminoazobenzene Integrated Into The Human Nav 1.4 Channel

In this section, we will use MoBioTools to generate input files for calculating the absorption spectrum of a photoswitch bound to a protein ion channel. These calculations will be performed using the QM/MM multistate complete active space second-order perturbation theory (MS-CASPT2) method. To achieve this goal, we will utilize the pyoverlaps.py script. This script serves the purpose of generating a Molcas input file for a chosen molecular geometry or a specified range of geometries. Subsequently, it conducts a CASSCF (Complete Active Space Self-Consistent Field) calculation within the Molcas framework. Afterward, it compares the active space orbitals resulting from the CASSCF calculation with those defined in a user-supplied reference active space. The script then performs necessary corrections to ensure that both sets of Molecular Orbitals (MOs) are in alignment, guaranteeing consistency in the computational results.

The system being investigated involves the presence of the p-diaminoazobenzene (p-DAZ) molecule within the voltage-gated ion channel  $Na_V1.4$  located in the human brain. The starting setup for this simulation was obtained from a previous research investigation[5]. Following that, an additional 100 nanoseconds of classical molecular dynamics (MD) simulations were carried out to create a collection of molecular configurations. However, for the context of this particular tutorial, we will supply you with both the system's topology and its trajectory data, simplifying your access to the necessary input for further analysis.

#### 3.1 Initial Structures

# 3.2 The Reference Molecular Orbitals for pyoverlaps.py

As the pyoverlaps.py script necessitates a reference set of MOs, our initial focus in this section will be on constructing and executing a geometry optimization procedure. This optimized geometry will serve as the initial structure for conducting a CASSCF single-point energy calcu-

lation. You can construct the molecule using your preferred molecular visualization software, such as Avogadro or IQmol, both of which are open-source and user-friendly. For this tutorial we will only require the free version of Molcas, called OpenMolcas, which can be dowloaded directly from the terminal command:

```
git clone https://gitlab.com/Molcas/OpenMolcas.git
```

For more information, please consult this link:

```
https://gitlab.com/Molcas/OpenMolcas
```

To begin, we will initiate the optimization of the molecule using the B3LYP/cc-pVDZ level of theory. In the directory tutorialopt you will locate both the non-optimized molecule file named paz.xyz and the Molcas input file designated for geometry optimization, which is named opt.inp:

```
&GATEWAY
Title= P-diaminoazobenzene
coord = paz.xyz; basis = cc-pvdz; group = c1
>>> Do while
    &SEWARD
    &SCF ; Title="P-diamino optimization"; KSDFT=B3LYP
    &SLAPAF &END
>>> EndDo
Listing 20: opt.inp
```

The location of the pymolcas.py script may vary depending on your installation. Regardless of its location, you can execute it using the following command:

```
pymolcas.py opt.inp
```

The geometry optimization process may require some time, but it's essential to obtain a reliable starting point for our reference wave function. Once the optimization is complete, please take the optimized geometry and copy it into the /wrong\_12\_10 folder. Please exit from the /opt folder and enter in the /wrong\_12\_10 folder.

In the full active space of p-diaminoazobenzene, you should include all the  $\pi$  and  $\pi^*$  orbitals, as well as the two  $\sigma$  orbitals corresponding to the lone pairs of the central nitrogens. However, for the purposes of this tutorial, we will simplify the active space to include only  $4\pi$  orbitals,  $2\sigma$  orbitals, and  $4\pi^*$  orbitals, totaling 12 electrons in 10 orbitals (12,10). Regrettably, performing a single CASSCF single point calculation doesn't always yield the expected active space, which is also the reason for the development of the pyoverlaps.py script. Inside the folder you'll see an input file called wrong\_12\_10.inp:

```
&GATEWAY
Title= Acrolein molecule
coord = opt.xyz; basis = CC-PVDZ; group = c1
&SEWARD;
&SCF
&RASSCF
```

```
LumOrb
nActEl = 12 0 0;
Inactive = 50; Ras2 = 10;
Symmetry = 1;
Spin = 1;
CiRoot = 10 10 1
```

Listing 21: wrong\_12\_10.inp

We will now conduct a series of calculations. First, we will initiate a Hartree-Fock (HF) calculation, followed by a subsequent CASSCF single-point calculation involving 12 electrons distributed among 10 orbitals. The CASSCF calculation will read the HF orbitals for its computation. Please execute this sequence of calculations as follows:

```
pymolcas.py wrong_12_10.inp
```

Please open the \*.molden file and look at the active orbitals. In the unique scenario involving p-diaminoazobenzene, there's a noteworthy aspect to consider. If you generate your MOs using the HF method and subsequently employ them as a checkpoint for a CASSCF single-point calculation, the resulting MOs may not align with your expectations, since one of the two  $\sigma$  orbitals in not present. To address this issue, two potential strategies emerge.

Firstly, you can explore the option of directly rotating the orbitals derived from the HF calculation. This approach involves manipulating the existing HF MOs to align them more closely with the desired electronic structure. Alternatively, if orbital rotation doesn't yield the desired outcome, you may consider expanding the active space encompassing the occupied orbitals. By increasing the active space of the occupied orbitals, you provide the calculation with a broader selection of orbitals from which it can construct the desired MOs. This expanded active space may facilitate the inclusion of the target MOs, simplifying subsequent orbital manipulations.

Now, we will proceed to run a Hartree-Fock (HF) calculation, followed by a CASSCF calculation with an active space of (16,10) for a single-point energy calculation. To do this, please exit the current folder and enter the 16\_10 folder. Within this folder, you will find the optimized geometry and an input file named 16\_10.inp:

```
&GATEWAY

Title = P-diaminoazobenzene

coord = opt.xyz; basis = CC-PVDZ; group = c1

&SEWARD;

&SCF

&RASSCF

LumOrb

nActEl = 16 0 0;

Inactive = 48; Ras2 = 10;

Symmetry = 1;

Spin = 1;

CiRoot = 10 10 1
```

Listing 22: 16\_10.inp

Please execute this sequence of calculations as follows:

```
pymolcas.py 16_10.inp
```

Upon reopening the \*rasscf.molden file, you may notice the presence of one of the  $\sigma$  orbitals, which was previously absent in the HF single-point calculation. This orbital now occupies position 49. However, in the case of a (12,10) calculation, it lies outside the active space. Thus, if we attempt to employ the CASSCF (16,10) orbitals directly for the (12,10) single-point calculation, this particular orbital would be excluded. Therefore, it becomes necessary to perform an orbital rotation, and this can be achieved using one of the higher-energy orbitals. To do this, please exit the current folder and enter the 12\_10 folder, where you will find the following input, 12\_10.inp

```
&GATEWAY
Title = P-diaminoazobenzene
coord = opt.xyz; basis = CC-PVDZ; group = c1
&SEWARD;

>>>COPY $CurrDir/16_10.RasOrb $WorkDir/INPORB

&RASSCF
LumOrb
nActEl = 12 0 0;
Inactive = 50; Ras2 = 10 ;
Symmetry = 1;
Spin = 1;
CiRoot = 10 10 1
alter = 1
1 49 52
```

Listing 23: 12\_10.inp

In order for the calculation to proceed, please copy the \*.RasOrb file from the 16\_10 folder into the current 12\_10 folder, which will allow the program to read the orbitals from the (16,10) calculation. Please execute this sequence of calculations as follows:

```
pymolcas.py 12_10.inp
```

Open the \*rasscf.molden file and look at the orbitals again. If everything proceeded well, you should have  $4\pi$ ,  $2\sigma$  and  $4\pi$  orbitals which we will be using for the QM/MM calculation. Please copy the \*rasscf.molden file into the tutorial3/pyoverlaps/casscf folder.

#### 3.3 The pyoverlaps.py Program

After this extensive process, you can now utilize the 12\_10.molden file as a reference file for the pyoverlaps.py script. As previously mentioned, the desired active space should consist of 4  $\pi$  orbitals, 2  $\sigma$  orbitals, and 4  $\pi^*$  (pi star) orbitals, resulting in a total of 12 electrons distributed among 10 orbitals, denoted as (12,10). Inside the folder, you'll discover the following files: system.parm7, which contains topology data; 100f.nc, a trajectory file comprising 100 frames; system.pdb, enabling you to review ligand and residue numbering; and molcas.inp which is a MOLCAS input file:

```
& GATEWAY
Title = p-diaminoazobenzene
Coord = azo.xyz
Group = C1
Basis set = cc-pvdz spherical all
XFIeld = charges.dat
&SEWARD
&SCF
&RASSCF
Title = rasscf
Symmetry = 1
Spin = 1
Nactel = 12 0 0
Inactive = 50
Ras2 = 10
CiRoot = 10 10 1
```

Listing 24: molcas.inp

The input for the current CASSCF calculation, which corresponds to the (12,10) active space, should remain identical to the previous calculation. In addition, it includes the "XField" keyword to read the point charge data, which you can extract using the pyoverlaps.py script. Now, you can proceed to execute the pyoverlaps.py script with the following command:

```
pyoverlaps.py -p system.parm7 -c 100f.nc -a true -r 12_10.molden
-tpl molcas.inp -rng 51 60 -cl yes -qm :LIG -n 100
```

The program will repeatedly execute CASSCF calculations until the active space of the geometry in question matches the reference. This "correction" of the active space is accomplished through the utilization of the algorithm outlined in reference [6], in conjunction with the alter keyword within the &RASSCF module of Molcas. Once all the geometries have been processed, a file named "results.dat" will emerge in the directory. This file will display the exchanged MOs (MOs) for every individual geometry. If the iteration count reaches 5, it implies that the program could not successfully rotate the MOs. However, in our scenario, 72 out of 100 geometries have achieved convergence. With the attainment of the accurate wave function at CASSCF (12,10), we can now proceed to execute MS-CASPT2 calculations and acquire reasonably accurate energy values. Let's leave the /casscf directory and navigate to the /caspt2 folder. Within this directory, we will utilize the main\_qminputs.py script to generate the CASPT2 inputs necessary for the upcoming QMMM (Quantum Mechanics/Molecular Mechanics) calculation. Notice that this is the script used in the other tutorials to generate the corresponding input files. main\_qminputs.py relies on two existing files in the folder, one of which is geom.inp:

```
&main
tpl = molcas
top = system.parm7
traj = 100f.nc
qmmask = :587
geoms = 0 100 1
```

&end

### Listing 25: geom.inp

It reads in the software for interfacing, topology data, trajectory information, and provides options to specify the QM region and the number of geometries to generate. Additionally, you will find the MOLCAS-specific file named molcas.tpl in the same folder:

```
&header
title = p-AZO molecule
group = C1
basis set = cc-pVDZ spherical all
RICD
&end
&seward
&end
&chgspin
0,1,0,1,0,1
&end
&externchg
&end
&guess
jobiph
&end
&caspt2
IPEA = 0.0
SHIFT= 0.0
IMAGinary= 0.2
maxiter=100
multistate = 10 1 2 3 4 5 6 7 8 9 10
PROP
&end
&rassi
Nr of JobIph = 1 10; 1 2 3 4 5 6 7 8 9 10
HEFF
&end
```

Listing 26: molcas.tpl

This script is designed to supply all the essential keywords required to execute a successful CASPT2 single-point calculation. Please execute it as:

```
main_qminputs.py -i geom.inp -t molcas.tpl
```

which will generate 100 geometries with the .xyz file of the QM region, the MM region and the MOLCAS input file. However, there is a critical component that is currently missing: the

CASSCF reference wave function. To rectify this, we can implement a loop that copies the \*.JobIph file from each geometry in the /casscf folder to the corresponding geometry in the /caspt2 folder. This step is essential to ensure that the CASPT2 calculations have access to the necessary CASSCF wave function information.

```
for i in {1..100}
do
cp casscf/geom$i/*.JobIph caspt2/geom$i/geom$i.in.guess.JobIph
done
```

Please run the calculation with pymolcas.py in the corresponding folder of each geometry. If everything proceeded well, you should now have a MS-CASPT2 single point for each geometry in which the orbital rotation of the pyoverlaps.py was successful. This is the end of the tutorial.

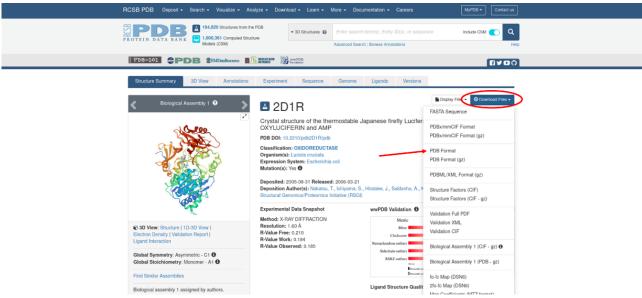
# 4 ABSORPTION SPECTRUM: LUCIFERINE/LUCIFERASE COMPLEX

#### 4.1 Setup

#### 4.1.1 Structure of the Protein

In this tutorial, we will simulate the absorption spectrum of the Luciferine/Luciferase complex, whose crystallographic structure (PDB code: 2D1R) can be retrieved from the e Protein Data Bank website (https://www.rcsb.org/, Figure 4). We will use PyMOL to pre-process it prior to generate the actual input files with MoBioTools. The PDB structure can either be retrieved from the previously mentioned website, or directly downloaded using the PyMOL graphical user interface by means of the fetch command (Figure 5).





(b) PDB ID: 2D1R section of the PDB.

Figure 4: Screenshots of the PDB.

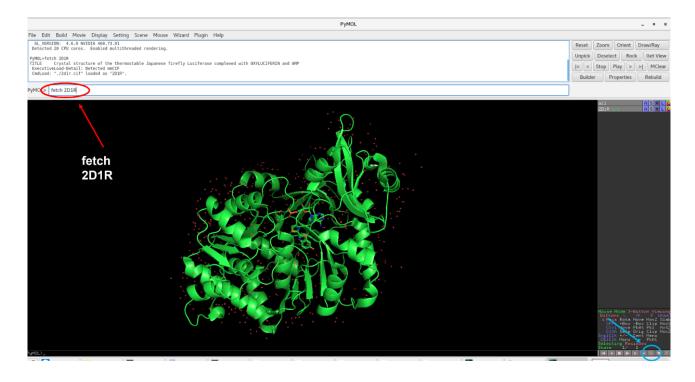


Figure 5: Preview of the 2D1R PDB structure with PyMOL.

As can be observed from Figure 5, the obtained structure contains some water molecules that need to be removed since a latter step involves the solvation of the whole system. For this, we need to display the sequence of amino acids and residues by clicking on the S letter circled in blue at the bottom-right of Figure 5. Then, we look for the sequence of oxygen atoms, select it and remove the atoms of the selection. Note that there are no hydrogen atoms in the structure, since the experimental technique does not allow their detection. To select the atoms, click on the first 0, as shown in Figure 6; slide the underlying bar until the last atom appears, press SHIFT on the keyboard and select everything until the end by clicking on the last atom. Then, these atoms can be easily removed by clicking in the A (action) of the sele section on the left and clicking on remove atoms, as shown in Figure 6.

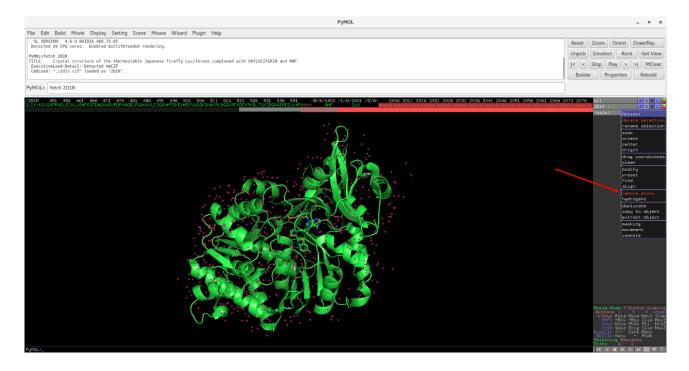


Figure 6: Selecting and removing residues of a PDB structure with PyMOL.

As a result, the structure on Figure 7 is obtained. Nevertheless, this structure is still missing the hydrogen atoms. The H atoms corresponding to the protein will be automatically added later on in the process, whereas those of the oxyluciferine (OLU) and the adenosine monophosphate (AMP) need to be added manually. To do so, please select the OLU and AMP residues in the upper sequence section, click on the A of the sele section on the left and then click on hydrogen atoms > add, as shown in Figure 7. This would generate the monoanion of the AMP and the phenol-enolate tautomer of the oxyluciferine chromophore.



Figure 7: Protonation of a structure with PyMOL.

To properly save it, the whole structure needs to be copied to a new object. To do so, select the residues of the protein by choosing the **chain** selection at the bottom-right of the window, as shown in Figure 8.

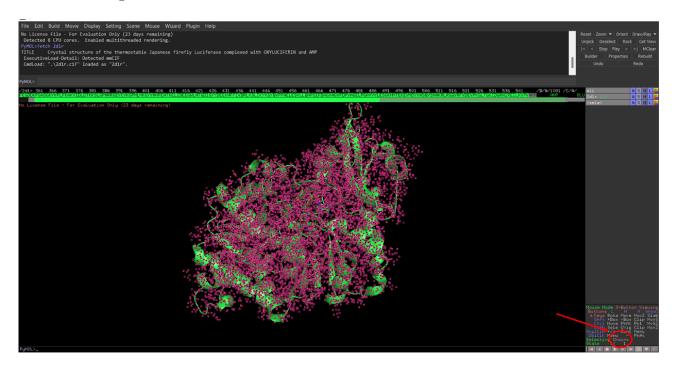


Figure 8: Enabling chain the selection option in PyMol.

Subsequently, click on the A letter of the selection > copy to object > new (see Figure 9).

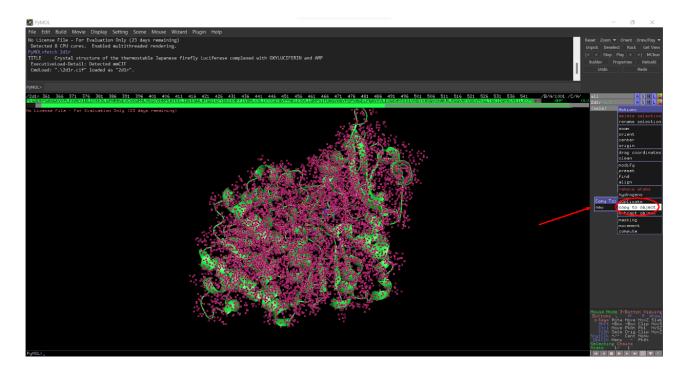


Figure 9: Adding a selection to a new object in PyMol.

By selecting the AMP residue and following the same procedure, we can add the selection to the previously generated object if, instead of selecting a new object, we select obj01. Re-

peating the same procedure for the OLU, then the new object will encompass the whole system.

Finally, save the PDB file (system2.pdb) by click on File > Export Molecule on the top-left and then, mark the option original atom order, as shown in Figure 10.

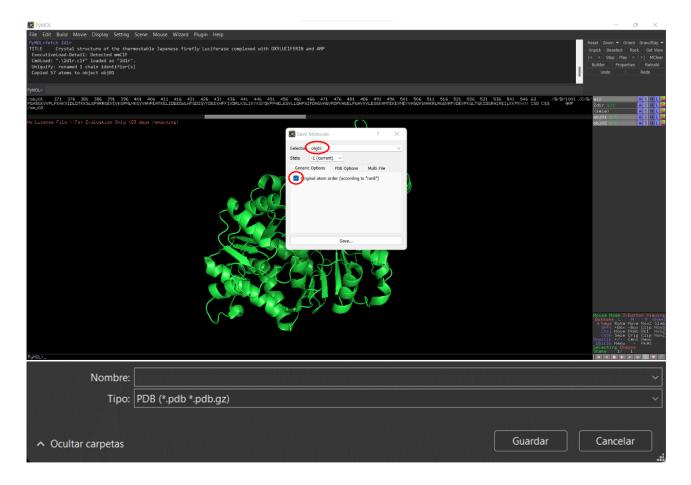


Figure 10: Saving options.

Additionally, we will copy the AMP and OLU residues to objects by individually selecting them and clicking again in A of the sele section > copy to object > new, and save both obj02 and obj03 as two PDB files since we will need them later on.

Some extra changes need to be done in the new PDB file so as to have it in a format compatible with antechamber, which will be used in the following section. Open the file with any text editor, for example vi:

#### vi system.pdb

Inside the file, look for the residues CSO 63 and CSO 64. For each of these residues, perform the following changes:

- CSO by CYS
- OD by HG

• HETATM by ATOM.

Moreover, for both the OLU and AMP residues we have to change:

- HETATM by ATOM.
- Atom type + atom number in the atom name column, the third one as shown in Figure 11.

```
90.00
                                               90.00
          57.612
CRYST1
                   181.402
                              52.743
                                                       90.00 P 21 21 21
                                                               1.00 32.06
MOTA
              01
                   OLU A2001
                                    16.670
                                             26.100
                                                      10.887
                                                                                  C
                                                                                       0
           1
MOTA
           2
              C2
                   OLU A2001
                                    15.028
                                             24.679
                                                       9.909
                                                               1.00
                                                                     30.05
                                                                                  C
                                                                                       C
                                                                     31.21
MOTA
           3
              C3
                   OLU A2001
                                    16.429
                                             25.049
                                                      10.303
                                                               1.00
                                                                                  C
                                                                                       C
MOTA
           4
              Ν4
                   OLU A2001
                                    17.314
                                             24.123
                                                       9.956
                                                               1.00
                                                                     30.21
                                                                                  C
                                                                                       N
                                                               1.00
MOTA
           5
              C5
                   OLU A2001
                                    16.764
                                             23.072
                                                       9.331
                                                                     30.07
                                                                                  C
                                                                                       C
                                                                                       S
S
MOTA
           6
              56
                   OLU A2001
                                    15.145
                                             23.066
                                                       9.100
                                                               1.00
                                                                     32.70
                                                                                  C
           7
              S7
                                    19.260
                                                       9.098
                                                                                  C
MOTA
                   OLU A2001
                                             22.005
                                                               1.00
                                                                     26.21
                                                                                       C
MOTA
           8
              C8
                   OLU A2001
                                             21.905
                                                       8.870
                                                                                  C
                                    17.572
                                                               1.00 28.51
MOTA
           9
              Ν9
                   OLU A2001
                                             20.761
                                                       8.333
                                                                                  C
                                                                                       Ν
                                    17.122
                                                               1.00 26.03
MOTA
          10
              C10 OLU A2001
                                    18.079
                                             19.867
                                                       8.068
                                                                                  C
                                                                                       C
                                                               1.00
                                                                     26.39
ATOM
          11
              C11 OLU A2001
                                    19.363
                                             20.481
                                                       8.469
                                                               1.00
                                                                     26.70
                                                                                  C
                                                                                       C
ATOM
          12
              C12 OLU A2001
                                    20.526
                                             19.726
                                                       8.302
                                                               1.00
                                                                     27.50
                                                                                  C
                                                                                       C
ATOM
          13
              C13 OLU A2001
                                    20.463
                                             18.435
                                                       7.744
                                                               1.00 27.89
                                                                                  C
                                                                                       C
ATOM
          14
              014 OLU A2001
                                    21.598
                                             17.690
                                                       7.564
                                                               1.00 29.92
                                                                                  C
                                                                                       0
ATOM
          15
              C15 OLU A2001
                                    19.241
                                             17.879
                                                       7.363
                                                               1.00 26.46
                                                                                  C
                                                                                       C
ATOM
          16
              C16 OLU A2001
                                    18.054
                                             18.591
                                                       7.519
                                                               1.00 23.27
                                                                                  C
                                                                                       C
ATOM
          17
              H17 OLU A2001
                                    14.124
                                             25.264
                                                      10.077
                                                               1.00 30.05
                                                                                  C
                                                                                       Н
MOTA
          18
              H18 OLU A2001
                                    21.488
                                             20.139
                                                       8.605
                                                               1.00 27.50
                                                                                  C
                                                                                       Н
MOTA
          19
              H19 OLU A2001
                                    21.852
                                             17.711
                                                       6.638
                                                               1.00 29.92
                                                                                  C
                                                                                       Н
MOTA
          20
              H20 OLU A2001
                                    19.215
                                             16.876
                                                       6.938
                                                               1.00 26.46
                                                                                  C
                                                                                       Н
MOTA
          21
              H21 OLU A2001
                                    17.107
                                             18.148
                                                       7.211
                                                               1.00 23.27
                                                                                  C
                                                                                       Н
CONECT
           1
                 3
           2
               17
                      3
CONECT
                            6
```

Figure 11: Changed atom name column.

These changes need to be done not only in the PDB file containing the whole system but also in the the OLU and AMP separated PDB files.

# 4.1.2 Solvation and Parameters

The Amber software already has predefined force field parameters (equilibrium values and force constants) to deal with proteins and water. However, the luciferin/luciferase complex contains two non-standard residues: AMP and OLU, for which it is necessary to determine extra parameters. The procedure is analogous for both of them, so we will describe in detail the case of OLU. At first, we transform the OLU.pdb file into a .mol2 file:

```
\label{lem:continuous} $$ \text{cantechamber -i OLU.pdb -fi pdb -o OLU.mol2 -fo mol2 -c bcc -at gaff -nc -1} \
```

where the options -i, -fi, -o, -fo, -c, and -nc correspond to the input name, input format, output name, output format, method to determine the charges, and the total charge of the system, respectively.

Then, we have to check whether there are parameters for the molecule in the selected force field and apply the proper corrections in the case that there aren't any:

```
parmchk2 -i OLU.mol2 -o OLU.frcmod -f mol2
```

where the options -i, -o, and -f indicate the input file, the output file and the input format, respectively. As stated above, this process needs to be repeated in the case of the AMP molecule:

```
antechamber -i AMP.pdb -fi pdb -o AMP.mol2 -fo mol2 -c bcc -at gaff -nc -1 parmchk2 -i AMP.mol2 -o AMP.frcmod -f mol2
```

Afterwards, we have to generate the library files for the ligands by means of tleap:

```
tleap -f lib.in
```

where the input file lib.in content is explained below:

```
source leaprc.gaff2  # source leaprc file for gaff

OLU=loadmol2 OLU.mol2  # load mol2 file of the oxyluciferine saveoff OLU OLU.lib  # create library of the oxyluciferine

AMP=loadmol2 AMP.mol2  # load mol2 file of the AMP saveoff AMP AMP.lib  # create library of the AMP
```

Listing 27: lib.in

Finally, we will generate the coordinates and the parameter files to be used during the MD simulations using again the tleap program with the solvation.in input file:

```
tleap -f solvation.in
source leaprc.gaff2
                                    # source leaprc file for gaff
source leaprc.protein.ff19SB
                                   # source leaprc file for ff19SB
                                      protein force field
source leaprc.water.tip3p
                                   # source leaprc file for TIP3P
                                      water model
loadamberparams OLU.frcmod
                                   # load OLU parameters
                                   # load AMP parameters
loadamberparams AMP.frcmod
loadoff OLU.lib
                                   # load library for OLU
loadoff AMP.lib
                                   # load library for AMP
SYS=loadpdb system2.pdb
                                   # load whole system
solvateOct SYS TIP3PBOX 5.0
                                   # solvate system with H2O with
                                      truncated octaedron box of 5 A
                                    # add ions - cell net charge
addIons2 SYS Na+ 0
                                      must be 0
```

Listing 28: solvation.in

If one opens the final PDB file, it is possible to observe the solvated system and the truncated octahedron shape of the cell.

# 4.2 Minimization, Equilibration and Production

Once the .prmtop (force field parameters) and the .rst7 (coordinate) files are prepared, the system is ready for us to perform the MD simulations. The inputs for the minimization, heating and equilibration/production simulations are shown below.

```
Minimization input file in explicit solvent
&cntrl
    ! Minimization options
   imin=1,
                  ! Turn on minimization
   ntx=1,
                   ! Read position but not velocities
                   !(5 for reading also velocities)
   irest=0,
                  ! Do not restart (=1) the simulation
   maxcyc=10000, ! Maximum number of minimization cycles
   ncyc=5000,
                   ! Method switched from steepest descent to
                    !conjugate gradient after NCYC
                   ! PBC cte V
   ntb=1,
    iwrap=1,
    ! Potential energy function options
               ! Nonbonded cutoff in Angstroms
    cut=9.0,
    fswitch=7.0, ! Van der Waal cutoff
    !Restraint options
   nmropt=1,
                   !Restraints
    ! Control how often information is printed to the output file
                  ! Print energies every 100 steps
   ntpr=100,
   ntwx=0,
                  ! No trajectory file
&wt type = 'END' /
                            Listing 29: 01_Min_luc.in
   Heat
 &cntrl
                 ! Turn off minimization
  imin=0,
  ntx=1,
                  ! Read position but not velocities (5 for reading
                  !also velocities)
```

```
! Do not restart (=1) the simulation
       irest=0,
       nstlim=50000, ! Perform MD for 50000 steps
       dt = 0.002,
                                                                 ! 2 fs time step
       ntf=2, ntc=2, ! Shake hydrogen atoms
                                                              ! Initial temperature
       tempi=0.0,
       temp0=300.0,
                                                                 ! Target temperature
       ! Control how often information is printed to the output file
                                                                   ! Energy information will be printed every ntpr steps
       ntpr=100,
       ntwx=100,
                                                                       ! Coordinates written to the mdcrd file
                                                                       !every ntwx steps
       ! Potential energy function options
       cut=9.0,
                                              ! Nonbonded cutoff in Angstroms
       fswitch=7.0,
                                                                 ! Van der Waal cutoff
       ! Ensemble control
       ntb=1,
                                                                  ! PBC cte V
                                                                 ! No pressure scaling (Default)
       ntp=0,
                                                                  ! Langevin dynamics
       ntt=3,
       gamma_ln=2.0, ! Collision frequency for the Langevin
                                                                      !dynamics in ps-1
       ! Wrap coordinates when printing them to the same unit cell
       nmropt=1,
       iwrap=1,
                                                                  ! Random seed to use to generate velocity distribution
       ig=-1,
! Change conditions during simulation
! Simulation increases T from 0 to 300K during steps 0 to 25000
&wt type='TEMP0', istep1=0, istep2=25000, value1=0.0, value2=300.0 /
! Keep T cte for the rest of the simulation % \left( 1\right) =\left( 1\right) +\left( 1\right)
&wt type='TEMP0', istep1=250001,
istep2=50000, value1=300.0,
value2=300.0 /
&wt type='END' /
                                                                                                             Listing 30: 02_Heat_luc.in
               A NPT simulation for common production-level simulations
   &cntrl
               imin=0,
                                                                                  ! No minimization
                                                                                ! This IS a restart of an old MD simulation
               irest=1,
                                                                                  ! So our inpcrd file has velocities
               ntx=5,
               ! Temperature control
                                                                                 ! Langevin dynamics
               ntt=3,
               gamma_ln=1.0,
                                                                               ! Friction coefficient (ps^-1)
```

```
temp0=303.15,    ! Target temperature
   ig=-1,
                   ! Seed for velocities
   ! Potential energy control
  cut=9.0,
               ! nonbonded cutoff, in Angstroms
   fswitch=7.0, ! Force-based switching
  ! MD settings
  nstlim=10000000, ! 20ns total
  dt = 0.002,
                   ! time step (ps)
  ! SHAKE
  ntc=2,
                    ! Constrain bonds containing hydrogen
  ntf=2,
                    ! Do not calculate forces of bonds containing
                    ! hydrogen
  ! Control how often information is printed
                   ! Print energies every 1000 steps
  ntpr=1000,
  ntwx = 25000,
                   ! Print coordinates every 25000 steps to the
                   !trajectory
  ntwr=10000,
                   ! Print a restart file every 10K steps (can be
                    !less frequent)
                   ! Write NetCDF format
  ntxo=2,
                   ! Write NetCDF format (always do this!)
  ioutfm=1,
   ! Wrap coordinates when printing them to the same unit cell
  iwrap=1,
  nmropt=1,
  ! Constant pressure control.
  barostat=1,
                ! Berendsen barostat
                   ! 1=isotropic, 2=anisotropic, 3=semi-isotropic
  ntp=1,
                    !w/ surften
                    ! Target external
  pres0=1.0,
                    !pressure, in bar
   ! Set water atom/residue names for
  !SETTLE recognition
  watnam='WAT', ! Water residues
                   !are named WAT
  owtnm='0',
                   ! Water oxygens
                   !are named O
&ewald
  vdwmeth = 0,
&wt
```

/

```
type='END'
/
```

Listing 31: 03\_Prod\_luc.in

#### 4.3 MoBioTools

Once the molecular dynamics simulation is finished, we will have to extract some geometries out of the trajectory to compute the absorption spectrum of the luciferine/luciferase complex. To do so, we will use the MoBioTools software that automatically takes the geometries and prepares the inputs. For this, we will need two files: the gaussian\_luc.main.in, that specifies the parameter and trajectory files to be read and the quantum-mechanical region, among other things; and the Gaussian template file, the gaussian\_luc.tpl.inp. In general, the script that prepares the quantum-mechanical input files can be run in the command line as:

```
main_qminputs.py -i gaussian_luc.main.in -t gaussian_luc.tpl.inp
&main

tpl = gaussian

top = final_system.prmtop

traj = 03_prod.nc

qmmask = :OLU

geoms = variabless
solvmask =:WAT
&end
```

Listing 32: gaussian\_luc.main.in

```
&header
%NprocShared=4
%mem=16GB
&end

&route
#p B3LYP/6-31G NoSym TD(singlets,nstates=10)
&end

&externchg
&end

&chgspin
-1,1
&end
```

Listing 33: gaussian\_luc.tpl.inp

However, for the generation of many geometries one can use the script shown below:

```
#!/bin/bash
```

```
for number in ` seq 199 2 399` ; do
    sed -i "s/variabless/$number/g" gaussian.main.in
    main_qminputs.py -i gaussian.main.in -t gaussian.tpl.inp
    sed -i "s/$number/variabless/g" gaussian.main.in
done
```

Listing 34: many\_geoms\_luc.sh

#### 4.4 Absorption Spectrum

Once the quantum-mechanical calculations are finished, we will retrieve the energies involved in the calculated electronic transitions. A simple script for this can be found below

```
#!/bin/bash
apwd='pwd'
                     Eexc (eV) (nm) f' > ${apwd}/SUMMARY_STEPS.dat
echo '# geom
               state
for i in `seq 199 2 399`
do
for j in `seq 9`
   exc=`grep "Excited State $j" geom${i}/geom${i}.log | awk '{print $5}'`
   f=`grep "Excited State $j" geom${i}/geom${i}.log | awk '{print $9}'`
                             $j" geom${i}/geom${i}.log | awk '{print $7}'`
   nm=`grep "Excited State
                                           $nm" >> ${apwd}/SUMMARY_STEPS.dat
   echo "
            ${i}
                   $ј
                         $exc
                                ${f#*=}
   done
   exc=`grep "Excited State 10" geom${i}/geom${i}.log | awk '{print $5}'`
   f=\ensuremath{`grep"Excited State 10" geom${i}/geom${i}.log | awk '{print $9}'`}
   nm=`grep "Excited State 10" geom${i}/geom${i}.log | awk '{print $7}'`
   echo "
            ${i}
                   10
                                ${f#*=}
                                           $nm" >> ${apwd}/SUMMARY_STEPS.dat
                         $exc
done
```

Listing 35: get\_energies.sh

With these energies and their corresponding intensities, one can place a Gaussian function on top of each of them in order to obtain the absorption spectrum using, for example the python script shown below:

```
x = np.linspace(0.5, 6.0, 300)
archivo="SUMMARY_STEPS.dat"
out="spectrum_"+archivo
out2="spectrum_nm_"+archivo
data= np.loadtxt(archivo)
ro=data[:,2]
fo=data[:,3]
gaus = np.zeros(len(x))
for i in range(0,len(ro)):
    gaus += gaussian(x, 0.3, ro[i], fo[i])
with open(out, 'w') as f:
    f.write('#E (eV)
                        Intensity')
    f.write('\n')
    for i in range(len(x)):
        f.write(str(x[i]))
        f.write('
                  ')
        f.write(str(gaus[i]))
        f.write('\n')
with open(out2, 'w') as f:
    f.write('#E (nm)
                        Intensity')
    f.write('\n')
    for i in range(len(x)):
        f.write(str((10000000*27.211/219474)/x[i]))
        f.write(' ')
        f.write(str(gaus[i]))
        f.write('\n')
```

Listing 36: spectrum.py

Finally, the generated file can be represented con any plotting utility. The script plot\_absorption.gpl shows a gnuplot example on how to do so.

```
set term png
set output 'plot.spectrum_SUMMARY_STEPS.png'

set key font ",20" spacing "1.5"

set xtics font ",15"
set ytics font ",15"

unset key

set xlabel "Absorption Energy (eV)" font ",20"
set ylabel "Intensity" font ",20" offset 2,0,0

p 'spectrum_SUMMARY_STEPS.dat' u 1:2 w 1
1c rgb "#D52D00" lw 3
```

Listing 37: plot.absorption.gpl

The script can be executed as follows:

gnuplot plot.absorption.gpl

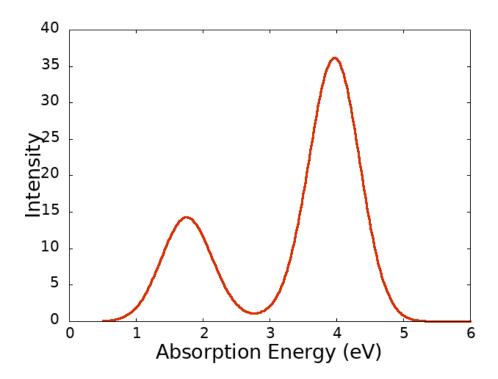


Figure 12: Absorption spectrum of the luciferine/luciferase complex

And with this, we come to the end of this tutorial.

#### References

- [1] N. Ramos-Berdullas, I. Pérez-Juste, C. Van Alsenoy and M. Mandado. Theoretical study of the adsorption of aromatic units on carbon allotropes including explicit (empirical) DFT dispersion corrections and implicitly dispersion-corrected functionals: the pyridine case. *Phys. Chem. Chem. Phys.*, 17(1):575–587, 2015.
- [2] M. Mandado and J. M. Hermida-Ramón. Electron Density Based Partitioning Scheme of Interaction Energies. J. Chem. Theory Comput., 7(3):633–641, 2011.
- [3] G. Cárdenas, A. Pérez-Barcia, M. Mandado, and J. J. Nogueira. Characterization of cisplatin/membrane interactions by QM/MM energy decomposition analysis. *Phys. Chem. Chem. Phys.*, 23:20533, 2021.
- [4] M. Ropo, M. Schneider, C. Baldauf, and V. Blum. First-principles data set of 45,892 isolated and cation-coordinated conformers of 20 proteinogenic amino acids. *Sci. Data*, 3, 2016.
- [5] Vito F Palmisano, Carlos Gómez-Rodellar, Hannah Pollak, Gustavo Cárdenas, Ben Corry, Shirin Faraji, and Juan J Nogueira. Binding of azobenzene and p-diaminoazobenzene to the human voltage-gated sodium channel na v 1.4. *Physical Chemistry Chemical Physics*, 23(5):3552–3564, 2021.

[6]	Gustavo Cárdenas and Juan J Nogueira. An algorithm to correct for the cassef active space in multiscale qm/mm calculations based on geometry ensembles. International Journal of Quantum Chemistry, $121(6)$ :e $26533$ , $2021$ .