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Start studying 11.1 Describing Chemical Reactions. Learn vocabulary, terms, and more with flashcards, games, and other study tools. ... if a Δ is used to increase the rate of a chemical reaction, its formula is written above the arrow ... symbolic way of describing a chemical reaction.

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Use (s) for a solid, (l) for a liquid, (g) for a gas, and (aq) for a substance in aqueous solution (a substance dissolved in water). (s) In many chemical reactions, a catalyst is added to the reaction mixture. A catalyst is a substance that speeds up the reaction but is not used up in the reaction.

11.1 Describing Chemical Reactions 11

The production of a new substance, a gas, is evidence of a chemical change. Two new substances are produced in this reaction, oxygen gas and liquid water. You could describe this reaction by saying, "Hydrogen peroxide decomposes to form water and oxygen gas." You could also write a word equation.

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Chemical equation (11.1) A representation of a chemical reaction; the formulas of the reactants (on the left) are connected by an arrow with the formulas of the products on the right i.e.

Chapter 11: chemical reactions Flashcards | Quizlet

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Chapter 11 Chemical Reactions 113 SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329)

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The heat and smoke are the products of a combustion reaction. Combustion is one of the five general types of chemical reactions. In this chapter, you will learn that if you can recognize a reaction as being a particular type, you may be able to predict the products of the reaction.

11.2 Types of Chemical Reactions - Henry County School ...

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