

Physics 1
(15)

2019

Prep Engineering
Level (1)

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مركز الإلهام والتطوير
مركز النسخ والتصوير

CH-5: Temperature

[A] Definitions (تعريفات عامة)

[1] Temperature:

→ Is the degree of hotness or coldness of a body and is related to the kinetic energy of molecules of substance.
 * درجة الحرارة: هي درجة سخونة أو برودة الجسم وترتبط بمطاقة حركة الجزيئات في البيئة.

[2] Thermal equilibrium: الإلتزان الحراري

→ A thermal equilibrium exists between two bodies when they are in thermal contact with each other and there is no net flow of heat between them.
 * يحدث الإلتزان الحراري بين جسمين إذا كانوا متصلين مع بعضهما وكانت محصلة انتقال الطاقة الحرارية بينهما تساوي صفر.

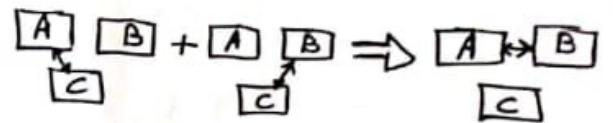
T_A	T_B
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if $T_A = T_B \Rightarrow \text{Heat} = \text{Zero}$

[3] Zeroth Law of thermodynamic:

القانون الصفري لديناميكا الحرارية

→ If two bodies A and B are in thermal equilibrium with a third body C, then A and B are in thermal equilibrium with each other.



← إذا كان جسمان A و B في اتزان حراري مع جسم ثالث C فإن A و B في اتزان حراري مع بعضهما.

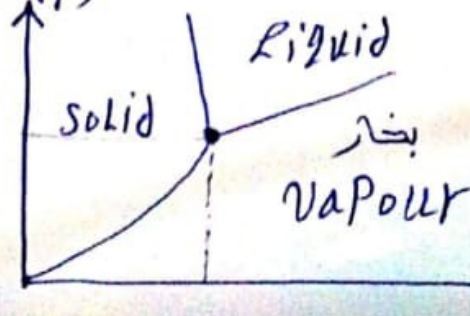
[4] Triple Point:

النقطة الثلاثية

→ Is the Point where the three Phases of the material (solid - liquid - gas) are exist in equilibrium.

← هي النقطة التي توجد عندها المادة في حالاتها الثلاث (صلب - سائل - غاز) في حالة اتزان.

(V) or (P)



$$T = 273.16 \text{ K}$$

Water

[B] Temperature scales: مقاييس الحرارة

مقاييس

(1) Kelvin scale: (thermodynamic scale)

$$T_K = T_C + 273 \Rightarrow dT_K = dT_C$$

مقاييس

(2) Fahrenheit scale: (فهرنهايت)

$$T_F = \frac{9}{5} T_C + 32 \Rightarrow dT_F = \frac{9}{5} dT_C$$

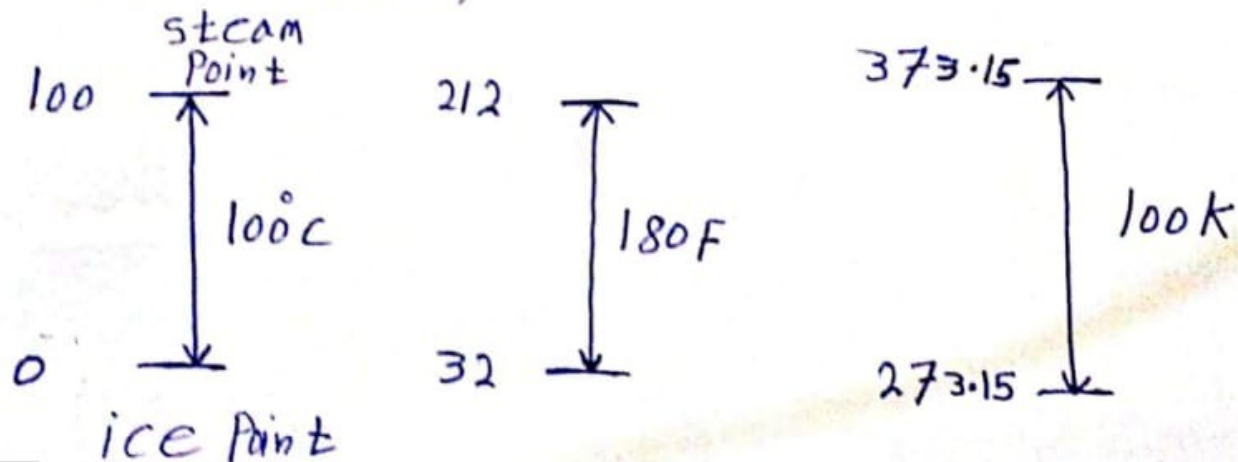
مقاييس

(3) Rankine scale:

(رانكن)

$$T_R = \frac{9}{5} T_K \Rightarrow dT_R = \frac{9}{5} dT_K$$

(Upper Fixed Point)



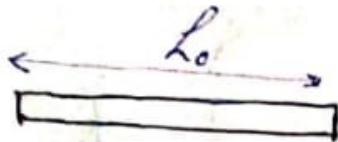
ice Point

(Lower Fixed Point)

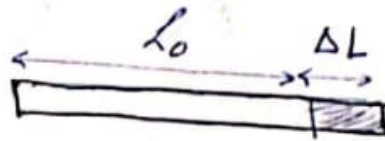
[C] Thermal expansion of solids:

التمدد الحراري للمواد الصلبة

(a) Linear expansion: التمدد الطولي



$$T = T_0$$



$$T = T$$

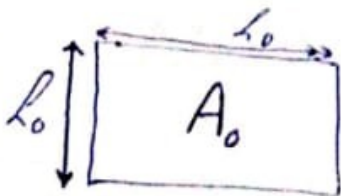
حفظ

$$\Delta L = \alpha L_0 \Delta T = \alpha L_0 (T - T_0)$$

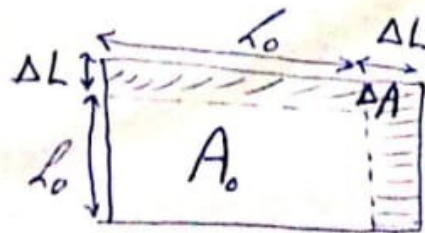
* Linear expansion coefficient (α):

معامل التمدد الطولي وهو معامل الزيادة في
الطول مع زيادة درجة الحرارة (°C) أو K

(b) surface expansion: التمدد السطحي



$$T = T_0$$



$$T = T$$

Chapter 5: Temperature

scale D. Water at 90°C is warmer than water at 202°F E. 0°F corresponds to -32°C

ans: C

10. Room temperature is about 20 degrees on the:

- A. Kelvin scale (B) Celsius scale C. Fahrenheit scale D. absolute scale
E. C major scale

ans: B

11. A thermometer indicates 98.6°C . It may be:

- A. outdoors on a cold day B. in a comfortable room (C) in a cup of hot tea
D. in a normal person's mouth E. in liquid air

ans: C

12. The two metallic strips that constitute some thermostats must differ in:

- A. length B. thickness C. mass D. rate at which they conduct heat
(E) coefficient of linear expansion

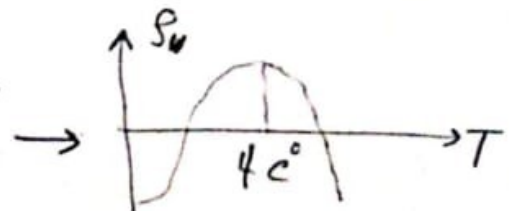
ans: E

13. It is more difficult to measure the coefficient of volume expansion of a liquid than that of a solid because:

- A. no relation exists between linear and volume expansion coefficients B. a liquid tends to evaporate
C. a liquid expands too much when heated
D. a liquid expands too little when heated (E) the containing vessel also expands

ans: E

Unusual behavior
of water



Note:

- (1) maximum density of water at 4°C
(2) Holes expands like solids.

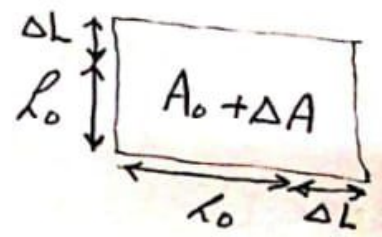
$$\Delta A = \gamma A_0 \Delta T = \gamma A_0 (T - T_0) \quad \text{حفظ}$$

تعريف * surface expansion coefficient (γ):
 معامل التمدد السطحي وهو معدل الزيادة في المساحة مع
 زيادة درجة الحرارة (K^{-1} or $^{\circ}C^{-1}$)

اثبات 1

* Prove that $\gamma = 2\alpha$

Proof:



$$A_0 + \Delta A = (l_0 + \Delta L)^2$$

$$A_0 + \Delta A = l_0^2 + 2l_0\Delta L + \Delta L^2$$

$$A_0 + \Delta A = A_0 + 2l_0(\alpha l_0 \Delta T) \quad \text{very small } \Delta L^2$$

$$\therefore \Delta A = 2\alpha l_0^2 \Delta T$$

$$\therefore \Delta A = 2\alpha A_0 \Delta T$$

$$\therefore \Delta A = \gamma A_0 \Delta T$$

$$\therefore \boxed{\gamma = 2\alpha}$$

(C) Volume expansion:
 التمدد الحجمي

$$\Delta V = \beta V_0 \Delta T = \beta V_0 (T - T_0) \quad \text{حفظ}$$

تعريف * Volume expansion coefficient (β):
 معامل التمدد الحجمي وهو معدل الزيادة في الحجم مع زيادة درجة الحرارة
 (K^{-1} or $^{\circ}C^{-1}$)

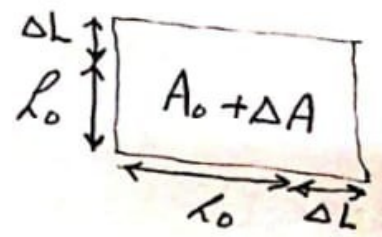
$$\Delta A = \gamma A_0 \Delta T = \gamma A_0 (T - T_0) \quad \text{حفظ}$$

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(C) Volume expansion:
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$$\Delta V = \beta V_0 \Delta T = \beta V_0 (T - T_0) \quad \text{حفظ}$$

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 (K^{-1} or $^{\circ}C^{-1}$)

اثبات 2

* Prove that $\beta = 3\alpha$ ~ proof ~

$$\Delta V + V_0 = (L_0 + \Delta L)^3$$

$$\therefore \Delta V + V_0 = L_0^3 + 3L_0^2 \Delta L + 3L_0 \Delta L^2 + \Delta L^3$$

$$\therefore \Delta V - \cancel{V_0} = \cancel{V_0} + 3L_0^2 \Delta L$$

very small
can be
neglected

$$\therefore \Delta V = 3L_0^2 \Delta L$$

$$\therefore \Delta L = \alpha L_0 \Delta T$$

$$\therefore \Delta V = 3L_0^2 (\alpha L_0 \Delta T)$$

$$= 3\alpha L_0^3 \Delta T$$

$$\therefore \Delta V = 3\alpha V_0 \Delta T$$

$$\therefore \Delta V = \beta V_0 \Delta T$$

$$\boxed{\beta = 3\alpha}$$

Temperature -- Problems

[1] An object has temperature of 45°C , What is the temperature in Fahrenheit?

~ answer ~

$$\therefore T_F = \frac{9}{5} T_C + 32$$

$$\therefore T_F = \frac{9}{5} (45) + 32 = \boxed{113^{\circ}\text{F}}$$

[2] The temperature on a warm summer day is 95°F , What is

(a) In degree Celsius

(b) In Kelvin

~ answer ~

$$(a) \therefore T_F = \frac{9}{5} T_C + 32$$

$$\therefore 95 = \frac{9}{5} T_C + 32$$

$$\boxed{T_C = 35^{\circ}\text{C}}$$

$$(b) \therefore T_K = T_c + 273.15$$

$$\therefore T_K = 35 + 273.15 = \boxed{308.15K}$$

[3] Find the temperature at which the Celsius scale equals the Fahrenheit scale.

~ answer ~

$$T = \frac{9}{5} T + 32$$

$$\therefore -\frac{4}{5} T = 32 \Rightarrow \boxed{T = -40}$$

[4] Modern eaves are constructed from a roll sheet of aluminium, what is the change in the length of 30m long AL trough.

$$\alpha = 23 \times 10^{-6} \text{ } ^\circ\text{C}^{-1}, \Delta T = 100^\circ\text{F}$$

~ answer ~

$$\Delta L = \alpha L_0 \Delta T_c$$

شريط صلب

6 steel tape 5m long is calibrated at a temperature of 20°C. what is its length on a hot summer day of 35°C.

~ answer ~

$$\Delta L = \alpha L_0 \Delta T \quad \alpha_{\text{steel}} = 1.2 \times 10^{-5}$$

$$\Delta L = (1.2 \times 10^{-5}) (5) (35 - 20)$$

$$= 9 \times 10^{-4} \text{ m}$$

$$L = L_0 + \Delta L = 5 + 9 \times 10^{-4} = \boxed{5.0009 \text{ m}}$$

وعاء

ملئ تماماً

7 glass flask of volume 200 cm³ is just filled with mercury at 20°C. How much mercury overflows when the temperature of the system is raised to 100°C.

$$\beta_{\text{glass}} = 1.2 \times 10^{-5} \text{ } ^\circ\text{C}^{-1}, \quad \beta_{\text{Hg}} = 0.000182$$

~ answer ~

$$\Delta V_{\text{over}} = \Delta V_{\text{Hg}} - \Delta V_{\text{glass}}$$

$$= \beta_{\text{Hg}} V \Delta T - \beta_{\text{glass}} V \Delta T$$

$$= (\beta_{\text{Hg}} - \beta_{\text{glass}}) V \Delta T$$

$$= [0.000182 - 1.2 \times 10^{-5}] * 200 * (100 - 20)$$

$$= \boxed{2.72 \text{ cm}^3}$$

8) A glass flask whose volume is 1000 cm^3 at 0°C is filled level full of mercury, When the flask and mercury are heated to 100°C , 15.2 cm^3 of mercury overflow. if coefficient of volume expansion for mercury is 0.000182 per Celsius degree
 → compute the coefficient of linear expansion of the glass (α)

~ answer ~

$$\begin{aligned}\Delta V_{\text{over}} &= \Delta V_{\text{Hg}} - \Delta V_{\text{glass}} \\ &= \beta_{\text{Hg}} V \Delta T - \beta_{\text{glass}} V \Delta T \\ &= V \Delta T (\beta_{\text{Hg}} - \beta_{\text{glass}})\end{aligned}$$

$$15.2 = 1000 (100^\circ \text{C} - 0^\circ \text{C}) [0.000182 - \beta_{\text{glass}}]$$

$$15.2 = 18.2 - 10^5 \beta_{\text{glass}}$$

$$\therefore \beta_{\text{glass}} = \frac{18.2 - 15.2}{10^5} = 3 \times 10^{-5} \text{ } ^\circ \text{C}^{-1}$$

$$\therefore \alpha = \frac{\beta}{3} \quad (\beta = 3\alpha)$$

$$\boxed{\alpha = 1 \times 10^{-5} \text{ } ^\circ \text{C}^{-1}}$$

" QUIZ 5 "

13

1. If two objects are in thermal equilibrium with each other:

- A. they cannot be moving B. they cannot be undergoing an elastic collision
C. they cannot have different pressures ☒ D. they cannot be at different temperatures
E. they cannot be falling in Earth's gravitational field

ans: D

2. When two gases separated by a diathermal wall are in thermal equilibrium with each other:

- A. only their pressures must be the same B. only their volumes must be the same
C. they must have the same number of particles D. they must have the same pressure and the same volume
☒ E. only their temperatures must be the same

ans: E

3. A balloon is filled with cold air and placed in a warm room. It is NOT in thermal equilibrium with the air of the room until:

- A. it rises to the ceiling B. it sinks to the floor ☒ C. it stops expanding
D. it starts to contract E. none of the above

ans: C

4. Suppose object C is in thermal equilibrium with object A and with object B. The zeroth law of thermodynamics states:

- A. that C will always be in thermal equilibrium with both A and B B. that C must transfer energy to both A and B
☒ C. that A is in thermal equilibrium with B D. that A cannot be in thermal equilibrium with B
E. nothing about the relationship between A and B

ans: C

5. The zeroth law of thermodynamics allows us to define:

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10

$$\therefore \Delta L = (23 \times 10^{-6}) \times (30) \times \left(\frac{500}{9}\right)$$

5] An iron bar is kept at constant length while it is heated from 0°C to 10°C

(a) Find the generated stress.
(b) Find the Force to prevent the extension.

~ answer ~

$$(b) F = A I \alpha \Delta T = 48 \times 10^2 N$$