

# Heat Transfer: Conduction

**Example: Clinical thermometer**

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# Learning goals

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## Clinical Thermometer

- ▶ Example of the lumped capacity model  
→ Procedure



[1]

[1] [sciencing.com/different-parts-of-a-mercury-thermometer-12073649.html](http://sciencing.com/different-parts-of-a-mercury-thermometer-12073649.html)

# How does a liquid thermometer work



[1]

The tip is heated by contact with the object to be measured.

## Clinical Thermometer:

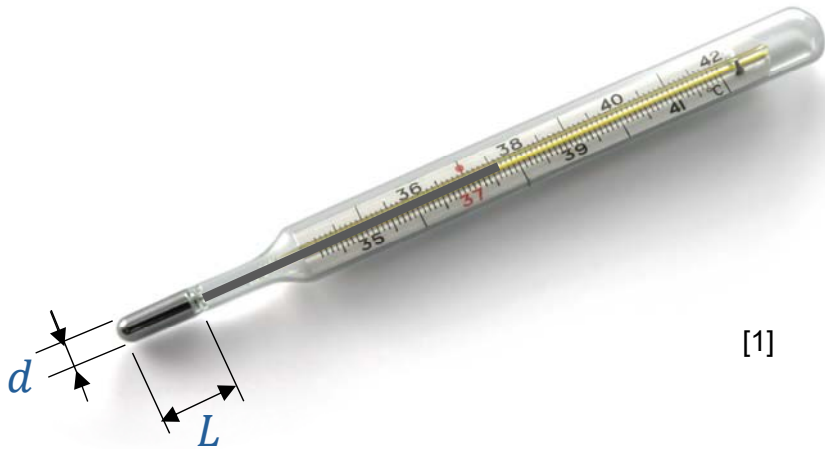
- ▶ Build with mercury as an indicator  
→ no longer in use today

### Principle:

linear volume expansion of the liquid through temperature rise

[1] [sciencing.com/different-parts-of-a-mercury-thermometer-12073649.html](http://sciencing.com/different-parts-of-a-mercury-thermometer-12073649.html)

# How does a liquid thermometer work



[1]

## Given parameter:

$$L = 30 \times 10^{-3} \text{ m}$$

$$d = 4 \times 10^{-3} \text{ m}$$

$$T_0 = 20^\circ \text{C}$$

$$c_{Hg} = 140 \text{ J}/(\text{kg} \cdot \text{K})$$

$$\rho_{Hg} = 14 \times 10^3 \text{ kg}/\text{m}^3$$

$$\lambda_{Hg} = 9 \text{ W}/(\text{m} \cdot \text{K})$$

## Problem discription:

- A child measures its body temperature at two different times:

$$t_1 = 40 \text{ s} \quad T_1 = 34^\circ \text{C} \quad t_2 = 100 \text{ s} \quad T_2 = 39^\circ \text{C}$$

- How high is the fever?
- What is the heat transfer coefficient  $\alpha$ ?
- How long does it take to determine the temperature with a precision of 0.1K?
- Are the assumptions of the calculation valid?

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# Calculation of the fever temperature

## a) How high is the fever?

Still unknown:

- ▶ Parameter  $m$
- ▶ „Fever temperature“  $T_f$

In principle calculable,  
requires iteration:

- ▶ simply start with good  
estimate:  $T_f = 40^\circ\text{C}$

Lumped capacity model:

$$\Theta^* = \frac{T - T_0}{T_f - T_0} = 1 - e^{-\underbrace{\frac{\alpha A}{\rho c_p V}}_m t}$$

$$1 - \frac{T - T_0}{T_f - T_0} = \frac{T_f - T}{T_f - T_0} = e^{-mt}$$

$$\ln \frac{T_f - T}{T_f - T_0} = -mt$$

$$\begin{cases} \ln \frac{T_f - T_1}{T_f - T_0} = -mt_1 & (1) \\ \ln \frac{T_f - T_2}{T_f - T_0} = -mt_2 & (2) \end{cases}$$

$$\frac{(1)}{(2)}: \frac{t_1}{t_2} = \frac{\ln \frac{T_f - T_1}{T_f - T_0}}{\ln \frac{T_f - T_2}{T_f - T_0}}$$

substitute 0.4

0.402 match

$$m = -\frac{1}{t_1} \ln \frac{T_f - T_1}{T_f - T_0} \approx 0.03$$

## Calculation of the heat transfer coefficient

b) How large is  $\alpha$  ?

Lumped capacity model:

$$m = \frac{\alpha A}{\rho c_p V} \approx 0.03$$

$$\text{with } A = \pi d L; V = \frac{\pi d^2 L}{4} \Rightarrow \frac{A}{V} = \frac{4}{d}$$

$$\alpha = \frac{m \rho c_p d}{4} = 58.8 \frac{\text{W}}{\text{m}^2 \text{K}}$$

d) Are the assumptions of the lumped capacity model correct?

$$Bi \ll 1 ?$$

$$Bi = \frac{\alpha \cdot \frac{d}{2}}{\lambda_{Hg}} = 0.013 \ll 1 \quad \checkmark$$

## Calculation of the measurement time

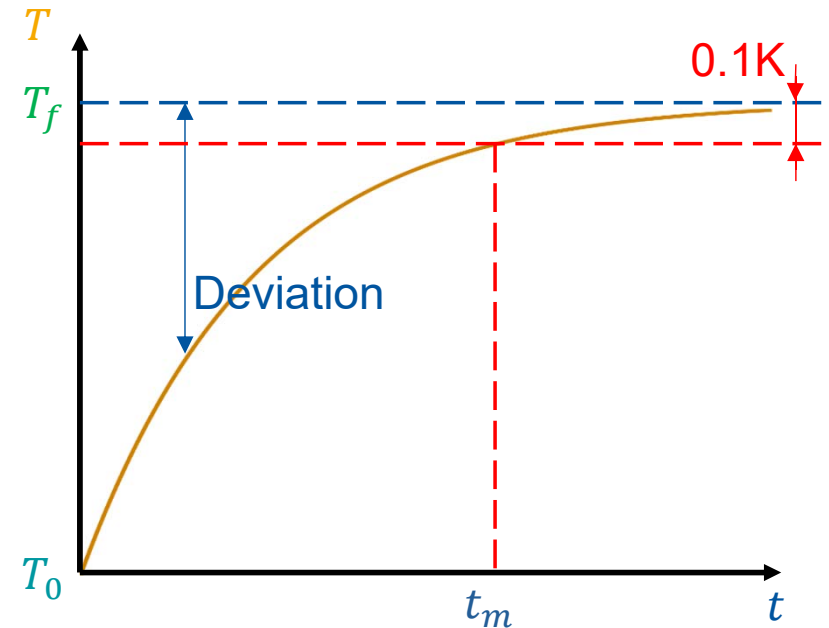
c) How long must be measured to determine the temperature with an accuracy of 0.1K?

Body with “high” thermal conductivity:

► Measuring time  $t_m$  for a 0.1K precision

$$t_m = -\frac{1}{m} \ln \frac{T_f - T}{T_f - T_0}$$

$$t_m = -\frac{1}{0.03} \ln \frac{0.1}{40-20} = 176.6\text{s}$$



## Comprehension questions

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For safety reasons, mercury thermometers are no longer available in retail. Thermometers filled with alcohol are also hardly used anymore.

**Why? What are the disadvantages of these measuring instruments?**

The standard devices currently in use are digital thermometers.

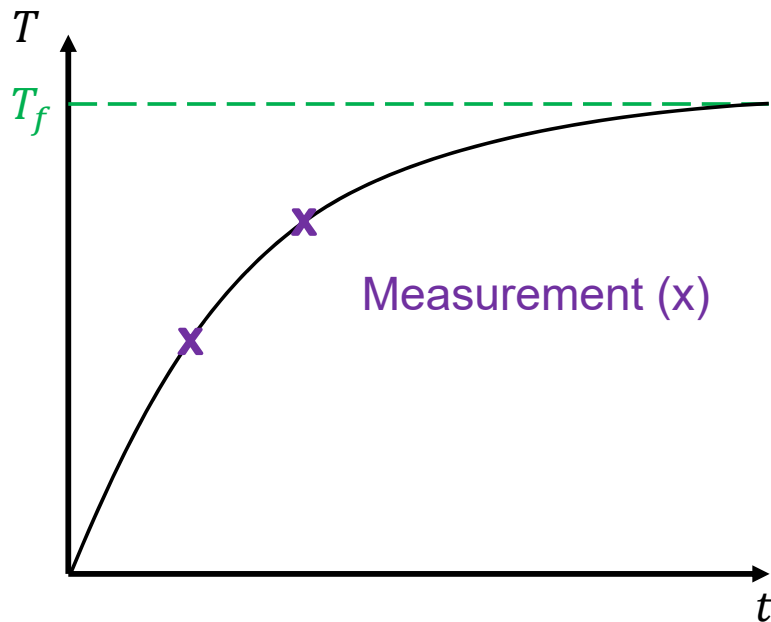
**How can the body temperature be determined with them?**

*I will present the solution after approx. 15 sec on the next slide (stop the video here if you want to think for yourself first).*



## Resolution to the question

### Digital device:



is stored on a chip in the device

with unknown  $m$  and  $T_f \rightarrow$   
2 measurements and  $T_f$  can **be calculated**  
by evaluating the stored function