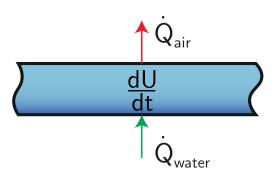


## EB - Cond. - Body 6

Ice is continuously increasing in thickness. Convection occurs on both sides. Derive the differential equation to describe the ice layer thickness s(t). The enthalpy of fusion is h. Take  $T_2 > T_w > T_1$ . The cooling of the ice, contrary to freezing, is energetically negligible,  $c_i \cdot \Delta T << h$ .



## Energy balance:

$$\frac{dU}{dt} = \dot{Q}_{water} - \dot{Q}_{air}$$

The heat transfer can be classified as transient, for that reason the change of internal energy over time equals the sum of the in and outgoing fluxes.

## Change of internal energy over time:

$$\frac{dU}{dt} = -\rho A h \frac{ds}{dt}$$

The internal energy can be described by use of the enthalpy of fusion. This is denoted with an negative sign, since solidification of water costs energy.

## Heat fluxes:

$$\dot{Q}_{water} = \alpha_2 A (T_2 - T_w)$$

$$\dot{Q}_{air} = (T_w - T_1)A\frac{1}{\frac{1}{\alpha_1} + \frac{s(t)}{\lambda}}$$