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# Oxide-mediated self-limiting recovery of field effect mobility in plasma-treated MoS<sub>2</sub>

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Jakub Jadwiszczak

Trinity College Dublin, Ireland

28 November 2017

Materials Research Society Fall Meeting 2017  
Boston, USA.

# MoS<sub>2</sub>-plasma interaction in the literature

## O<sub>2</sub> plasma

- Oxygen forms MoO<sub>3</sub> through sulfur-replacing reaction.
- This hampers device conductivity and causes p-type doping.

## Ar plasma

- Argon won't react chemically, but will make sulfur vacancies.
- It has been found to cause 2H → 1T polytype shift by displacing top sulfuric layer.

For our experiment we used a 1:3 mixture of O<sub>2</sub>:Ar gas.

# MoS<sub>2</sub>-plasma interaction in the literature

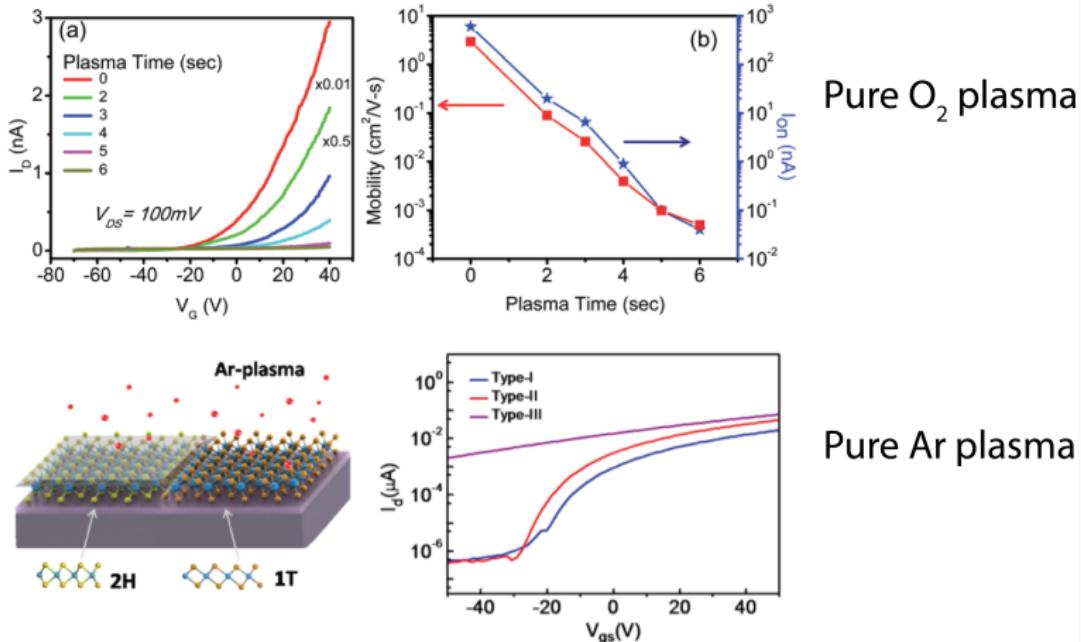


Figure 1: Top panel: Islam, M. et al. Nanoscale 6.17 (2014): 10033-10039.

Bottom panel: Zhu, J. et al. Journal of the American Chemical Society 139.30 (2017): 10216-10219.

# Electrical testing with increasing plasma exposure

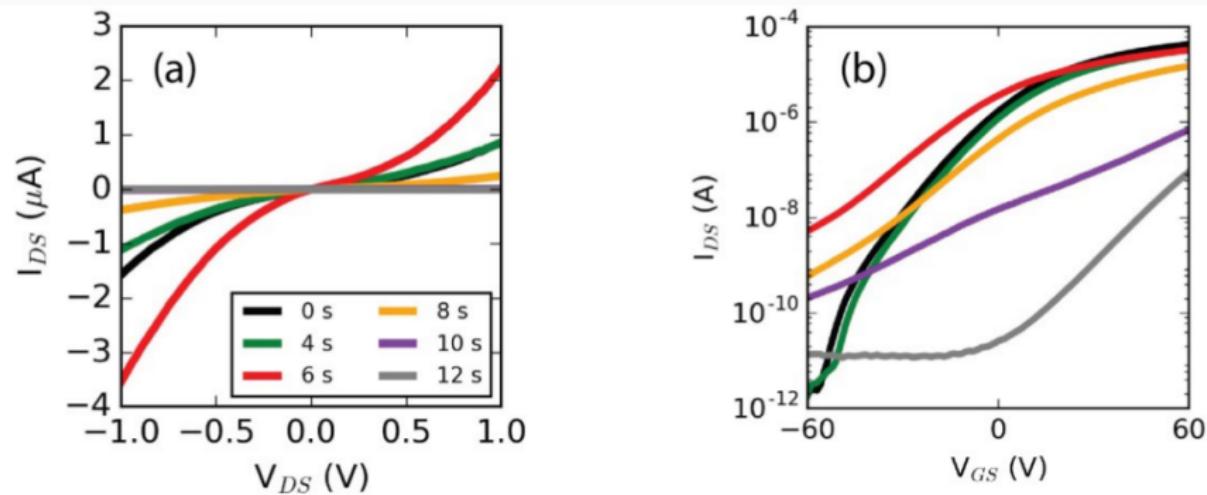
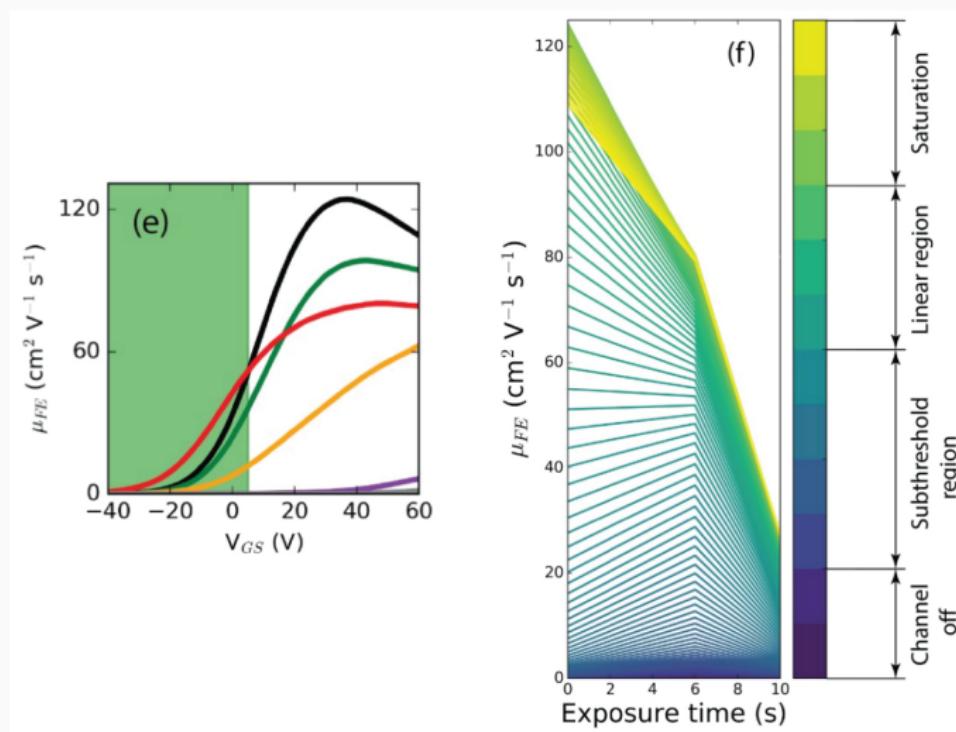


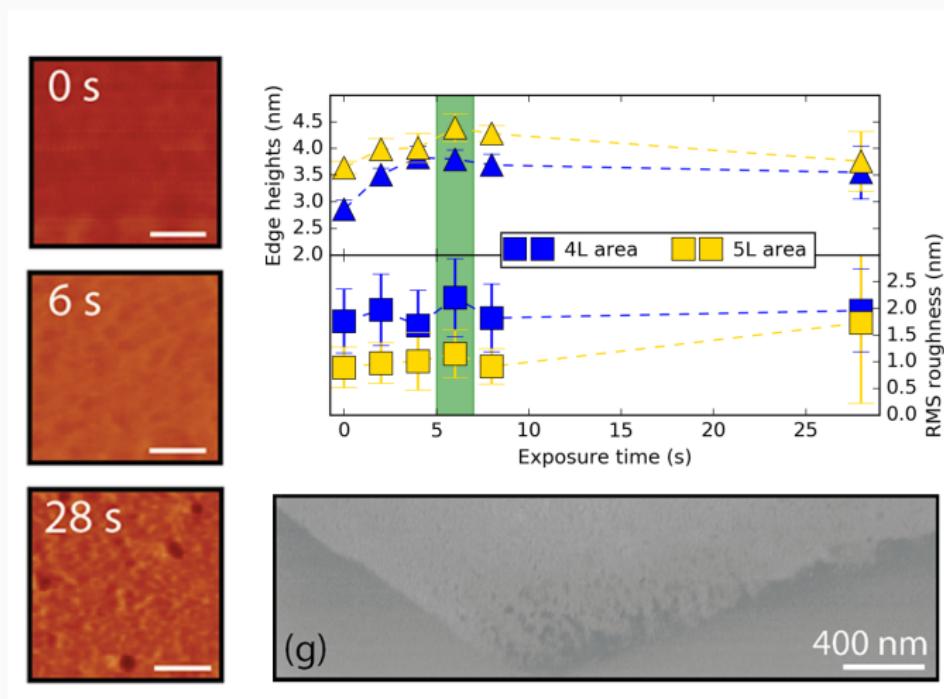
Figure 2: (a) IV sweeps at no gate bias. (b) Gate sweeps at drain-source bias of 1V.

# Mobility boost observed at 6 seconds



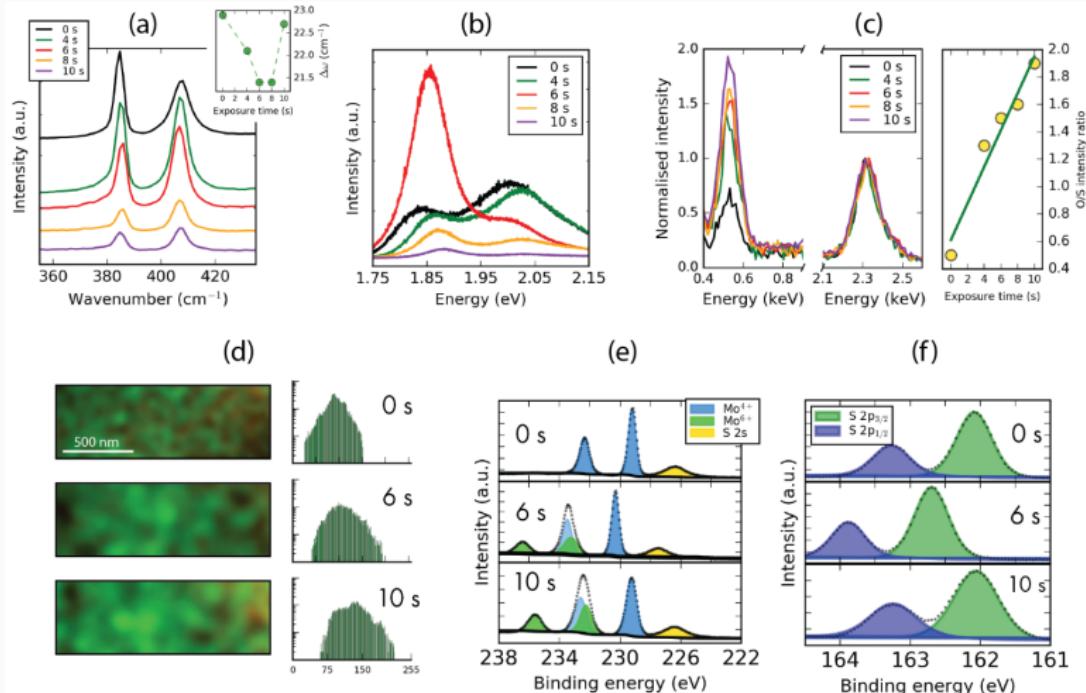
**Figure 3:** (e) Mobility across the whole gate bias range. (f)  $\mu_{FE}$  extracted over time at each gate bias.

# AFM & SEM indicate material change to flake surface



**Figure 4:** Flake increases in height with plasma treatment; patches of oxide form on the surface.

# Spectroscopies confirm presence of $\text{MoO}_3$



**Figure 5:** (a) Raman and (b) PL suggest decoupling of top layer. (c)-(d) EDX mapping shows increased oxygen content relative to sulfur. (e)-(f) XPS confirms  $\text{MoO}_3$  presence on the surface and removal of sulfur.

# Surface damage surveyed by TEM

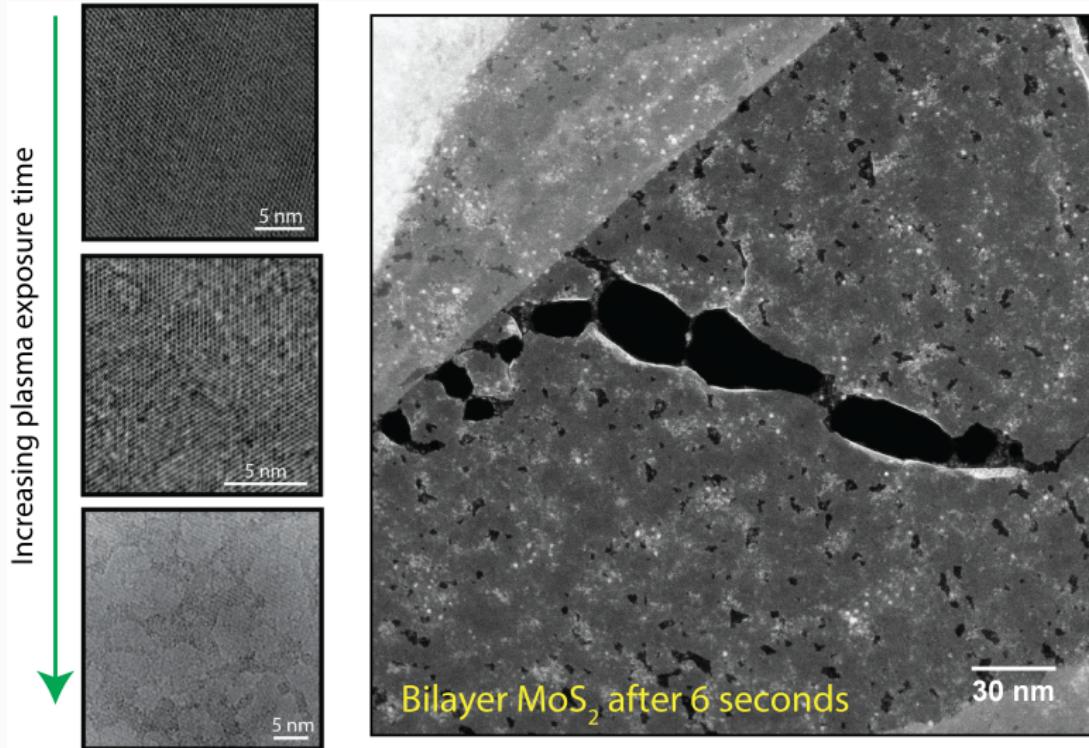
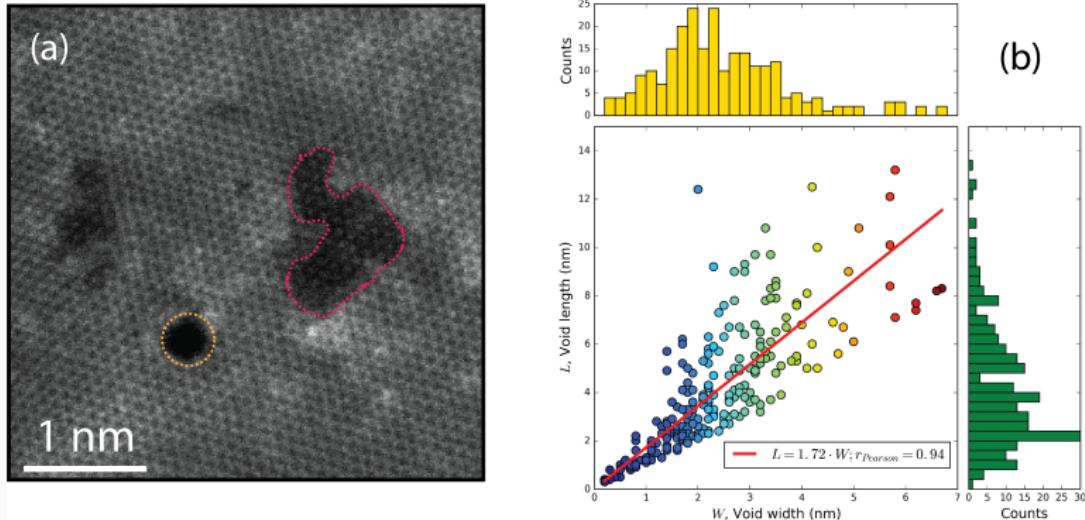


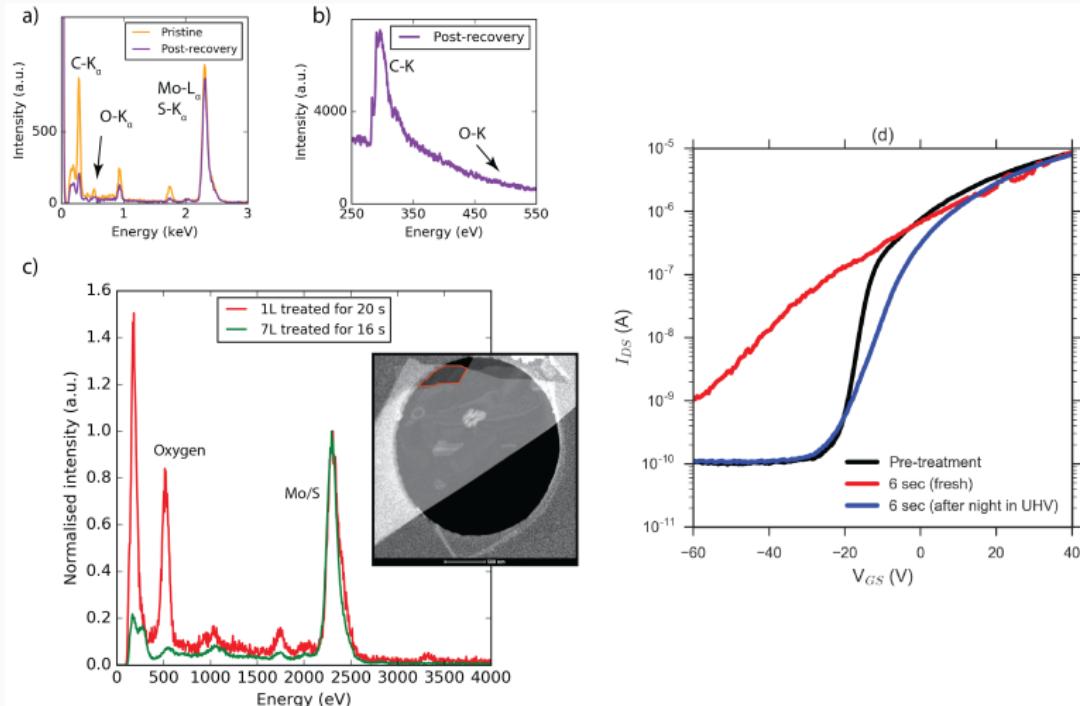
Figure 6: (left) HRTEM, (right) HAADF ACSTEM.

# Size distribution of oxide-mediated etched pits at 6 seconds



**Figure 7:** (a) Perforation circled in orange, pit circled in magenta. (b) Lateral void dimensions increase isotropically with time.

# Oxides desorb from pits over time at UHV



**Figure 8:** (a) EDX in ACSTEM. (b) EELS in ACSTEM. (c) EDX in TEM. (d) Electrical testing overnight.

# Proposed model of mobility enhancement

● Oxygen   ● Sulfur   ● Molybdenum

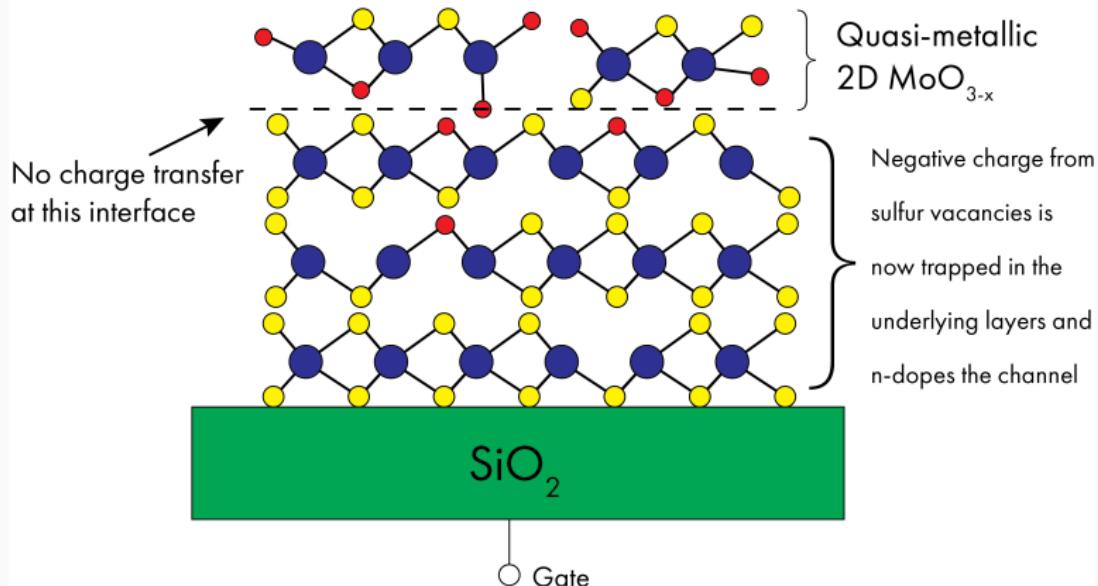
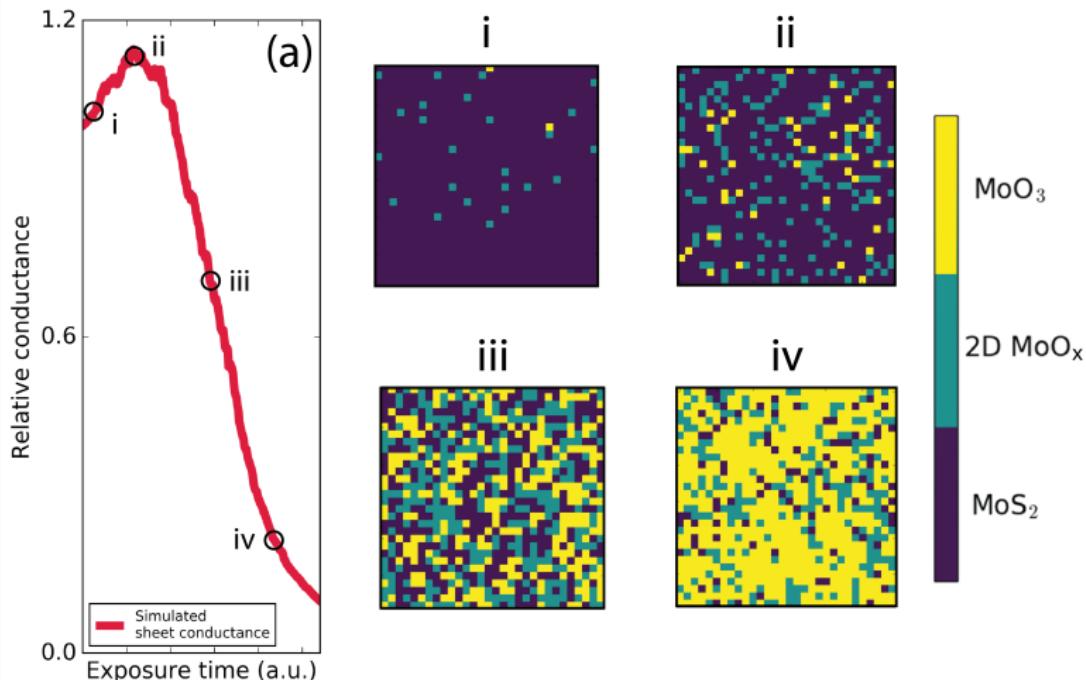


Figure 9: f

# Resistor network simulation



**Figure 10:** Conductance change across network as nodes convert from semiconducting  $\text{MoS}_2$  to highly conductive sub-stoichiometric  $\text{MoO}_x$  and into insulating  $\text{MoO}_3$ .

## Conclusions:

2D  $\text{MoO}_{3-x}$  forms at 6 seconds on the top layer, screens the  $\text{MoS}_2$  channel and enhances conductivity.

It is qualitatively different from the insulating  $\text{MoO}_3$  phase which forms at longer treatment times, and is unstable at high vacuum.

Its existence may be exploited in future van der Waals devices.

## Acknowledgements:

- Prof. H. Zhang, Dr. D. Fox, Pierce Maguire (**TEM, SEM, EDX**)
- Prof. M. Ferreira, Colin O'Callaghan, Eamonn Weitz (**simulations**)
- Prof. J. Boland, Darragh Keane (**AFM**)
- Prof. G. Duesberg, Conor Cullen (**Raman & PL**)
- Prof. L. Bradley, John Gough (**PL**)
- Prof. C. McGuinness (**XPS**)
- Prof. Y. Zhou, Ian O'Reilly (**electrical**)
- Prof. V. Nicolosi, Dr. A. Shmeliov, Clive Downing (**ACSTEM, EELS**)