



Trinity College Dublin
Coláiste na Tríonóide, Baile Átha Cliath
The University of Dublin

Oxide-mediated self-limiting recovery of field effect mobility in plasma-treated MoS₂

Jakub Jadwiszczak

Trinity College Dublin, Ireland

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MoS₂-plasma interaction in the literature

O₂ plasma

- Oxygen forms MoO₃ through sulfur-replacing reaction.
- This hampers device conductivity and causes p-type doping.

Ar plasma

- Argon won't react chemically, but will make sulfur vacancies.
- It has been found to cause 2H → 1T polytype shift by displacing top sulfuric layer.

For our experiment we used a 1:3 mixture of O₂:Ar gas.

MoS₂-plasma interaction in the literature

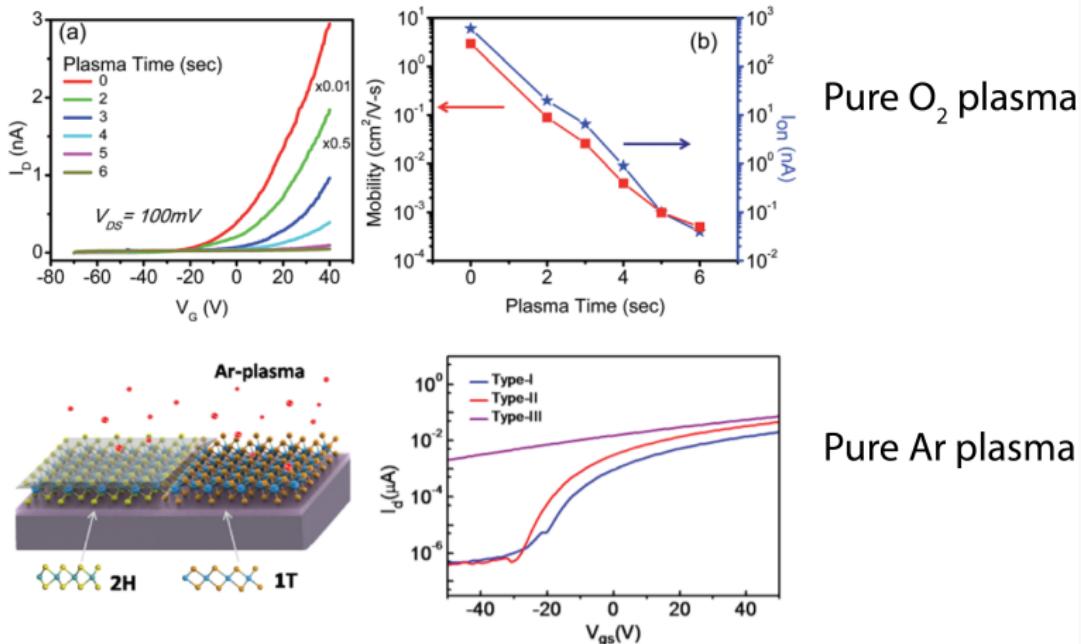


Figure 1: Top panels: Islam, M. et al. Nanoscale 6.17 (2014): 10033-10039.

Bottom panels: Zhu, J. et al. Journal of the American Chemical Society 139.30 (2017): 10216-10219.

Electrical testing with increasing plasma exposure

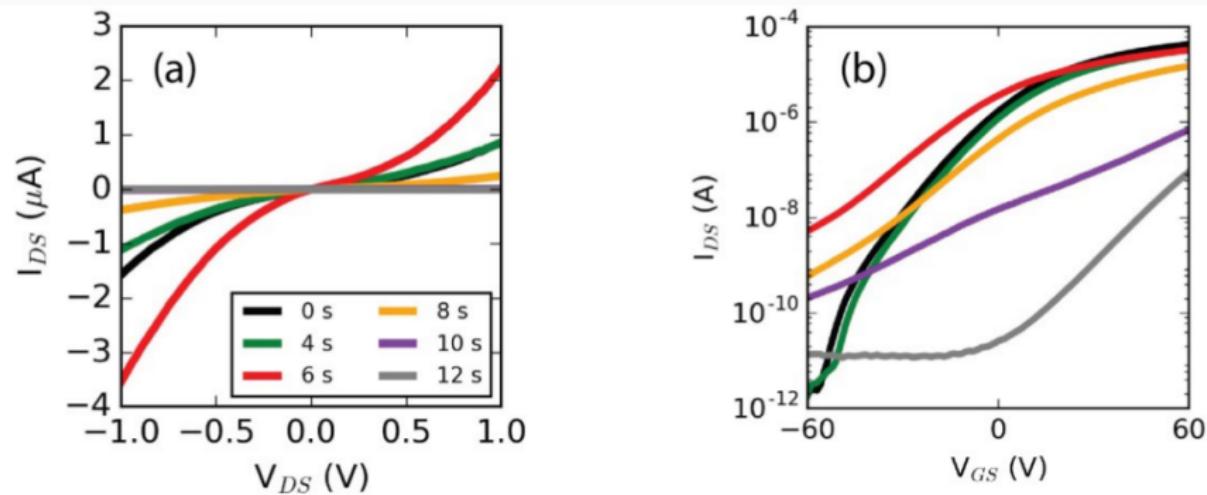


Figure 2: (a) IV sweeps at no gate bias. (b) Gate sweeps at drain-source bias of 1V.

Mobility boost observed at 6 seconds

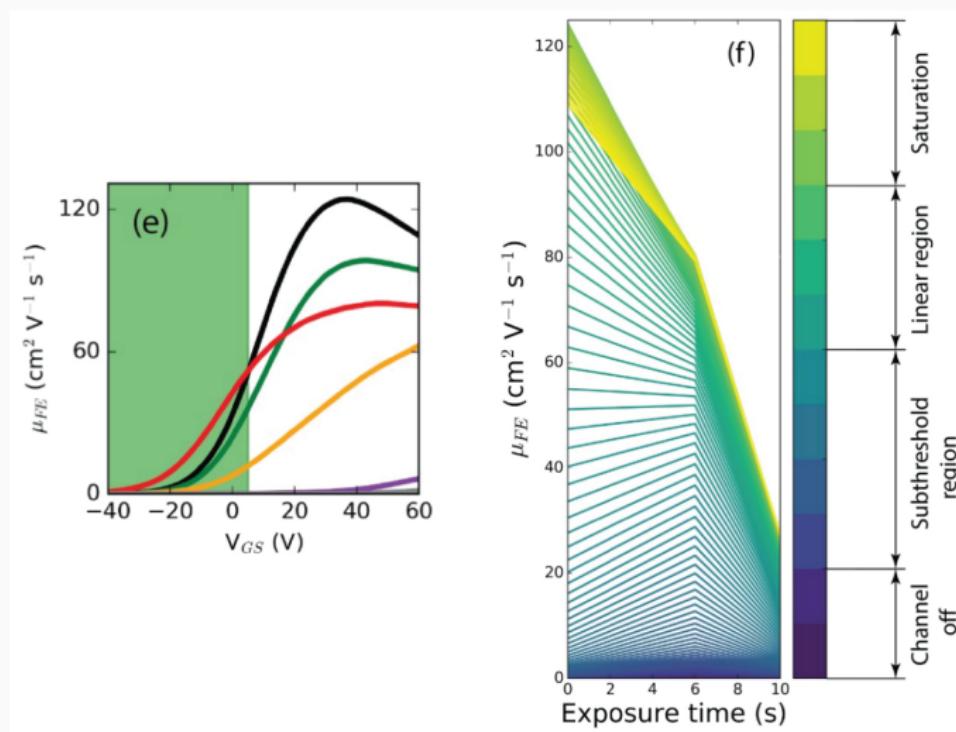


Figure 3: (e) Mobility across the whole gate bias range. (f) μ_{FE} extracted over time at each gate bias.

AFM & SEM indicate material change to flake surface

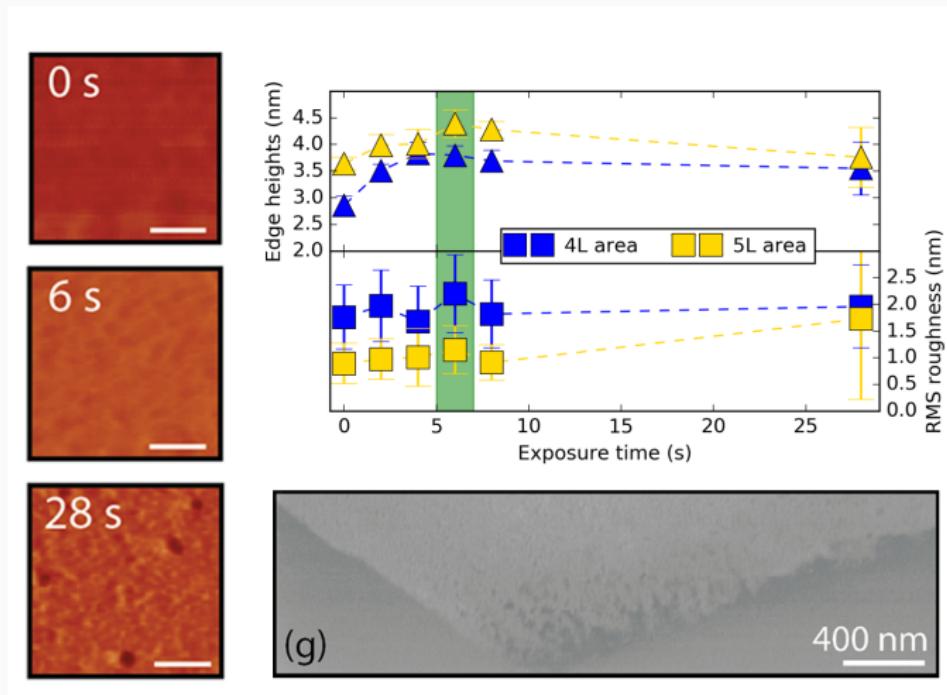


Figure 4: Flake increases in height with plasma treatment; patches of oxide form on the surface. Scale bar in AFM maps is 200 nm.

Spectroscopies confirm presence of MoO_3

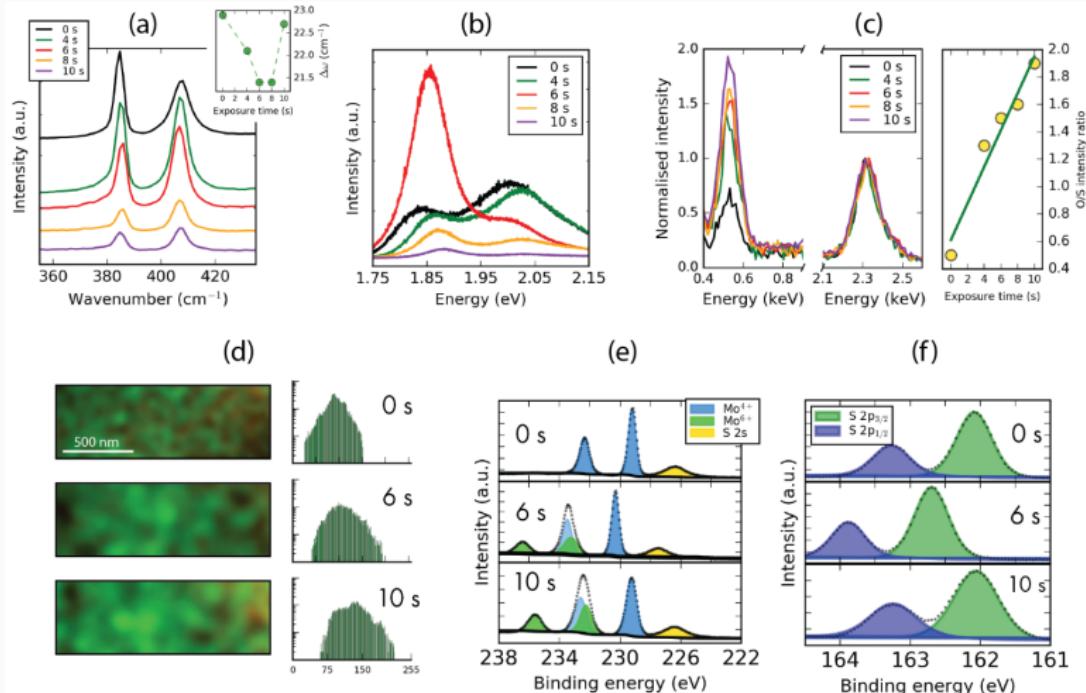


Figure 5: (a) Raman and (b) PL suggest decoupling of top layer. (c)-(d) EDX mapping shows increased oxygen content relative to sulfur. (e)-(f) XPS confirms MoO_3 presence on the surface and removal of sulfur.

Surface damage surveyed by TEM

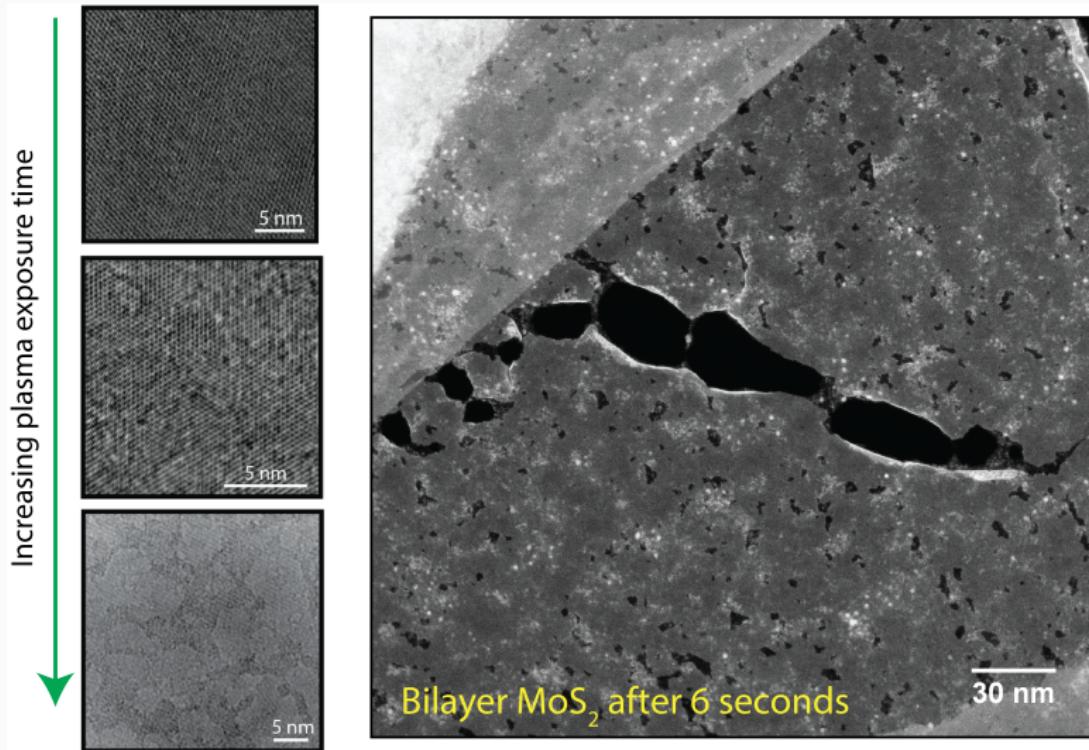


Figure 6: (left) HRTEM, (right) HAADF ACSTEM.

Size distribution of oxide-mediated etched pits at 6 seconds

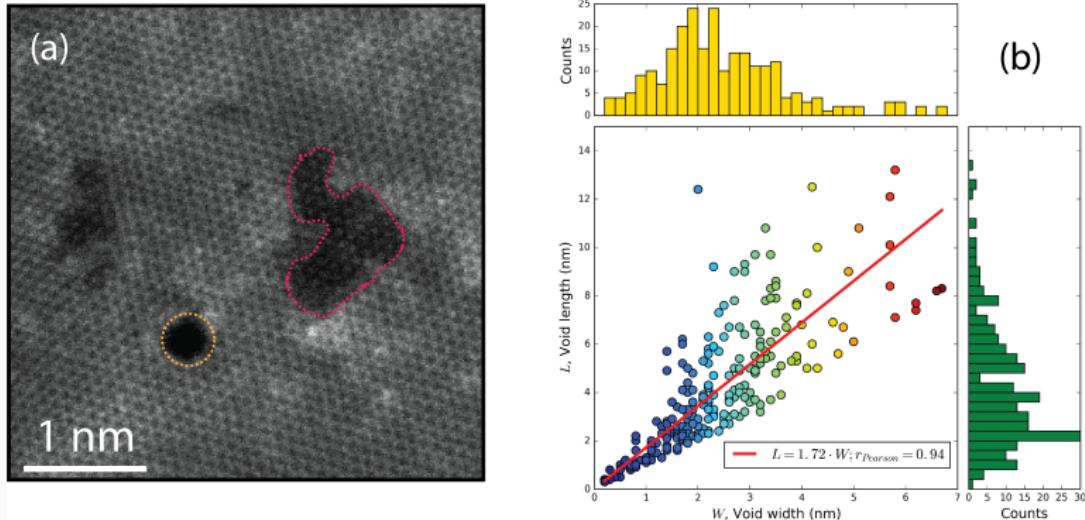


Figure 7: (a) Perforation circled in orange, pit circled in red. (b) Lateral void dimensions increase isotropically with time.

Oxides desorb from pits over time at UHV

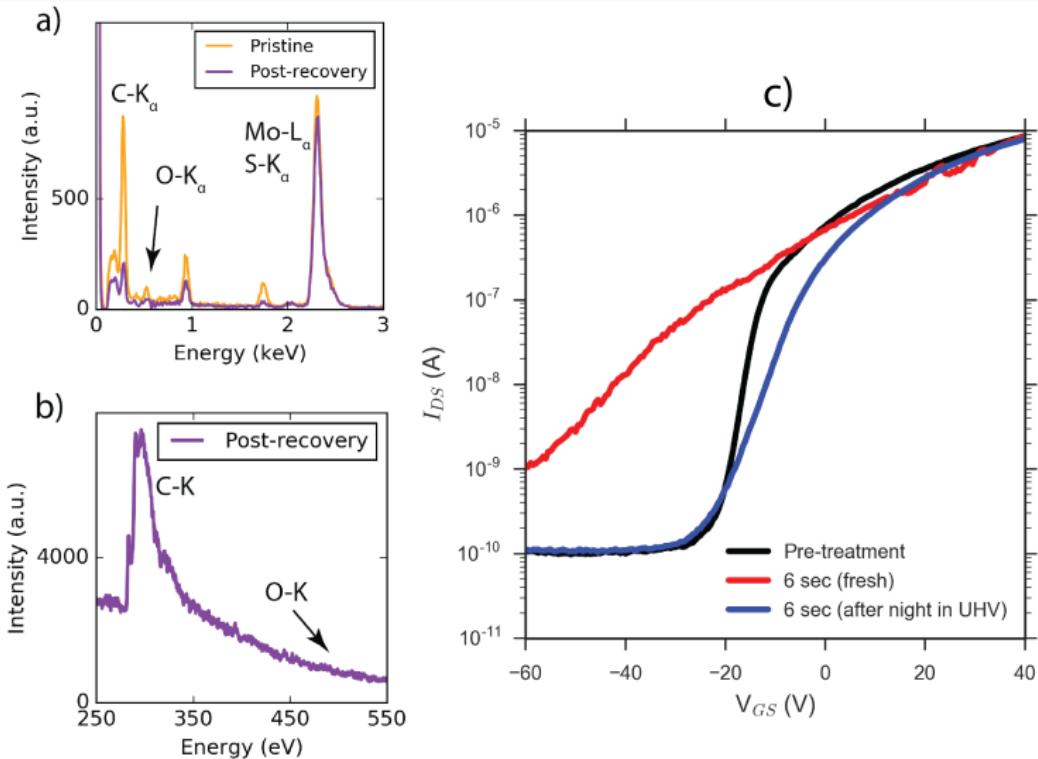


Figure 8: (a) EDX in ACSTEM. (b) EELS in ACSTEM. (c) Electrical testing overnight.

Proposed model of mobility enhancement

● Oxygen ● Sulfur ● Molybdenum

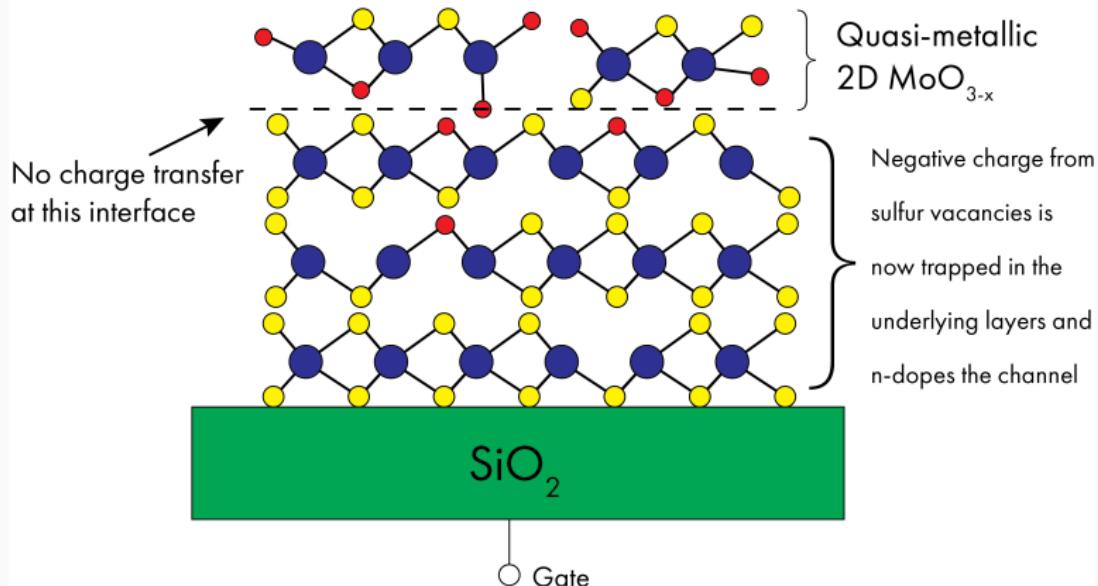


Figure 9: Cartoon of oxide-capped 4L MoS_2 device.

Conclusions:

2D MoO_{3-x} phase forms at 6 seconds on the top layer, electrically shields the MoS_2 channel and enhances conductivity.

It is qualitatively different from the insulating MoO_3 phase which forms at longer treatment times (volatility, electrical properties, etc.)

Its existence may be exploited in future van der Waals devices.

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