

**Summary of Problem Statement****Problem #** 3

Create a program that will help consumers determine the efficiency of their stoves.

**Known / Input**

Temp. of water (Initial\_temp) [deg F]  
Time (Time) [min]  
Power (P) [W]

**Unknown / Output**

Thermal Energy (Q) [J]  
Power (Power\_Burner) [W]  
Efficiency (Burner Efficiency) [%]

**Assumptions**

Air Pressure will not impact any results.  
The pan on the stove does not impact results (by absorbing heat and also because everyone has different pans at home).  
D = 1000 Density of water [Kg/m<sup>3</sup>] I put this in assumptions because in real life, the density of water varies at different temps.  
C = 4186 = Specific heat of water [J/Kg\*°C]

**Other Variables**

D = Density of water, M = D \* V  
Convert temp to Celsius: Initial\_temp = (Initial\_temp - 32) / 1.8  
Final\_temp = 100 [°C]  
Delta\_temp = (Final\_temp - Initial\_temp)

**Algorithm**

Inputs: User defined data about: Temp of water, Name and model of motor, Time it takes for the water to boil and the power that an oven is capable of putting in. Using input method.

Convert temp to Celsius using  $5/9(\text{Temp (deg F)} - 32)$ .

Convert time from min to seconds. Time = Time \* 60;

Convert volume from gallons into m<sup>3</sup> to solve for mass:  $V = (1 / .264 * 1000 / 100^3)$ ; [m<sup>3</sup>]

Define all intermediate values used for final calculations: Specific heat of water, Density of water

Change in temperature (deg C) = 100 - initial temp

Outputs: Thermal Energy of the stove, power output of the burner and efficiency of the stove

Calculations:

$Q = M * C * \text{Delta\_temp}$ ; Thermal Energy [J]

Power\_Burner = Q / Time; Power of the burner [W]

Efficiency = (Power\_Burner / P) \* 100; Percent efficiency of the burner

**Test Cases**

Using information from the test case with Stove Model: Charbake 5

Temp =  $(100 - (5/9)(69-32)) = 79.45$  [deg C]    T = 18\*60 = 1,080 [s]    V =  $(1 / .264) * (1000 / 100^3) = .00378$  [m<sup>3</sup>]

M = 1000 [kg/m<sup>3</sup>] \* .00378 [m<sup>3</sup>] = 3.78 [kg]

Q = 3.7878 [kg] \* 4186 [J/Kg\*°C] \* 79.444 [°C] = 1259642 [J] <<<< This number varies in a range of almost 1000 [J] based on the number of decimal points I use.

Power = 1259642 [J] / 1080 [s] = 1166 [W]

Efficiency =  $(1166 \text{ [W]} / 1350 \text{ [W]}) * 100 = 86.4\%$