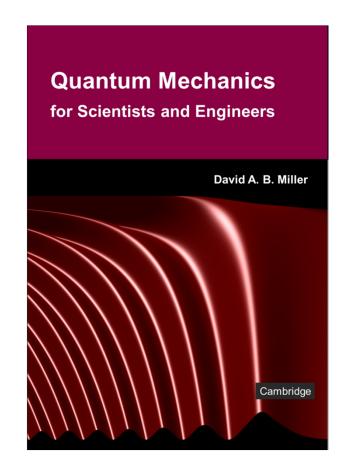
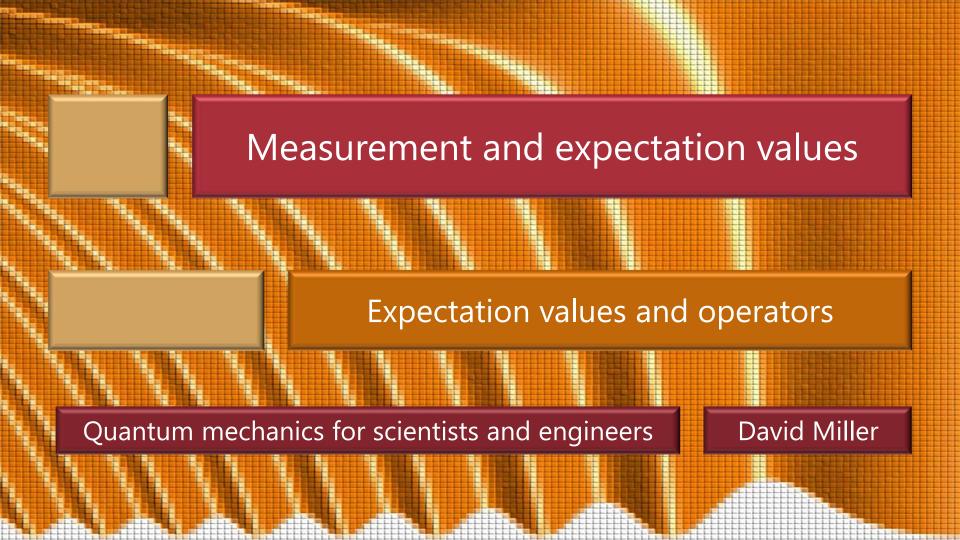
4.3 Measurement and expectation values

Slides: Video 4.3.3 Expectation values and operators

Text reference: Quantum Mechanics for Scientists and Engineers

Sections 3.9 – 3.10





Hamiltonian operator

In classical mechanics, the Hamiltonian is a function of position and momentum representing the total energy of the system

In quantum mechanical systems that can be analyzed by Schrödinger's equation t^2

we can define the entity
$$\hat{H} = -\frac{\hbar^2}{2m}\nabla^2 + V(\mathbf{r},t)$$

so we can write the Schrödinger equations as

$$\hat{H}\psi(\mathbf{r}) = E\psi(\mathbf{r}) \text{ and } \hat{H}\Psi(\mathbf{r},t) = i\hbar \frac{\partial \Psi(\mathbf{r},t)}{\partial t}$$

Hamiltonian operator

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The entity \hat{H}
  is not a number
     is not a function
        It is an "operator"
          just like the entity d / dz is a spatial derivative
            operator
             We will use the notation with a "hat" above
               the letter to indicate an operator
The most general definition of an operator is
  an entity that turns one function into another
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Hamiltonian operator

The particular operator \hat{H} is called the Hamiltonian operator Just like the classical Hamiltonian function it is related to the total energy of the system This Hamiltonian idea extends beyond the specific Schrödinger-equation definition we have so far which is for single, non-magnetic particles In general, in non-relativistic quantum mechanics the Hamiltonian is the operator related to the total energy of the system

Operators and expectation values

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Now we show a simple, important and
 general relation between
   the Hamiltonian operator
      the wavefunction, and
         the expectation value of the energy
To do so
   we start by looking at the integral
          I = \int \Psi^* (\mathbf{r}, t) \hat{H} \Psi (\mathbf{r}, t) d^3 \mathbf{r}
     where \Psi(\mathbf{r},t) is the wavefunction of
       some system of interest
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Operators and expectation values

In looking at this integral
$$I = \int \Psi^*(\mathbf{r},t) \hat{H} \Psi(\mathbf{r},t) d^3 \mathbf{r}$$

we will expand the wavefunction $\Psi(\mathbf{r},t)$ in

the (normalized) energy eigenstates $\psi_n(\mathbf{r})$

$$\Psi(\mathbf{r},t) = \sum c_n(t)\psi_n(\mathbf{r})$$

$$\hat{H}\Psi(\mathbf{r},t) = \left[-\frac{\hbar^2}{2m} \nabla^2 + V(\mathbf{r},t) \right] \Psi(\mathbf{r},t)$$

$$= \left[-\frac{\hbar^2}{2m} \nabla^2 + V(\mathbf{r}, t) \right] \sum_{n} c_n(t) \psi_n(\mathbf{r}) = \sum_{n} c_n(t) E_n \psi_n(\mathbf{r})$$

Operators and expectation values

So the integral becomes

$$\int \Psi^*(\mathbf{r},t) \hat{H} \Psi(\mathbf{r},t) d^3 \mathbf{r} = \int_{-\infty}^{\infty} \left[\sum_{m} c_m^*(t) \psi_m^*(\mathbf{r}) \right] \times \left[\sum_{n} c_n(t) E_n \psi_n(\mathbf{r}) \right] d^3 \mathbf{r}$$

Because of the orthonormality of the basis functions $\psi_n(\mathbf{r})$ the only terms in the double sum that survive

are the ones for which n = m

so
$$\int \Psi^*(\mathbf{r},t) \hat{H} \Psi(\mathbf{r},t) d^3 \mathbf{r} = \sum E_n |c_n|^2$$

But this is just the expectation value of the energy, so

$$\langle E \rangle = \int \Psi^* (\mathbf{r}, t) \hat{H} \Psi (\mathbf{r}, t) d^3 \mathbf{r}$$

Benefit of the use of operators

Question:

if we already knew how to calculate $\langle E \rangle$

from
$$\langle E \rangle = \sum_{n} E_{n} P_{n} = \sum_{n} E_{n} |c_{n}|^{2}$$

why use the new relation?

$$\langle E \rangle = \int \Psi^* (\mathbf{r}, t) \hat{H} \Psi (\mathbf{r}, t) d^3 \mathbf{r}$$

Answer:

We do not have to solve for the eigenfunctions of the operator to get the result

