

Nature of Invention: Process design

Applicant: Catalysta Industries Pvt. Ltd

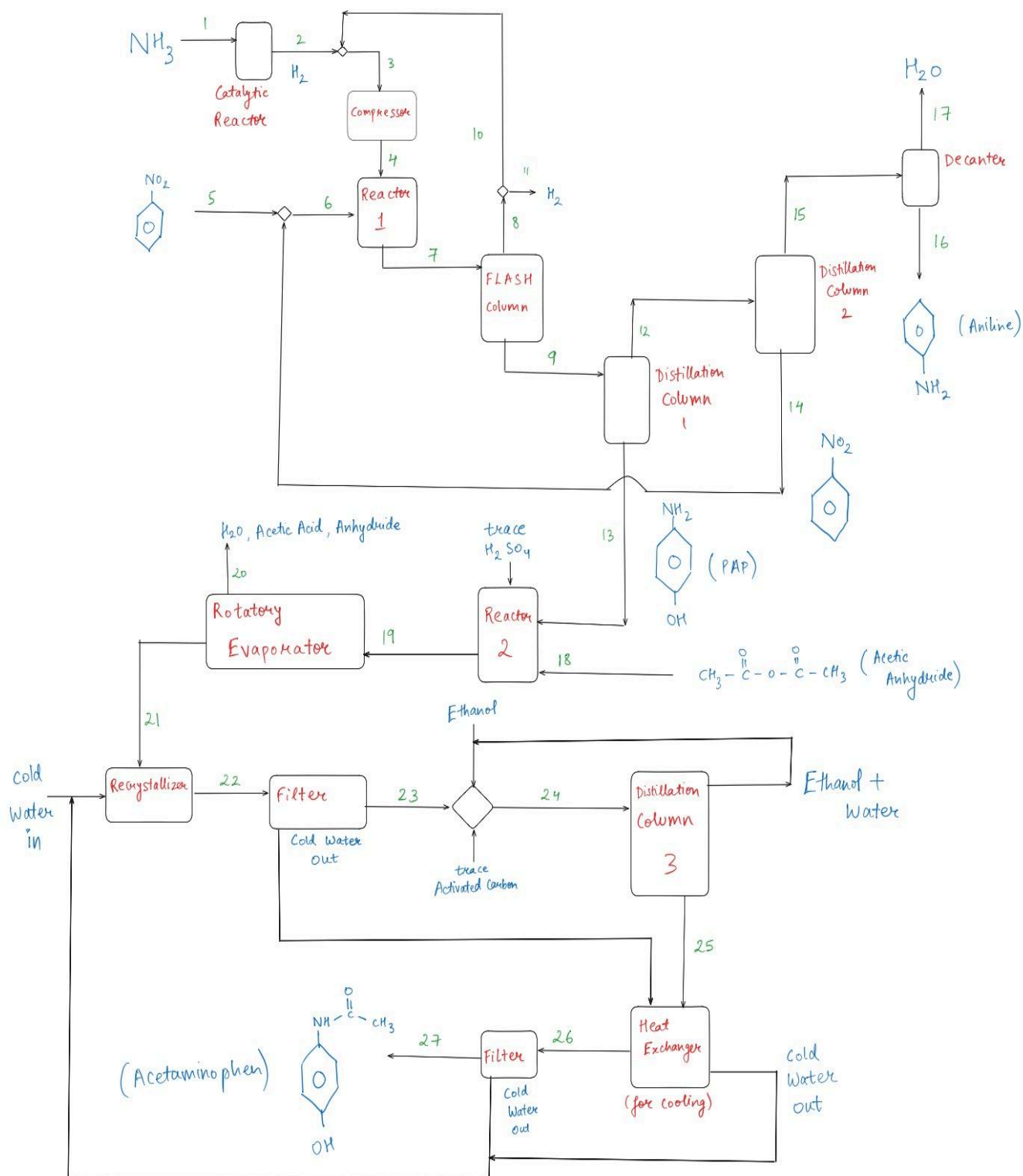
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Chemical Formula: $C_8H_9NO_2$ ($HOC_6H_4NHCOCH_3$)

Chemical Name: Acetaminophen

Process Title: Integrated Process for the Production and Purification of Paracetamol from Nitrobenzene.



Unit operations and process conditions:

unit operation	design gauge pressure (barg)	design temperature(C)
Reactor 1	4.71	121.11
Reactor 2	0	121.11
Flash drum	4.71	121.11

unit operation	design gauge pressure bottom(barg)	design temperature bottom(C)	operating bottom temperature(C)	bottom tray type
Distillation column 1	1.034	311.264	283.487	sieve
Distillation column 2	1.034	236.729	208.952	sieve
Distillation column 3	1.034	150	130	sieve

unit operation	design gauge pressure (barg)	vacuum design gauge pressure(barg)	operating temperature (C)	design temperature (C)
Decanter	1.03425	-1.00667	30	121.11

Material balance scaled-up process:**Specie data:**

Species	Molar Weight	Average Density (kg/m ³)
Ammonia	17	701
Hydrogen	2	0.08988
Nitrobenzene	123.11	1109
Aniline	93.13	998
Water	18	947.99
Para aminophenol	109.13	1250
Acetic Anhydride	102.09	1080
Acetic Acid	60.05	1050
Ethanol	46.07	789
Acetaminophen	151.163	1263

Material balance:**Assumptions:**

- 100% conversion of NH₃ to H₂ and N₂ is separated out.
- In flash drum 100% hydrogen move to stripper.
- In Distillation column 1 100% para aminophenol gets separated in stream 13 and all other species move to stream 12.
- In Distillation column 2 100% nitrobenzene is separated in stream 14 and all other species move to stream 15.

			kmol /day	kg/d ay		kmol /day	kg/d ay	kg/m ³	m ³ / day
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Stream	Remarks	Relevant Species	Molar Flow Rate	Mass Flow Rate	Mass Fraction	Total Molar Flow	Total Mass Flow	Average Density	Volume Flow Rate
1	90% conversion to pure H ₂	Ammonia	24.49	416.28	1	24.49	416.28	701	0.5938440377
2		Hydrogen	33.06	66.12	1	33.06	66.12	0.08988	735.6008077
3		Hydrogen	94.88	189.77	1	94.88	189.77	0.08988	2111.368564
4	PAP formed = $0.7 * 0.6$ * every 2 Hydrogen required; Aniline formed = $0.3 * 0.6$ * every 3 Hydrogen Required	Hydrogen	94.88	189.77	1	94.88	189.77	0.08988	2111.368564
5		Nitrobenzene	11.39	1401.75	1	11.39	1401.75	1109	1.263979849
6	Conversion of NB = 60%	Nitrobenzene	18.98	2336.26	1	18.98	2336.26	1109	2.106633082
7	NB used = PAP and Aniline formed	Paraaminophenol	7.97	869.80	0.3443052238	102.48	2526.25	1066.278673	2.369224558
	Selectivity of PAP = 70%	Aniline	3.42	318.12	0.1259250913				
	Used H ₂ = 2*PAP + 3*Aniline	Hydrogen	68.70	137.39	0.05438620213				

	Unreacted = 40% of Fed	Nitroben zene	7.59	934. 50	0.36991 63174				
	Water = 2*Aniline + PAP	Water	14.8 0	266. 44	0.10546 71655				
8		Hydrogen	68.7 0	137. 39	1	68.7 0	137. 39	0.0898 8	1528.63 084
9		Paraamin ophenol	7.97	869. 80	0.36410 76564	33.7 8	2388 .86	1127.5 99647	2.11853 5851
		Aniline	3.42	318. 12	0.13316 75696				
		Nitroben zene	7.59	934. 50	0.39119 1751				
		Water	14.8 0	266. 44	0.11153 3023				
10		Hydrogen	61.8 3	123. 65	1	61.8 3	123. 65	0.0898 8	1375.76 7756
11		Hydrogen	6.87	13.7 4	1	6.87	13.7 4	0.0898 8	152.863 084
12		Aniline	3.42	318. 12	0.20941 84196	25.8 1	1519 .06	1057.5 14033	1.43644 2367
		Nitroben zene	7.59	934. 50	0.61518 55026				
		Water	14.8 0	266. 44	0.17539 60778				
13	PAP needed = Acetaminop hen formed	Paraamin ophenol	7.97	869. 80	1	7.97	869. 80	1250	0.69584 1854
14		Nitroben zene	7.59	934. 50	1	7.59	934. 50	1109	0.84265 32329
15		Aniline	3.42	318. 12	0.54420 61591	18.2 2	584. 56	975.20 575	0.59941 76366

		Water	14.8 0	266. 44	0.45579 38409				
16		Aniline	3.42	318. 12	1	3.42	318. 12	998	0.31875 62299
17		Water	14.8 0	266. 44	1	14.8 0	266. 44	947.99	0.28105 44503
18	PAP : Anhydride = 1 : 3	Acetic Anhydride	23.9 1	2441 .07	1	23.9 1	2441 .07	1080	2.26025 3281
19	Post Filter Yield : 83%	Acetamin ophen	7.97	1204 .82	0.36390 35805	31.8 8	3310 .82	1142.2 57498	2.89848 8365
		Acetic Anhydride	15.9 4	1627 .38	0.49153 45227				
		Acetic Acid	7.97	478. 62	0.14456 18968				
20		Acetic Anhydride	15.9 4	1627 .38	0.77273 5874	23.9 1	2106 .00	1073.1 82076	1.96238 9084
		Acetic Acid	7.97	478. 62	0.22726 4126				
21		Acetamin ophen	7.97	1204 .82	1	7.97	1204 .82	1263	0.95393 45029
22		Acetamin ophen	7.97	1204 .82	1	7.97	1204 .82	1263	0.95393 45029
23	Recrystall isation Filter causes drop by 10%	Acetamin ophen	7.17	1084 .34	1	7.17	1084 .34	1263	0.85854 10526
24		Acetamin ophen	7.17	1084 .34	1	7.17	1084 .34	1263	0.85854 10526

25		Acetaminophen	7.17	1084.34	1	7.17	1084.34	1263	0.8585410526
26	Last Filter causes remaining drop	Acetaminophen	7.17	1084.34	1	7.17	1084.34	1263	0.8585410526
27	Basis Given	Acetaminophen	6.62	1000.00	1	6.62	1000.00	1263	0.7917656374

Capital cost (only for the reactor):

Equipment	Design Capacity (L)	No. of units	Cost/unit (\$ for year 2014)	Total Cost (\$ for year 2014)
Reactor 1 (Jacketed reactor, agitated, Carbon steel, 25 - 150psi pressure)	118.46	1	10,000	10,000
Reactor 2 (Jacketed reactor, agitated, Carbon steel, atm-25 psi pressure)	207.034	1	8,100	8,100

References: Provide reference for a research paper or an actual patent.

1. [Matches' Reactor cost - autoclave, fermenter, kettle, mixer settler.](#)
2. e-journal: [Multistep Synthesis of Paracetamol in Continuous Flow](#)

3. Patent: [FR3109581A1 - Continuous paracetamol synthesis process - Google Patents](#)
4. link: [Synthesis of paracetamol from p-aminophenol - Labmonk](#)
5. <https://arxiv.org/ftp/arxiv/papers/2110/2110.15750.pdf>

List the contributions of each author:

- Authors 1 and 2 contributed to the block diagram and calculations.
- Author 3 contributed to block diagram and deciding process conditions.
- Author 4 calculated the capital cost of the reactors.
- Author 5 helped find good resources and references.
- Author 6 proofread the document and helped in correcting some information.

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