

# General Chemistry Lab

**Dr. Mohammed Kabir Uddin**

**PhD (Australia), MS (Canada), MPhil in Organic Chemistry (BUET), M.Sc in Organic Chemistry & B.Sc (CU)**

Assistant Professor

Department of Biochemistry and Microbiology

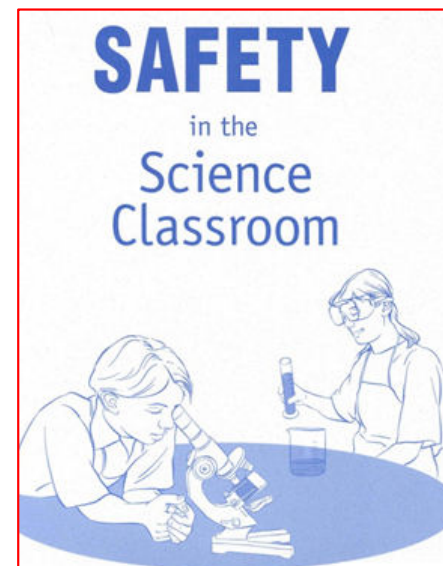
School of Health & Life Sciences

Ph: 01796585904

[kabirmuddin@gmail.com](mailto:kabirmuddin@gmail.com)

## Laboratory procedures

- **Practical laboratory science has many objectives and opinion:**
- **The growth of experience and knowledge in laboratory safety**
- **Experience in scientific instrumentation**



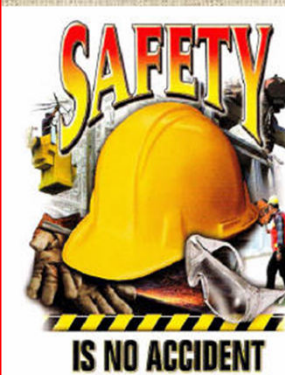
## Laboratory Attendance

- **All laboratory classes must be attended except in cases of illness**

## Laboratory Hours

- **No student may work in the laboratory outside the allotted class periods without permission from the laboratory supervisor**

## Safety and Rules of the Lab



## Safety

### Accidents

- From time to time accidents do occur in the laboratory
- If you receive a chemical burn from acid, alkali, etc., immediately flood the affected area with water
- For all accidents, immediately informed to demonstrator



## Allergies

- If you are aware that you suffer from allergic reactions to any type of chemical, please inform the Unit Coordinator and your laboratory demonstrator before the commencement of laboratory classes

## Safety Glasses

- In order to minimise the risk of eye damage, safety glasses must be worn in the laboratory by all persons at all times





## Chemical Safety

- Read all labels twice before removing a chemical from the container
- Only use the type and amount of chemical instructed to use
- Never touch, taste, or smell a chemical unless instructed by the teacher
- Never mix chemicals unless instructed to do so



## Toxic and Corrosive Chemicals

- Splashing by **corrosive chemicals** most frequently arises when **test tubes are heated too rapidly**
- Test tubes should never be placed inside a **Bunsen flame**, but held above it and **shaken gently**
- **Concentrated acids** should always be diluted by pouring the acid carefully into water; **never add water to concentrated acid solutions**



## Toxic and Corrosive Chemicals

- If either **skin or clothing** is **splashed with any corrosive liquid**, immediately douse with large quantities of water over
- *No liquids* are to be **pipetted by mouth**; manually operated **pipette pumps** are available in the laboratory





## Forbidden

- **Smoking, eating and drinking** - are absolutely forbidden at all times in the laboratory



## Oxidising Agents

- **Chemicals such as nitric acid and hydrogen peroxide, permanganates, persulfates, chlorates and (particularly) perchlorates, can be very dangerous when in contact with oxidisable materials**





## Reducing Agents

- Some reducing agents, such as sodium metal and lithium aluminium hydride, can react with explosive violence on contact with water



## Fire

- Fires can arise from incompletely extinguished matches or spills being placed in waste paper baskets
- For fire-fighting, every laboratory room is provided with Bromochlorodifluoromethane (BCF) or carbon dioxide extinguishers



Going to the Lab?  
Dress for the occasion.



Wear appropriate footwear!

IOWA STATE UNIVERSITY  
Department of Environmental Health and Safety  
515-284-0359 www.ehs.iastate.edu

## Clothing

- You must be adequately clothed to be admitted to a laboratory. Covered shoes (not sandals or thongs) must always be worn.



## Glassware

- Dispose of **broken glassware** safely after notifying your demonstrator



## Summary of important rules

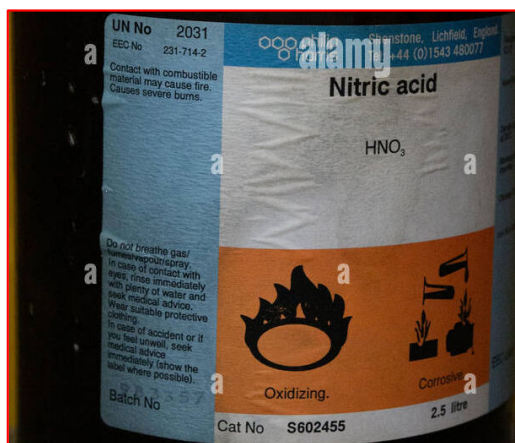
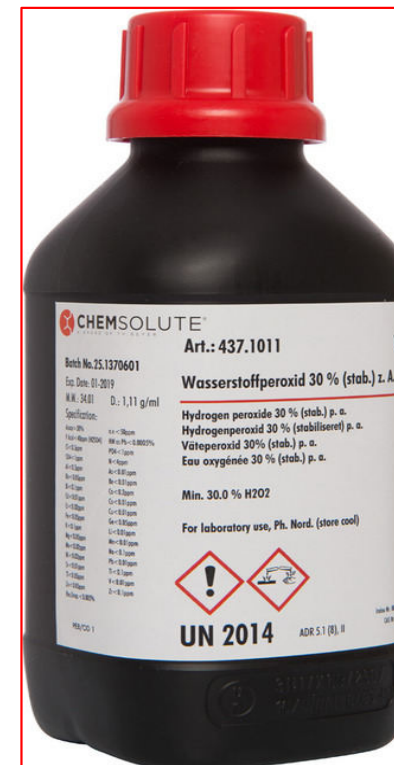
- **You MUST:** wear shoes, wear safety glasses and a laboratory coat, tie back long hair, report all accidents immediately
- **You MUST NOT:** drink, eat, smoke, work unsupervised, use volatile solvents near a flame



# CHE 101L



## Chemical Hazard Symbols





## Glass Apparatus & Instruments



Electronic balance



Distilled water



Dil.  $\text{H}_2\text{SO}_4$



Test tube



Weighing bottle



Potassium permanganate solution



250 ml Conical flask



Tripod stand



Spatula



Oxalic Acid



250 ml Beaker



250 ml Standard flask



White tile



Bunsen burner



50 ml Burette



Funnel



20 ml pipette



Burette stand

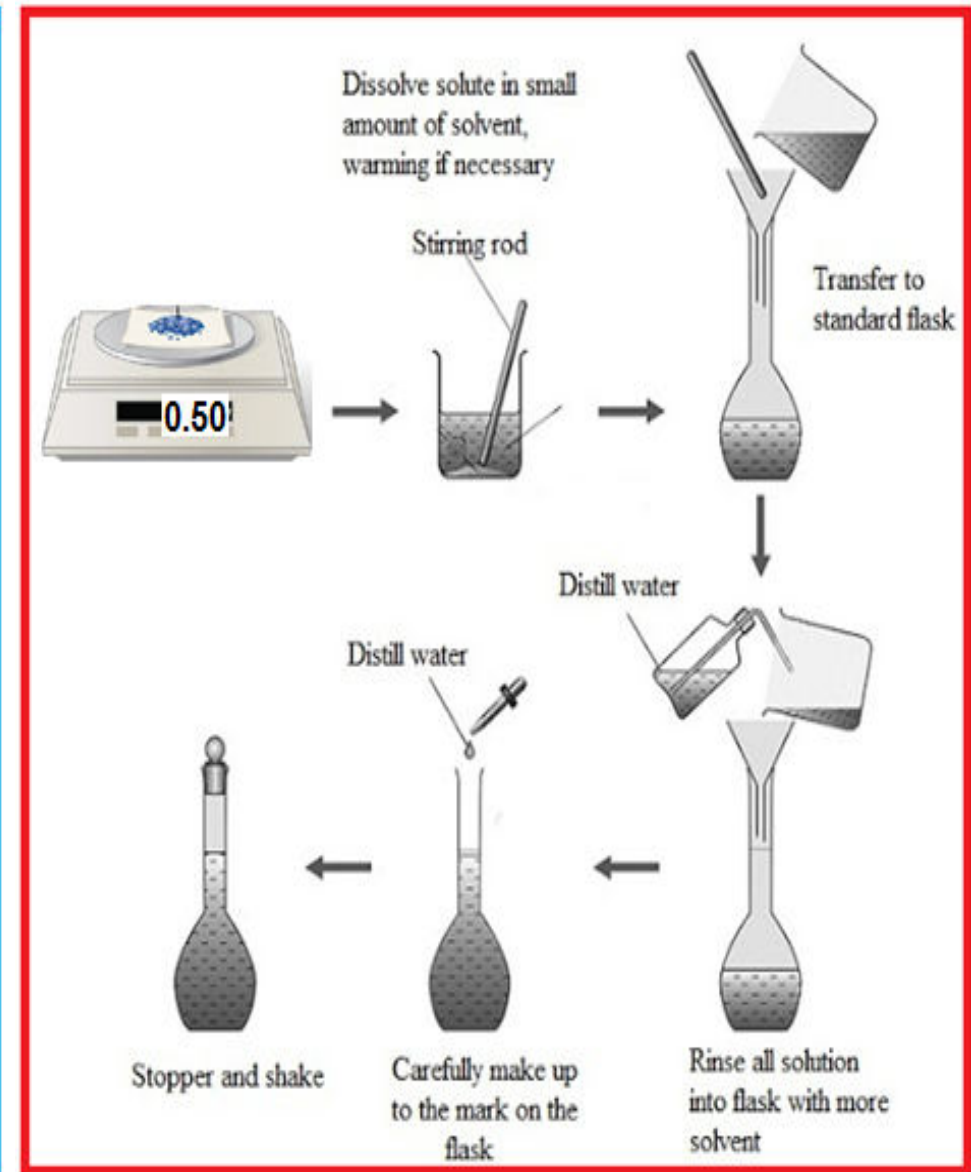
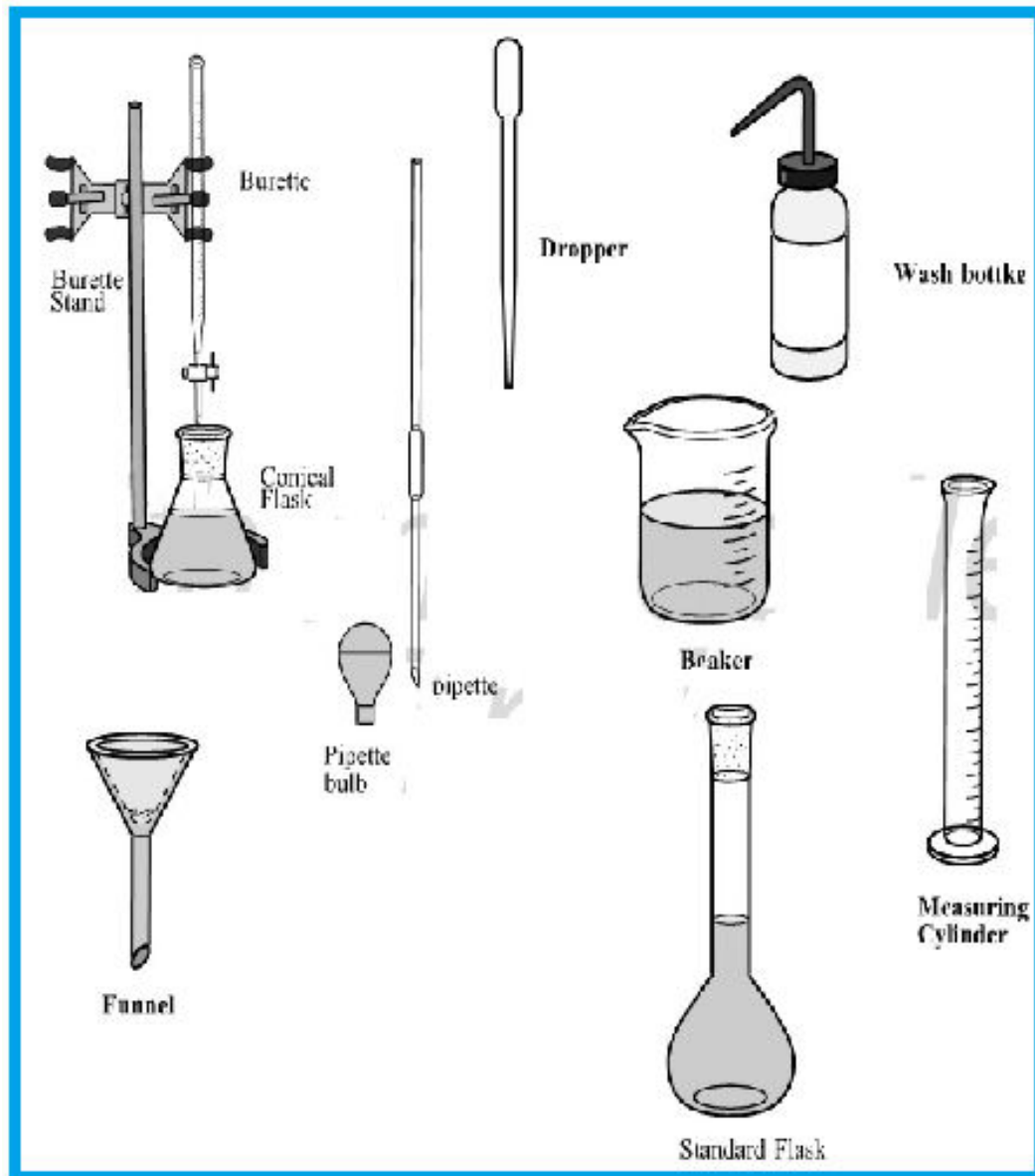
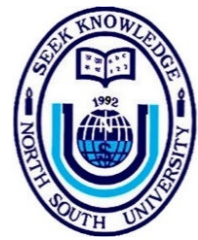


Glass rod



Wire gauze

# CHE 101L



# CHE 101L



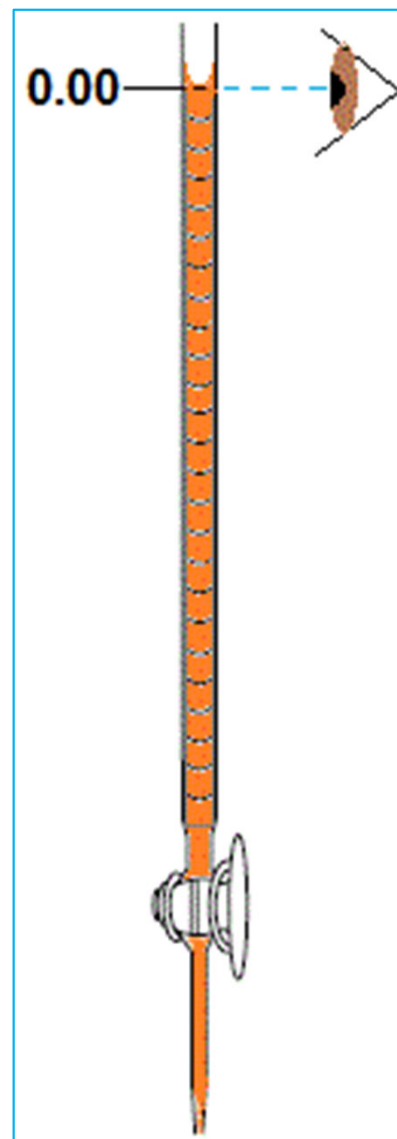
A drop of potassium manganate(VII) is added

The colour appears in the flask

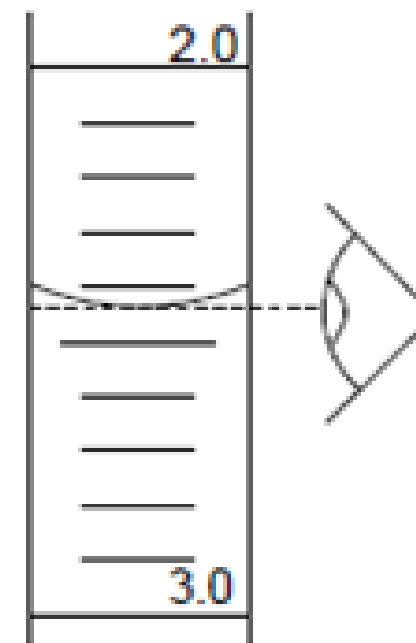
On swirling the potassium manganate(VII) reacts and the colour disappears

More potassium manganate(VII) is added

Until the colour remains after swirling



## Burette



2.45 cm<sup>3</sup>

**NOT**

3.55 cm<sup>3</sup>

Thank You for Listening  
Τυχαίοντες για την παρουσία



Any Question  
Τις ερωτήσεις