



THE UNIVERSITY
of EDINBURGH

Classical Molecular Dynamics Simulations

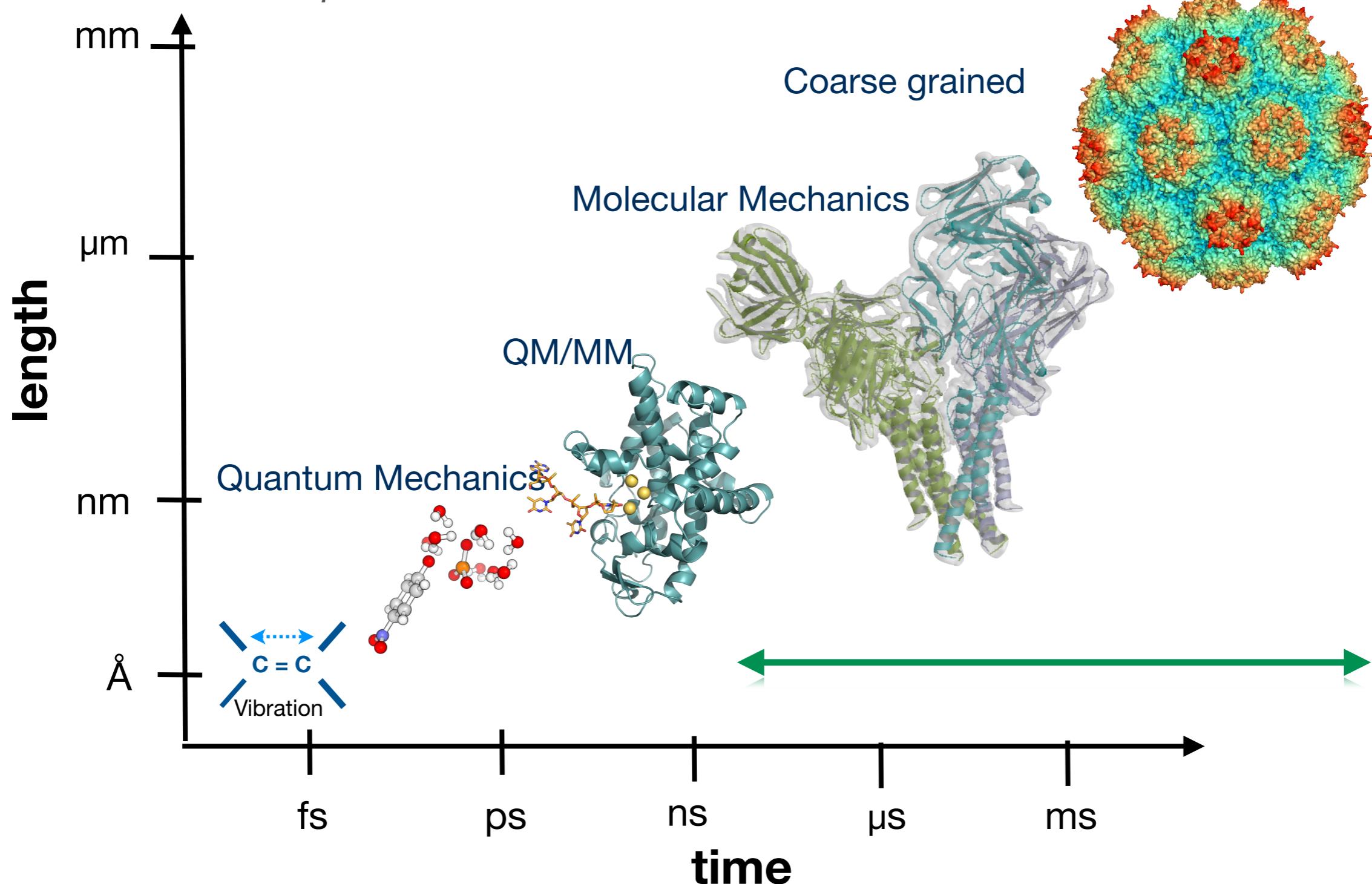
Antonia Mey – antonia.mey@ed.ac.uk

Slides adapted from Dr Fernanda Duarte

School of Chemistry
University of Edinburgh, UK

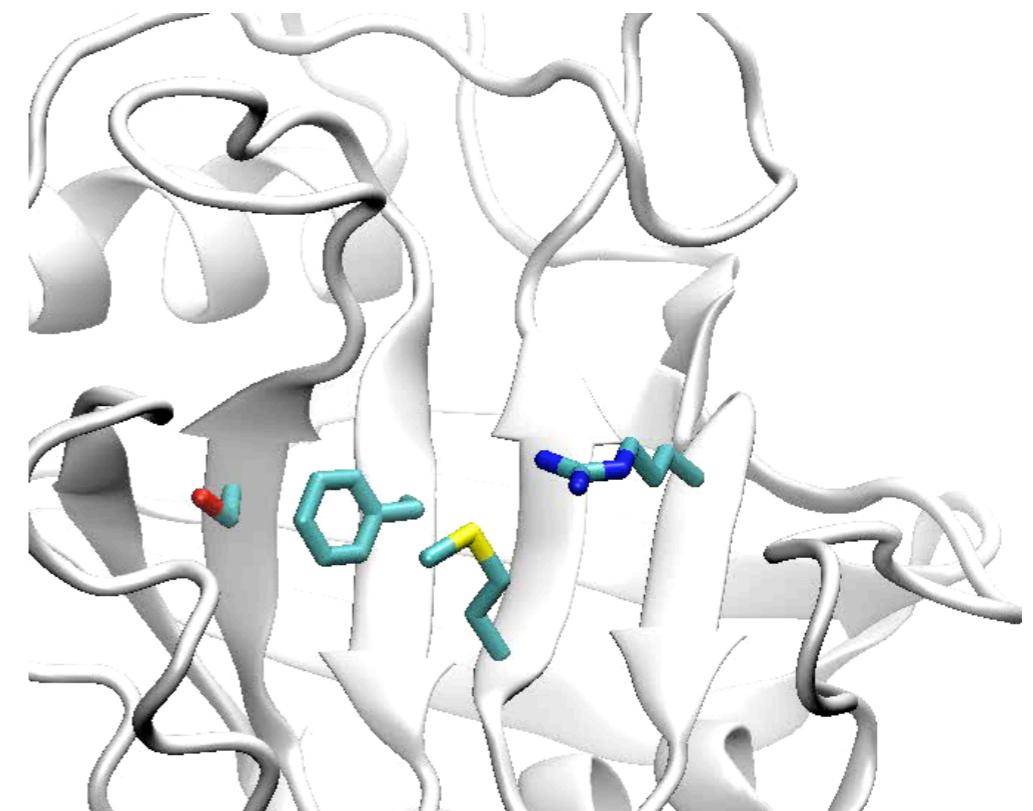
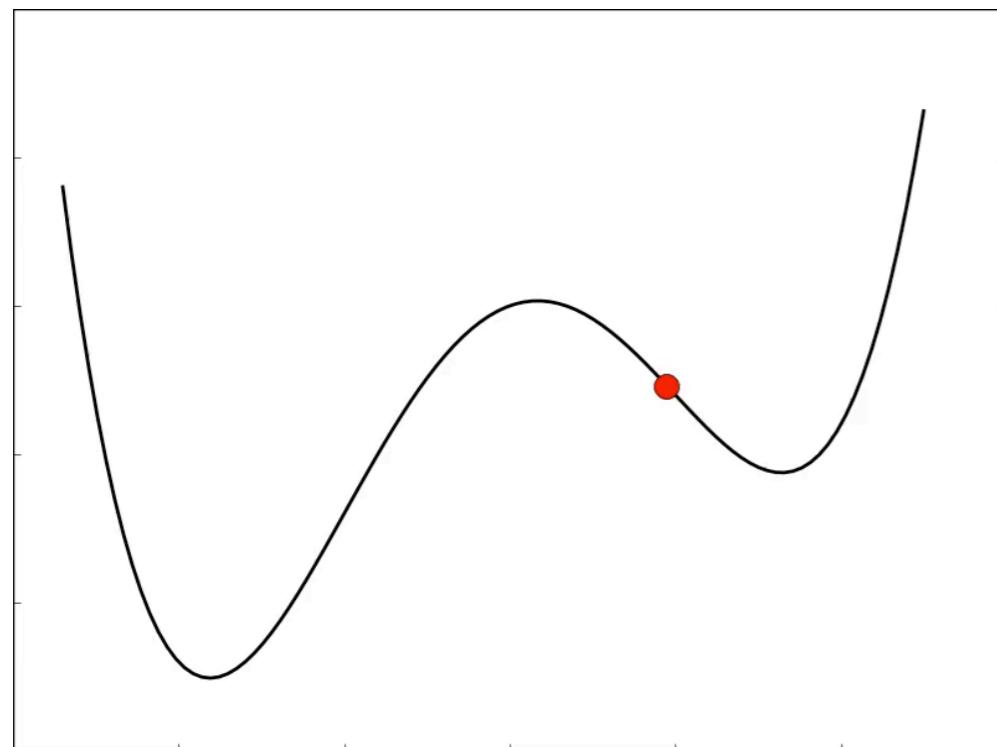
Computational Modelling

“Everything should be made as simple as possible, but not simpler.”



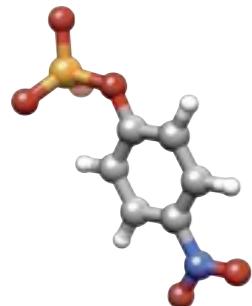
Overview

- What is - and why run - a Classical Simulation?
- Force fields/Atom Types/Solvent Models
- Molecular Dynamics–Framework and Steps Involved



Computational Modelling

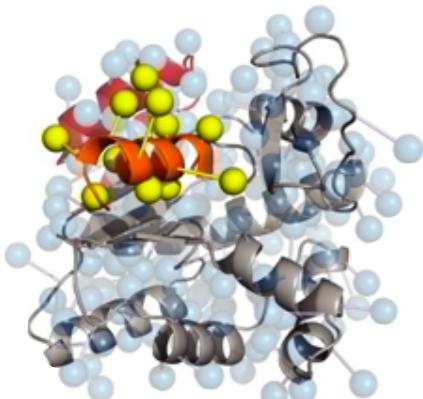
“Everything should be made as simple as possible, but not simpler.”



Quantum Mechanics

Atomistic
Electronic Structure
(Schrödinger equation)

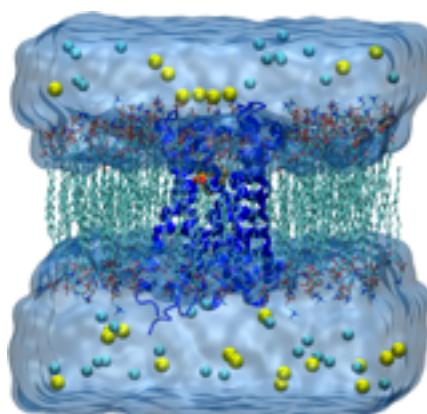
Accurate \Rightarrow Expensive
(DFT N^3 - CCSD(T) N^7)



Molecular Dynamics Monte Carlo

Atomistic
Empirical Forces
(Newton's equations/
Random move)

LESS accurate \Rightarrow Faster
($N \log N/N^2$)



Coarse grained

Not atomistic
Reduction of number of
degrees of freedom
Reaction and diffusion

Can be even faster

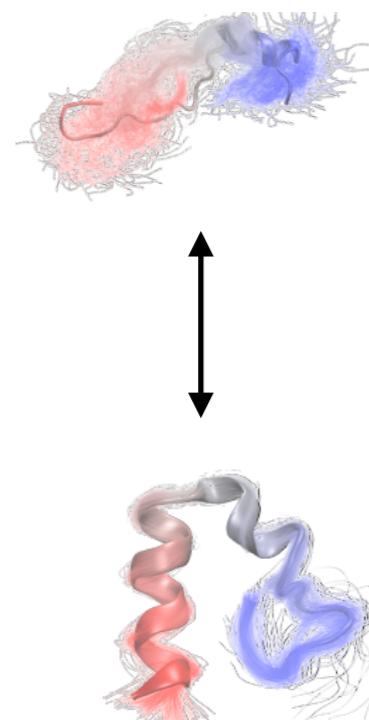
Computational Modelling

What is the scientific question we want to address?

How much detail do we need to describe the phenomenon?

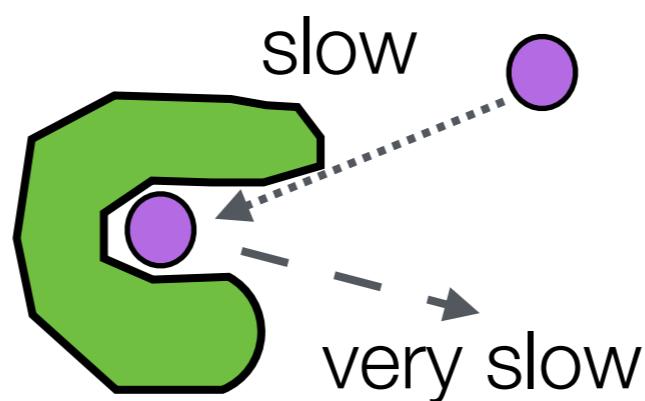
What are the computational resources?

Large or small
conformational
dynamics



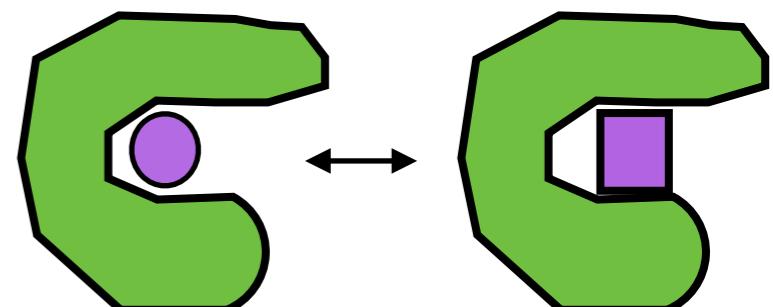
Standard MD

Binding
thermodynamics/
dynamics



Enhanced
sampling

Reaction dynamics



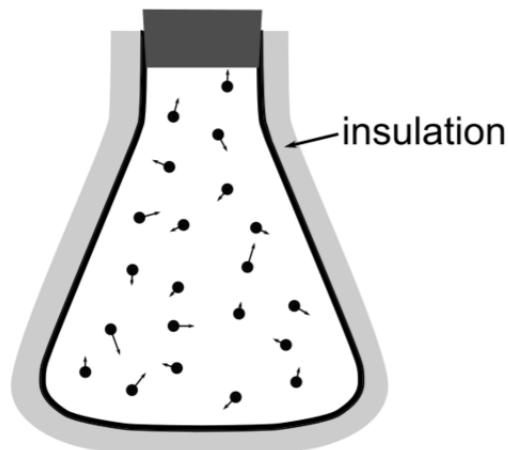
QM/MM

Computational Modelling

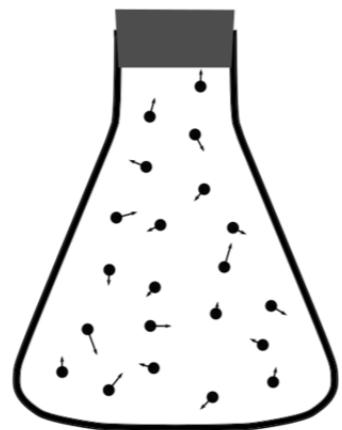
Experiments measure **ensemble averages**, of macroscopic properties



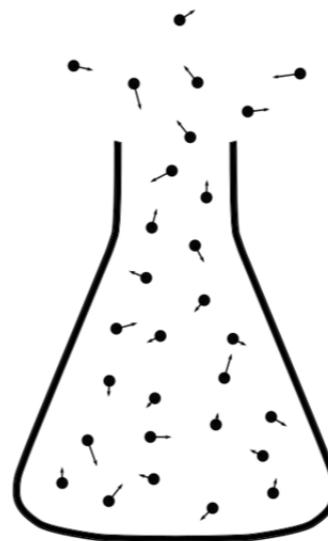
The knowledge of a **single** structure, even of a global energy minimum is **NOT sufficient**.



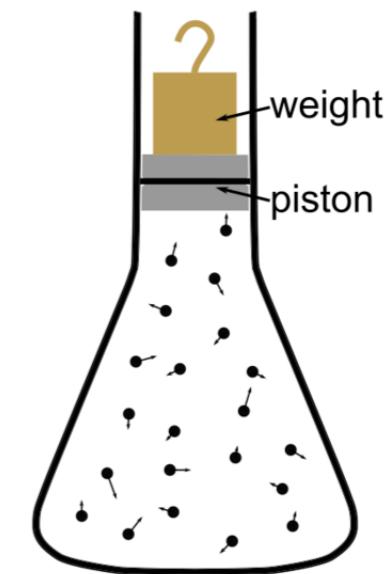
Microcanonical
(NVE)



Canonical
(NVT)



Grand canonical
(μ VT)



Isobaric-Isothermal
(NPT)

Computational Modelling

Statistical mechanics to the rescue

Macroscopic Observables

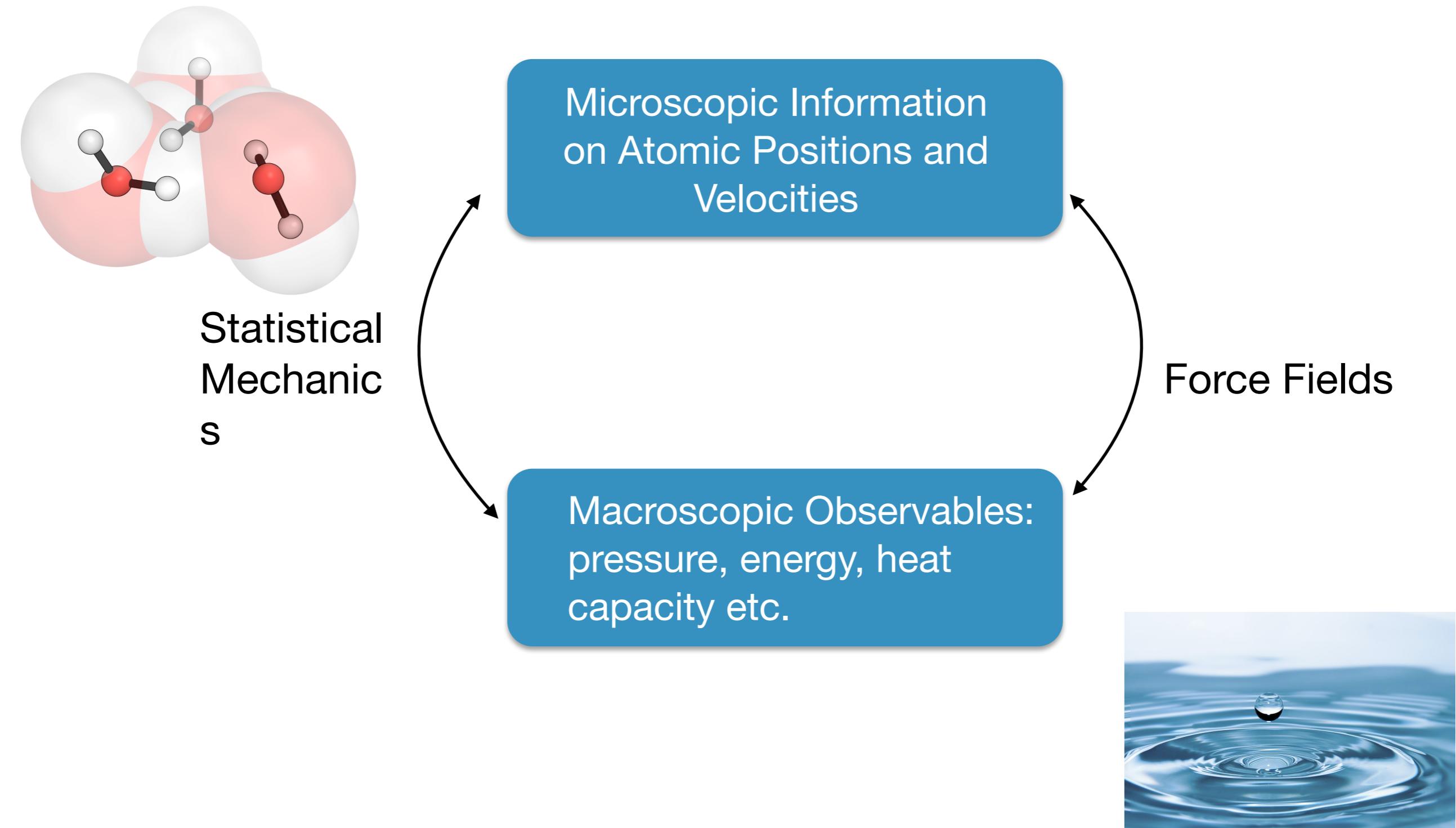
Equilibrium Properties

- Binding Constant of an Inhibitor
- Potential Energy of a System
- Structure of a Liquid

Non-Equilibrium Properties

- Viscosity of a Liquid
- Diffusion Processes in Membranes
- Reaction Kinetics

Computational Modelling



Generation of a Representative Equilibrium Ensemble

Metropolis Monte Carlo (MC)

Random Walk algorithm

No true analogue of time

Faster, MC compare energies no forces.

Thermodynamics and structural properties

Molecular Dynamics (MD):

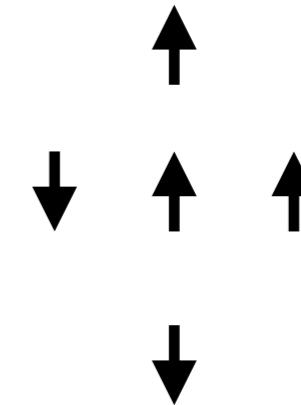
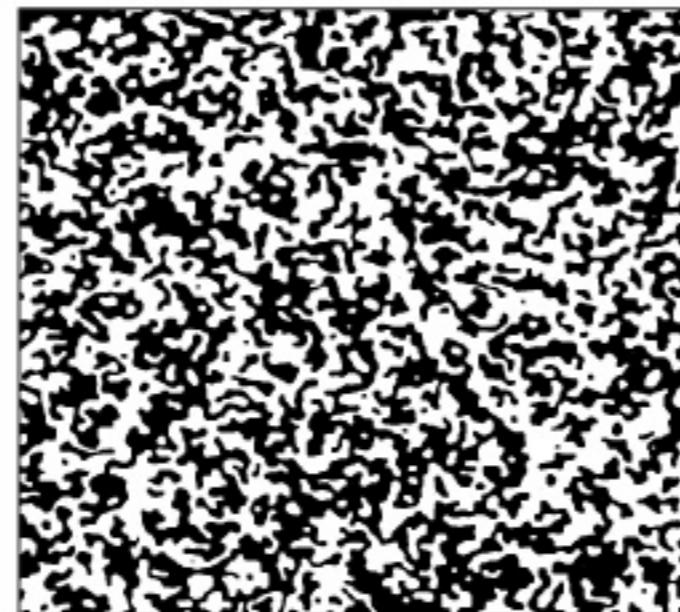
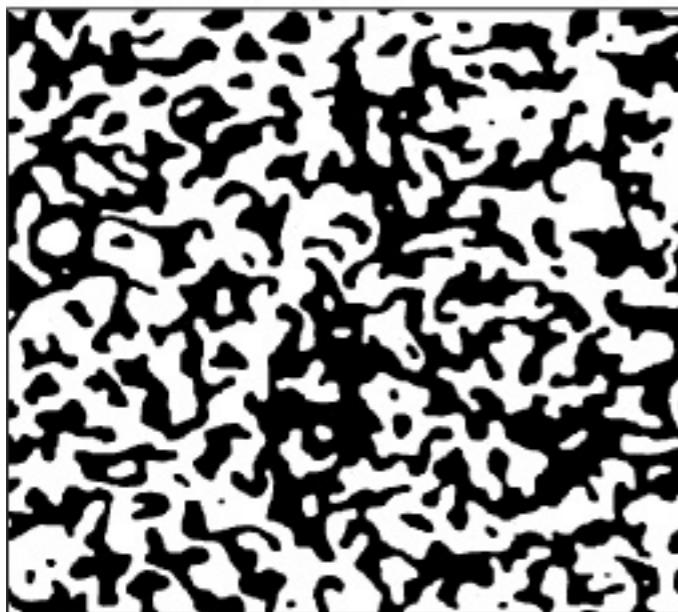
Newton's Equation

Simulation times from ps to ms

Require more computer time to achieve same level of convergence

Thermodynamics, structural and dynamic (equilibrium and non-equilibrium) properties

Monte Carlo – Ising model



What is this E?

$$P(\text{accept}) = \min(1, \exp(-\beta[E_f(x) - E_i(x)]))$$

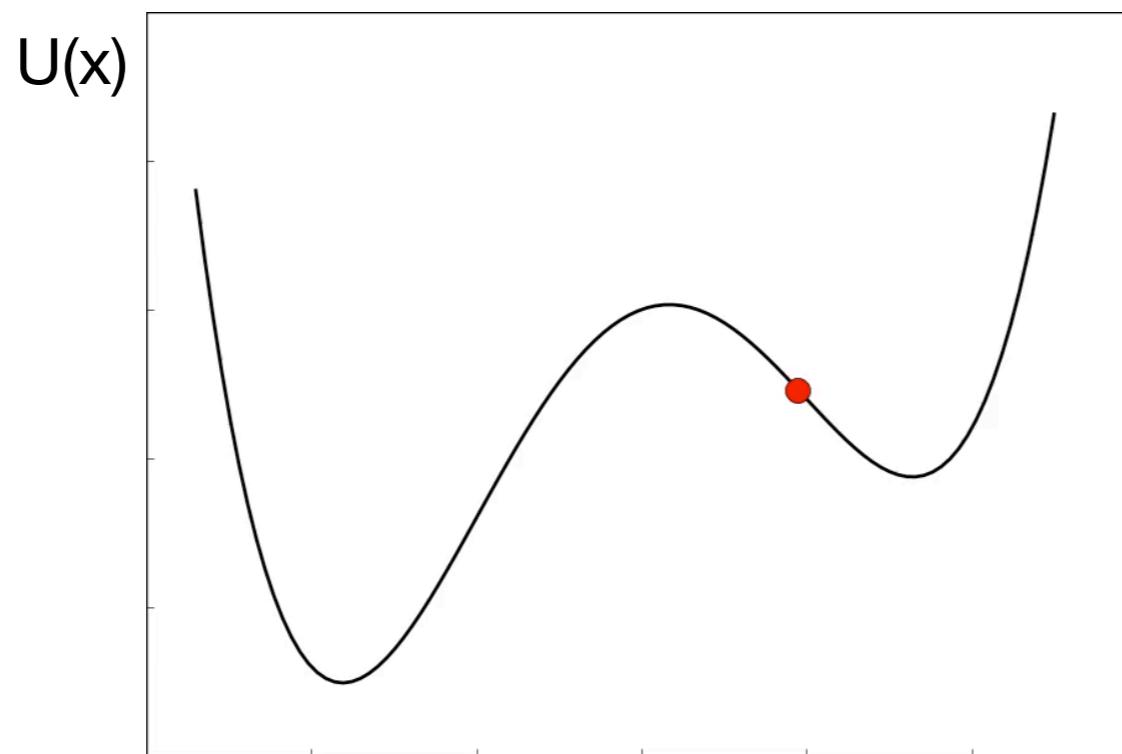
Langevin dynamics

Langevin Dynamics (LD)

Stochastic differential equation in which two force terms have been added to Newton's second law to approximate the effects of neglected degrees of freedom.

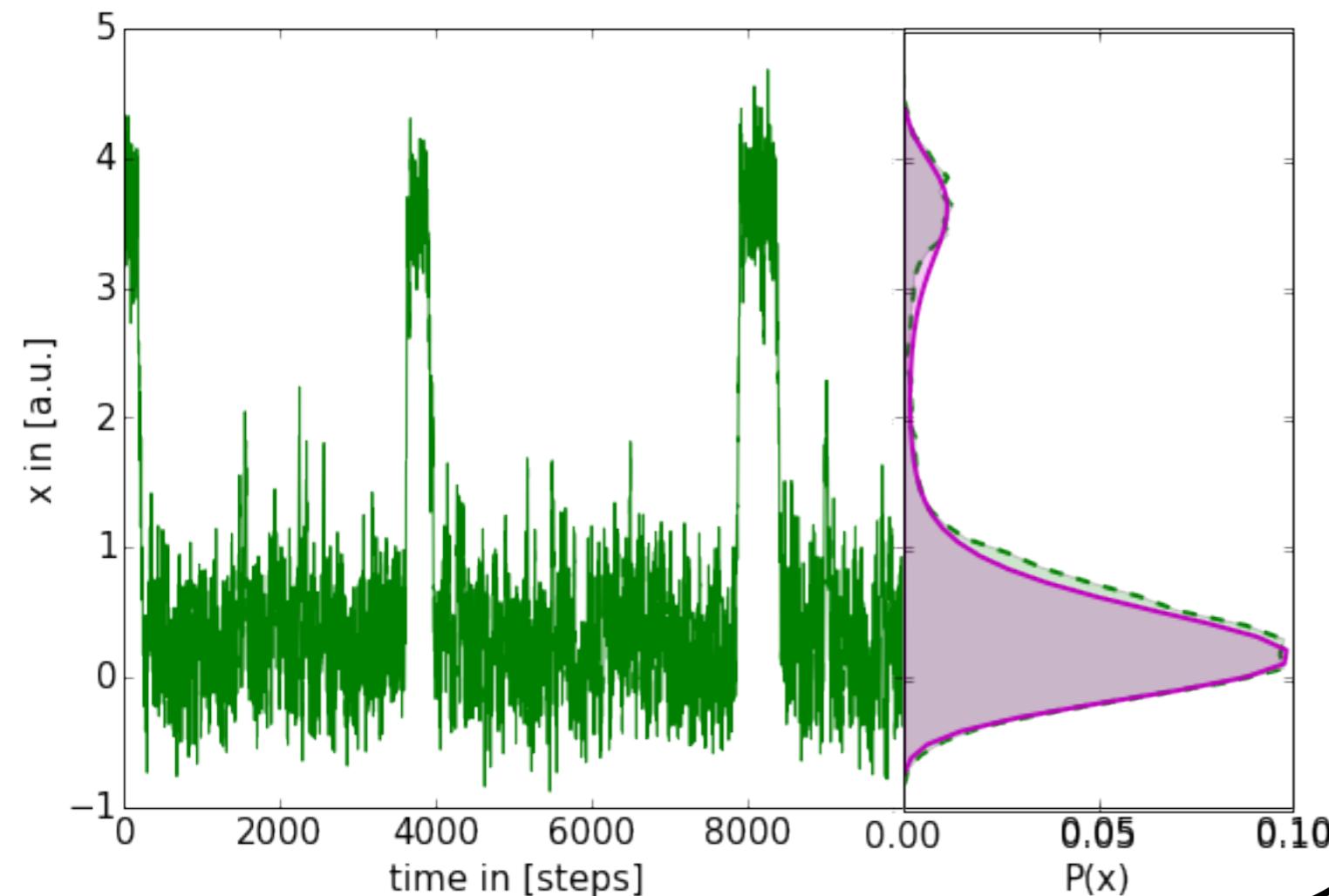
$$\vec{F}_i - \gamma_i \vec{v}_i + \vec{R}_i(t) = m_i \vec{a}_i$$

Frictional force Random force



A closer look at LD?

Boltzmann tells us, why it is easier to sample this potential at higher temperature



random force will sample both wells

Boltzmann factor

$$P(x) = \frac{\exp(-U(x)/kT)}{\int_{-\infty}^{+\infty} \exp(-U(x)/kT)}$$

Partition function: Z

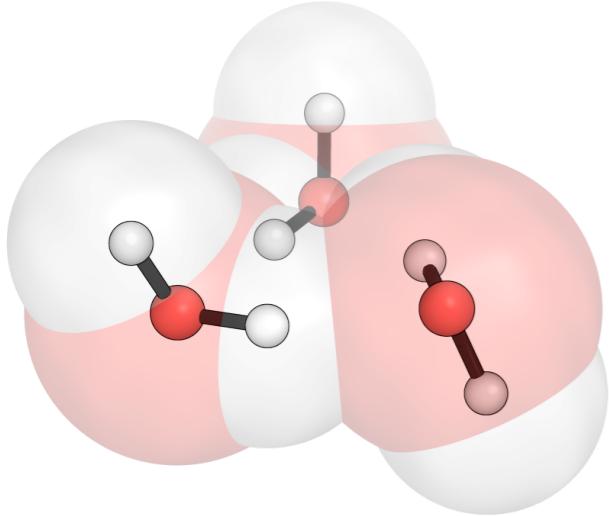
Free Energy:

$$F = -\ln Z$$

Possible to compute free energy differences between configurations, or thermodynamic states.

Molecular force fields

Example of a potential energy (force field) function:



- It contains all the parameters for this expression plus masses
- Parameters defined in terms of atom type (e.g. sp^3 , sp^2 and sp atoms).
- In most cases pair-additive, non-polarisable.

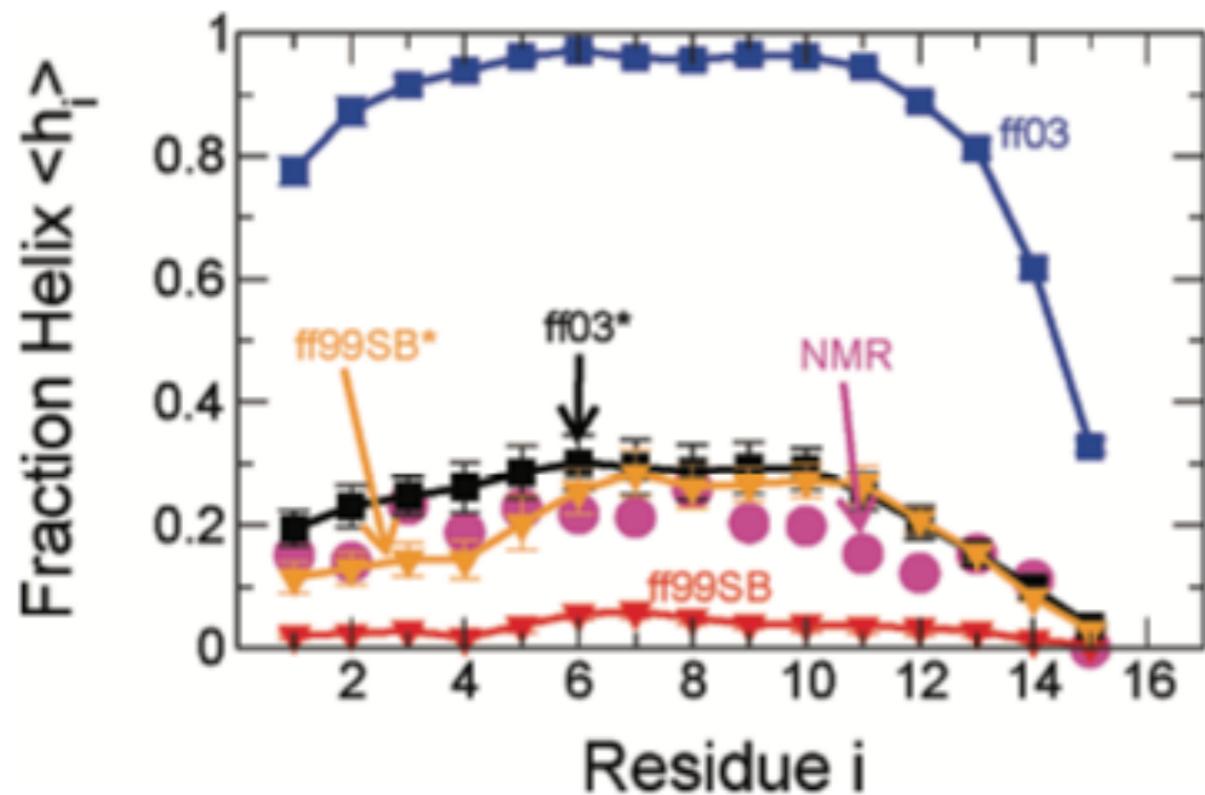
$$U = \sum_{\text{bonds}} K(\mathbf{r} - \mathbf{r}_{\text{eq}})^2 + \sum_{\text{angles}} K_\theta (\theta - \theta_{eq})^2 + \sum_{\text{dihedral}} \frac{V_n}{2} [1 + \cos(n\phi - \gamma)] + \sum_{\text{non-bonded}} \left[\frac{A_{ij}}{R_{ij}^{12}} - \frac{B_{ij}}{R_{ij}^6} + \frac{q_i q_j}{\epsilon R_{ij}} \right]$$

Molecular force fields – so many of them

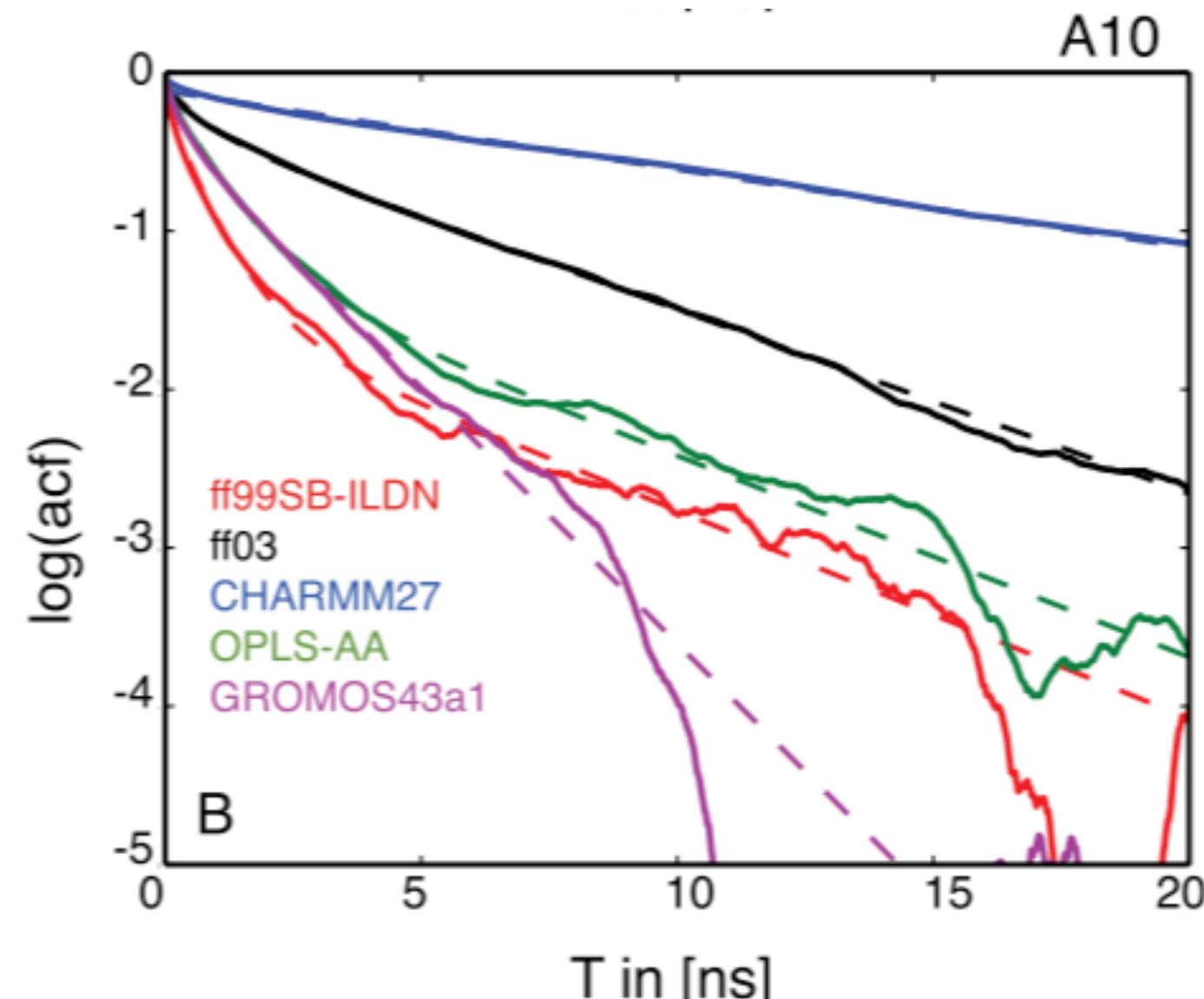
- They can be modified, different implementations depending on the software
- Pay attention to conversion rules!

Amber	DNA, proteins and lipids. Generalized Amber Force Field (GAFF) covers most organic space.
OPLS	Organic molecules in the liquid phase. Available as all-atom (AA) or united-atom (UA) form.
CHARMM	DNA, proteins, lipids, sugars.
GROMOS	General purpose. Organic and biochemical space.
Dreiding	General purpose. Includes some parameters for metals and main group elements.
MM2/MM3	General purpose. MM2 used for hydrocarbons. MM3 includes most of organic space plus some other main group and metal atom types.
Water	Rigid and flexible models available. TIP3P widely used in biological simulation.

Molecular force fields – which one to use?



Best et al. J. Phys. Chem. B 2009, **113**,
9004–9015



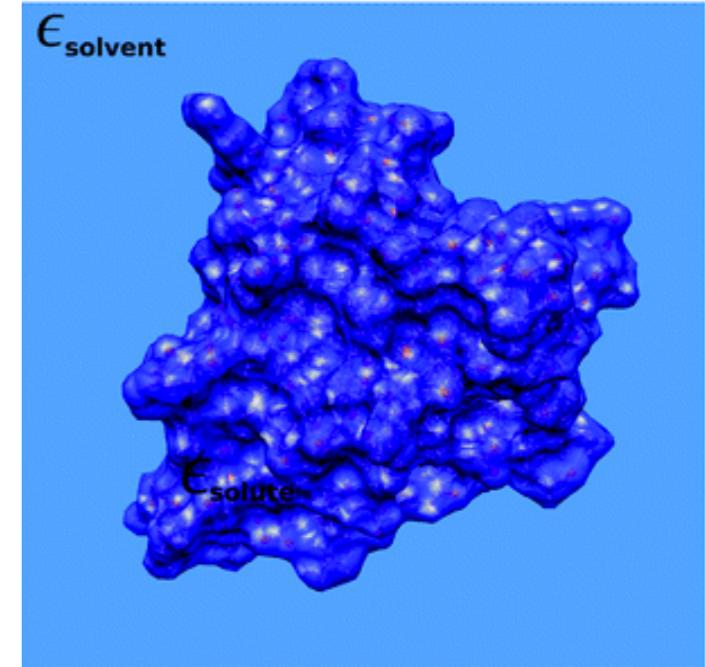
Vitalini, Mey et al, J. Chem. Phys. **142**,
084101 (2015)

What about water?

Implicit Solvent Model

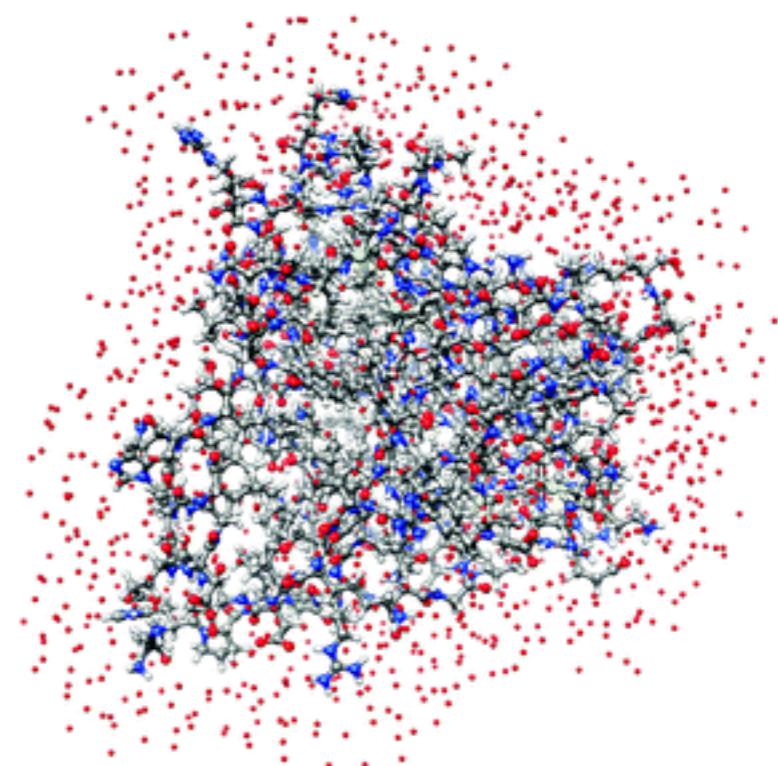
- Macromolecule interacts only with itself.
- Solvent effects accounted for in dielectric constant.

Generalised Born (GB): A more advanced implicit solvent model. GBSA is a GB model augmented with the hydrophobic solvent accessible surface area (SA) term.



Explicit Solvent Model

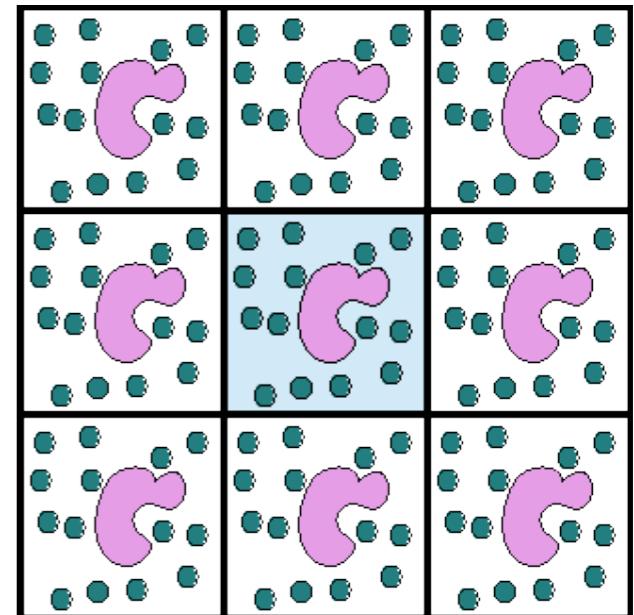
- Macromolecule is surrounded by solvent molecules (e.g. water).
- Specific non-bonded interactions are calculated.



Periodic boundary conditions

- Boundary Conditions are unnatural however, we cannot simulate infinite systems.

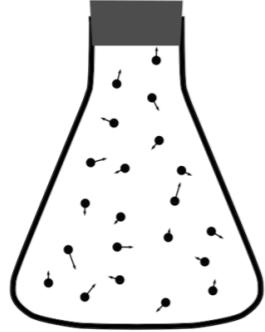
- We use PBC to avoid real phase boundaries.



- Minimum-Image Convention:** cut-off radius for the Lennard-Jones (Coulomb) interactions cannot exceed half the box size.

- Truly bad things can happen in charged systems (accumulation of charges at the cut-off boundary, wrong energies). **Particle mesh Ewald algorithm.**

Barostat/Thermostat



Canonical (NVT)

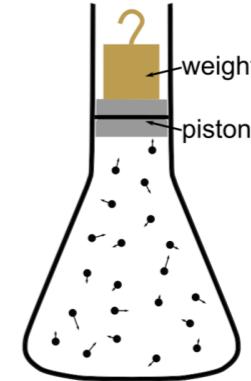
Requires a thermostat to make sure the temperature of the system stays the same.

Careful!

While Berendsen T and P control are simple to implement and use, they can steadily drive the system state far from equilibrium toward equilibrated state.

Good choice!

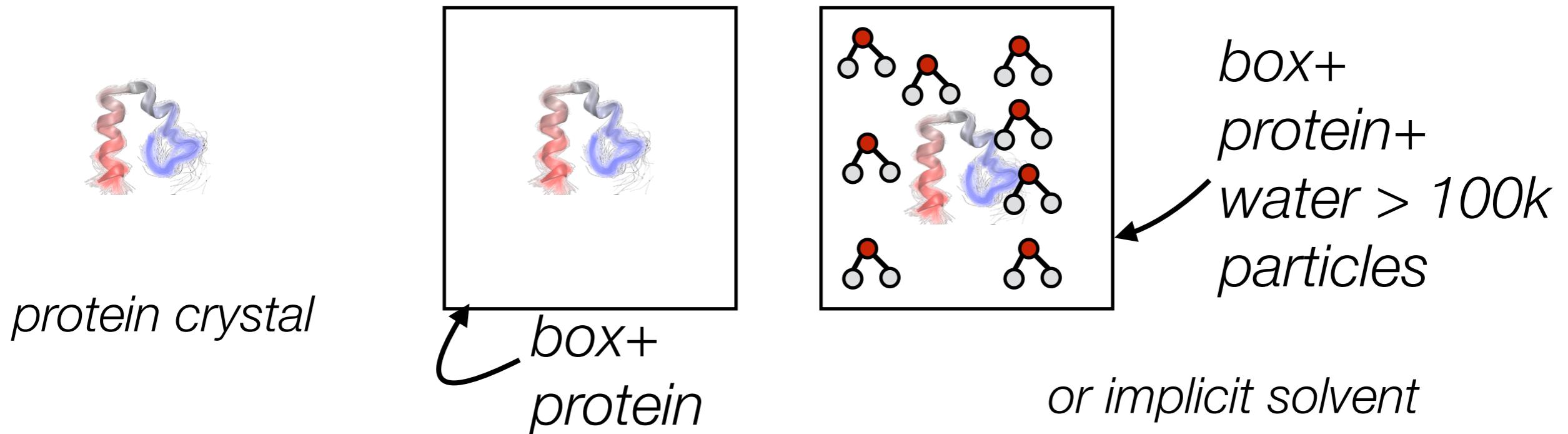
Nosé-Hoover thermostat with Rahman-Parrinello barostat



Isobaric-Isothermal (NPT)

Requires a thermostat and barostat to make sure the temperature and pressure of the system stays the same.

Putting everything together



Decide on: *Thermostat, Barostat and integrator*

Each atom has a position and velocity and is integrated according to Newtonian dynamics, with a timestep of **2 fs**.

– ***Reaching interesting timescales means millions of MD steps***

Timestep and integrator

Depends on the system that you are interested in

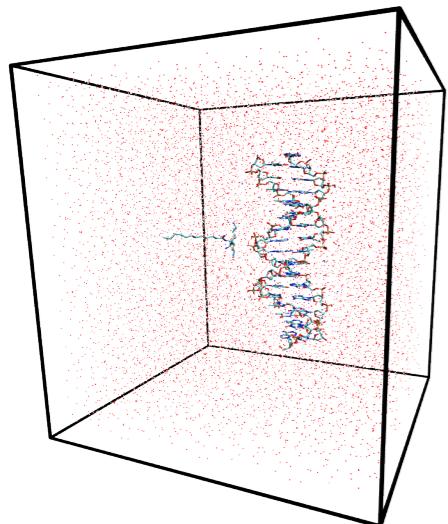
- Too small - only a small region of phase space will be covered
- Too large - instabilities may arise due to atom overlaps

For flexible molecules the time step should be $\sim 1/10$ th the time of the fastest motion (often C-H vibrations).

System	Types of motion present	Suggested time step (s)
Atoms	Translation	10^{-14}
Rigid molecules	Translation and rotation	5×10^{-15}
Flexible molecules, rigid bonds	Translations, rotation, torsion	2×10^{-15}
Flexible molecules, flexible bonds	Translation, rotation, torsion, vibration	10^{-15} or 5×10^{-16}

Simple example of MD workflow

System prep

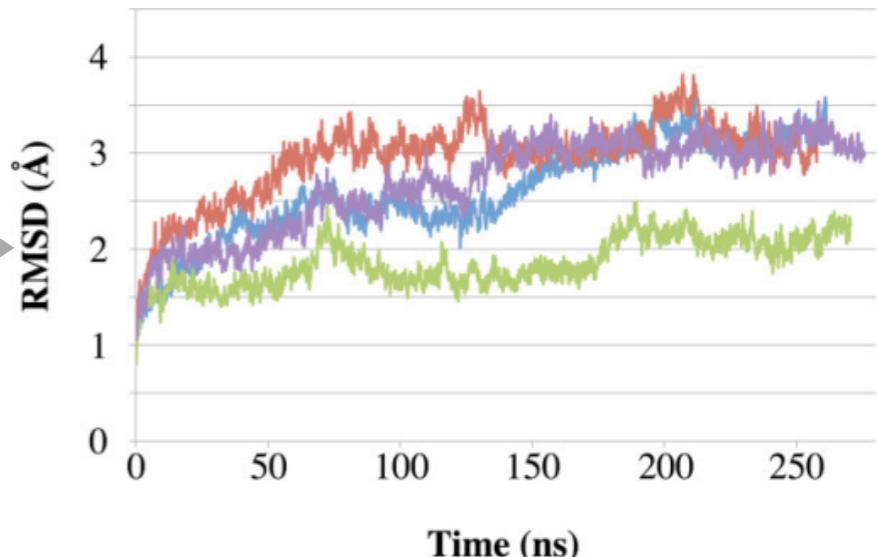


Simulation



The IBM Blue Gene/P supercomputer

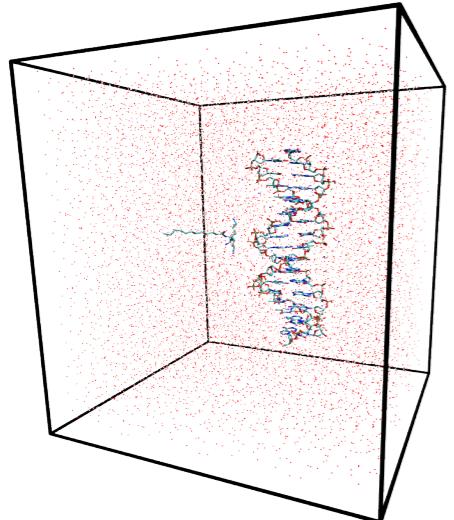
Analysis



Cloud computing

Simple example of MD workflow

System prep



Simulation

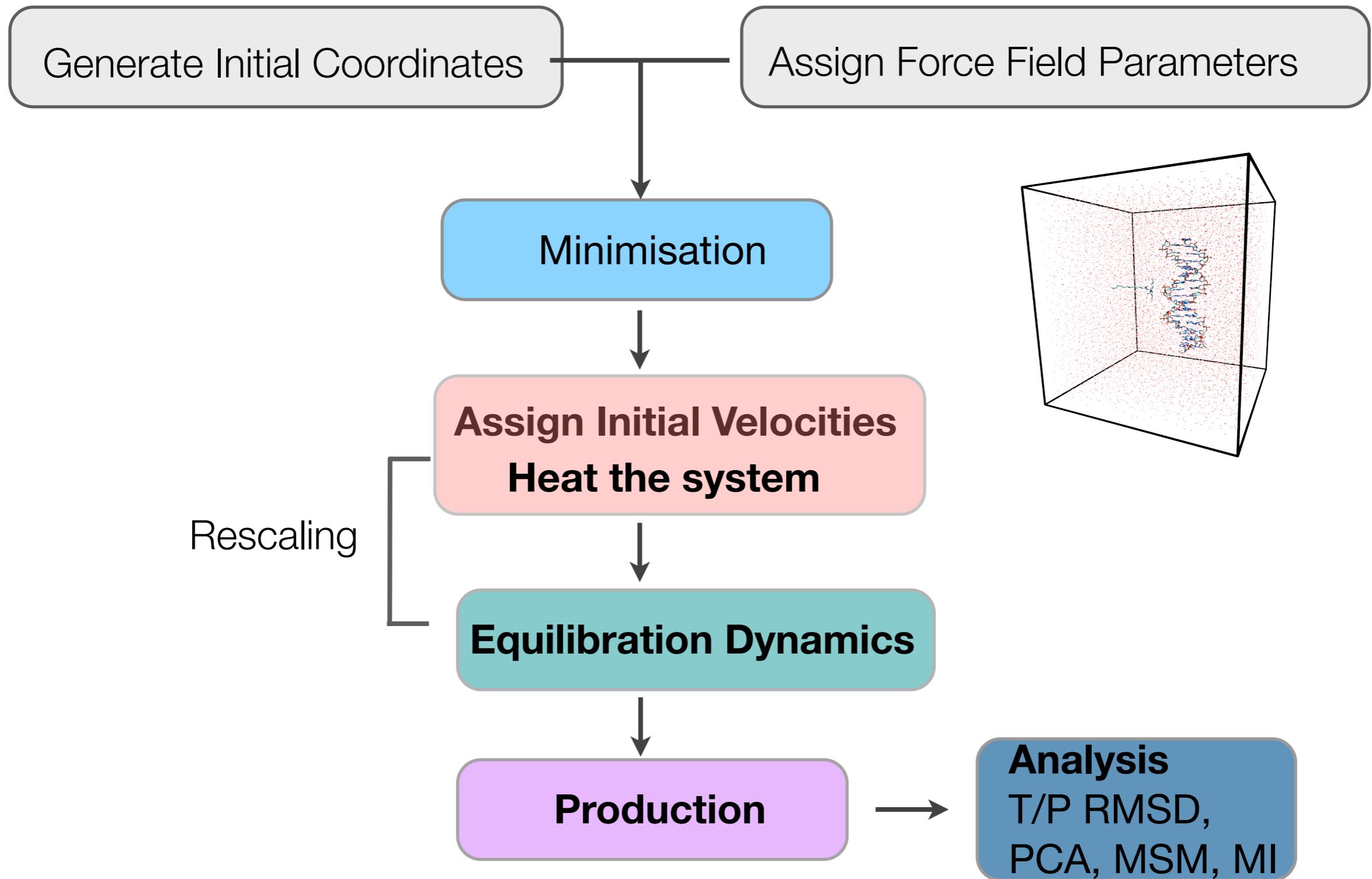


The IBM Blue Gene/P supercomputer

What will actually happen

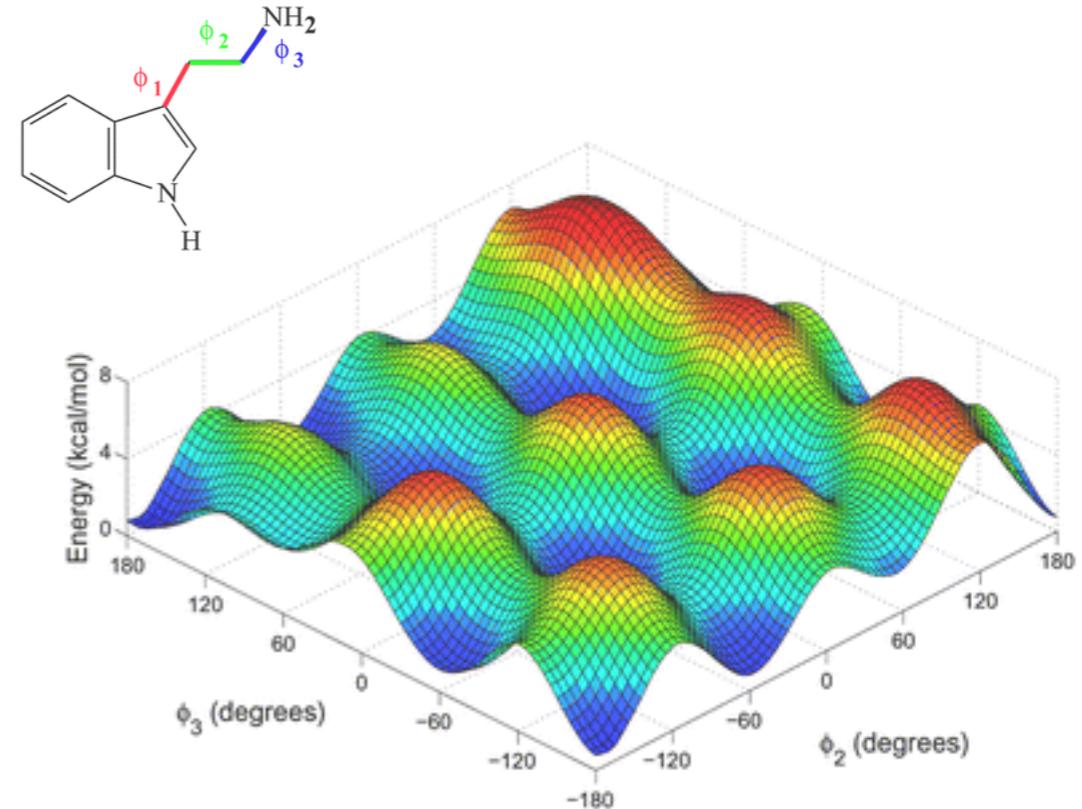


More detailed actual workflow



Minimisation

The potential energy function of a molecular system is a very complex landscape.



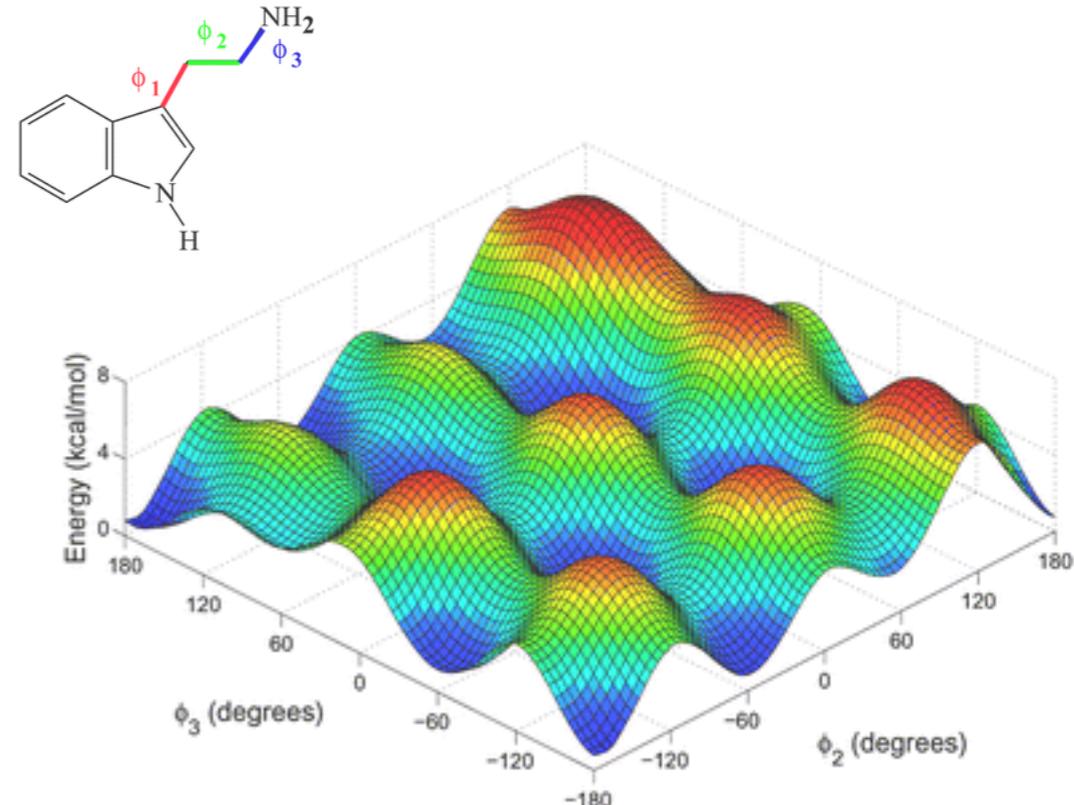
It has a deepest point, **global minimum**, and a very large number of **local minima**.

The **gradient** ($-\nabla V(r^N)$) is zero & the **Hessian matrix** ($H(r^N)$) has non-negative eigenvalues.

In between local minima there are saddle points: $H(r^N)$ has **only one** negative eigenvalue.

Minimisation

The potential energy function of a molecular system is a very complex landscape.



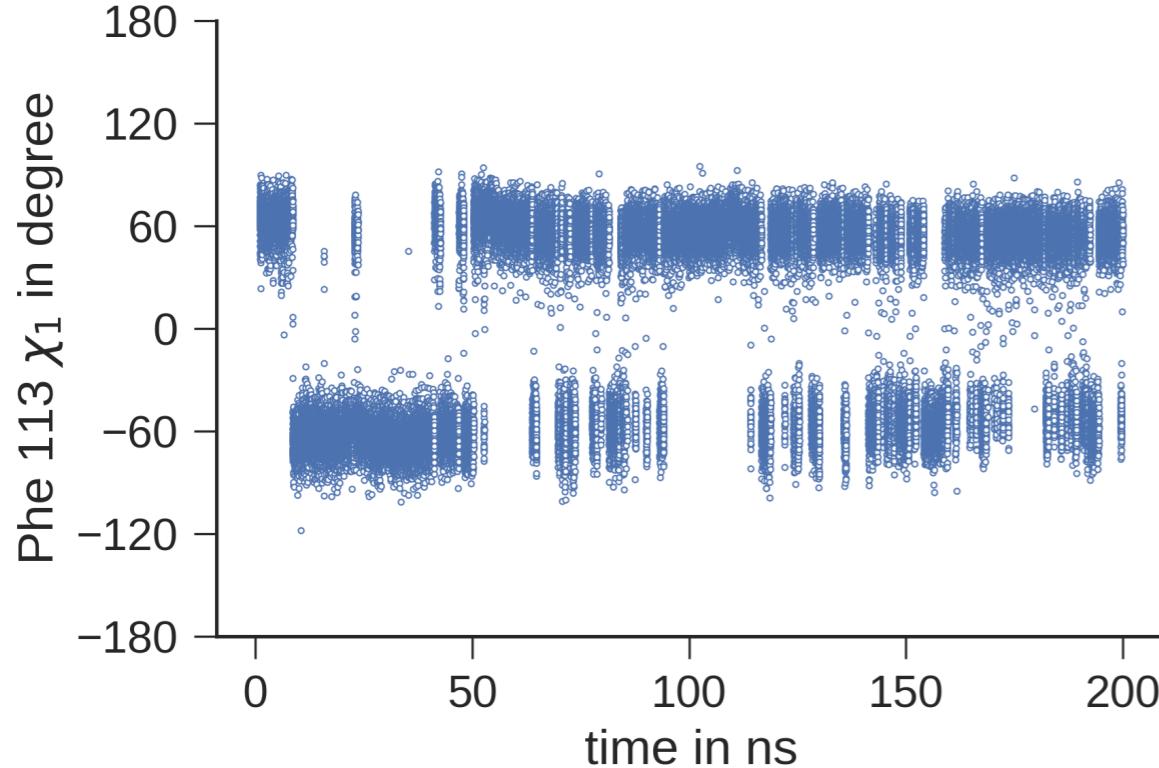
NO minimisation method can guarantee the determination of the global minimum

Finding this nearest local minimum is all these methods can do for you!

Methods

- Ones that require only function evaluations
- Ones that use derivative information. **Conjugate Gradient/ Steepest Descent**
- Ones that use second derivative information as well (QM approaches)

Analysis

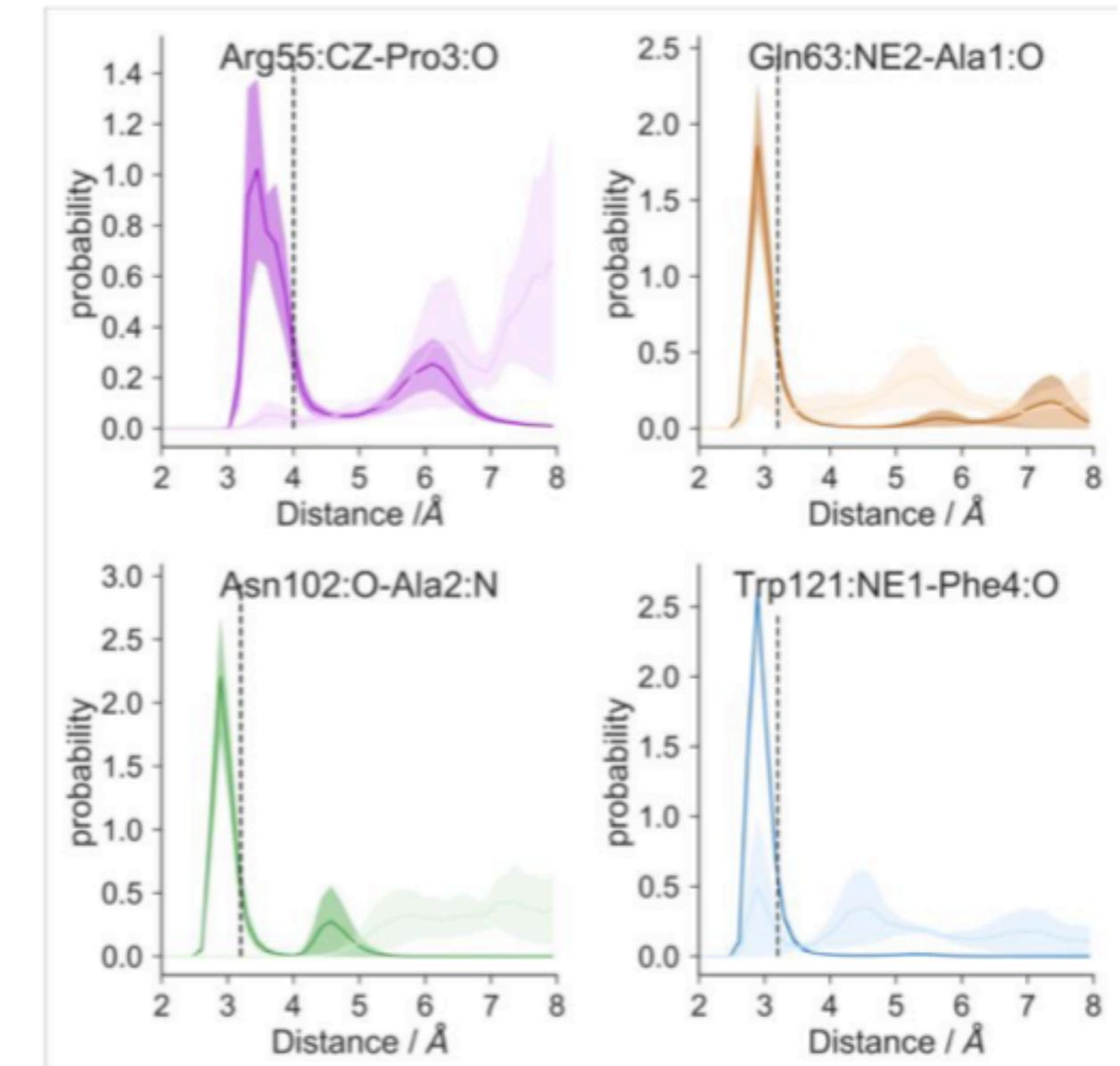


Enhanced analysis techniques:

PCA

Markov State models

Mutual information

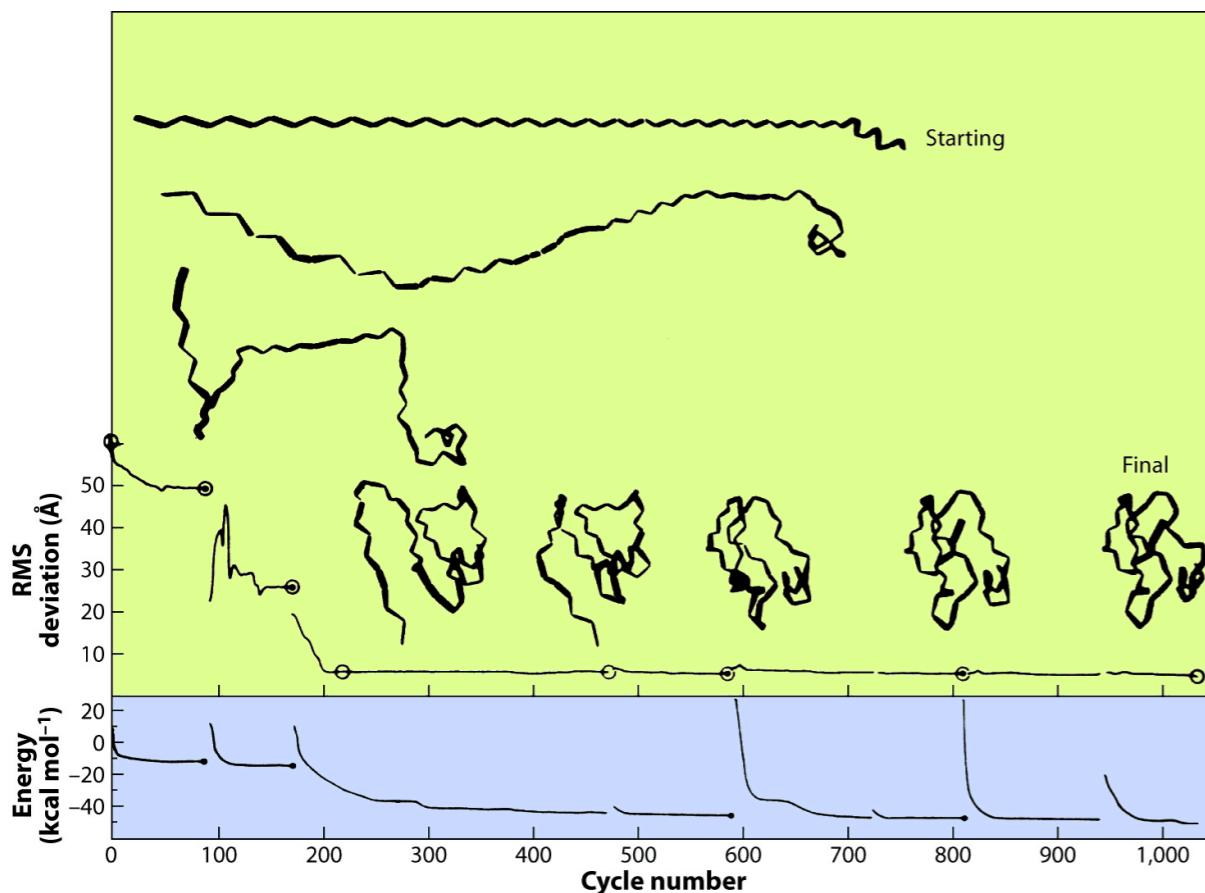


- > Also enhanced simulation techniques

Applications

Computer Simulations of Protein Folding

First MD simulation of a biological folding process (BPTI) (Levitt & Warshel, 1975)



The Nobel Prize in Chemistry 2013



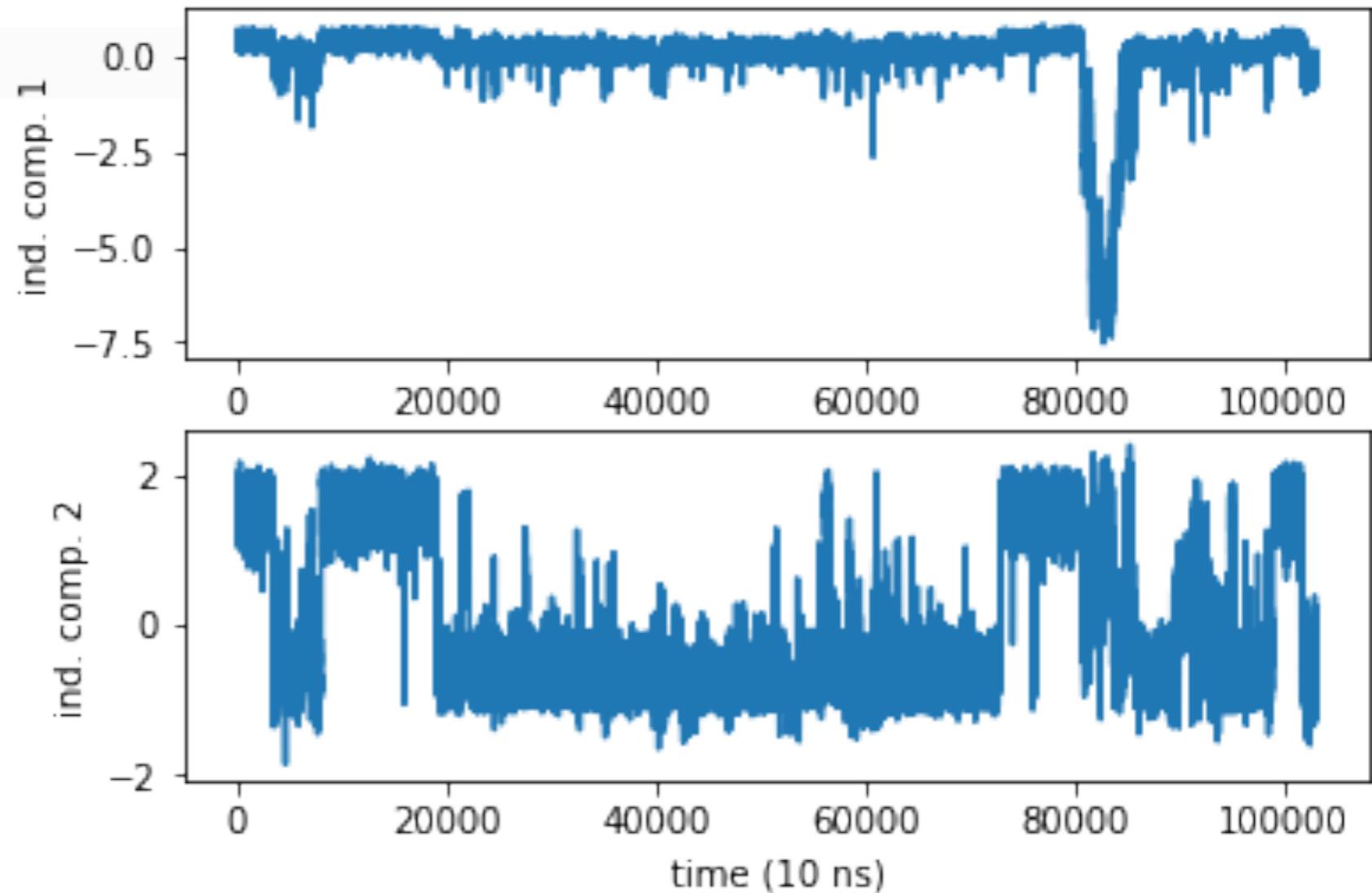
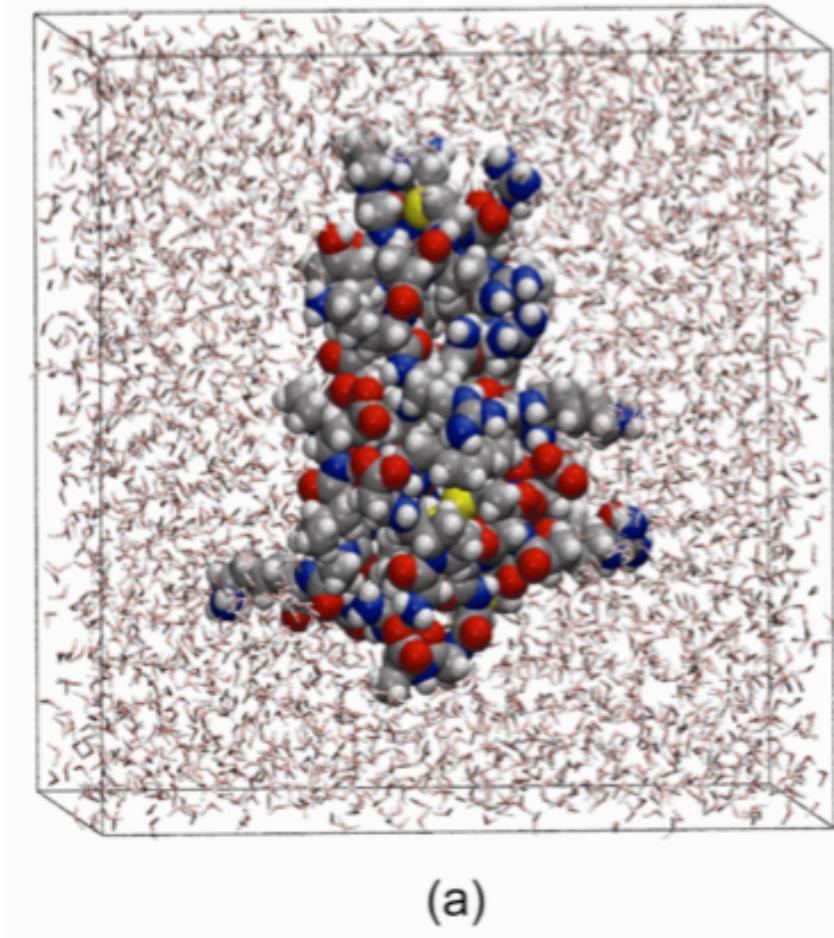
Martin Karplus Michael Levitt Arieh Warshel

https://youtu.be/_hMa6G0ZoPQ?t=40

Applications

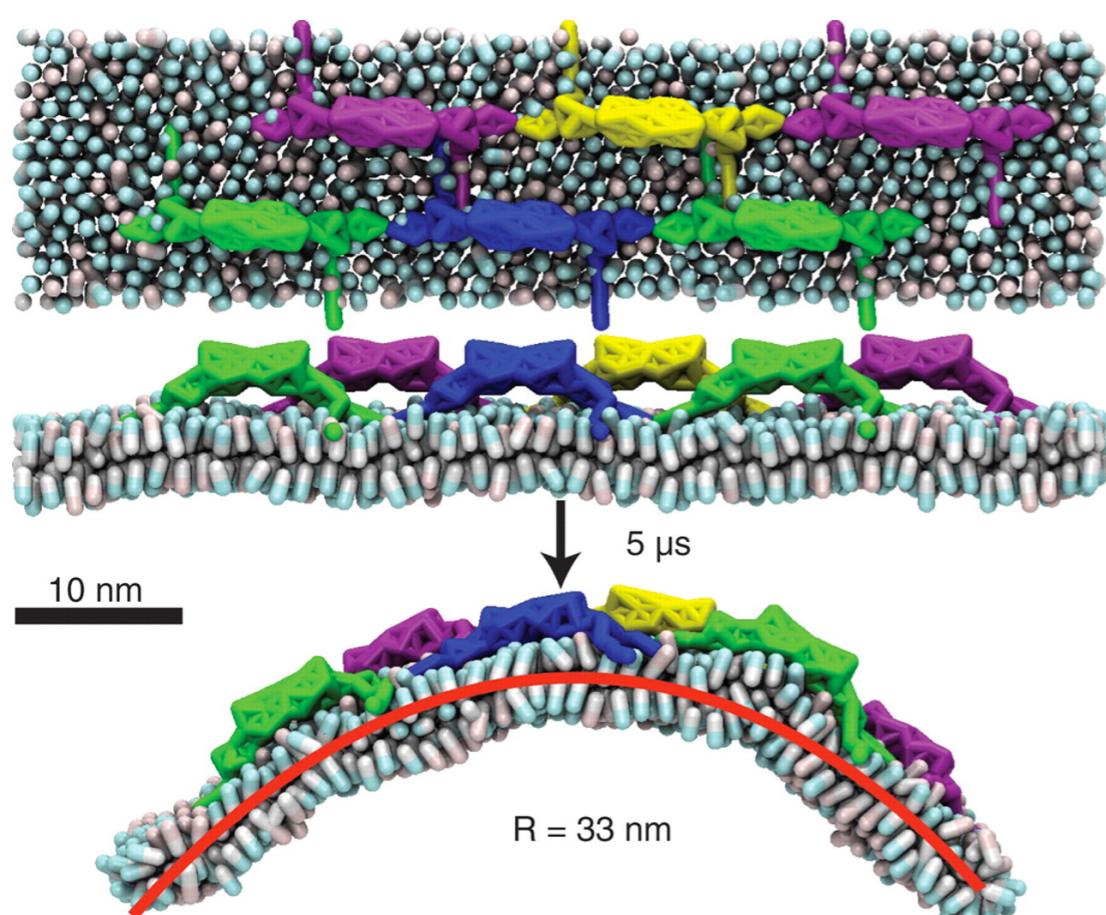
Computer Simulations of Protein Folding

BPTI revisited – 1st millisecond simulation using Anton



Applications

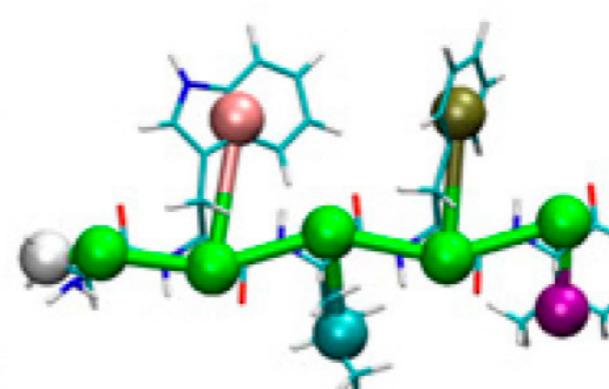
Four-Scale Description of Membrane Sculpting by BAR Domains



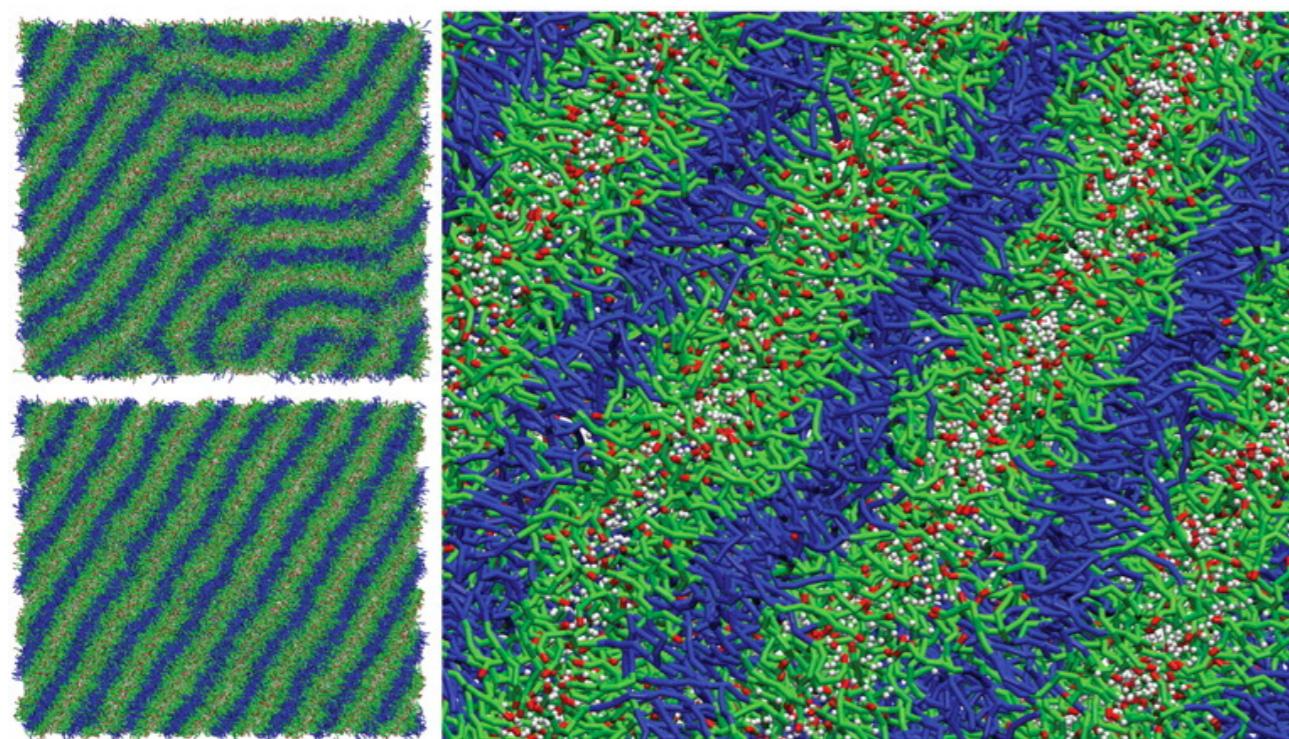
Computation was done using four levels of description

- All-atom MD,
- Residue-based CG (RBCG) MD ~ 10 atoms per CG bead
- Shape-based CG (SBCG) MD ~ 150 atoms per CG bead
- Continuum elastic membrane model.

NAMD Software



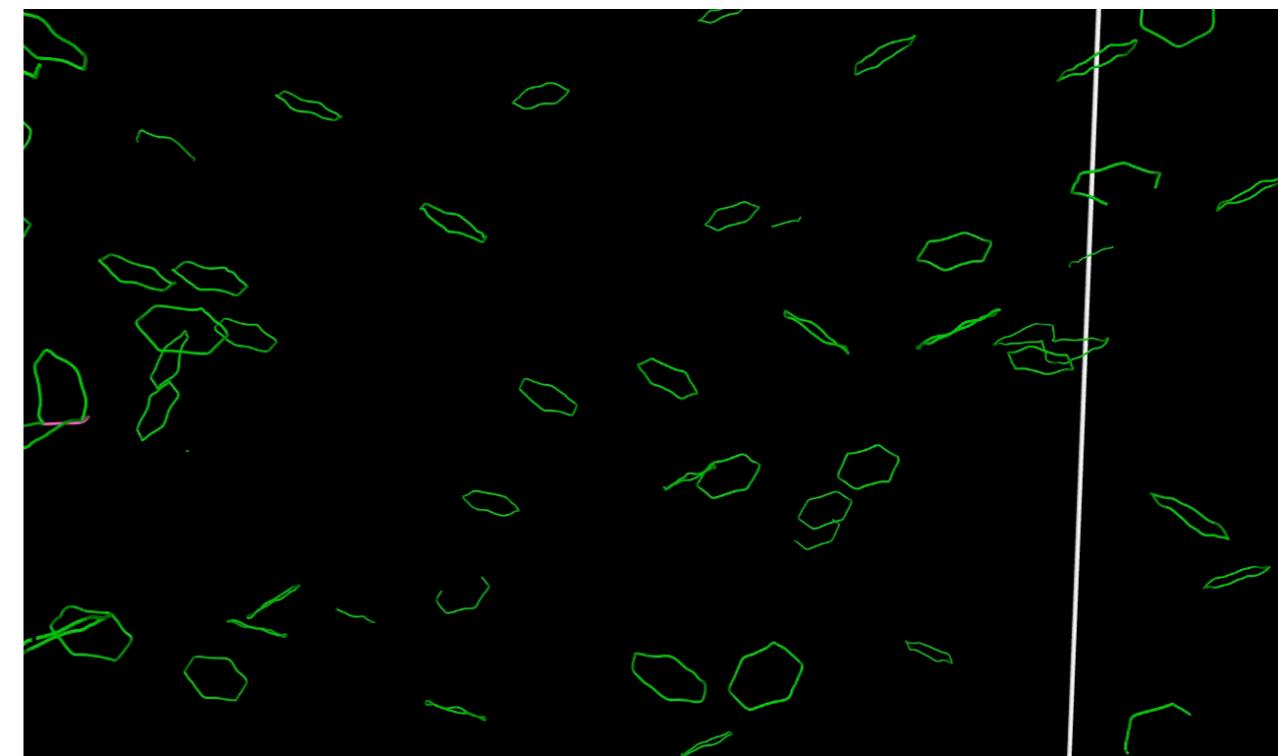
Applications



**Large-Scale Molecular Dynamics
Simulations of Self-Assembling
Systems**

Science, 2008, 321, 798-800

Software: LAMMPS
snapshots of a solution with 80% surfactant
(C12E6) and 20% water
800,000 Coarse grained particles.



**Probing the limits of metal plasticity
with molecular dynamics simulations**
Nature, 2017, 550, 492

Software: LAMMPS
Fragments of tantalum single crystals embedded into an
infinite crystal under PBC (about 268million atoms)
The crystal was compressed either at a constant or variable
rate

Useful Reading

- GROMACS manual www.gromacs.org
- Gromacs Tutorial: <http://www.bevanlab.biochem.vt.edu/Pages/Personal/justin/gmx-tutorials/>
- OpenMM Tutorial: <http://openmm.org/tutorials/index.html>
- LAMMPS Tutorials: <http://lammps.sandia.gov/tutorials.html>
- Charmm Tutorial - http://www.ch.embnet.org/MD_tutorial/
- Allen, M.P., Tildesley, D.J. “Computer Simulation of Liquids”, 1987.
- Frenkel, D., Smit B. “Understanding Molecular Simulation”, Academic Press, 1996.
- Keffer, D.J., “A Working Person’s Guide to Molecular Dynamics”, http://utkstair.org/clausius/docs/che548/pdf/md_sim.pdf.

Useful Tools

Classical simulation codes

- Gromacs www.gromacs.org
- OpenMM <http://openmm.org/>
- Amber – <http://ambermd.org/>
- NAMD <http://www.ks.uiuc.edu/Research/namd/>
- CHARMM - <http://www.charmm.org>
- DL_POLY – <http://www.stfc.ac.uk/cse/25526.aspx>
- LAMMPS – <http://lammps.sandia.gov>
- BioSimSpace – biosimspace.org (Workflow design tool)

Visualisation software

- VMD – <http://www.ks.uiuc.edu/Research/vmd/>
- Pymol - <http://www.pymol.org>
- Chimera <https://www.cgl.ucsf.edu/chimera/>