Notes on the function gsw_alpha_wrt_pt_t_exact(SA,t,p)

This function, **gsw_alpha_wrt_pt_t_exact**(SA,t,p), evaluates the thermal expansion coefficient with respect to potential temperature θ , α^{θ} ,

$$\alpha^{\theta} = -\frac{1}{\rho} \frac{\partial \rho}{\partial \theta} \Big|_{S_{A}, p} = \frac{1}{\nu} \frac{\partial \nu}{\partial \theta} \Big|_{S_{A}, p}, \tag{1}$$

with the input temperature being *in situ* temperature t. The potential temperature θ being considered here has its reference pressure being 0 dbar. This function uses the full TEOS-10 Gibbs function $g(S_A,t,p)$ of IOC et~al. (2010), being the sum of the IAPWS-09 and IAPWS-08 Gibbs functions. This function evaluates α^{θ} from the expression derived in appendix A.15 of the TEOS-10 Manual (IOC et~al., 2010), namely Eqn. (A.15.2), repeated here,

$$\alpha^{\theta} = \frac{1}{v} \frac{\partial v}{\partial \theta} \bigg|_{S_{A,p}} = \frac{1}{v} \frac{\partial v}{\partial T} \bigg|_{S_{A,p}} \left(\frac{\partial \theta}{\partial T} \bigg|_{S_{A,p}} \right)^{-1} = \frac{g_{TP}(S_A, t, p)}{g_P(S_A, t, p)} \frac{g_{TT}(S_A, \theta, p_r)}{g_{TT}(S_A, t, p)}.$$
(A.15.2)

The first step in the code is to evaluate the potential temperature θ with respect to $p_r = 0$ dbar. Then **gsw_alpha_wrt_pt_t_exact**(SA,t,p) evaluates α^{θ} via the four partial derivatives of the Gibbs function in the right-hand most part of Eqn. (A.15.2).

<u>References</u>

IAPWS, 2008: Release on the IAPWS Formulation 2008 for the Thermodynamic Properties of Seawater. The International Association for the Properties of Water and Steam. Berlin, Germany, September 2008, available from www.iapws.org. This Release is referred to in the text as IAPWS-08.

IAPWS, 2009: Supplementary Release on a Computationally Efficient Thermodynamic Formulation for Liquid Water for Oceanographic Use. The International Association for the Properties of Water and Steam. Doorwerth, The Netherlands, September 2009, available from http://www.iapws.org. This Release is referred to in the text as IAPWS-09.

IOC, SCOR and IAPSO, 2010: The international thermodynamic equation of seawater – 2010: Calculation and use of thermodynamic properties. Intergovernmental Oceanographic Commission, Manuals and Guides No. 56, UNESCO (English), 196 pp. Available from http://www.TEOS-10.org

Here follows section 2.18 and appendix A.15 of the TEOS-10 Manual (IOC et al., 2010).

2.18 Thermal expansion coefficients

The thermal expansion coefficient α^t with respect to *in situ* temperature t, is

$$\alpha^{t} = \alpha^{t} \left(S_{A}, t, p \right) = -\frac{1}{\rho} \frac{\partial \rho}{\partial T} \Big|_{S_{A}, p} = \frac{1}{\nu} \frac{\partial \nu}{\partial T} \Big|_{S_{A}, p} = \frac{g_{TP}}{g_{P}}. \tag{2.18.1}$$

The thermal expansion coefficient α^{θ} with respect to potential temperature θ , is (see appendix A.15)

$$\alpha^{\theta} = \alpha^{\theta} \left(S_{A}, t, p, p_{r} \right) = -\frac{1}{\rho} \frac{\partial \rho}{\partial \theta} \bigg|_{S_{A}, p} = \frac{1}{\nu} \frac{\partial \nu}{\partial \theta} \bigg|_{S_{A}, p} = \frac{g_{TP}}{g_{P}} \frac{g_{TT} \left(S_{A}, \theta, p_{r} \right)}{g_{TT}}, \qquad (2.18.2)$$

where p_r is the reference pressure of the potential temperature. The g_{TT} derivative in the numerator is evaluated at (S_A, θ, p_r) whereas the other derivatives are all evaluated at (S_A, t, p) .

The thermal expansion coefficient α^{Θ} with respect to Conservative Temperature Θ , is (see appendix A.15)

$$\alpha^{\Theta} = \alpha^{\Theta} \left(S_{A}, t, p \right) = -\frac{1}{\rho} \frac{\partial \rho}{\partial \Theta} \Big|_{S_{A}, p} = \frac{1}{\nu} \frac{\partial \nu}{\partial \Theta} \Big|_{S_{A}, p} = -\frac{g_{TP}}{g_{P}} \frac{c_{p}^{0}}{\left(T_{0} + \theta \right) g_{TT}}. \tag{2.18.3}$$

Note that Conservative Temperature Θ is defined only with respect to a reference pressure of 0 dbar so that the θ in Eqn. (2.18.3) is the potential temperature with $p_r = 0$ dbar. All the derivatives on the right-hand side of Eqn. (2.18.3) are evaluated at (S_A, t, p) . The constant c_p^0 is defined in Eqn. (3.3.3) below.

A.15 Derivation of the expressions for α^{θ} , β^{θ} , α^{Θ} and β^{Θ}

This appendix derives the expressions in Eqns. (2.18.2) – (2.18.3) and (2.19.2) – (2.19.3) for the thermal expansion coefficients α^{θ} and α^{Θ} and the haline contraction coefficients β^{θ} and β^{Θ} .

In order to derive Eqn. (2.18.2) for α^{θ} we first need an expression for $\partial \theta / \partial T|_{S_A, p}$. This is found by differentiating with respect to *in situ* temperature the entropy equality $\eta(S_A, t, p) = \eta(S_A, \theta[S_A, t, p, p_r], p_r)$ which defines potential temperature, obtaining

$$\frac{\partial \theta}{\partial T}\Big|_{S_{A},p} = \frac{\eta_{T}\left(S_{A},t,p\right)}{\eta_{T}\left(S_{A},\theta,p_{r}\right)} = \frac{g_{TT}\left(S_{A},t,p\right)}{g_{TT}\left(S_{A},\theta,p_{r}\right)} = \frac{\left(T_{0}+\theta\right)}{\left(T_{0}+t\right)} \frac{c_{p}\left(S_{A},t,p\right)}{c_{p}\left(S_{A},\theta,p_{r}\right)}.$$
(A.15.1)

This is then used to obtain the desired expression Eqn. (2.18.2) for α^{θ} as follows

$$\alpha^{\theta} = \frac{1}{v} \frac{\partial v}{\partial \theta} \bigg|_{S_{A}, p} = \frac{1}{v} \frac{\partial v}{\partial T} \bigg|_{S_{A}, p} \left(\frac{\partial \theta}{\partial T} \bigg|_{S_{A}, p} \right)^{-1} = \frac{g_{TP}(S_{A}, t, p)}{g_{P}(S_{A}, t, p)} \frac{g_{TT}(S_{A}, \theta, p_{r})}{g_{TT}(S_{A}, t, p)}.$$
(A.15.2)

In order to derive Eqn. (2.18.3) for α^{Θ} we first need an expression for $\partial \Theta/\partial t\big|_{S_A,p}$. This is found by differentiating with respect to *in situ* temperature the entropy equality $\eta(S_A,t,p) = \hat{\eta}(S_A,\Theta[S_A,t,p])$ obtaining

$$\left. \frac{\partial \Theta}{\partial T} \right|_{S_{A}, p} = \left. \eta_{T} \left(S_{A}, t, p \right) \frac{\partial \Theta}{\partial \eta} \right|_{S_{A}} = -\left(T_{0} + \theta \right) \frac{g_{TT} \left(S_{A}, t, p \right)}{c_{p}^{0}} = \frac{\left(T_{0} + \theta \right)}{\left(T_{0} + t \right)} \frac{c_{p} \left(S_{A}, t, p \right)}{c_{p}^{0}}, \quad (A.15.3)$$

where the second part of this equation has used Eqn. (A.12.4) for $\Theta_{\eta}|_{S_A}$. This is then used to obtain the desired expression Eqn. (2.18.3) for α^{Θ} as follows

$$\alpha^{\Theta} = \frac{1}{v} \frac{\partial v}{\partial \Theta} \Big|_{S_{A}, p} = \frac{1}{v} \frac{\partial v}{\partial T} \Big|_{S_{A}, p} \left(\frac{\partial \Theta}{\partial T} \Big|_{S_{A}, p} \right)^{-1} = -\frac{g_{TP} \left(S_{A}, t, p \right)}{g_{P} \left(S_{A}, t, p \right)} \frac{c_{p}^{0}}{\left(T_{0} + \theta \right) g_{TT} \left(S_{A}, t, p \right)}. \tag{A.15.4}$$

In order to derive Eqn. (2.19.2) for β^{θ} we first need an expression for $\partial \theta/\partial S_A|_{T,p}$. This is found by differentiating with respect to Absolute Salinity the entropy equality $\eta(S_A,t,p)=\eta(S_A,\theta[S_A,t,p,p_r],p_r)$ which defines potential temperature, obtaining

$$\frac{\partial \theta}{\partial S_{A}}\Big|_{T,p} = \theta_{\eta}\Big|_{S_{A}}\Big[\eta_{S_{A}}(S_{A},t,p) - \eta_{S_{A}}(S_{A},\theta,p_{r})\Big]$$

$$= \frac{(T_{0}+\theta)}{c_{p}(S_{A},\theta,p_{r})}\Big[\mu_{T}(S_{A},\theta,p_{r}) - \mu_{T}(S_{A},t,p)\Big]$$

$$= \Big[g_{S_{A}T}(S_{A},t,p) - g_{S_{A}T}(S_{A},\theta,p_{r})\Big]\Big/g_{TT}(S_{A},\theta,p_{r}),$$
(A.15.5)

where Eqns. (A.12.5) and (A.12.7) have been used with a general reference pressure $p_{\rm r}$ rather than with $p_{\rm r}=0$. By differentiating $\rho=\tilde{\rho}\left(S_{\rm A},\theta\left[S_{\rm A},t,p,p_{\rm r}\right],p\right)$ with respect to Absolute Salinity it can be shown that (Gill (1982), McDougall (1987a))

$$\beta^{\theta} = \frac{1}{\rho} \frac{\partial \rho}{\partial S_{A}} \bigg|_{\theta, p} = \frac{1}{\rho} \frac{\partial \rho}{\partial S_{A}} \bigg|_{T, p} + \alpha^{\theta} \frac{\partial \theta}{\partial S_{A}} \bigg|_{T, p}, \tag{A.15.6}$$

and using Eqn. (A.15.5) we arrive at the desired expression Eqn. (2.19.2) for β^{θ}

$$\beta^{\theta} = -\frac{g_{S_{A}P}(S_{A},t,p)}{g_{P}(S_{A},t,p)} + \frac{g_{TP}(S_{A},t,p) \left[g_{S_{A}T}(S_{A},t,p) - g_{S_{A}T}(S_{A},\theta,p_{r})\right]}{g_{P}(S_{A},t,p)g_{TT}(S_{A},t,p)} . \tag{A.15.7}$$

Note that the terms in the natural logarithm of the square root of Absolute Salinity cancel from the two parts of the square brackets in Eqns. (A.15.5) and (A.15.7).

In order to derive Eqn. (2.19.3) for β^{Θ} we first need an expression for $\partial \Theta/\partial S_A|_{T,p}$. This is found by differentiating with respect to Absolute Salinity the entropy equality $\eta(S_A,t,p)=\hat{\eta}(S_A,\Theta[S_A,t,p])$ obtaining (using Eqns. (A.12.4) and (A.12.8))

$$\frac{\partial \Theta}{\partial S_{\mathbf{A}}}\Big|_{T,p} = \Theta_{\eta}\Big|_{S_{\mathbf{A}}}\Big[\eta_{S_{\mathbf{A}}}(S_{\mathbf{A}},t,p) - \hat{\eta}_{S_{\mathbf{A}}}\Big|_{\Theta}\Big]$$

$$= \Big[\mu(S_{\mathbf{A}},\theta,0) - (T_{0}+\theta)\mu_{T}(S_{\mathbf{A}},t,p)\Big]/c_{p}^{0}$$

$$= \Big[g_{S_{\mathbf{A}}}(S_{\mathbf{A}},\theta,0) - (T_{0}+\theta)g_{S_{\mathbf{A}}T}(S_{\mathbf{A}},t,p)\Big]/c_{p}^{0}.$$
(A.15.8)

Differentiating $\rho = \hat{\rho}(S_A, \Theta[S_A, t, p], p)$ with respect to Absolute Salinity leads to

$$\beta^{\Theta} = \frac{1}{\rho} \frac{\partial \rho}{\partial S_{A}} \bigg|_{\Theta, p} = \frac{1}{\rho} \frac{\partial \rho}{\partial S_{A}} \bigg|_{T, p} + \alpha^{\Theta} \frac{\partial \Theta}{\partial S_{A}} \bigg|_{T, p}, \tag{A.15.9}$$

and using Eqn. (A.15.8) we arrive at the desired expression (2.19.3) for β^{Θ} namely

$$\beta^{\Theta} = -\frac{g_{S_{A}P}(S_{A},t,p)}{g_{P}(S_{A},t,p)} + \frac{g_{TP}(S_{A},t,p) \left[g_{S_{A}T}(S_{A},t,p) - g_{S_{A}}(S_{A},\theta,0) / (T_{0}+\theta)\right]}{g_{P}(S_{A},t,p) g_{TT}(S_{A},t,p)} . \quad (A.15.10)$$

Note that the terms in the natural logarithm of the square root of Absolute Salinity cancel from the two parts of the square brackets in Eqns. (A.15.8) and (A.15.10).