

Xenon-Doped Liquid Argon Scintillation Light

UNM DEPARTMENT OF PHYSICS & ASTRONOMY



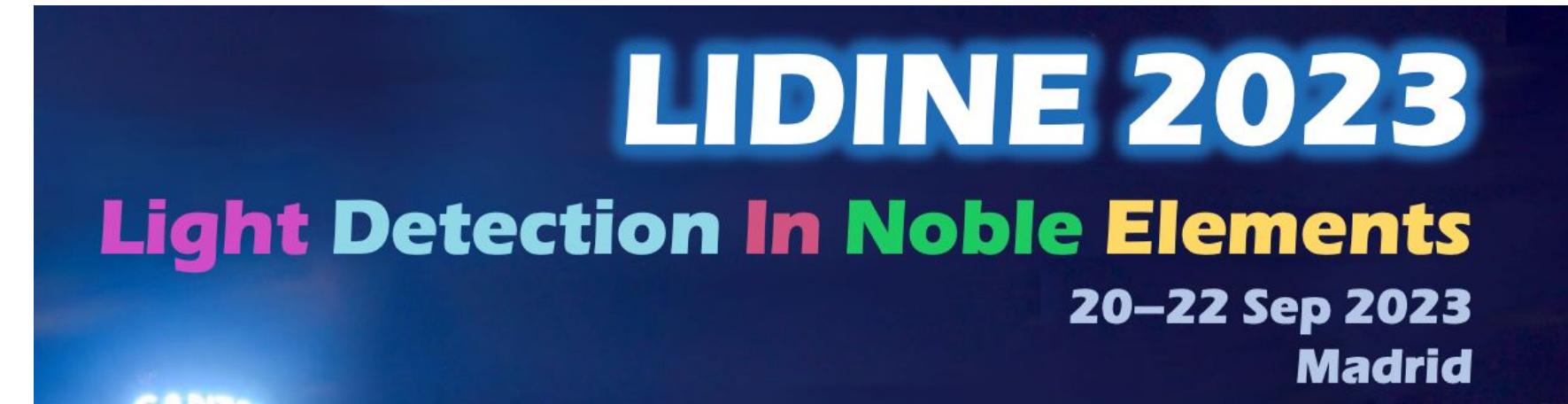
D.E. Fields, M. Gold, T. Reza, Luis
Flores-Sanchez (UNM)
S.R. Elliott, R. Massarczyk, A. Mazumdar, C.
Romo Luque, T. Thorpe, N. O'Brien, B. Turner
(LANL)

<https://arxiv.org/pdf/2009.10755.pdf>

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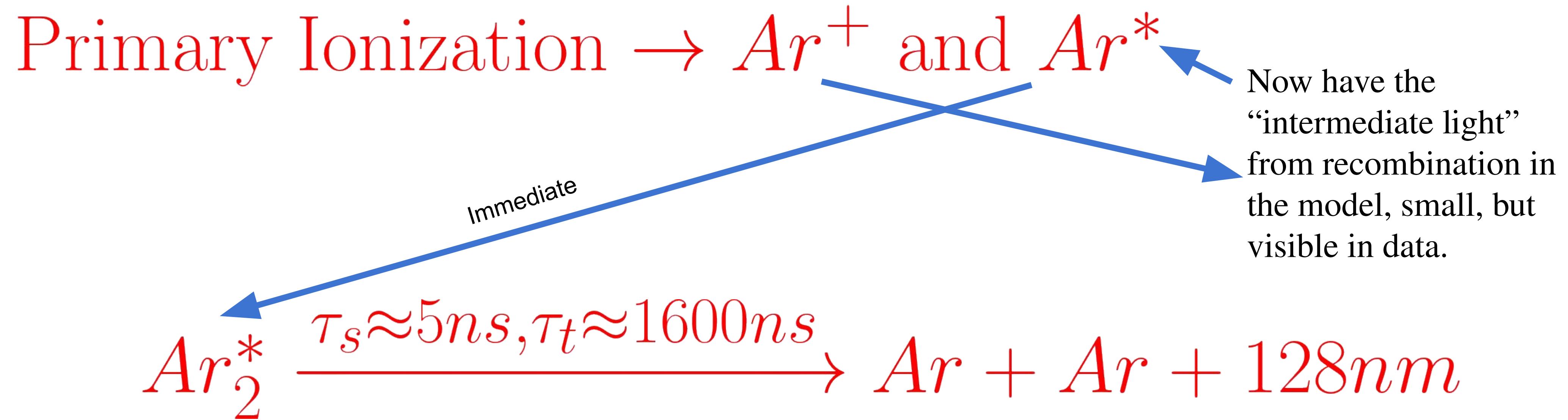


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Sep. 22, 2023

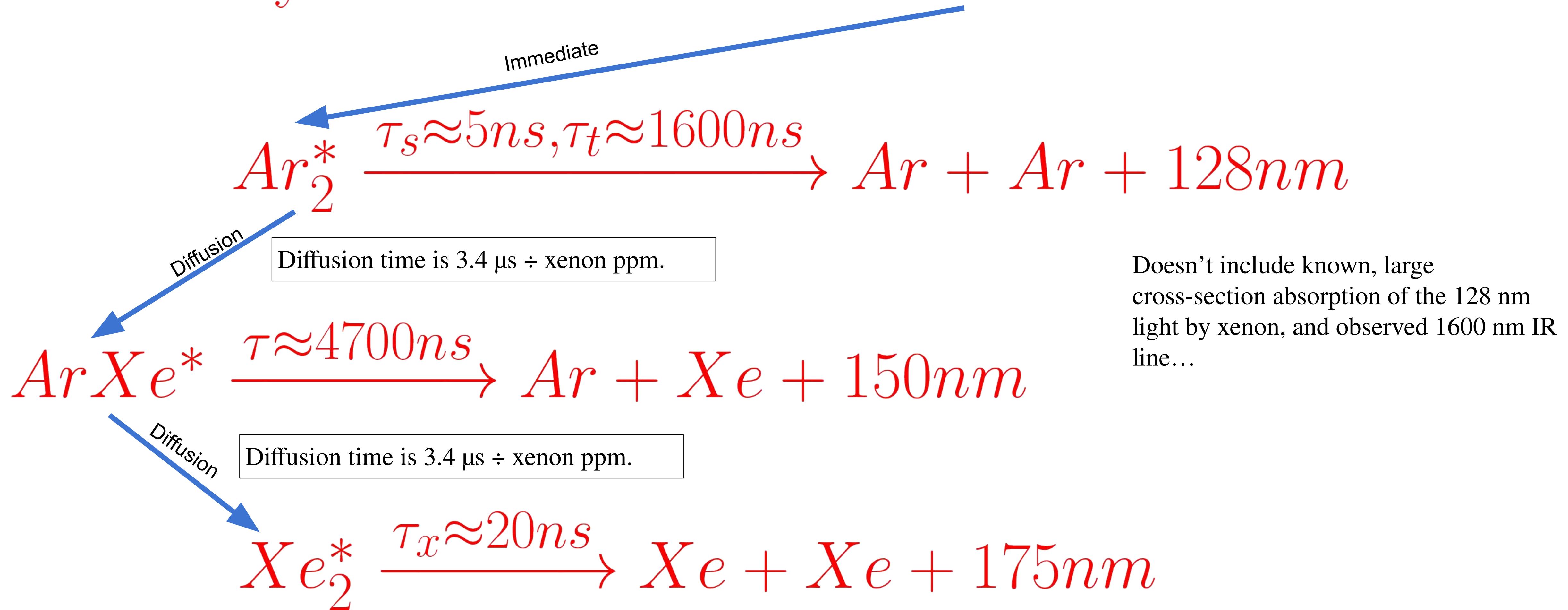
Evolution of Our Understanding: *Pure LAr*



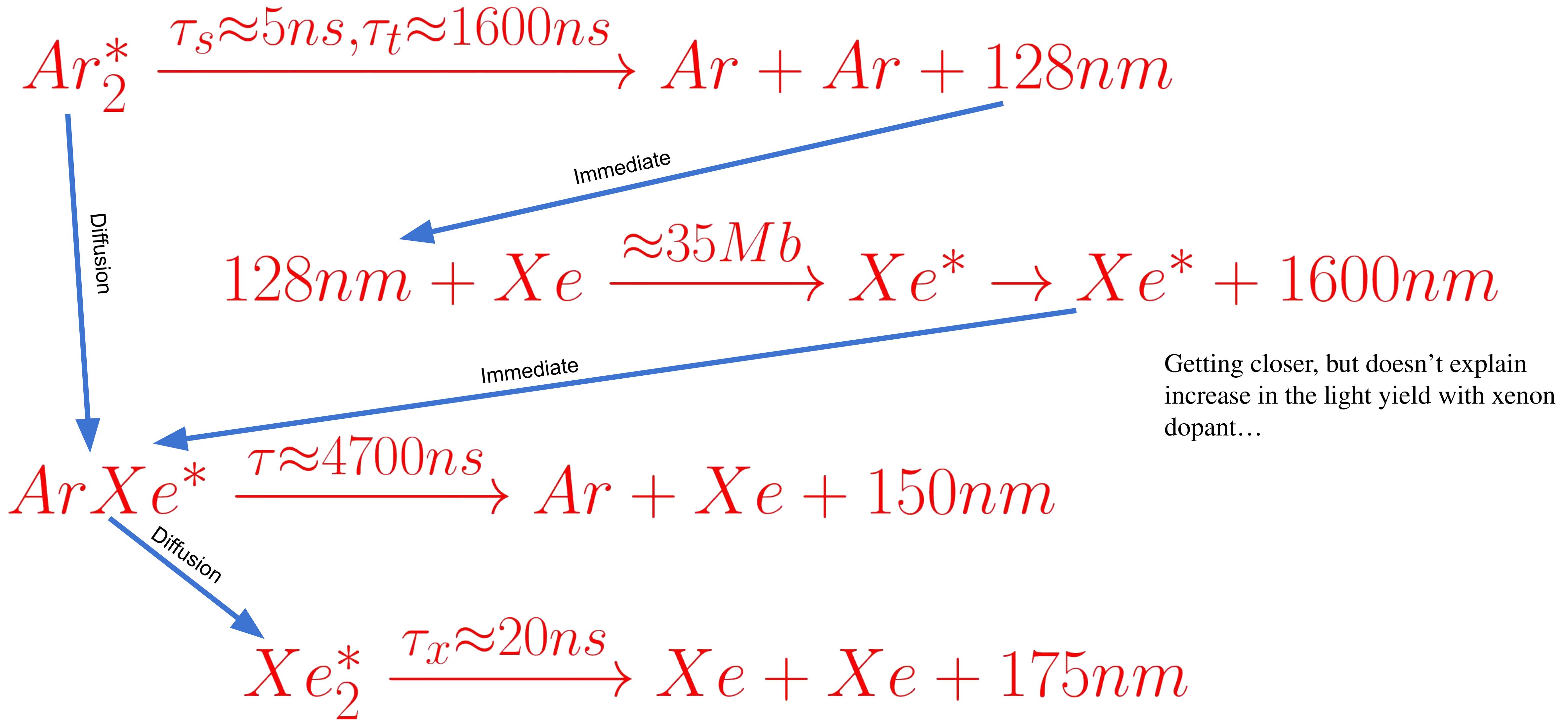
Very different measurements of the long, triplet lifetime from 1100 to 2100 ns. Ratio of triplet to singlet different for different ionizing particles and can change in the presence of an electric field. Doping with xenon has dramatic effects...

Evolution of Our Understanding: Xe-doped LAr

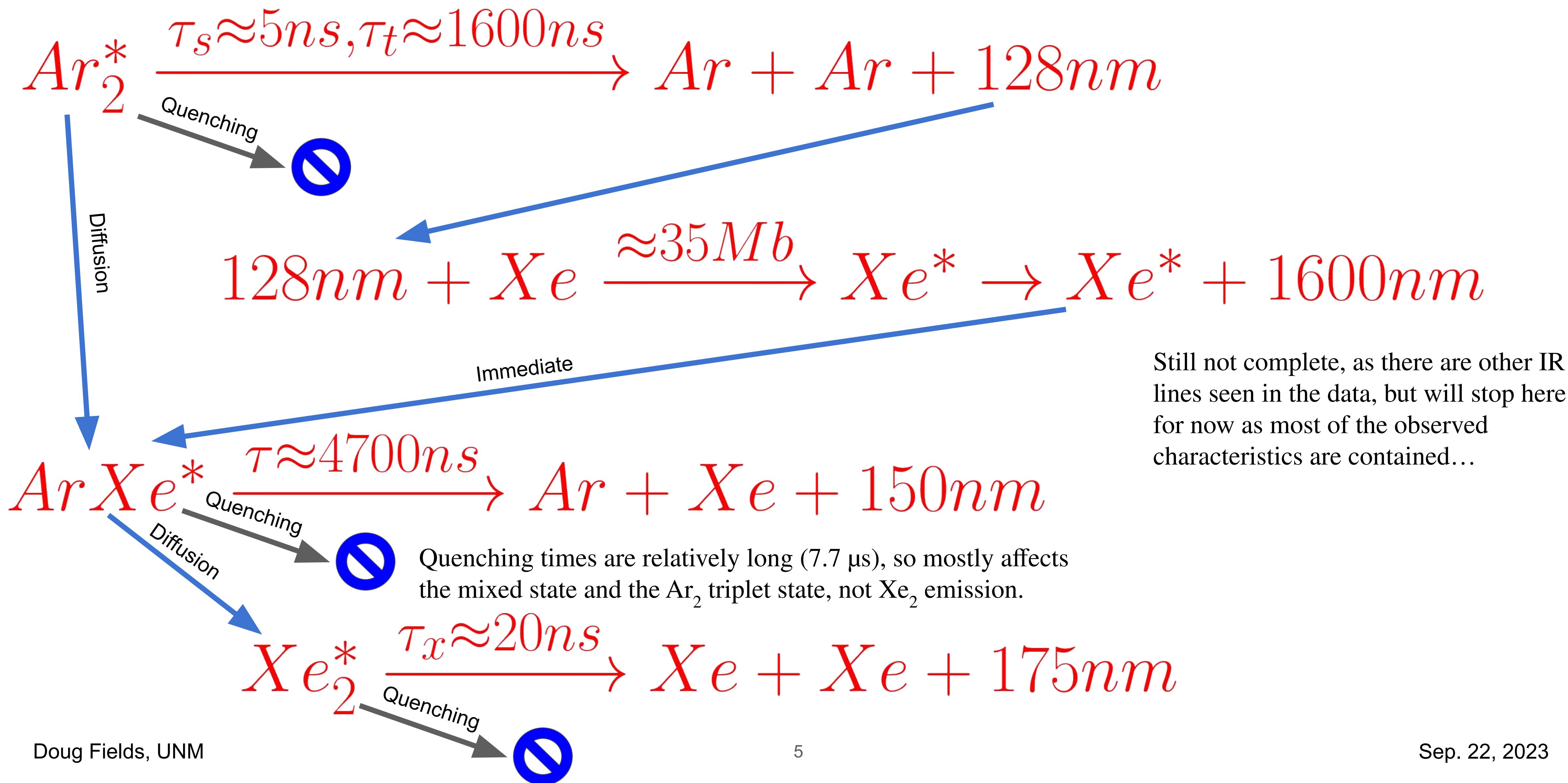
Primary Ionization $\rightarrow Ar^+$ and Ar^*



Evolution of Knowledge: Absorption



Evolution of Knowledge: Quenching

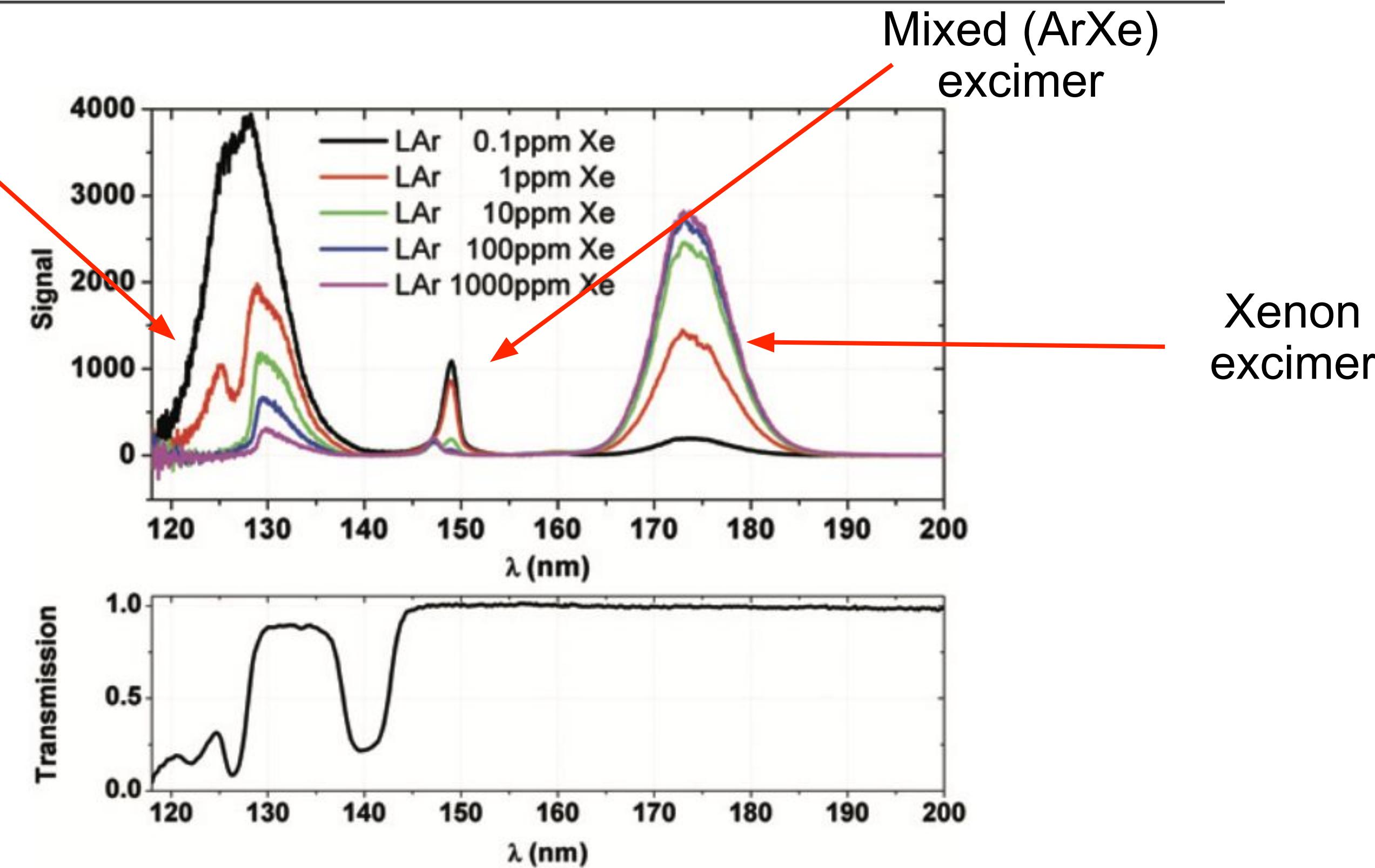


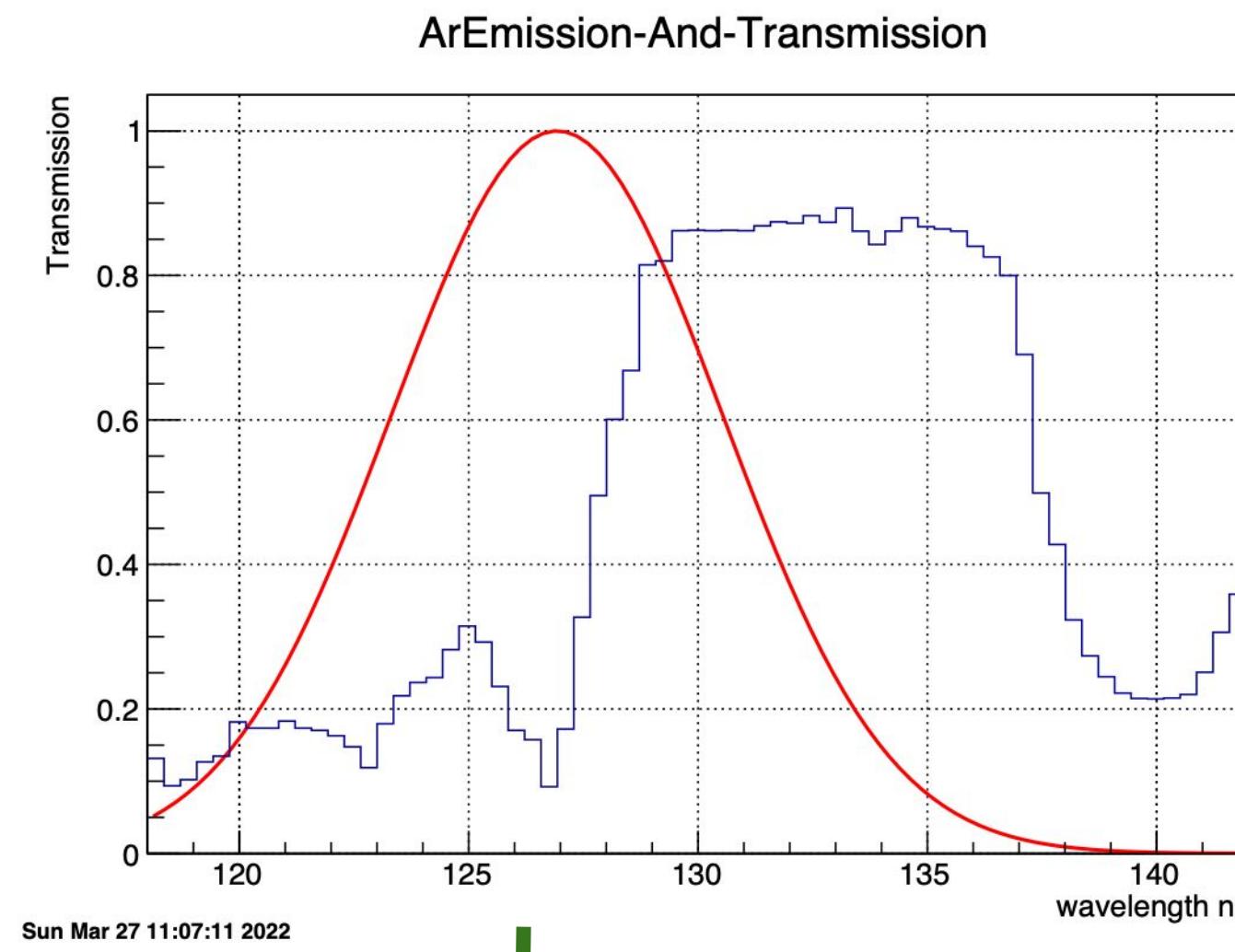
Absorption is not straightforward

A. Neumeier *et al.*

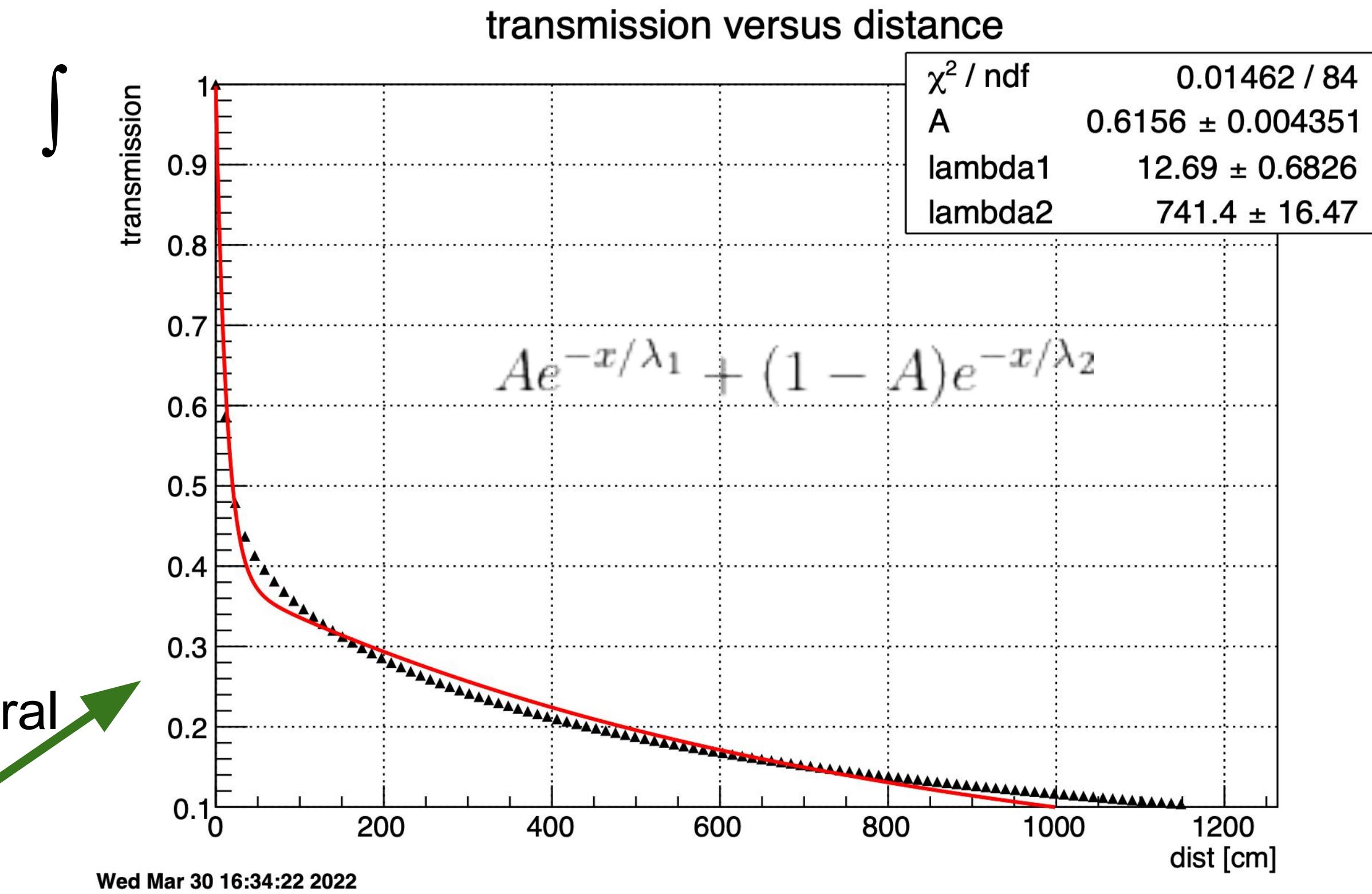
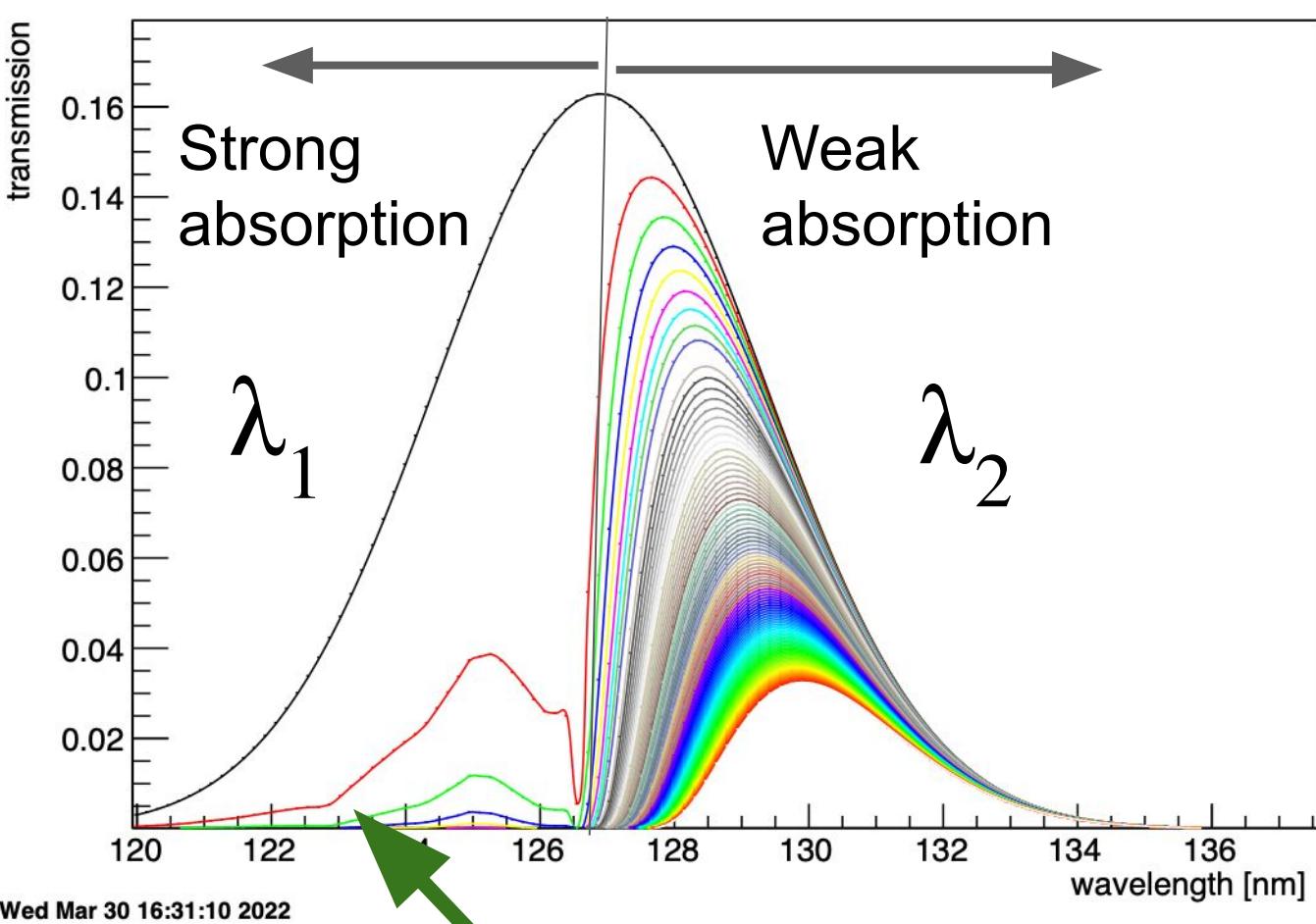
The VUV emission of
electron-beam excited liquid
argon doped (2 mm)

Transmission of liquid argon
doped with 0.1 ppm xenon
measured with a deuterium light
source and a length of the optical
path of 11.6 cm





Iterate convolution



Integral

For $\lambda_1 = 12.7$ cm,
 $\sigma_{\text{abs}} = 47.6/(12.7 \text{ cm})/(0.1 \text{ ppm}) = 37.5 \text{ Mb}$
 in agreement with Calvo *et al.* ArDM

Our Model

time dependencies of the different states. Our model consists of the following coupled differential equations for the number of molecules as a function of time t for the dimer states argon singlet $S(t)$, argon triplet $T(t)$, mixed $M(t)$ and xenon $X(t)$:

Note: We are adding the intermediate light in the model (can see the effect).

$$\dot{S} = -S/\tau_S - (k_x + k_q) S \equiv -\lambda_1 S \quad (1)$$

$$\dot{T} = -T/\tau_T - (k_x + k_q) T \equiv -\lambda_3 T \quad (2)$$

$$\dot{M} = - (1/\tau_M + k_x + k'_q) M + (k_x + A/\tau_S) S + (k_x + A/\tau_T) T \quad (3)$$

$$\dot{X} = -X/\tau_x + k_x M \quad (4)$$

We take A from the fit to the absorption curve.

k_x is calculated diffusion limited reaction rate in LAr for Ar, to Xe (scales with Xe concentration) $k_x = 2.9 \times 10^{-4} \times [\text{ppm}] \text{ ns}^{-1}$

Note: Below 0.5 ppm, quenching is faster than diffusion.

$k_{q'} \sim k_q = 1.3 \times 10^{-4} \text{ ns}^{-1}$ are quenching rates (Segreto, *Phys.Rev.D* 103 (2021) 4, 043001)

Light yield (time)

Then these equations are easily integrated giving $S(t)$, $T(t)$, $M(t)$ and $X(t)$, and the light seen ($128\text{ nm} + 150\text{ nm} + 175\text{ nm}$) is given by:

$$\ell(t) = N_1(1 - A)e^{-t\lambda_1}/\tau_S + N_3(1 - A)e^{-t\lambda_3}/\tau_T + M(t)/\tau_M + X(t)/\tau_X$$

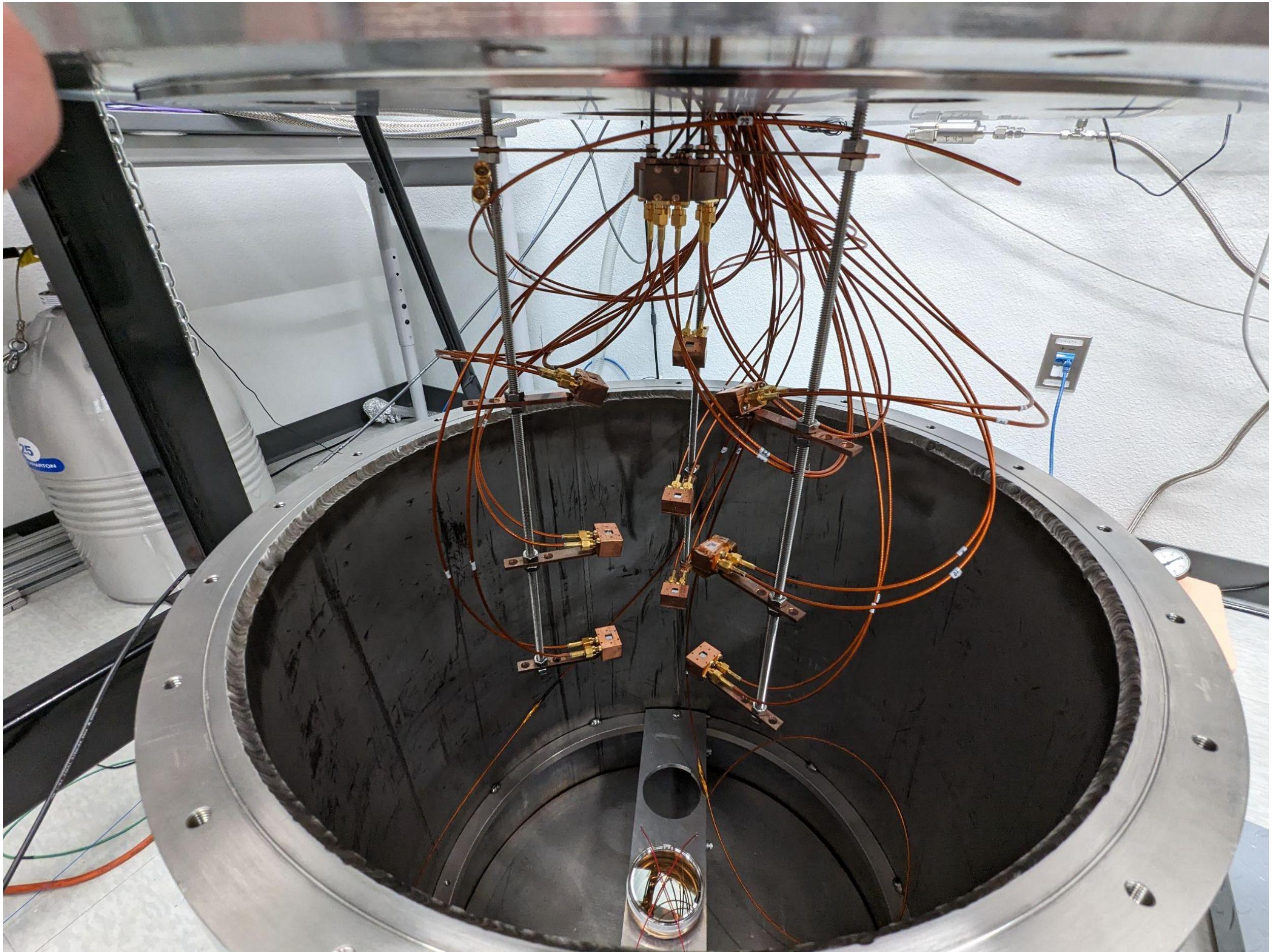
In terms of singlet fraction (sfrac) $N_1 = N^*sfrac$, $N_3 = N(1-sfrac)$

N is the total number of dimers created by initial ionization

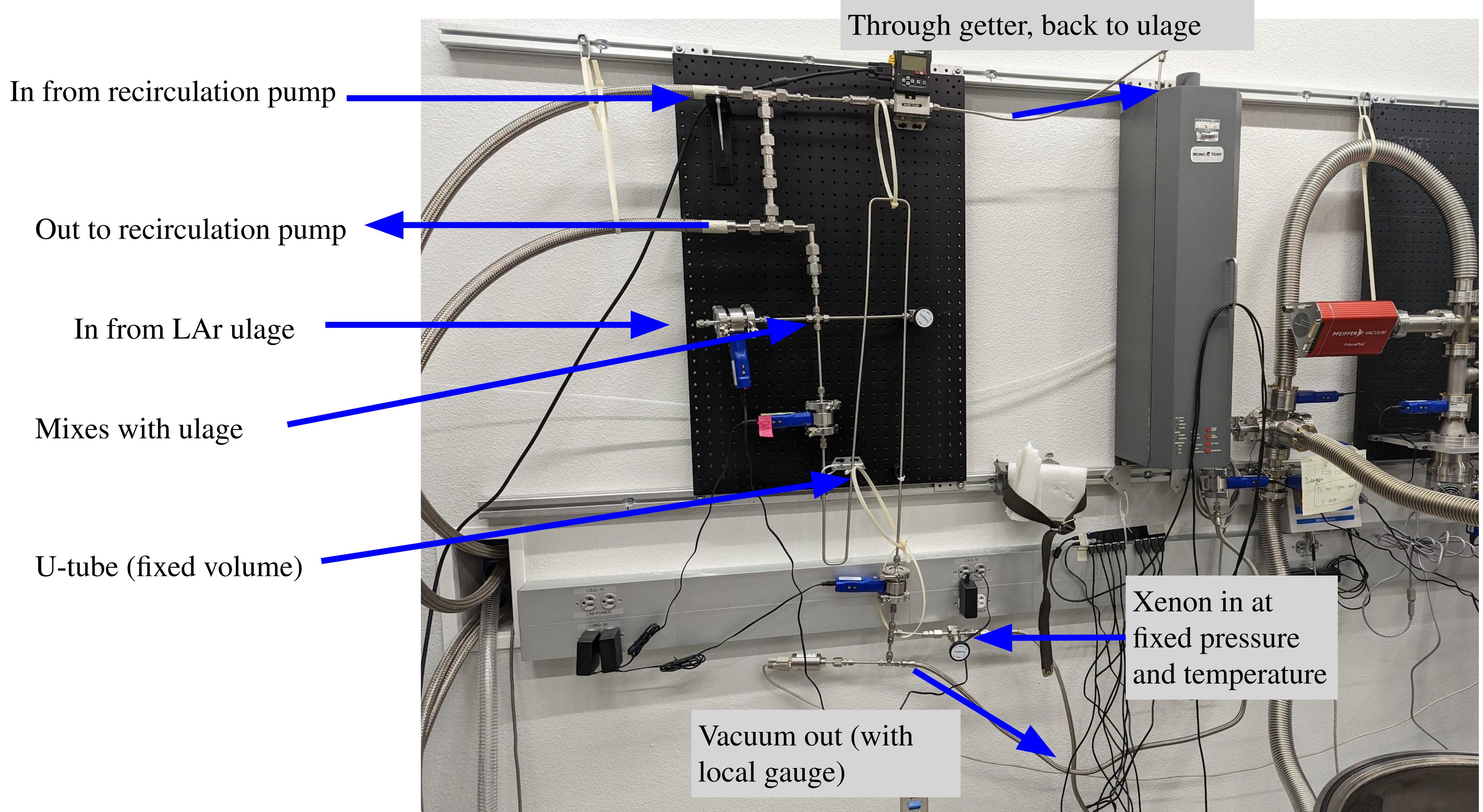
Similar to Segretto, but includes absorption, and is based on diffusion times

New Setup

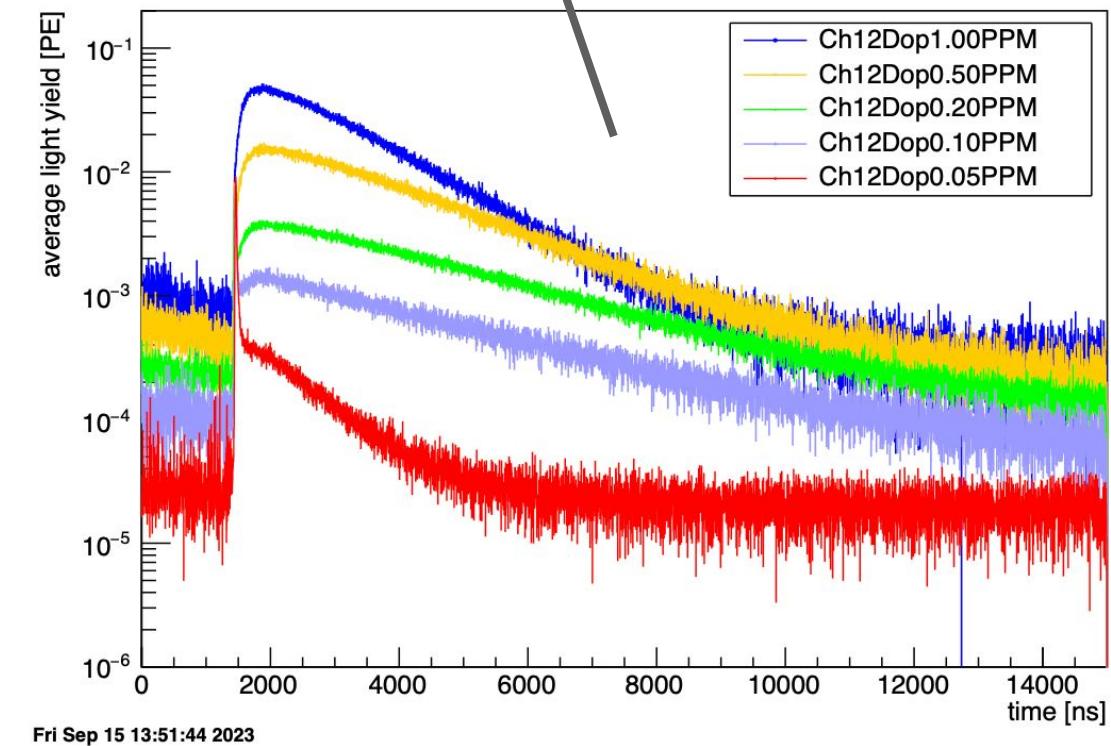
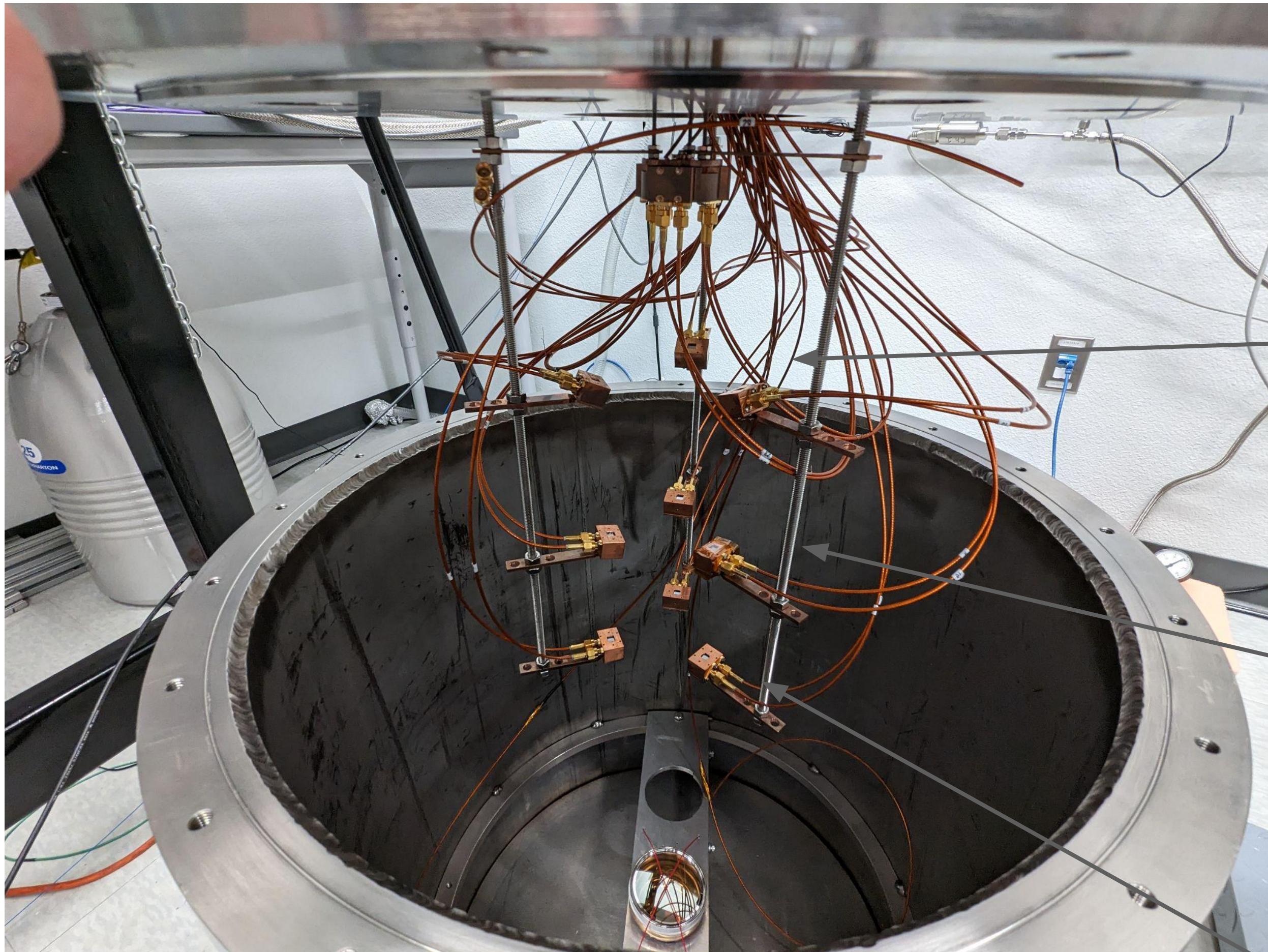
- One Hamamatsu R11410 (sensitive to 175 nm, but not 128 nm).
- Triggered ^{241}Am source aka Llama (LEGEND LAr monitoring system from TUM).
- Three levels of 3 SiPM (Hamamatsu 13370) detectors sensitive to both 128 nm and 175 nm.
- One SiPM covered with Suprasil glass to block 128 nm light.



Doping Procedure

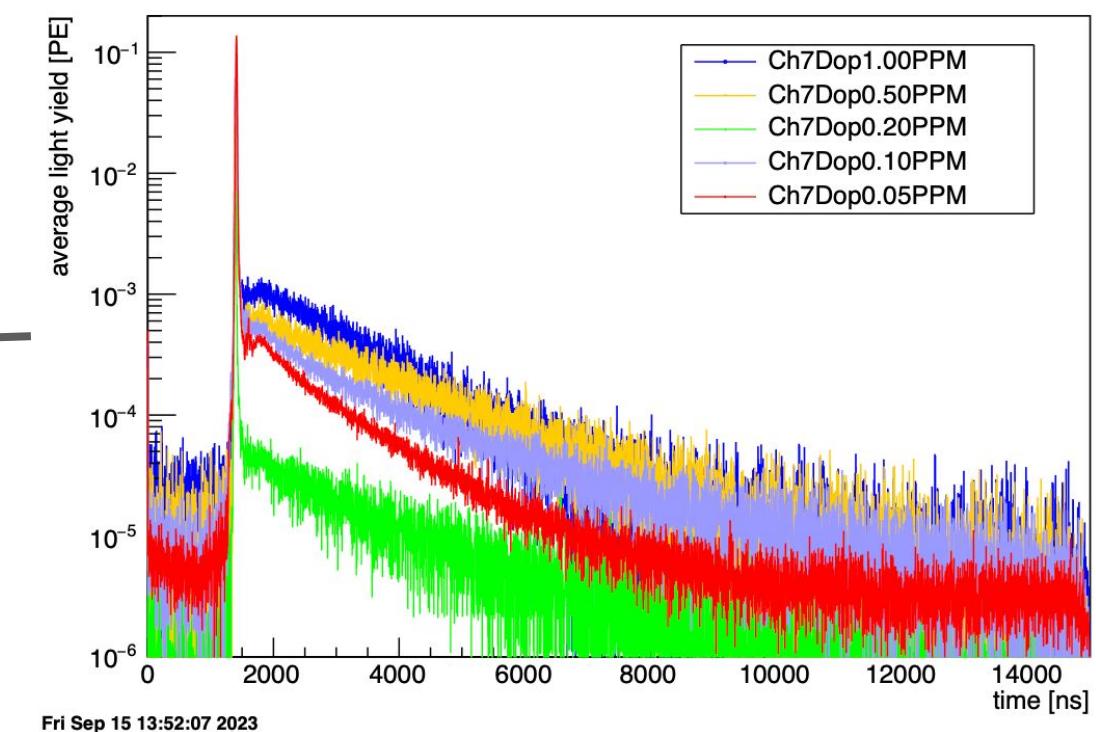


Preliminary Data

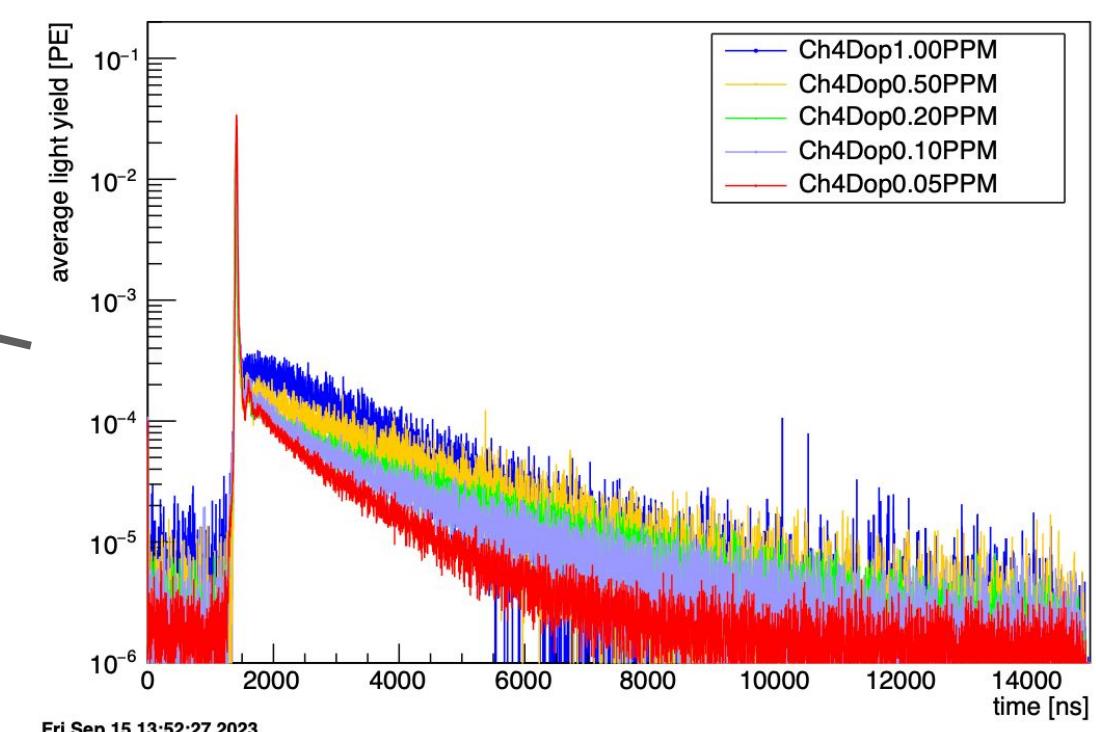


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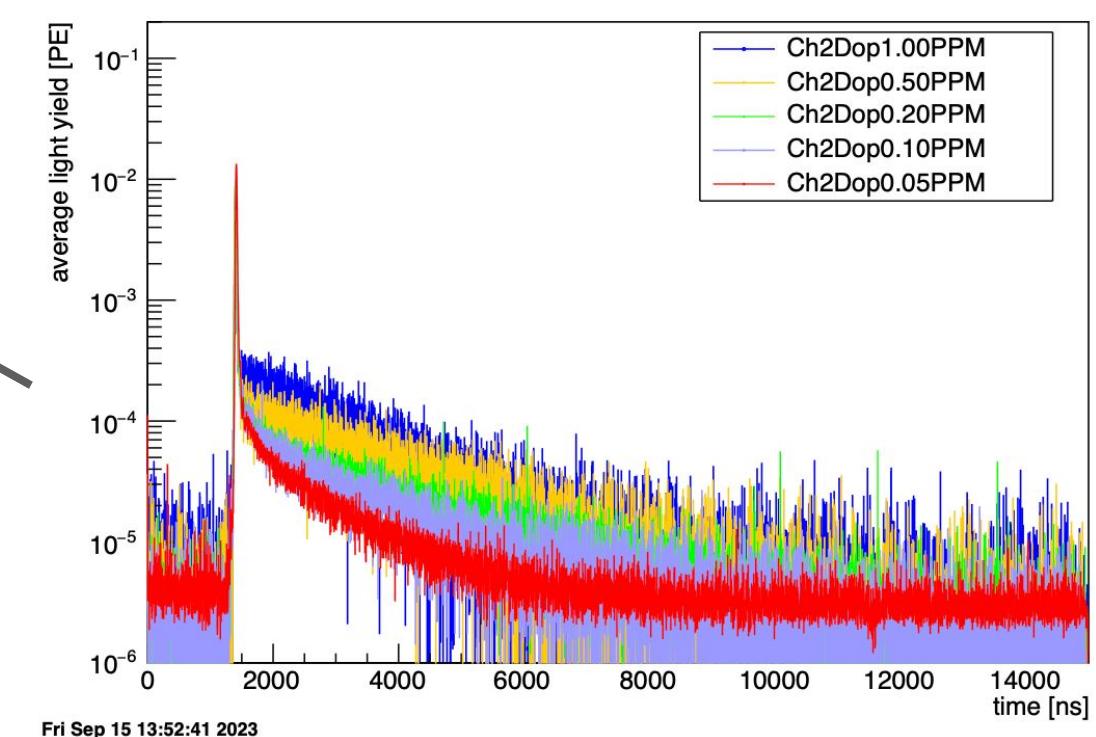
12



Undoped 0.01 -
0.07 ppm



Doped 0.1 ppm



Doped 0.2 ppm

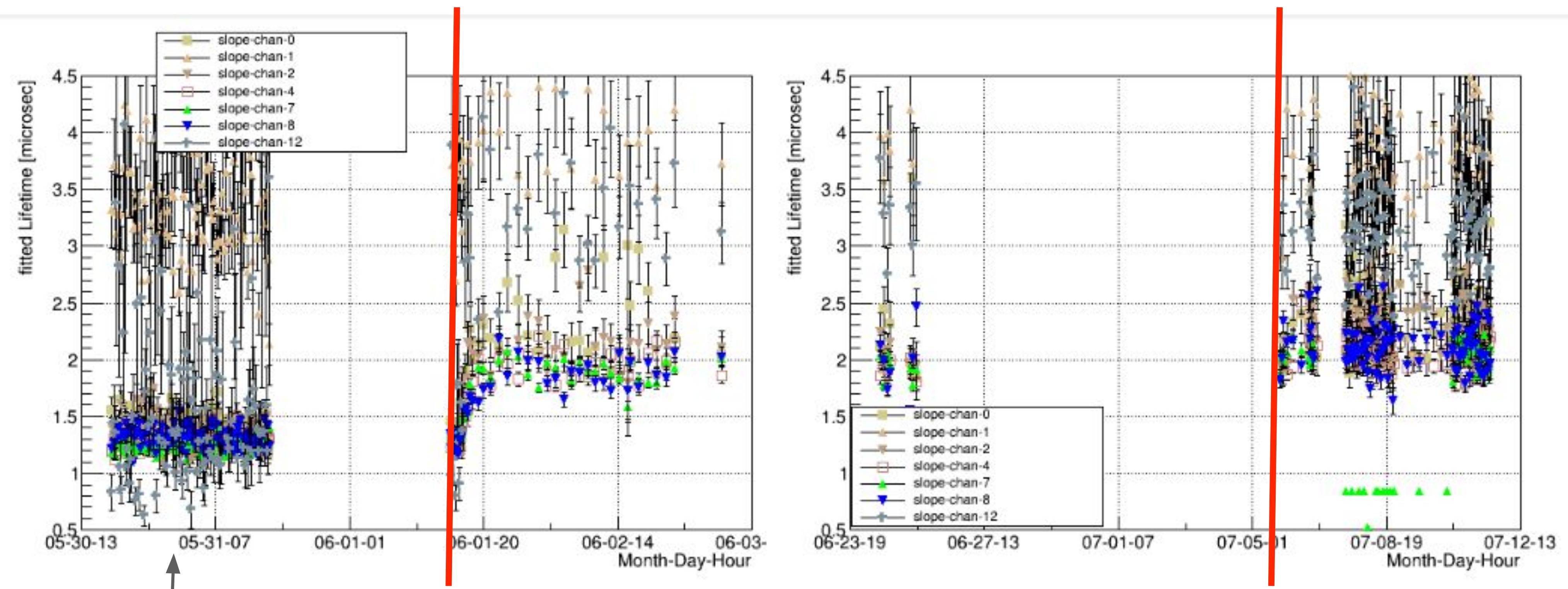
Doped 0.5 ppm

Doped 1.0 ppm

ep. 22, 2023

Preliminary Xe-Doping Data

Slopes from ~1.5 μ s after singlet to 5 μ s



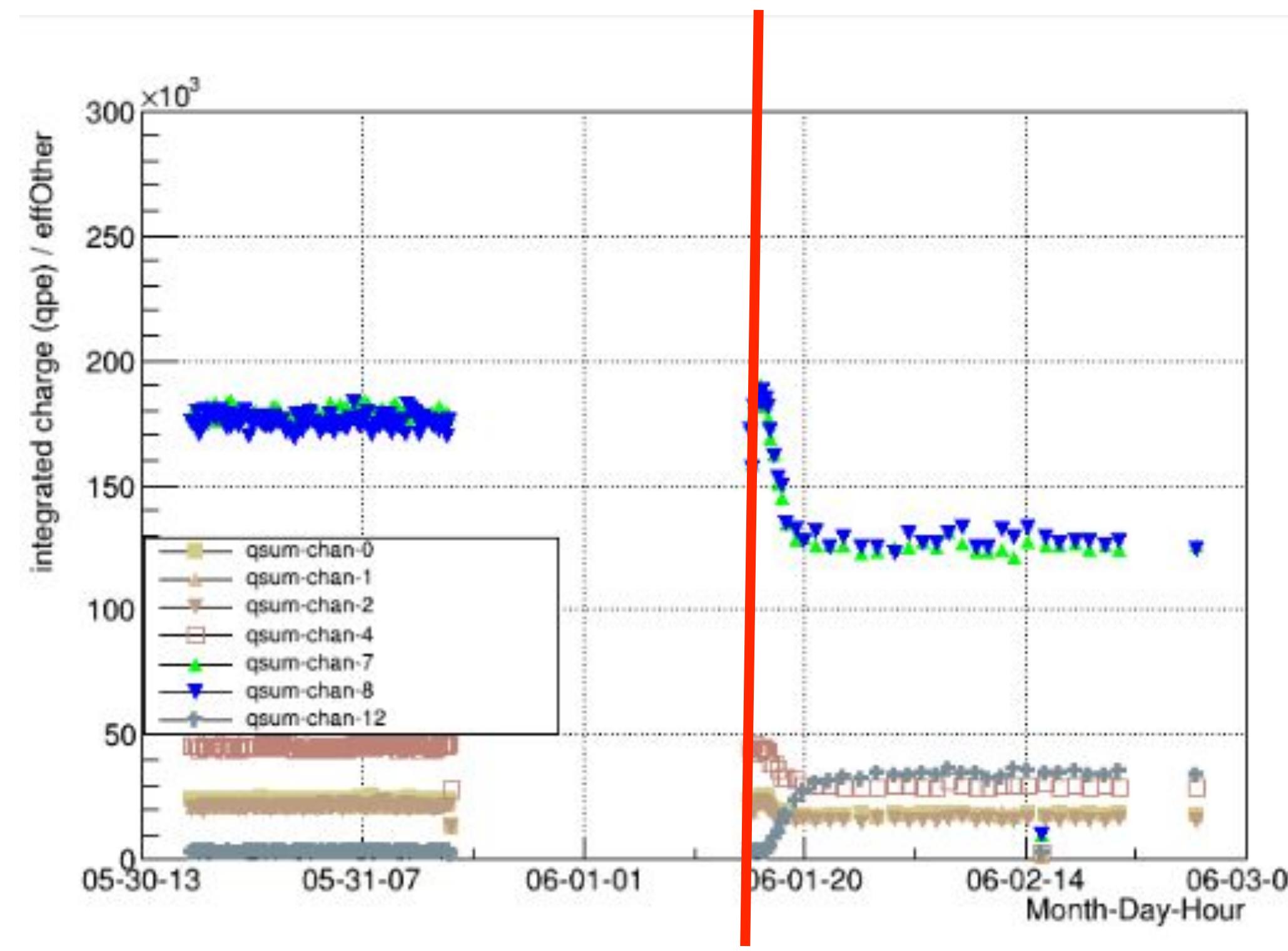
initial xenon
concentration in
ulage measured
to be $0.01 \pm$
 0.07 ppm

0.1 ppm (added
to initial)

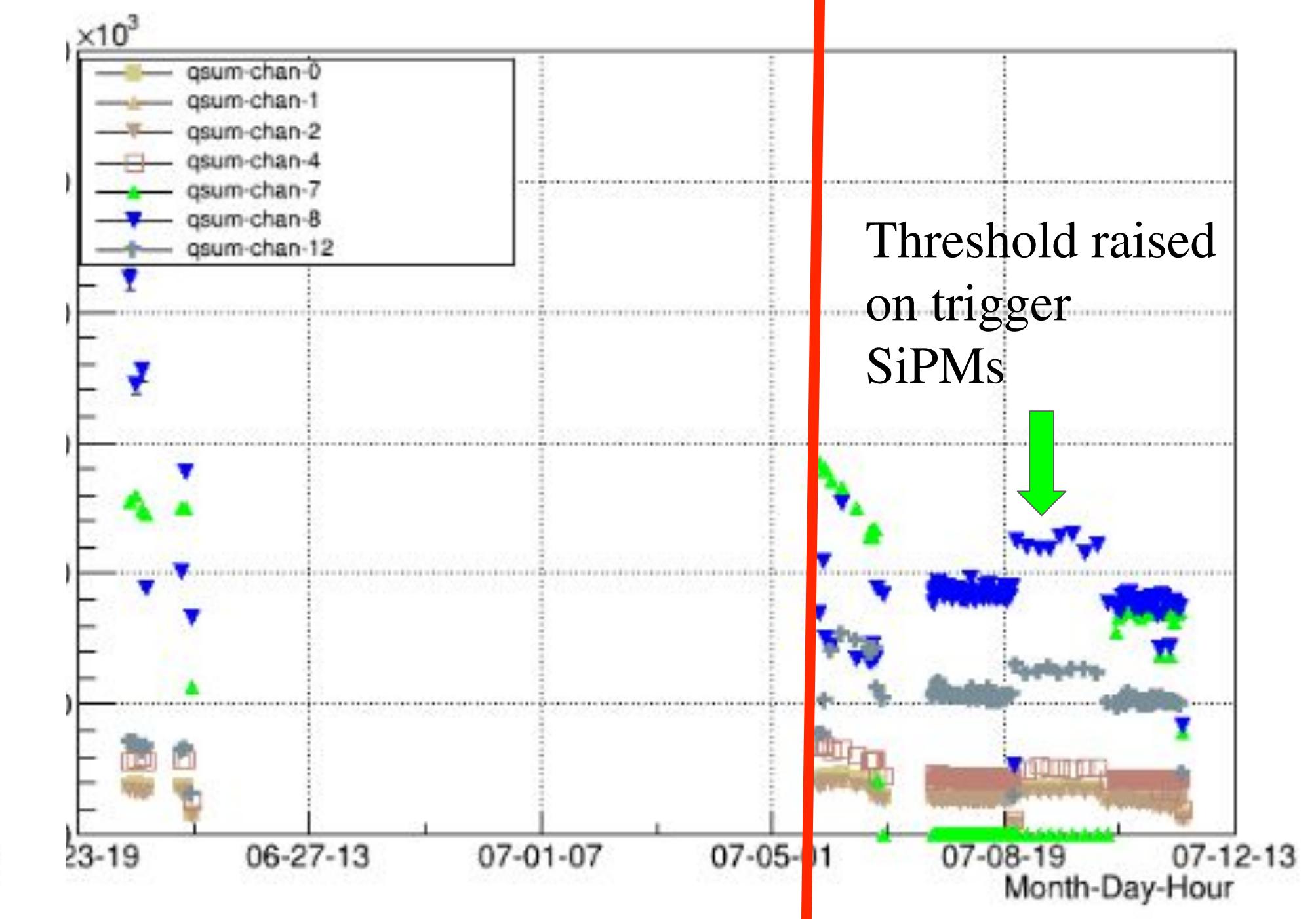
0.2 ppm total
added

Preliminary Xe-Doping Data

Integrated charge from ~1.5 μ s after singlet to 5 μ s



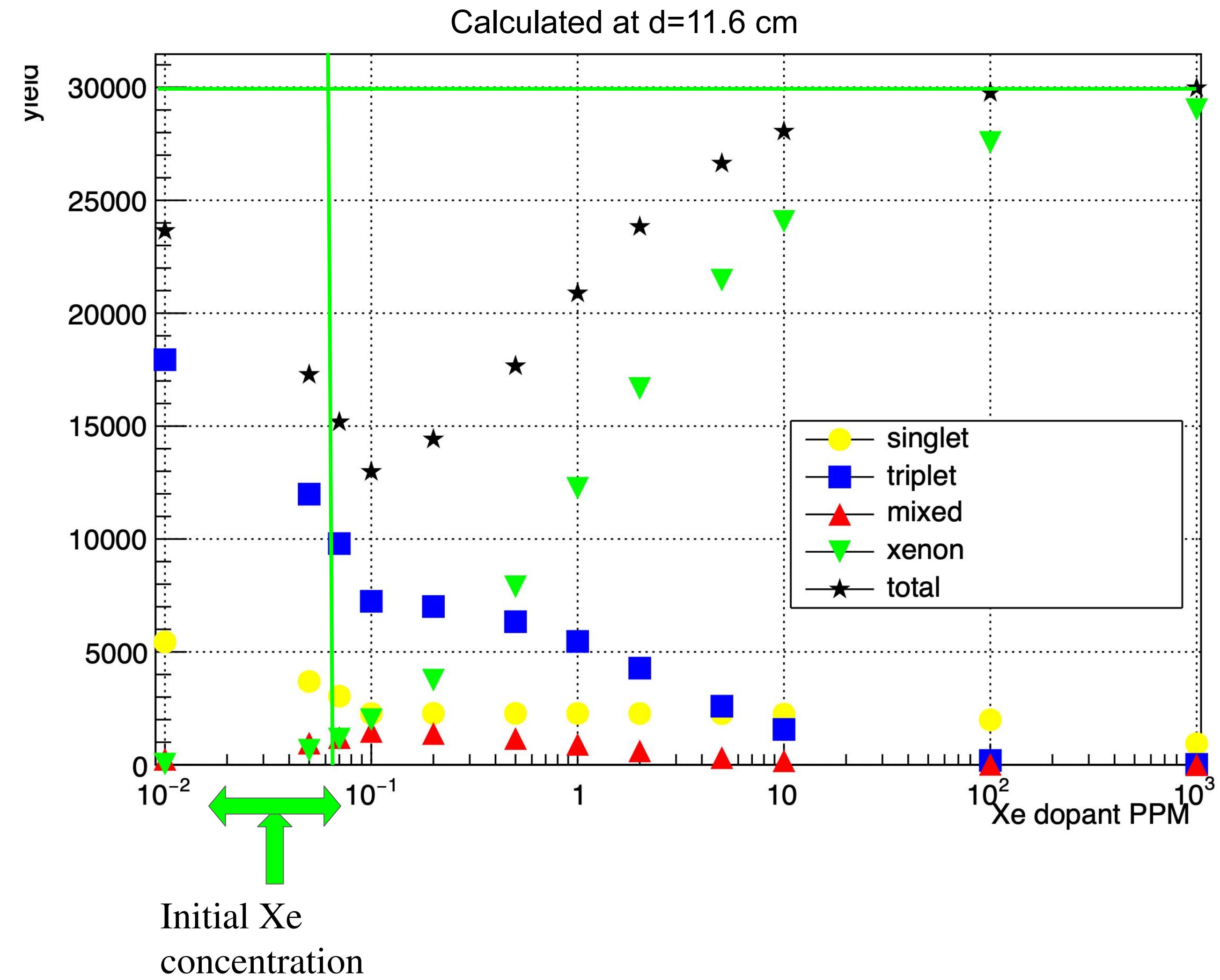
0.1 ppm (added
to initial)



0.2 ppm total
added

Model Prediction (NOT yet tuned to data)

- At doping levels greater than 0.1, we take A to be constant (as in our fits), but at very low PPM, the assumption of constant absorption is no longer valid, so we use values of A from the previous fit.
- From the model we can understand the increase in light given the initial concentration of Xe in the LAr.
- Notice that there is still very strong reduction in light at 0.1 ppm Xe
- Even at 0.01 ppm, the yield is still only 80% of original (number of excitons).
- At 10 ppm Xe, light is recovered to >90% of original.



Conclusions

- Xenon absorption of Argon 128 nm scintillation light is complicated and has a significant effect on the time spectra of emission and light yields even at very low concentrations.
- Quenching of long-lived states can account for increased light yields at Xe doping levels above 1 ppm.
- Our upcoming runs should narrow the parameter space for lifetimes and quenching times, and we can examine the effect on the fast/slow ratio for PID.
- Can also add N_2 , etc. to the model...
- Thank you for staying to the end!

Backup

Scintillation light in LAr

There have been many papers and much work done to characterize and understand the amount and time distribution of scintillation light in liquid argon...

Proc. Roy. Soc. Lond. A. 317, 113–131 (1970)
Printed in Great Britain

Experimental evidence
in liquid argon
By B. RAZ
Department of Chemistry, I
(Communicated by Sir Nevill 1

In connexion with studies of the electron capture process it was found whether there exist exciton states in simple vacuum ultraviolet spectroscopic study of liquid argon. Experimental evidence was obtained for Wannier exciton states in the excited states of the absorption spectra of the doped liquid argon. The following experimental results are reported:
(a) In the Xe/Ar liquid two absorptive $^1S_0 \rightarrow ^1P_1$ transitions (or alternatively $^1S_0 \rightarrow ^3P_1$) at 141 nm (8.80 eV)† and at 123 nm (10.1 eV).
(b) In the Xe/Kr liquid three absorptions at 125.5 nm (9.89 eV) and 129 nm (9.6 eV).
(c) The absorption spectra of the dopants.

16v3 [physics.ins-det] 21 Sep 2019

Attenuation of vacuum ultraviolet light in pure and xenon-doped liquid argon - an approach to an assignment of the near-infrared emission from the mixture

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Published in EPL (2015)

Abstract. Results of transmission experiments of vacuum ultraviolet light through a 11.6 cm long cell filled with pure and xenon-doped liquid argon are described. Pure liquid argon shows no attenuation down to the experimental short-wavelength cut-off at 118 nm. Based on a conservative approach, a lower limit of 1.10 m for the attenuation length of its own scintillation light could be derived. Adding xenon to liquid argon at concentrations on the order of parts per million leads to strong xenon-related absorption features which are used for a tentative assignment of the recently found near-infrared emission observed in electron-beam excited liquid argon-xenon mixtures. Two of the three absorption features can be explained by perturbed xenon transitions and the third one by a trapped exciton (Wannier-Mott) impurity state. A calibration curve connecting the equivalent width of the absorption line at 140 nm with xenon concentration is provided.

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24 Nov 2015

Fast component re-emission in Xe-doped liquid argon

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ABSTRACT: We present the first direct experimental confirmation of the fast component re-emission in liquid argon (LAr) doped with xenon (Xe). This effect was studied at various Xe concentrations up to ~ 3000 ppm. The rate constant of energy transfer for the fast component was quantified. It was shown that LAr doped with a high concentration of Xe without TPB has a better Pulse Discrimination (PSD) efficiency than pure LAr or Xe-doped LAr with TPB. The stability of the mixture was tested for the first time at high Xe concentration for long continuous runtimes.

Novosibirsk State University - Novosibirsk, 630090, Russia

Emission

nas - UNICAMP
São Paulo, Brazil

In Matter experiments at the Ar-Xe scintillator. Liquid argon and 127 nm and with the decay of the lowest lying bound state. A model is proposed through the excimer Ar_2^+ ions. The dependence on the intensity of the emission is due to the abundance of the ions in the mixture. The experimental results are presented.

doi: 10.1209/0295-5075/109/12001

Intense vacuum ultraviolet and infrared scintillation of liquid Ar-Xe mixtures

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Table-top setup for investigating the scintillation properties of liquid argon

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IST (2011)

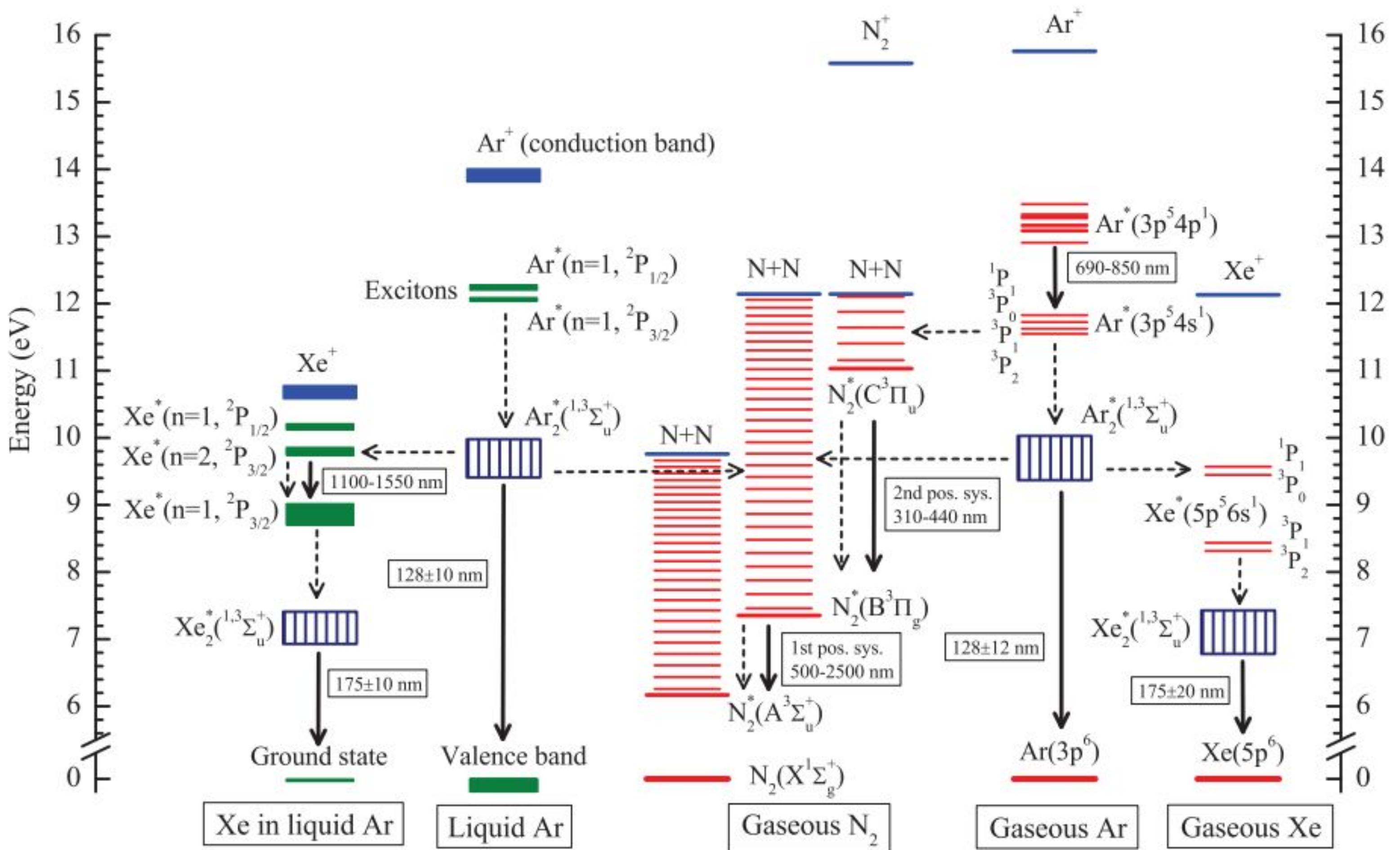
The light and temporal light emission properties of liquid argon have been studied in the context of rare-gas detectors for detecting Dark Matter particles in astronomy. A table-top setup for continuous and pulsed low energy electron beam excitation is used to stimulate light emission from 110 to 1000 nm in wavelength is covered by the detection system with a time resolution of 1 ns.

In xenon-doped liquid argon is described in the concentrations in liquid argon from 0.1 ppm to 1000 ppm. From the second excimer continuum of argon non (~ 174 nm) is observed by recording optical spectra. At a xenon concentration of ~ 10 ppm for which, a peak wavelength of $1.17 \mu\text{m}$ with 13000 ± 4000 is found. The corresponding value for the VUV excimer continuum of xenon is determined to be $1.17 \mu\text{m}$. Under these excitation conditions pure electron energy deposited at a peak wavelength spectrum for the 10 ppm Ar-Xe liquid mixture UV emission spectra from xenon-doped liquid ions from 0.1 ppm to 1000 ppm are also shown. At well-defined wavelength positions in the are presented.

Starting Point

A. Buzulutskov

Not working in an
(academic) vacuum,
but until now, noone
has put all the pieces
together...



Plus Nitrogen

- The addition of Nitrogen involves a complicated cascade of reactions, some of which emit in the IR and UV...
- BUT, decay times are very long (from $10\ \mu s$ to several ms), hence collisional quenching dominates and one can take excitation energy transfer to Nitrogen excimers as non-photonic.
- So, one could just add another loss term with a time constant equal to the diffusion time times the nitrogen concentration.