



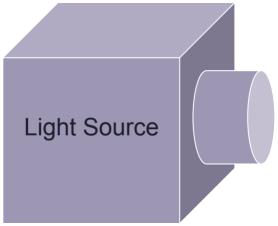






### Transition!

### No Transition



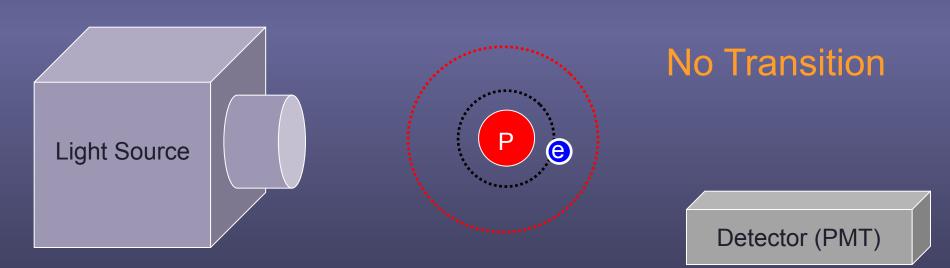






## How do we study atoms?

- Spectroscopy how light interacts with atoms
- Energy of a photon proportional to frequency (color); E=hv
- First, we excite an atom using light
- The atom then decays back to the ground state
- We detect fluorescence with a PMT



If the color is not right, we don't get a transition

# What's that frequency?

