

Engineering Yeast for Heavy Metal Waste Remediation

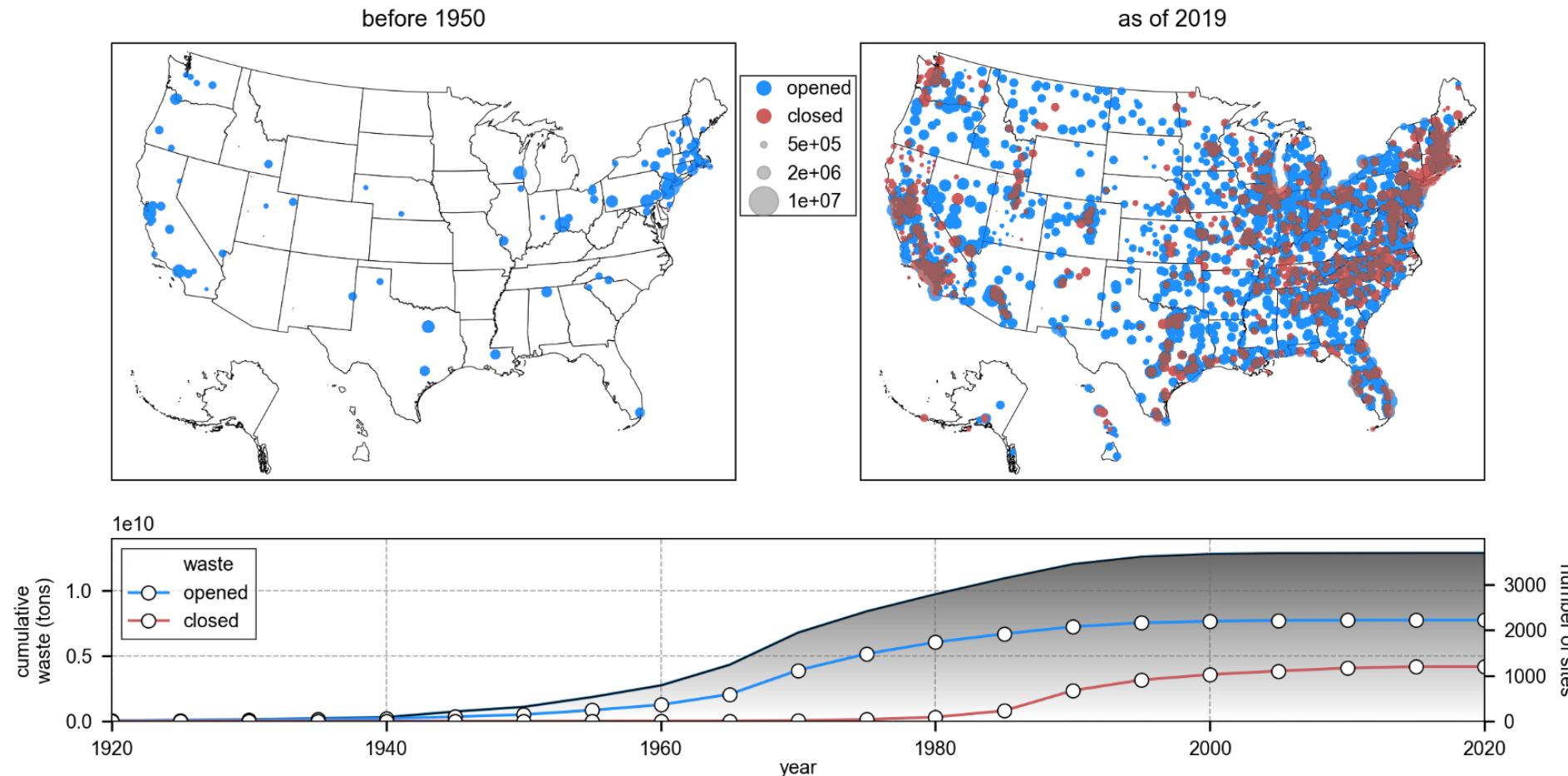


Talking about waste

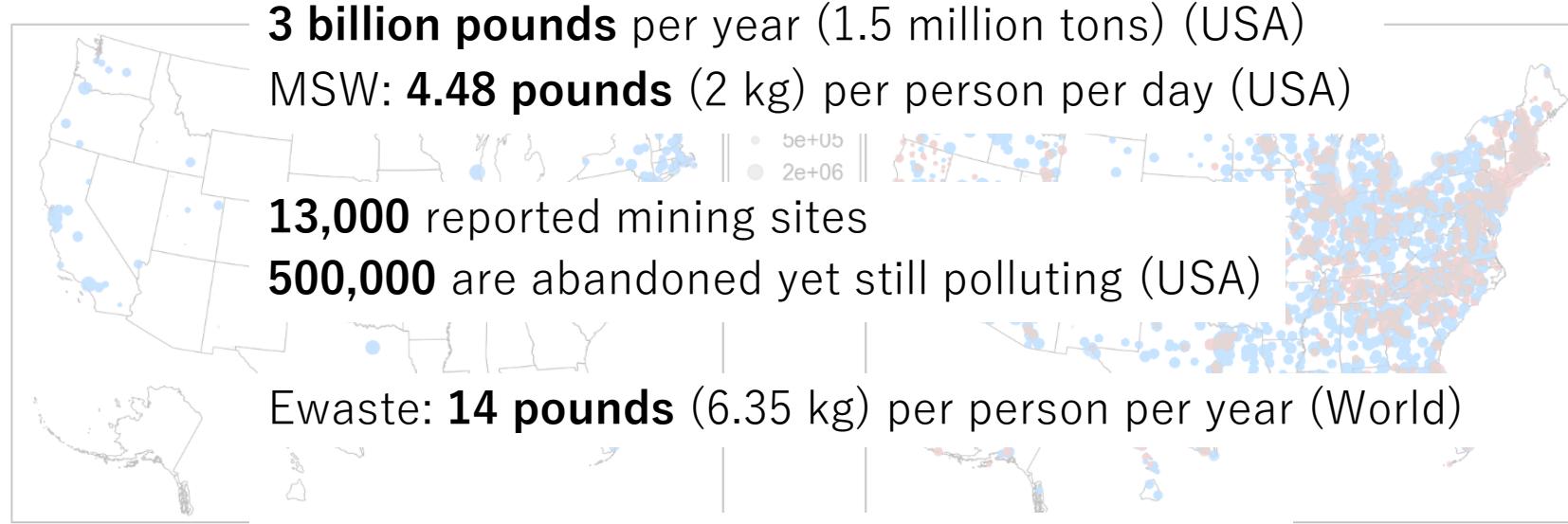


Who cares?

Talking about waste – by numbers



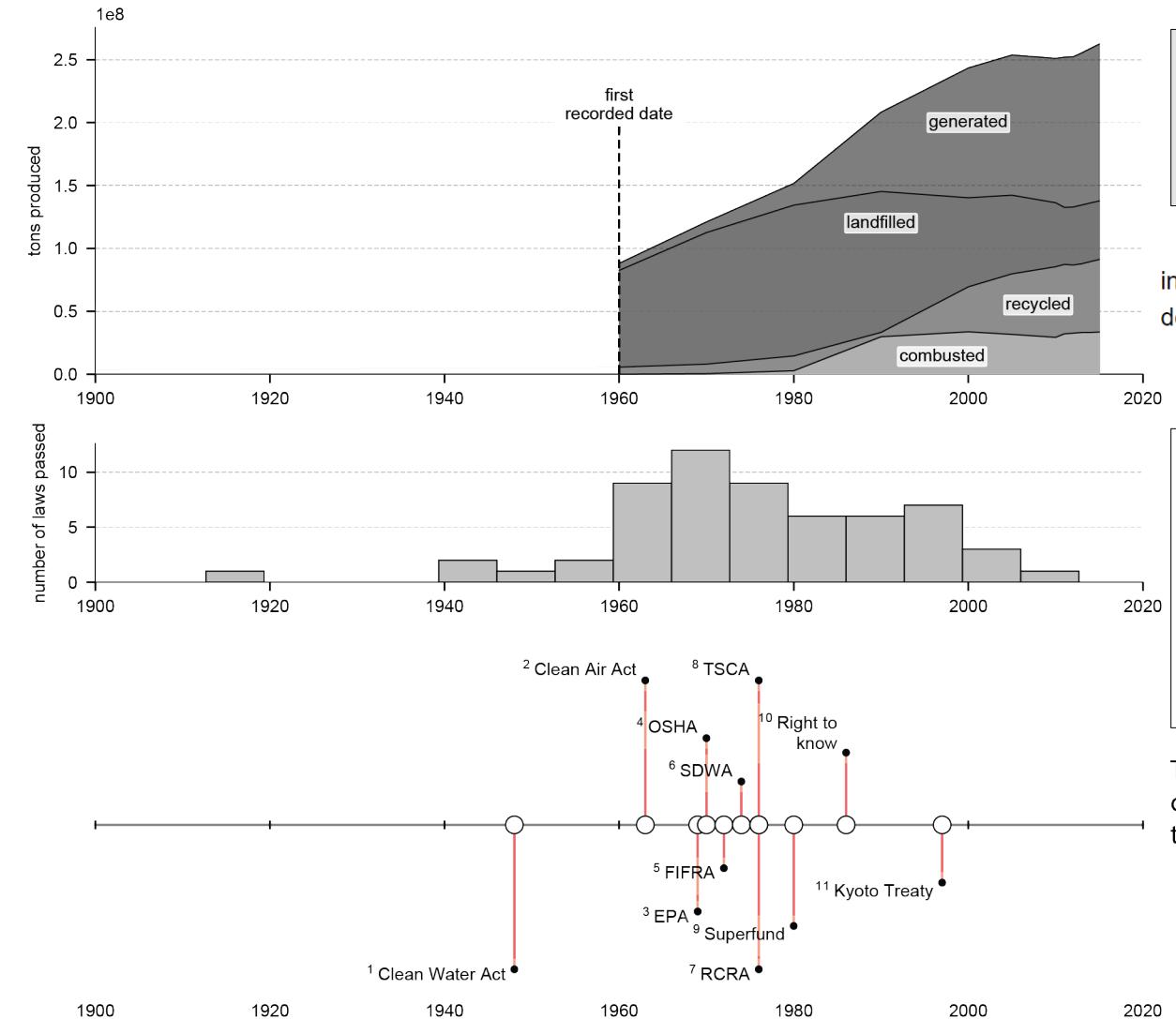
Talking about waste – by numbers



<https://www.epa.gov/facts-and-figures-about-materials-waste-and-recycling/national-overview-facts-and-figures-materials>

DeGraff, Jerome V. 2007. *Understanding and Responding to Hazardous Substances at Mine Sites in the Western United States*. Geological Society of America.

Talking about waste – by policy



American Environmental Policy since 1964 ↗

Richard N. L. Andrews

Subject: 20th Century: Post-1945, Political History, Environmental History | Online Publication Date: Feb 2018
DOI: 10.1093/acrefore/9780199329175.013.408

Beginning in the 1980s, the long-standing Progressive ideal of the “public interest” was increasingly supplanted by a narrative of “government overreach,” and the 1990s witnessed campaigns to delegitimize the underlying evidence justifying environmental policies by labeling it “junk science” or a “hoax.”

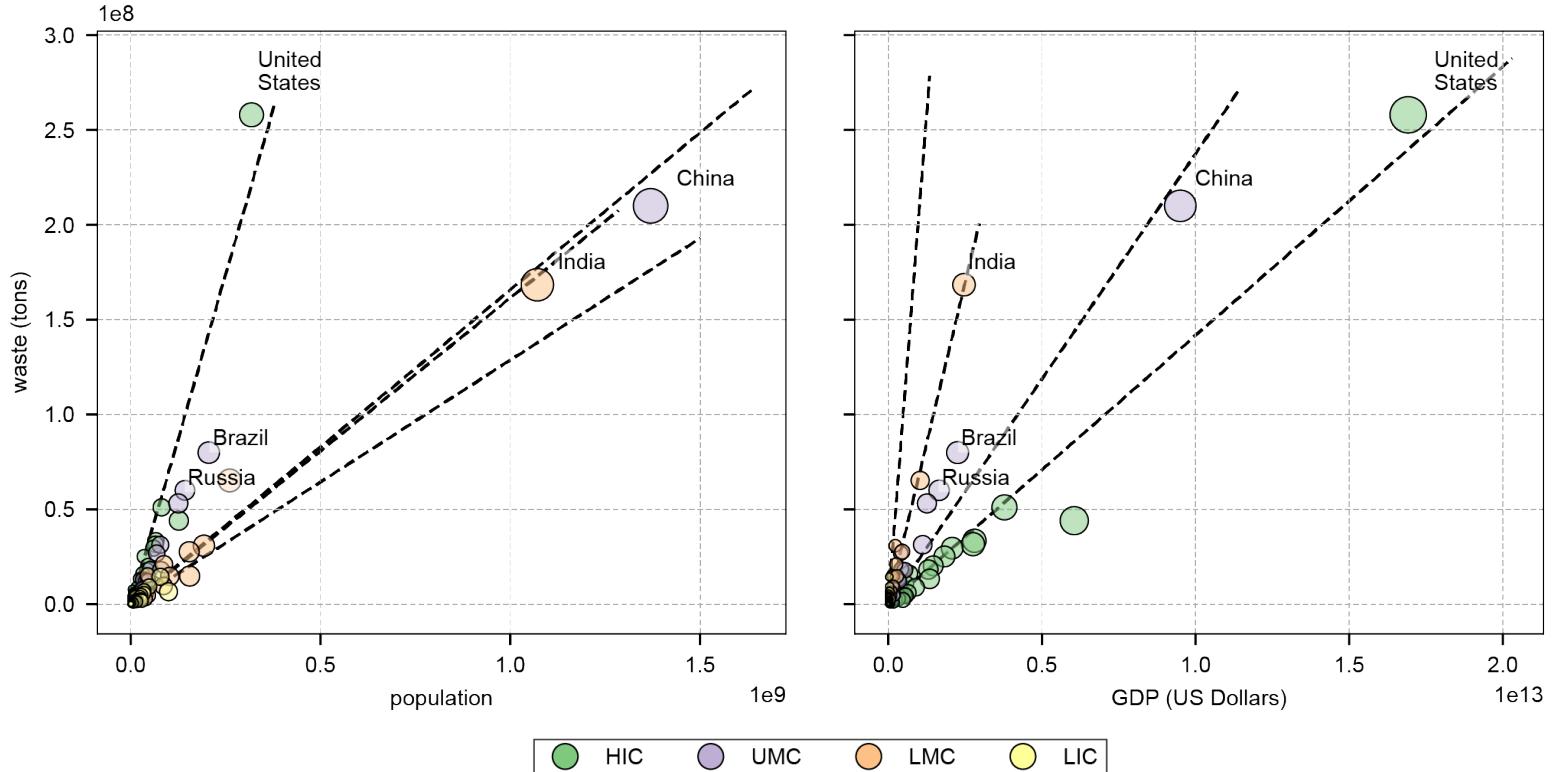
Does It Really Pay to Be Green? An Empirical Study of Firm Environmental and Financial Performance: An Empirical Study of Firm Environmental and Financial Performance

Andrew A. King, Michael J. Lenox ✉

First published: 08 February 2008 | <https://doi.org/10.1162/108819801753358526> | Cited by: 28

The relationship between underlying capabilities and environmental management is likely to be complex and contingent... It suggests that “When does it pay to be green?” may be a more important question than “Does it pay to be green?”

Talking about waste – using trends



- HIC = high income country
(e.g. USA)
- UMC = upper middle-income country
(e.g. China)
- LMC = lower middle-income country
(e.g. India)
- LIC = lower income country
(e.g. Zambia)

	tons per 1000 people	R ²
HIC	69.4	0.94
UMC	16.6	0.91
LMC	16.1	0.97
LIC	12.8	0.68

	tons per million \$	R ²
HIC	14.2	0.96
UMC	23.7	0.96
LMC	67.8	0.99
LIC	206	0.59

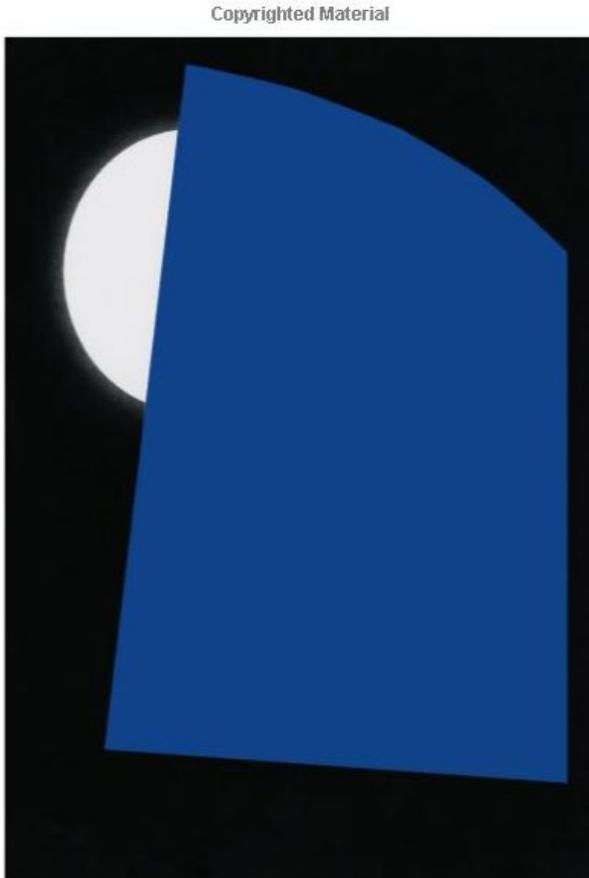
An explanation of waste production under the lens of societal and cultural changes

The Historical Nature of Cities: A Study of Urbanization and Hazardous Waste Accumulation

James R. Elliott^a and Scott Frickel^b

Major conclusions

- “Out of site, out of mind”
- “Someone else will do it”



Risk Society
Towards a New Modernity

Ulrich Beck

“…a society increasingly preoccupied with the future (and also with safety), which generates the notion of risk”

“…risk society; that is, a condition wherein the growth of knowledge has everywhere created ‘manufactured uncertainty’ (for example the risk of ecological disaster), so that scientific expertise is increasingly called upon to alleviate the effects of earlier applications of science.”

We have trash in space

Who cares?



<https://www.bbc.com/news/av/science-environment-38091707/time-to-take-out-the-space-trash>

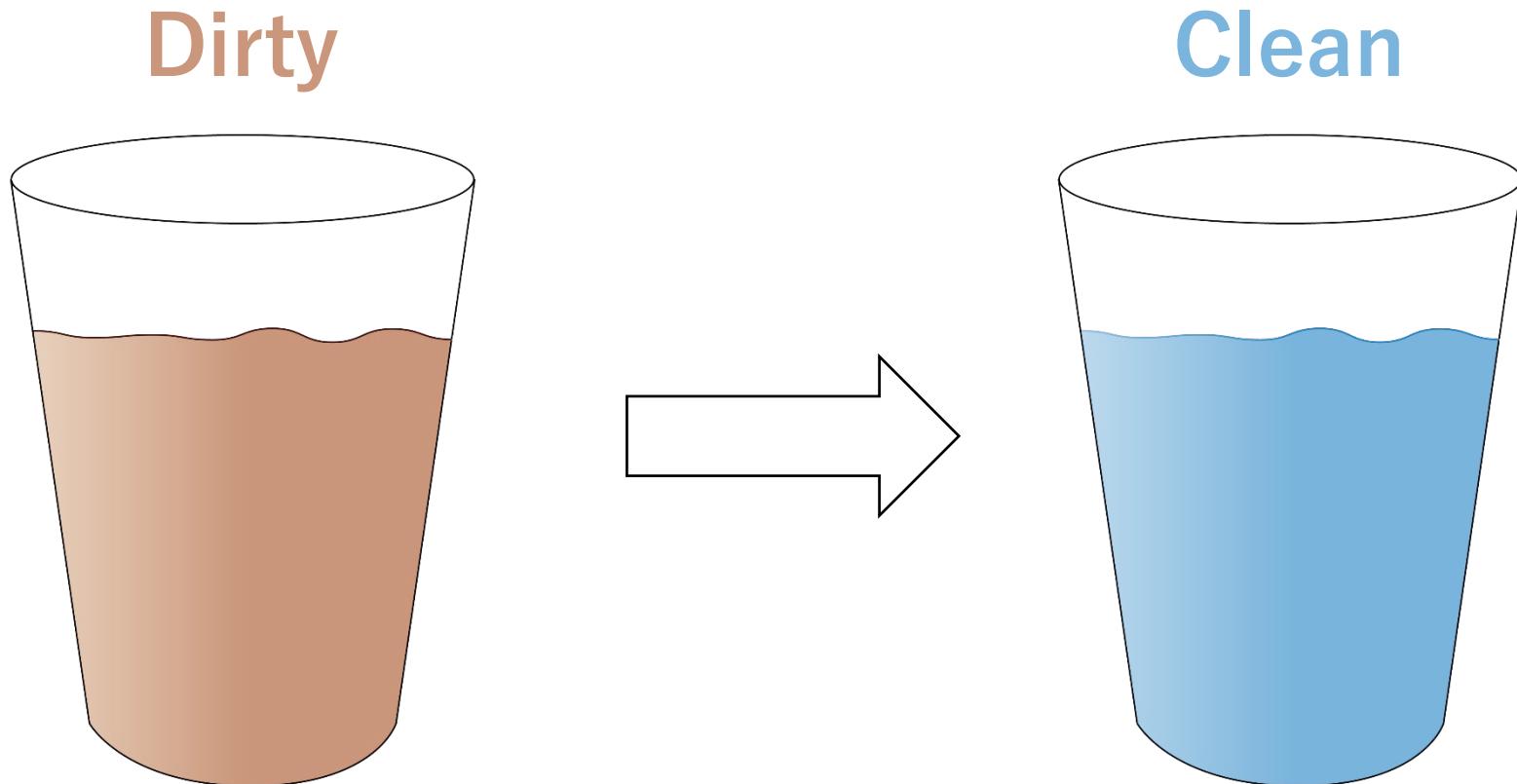
Who should care?

the engineers and scientist

Who should care?

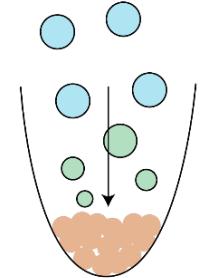
more people should try

The goal – clean water



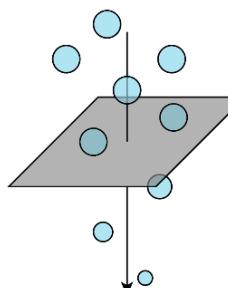
Current remediation strategies rely on physicochemical methods

Chemical
Precipitation



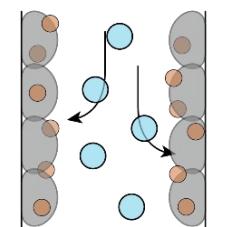
React

Absorption



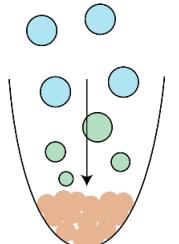
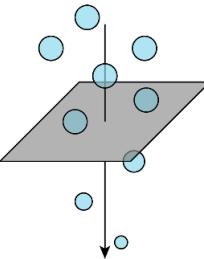
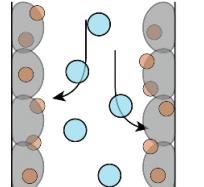
Absorb

Ion-exchange



Bind

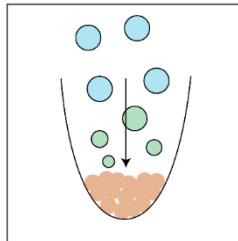
Cost, specificity, and technology accessibility are limitations for adoption

	specificity	re-usability	cost to make	cost to operate	
Chemical Precipitation		LOW	NO	LOW	HIGH
Absorption		LOW	NO	VARIABLES	VARIABLES
Ion-exchange		VARIABLES	YES	HIGH	HIGH

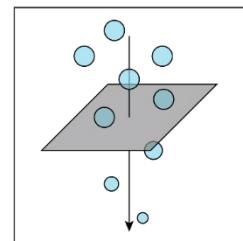
Designing biological analogies to physicochemical processes

Synthetic

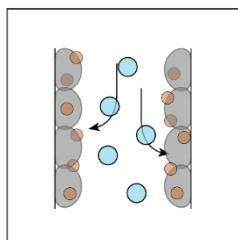
Chemical precipitation



Absorption



Ion-exchange



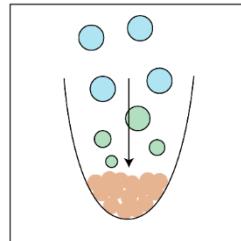
Biological

- Sustainable
- Scalable
- Autonomous
- Genetically encoded

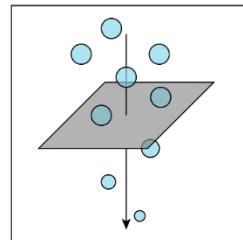
Designing biological analogies to physicochemical processes

Synthetic

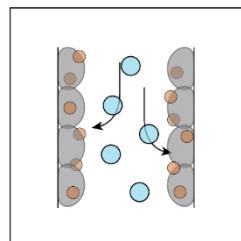
Chemical precipitation



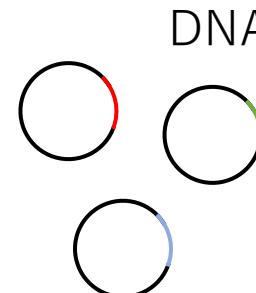
Absorption



Ion-exchange

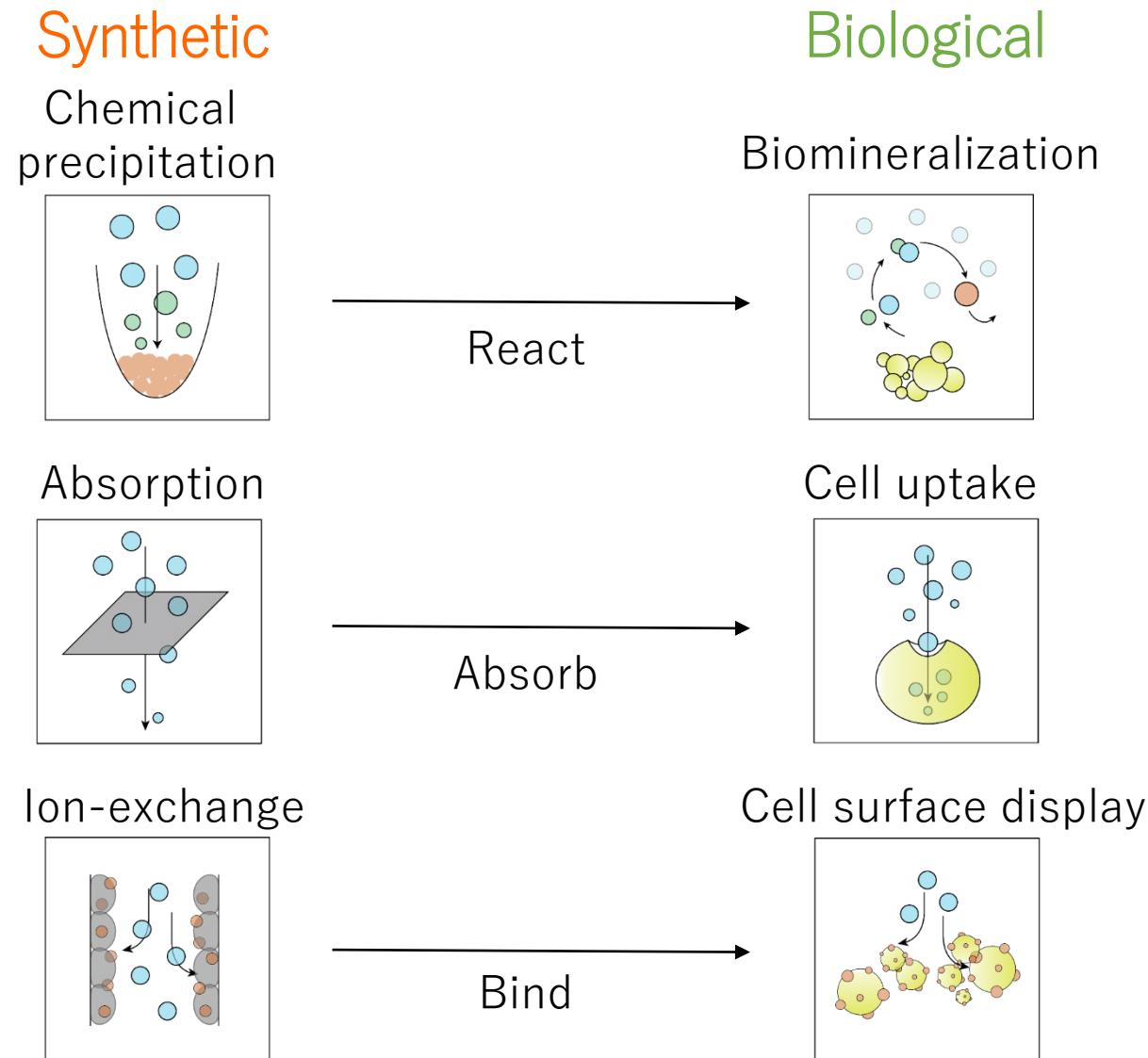


Biological

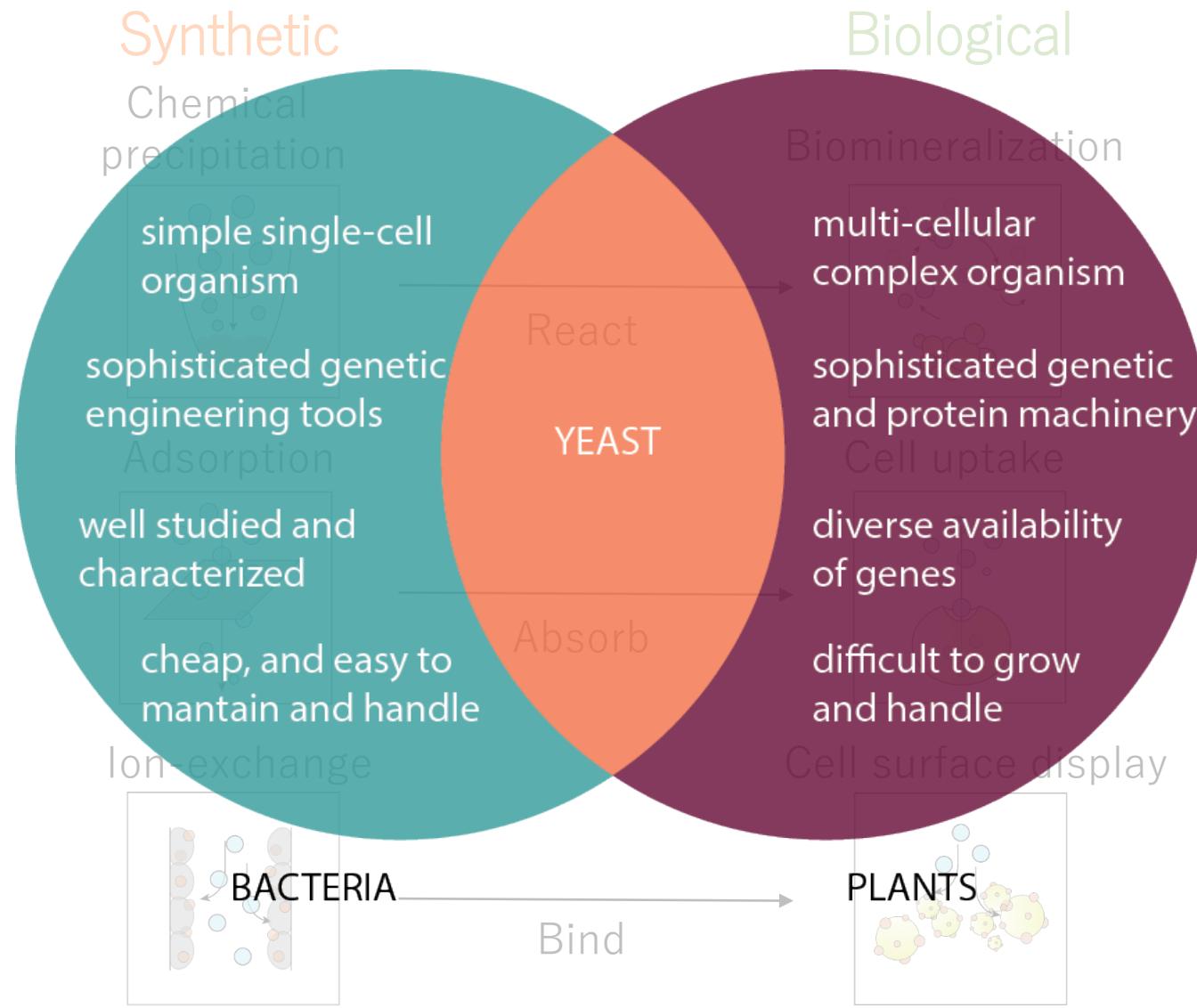


FUNCTION

Designing biological analogies to physicochemical processes



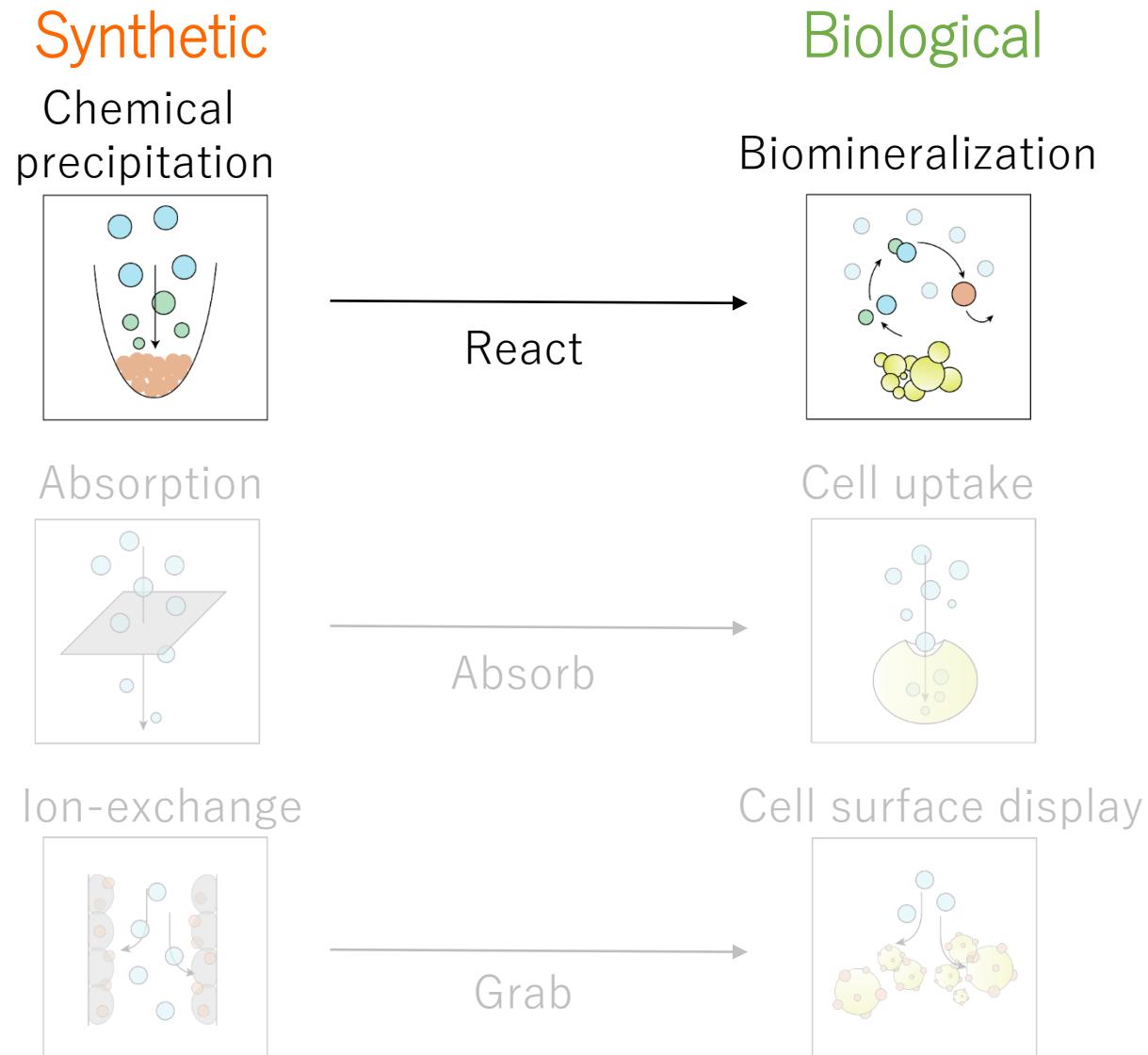
Designing biological analogies to physicochemical processes





Why use yeast?

Biomineralization as an analogy to chemical precipitation



Biomineralization as an analogy to chemical precipitation



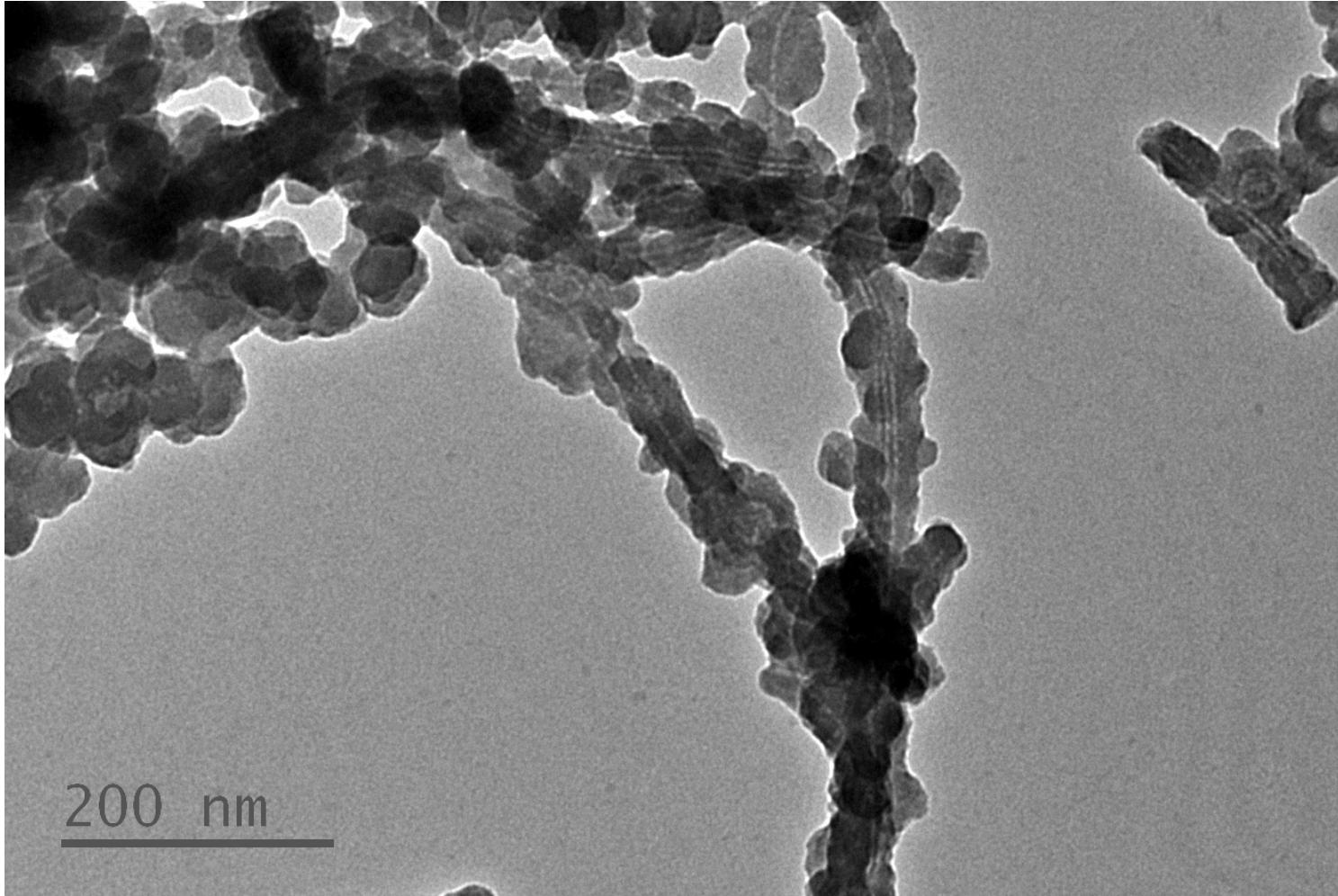
pH > 8

sacrificial chemicals:

- Lime (CaOH)
- Iron (Fe(III))
- Sulfides (H_2S)

production of sludge

Biomineralization as an analogy to chemical precipitation



rarely occurs

highly controlled for specific tissues (bones, shells, etc.)

sacrificial chemicals:

- phosphates (PO_4^{3-})
- carbonates (CO_3^{2-})
- sulfates (SO_4^{2-})

Inspiration for chemical precipitation from the wine industry

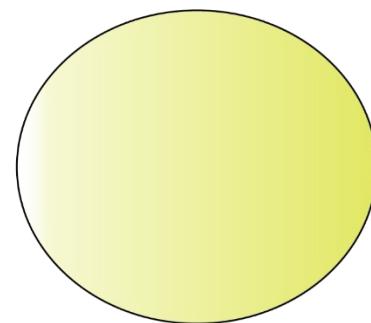


poor winemaking generated odorous and distasteful wines

chemical culprit:

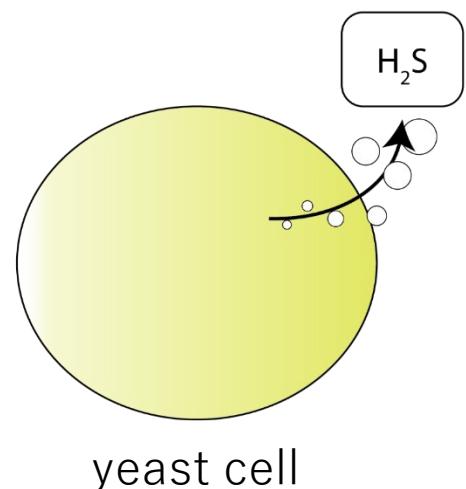
- hydrogen sulfide (H_2S)

Engineering H₂S production in yeast

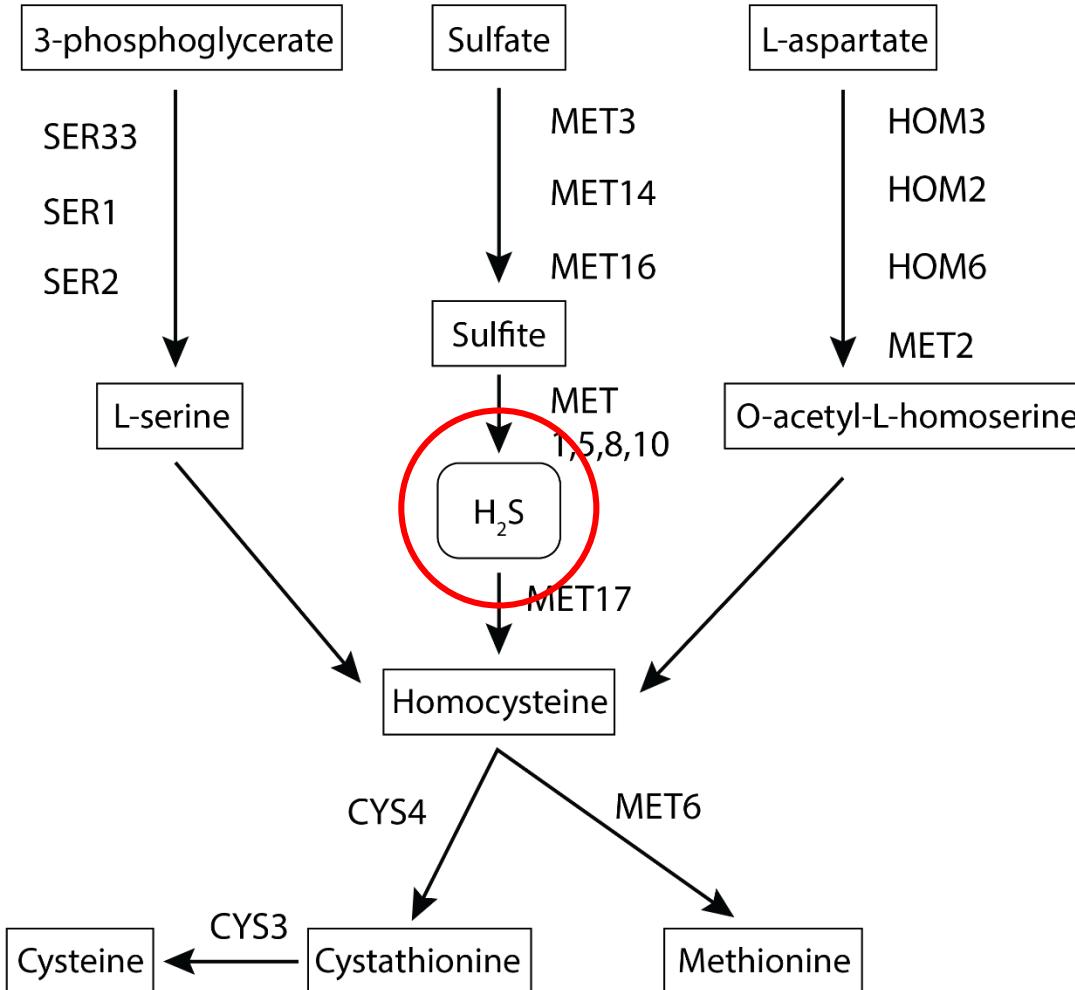


yeast cell

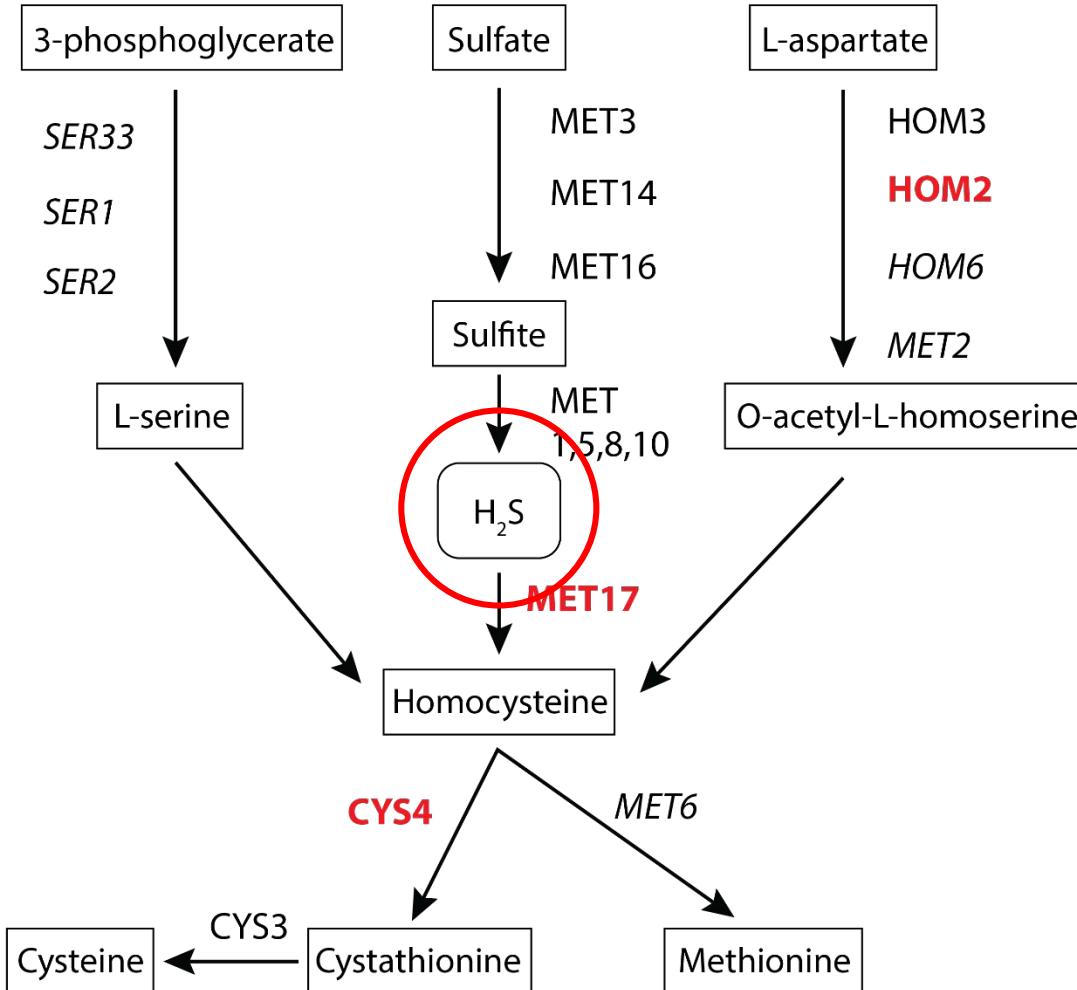
Engineering H₂S production in yeast



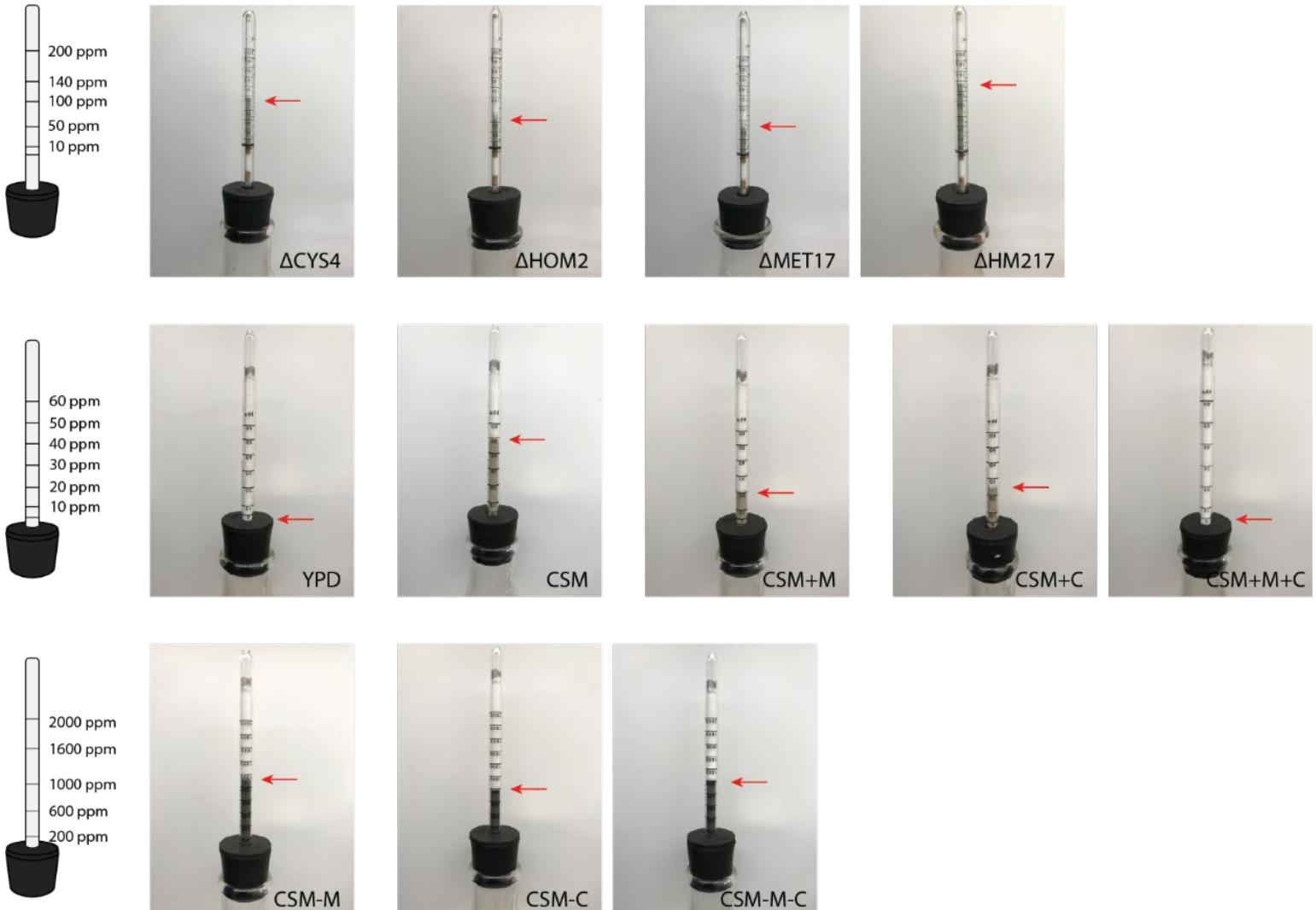
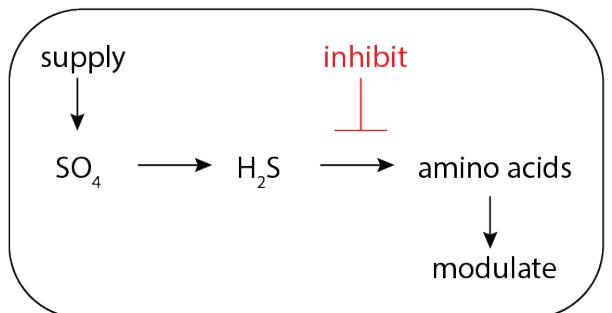
Engineering H₂S production in yeast



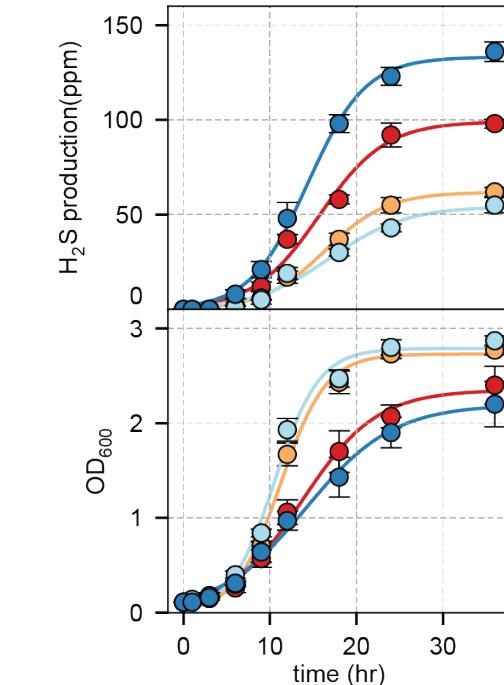
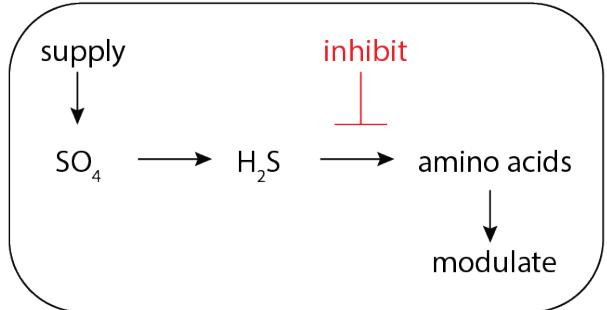
Engineering H₂S production in yeast



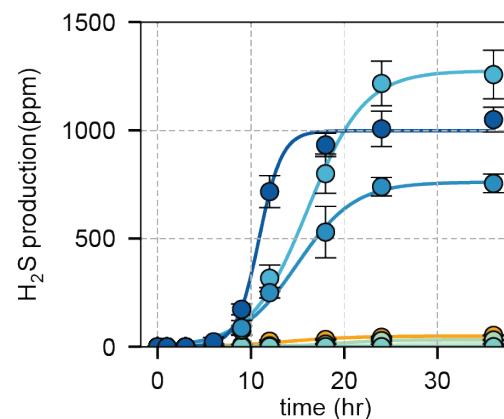
Quantifying and controlling H₂S production using simple culturing methods



Controlling yeast H₂S production with different knockouts and culture conditions

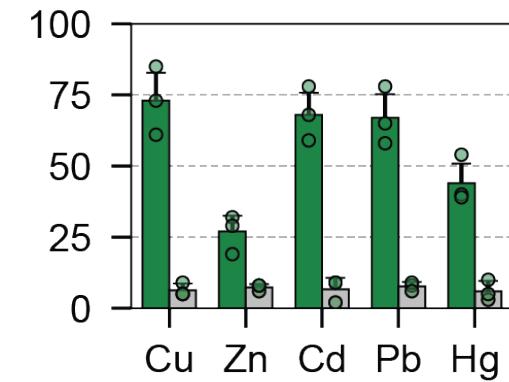
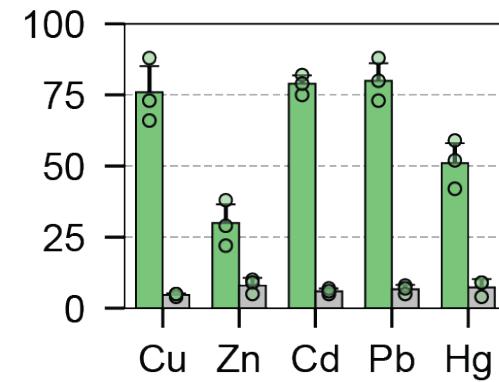
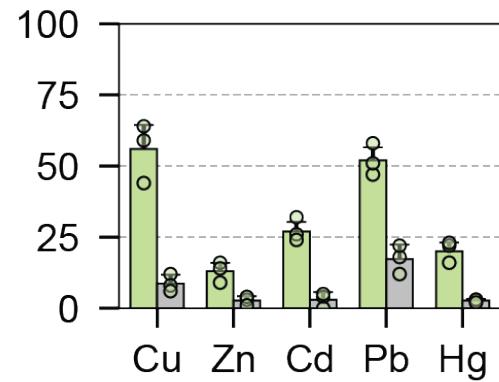
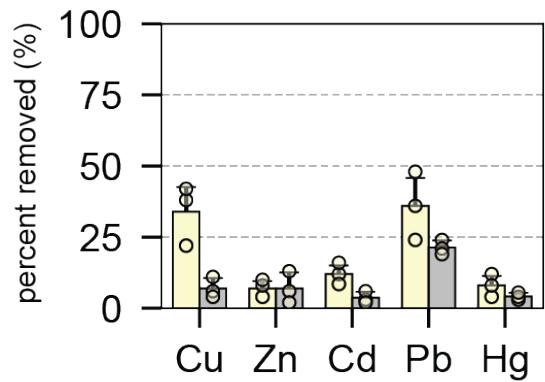
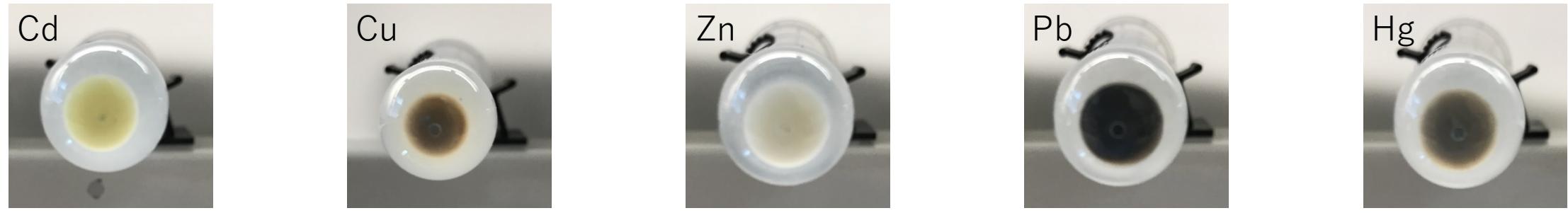


	A (ppm)	t _{1/2} (hr)	r (ppm hr ⁻¹)
CYS4	99 ± 3.0	16 ± 0.48	2.36 ± 0.22
HOM2	62 ± 2.5	16 ± 0.66	1.57 ± 0.21
MET17	54 ± 4.5	17 ± 0.94	1.12 ± 0.13
HM217	133 ± 2.4	14 ± 0.46	3.62 ± 0.36



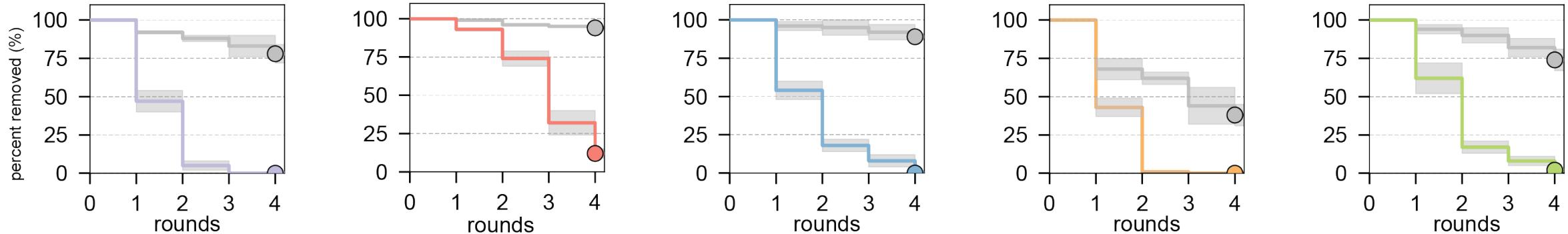
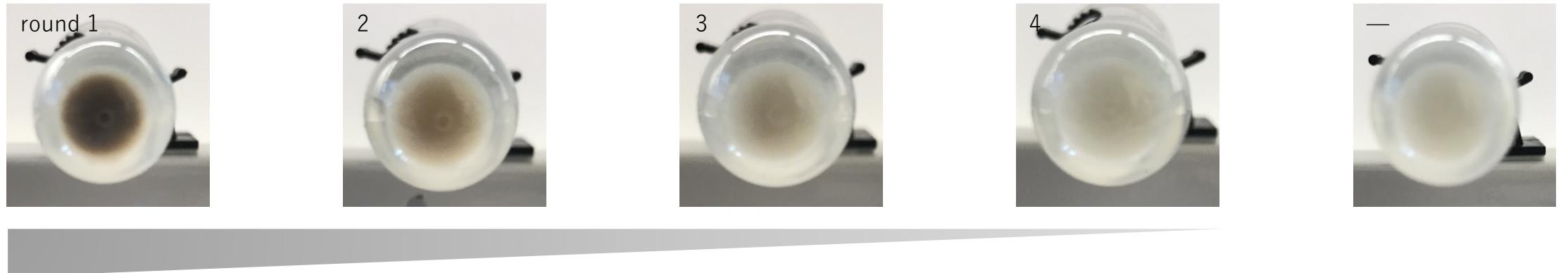
	A (ppm)	t _{1/2} (hr)	r (ppm hr ⁻¹)
YPD	0 ± 0	0 ± 0	0 ± 0
CSM	50 ± 7.4	14 ± 2.3	1.26 ± 0.54
CSM+M	12 ± 6.4	23 ± 1.0	0.31 ± 0.82
CSM+C	32 ± 4.2	19 ± 1.3	1.07 ± 0.70
CSM+M+C	0 ± 0	0 ± 0	0 ± 0
CSM-M	1275 ± 80	16 ± 2.0	38.2 ± 7.51
CSM-C	761 ± 102	15 ± 1.4	22.4 ± 5.12
CSM-M-C	998 ± 92	11 ± 4.2	74.9 ± 17.7

Removing heavy metals using H_2S generated from yeast

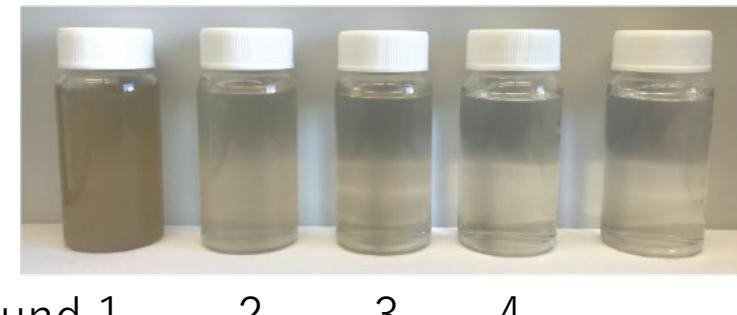
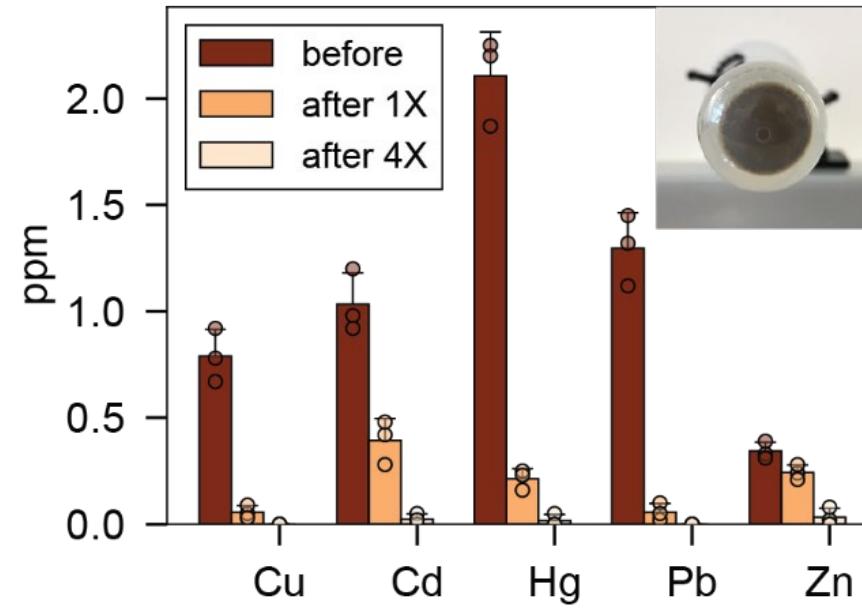
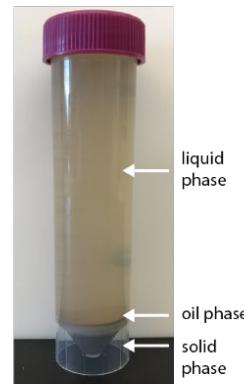


■ YPD ■ CSM ■ CSM-M ■ CSM-C ■ ctrl

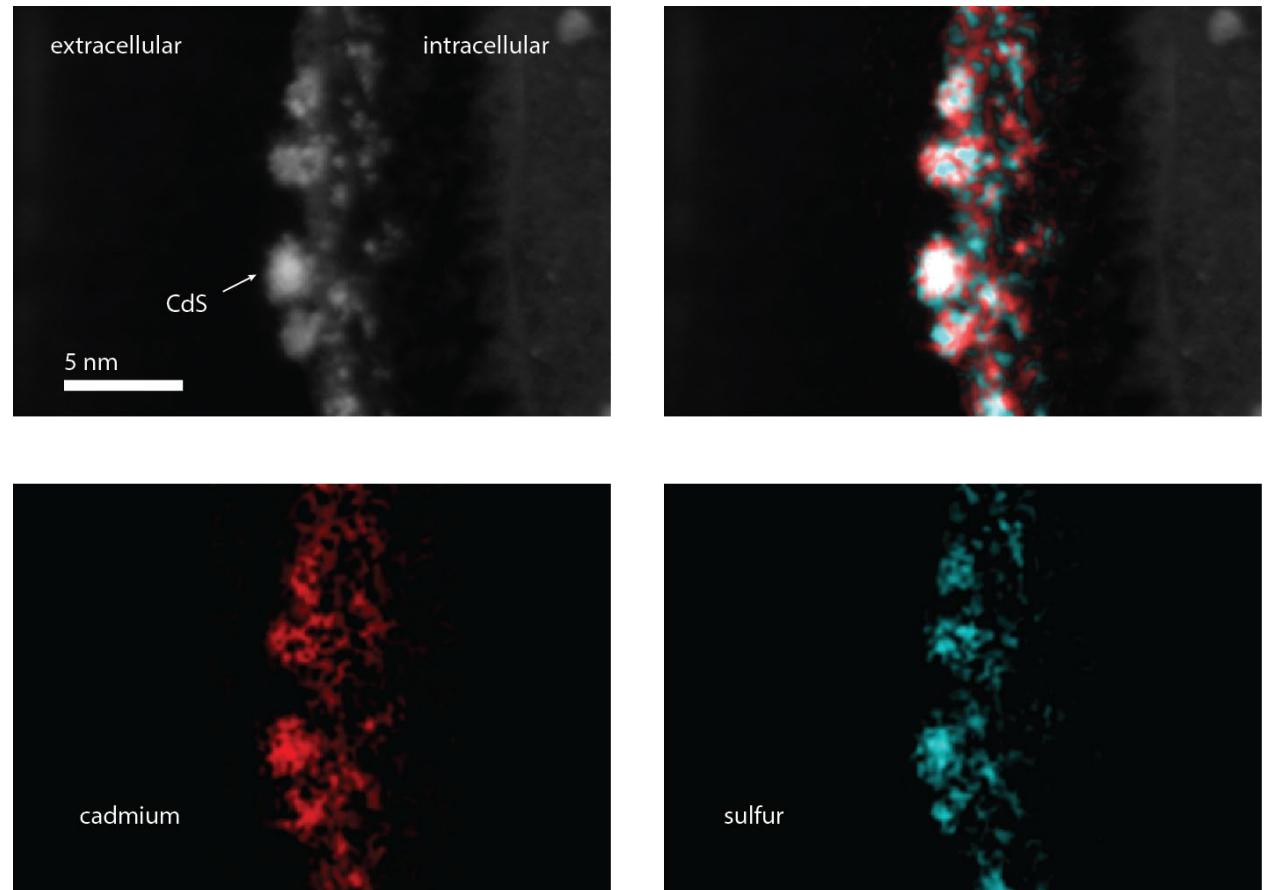
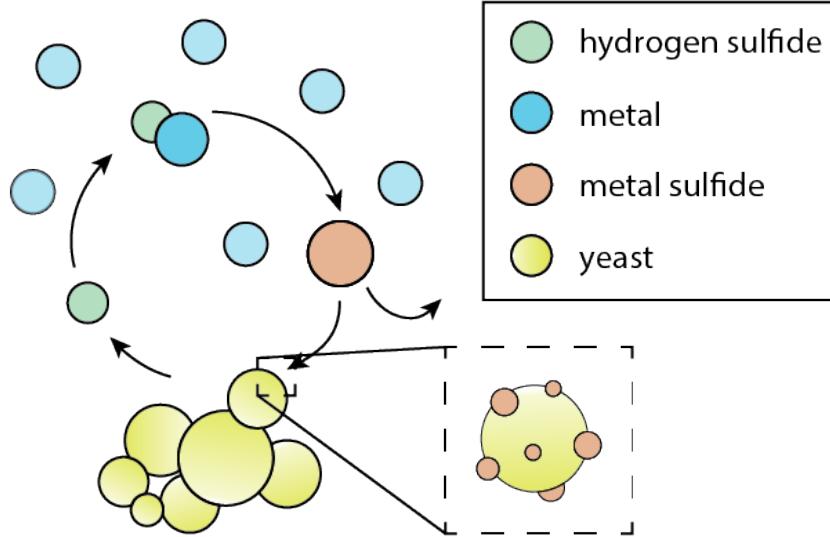
Multiple rounds of yeast mediated H₂S precipitation can completely eliminate metal



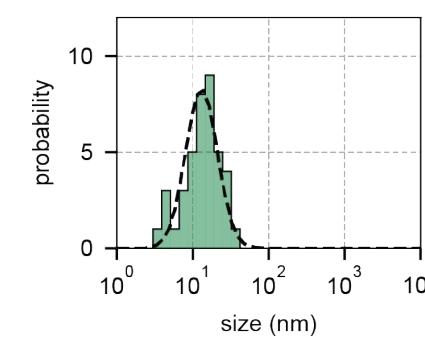
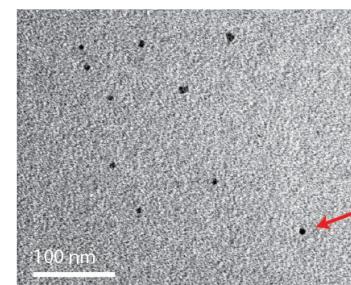
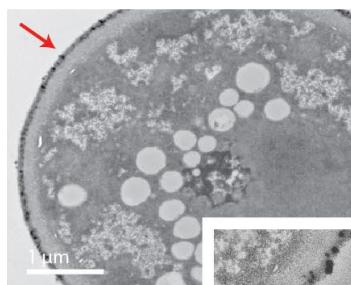
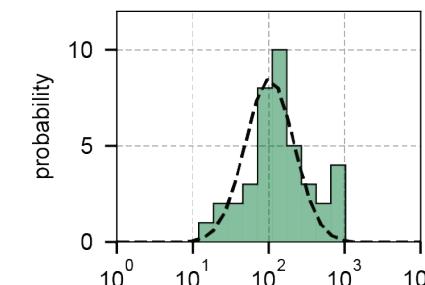
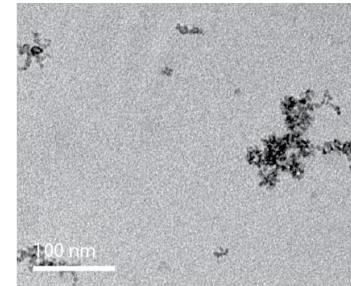
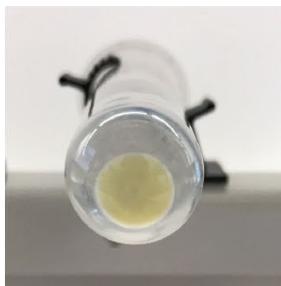
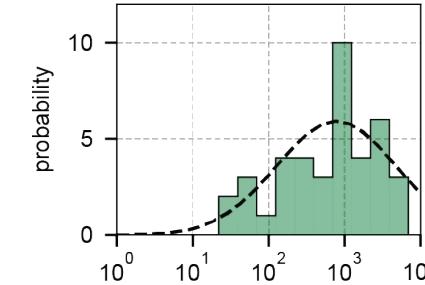
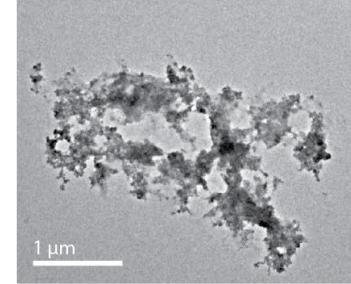
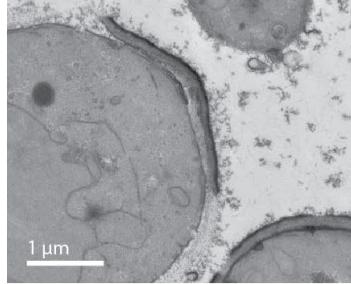
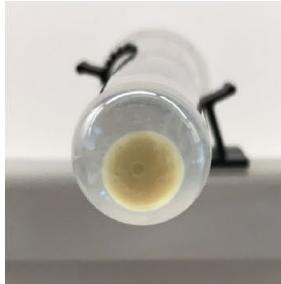
Removing heavy metals from the Canadian Athabasca Oil Sands



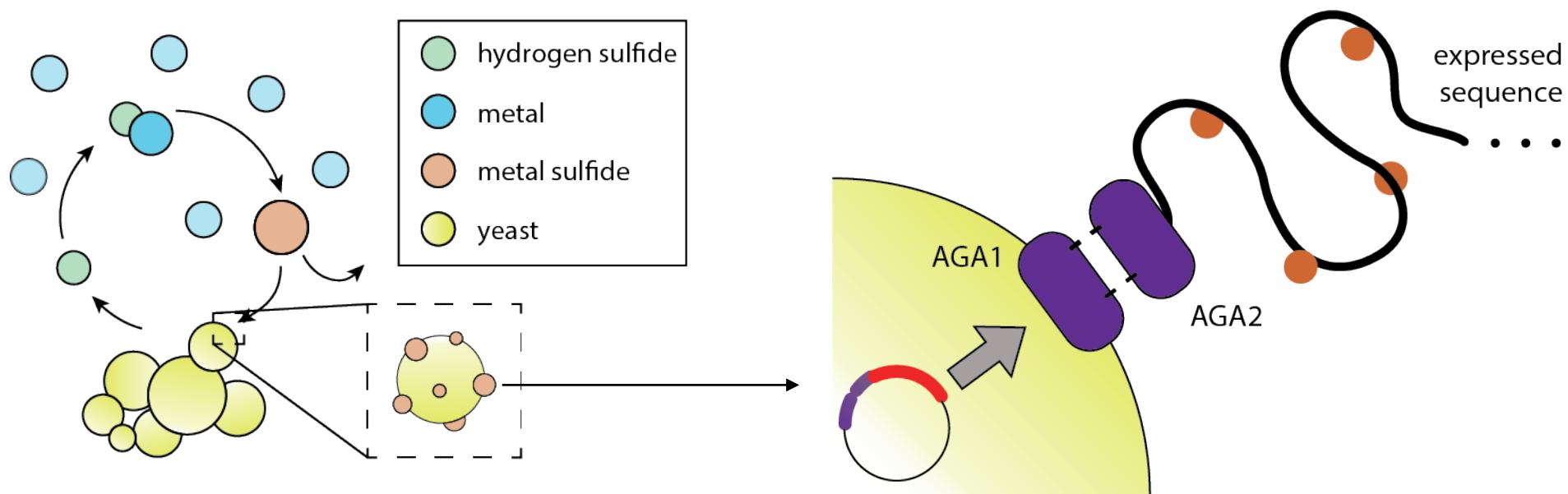
Controlling metal sulfide precipitation based on sulfur production rate and yeast display



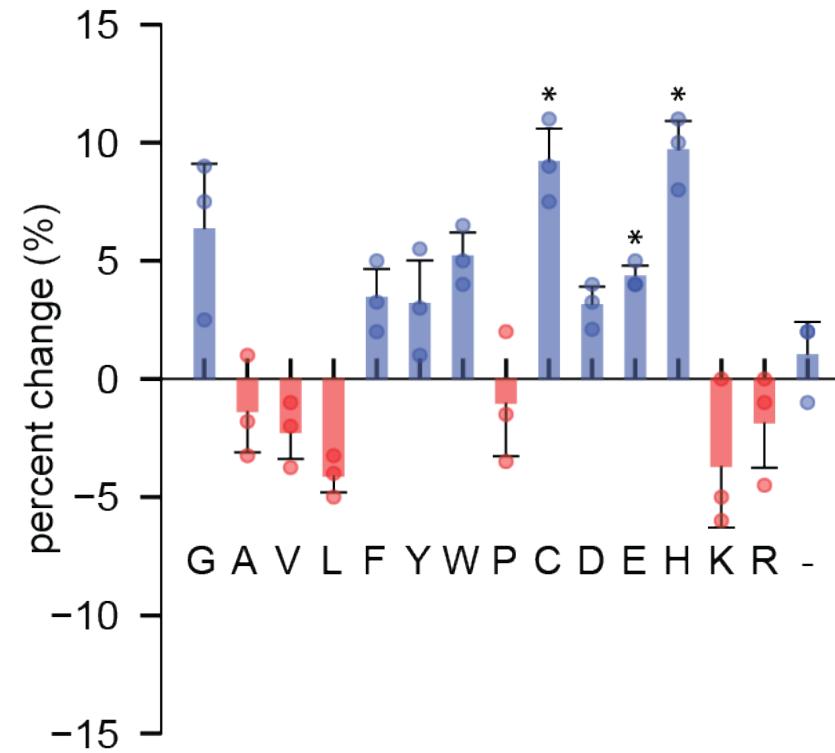
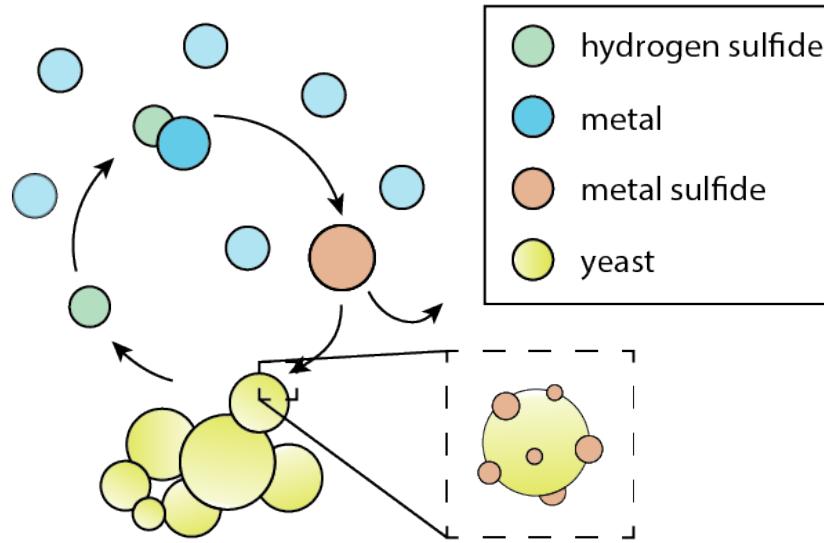
Controlling H₂S production rates improves metal sulfide size distribution formation



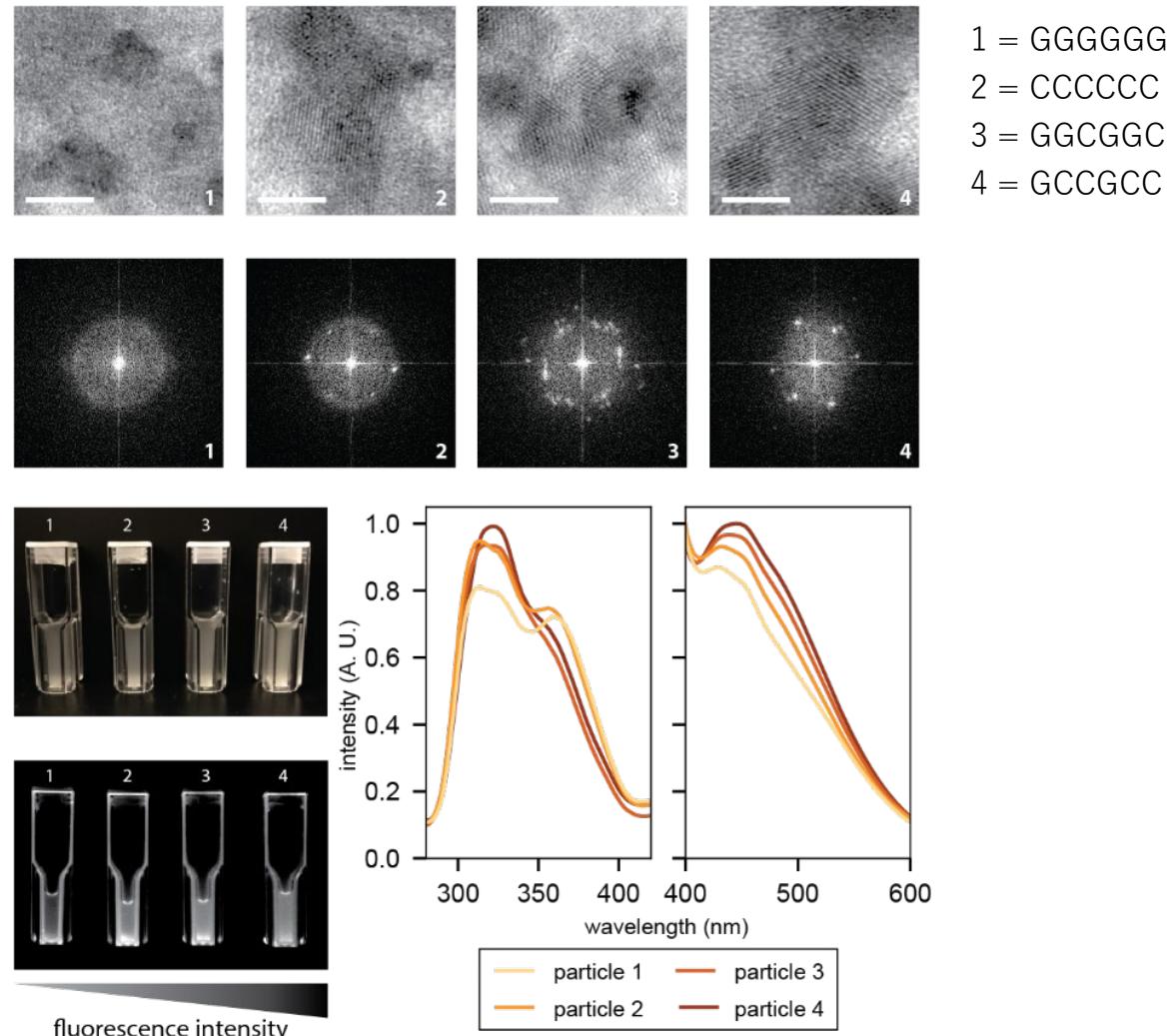
Controlling metal sulfide precipitation based on sulfur production rate and yeast display



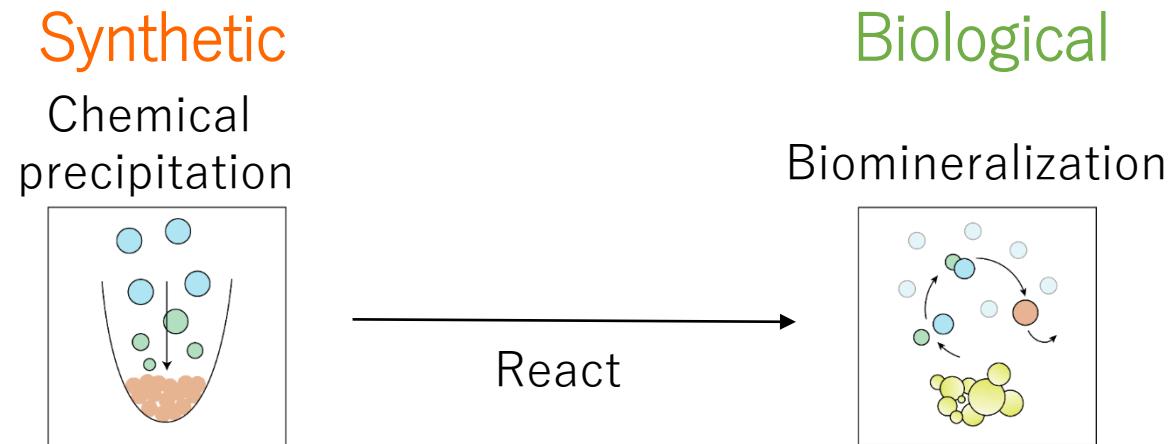
Controlling metal sulfide precipitation based on sulfur production rate and yeast display



Improving metal sulfide morphology and crystallinity using yeast display



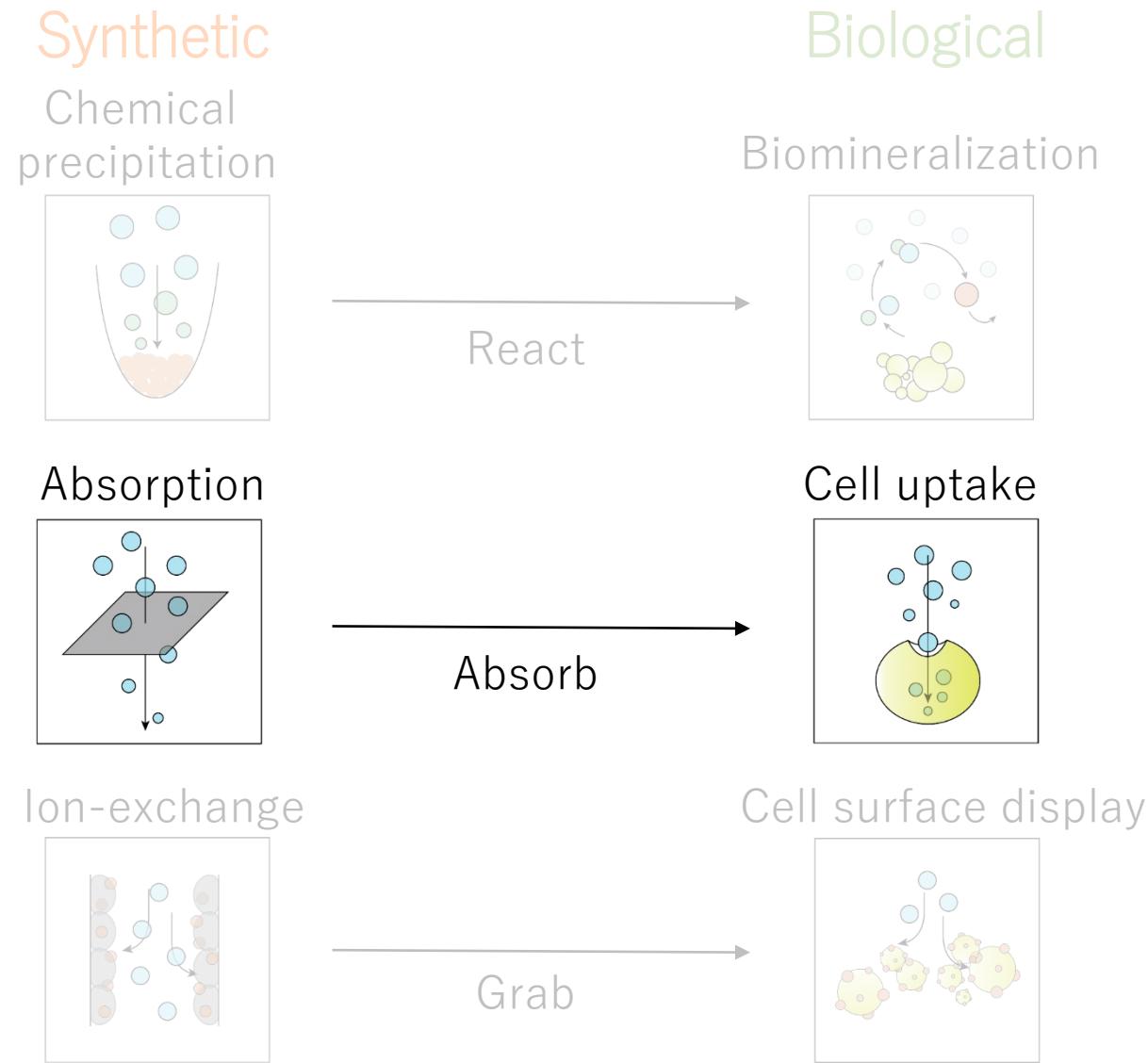
Takeaway



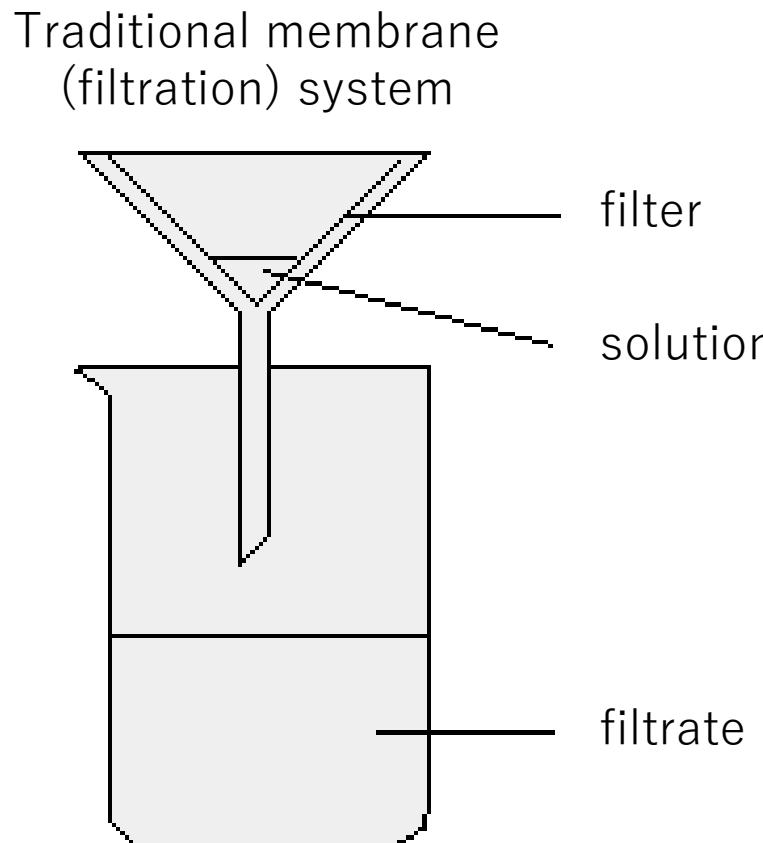
TAKEAWAY

- Engineering S²⁻ production
- Knockouts and culture conditions tune H₂S production
- Metal precipitation of Cu, Zn, Cd, Hg, and Pb > 50% for 1 round
- Controlled metal sulfide particle formation, compared to production of sludge

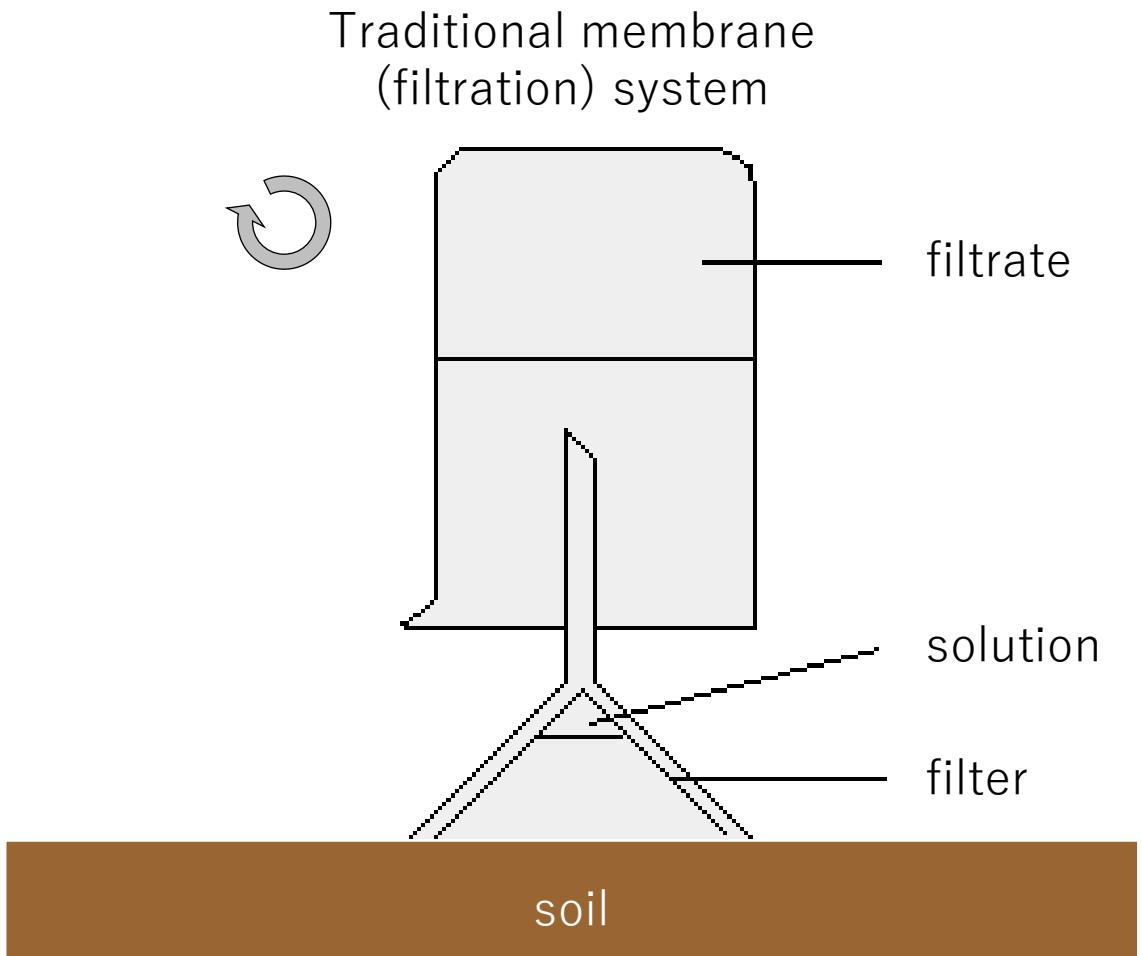
Cellular metal uptake as an analogy to absorption



Absorption requires a filter and a sequestration body to retain contaminants



Absorption requires a filter and a sequestration body to retain contaminants



Plant (phyto-) remediation

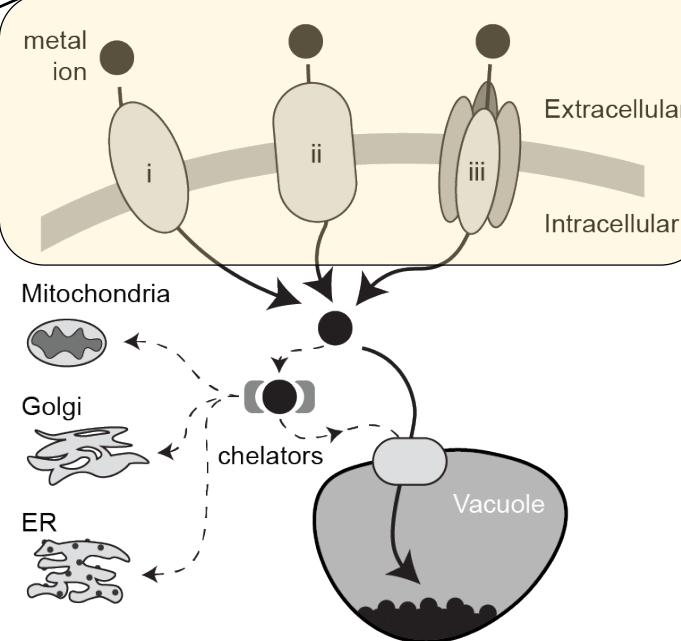


Hyperactive membrane transporters confer hyperaccumulation in plants, and yeast.

Plants



Gatekeepers: metal transporters

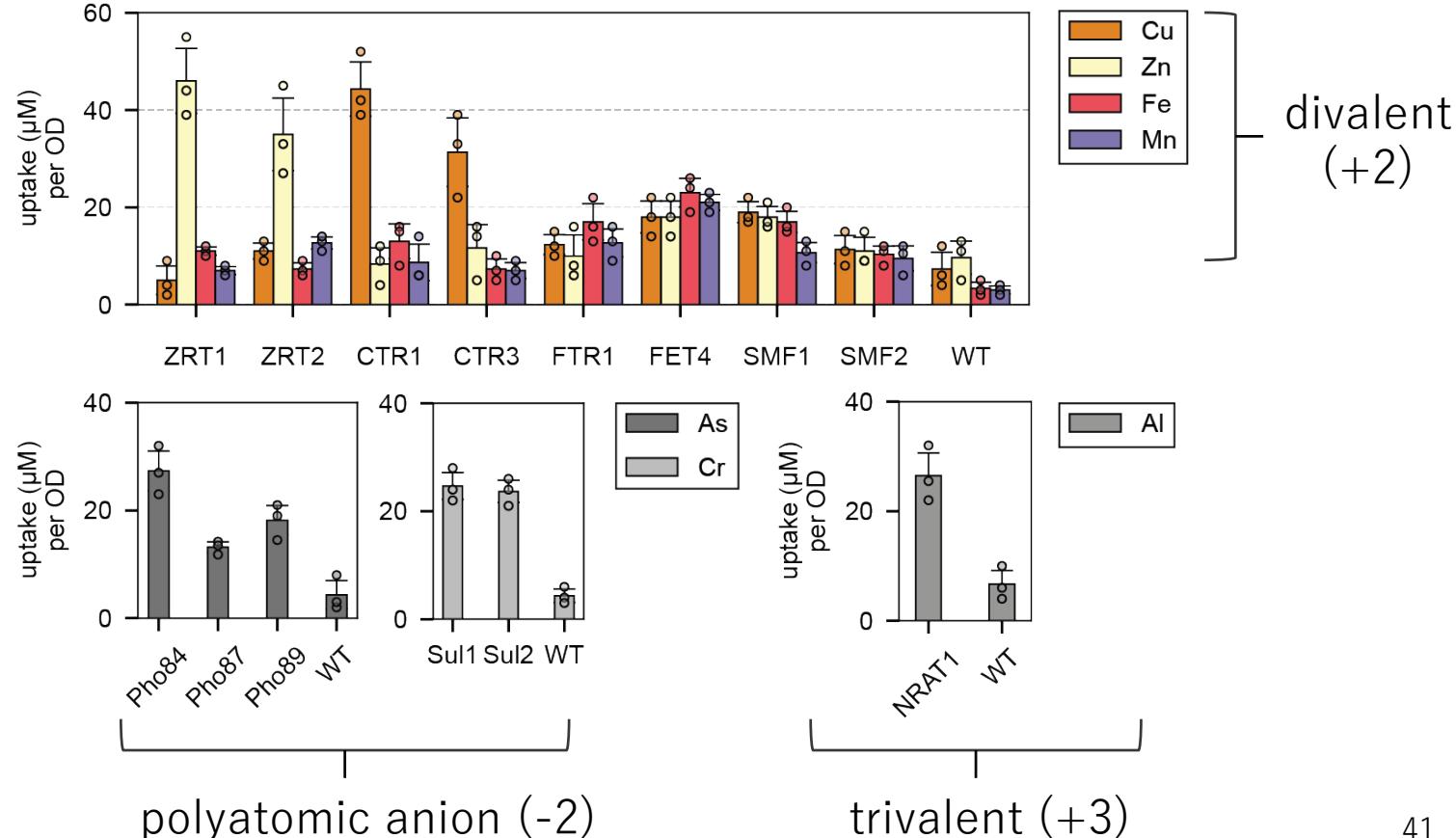


i = native transporters

ii = permeases

iii = non-native transporters

Yeast



polyatomic anion (-2)

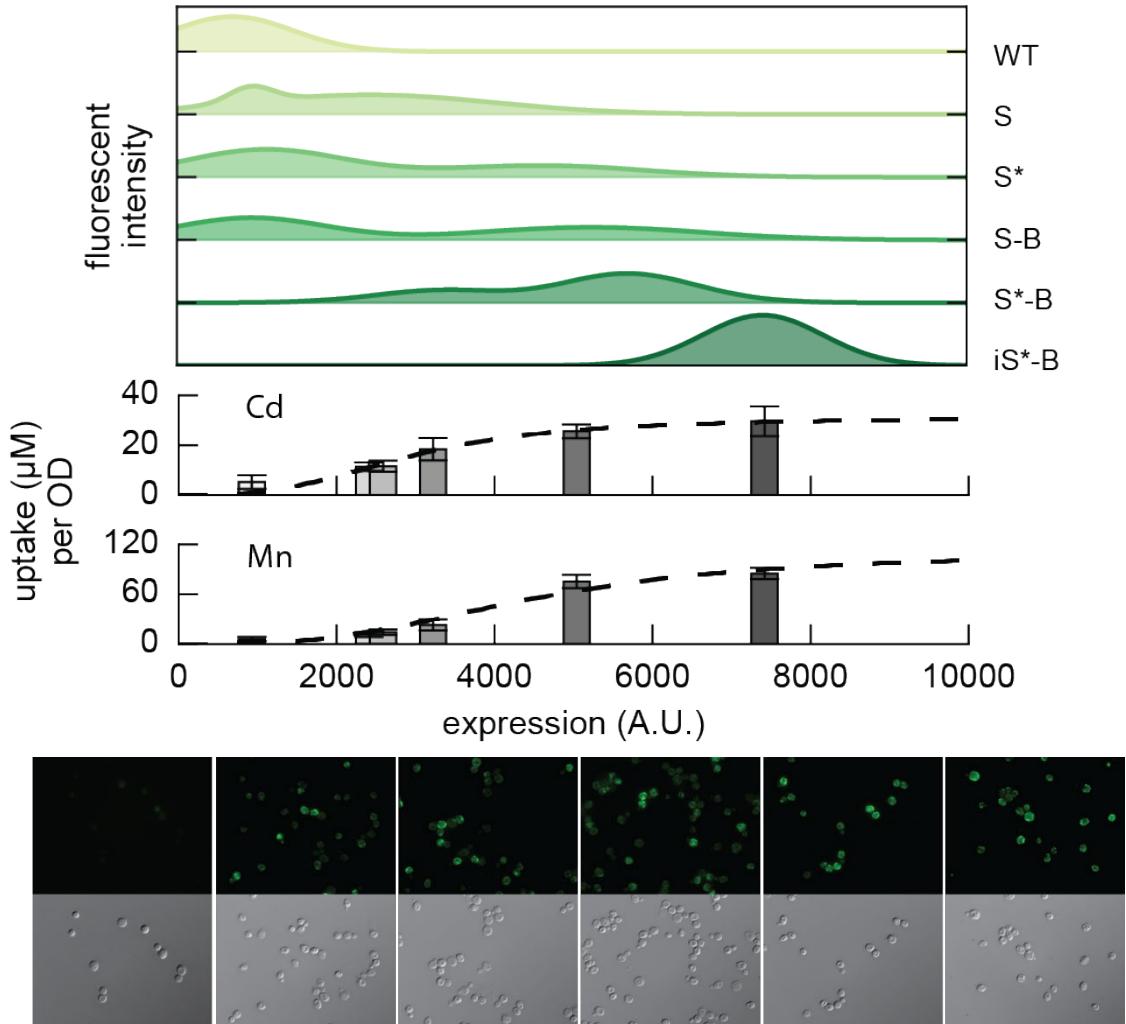
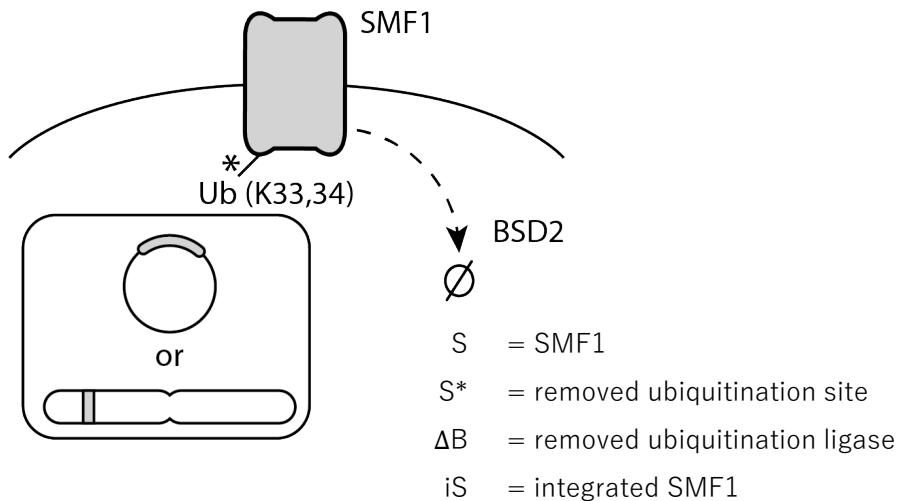
trivalent (+3)

Establishing engineering requirements for enhancing metal uptake

1. Overexpress transporter
2. Increase expression longevity
3. Access other compartments
(i.e. vacuole)
4. Increase metal tolerance

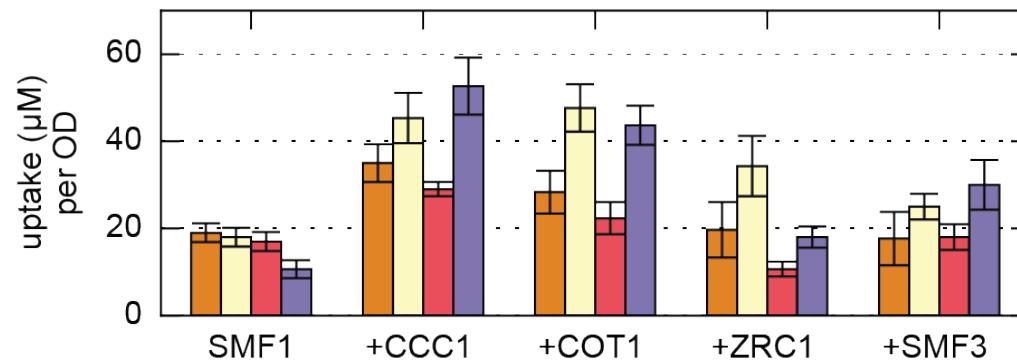
Increasing membrane transporter expression

1. Overexpress transporter
2. Increase expression longevity
3. Access other compartments
(i.e. vacuole)
4. Increase metal tolerance

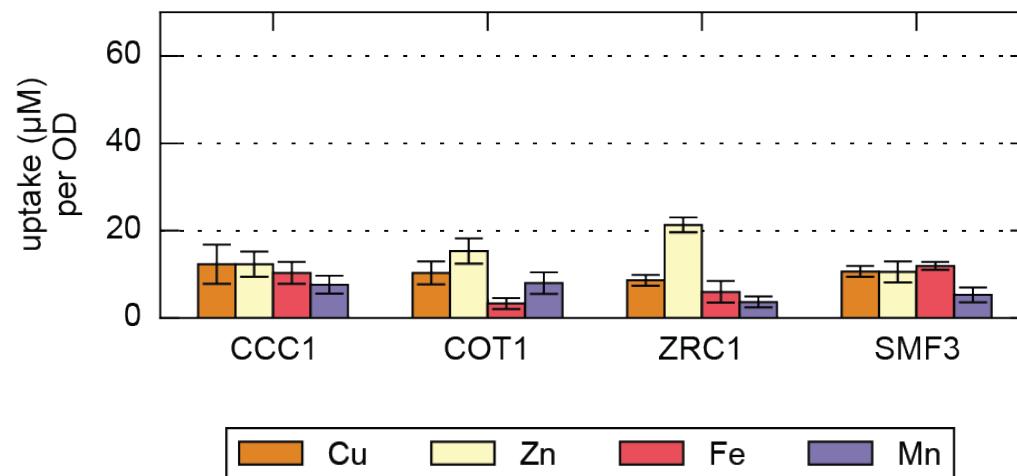


Metals can be further compartmentalized in vacuoles

1. Overexpress transporter
2. Increase expression longevity
3. Access other compartments (i.e. vacuole)
4. Increase metal tolerance



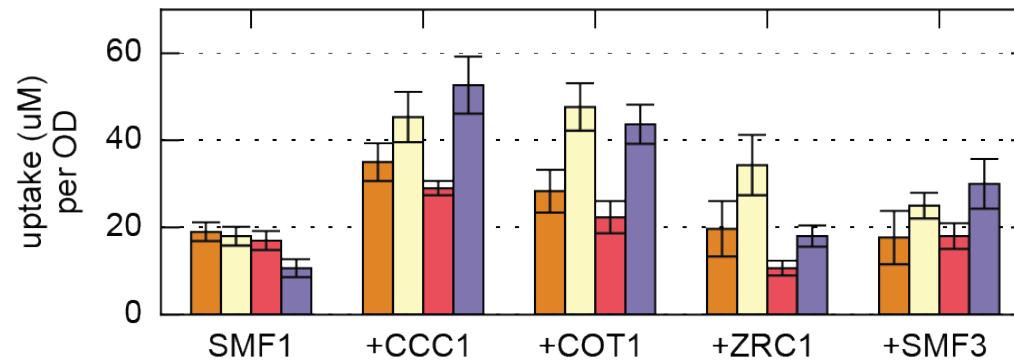
overexpression
of vacuole
transporter
= ↑ uptake
capacity



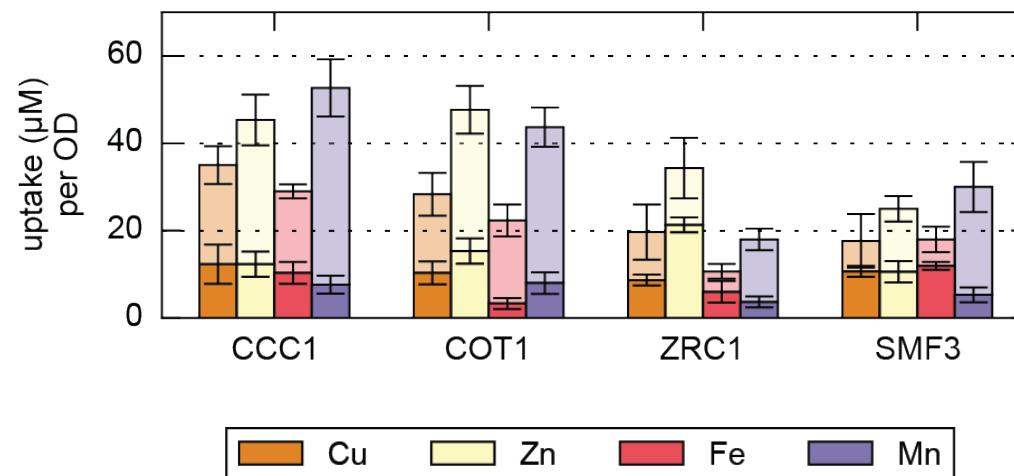
surface
transport is the
limiting
“reaction”

Metals can be further compartmentalized in vacuoles

1. Overexpress transporter
2. Increase expression longevity
3. Access other compartments (i.e. vacuole)
4. Increase metal tolerance



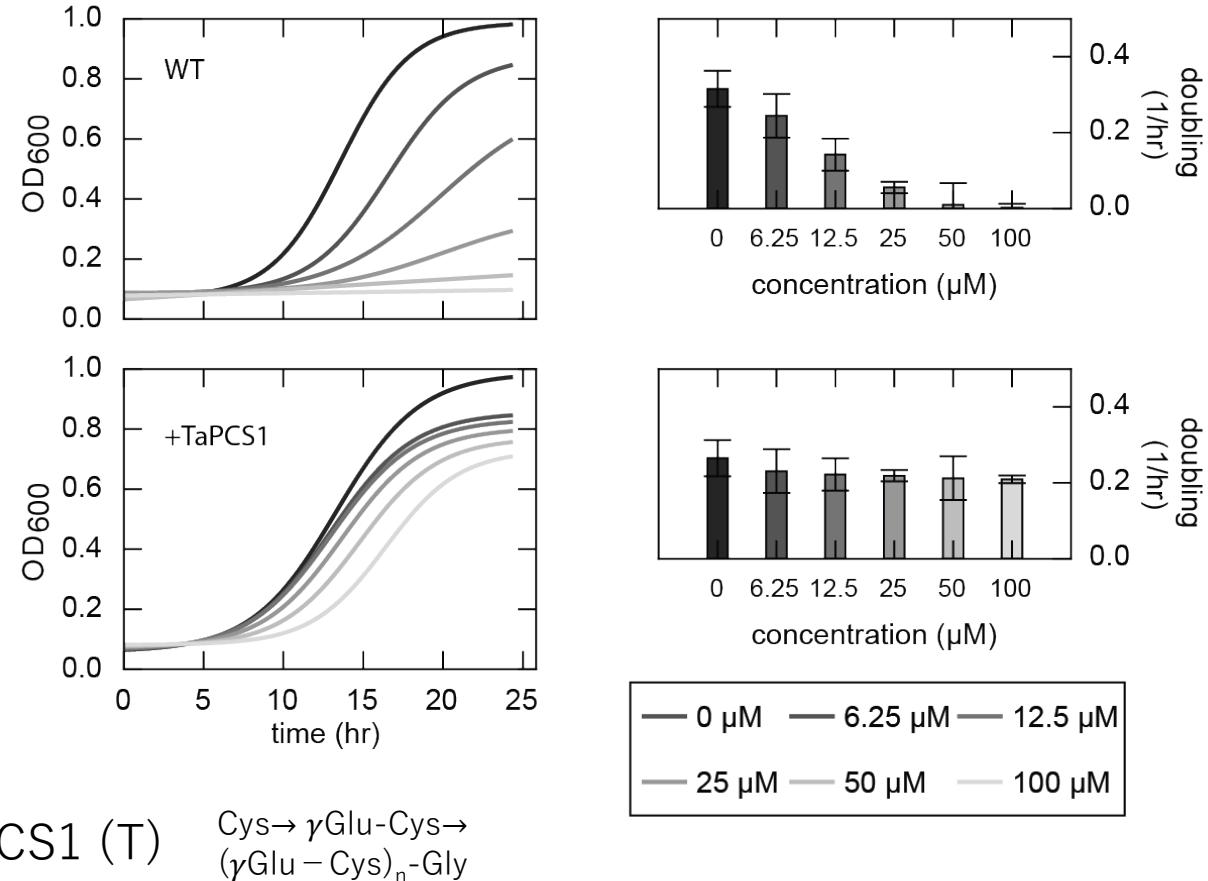
overexpression of vacuole transporter = ↑ uptake capacity



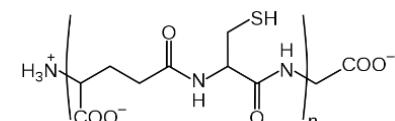
surface transport is the limiting “reaction”

Enhancing cell tolerance to internalized heavy metals

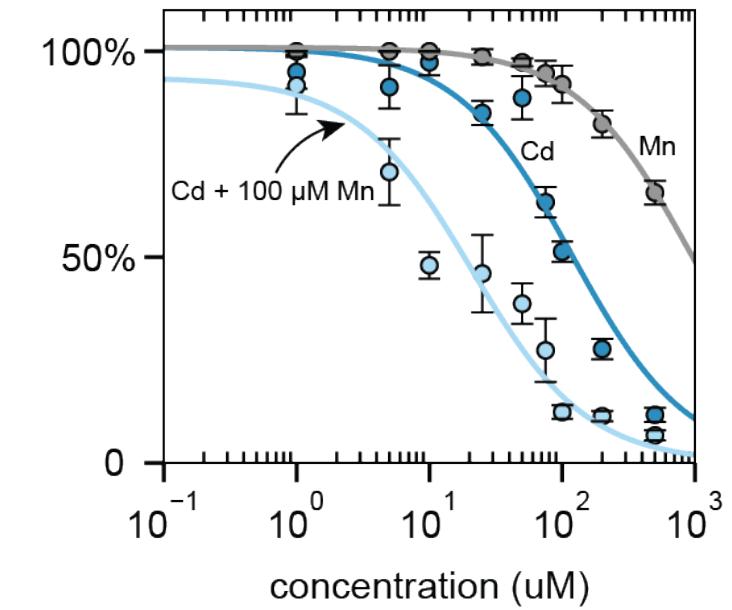
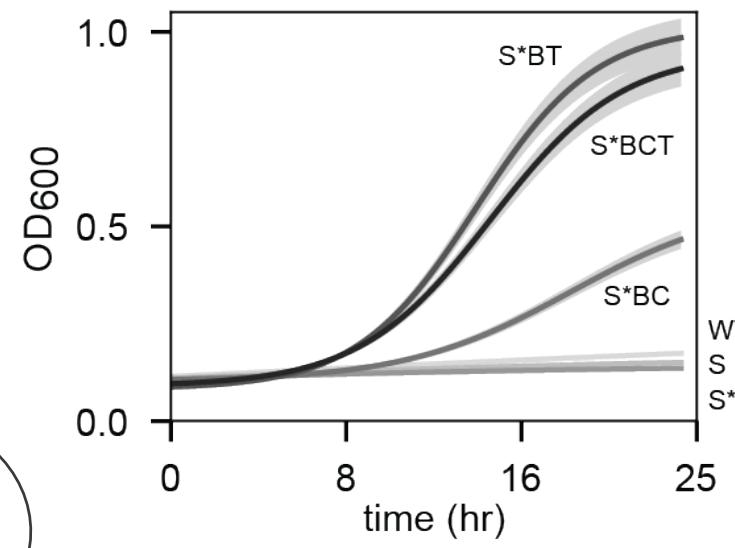
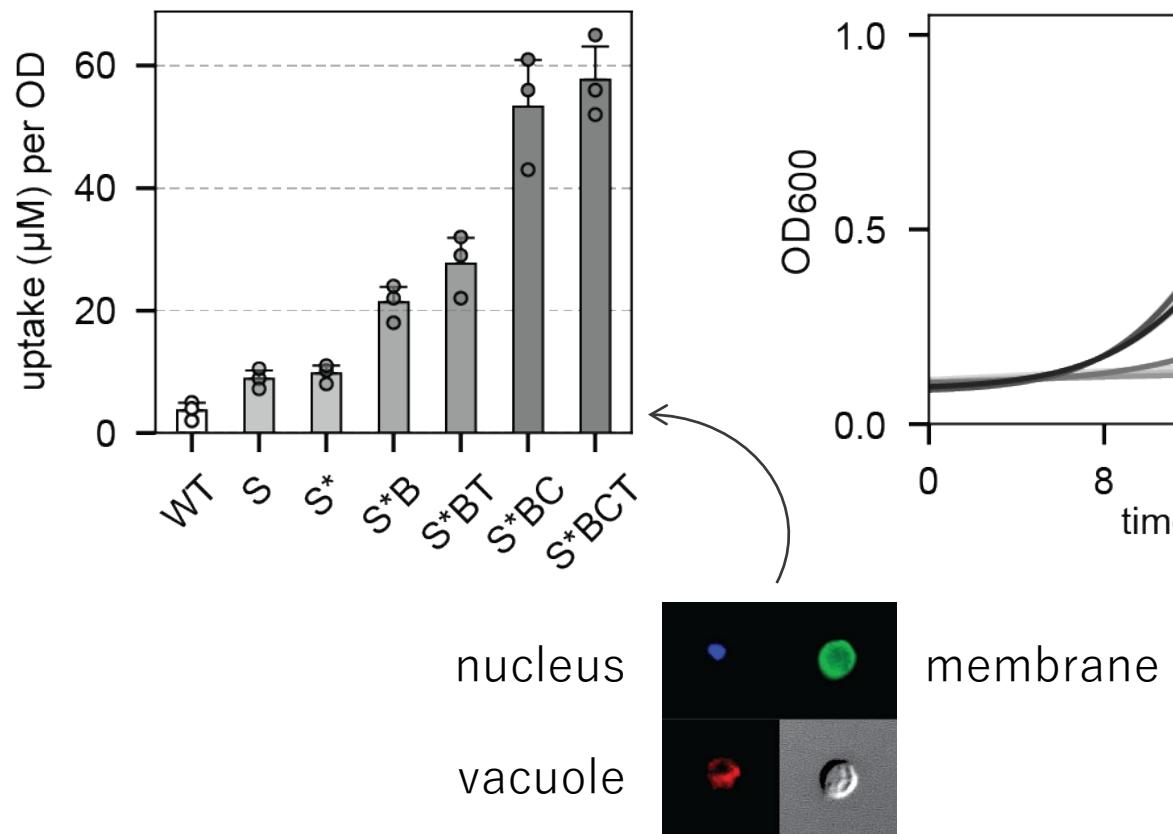
1. Overexpress transporter
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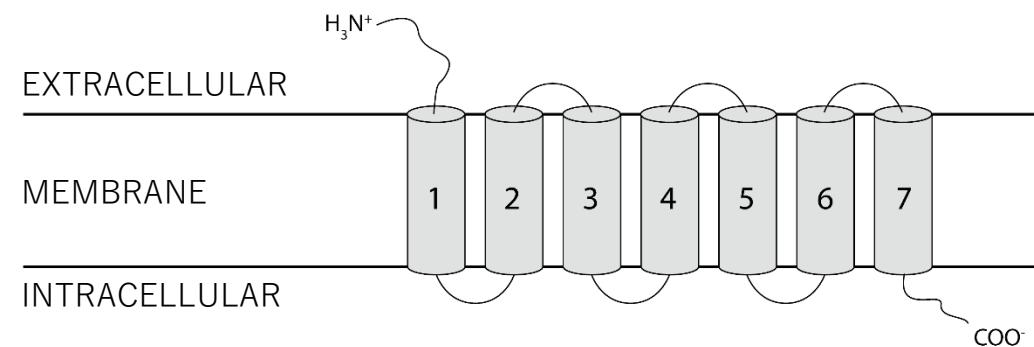
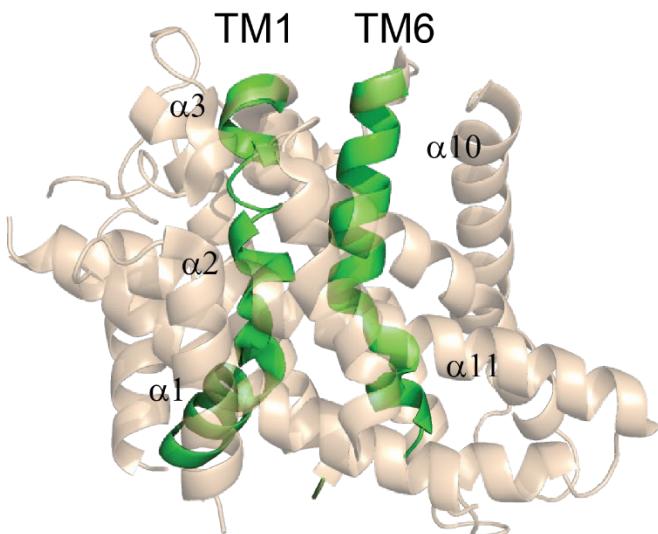
+TaPCS1 (T) Cys → γGlu-Cys →
 (γGlu-Cys)_n-Gly



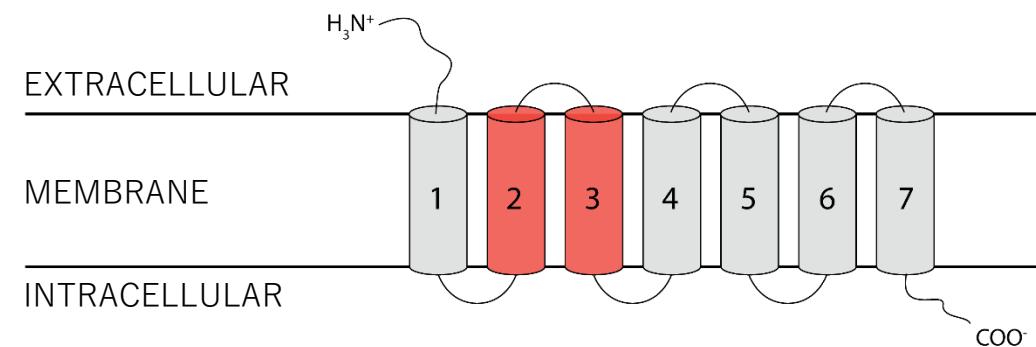
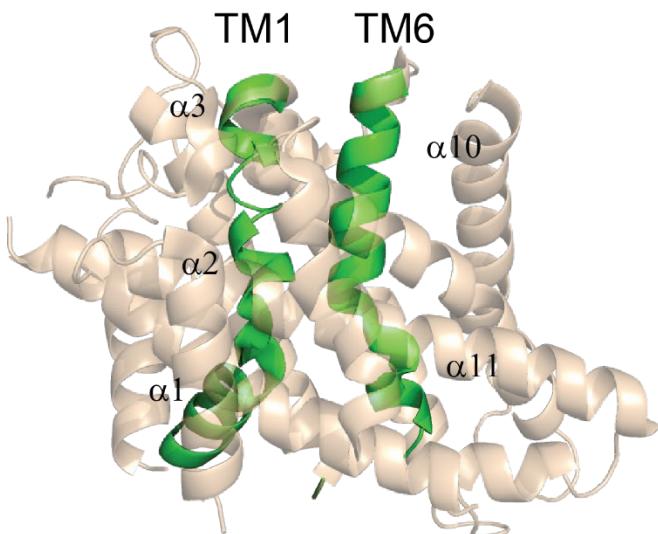
Combining design criteria into a single yeast strain for enhanced metal removal



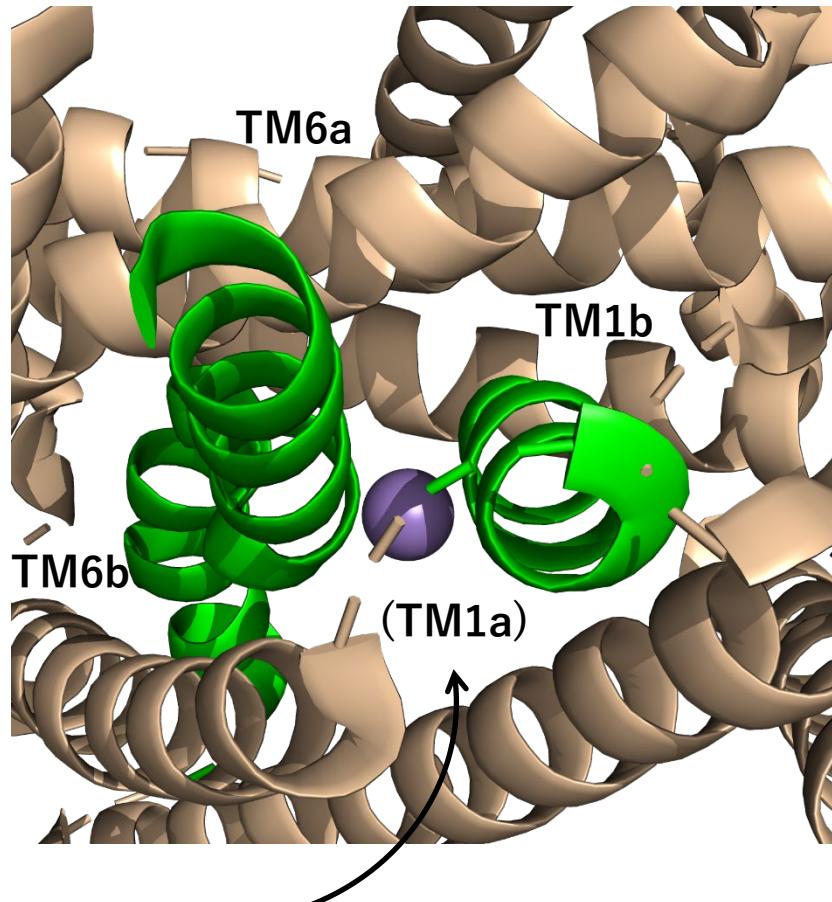
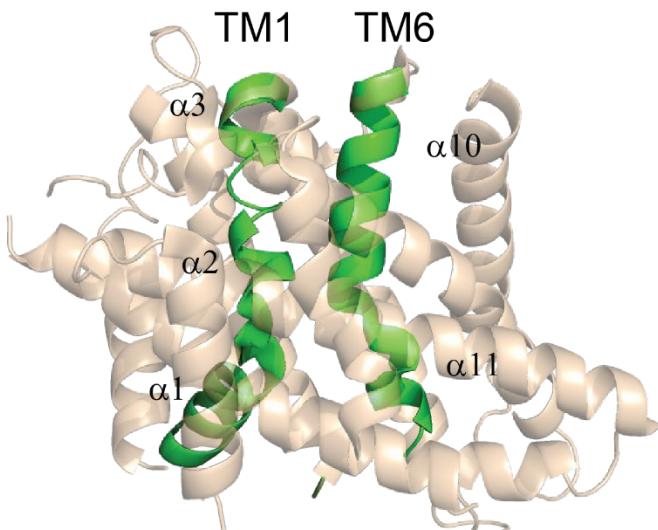
Altering metal transport specificity by engineering membrane transporters



Altering metal transport specificity by engineering membrane transporters

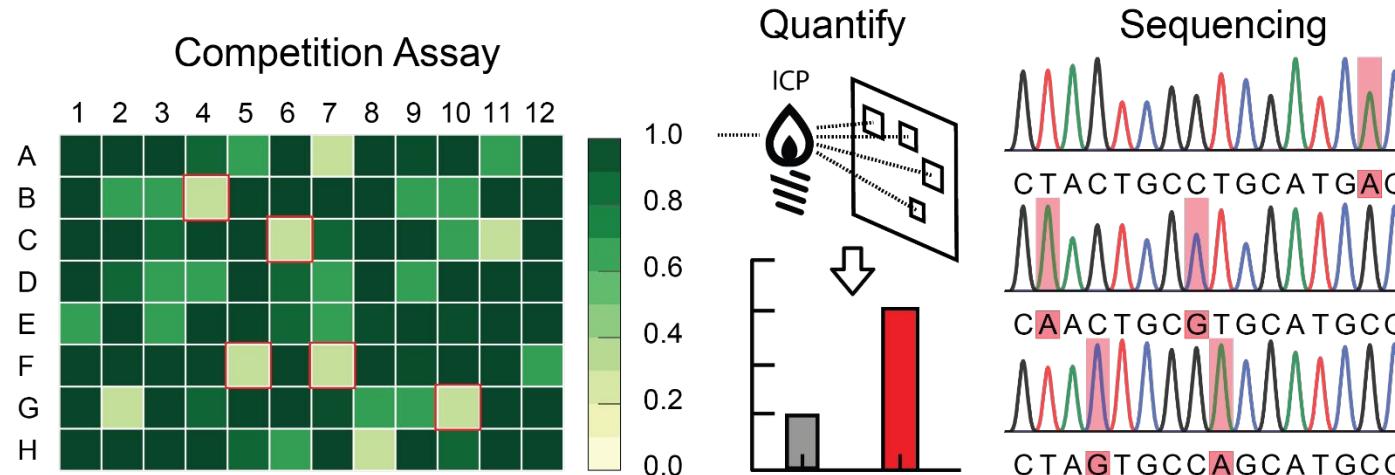
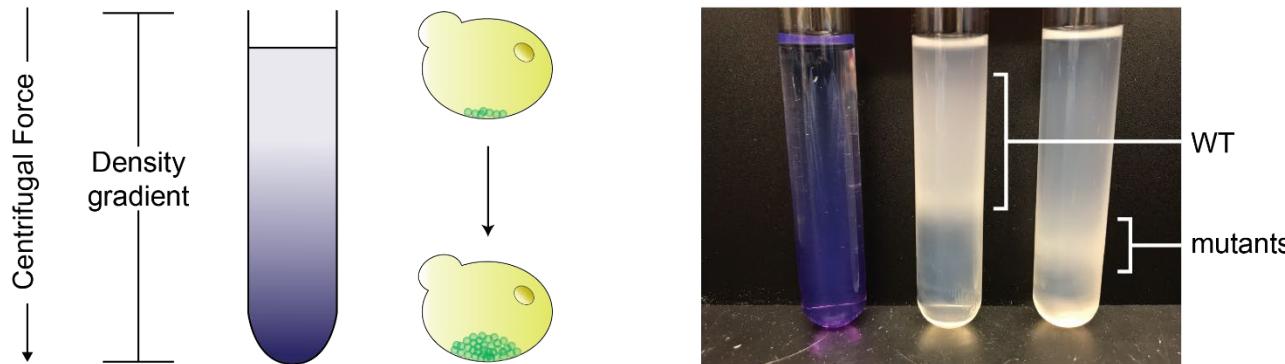


Altering metal transport specificity by engineering membrane transporters

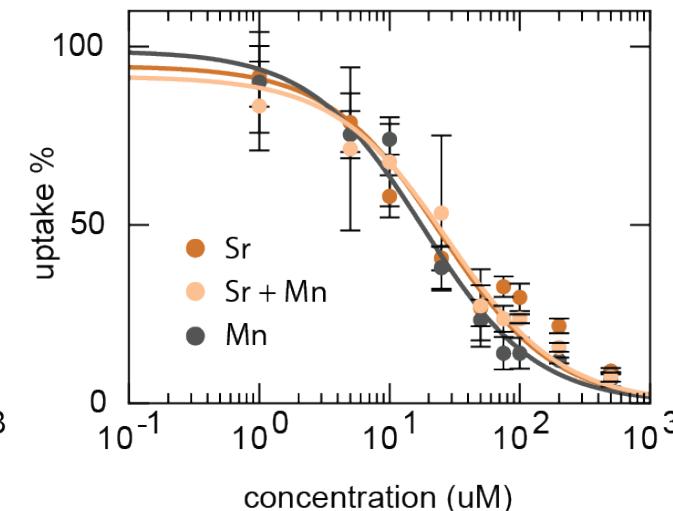
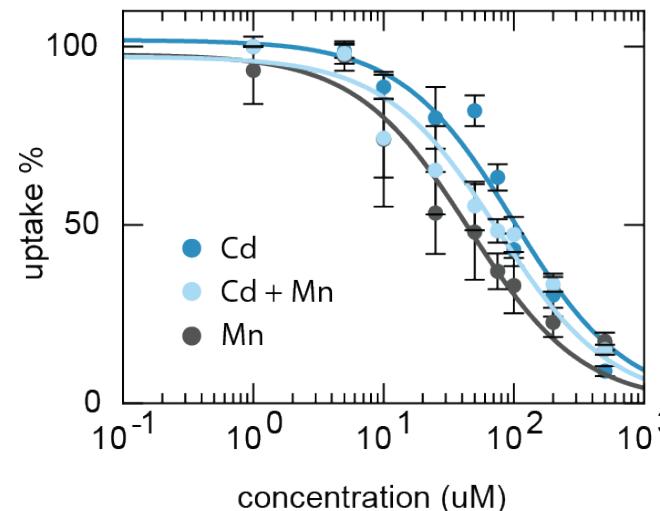
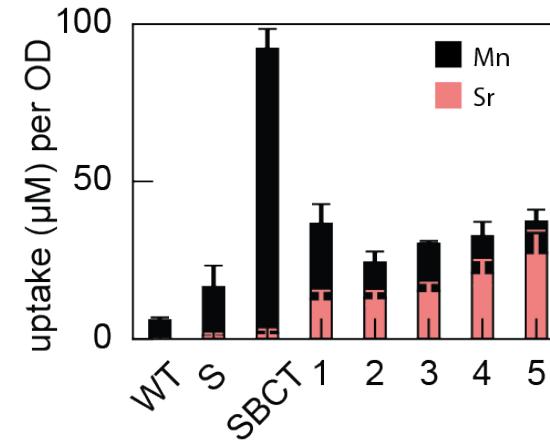
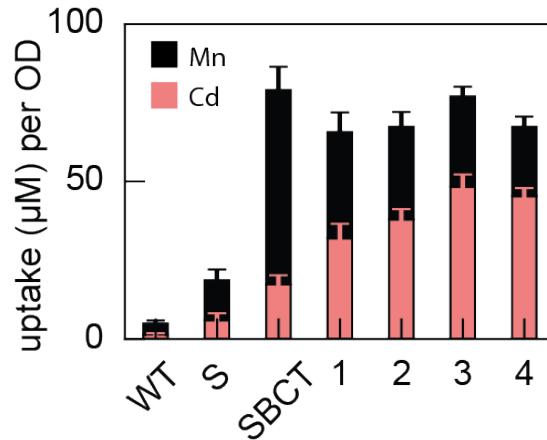


Rationally mutate

Designing a high throughput method for screening mutant metal transporters



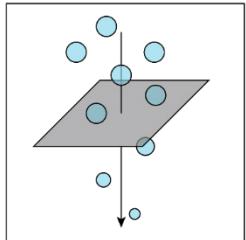
Creating cadmium and strontium transporters with reduced manganese uptake



Takeaway

Synthetic

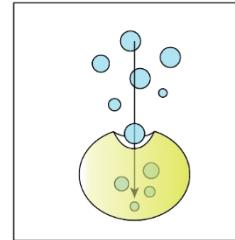
Absorption



Absorb

Biological

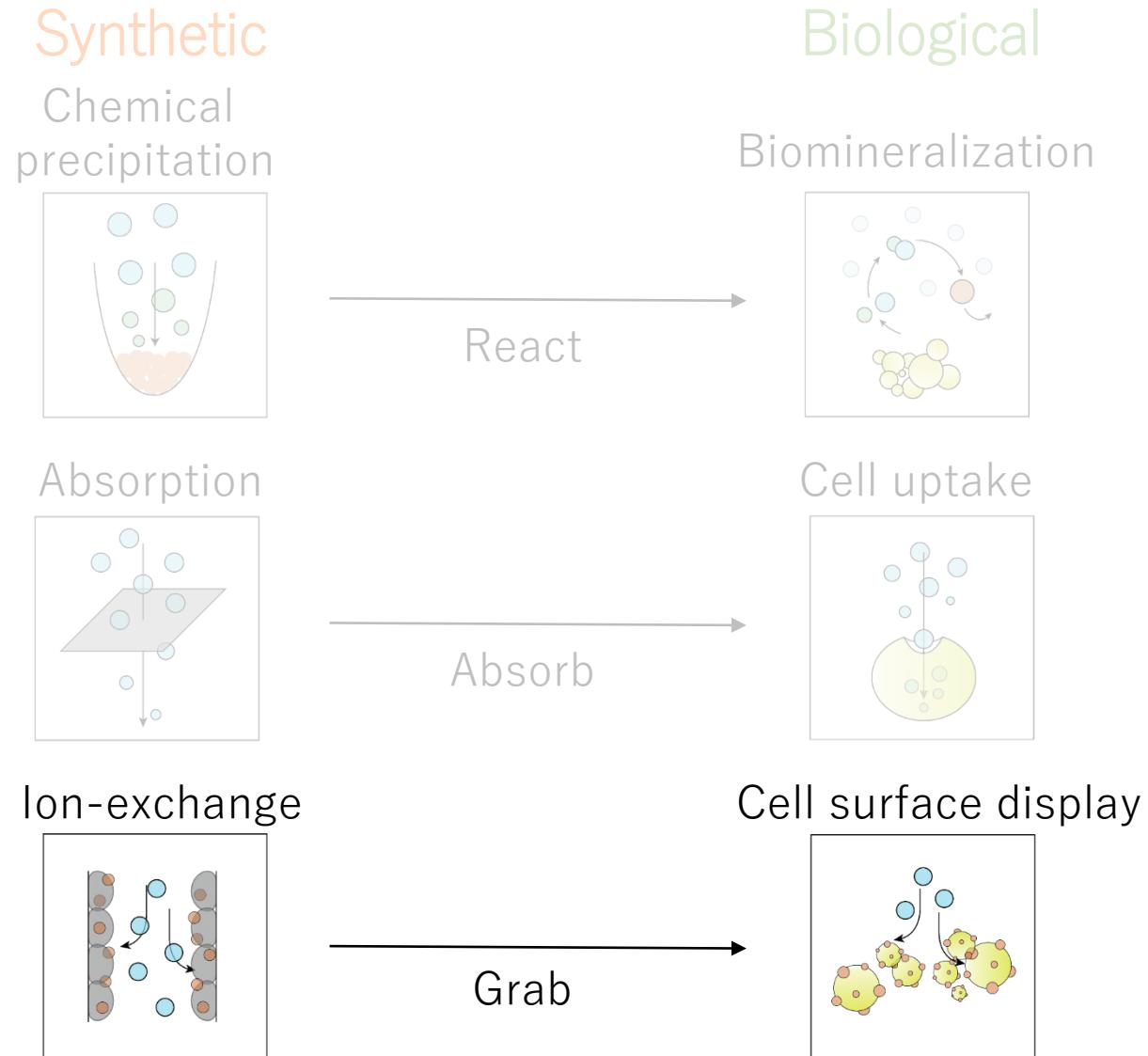
Cell uptake



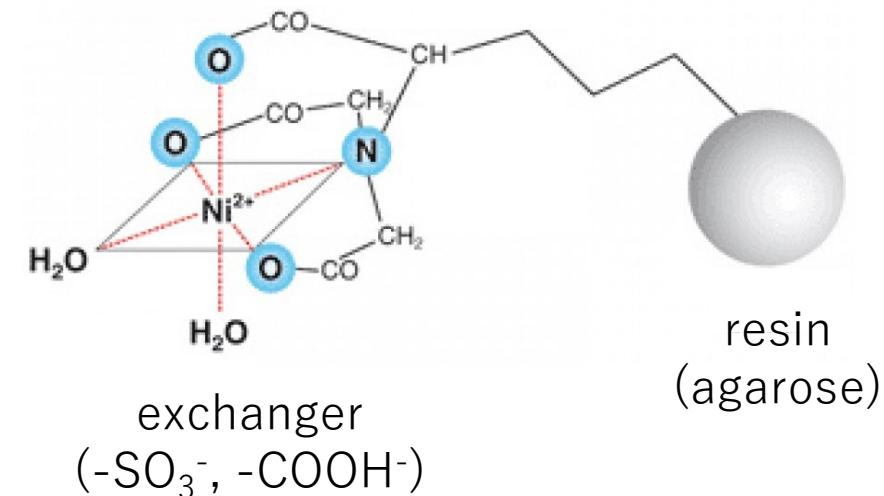
TAKEAWAY

- Membrane transporters confer metal hyperaccumulation
- Increasing transporter expression, longevity, and metal tolerance enhances metal uptake
- Membrane transporters offer high metal specificity, and can be engineered for other metals

Cell and protein chelation as an analogy to ion-exchange



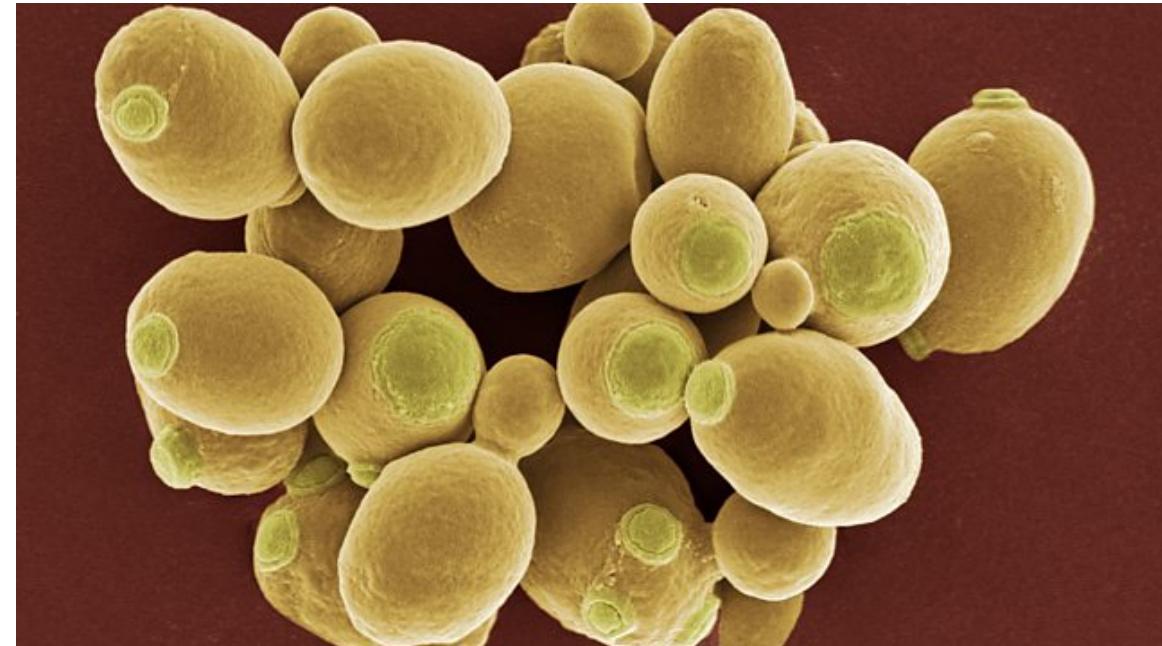
Ion-exchange has two primary components:
the resin and the exchanger



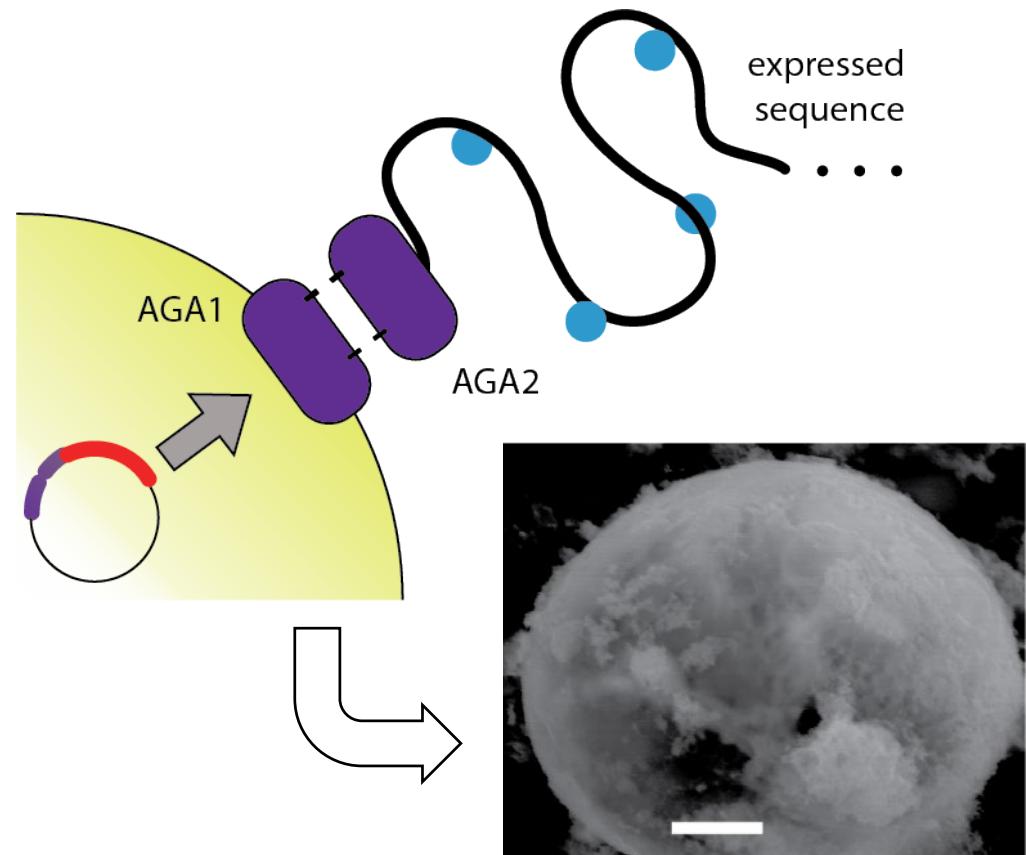
exchanger
($-\text{SO}_3^-$, $-\text{COOH}^-$)

resin
(agarose)

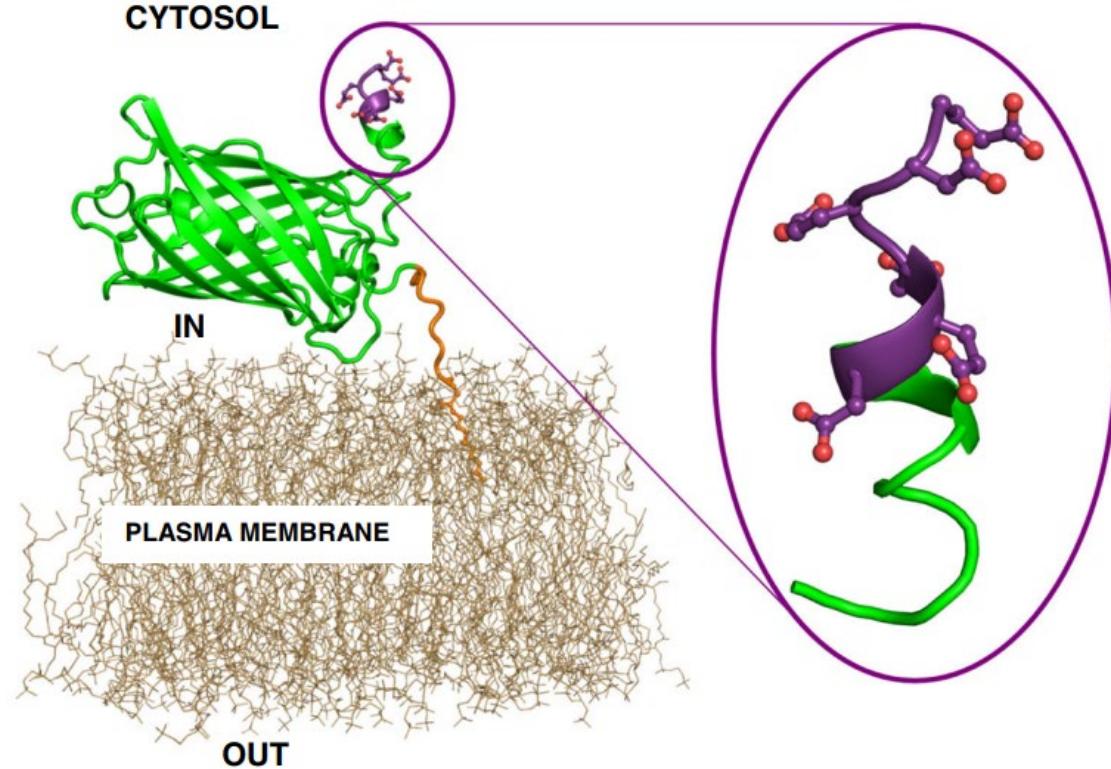
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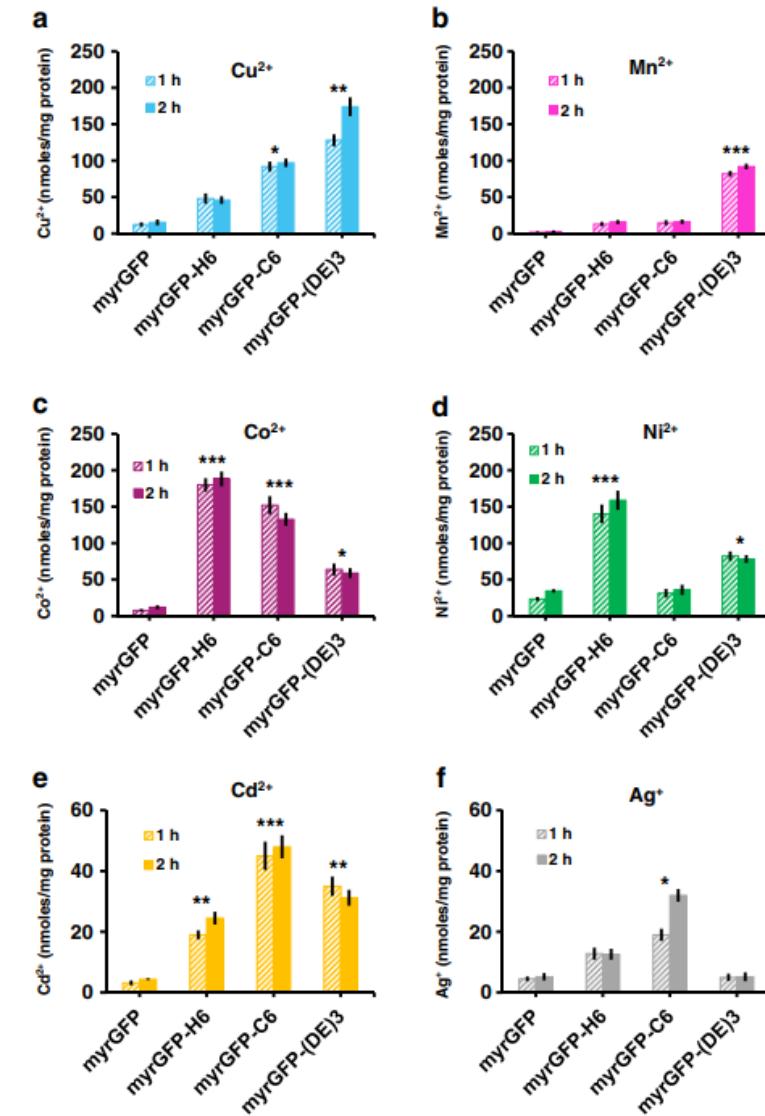
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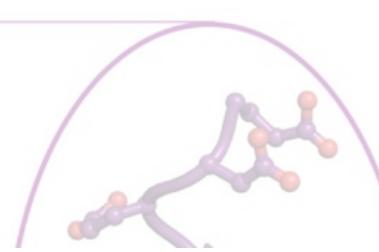
Past studies using display technology showed metal removal, but may not be enough



Ruta, Lavinia Liliana, et al. "Heavy metal accumulation by *Saccharomyces cerevisiae* cells armed with metal binding hexapeptides targeted to the inner face of the plasma membrane." (2017)



Past studies using display technology showed metal removal, but may not be enough



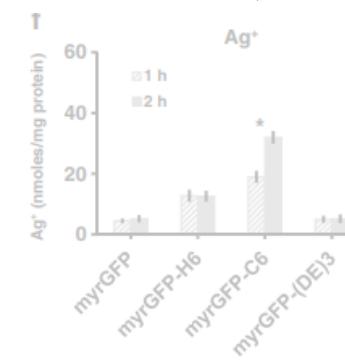
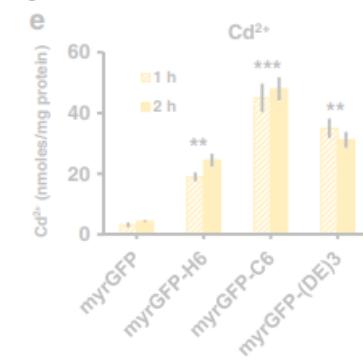
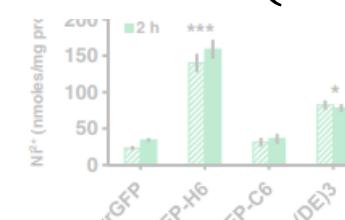
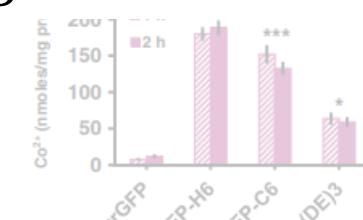
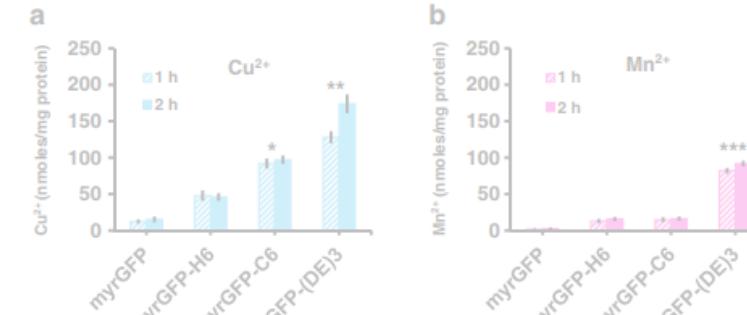
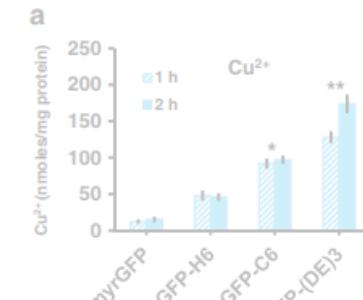
$$\# \text{ cells} \times \# \text{ displayed peptides} \times \# \text{ binding sites} \approx 2 \text{ nano molar (10}^{-9}\text{)}$$



$$\# \text{ resins} \times \# \text{ functional groups} \times \# \text{ binding sites} > \text{milli molar (10}^{-3}\text{)}$$



Ruta, Lavinia Liliana, et al. "Heavy metal accumulation by *Saccharomyces cerevisiae* cells armed with metal binding hexapeptides targeted to the inner face of the plasma membrane." (2017)



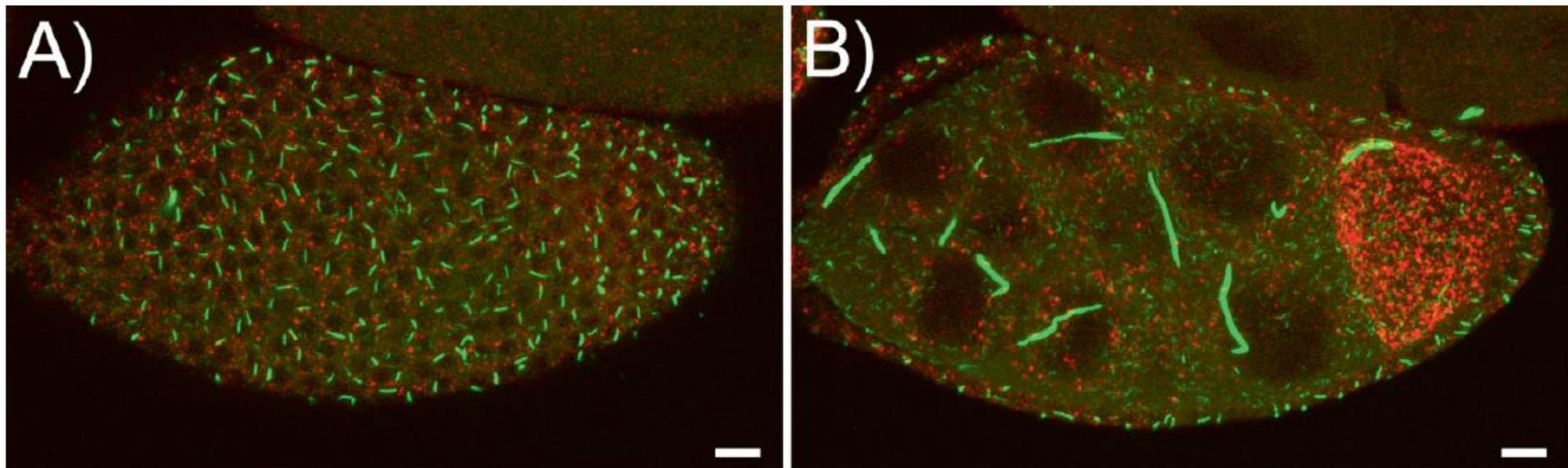
Discovery of aggregating proteins can inspire new methods for metal-exchange removal



Insights & Perspectives | Free Access

The enigmatic cytoophidium: Compartmentation of CTP synthase via filament formation

Ji-Long Liu

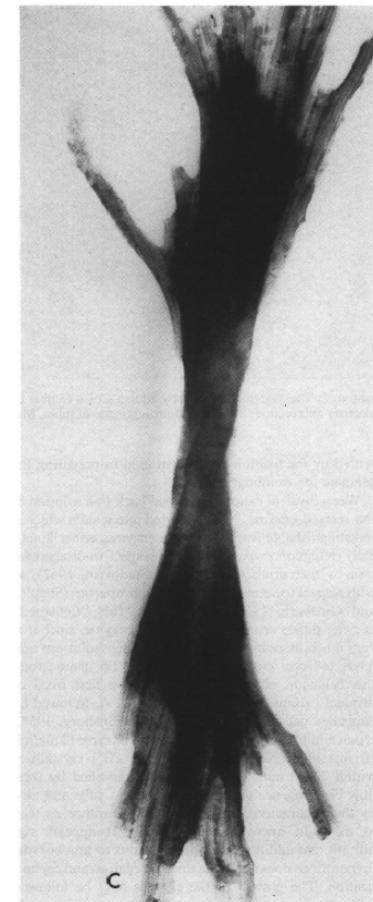
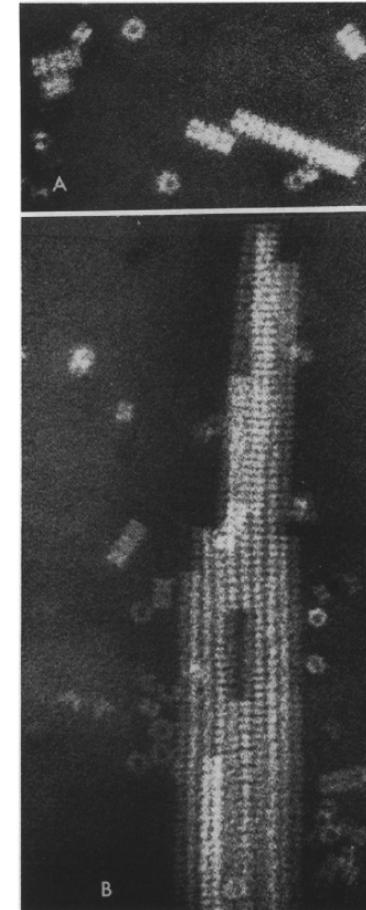


Numerous proteins with known enzymatic function have unique aggregating properties

Table I. Proteins that assemble into intracellular structures

Foci-forming protein	Biological process
Prs4p	5-phosphoribose 1-diphosphate biosynthetic process
Acs1p	Acetate fermentation/acetyl-CoA biosynthetic process/histone acetylation
Glt1p	Ammonia assimilation cycle/glutamate biosynthetic process
Ssd1p	Cell wall organization/chronological cell aging/relicative cell aging
Ura7p	CTP synthesis
Hsp42p	Cytoskeleton organization/response to stress
Rnr4p	Deoxyribonucleotide biosynthetic process
Rnr2p	Deoxyribonucleotide biosynthetic process
Psa1p	GDP-mannose biosynthetic process/protein amino acid glycosylation
Gln1p	Glutamine biosynthetic process/nitrogen compound metabolic process
Dug2p	Glutathione catabolic process
Gly1p	Glycine biosynthetic process/threonine catabolic process
Gsy2p	Glycogen biosynthetic process
Gdb1p	Glycogen catabolic process
Gph1p	Glycogen catabolic process
Hem2p	Heme biosynthetic process
His4p	Histidine biosynthetic process
Hek2p	Intracellular mRNA localization/telomere maintenance via telomere
Rim20p	Invasive growth in response to glucose limitation/protein processing/proteolysis
Sam1p	Methionine metabolic process/S-adenosylmethionine biosynthetic process
Sam2p	Methionine metabolic process/S-adenosylmethionine biosynthetic process
Hsp104p	Protein folding
Ssa1p	Protein folding
Sse2p	Protein folding
Ssa2p	Protein folding
Sis1p	Protein folding
Rpn9p	Proteosome assembly/ubiquitin-dependent protein catabolic process
Gcd6p	Regulation of translation initiation
Sui2p	Regulation of translation initiation
Gcd2p	Regulation of translation initiation
Gcd7p	Regulation of translation initiation
Gcn3p	Regulation of translation initiation
Sgt2p	Response to heat (glutamine-rich cytoplasmic protein of unknown function)
Thr1p	Threonine metabolic process
YAR009Cp	Transposition, RNA mediated
YLR143Wp	Unknown
YMR253Cp	Unknown

Narayanaswamy, R. et al. Widespread reorganization of metabolic enzymes into reversible assemblies upon nutrient starvation. *PNAS* 106, 10147–10152 (2009).

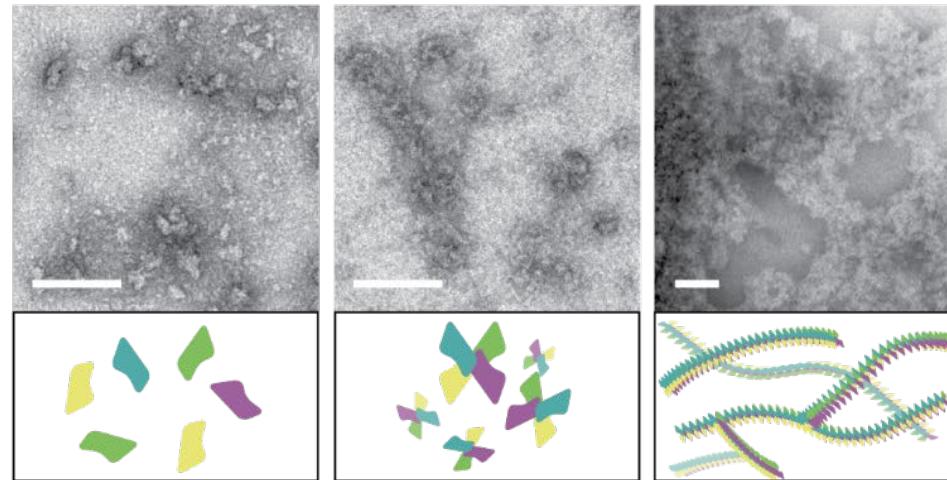
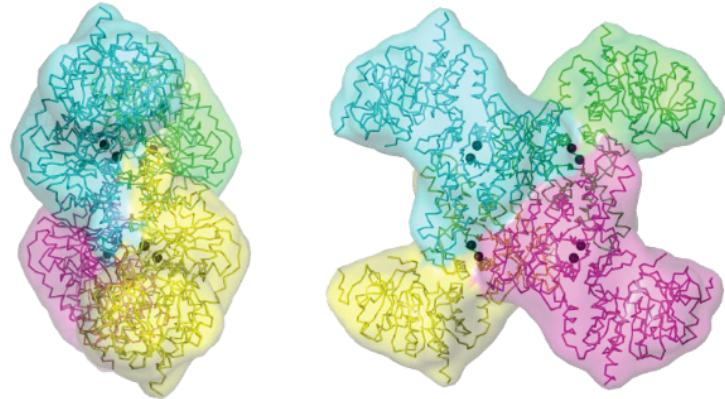


B
C

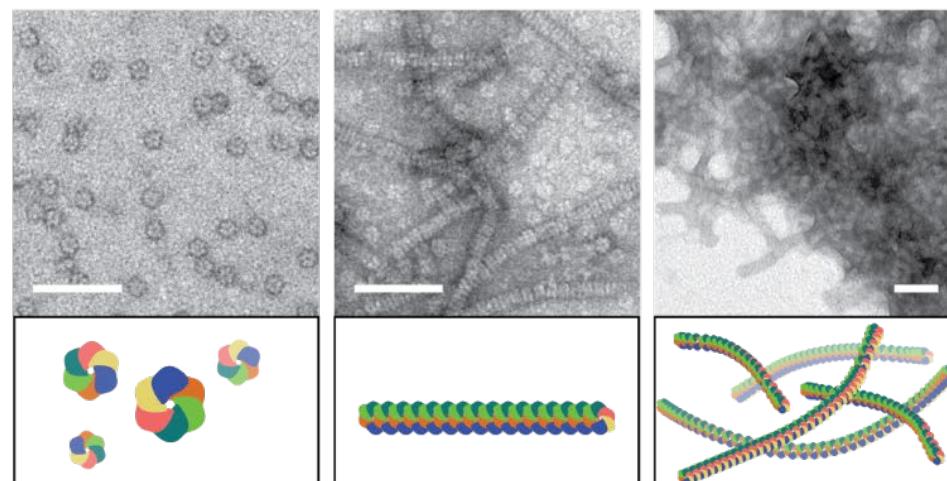
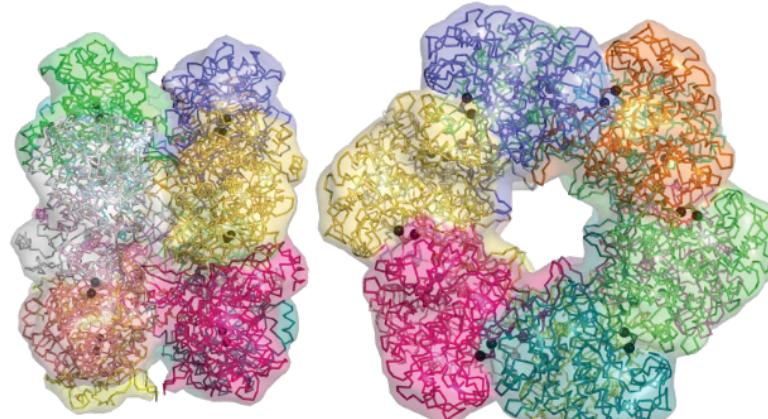
Valentine, R. C., Shapiro, B. M. & Stadtman, E. R. Regulation of glutamine synthetase. XII. Electron microscopy of the enzyme from *Escherichia coli*. *Biochemistry* 7, 2143–2152 (1968).

pyrG and glnA from *E. coli* have metal-inducible aggregation behavior

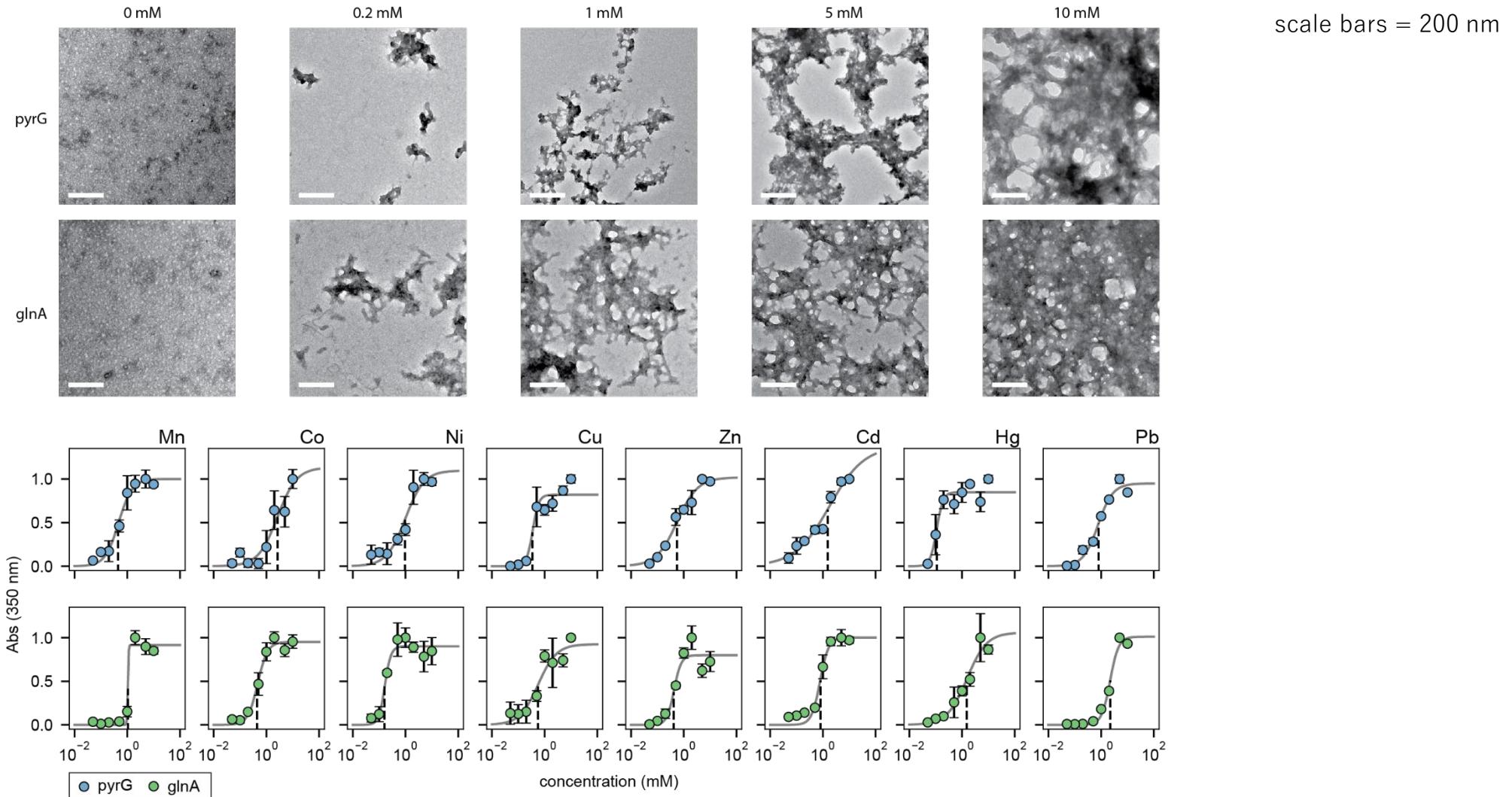
pyrG



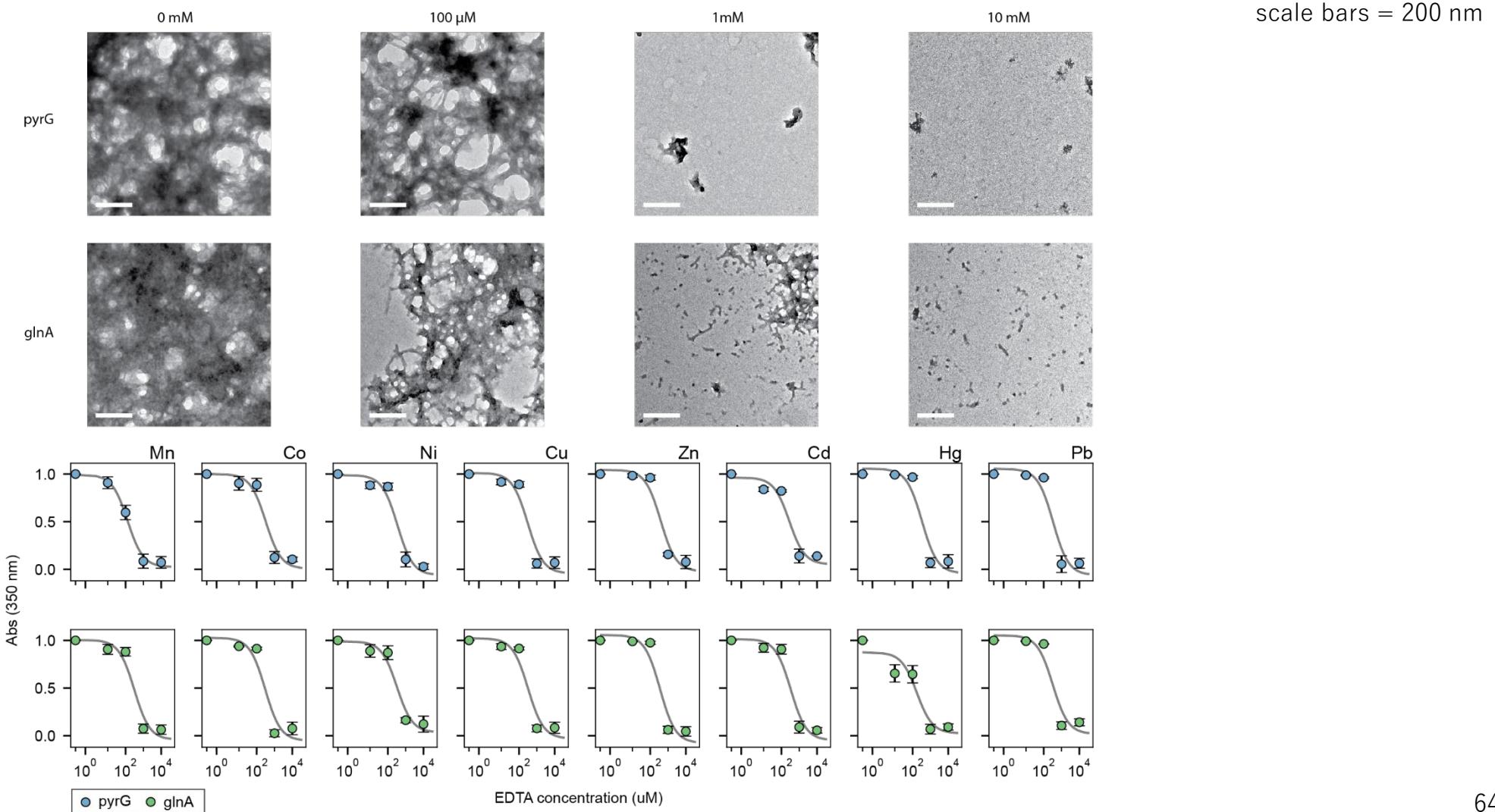
glnA



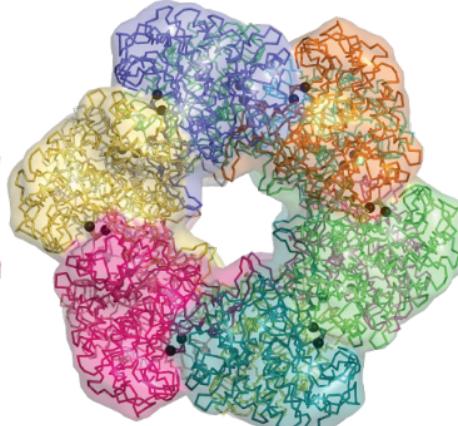
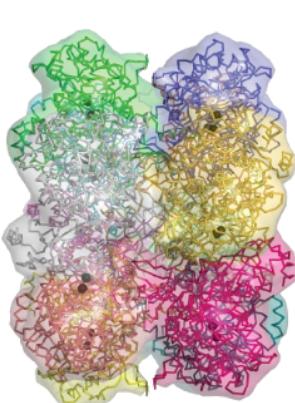
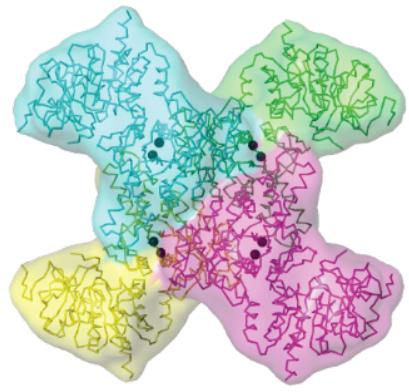
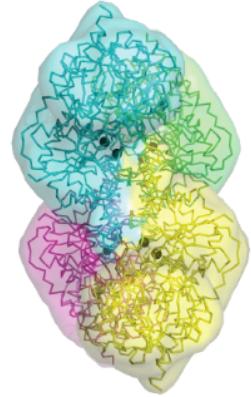
Aggregation intensity is induced by metal concentration



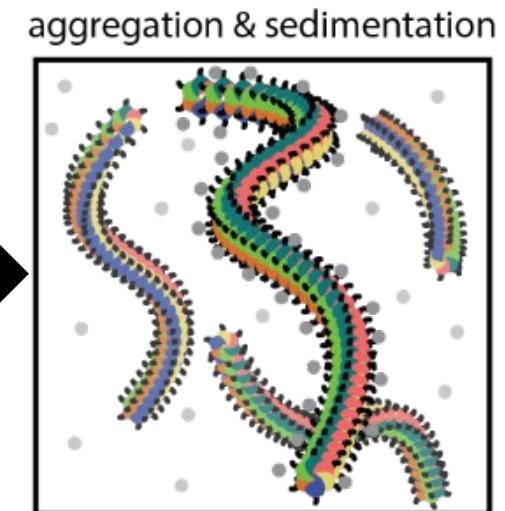
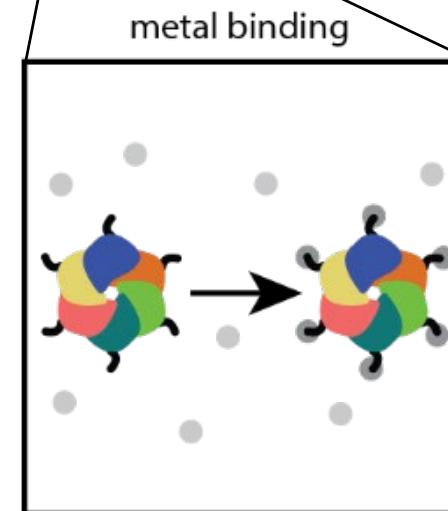
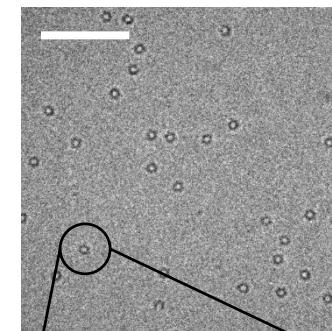
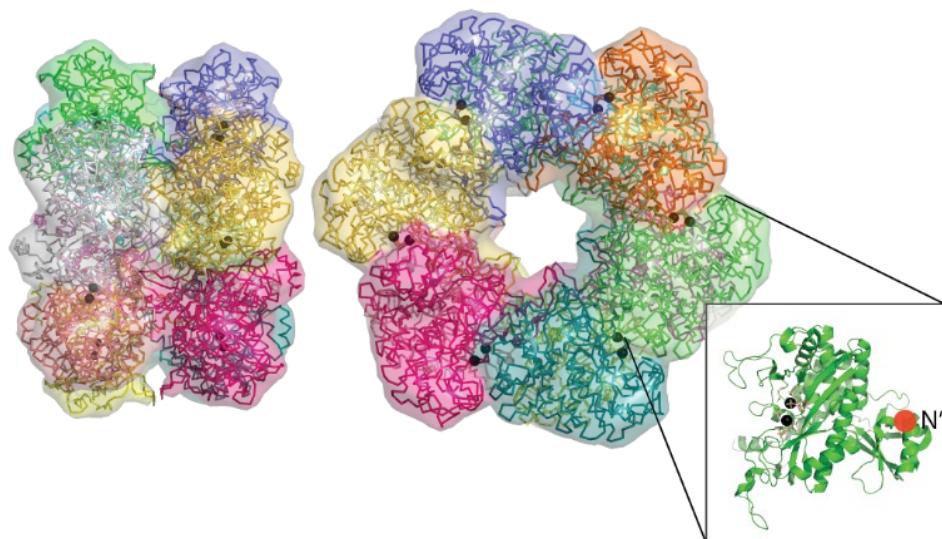
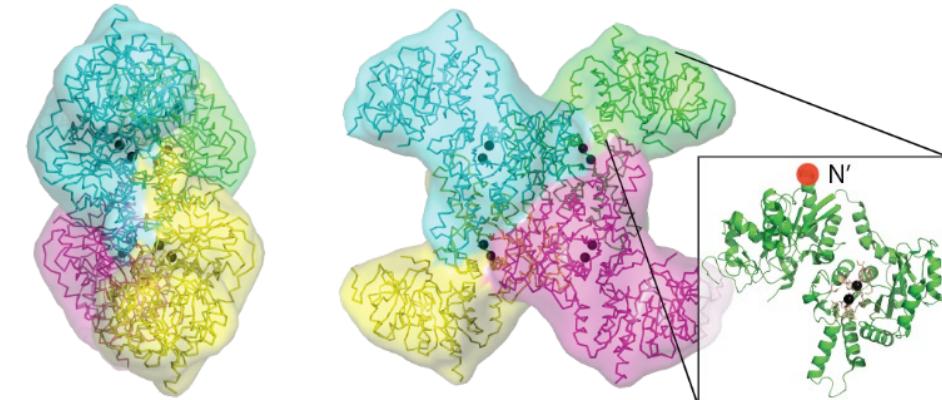
Absence or competition of metals reverses aggregation



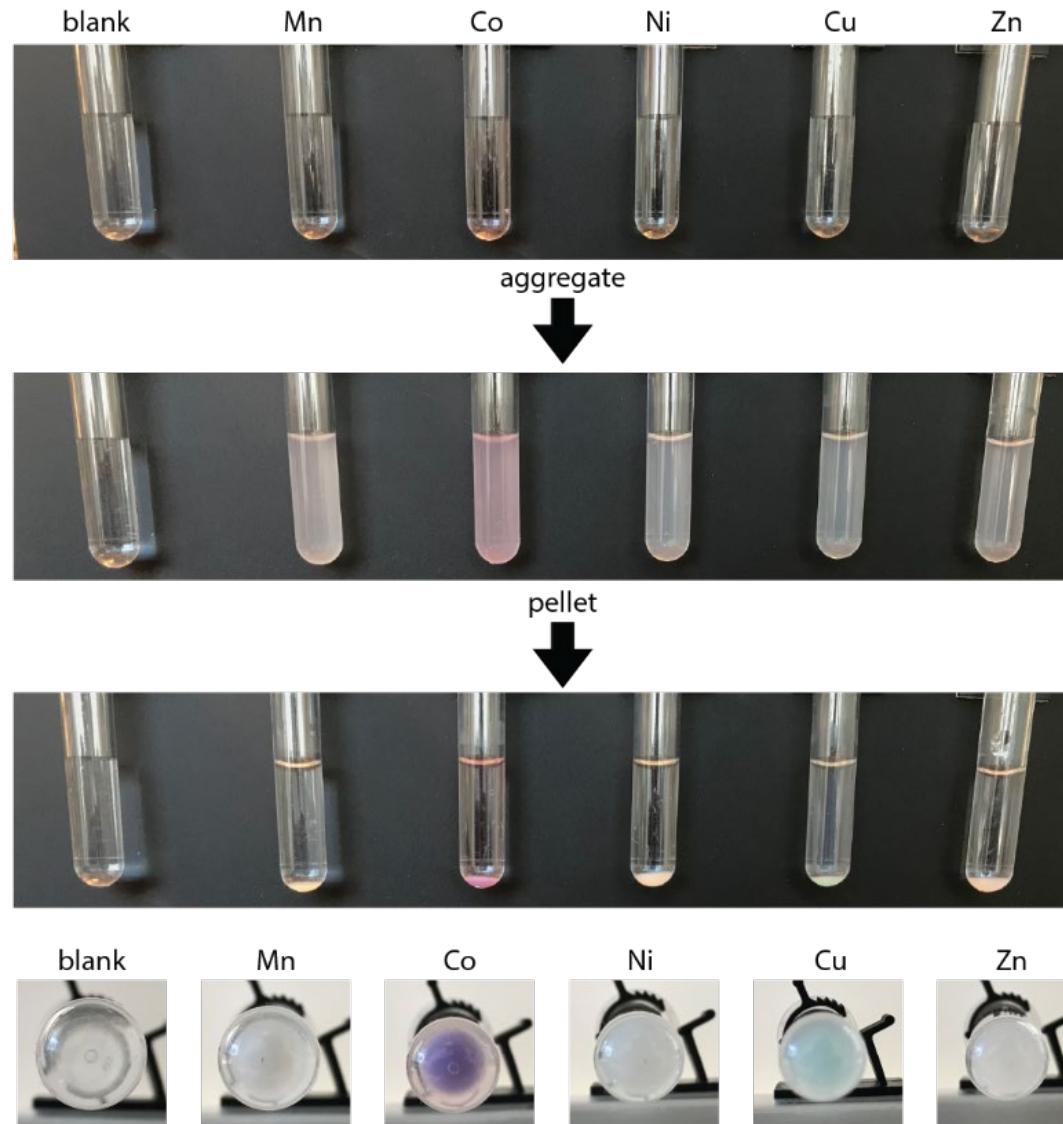
Engineering these proteins to act as both
'resin' and 'exchanger'



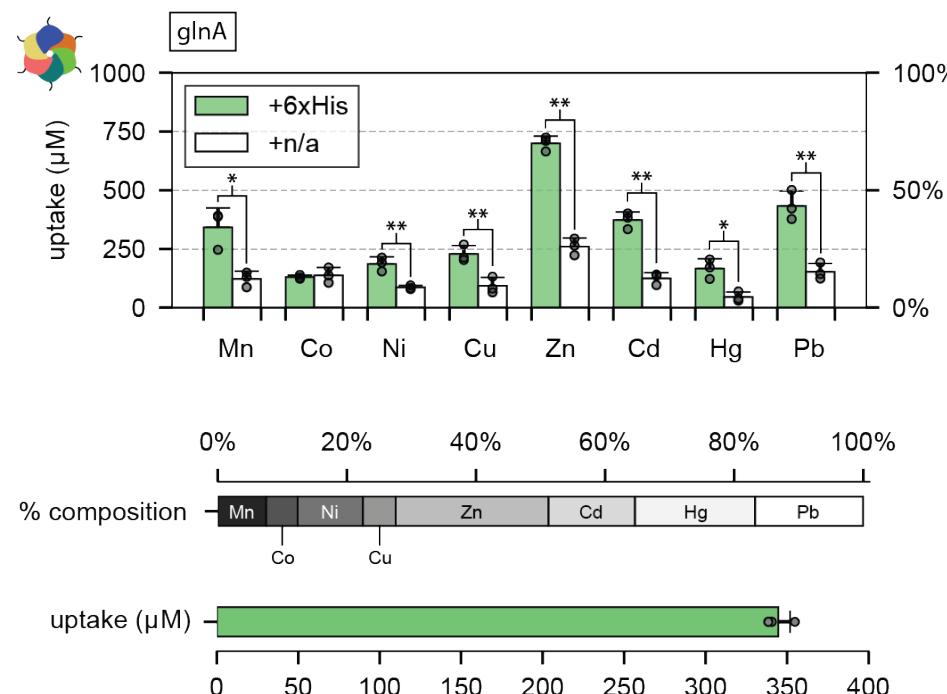
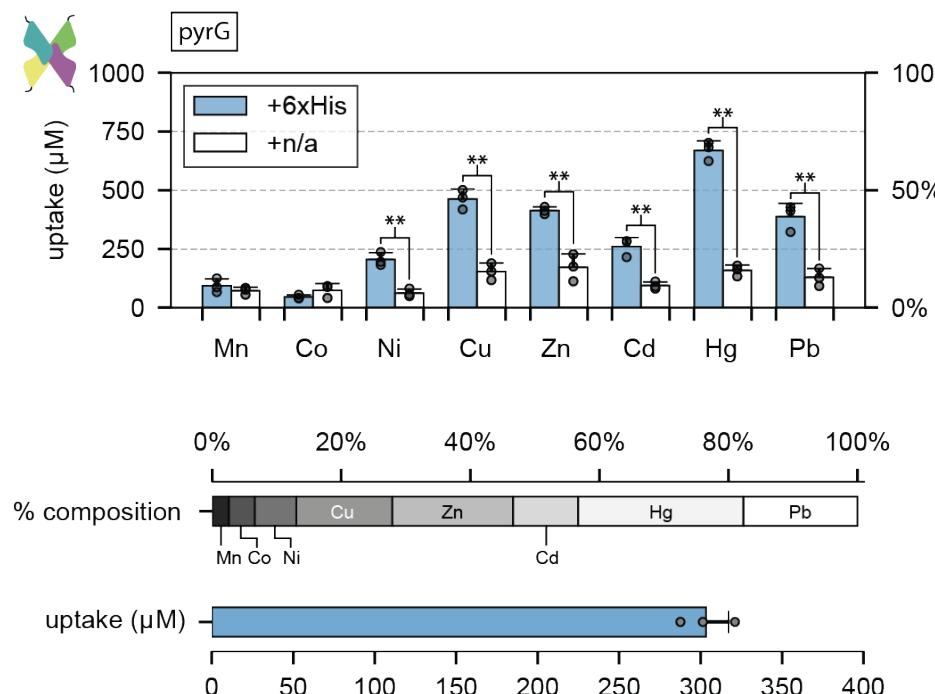
Engineering these proteins to act as both 'resin' and 'exchanger'



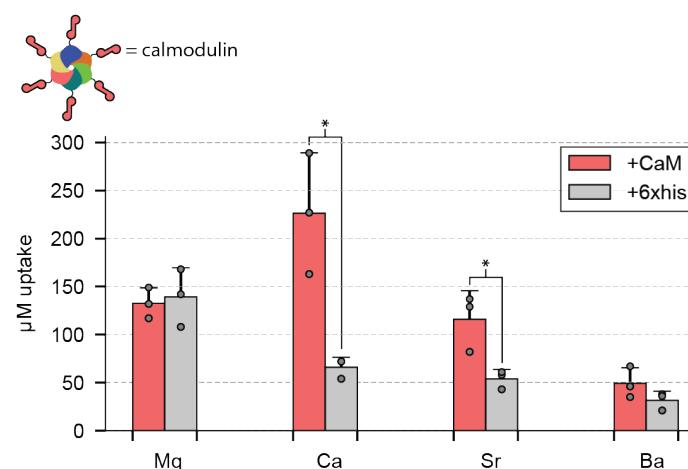
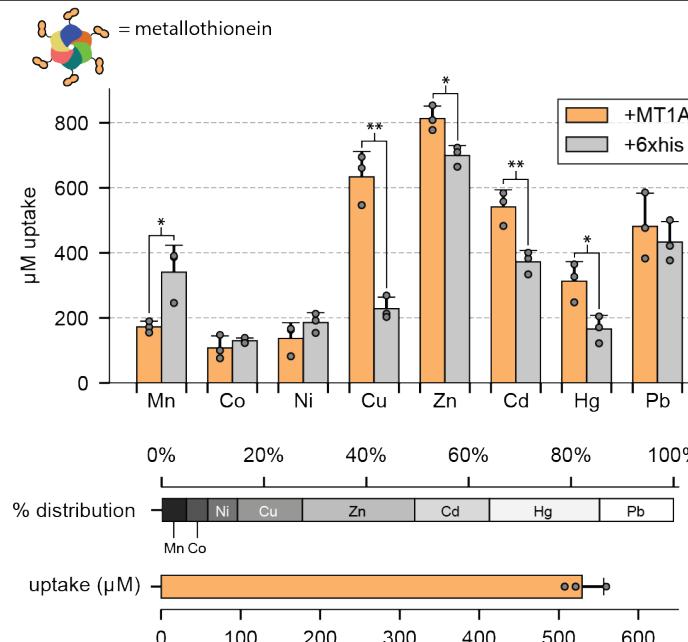
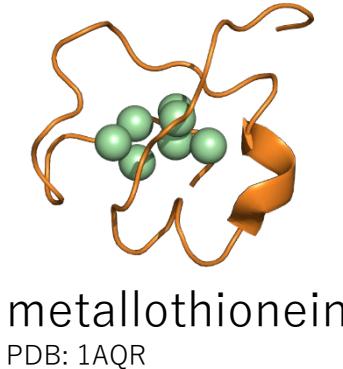
Surface modification with a His-tag helps chelate and sediment bound metals



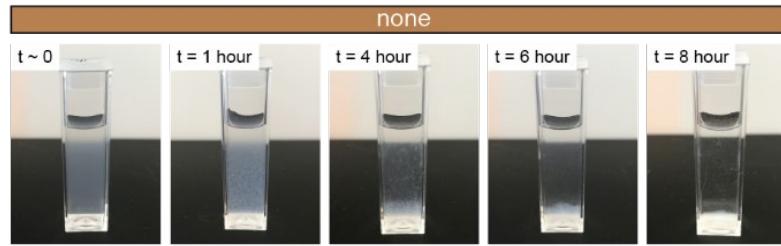
Metal removal profile is determined by aggregation propensity and exchanger



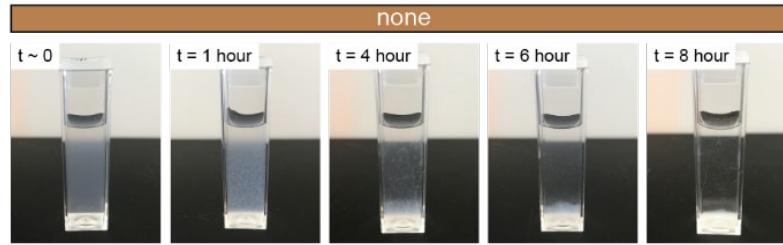
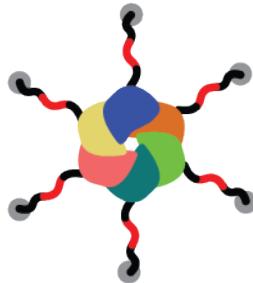
Changing the metal binding domain alters the metal removal profile



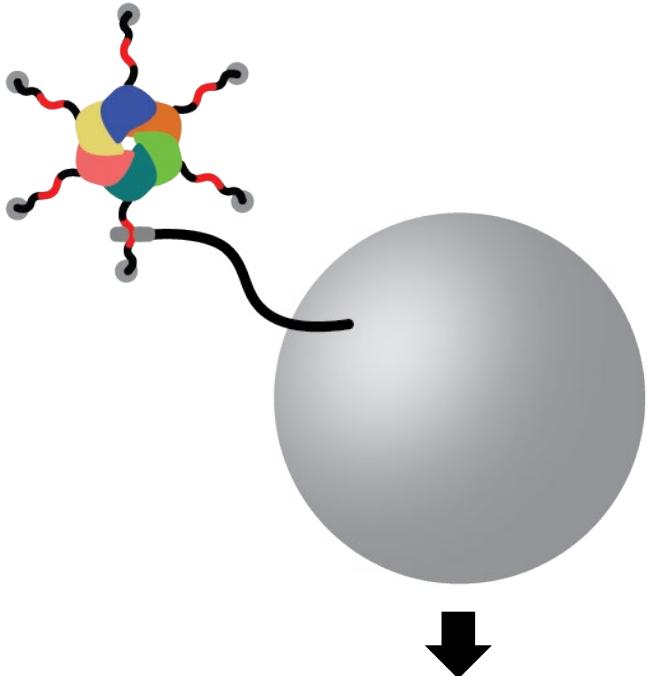
Further addition of a binding motif allows for anchorage onto denser substrates



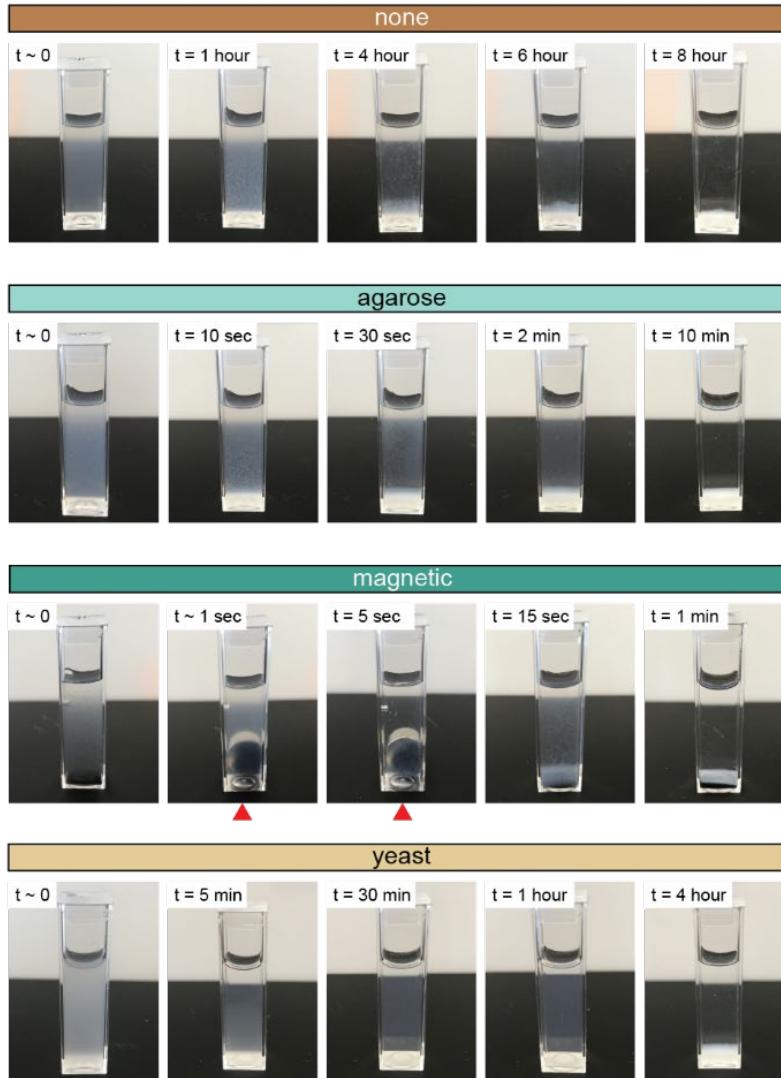
Further addition of a binding motif allows for anchorage onto denser substrates



Further addition of a binding motif allows for anchorage onto denser substrates

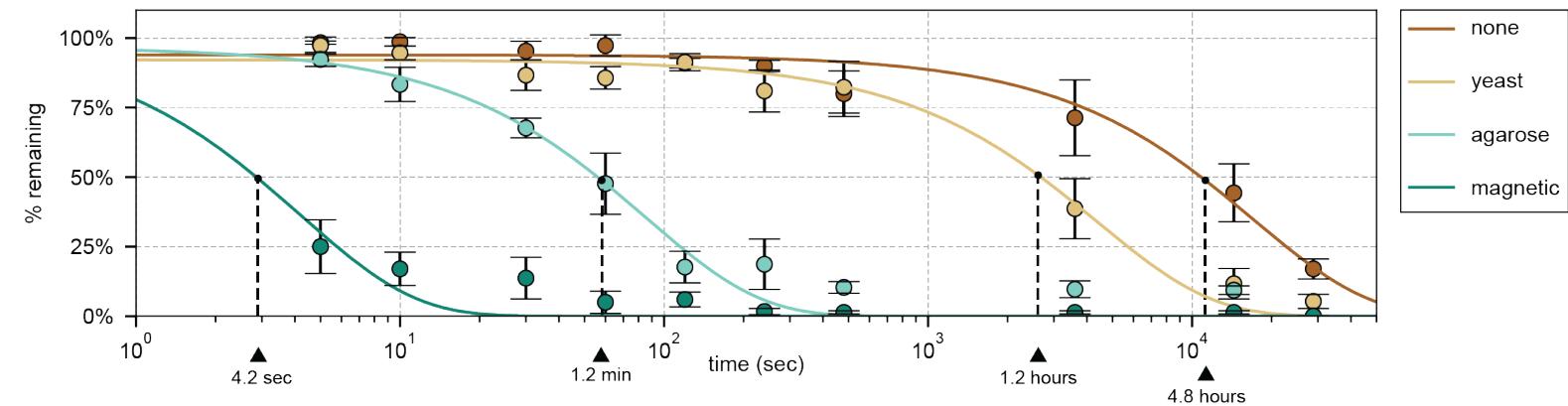
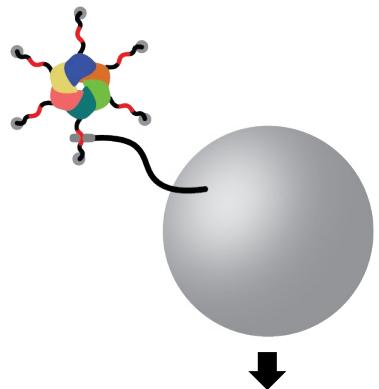


gravity
magnetically
biologically

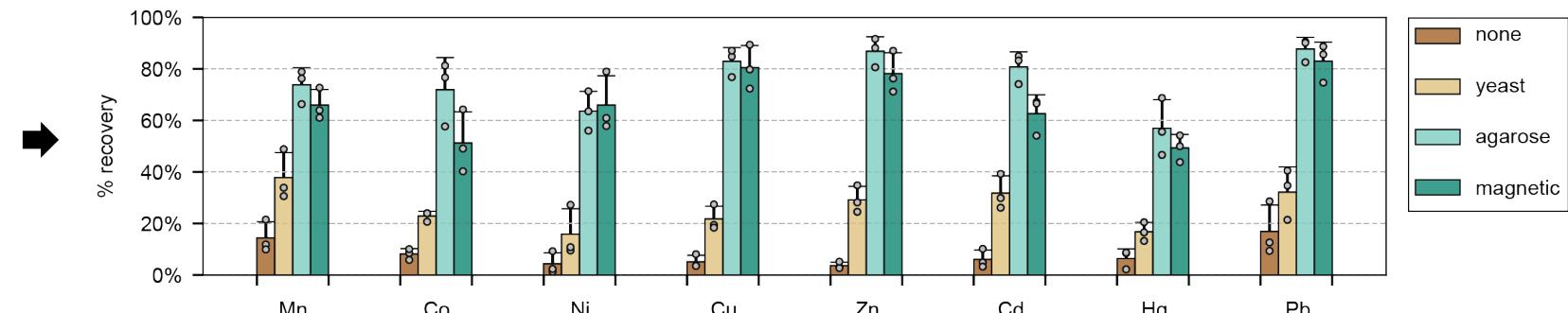
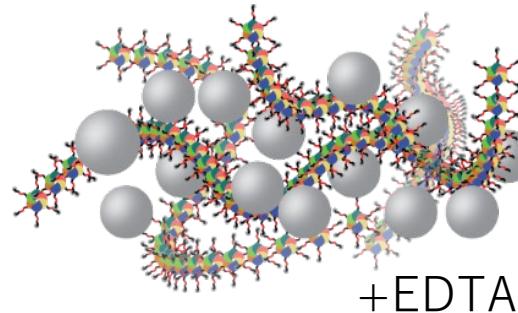


Further addition of a binding motif allows for anchorage onto denser substrates

sedimentation



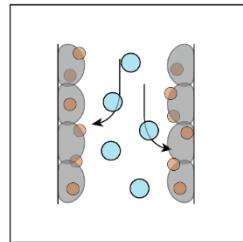
recovery



Takeaway

Synthetic

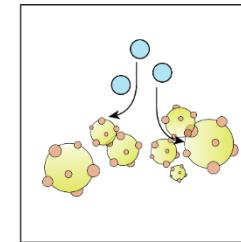
Ion-exchange



Grab

Biological

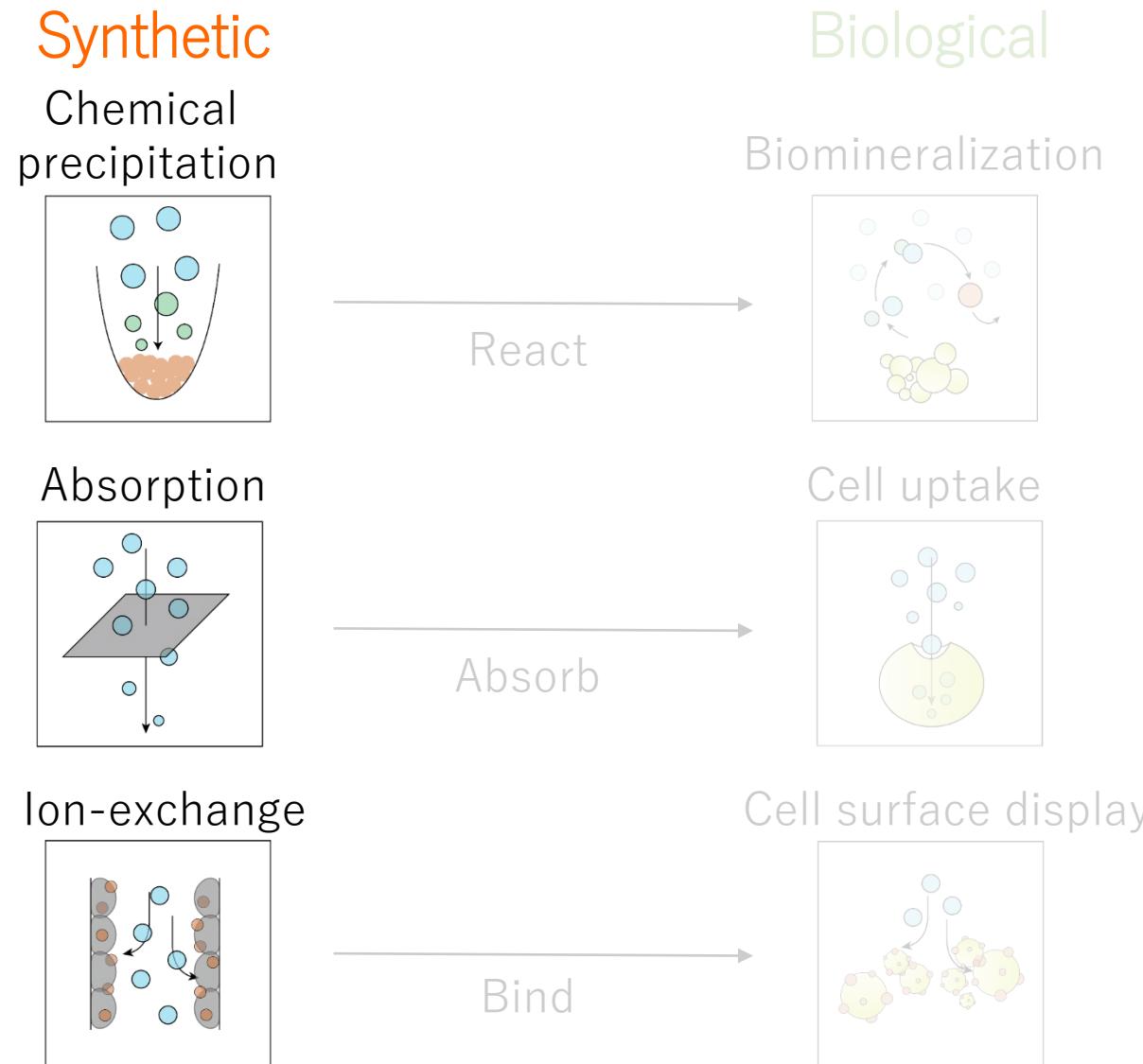
Cell surface display



TAKEAWAY

- Display technology is possible, but does not scale
- Aggregation behavior of recently discovered proteins can be engineered to be ion-exchange like
- Surface modifications offer new ways to remove metals

The bigger picture: insight on industrial logistics, scalability, and accessibility



The infrastructure for physicochemical processes is complex

React



Absorb



Bind



Infrastructure



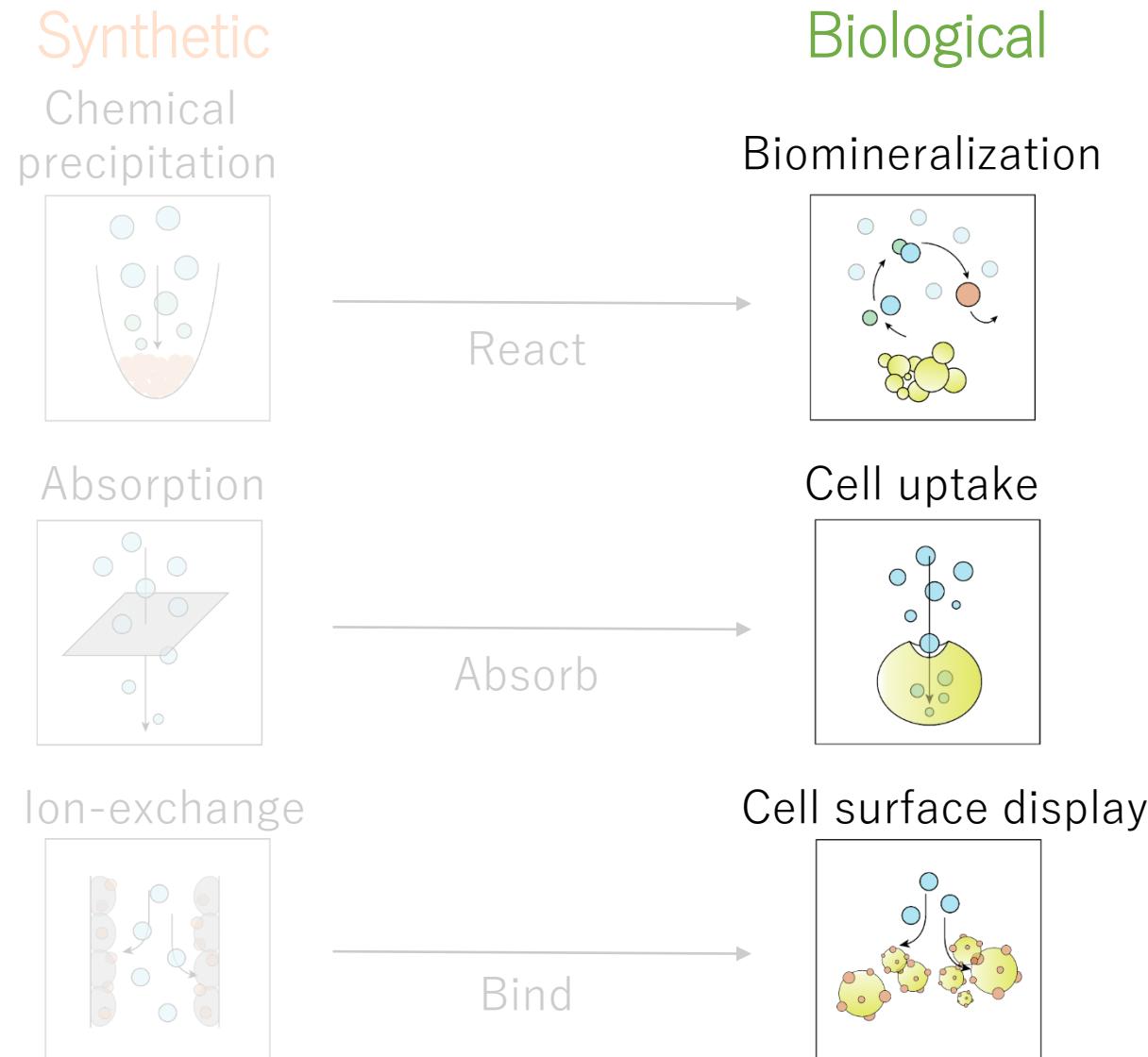
Modularity



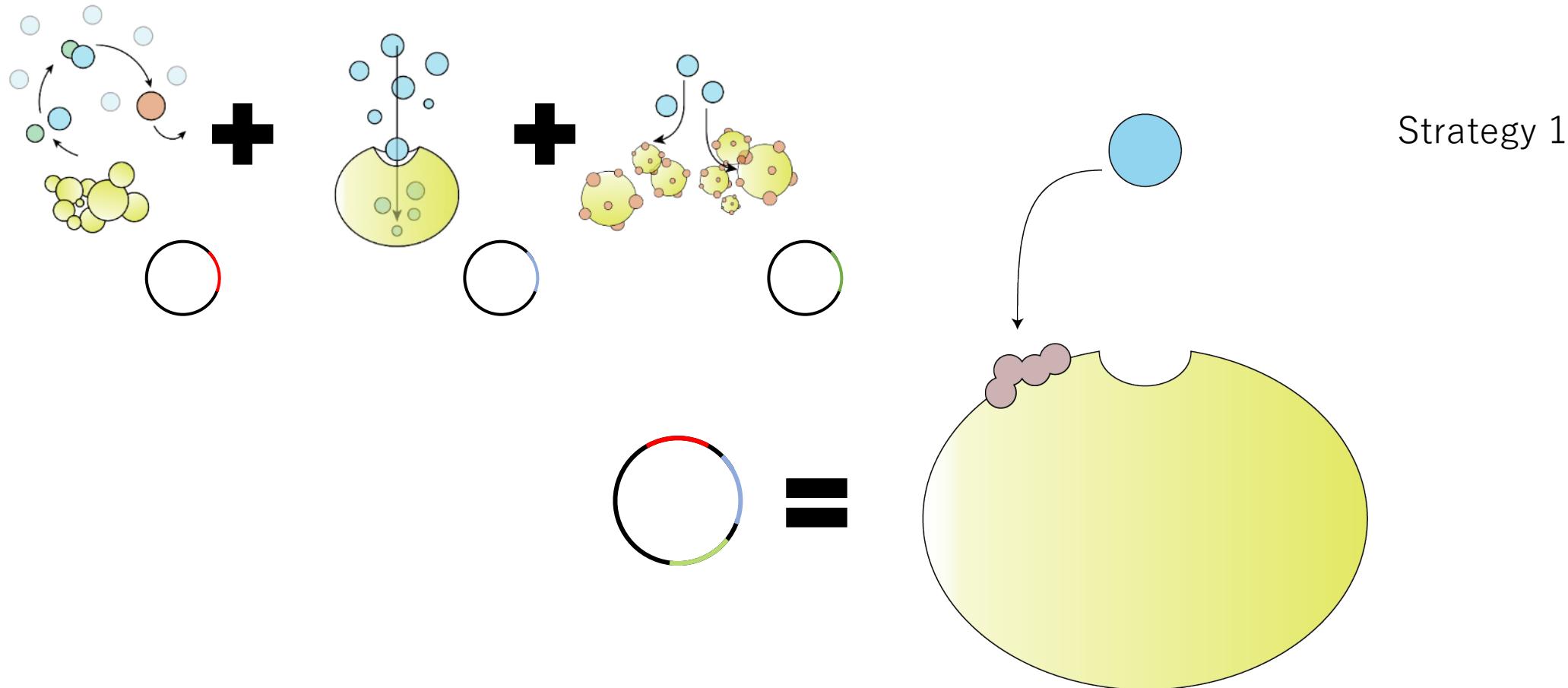
Maintenance



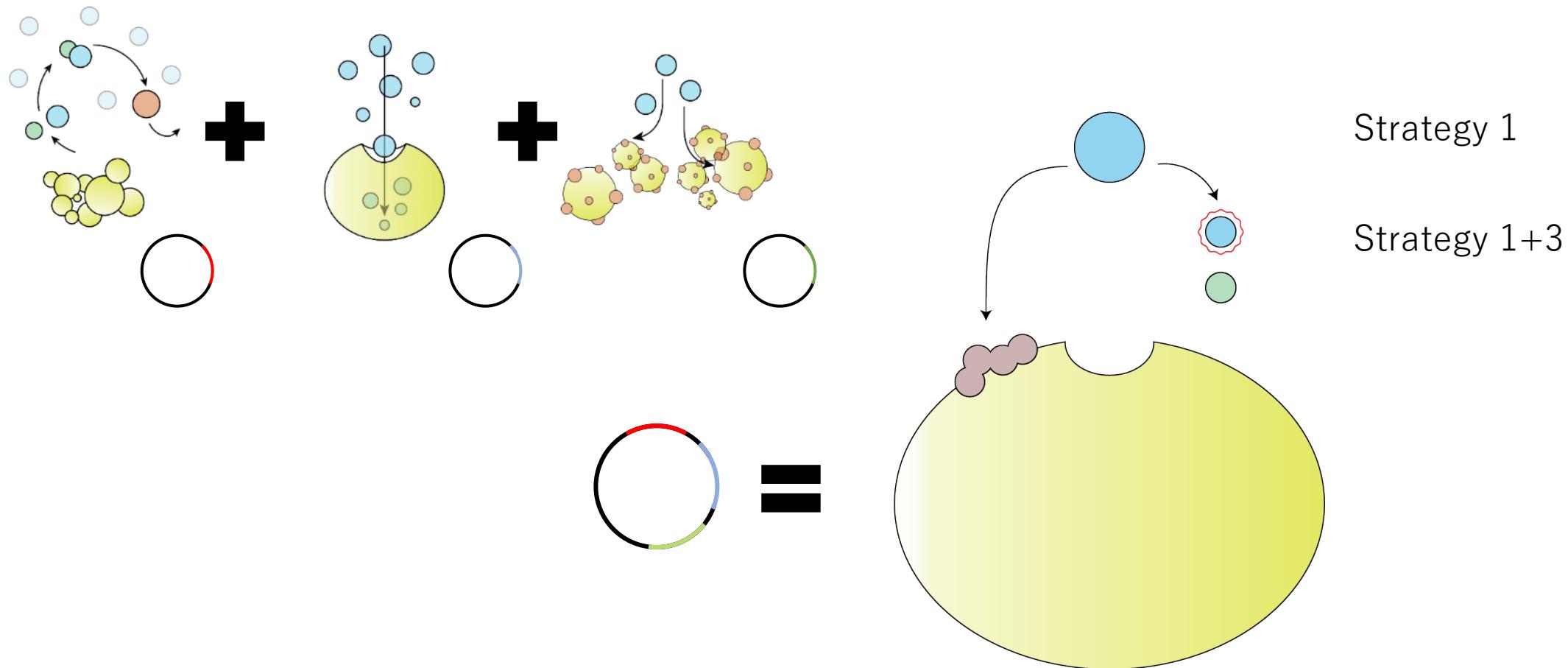
The bigger picture: insight on industrial logistics, scalability, and accessibility



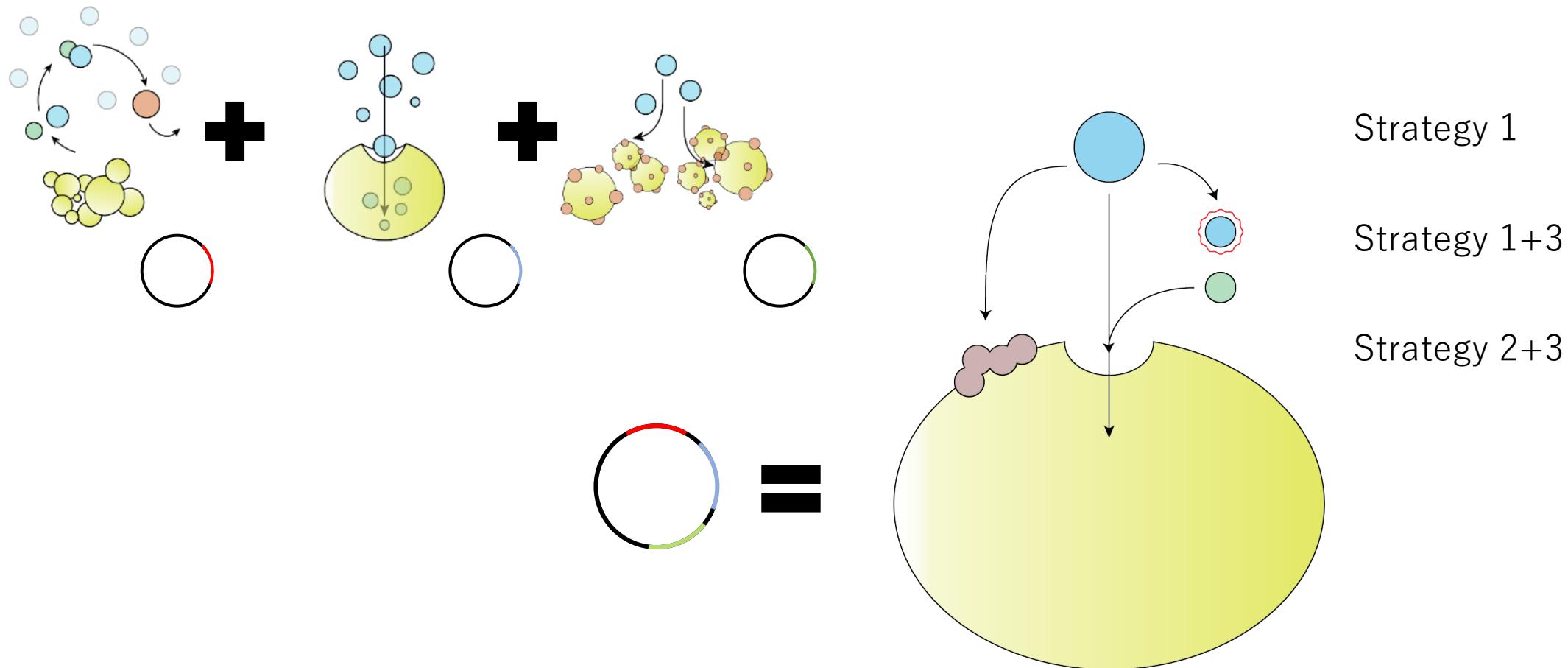
Yeast are both the factories and engineers,
they only need instructions



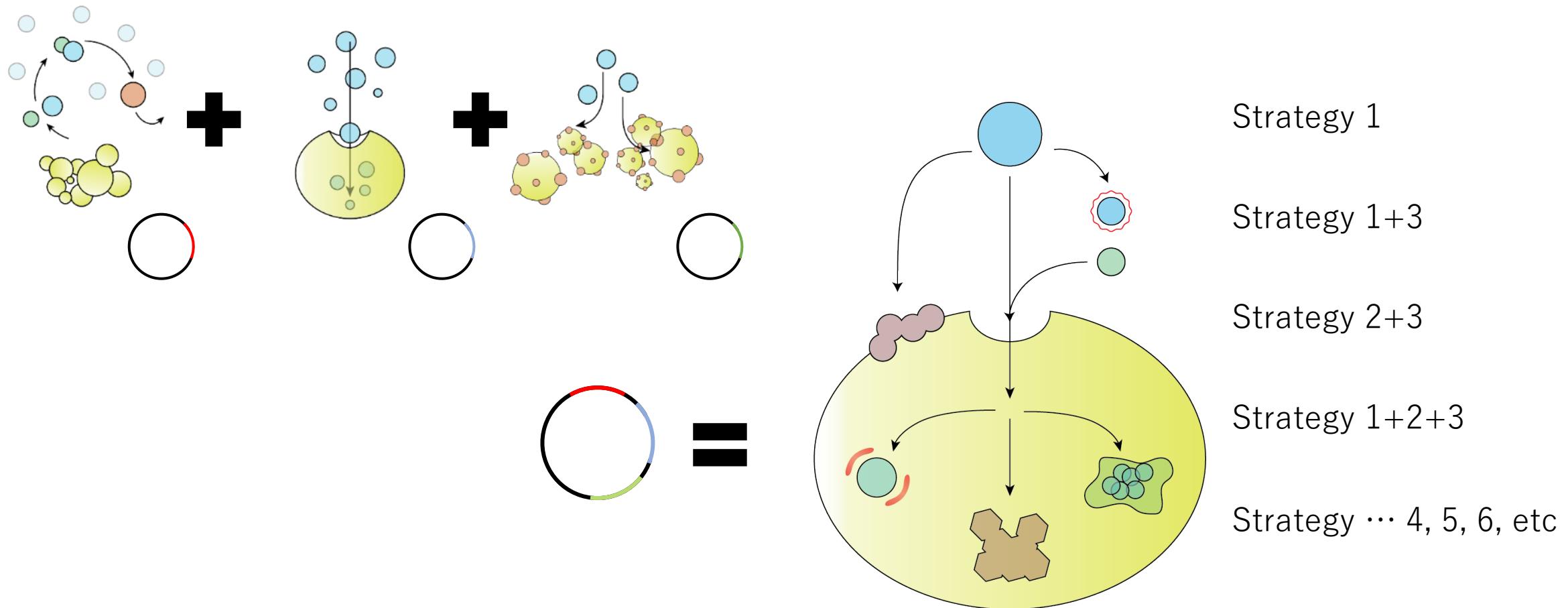
Yeast are both the factories and engineers,
they only need instructions



Yeast are both the factories and engineers,
they only need instructions



Yeast are both the factories and engineers, they only need instructions



Cost of physicochemical processes versus expected cost of yeast

¹

method	cost (US\$)
chemical precipitation	\$ 20–500
ab(d)sorption	\$ 50–150
ion-exchange	\$ 50–200

metal	EPA limit (ppm)	metal mass (g)	yeast mass (g)	culture volume (L)	cost (US\$)	10 OD ₆₀₀ 1% yeast biomass ² 16 cents per L
Cr	0.1	100	10000	20000	\$ 3200	
Mn	0.05	50	5000	10000	\$ 1600	
Fe	0.3	300	30000	60000	\$ 9600	
Co*	0.1	100	10000	20000	\$ 3200	
Ni	0.1	100	10000	20000	\$ 3200	
Cu	1.3	1300	130000	260000	\$ 41600	
Zn	5	5000	500000	1000000	\$ 160000	
As	0.01	10	1000	2000	\$ 320	
Cd	0.005	5	500	1000	\$ 160	
Hg	0.002	2	200	400	\$ 64	
Pb	0.015	15	1500	3000	\$ 480	

¹ Gupta, Vinod Kumar, et al. "Chemical treatment technologies for waste-water recycling—an overview." *Rsc Advances* 2.16 (2012): 6380-6388.

² Harrison, Roger G., Paul Todd, Paul W. Todd, Scott R. Rudge, and Demetri P. Petrides. 2015. *Bioseparations Science and Engineering*. Oxford University Press.

* cobalt toxicity limit was inferred

Cost of physicochemical processes versus expected cost of yeast

¹

method	cost (US\$)
chemical precipitation	\$ 20–500
ab(d)sorption	\$ 50–150
ion-exchange	\$ 50–200

metal	EPA limit (ppm)	metal mass (g)	yeast mass (g)	culture volume (L)	cost (US\$)	10 OD ₆₀₀ 1% yeast biomass ² 16 cents per L
Cr	0.1	100	10000	20000	\$ 3200	
Mn	0.05	50	5000	10000	\$ 1600	
Fe	0.3	300	30000	60000	\$ 9600	
Co*	0.1	100	10000	20000	\$ 3200	
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10 OD₆₀₀
1% yeast biomass
² 16 cents per L

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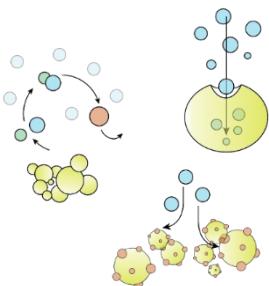
* cobalt toxicity limit was inferred

The potential impact of the yeast market on water remediation



³ 195 billion liters of yeast

1.07 million tons of yeast



1% capture capacity dry weight

10,000 tons of metal removed

9.8 trillion liters of water remediated
at 1 ppm (EPA limit)

or

~50X ratio of yeast culture to water
remediated



⁴ **3.9 million** olympic swimming pools

⁵ 309 thousand pools &
10 million residential pools in the US

³ Barth-Haas Group. 2019. "Worldwide Beer Production, 2017 | Statistic." Statista. May 12, 2019.
<https://www.statista.com/statistics/270275/worldwide-beer-production/>

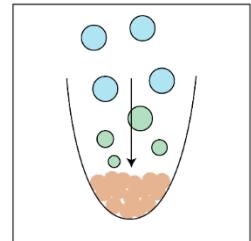
⁴ <https://www.livestrong.com/article/350103-measurements-for-an-olympic-size-swimming-pool/>

⁵ <https://www.thespruce.com/facts-about-pools-spas-swimming-safety-27371>

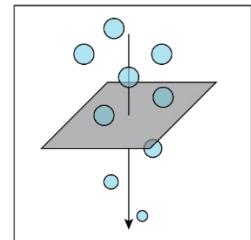
Biological platforms provide a natural alternative to physicochemical technologies

Synthetic

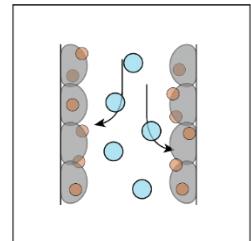
Chemical precipitation



Absorption

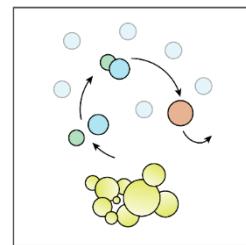


Ion-exchange

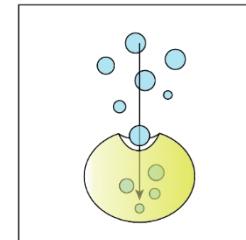


Biological

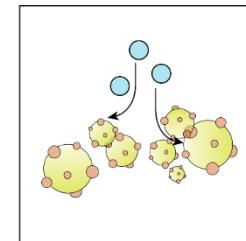
Biomineralization



Cell uptake



Cell surface display

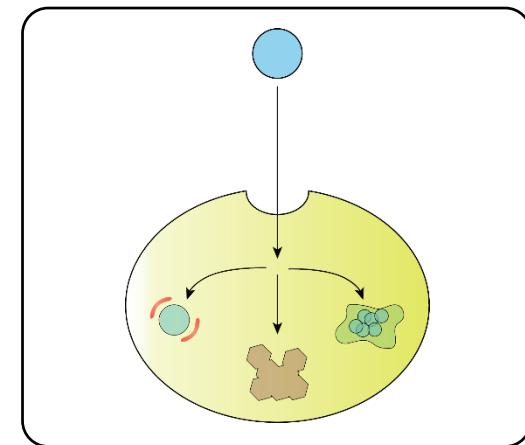


React

Absorb

Bind

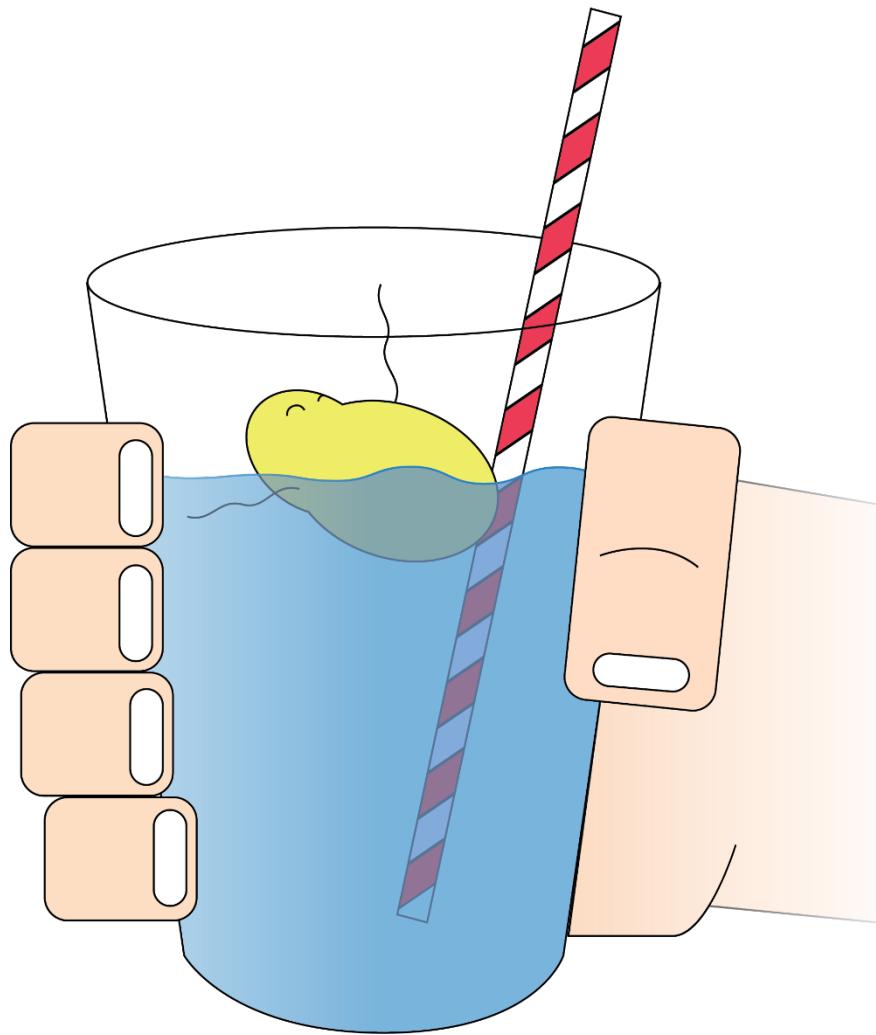
Fully customized yeast strain



Real world waste treatment

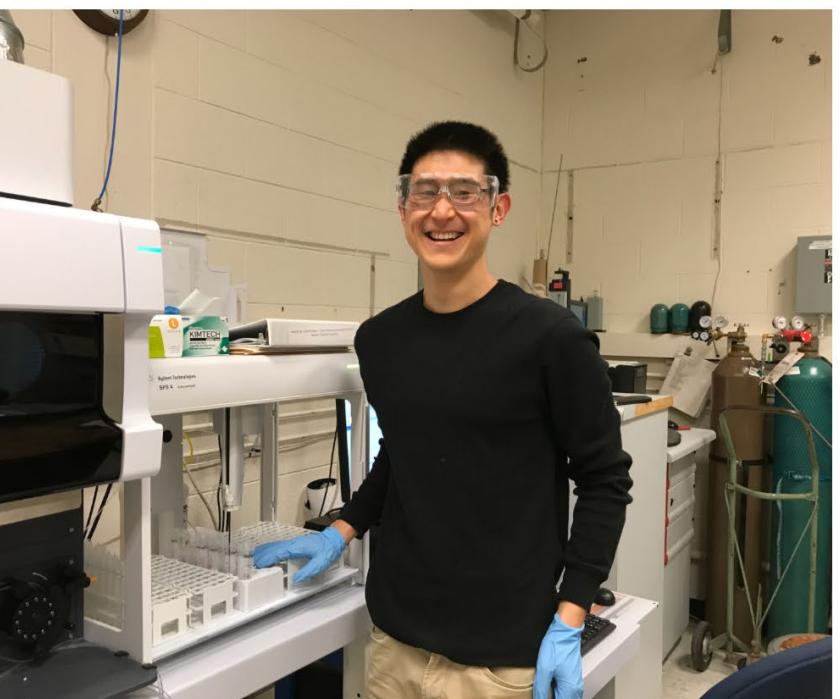


Happy Drinking



Thank you

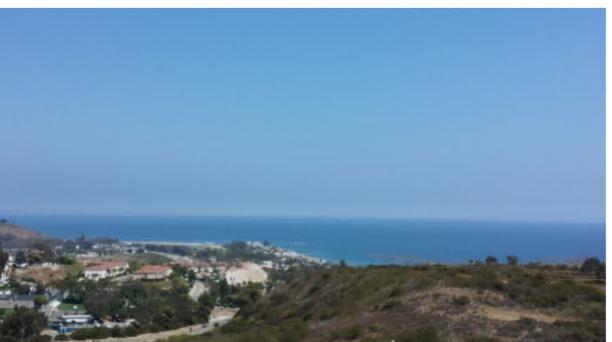
Belcher Lab	Committee Members	Facilities
• Nimrod Heldman • Jifa Qi • Hiroshi Atsumi • Briana Dunn • Neel Bardhan • Griffin Clausen • Shengnan Huang • Swati Kataria • Alan Ransil • Uyanga Tsedev • Geran Zhang • Jackie Ohmura • William Records	• Tyler Toth • Eric Lehnardht • Shalmalee Pandit • Jelle van Der Hilst • Jeremy Zhong • Ching-Wei Lin • Shuya Wei • Peter Jansen • Ngozi Eze • Erin Reynolds	• Angela Belcher (Advisor) • Dane Wittrup (Chair) • Cathy Drennan
		• Amon Lab
	• Christopher Brennan • Summer Morrill • Dan Corbi • Hilla Weidberg	• CMSE – ICP, TEM, SEM, XRD, Fluor. • DCIF – NMR • Koch Flow Core • Koch Microscopy Core • Koch High Throughput Core • Koch Nanotechnology Core • Whitehead Keck Microscopy
	• BE Comm Lab	• Funding
	• Prerna Bhargava • Diana Chien • Sean Clarke • Comm Fellows	• NSF Graduate fellowship • Bose Grant • CEHS Pilot Grant • Shell

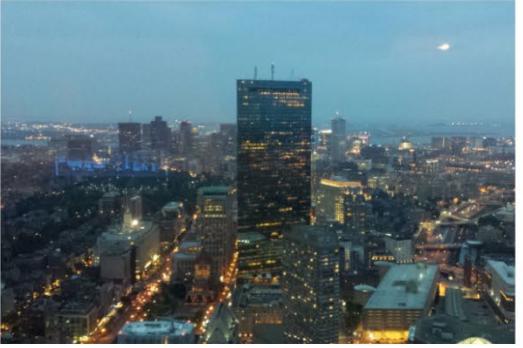


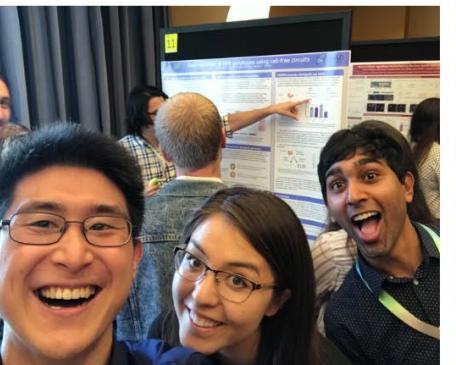












Questions

