



Final Report

AI Guided Alloy Design

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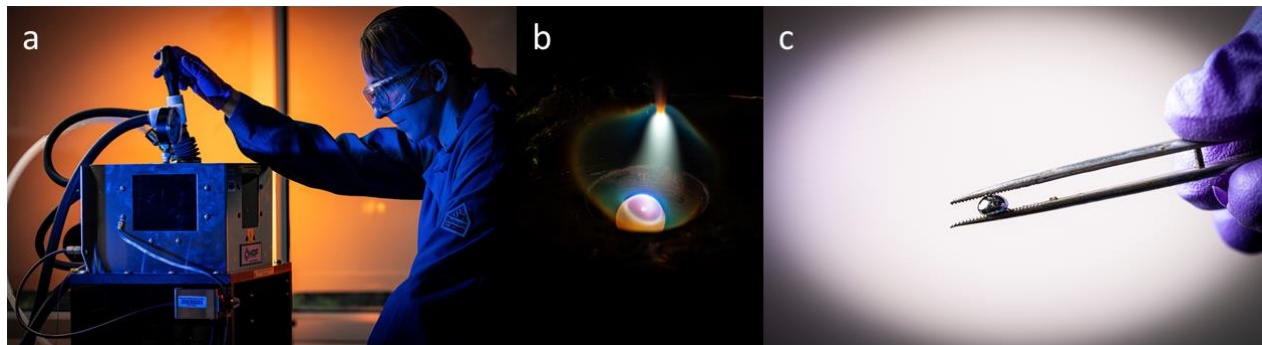


Figure 1: An overview of the arc melting process including (a) the MRF arc melting furnace (b) a close up of a molybdenum alloy being melted and (c) the final form factor of arc melted ingots

I. Project Overview

Multiple principal element alloys (MPEAs) are a class of alloy that frequently have desirable properties such as high strength, hardness, and temperature stability. By mixing roughly equi-atomic amounts of four or more elements, targeted FCC and BCC structures are formed, theoretically enabled by the high entropy of mixing multiple base metals. While these materials have shown promise, the potential compositional variations are massive, making machine learning guided exploration of the space promising. However, this is complicated by the relatively limited availability of materials properties data.

The Clancy Group at Johns Hopkins University Whiting School of Engineering has successfully demonstrated the use of the Physical analytics pipeline, PAL 2.0 model, to efficiently navigate complex compositional spaces compared to Edisonian approaches. PAL 2.0 combines a physics-based surrogate model with Bayesian optimization. PAL 2.0 uses a physics-based hypothesis to make predictions using neural networks. Prior work has generated predictions of high hardness MPEAs, however, these predictions are largely unverified with synthesis and experimental characterization.

The project had two main objectives:

1. Fabricate samples of compositions predicted by the Clancy Group. Predictions of high hardness (greater than 1000 Vickers hardness) MPEA compositions were made using the PAL 2.0 framework from existing data sets. JHU/APL arc melted small MPEA ingots. Raw material forms (powders, foils) were mixed in mass percentages predicted by the models. These materials were then melted to form a single alloy ingot using an MRF SA-200 arc melter.
2. Validate the PAL 2.0 model for high hardness MPEAs by measuring and comparing to experimental data. In this closed-loop discovery process, improvements are made by informing the model with actual experimental data. The impact of this process is measured by its ability to predict experimentally validated material properties. In this effort, MPEAs were characterized with Vickers hardness testing.

II. Technical Accomplishments

Task 1: Experimental synthesis of proposed MPEAs

A total of 20 samples was fabricated from the recommendations list provided by the Clancy Group (Attachment 1). Samples were prioritized for synthesis based on availability of elements (no

materials were purchased on this project) and perceived ease of synthesis. For example, samples containing Manganese tend to vaporize during melting making synthesis of a targeted compositions difficult.

Table 1 contains a list of samples that were synthesized, the starting weight of the raw elements, and the final ingot weight.

Some of the samples lost significant mass during synthesis, particularly samples that contain a mixture of light elements (aluminum and manganese) with refractory elements (molybdenum and tungsten). This is expected given that aluminum boils at 2,470 °C and tungsten doesn't boil until temperatures above 3,000 °C. In the future, alloying strategies can be used to mitigate these losses, but these investigations were outside the scope of this project.

Table 1: Synthesized samples including model-recommended elemental weight percentage. The target mass for each sample was 700 mg.

Sample #	Recommended Elemental Comp (wt%)	Sample Masses (mg)				Post Melt Weight (mg)
170	Fe Mo Si 31 55 14		Fe	Mo	Si	Total
		Target:	218	382	100	700
		Measured:	217	382	102	701
118	Al Fe Mo 15 29 55		Al	Fe	Mo	Total
		Target:	108	206	386	700
		Measured:	110	207	385	702
283	Al Mn Mo 16 32 51		Al	Mn	Mo	Total
		Target:	114	226	360	700
		Measured:	115	225	362	702
42	Co Si Ta 23 11 67		Co	Si	Ta	Total
		Target:	158	74	468	700
		Measured:	161	72	468	702
114	Co Fe Si 39 41 20		Co	Fe	Si	Total
		Target:	271	285	143	700
		Measured:	271	287	146	704
53	Co Si Ta 22 11 67		Co	Si	Ta	Total
		Target:	153	75	472	700
		Measured:	155	78	472	705
	Al Mn Mo		Al	Mn	Mo	Total

283 Redo	16 32 51					Target:		114	226	360	700		580			
						Measured:		114	255	363	732					
3	Co	Fe	Si	Ti	W			Co	Fe	Si	Ti	W	Total	656		
	5	6	4	17	68	Target:		38	42	27	119	475	700			
						Measured:		41	43	29	121	475	708			
89	Al	Co	Fe	Mo	Si	Ta			Al	Co	Fe	Mo	Si	Ta	Total	702
	5	3	6	28	3	55	Target:		36	22	45	195	18	383	700	
						Measured:		36	26	48	198	18	383	708		
954	Al	Co	Mo	V					Al	Co	Mo	V	Total		711	
	17	38	29	15	Target:		121	269	202	108	700					
						Measured:		120	269	203	109	701				

780	Al	Cr	Mo					Al	Cr	Mo			Total	
	15	31	54				Target:	107	214	379			700	
							Measured:	107	214	379			699	668
72	Fe	Si	V	W				Fe	Si	V	W		Total	
	13	3	19	65			Target:	90	23	134	453		700	
							Measured:	89	23	138	456		705	644
9	Al	Co	Ni	Si				Al	Co	Ni	Si		Total	
	22	31	33	14			Target:	154	215	230	101		700	
							Measured:	156	218	231	101		706	432
420	Cr	Mn	Mo					Cr	Mn	Mo			Total	
	26	27	47				Target:	184	186	329			700	
							Measured:	186	202	332			720	530
461	Al	Co	V					Al	Co	V			Total	
	20	44	36				Target:	138	310	253			700	
							Measured:	135	311	253			699	646
233	Al	Co	Ti					Al	Co	Ti			Total	
	20	45	35				Target:	140	316	244			700	
							Measured:	140	318	244			702	698
430	Al	Co	Mo	Ti				Al	Co	Mo	Ti		Total	
	17	40	24	18			Target:	121	280	171	127		700	
							Measured:	123	284	175	126		706	701
489	Al	Co	V					Al	Co	V			Total	
	20	45	36				Target:	137	313	251			700	

		Measured:	137	310	251	699	675
149	Cr Hf Si 20 70 10	Cr Hf Si Target: 138 491 71 Measured: 136 492 72	Total 700 700				639
130	Al V W 10 20 70	Al V W Target: 67 141 493 Measured: 68 143 490	Total 700 701				300

Task 2: Characterization and analysis of synthesized alloys

A total of seven samples were chosen for testing with Vickers hardness testing to compare to the predictions. Samples were prioritized from Table 1 based on how close the final mass to the target mass, thus ensuring the composition of the sample is close to that predicted by the model. Samples that lost significant mass were not tested because their final composition was unknown. Between three and five indents were taken on each sample. Indents were performed at either 300 or 500 grams of indent force. In previous efforts all samples were tested at 500 grams. However, some of the very high hardness samples experienced brittle cracking during indentation that can be mitigated by decreasing indentation load. The results from these tests are listed in Table 2.

Table 2: Measured and predicted hardness values for synthesized materials

Sample #	Indent	Force (g)	Measured VHN	Predicted VHN
42	1	300	1185	
	2	300	1101	
	3	300	1093	
	4	300	1175	
	Average	300	1137	
53	1	500	1170	
	2	500	1245	
	3	500	1363	
	4	500	1309	
	Average	500	1269	
72	1	300	880	
	2	300	992	
	3	300	872	
	4	300	864	
	Average	300	900	
114	1	500	974	
	2	500	952	
	3	500	940	

	4	500	933	
	Average	500	950	894
118	1	300	750	
	2	300	728	
	3	300	711	
	4	300	653	
	Average	300	709	
170	1	500	1297	
	2	500	1323	
	3	500	1387	
	4	500	1081	
	Average	500	1263	
489	1	300	475	
	2	300	479	
	3	300	455	
	4	300	494	
	Average	300	476	

III. Important Findings and Conclusions

- The PAL 2.0 model is able to predict high hardness materials in the MPEA design space. Through three rounds of predictions (the first two happening on previous efforts), the hardness values for synthesized materials has increased. Multiple materials had Vickers hardness values (VHN) over 1200 VHN, exceeding anything in the initial training data set (Figure 1).
- Additional considerations need to be paid to the idea of “synthesizability”. Several interesting compositions predicted by the model contain elements of very different melting temperatures making JHU/APL’s rapid arc melting approach difficult.

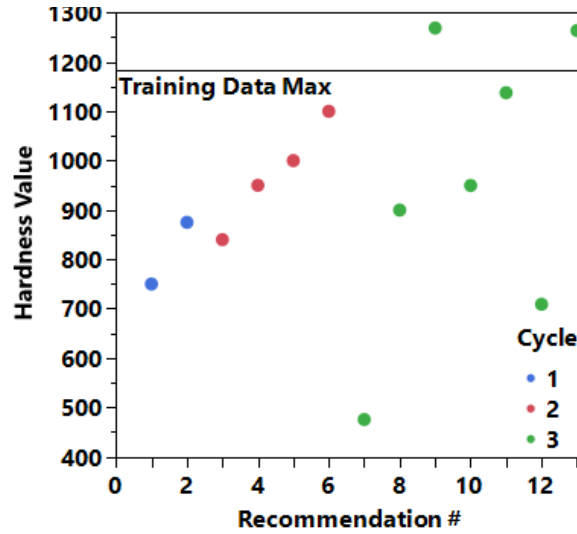


Figure 2: Experimentally measured hardness values for each recommended composition. Plot includes prior data from cycles 1 and 2. Within each cycle, samples are ordered by the model predicted hardness from low to high

IV. Implications for Further Research

1. Two proposed approaches can be taken to more accurately synthesize some of the predicted compositions. The first is to incorporate some of the physics behind synthesizability (i.e., melting temperature) into the model; the second is to explore additional processing routes (quartz ampule melting, multiple alloying steps) into the synthesis approach.
2. Given the model has predicted high hardness materials that are relatively cheap, light, and straightforward to synthesize, effort should be taken to scale up the material from the small ingot used for *discovery* to a *useful material system* in an application like hard coating for wear resistance.
3. There are several other metallic microstructures like B2 phases for strengthening that potentially possess similarities to the design framework used for this project. These materials have demonstrated high temperature strength and would be useful for several applications where room temperature ductility is also needed. Taking the lessons learned from this project and applying it to a more complex, high impact material class is worth investigating.