# Contribution of Lattice Distortion to Solid Solution Strengthening in a Series of Refractory High Entropy Alloys



H. CHEN, A. KAUFFMANN, S. LAUBE, I.-C. CHOI, R. SCHWAIGER, Y. HUANG, K. LICHTENBERG, F. MÜLLER, B. GORR, H.-J. CHRIST, and M. HEILMAIER

We present an experimental approach for revealing the impact of lattice distortion on solid solution strengthening in a series of body-centered-cubic (bcc) Al-containing, refractory high entropy alloys (HEAs) from the Nb-Mo-Cr-Ti-Al system. By systematically varying the Nb and Cr content, a wide range of atomic size difference as a common measure for the lattice distortion was obtained. Single-phase, bcc solid solutions were achieved by arc melting and homogenization as well as verified by means of scanning electron microscopy and X-ray diffraction. The atomic radii of the alloying elements for determination of atomic size difference were recalculated on the basis of the mean atomic radii in and the chemical compositions of the solid solutions. Microhardness ( $\mu$ H) at room temperature correlates well with the deduced atomic size difference. Nevertheless, the mechanisms of microscopic slip lead to pronounced temperature dependence of mechanical strength. In order to account for this particular feature, we present a combined approach, using  $\mu$ H, nanoindentation, and compression tests. The athermal proportion to the yield stress of the investigated equimolar alloys is revealed. These parameters support the universality of this aforementioned correlation. Hence, the pertinence of lattice distortion for solid solution strengthening in bcc HEAs is proven.

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# I. INTRODUCTION

SINCE the introduction of high entropy alloys (HEAs),<sup>[1]</sup> the focus has especially been on their microstructural and mechanical properties.<sup>[2]</sup> Solid solution strengthening in such chemically complex alloys is of particular interest,<sup>[3]</sup> as these systems typically exceed the constitutional range spanned by classical theories for the description of solid solution strengthening in dilute<sup>[4,5]</sup> and more concentrated<sup>[6]</sup> binary alloys. Accordingly, some efforts were made to develop models succeeding the current state of the description of solid solution strengthening.<sup>[7–10]</sup> The approach presented by

parameter and, thus, the misfit with composition in binary solid solution towards multicomponent systems. In contrast, Varvenne et al.[10] presented a mechanistic, parameter-free, and predictive theory for the yield strength of random alloys and face-centered-cubic (fcc) HEAs, respectively, which was validated against data available for the CoCrFeMnNi system mainly contributed by Wu et al.[11] The fundamental assumption is that dislocations move through an effective medium having average lattice parameter, shear modulus, stacking fault energy, *etc.*<sup>[10]</sup> Strengthening accordingly arises from the interaction of dislocations with random, local concentration fluctuations around the average composition of the HEA.<sup>[10]</sup> This especially accounts for the fact that in the case of HEAs, solute and solvent cannot be distinguished due to their equal concentration. In the conclusion of Reference 10, alloys with maximized strength can be obtained by maximizing the concentration-weighted, mean-square misfit volume or the shear

modulus. Assuming the validity of Vegard's law and

using reasonable values for individual atomic radii, the

concentration-weighted, mean-square misfit volume,

Toda-Caraballo and P.E.J. Rivera-Díaz-del-Castillo<sup>[8]</sup> is

based on the theory by Labusch, [6] while it uses a

generalization of the gradual changes of the lattice

H. CHEN, A. KAUFFMANN, S. LAUBE, Y. HUANG, K. LICHTENBERG, and M. HEILMAIER are with the Institute for Applied Materials (IAM-WK), Karlsruhe Institute of Technology (KIT), Engelbert-Arnold-Str. 4, 76131 Karlsruhe, Germany. Contact email: hans.chen@kit.edu I.-C. CHOI and R. SCHWAIGER are with the Institute for Applied Materials (IAM-WBM), Karlsruhe Institute of Technology (KIT), Hermann-von-Helmholtz-Platz 1, 76344 Eggenstein-Leopoldshafen, Germany. F. MÜLLER, B. GORR, and H.-J. CHRIST are with the Institut für Werkstofftechnik, Universität Siegen, Paul-Bonatz-Str. 9-11, 57068 Siegen, Germany.

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which is used in this theory, can directly be transformed into the more common atomic size difference  $\delta^{[10]}$ :

$$\delta = \sqrt{\sum_{i} x_i \left(1 - \frac{r_i}{\bar{r}}\right)^2},\tag{1}$$

where  $x_i$  is the molar fraction of the alloying element,  $r_i$  is the atomic radius of the respective alloying element, and  $\bar{r}$  is the average atomic radius of the alloy. An extension of this procedure was introduced by Okamoto  $et\ al.^{[9]}$  by taking the individual mean-square atomic displacements of the alloying elements into account. They propose a linear relationship of the yield strength and the averaged mean-square atomic displacement.

Strictly, the proposed theory by Varvenne et al.[10] does only apply to dislocation slip with low impact of Peierls stress as it is observed for fcc metals and alloys. Nevertheless, lattice distortion might significantly attribute to solid solution strengthening in body-centered-cubic (bcc) HEAs as well, and the simplicity and elegance of the atomic size difference approach make it worthwhile testing for correlation of  $\delta$  with the plastic yield of bcc HEAs. In this case, the contribution to solid solution strengthening might not be driven by the interaction of the elastic distortion field of dislocations and the one of solute atoms but by the interaction of dislocation cores with random, local concentration fluctuations, as suggested by Rao et al.[12] Deviations from significant correlation might lead to the identification of (1) interesting model alloys with nonequimolar composition or (2) additional contributions by short-range ordering<sup>[13]</sup> as well as variations of local dislocation core structures<sup>[12]</sup> specific to certain alloys.

Recently, we presented a suitable processing route for obtaining an almost single-phase, bcc HEA NbMoCr-TiAl. [14] The basic design goal was to combine elements with high melting points (Nb, Mo) and, thus, high intrinsic creep strength with protective oxide scale forming elements (Cr, Al) while simultaneously reducing density (Al, Ti) by a proper choice of elements. Arc melting and homogenization treatment at 1300 °C for 20 hours lead to an almost single-phase microstructure that possesses significant plasticity above 600 °C. Moreover, the formation of combined  $\langle 111 \rangle - \langle \langle 001 \rangle$ -fiber texture components with respect to the loading axis developed during compression testing at these temperatures reveals dislocation slip with (111) Burgers vectors, as expected for bcc materials. The alloy exhibits a reasonable high-temperature strength. Furthermore, the capability of increasing oxidation resistance by adding Cr and Al has been verified for the present and related alloys. [15-18]

In general, large differences in atomic size, which are expected for Nb-Mo-Cr-Ti-Al, as well as its ability to deform by dislocation slip makes it a favorable model system for studying the impact of the contribution of lattice distortion to solid solution strengthening in bcc HEAs. For this purpose, suitable changes in alloy composition and the according homogenization treatment for obtaining single-phase microstructure were established and are presented in this article. In order to

quantitatively describe the lattice distortion in the present alloys, lattice parameters and, thus, interatomic spacings were determined by means of X-ray diffraction (XRD). These data are used to recalculate the individual atomic radii of the alloying elements with the coordination of a bcc lattice. Finally, the according atomic size difference is correlated with mechanical properties exhibited by the alloy series. In order to account for the temperature-dependent strength of bcc metals, a suitable combination of microhardness ( $\mu$ H) for screening at room temperature (RT) as well as nanoindentation up to 400 °C and compression tests for temperatures beyond 400 °C were exemplarily conducted. The relevance of the  $\mu$ H results at RT of the respective alloys is stressed.

#### II. EXPERIMENTAL

The alloys investigated in this work were manufactured from elemental bulk material. Nb, Mo, Cr, Ti, and Al with a purity of 99.9, 99.96, 99, 99.8, and 99.9 pct, respectively, were mixed in the desired concentrations and then melted under Ar atmosphere, using an AM/0.5 arc melting furnace by Edmund Bühler GmbH (Germany). The furnace chamber was set to a pressure of  $5 \times 10^{-2}$  mbar and, alternatively, evacuated and flooded with Ar three times before pumping to a high vacuum of less than  $1 \times 10^{-4}$  mbar. The Ar pressure for arc melting was set to 600 mbar. A Zr getter was liquefied prior to each melting step to reduce residual oxygen content. The melted buttons were flipped and remelted at least five times and optionally cast into a water-cooled rod-shaped Cu mold with a diameter of 12 mm and length of 60 mm. Compositions of the equiatomic alloys in as-cast state were determined by inductively coupled plasma-optical emission spectrometry (ICP-OES). O and N contents were analyzed by means of carrier gas hot extraction analysis by a TC500 device (Leco Instrumente GmbH). Depending on the alloy composition, homogenization was performed under Ar atmosphere in a Gero HTRH 70-600/18 resistance tube furnace for 20 hours at 1200 °C, 1300 °C, 1400 °C, and 1500 °C. For this purpose, buttons and rods were cut into platelets with a thickness of 5 mm. The heating and cooling rate was 4.2 K/min. Subsequent to face grinding, XRD analyses (step width for acquisition 0.01 deg, acquisition time of 384 s/step) were carried out on a D2 Phaser system by Bruker equipped with a Lynxeye line detector. The lattice parameter was determined by extrapolating towards  $\theta = 90 \deg$ using the weighting function  $\frac{1}{2}(\cot^2\theta + \cot\theta \cdot \cos\theta)$ , according to Nelson-Riley. [19] For microstructural investigations, the samples were prepared by a standard metallographic procedure. After grinding with a series of abrasive papers of decreasing grain size, the samples were polished, etched with NH<sub>3</sub>, H<sub>2</sub>O<sub>2</sub>, and distilled water in a ratio of 1:1:1 at 50 °C for 5 minutes, and finalized by a vibratory polishing step for 8 hours, using a noncrystallizing oxide polishing suspension with pH 9.8 supplied by Struers. Scanning electron microscopy (SEM) and energy dispersive X-ray spectroscopy (EDX) analysis were performed on a Zeiss EVO50 system equipped with a Thermo Scientific EDX system. For each equiatomic alloy, porosity was investigated in terms of area fraction, which was determined from several binarized SEM micrographs. The investigated area was 1.2 mm², respectively.

Mechanical properties were characterized in terms of yield stress and hardness. Yield stress  $\sigma_y$  during compression tests was determined for 0.2 pct plastic strain at elevated temperatures in a range of 400 °C to 1200 °C at an initial engineering strain rate of  $10^{-3}$  s<sup>-1</sup>. The samples of  $(3 \times 3 \times 4.5)$  mm<sup>3</sup> were tested under vacuum in a Zwick Z100 electromechanical universal testing machine equipped with a vacuum furnace by Maytec. Strain was determined using strain gauges attached to the samples. Vickers hardness ( $\mu$ H) was investigated by a Qness Q10A+ microhardness tester with a load of 0.98 N. Mean values and standard deviations were obtained by analyzing 16 hardness indentations on each sample. Furthermore, a nanoindenter G200XP by Keysight Technologies GmbH equipped with a laser heating stage by SUR-FACE systems + technology GmbH was used to measure the temperature-dependent nanohardness (nH) between RT and 400 °C at a constant indentation strain rate of  $5 \times 10^{-2} \, \mathrm{s}^{-1}$ . Strain-rate sensitivity  $m = \left(\frac{\partial \ln \sigma_{y}}{\partial \ln \dot{\varepsilon}}\right)_{\varepsilon,T} = \left(\frac{\partial \ln \left(\frac{\mathrm{nH}}{C}\right)}{\partial \ln \dot{\varepsilon}}\right)_{\varepsilon,T}^{[20,21]} (\sigma_{y}: \text{ yield stress, } \dot{\varepsilon}:$ 

strain rate, T: temperature, nH: nanohardness, and C: constraint factor, usually  $C \approx 3^{[22]}$ ) was analyzed from RT to 300 °C by strain-rate jump tests with strain rates of  $5 \times 10^{-2}$ ,  $1.6 \times 10^{-2}$ , and  $5 \times 10^{-3}$  s<sup>-1</sup>. Shear modulus G of selected alloys was calculated by  $G = \frac{1}{1-p} \cdot \rho \cdot v_{\rm T}^2$  (p: porosity,  $\rho$ : density, and  $v_{\rm T}$ : velocity of transverse ultrasonic waves). The experimentally determined density  $\rho$  comprises an error that arises from the sample porosity p and is corrected by the factor  $\frac{1}{1-p}$  of the aforementioned equation. The density was determined by Archimedes' principle;  $v_{\rm T}$  was analyzed by ultrasonic phase spectroscopy on cubic samples with an edge length of 5 mm within the frequency range of 5 kHz to 10 MHz using two identical broadband ultrasonic wave transducers (Panametrics model V155) and a R3754A network analyzer by Advantest. Reference 23 provides further experimental details of ultrasonic phase spectroscopy.

# III. RESULTS AND DISCUSSION

# A. Microstructure

Prerequisite for the accurate assessment of solid solution strengthening is a homogenous microstructure of the alloys under investigation. In the following, this is exemplified with the three equimolar compositions that were manufactured for this work. All nonequiatomic compositions were treated in similar ways in order to ensure homogeneity of the solid solution. The starting point for the development was the aforementioned

homogeneous, quinary, equimolar NbMoCrTiAl alloy, which can be manufactured by arc melting and subsequent homogenization treatment. After heat treatment of the as-cast state at 1300 °C for 20 hours, an almost single-phase microstructure, as presented in Figure 1(b), can be obtained for this alloy. The XRD pattern in Figure 2(b) reveals the bcc crystal structure (W prototype, Strukturbericht A2).

During development of this alloy, several phases competing with the bcc solid solution were identified. Most importantly, the formation of the hexagonal Cr<sub>2</sub>Nb-Laves phase (MgZn<sub>2</sub> prototype, Strukturbericht C14) has to be suppressed to retain single-phase bcc. In NbMoCrTiAl, the Laves phase appears for homogenization heat treatments below 1300 °C as well as during compression testing of as-homogenized material at 1200 °C. Moreover, (Mo, Nb)<sub>3</sub>Al (Cr<sub>3</sub>Si prototype, Strukturbericht A15) forms during annealing at all temperatures tested for homogenization up to now.\*

Nevertheless, this phase appears only as a fringe of tiny dimension in the vicinity of some grain boundaries. For 1300 °C and 20 hours, its volume fraction is below 0.3 vol pct in total and, thus, can be considered to be not a relevant factor for the mechanical properties determined in this work. Some trace contents of (Al, Cr)<sub>3</sub>Ti (Cu<sub>3</sub>Au prototype, Strukturbericht D0<sub>9</sub>) as well as B2 superstructure can occasionally be found for a variety of processing conditions but are believed as well to be not relevant for mechanical property determination. All aforementioned phases were identified based on global XRD patterns as well as locally by means of EBSD patterns.

Since most of the competing phases are expected for varying alloy compositions derived from NbMoCrTiAl, only specific changes of alloying elements were conducted in this research. For the suppression of Laves phase formation, either Nb or Cr content was successively reduced toward the quaternary, equimolar alloys NbMoTiAl and MoCrTiAl. In the case of NbMoTiAl, a significantly higher homogenization temperature was needed in order to achieve a homogeneous bcc solid solution, namely, 1500 °C for 20 hours. The according microstructure and the XRD pattern are presented in Figures 1(a) and 2(a), respectively. In contrast, annealing at only 1200 °C for 20 hours is necessary to obtain single-phase, bcc MoCrTiAl, as shown in Figures 1(c) and 2(c). The concentrations of Mo, Ti, and Al were kept constant. It was expected that increasing Mo, Ti, and Al in Nb-Mo-Cr-Ti-Al could favor the formation of (Mo, Nb)<sub>3</sub>Al and (Al, Cr)<sub>3</sub>Ti, while reducing the amount of these particular elements could lead to stabilization of Cr<sub>2</sub>Nb.

Table I summarizes all investigated alloys, their compositions, and the necessary heat treatments to obtain almost single-phase microstructures with residual phases having a total volume fraction of less than 0.5 vol pct. In certain cases, higher homogenization

<sup>\*</sup>This phase remained unidentified in our previous investigation.<sup>[14]</sup>

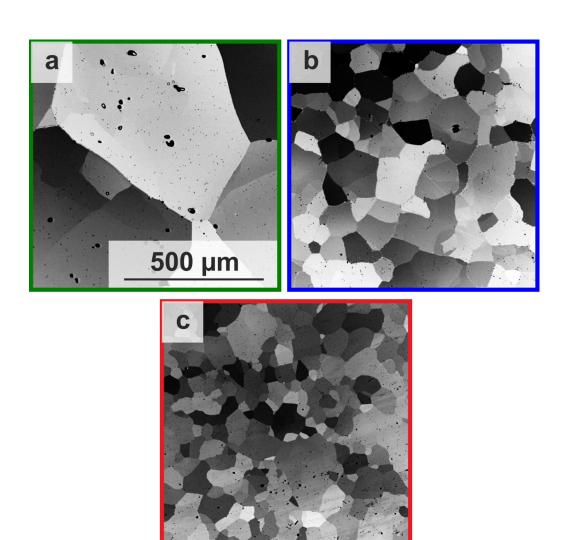


Fig. 1—Microstructure of alloys with equiatomic composition in homogenized condition, obtained by SEM using BSE contrast imaging (same magnification for all images): (a) NbMoTiAl annealed at 1500 °C, (b) NbMoCrTiAl annealed at 1300 °C, and (c) MoCrTiAl annealed at 1200 °C, each for 20 h. Certain porosity (black features within the grains) is typical for cast material (<2 vol pct in all presented alloys) and was randomly distributed in both as-cast (not presented in this article) and as-homogenized states.

temperatures were used as well in order to exclude effects by the grain size of the material and possible influences by the homogenization temperature (e.g., different extent of any type of ordering) on the mechanical response. For NbMoCrTiAl, several batches were tested for assessment of the reproducibility.

# B. Atomic Radii of the Alloying Elements

In principle, three degrees of complication of the description of the variation of the lattice in a HEA are possible. The easiest assumption is that the HEA has a defined lattice parameter without a variation on atomic length scale at all. This assumption is physically not meaningful and would lead to the conclusion that there is no hardening since the solute atoms do not have a distortion field. Further complication is possible by assuming an effective lattice parameter of the alloy as well as its variation caused by the individual size of the alloying elements; this variation might be used as a

measure for the lattice distortion. Even more complex is a description where each species of alloying elements has an average size and an according variation. In this case, the exact size of an atom is determined by the actual environment of the atom (not by its species alone as assumed in the previous case). Despite the latter case being the most realistic description and providing several quantities for the lattice distortion, the second assumption already allows for a possible description of solid solution strengthening. Furthermore, the second case is rather easy to be assessed when the atomic radii of the alloving elements within the solid solution are known. As pointed out by Varvenne et al., [10] the assessment of atomic size difference based on the (tabulated) atomic radii of the pure elements might lead to arbitrary results when the pure elements crystallize in a different crystal structure as compared to the HEA solid solution. In such case, the binding state of the alloying element is different and, accordingly, a different contribution to the lattice distortion has to be expected.

In the present work, we try to obtain reasonable atomic radii by analyzing all obtained bcc alloys (coordination number CN = 8). In the case of the crystal structure being constant, the binding state remains rather constant. Furthermore, only the refractory elements Nb and Cr are changed in their concentrations, which should further reduce the possibility of varying amounts of directed binding. The situation might be different when Al or Ti content is changed over a wide composition range. For the analyses of the data, an overdetermined set of equations is formulated on the basis of (1) the chemical compositions  $x_{i,i}$  presented in Table I, where i is the respective alloying element and j the respective alloy; and (2) the mean atomic radius  $\bar{r}_i$  in the alloys obtained from  $\bar{r}_j = \sqrt{3}/4 \cdot a_j^{\text{bcc}}$  (CN = 8), where the  $a_j^{\text{bcc}}$ values are the lattice parameters presented in Table I, which were determined by XRD. Moreover, the atomic radii of the bcc crystallizing elements are included in the calculation as well (CN = 8). The overdetermined set of equations, thus, reads

$$\begin{pmatrix} 1 & 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & 0 & 0 \\ 0 & 0 & 1 & 0 & 0 \\ x_{\text{Nb},1} & x_{\text{Mo},1} & x_{\text{Cr},1} & x_{\text{Ti},1} & x_{\text{Al},1} \\ x_{\text{Nb},2} & x_{\text{Mo},2} & x_{\text{Cr},2} & x_{\text{Ti},2} & x_{\text{Al},2} \\ x_{\text{Nb},3} & x_{\text{Mo},3} & x_{\text{Cr},3} & x_{\text{Ti},3} & x_{\text{Al},3} \\ & \vdots & & & & & & & & & & & & & & \\ \end{pmatrix} \begin{pmatrix} r_{\text{Nb}} \\ r_{\text{Mo}} \\ r_{\text{Cr}} \\ r_{\text{Ti}} \\ r_{\text{Al}} \end{pmatrix} = \begin{pmatrix} r_{\text{Nb}} \\ r_{\text{Mo}} \\ r_{\text{Cr}} \\ \bar{r}_{1} \\ \bar{r}_{2} \\ \bar{r}_{3} \\ \vdots \end{pmatrix}$$

and is solved with respect to  $r_{Nb}$ ,  $r_{Mo}$ ,  $r_{Cr}$ ,  $r_{Ti}$ , and  $r_{Al}$  by a least-squares procedure. The overdetermined set of equations is, in principle, a multicomponent extension of Vegard's law. Figure 3 shows the comparison of the atomic radii of the elements (CN = 12 for Al and Ti, CN = 8 for Mo, Cr, and Nb as well as for an extrapolation for Ti designated as Ti\*), the mean radii

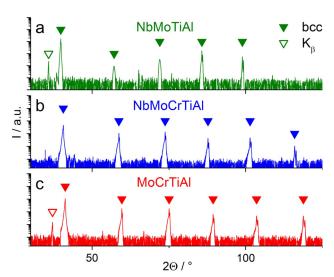


Fig. 2—XRD patterns with logarithmic intensity scale of alloys with equiatomic composition in homogenized condition: (a) NbMoTiAl annealed at 1500 °C, (b) NbMoCrTiAl annealed at 1300 °C, and (c) MoCrTiAl annealed at 1200 °C, each for 20 h. Peaks arising from residual Cu-K $\beta$  radiation are indicated by open triangles.

Composition in Atomic Percent of Solid Solutions in Annealed Condition Determined by Standard-Related EDX as well as Lattice Parameters of the Bcc Structure; O Contents were Exemplarily Analyzed for NbMoTiAl, NbMoCrTiAl, and MoCrTiAl: 780, 312, and 289 at. ppm, Respectively\*\*

|   |  |  |                         |  | $\chi_{i,j}$          |                      |                   |                           |
|---|--|--|-------------------------|--|-----------------------|----------------------|-------------------|---------------------------|
| j | Nominal Composition  | Annealing Condition                      | Nb                      | Mo   | Cr                    | Ti                   | Al                | $a_j^{\rm bcc}$ (nm)      |
| _ | NbMoTiAl   | 1400 °C, 20 h                            | $0.251 \pm 0.001$       | $0.248 \pm 0.002$  | 1                     | $0.253 \pm 0.002$    | $0.248 \pm 0.001$ | 1                         |
|   |  | $1500  ^{\circ}\text{C},  20  \text{h*}$ | $0.252 \pm 0.001$       | $0.249 \pm 0.001$  |                       | $0.252 \pm 0.001$    | $0.247 \pm 0.001$ | $3.208^{+0.018}_{-0.007}$ |
| 7 | $NbMoCr_{0.25}TiAl$  | 1300 °C, 20 h*                           | $0.259 \pm 0.00.1$      | $0.233 \pm 0.00.1$   | $0.058 \pm 0.001$     | $0.230 \pm 0.001$    | $0.220 \pm 0.000$ | $3.198^{+0.002}_{-0.058}$ |
| 3 | NbMoCr <sub>0.5</sub> TiAl                                       | 1300 °C, 20 h*                           | $0.234 \pm 0.001$       | $0.226 \pm 0.003$  | $0.099 \pm 0.001$     | $0.216 \pm 0.001$    | $0.225 \pm 0.001$ | $3.175^{+8.018}_{-0.013}$ |
|   |  | 1400 °C, 20 h                            | $0.231 \pm 0.000$       | $0.223 \pm 0.001$  | $0.101 \pm 0.001$     | $0.216 \pm 0.000$    | $0.228 \pm 0.000$ |                           |
| 4 | NbMoCr <sub>0.75</sub> TiAl                                      | 1300 °C, 20 h*                           | $0.234 \pm 0.001$       | 0  | $0.157 \pm 0.003$     | $0.204 \pm 0.001$    | $0.197 \pm 0.001$ | $3.165^{+0.035}_{-0.005}$ |
| 2 | NbMoCrTiAl   | 1300 °C, 20 h*                           | $0.219 \pm 0.001$       | $0.201 \pm 0.002$  | $0.200 \pm 0.002$     | $0.192 \pm 0.000$    | $0.187 \pm 0.001$ | $3.149^{+8:821}_{-0.004}$ |
|   |  | 1400 °C, 20 h                            |                         | $0.197 \pm 0.001$  | $0.195 \pm 0.001$     | $0.191 \pm 0.000$    | $0.198 \pm 0.000$ | 10000                     |
|   |  | 1300 °C, 20 h♦                           | $0.205 \pm 0.001$       | $0.195 \pm 0.000$  | $0.197 \pm 0.001$     | $0.205 \pm 0.002$    | $0.199 \pm 0.001$ |                           |
| 9 | $Nb_{0.75}MoCrTiAl$  | 1300 °C, 20 h*                           | $0.173 \pm 0.001$       | $0.214 \pm 0.001$  | $0.208 \pm 0.000$     | $0.205 \pm 0.000$    | $0.200 \pm 0.001$ | $3.141^{+0.048}_{-0.011}$ |
| 7 | Nb <sub>0.5</sub> MoCrTiAl                                       | 1300 °C, 20 h*                           | $0.112 \pm 0.000$       |  | $0.221 \pm 0.001$     | $0.215 \pm 0.001$    | $0.220 \pm 0.000$ | $3.127^{+8:644}_{-0.011}$ |
|   |  | 1400 °C, 20 h                            | $0.110 \pm 0.000$       | $0.228 \pm 0.001$  | $0.218 \pm 0.001$     | $0.218 \pm 0.001$    | $0.226 \pm 0.002$ |                           |
| ∞ | $Nb_{0.25}MoCrTiAl$  | 1300 °C, 20 h*                           | $0.065 \pm 0.001$       | $0.241 \pm 0.002$  | $0.238 \pm 0.002$     | $0.234 \pm 0.001$    | $0.222 \pm 0.001$ | $3.114^{+0.006}_{-0.009}$ |
| 6 | MoCrTiAl   | $1200  ^{\circ}\text{C},  20  \text{h*}$ |                         | $0.253 \pm 0.001$  | $0.248 \pm 0.000$     | $0.245 \pm 0.001$    | $0.254 \pm 0.001$ | $3.101^{+8.018}_{-0.006}$ |
|   |  | 1300 °C, 20 h                            |                         | $0.254 \pm 0.002$  | $0.247 \pm 0.002$     | $0.242 \pm 0.000$    | $0.257 \pm 0.001$ |                           |
| * | *Used for recalculation of the atomic radii, **N content was bel | mic radii, **N content was be            | low the detection limit | low the detection limit for all tested alloys, ◆ highlights EDX on a different batch. [14] | highlights EDX on a c | lifferent batch.[14] |                   |                           |

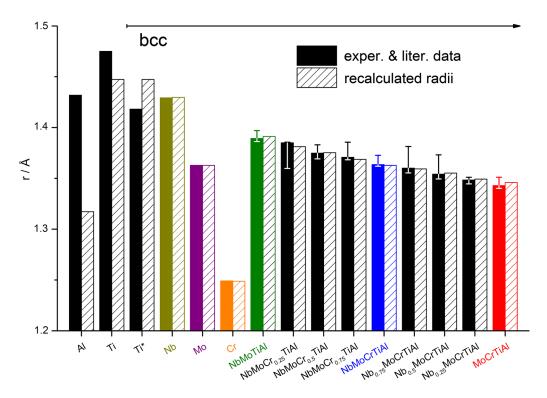


Fig. 3—Comparison of experimental and recalculated atomic radii in the elements  $r_i$  as well as investigated alloys  $\bar{r}_j$ . Literature data for Al, Ti (using CN = 12, respectively), Nb, Mo, and Cr (using CN = 8), represented by solid bars, are all taken from Ref. [25]. "Ti\*" designates an extrapolation of the atomic radius of bcc Ti corrected by thermal expansion down to RT (CN = 8) according to Ref. [24].

in the alloys deduced from the bcc lattice parameter (CN = 8), and the values obtained from solving the aforementioned overdetermined set of equations. Since Ti exhibits a bcc high-temperature modification, a second atomic radius for Ti based on the lattice parameter of the high-temperature bcc phase (CN = 8) and compensated by thermal expansion down to RT taken from Reference 24 is included as a reference. It is obvious that the atomic radii of all elements that are deduced from non-bcc crystal structures, namely, Al and Ti, are quite different from the recalculated atomic radii, whereas the radii of the bcc elements are well reflected by this approach. Good agreement of the recalculated atomic radii with the experimentally observed mean atomic radii in the alloys of the present contribution is found. It is pointed out that the error indication for the experimental mean atomic radii is conservatively chosen (taking the peak width into account) in all cases and might be significantly smaller in reality. Thus, the description of alloys within the Nb-Mo-Cr-Ti-Al system on the basis of  $r_{\text{Nb}} = 1.4297 \text{ Å}, \quad r_{\text{Mo}} = 1.3627 \text{ Å}, \quad r_{\text{Cr}} = 1.2488 \text{ Å}, \\ r_{\text{Ti}} = 1.4474 \text{ Å}, \text{ and } r_{\text{Al}} = 1.3172 \text{ Å} \text{ is rather good}$ and the linearity by Vegard's law seems to be reasonable—at least for variations of the Cr and Nb concentration. Further details of the binding states in such alloys have to be investigated in the future, especially with respect to individual variation of the radii of the elements, which will not be considered for the further course of this article. Also, the actual misfit volume was not experimentally determined within this work but would describe the extent of lattice distortion in an even more accurate way.

## C. Temperature-Dependent Mechanical Properties

Depending on the homologous temperature, bcc metals and substitutional solid solutions exhibit an intrinsic temperature-dependent yield stress that can be divided into a low-temperature, a plateau, and a high-temperature regime, as exemplarily shown for Fe-3 wt pct Si in References 26 and 27. This is schematically presented in Figure 4(a). At low temperatures, plastic deformation of bcc metals and alloys is thought to be mediated by the thermally activated formation and movement of kink pairs. The movement of kink pairs itself can be controlled by (1) either line tension when the critical kink width is very small and of the order of the dimension of the dislocation cores or the distance between minima in the Peierls potential or (2) by the elastic interaction of the kinks when the kink width is necessarily high.<sup>[28]</sup> Consequently, strength increases considerably with (1) decreasing temperature and (2) increasing strain rate. Above a certain temperature, usually referred to as "critical temperature" or "knee temperature,"[21,28] a strength plateau is reached. In this temperature regime, strength becomes virtually strainrate independent. The knee temperatures of the bcc principal elements of the present HEA system are about 130 °C for Nb, 275 °C for Mo, and 300 °C for Cr. [21,30] For the investigated alloys, a similar temperature range for the knee temperature is expected. When exceeding about 0.4·T<sub>m</sub> (T<sub>m</sub>: melting temperature), diffusional processes lead to rapid decrease of the yield stress. Mechanical strength becomes strain-rate dependent again. Thus, for the assessment of solid solution strengthening in bcc alloys, the strain-rate independent

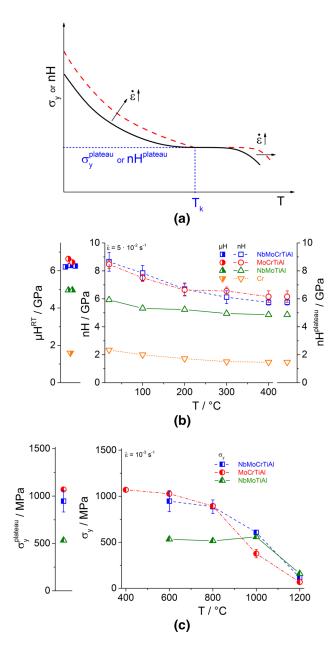


Fig. 4—(a) Scheme of the temperature-dependent yield stress of bcc metals and alloys. Experimental results for the temperature dependence of the equiatomic alloys MoCrTiAl, NbMoTiAl, and NbMoCrTiAl; (b) microhardness  $\mu H^{RT}$  was obtained at RT. Multiple data points represent  $\mu H^{RT}$  obtained for varying homogenization parameters and on different batches (Table I). Nanohardness nH was determined between RT and 400 °C in order to obtain the plateau nanohardness nH<sup>plateau</sup> = nH(400 °C). (c) Compression tests were performed at a consistent strain rate of  $10^{-3}~\rm s^{-1}$  between 400 °C and 1200 °C depending on the macroscopic ductility of the polycrystalline material in order to obtain the plateau yield stress  $\sigma_y^{\rm plateau}$  equal to  $\sigma_y^{600\,^{\circ}{\rm C}}$  (NbMoCrTiAl, NbMoTiAl) and  $\sigma_y^{400\,^{\circ}{\rm C}}$  (MoCrTiAl). The connecting lines are to guide the eyes. Error bars that do not exceed the symbol size are omitted.

plateau strength, frequently referred to as athermal strength, ideally should be investigated.

In order to account for temperature dependence of the mechanical properties, several mechanical tests were performed and are summarized in Figures 4(b) and (c), respectively. Compression tests performed from 400 °C

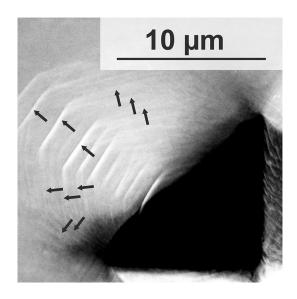


Fig. 5—Secondary electron contrast image of a Vickers indent (lower right corner) in NbMoCr<sub>0.5</sub>TiAl at RT, revealing slip traces in various directions in the vicinity of the indent; slip traces are highlighted by black arrows.

to 1200 °C on the equiatomic alloys in Figure 4(c) reveal the transition from plateau-like yield stress to creep-controlled strength. As shown in previous investigations, the predominant deformation mechanism in NbMoCrTiAl is dislocation slip up to a deformation temperature of 1200 °C. [14] Hence, this quinary alloy and its derivatives are well suited to discuss the temperature-dependent yield stress, also in the creep regime. Onset of the drop of strength is observed between 600 °C (MoCrTiAl) and 1000 °C (NbMoCrTiAl).

Due to microstructural features, NbMoCrTiAl and its derivatives do not reveal significant plasticity on a macroscopic scale at temperatures lower than 600 °C (NbMoTiAl, NbMoCrTiAl) and 400 °C (MoCrTiAl). As shown in Figure 1 and discussed in Reference 14, abnormal grain growth appears during homogenization, which might lead to increasing compatibility stresses between these large grains and the surrounding microstructure during compression testing. This is a possible reason for a limited ductility of the presented alloys. In addition, pores within the compression samples, developed during solidification after casting, further lead to an increase of internal stresses due to localization. Thus, crack formation is initiated, although, macroscopically, yield stress has not been reached. Nevertheless, µH and nanoindentation reveal plasticity (without cracking) on a local scale. Figure 5 exemplarily shows slip traces of NbMoCr<sub>0.5</sub>TiAl in the vicinity of a  $\mu$ H indent at RT.

By using nanoindentation, a temperature range between RT and 400 °C could be covered in order to reveal the plateau region of the yield stress in Figure 4(b). Further details about the temperature-dependent nH of pure Cr can be found elsewhere.<sup>[21]</sup> It has to be taken into account that indentation tests are typically affected by varying work-hardening behavior due to higher plastic strains when compared to uniaxial tests for yield stress determination. Typically, hardness

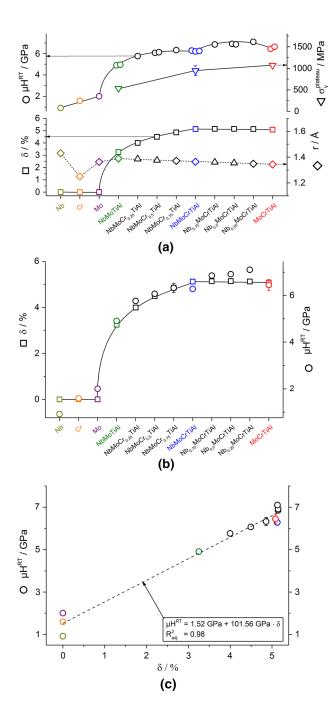


Fig. 6—(a) Assessment of atomic size r and atomic size difference  $\delta$  using recalculated radii (lower part of the figure) as well as of microhardness at RT  $\mu$ H<sup>RT</sup> and plateau yield stress  $\sigma_y^{\text{plateau}}$  (upper part of the figure). Multiple microstructural conditions (higher homogenization temperature, different batches) for various alloys are included here and do not reveal significant variation. (b) Correlation of  $\delta/\mu$ H<sup>RT</sup> and according element/alloy composition. (c)  $\mu$ H<sup>RT</sup> evolution with respect to  $\delta$  reveals linear correlation with  $\mu$ H<sup>RT</sup> = 1.52 GPa + 101.56 GPa· $\delta$  and  $R_{\text{adj}}^2 = 0.98$ . (d) Correlation of  $\delta$  and  $\sigma_y^{\text{plateau}}$  as well as nH<sup>plateau</sup> from Figs. 4(b) and (c), respectively. For all diagrams, error bars that do not exceed the symbol size are omitted. Color coding is consistent for all diagrams.

determined with a Berkovich indenter corresponds to a yield stress at about 7 pct plastic strain. [22] Hence, contribution of work hardening to strengthening that occurs from indentation eventually varies with alloy composition. Further alloy development with respect to

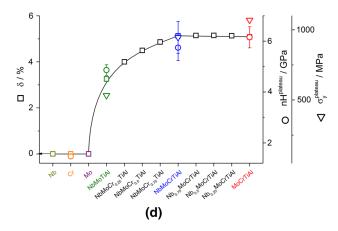


Fig. 6-continued.

ductility might provide chances to assess this in the plateau temperature range. In agreement with the observations from compression testing, the nH data indicate a plateau of the mechanical strength at temperatures higher than approximately 300 °C for the quaternary and quinary alloys. The temperature-dependent strain-rate sensitivity was exemplarily investigated for NbMoCrTiAl in order to verify that the plateau hardness is indeed achieved.

Compared to the rate-sensitive deformation behavior of pure bcc-structured elements in the low-temperature regime, such as Cr with a strain-rate sensitivity of about  $m^{\rm RT}=0.08^{[21,29]}$  at RT, NbMoCrTiAl shows an already, comparably low  $m^{\rm RT}=0.018\pm0.004$  at RT. Nevertheless, as is in the case of pure Cr, [21,29] the strain-rate sensitivity of NbMoCrTiAl almost vanishes approaching the knee temperature:  $m^{300 \text{ °C}} = 0.007 \pm 0.003$ . Thus, clearly, the plateau strength  $\sigma_{\nu}^{\text{plateau}}$  is reached in the temperature range between 300 °C and 600 °C in NbMoCrTiAl. This is in agreement with our observations by compression tests at elevated temperatures. In accordance, the chosen values for nH<sup>plateau</sup> and  $\sigma_{\nu}^{\text{plateau}}$  are included as separate charts in Figures 4(b) and (c), respectively.

The assessment of solid solution strengthening in the last part of the present article will be mainly based on  $\mu H$  measurements at RT  $\mu H^{RT}$  due to their simplicity and reproducibility.  $\mu H^{RT}$  does not significantly depend on alloy batch or varying homogenization parameters. Hence, it is independent of grain size and orientation and RT hardness is reproducible. As shown in Figure 4(b), strength and hardness at RT are already influenced by the thermally activated slip processes. Nevertheless, microhardness  $\mu H^{RT}$  at RT, on the one hand, and  $n H^{\text{plateau}}$  as well as  $\sigma_y^{\text{plateau}}$ , on the other hand, exhibit a very reasonable correlation (compare Figures 4(b) and (c)).

## D. Assessment of Solid Solution Strengthening

By using the atomic radii presented in Figure 3, the atomic size difference  $\delta$  was calculated. While the mean atomic radius, calculated from the lattice parameters presented in Table I, monotonically decreases with increasing Cr content from NbMoTiAl to NbMoCrTiAl

Table II. Relevant Materials Properties for the Evaluation of Composition-Depending Shear Modulus G

| j      | Nominal Composition                       | Annealing Condition            | Porosity   | $\rho  (\mathrm{kg/m^3})$      | $v_{\rm T}~({\rm m/s})$       | G (GPa)                          |
|--------|---|--------------------------------|--|--------------------------------|-------------------------------|----------------------------------|
| 1      | NbMoTiAl                                  | 1500 °C, 20 h                  | $0.0078 \pm 0.0018$  | $6637 \pm 2$                   | $2940 \pm 15$                 | $57.8 \pm 2.2$                   |
| 5<br>6 | NbMoCrTiAl<br>Nb <sub>0.75</sub> MoCrTiAl | 1300 °C, 20 h<br>1300 °C, 20 h | $0.0116 \pm 0.0027$<br>$0.0127 \pm 0.0110$                           | $6594 \pm 5$<br>$6501 \pm 46$  | $3150 \pm 18$<br>$2990 \pm 2$ | $66.2 \pm 1.0$<br>$58.9 \pm 2.5$ |
| 7      | Nb <sub>0.5</sub> MoCrTiAl                | 1300 °C, 20 h                  | $0.0027 \pm 0.0110$ $0.0083 \pm 0.0025$                              | $6420 \pm 16$                  | $3130 \pm 1$                  | $63.4 \pm 3.4$                   |
| 8      | Nb <sub>0.25</sub> MoCrTiAl<br>MoCrTiAl   | 1300 °C, 20 h<br>1200 °C, 20 h | $\begin{array}{c} 0.0080 \pm 0.0045 \\ 0.014 \pm 0.0070 \end{array}$ | $6245 \pm 40$<br>$6165 \pm 57$ | $2880 \pm 1$<br>$2927 \pm 29$ | $52.2 \pm 1.0$<br>$53.6 \pm 0.5$ |
| 9      | MOCITIAL                                  | 1200 C, 20 H                   | $0.014 \pm 0.0070$   | $0103 \pm 37$                  | 2921 ± 29                     | $33.0 \pm 0.3$                   |

and further decreases by decreasing the Nb content from NbMoCrTiAl to MoCrTiAl, the atomic size difference exhibits a different trend. Figure 6(a) shows that for alloys between NbMoTiAl and NbMoCrTiAl, a monotonic increase of  $\delta$  is observed, while it is almost constant for alloys between NbMoCrTiAl and MoCrTiAl. This is caused by the significant difference of the size of Cr compared to the average atomic size of the alloy, while Nb exhibits a size similar to mean atomic radii of the prepared alloys and, thus, provides negligible potential for additional lattice distortion. Nevertheless, all presented alloys based on Nb-Mo-Cr-Ti-Al exhibit a  $\delta$  ranging from 3.25 to 5.15 pct. Thus, a significant variation of lattice distortion for investigating solid solution strengthening is provided. For reference, Cr, Mo, and Nb without any lattice distortion are investigated, too.

For the assessment of the effect of  $\delta$  on the mechanical properties,  $\mu H$  at RT  $\mu H^{RT}$  and plateau yield stress  $\sigma_{\nu}^{\text{plateau}}$  are plotted in a unified way in Figure 6(a). Varying homogenization treatments as well as varying alloy batches (different data points in Figure 6(a); Table I provides details) do not significantly alter the  $\mu$ H of the alloys. For reference,  $\mu$ H<sup>RT</sup> values of Nb, Cr, and Mo are presented as well. Indeed, atomic size difference and µH seem to correlate. In the case of the alloy series ranging from NbMoCrTiAl to MoCrTiAl, where constant mechanical strength should be expected, an excess hardness of about 0.8 to 0.9 GPa above the average µH of NbMoCrTiAl to MoCrTiAl of 6.3 GPa is observed. Staying approximately constant from NbMoTiAl over NbMoCrTiAl towards MoCrTiAl, the gradual change of lattice parameter with composition  $\frac{1}{a}\frac{\partial a}{\partial x_i}$ , serving as a classical measure for the extent of lattice distortion in binary alloys, [4] apparently does not reflect the present evolution of  $\mu$ H.

Figure 6(b) presents the determined correlation of  $\mu H^{RT} = 1.52 \text{ GPa} + 101.56 \text{ GPa} \cdot \delta$  with  $R_{\text{adj}}^2 = 0.98$ .
\*\* As pointed out in Section III,  $\mu H^{RT}$  is altered by a

possibly different hardness increase due to thermally activated slip among the alloy series. For this reason, Figure 6(c) exemplarily incorporates the plateau values for nanohardness  $nH^{plateau}$  as well as yield stress  $\sigma_v^{plateau}$ 

for the quaternary and quinary alloys. Even then, a reasonable correlation of  $\delta$  and mechanical strength is observed for the present HEA system.

As already mentioned, the hardness of  $Nb_xMoCrTiAl$  alloys with x=0.25, 0.5, and 0.75 with almost constant atomic size difference apparently exceeds the average hardness of NbMoCrTiAl and MoCrTiAl. In addition to the severely distorted lattice as a major contribution to solid solution strengthening, additional factors might cause the exceeding  $\mu H$  of  $Nb_{0.25}MoCrTiAl,$   $Nb_{0.5}MoCrTiAl,$  and especially  $Nb_{0.75}MoCrTiAl.$ 

A change in shear modulus, which correlates with the line energy of dislocations, could influence dislocation motion. The most obvious deviations from the correlation of  $\mu H^{RT}$  and  $\delta$  are observed for the alloys with almost constant atomic size difference. Thus, the shear modulus G was determined for NbxMoCrTiAl with x = 0, 0.25, 0.5, 0.75,and 1 and NbMoTiAl also in order to cover the entire composition range of the investigated alloys (Table II). Shear moduli exhibit only minor changes from about 52 GPa to about 66 GPa and do not seem to provide a correlation with the observed RT microhardness  $\mu H^{RT}$ . Hence, we assume that within the Nb-Mo-Cr-Ti-Al system with varying Cr and Nb concentration, the shear modulus does not play a major role for solid solution strengthening. Although no global superstructure was detected by XRD analyses, different types of ordering might additionally strengthen some of the alloys.

## IV. CONCLUSIONS

Providing almost single-phase microstructure, dislocation mediated plasticity with  $\langle 111 \rangle$  Burgers vectors, and varying atomic size differences between 3.25 and 5.15 pct, NbMoCrTiAl and its derivatives are very well suited for investigating the impact of lattice distortion on the solid solution strengthening in concentrated bcc alloys. In the present article, a correlation between the atomic size difference  $\delta$  and RT microhardness  $\mu H^{RT}$ , namely,  $\mu H^{RT}=1.52$  GPa + 101.56 GPa· $\delta$ , was found for a series of alloys Nb-Mo-Cr-Ti-Al, including two of quaternary and one of quinary, equimolar composition. The following conclusions can be drawn.

(1) Bcc solid solutions were stabilized by appropriate heat treatments in all investigated alloys, and competing phases were accordingly suppressed when gradually reducing the concentration of Nb and Cr.

<sup>\*\*</sup> $R_{\mathrm{adj}}^2 = 1 - \left(\frac{(1-R^2)\cdot(n-1)}{n-k-1}\right)$  is a modified coefficient of determination  $R^2$  taking the degrees of freedom k and the number of data points for data fitting n into account. The coefficient of determination  $R^2$  corresponds to the ratio of the covariance and the product of the standard deviations of the fitted data.

- (2) Individual atomic radii of the alloying elements were determined by evaluation of experimentally observed mean atomic radii of all investigated alloys.
- (3) Although  $\mu H^{RT}$  reasonably correlates with the atomic size difference  $\delta$ , the magnitude of  $\mu H^{RT}$  is altered by the well-known feature of thermally activated dislocation slip in bcc metals and solid solutions. Thus, combined temperature-dependent mechanical characterization by both nanoindentation and compression tests was used to exemplarily assess the athermal strength of the equimolar alloys. Both  $\sigma_y^{\text{plateau}}$  and  $nH^{\text{plateau}}$  further support the correlation with  $\delta$  found for  $\mu H^{RT}$ .
- (4) Variation of shear moduli remains small within the composition range of the investigated alloys and cannot account for the observed dependence of the hardness of the alloys.

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